Soil, Groundwater, Surface Water, and Sediments of Kennedy Space Center, Florida: Background Chemical and Physical Characteristics

June 2000



		₩	
		_	

Soil, Groundwater, Surface Water, and Sediments of Kennedy Space Center, Florida: Background Chemical and Physical Characteristics

Prepared by

Paul A. Schmalzer, Melissa A. Hensley, Mario Mota, Carlton R. Hall, and
Colleen A. Dunlevy
Dynamac Corporation
Mail Code DYN-2
Kennedy Space Center, FL 32899

For NASA Environmental Progam Office Mail Code TA-C3 Kennedy Space Center, FL 32899

		
		•
		•
	•	

Table of Contents

Section	Page
Abstract	V
List of Figures	ix
List of Tables	X
List of Acronyms	xiii
Acknowledgements	χV
Executive Summary	xvi
Section 1. Introduction 1.1. Location 1.2. Climate 1.3. Geology 1.4. Soils 1.5. Groundwater 1.6. Surface Water 1.7. Sediment 1.8. Human History 1.9. Reasons for Study	1 1 1 1 3 4 5 6 7 8
Section 2. Soil 2.1. Methods 2.1.1. Site Selection 2.1.2. Sampling 2.1.3. Chemical Analysis 2.1.4. Data Analysis 2.2. Results 2.2.1. Soil Distribution 2.2.2. Soil Chemistry 2.2.3. Soil Physical Parameters 2.3.1. Soil Distribution 2.3.2. Soil Chemistry 2.3.3. Subsurface Soils	14 14 15 15 16 17 17 17 19 20 20 29

Table of Contents (Cont.)

Section	
	Page
Section 3. Groundwater	
3.1. Methods	53
3.1.1. Site Selection	53
3.1.2. Sampling	53
3.1.3. Chemical Analysis	53
3.1.4. Data Analysis	54
3.2. Results	54
3.2.1. Groundwater Chemistry	55
3.3. Discussion	55
- 10000000	56
Section 4. Surface Water	
4.1. Methods	79
4.1.1. Sample Site Selection	79
4.1.2. Sampling	79
4.1.3. Chemical Analysis	79
4.1.4. Data Analysis	80
4.2. Results and Discussion	80
The state and Discussion	81
Section 5. Sediment	
5.1. Methods	106
5.1.1. Sample Site Selection	106
5.1.2. Sampling	106
5.1.3. Chemical Analysis	106
5.1.4. Data Analysis	106
5.2. Results and Discussion	107
The state and Discussion	108
Section 6. Literature Cited	
- India	133

Appendices

Appendix		Page
Appendix A:	KSC Background Screening Tables for Soils, Groundwater, Surface Water, and Sediments of Kennedy Space Center	143
Appendix B:	Soil Data for Kennedy Space Center	164
Appendix C:	Groundwater Data for Kennedy Space Center	254
Appendix D:	Universal Engineering Well Completion Report	280
Appendix E:	Surface water data for Kennedy Space Center	437
• •	Sediment data for Kennedy Space Center	477
	Data validation	513

Abstract

This study documents background chemical composition of soils, groundwater, surface water, and sediments of John F. Kennedy Space Center (KSC). Kennedy Space Center is a complex environment; the chemical composition of these media is a product of geologic history, climate, interactions with biota, and anthropogenic effects such as citrus agriculture and ranching.

Fifty-eight soil series and land types occur on KSC. We developed a classification of ten soil classes: coastal, acid scrub, coquina scrub, flatwoods, hammocks, freshwater wetland, saltwater wetlands, citrus scrub, citrus hammock, and disturbed. These classes represent major divisions including upland (well-drained, poorly drained, acid, circumneutral, alkaline), wetland (freshwater, saltwater), agriculture, and disturbed. We selected twenty sample locations in each soil class, a total of 200 locations. A 0-30 cm layer composite sample was collected for analysis. Samples were also collected for bulk density and soil texture analyses. Standard sampling protocols were followed.

Soil samples were analyzed for organochlorine pesticides, aroclors, chlorinated herbicides, polycyclic aromatic hydrocarbons (PAH), total metals, pH, cation exchange capacity (CEC), bulk density, resistivity, and soil texture. Sixteen organochlorine pesticides, all (6) aroclors, 15 chlorinated herbicides, one PAH, and four metals were below detection in all samples. Frequencies of organochlorine pesticides above detection limits in all soils were low; however, frequencies were higher in agricultural (citrus scrub, citrus hammock), disturbed, and saltwater wetland soil classes. This probably reflects past use for mosquito control and in citrus agriculture. Frequencies of chlorinated herbicides above detection limits were low. PAHs were above detection limits in relatively few soil samples; frequencies were greater in agricultural (citrus scrub, citrus hammock) and disturbed soil classes. However, PAH values above detection occurred in other soil classes. PAHs have both natural and anthropogenic origins. Among the metals, Ba, Cd, Se, Ag, and TI occurred in few samples above detection limits. As, Cu, and Hg were above detection limits with an intermediate frequency, while the other metals were frequently above detection limits. Kruskal-Wallis tests indicated that all soil parameters differed among soil classes (p< 0.001). However, given the low frequency of detection of organics and some metals, analysis of variance (ANOVA) was conducted only for metals where >25 % of values were above detection and for pH, bulk density, resistivity, and CEC. All metals analyzed (Al, As, Ca, Cr, Cu, Fe, Pb, Mg, Mn, Hg, K, Na, V, Zn) were significantly different among types (p < 0.001). Patterns of differences varied among metals. Metal concentrations in KSC soils reflected both the origins of the soils and subsequent changes from weathering and leaching. Concentrations of most metals were low compared to other regions. Cr, Cu, Mn, and Zn had higher values in agricultural soils probably from sources in fertilizers or pesticides. Soil pH was highest in coastal soils and lowest in flatwoods. Coastal, acid scrub, and coquina scrub soils had very low CEC values consistent

with the low organic matter content of these soils; values in hammocks and wetlands, where organic matter accumulates, were substantially higher. Bulk density was low where organic matter accumulates in surface soils. Soil texture analysis indicated that coastal and coquina scrub soil classes were medium sands in the Wentworth classification and the other soil classes were fine sands.

The aquifer of primary concern for contamination issues on KSC is the Surficial aquifer. The Surficial aquifer can be divided into four subsystems: Dune, Dune-Swale, West Plain, and Marsh that differ in area, amount of recharge, and degree of interaction with saline water of the Indian River Lagoon and Atlantic Ocean.

We located six sample sites in each subsystem, 24 total sites. The sampling plan designated that a shallow well (4.6 m) was to be installed at each site. Intermediate wells (10.7 m) were to be installed at four sites per subsystem (16 total); deep wells (15.2 m) were to be installed at three sites per subsystem (12 total). A total of 52 wells were planned. Due to the depth of the confining unit at one location, the deep well was not installed there. Therefore, a total of 51 wells were installed at varying depths. Groundwater samples were collected using standard protocols. Groundwater samples were analyzed for organochlorine pesticides, aroclors, chlorinated herbicides, PAH, total metals, dissolved oxygen (DO), turbidity, pH, specific conductivity, temperature, total dissolved solids (TDS), and total organic carbon (TOC).

All organochlorine pesticides (25), all aroclors (6), and all chlorinated herbicides (18) were below detection in all samples. Ten PAHs were above detection in one to four samples. Co and Hg were always below detection. Be, Ag, and Zn were above detection in just one sample each. For some parameters, frequency of occurrence above detection varied with subaquifer or depth. PAHs were above detection only in shallow wells of the Dune, Dune-Swale, and West Plain subaquifers. As and Cd were most frequently above detection in the Dune subaquifer.

Among metals only Al, Ca, Cl, Fe, Mg, Mn, K, and Na were above detection in sufficient samples to compare concentrations. In addition, TDS, TOC, specific conductivity, pH, DO, and field temperature could be compared. Univariate, two-way ANOVA where subaquifer and depth were fixed effects was conducted using temperature and log-transformed chemistry data. Models were not significant for Al and DO; the significance level for the model (p=0.057) for Fe was marginal. Interactions were significant for pH. TOC differed across subaquifers but not depths. Ca, Cl, Mg, Mn, K, Na, TDS, and conductivity differed with subaquifer and depth.

For screening purposes, a simpler classification of groundwater was desired. The regulatory criteria (Florida Department of Environmental Protection) for dividing groundwater into classes G2, with total dissolved solids <10,000 mg/L, and G3, with total dissolved solids >10,000 mg/L, were adopted. Some

parameters were more frequently above detection in class G2 (Cd, Fe, Tl) and some were more frequently above detection in class G3 (Sb, As, Pb). Ca, Cl, Mg, K, Na, total dissolved solids, and conductivity were substantially higher in class G3, as would be expected.

The baseline data suggest that widespread contamination of the Surficial aquifer on KSC has not occurred. The chemical parameters varying most with subaquifer and depth were Ca, Cl, Mg, K, and Na, and conductivity and total dissolved solids that are related to these cations and anions. Concentrations increased with depth within a subaquifer. At a given depth, concentrations in the Dune-Swale and West Plain subaquifers were lower than in the Dune and Marsh subaquifers. These trends reflect increased mineralization with depth and differences between the fresh water Dune-Swale and West Plain subaquifers and the more saline Dune and Marsh systems.

KSC is surrounded by the Indian River Lagoon System that extends along the East Coast of Florida from Ponce de Leon Inlet to St. Lucie Inlet near Stuart. This system includes the Indian River, Banana River, and Mosquito Lagoon. These basins are shallow, aeolian, lagoons with depths averaging 1.5 m; maximums of 9 m are restricted generally to dredged basins and channels.

Location of the surface water sampling stations was determined based on the watershed basins. Forty stations were selected to incorporate samples from open lagoonal water, rivers, creeks, ditches, borrow pits, and impoundments. Samples were collected using standard sampling protocols. Basins included Banana Creek, Banana River, Indian River Lagoon, Mosquito Lagoon, saline ditches (salinity > 6 ppt), and freshwater ditches (salinity < 6 ppt).

Surface water was analyzed for organochlorine pesticides, aroclors, chlorinated herbicides, polycyclic aromatic hydrocarbons, and metals. Field parameters such as pH, temperature, turbidity, DO, and conductivity were also measured at each sampling location.

All of the aroclors (6) and chlorinated herbicides (18) were below detection. One of 25 organochlorine pesticides (Dieldrin) was above detection as were five of 17 PAHs. The occurrence of Dieldrin is probably related to past agricultural use. Concentrations of PAHs were low; these may result from natural sources or regional deposition. Sixteen of 24 metals were above detection limits; eight (Ba, Cd, Cr, Co, Hg, Ni, Vn, and Zn) were always below detection. Two PAHs (naphthalene, fluorene) and one metal (Cu) were above detection in only one sample. Nine metals (Sb, As, Be, Cu, Pb, Mn, Se, Ag, and Tl) were above detection in too few samples to test for differences among watershed basins. Seven metals commonly above detection limits (Al, Ca, Cl, Mg, Fe, K, and Na) differed among basins (ANOVA, p<0.05).

Sediment sampling stations were co-located with surface water sampling locations. A total of 40 stations were selected to incorporate samples from open lagoonal water, rivers, creeks, ditches, burrow pits, and impoundments. Stations included 18 ditches with five located near KSC's industrial areas.

Sediments were analyzed for for organochlorine pesticides, aroclors, chlorinated herbicides, PAH, and metals. Other parameters such as percent solids, pH, resistivity, texture, and TOC were also analyzed.

All of the aroclors (6), 16 of 18 chlorinated herbicides, 24 of 25 organochlorine pesticides, 6 of 18 PAHs, and 7 of 23 metals were always below detection. Chlorinated herbicides, organochlorine pesticides, and PAHs that were above detection were at low to very low frequencies. Ten metals (Al, Ca, Cr, Fe, Pb, Mg, Mn, K, Na, and V) were above detection in sufficient samples to analyze for difference among watershed basins; none were significant (ANOVA, p < 0.05).

An extensive evaluation and validation of the data was conducted. Metal analyses were of high quality with few problems noted. Organic analyses were more variable than metals. However, the organic analyses results were of generally good quality, and most organics were below detection.

List of Figures

Clavino	
Figure	Page
Figure 1.1. Location of Kennedy Space Center.	10
Figure 1.2. East-west geologic cross section, Kennedy Space Center.	11
Figure 1.3. Geohydrologic units on Kennedy Space Center.	12
Figure 1.4. Groundwater subaquifers of Kennedy Space Center.	13
Figure 2.1. Soil sampling locations, Kennedy Space Center.	30
Figure 2.2. Distribution of soil classes on Kennedy Space Center.	31
Figure 2.3. Phi size distribution for coastal soils.	32
Figure 3.1. Groundwater well sampling locations, Kennedy Space Center.	58
Figure 4.1. Basins, surface water, and sediment sampling locations	97
Figure 4.2. Surface water and sediment sampling locations for the KSC Background Study.	98

List of Tables

Table		Page
Table 2.1.	Soil classification for Kennedy Space Center.	33
Table 2.2.	List of parameters, the EPA method numbers, and detection limits, used in soil samples collected for the KSC Baseline Study.	35
Table 2.3.	Area of soil classes.	38
Table 2.4.	Parameters below detection limits (non-detect) in all soil samples.	39
Table 2.5.	Frequency (percent) of values greater than detection limits for chemical parameters in soils.	40
Table 2.6.	Parameters that were determined by the Remediation team to be treated as below detection limits (non-detect) for all soil locations based on low frequency of detection.	43
Table 2.7.	Means and standard deviations (in parenthesis) of chemical parameters in soils.	444
Table 2.8.	Analysis of variance of soil metals among soil classes.	48
Table 2.9.	Analysis of variance among soil classes for pH, bulk density, resistivity, and CEC.	49
Table 2.10.	Mean grain size and classification of soil types.	50
Table 2.11.	Concentrations of selected elements in typical surface soils of Florida.	51
Table 2.12.	Background levels of trace elements in surface soils.	52
Table 3.1.	List of parameters, EPA method number, and lab reporting limit for parameters analyzed for all groundwater samples.	59
Table 3.2.	Frequency (percent) of values greater than detection limits for chemical parameters in groundwater.	62
Table 3.3.	List of parameters that were below detection limits (non-detect) for all groundwater wells.	65

List of Tables (continued)

Table	(continued)	
·		Page
Table 3.	4. List of parameters that were determined by Remediation team consensus to be treated as below detection limits (non-detect)for all groundwater locations based on low frequency of detection.	669
Table 3.5	5. Frequency (percent) of values greater than detection limits for chemical parameters in groundwater by subaquifer and depth.	67
Table 3.6	 Chemical parameters in groundwater by subaquifer and depth. 	69
Table 3.7	 Summary of two-way analysis of variance of selected groundwater parameters. 	73
Table 3.8	Frequency (percent) of values greater than detection limits for chemical parameters in groundwater with one or more samples above detection limits by groundwater class.	74
Table 3.9.	Means and standard deviations (in parenthesis) for chemical parameters in groundwater by groundwater class.	75
Table 3.10	 Groundwater chemistry data for the Surficial Aquifer on KSC from Clark (1987c). 	77
Table 4.1.	List of parameters, EPA methods, and detection limits used to analyze surface water samples collected for the 1998 KSC Background Study. * = measurement made with a calibrated field instrument (YSI).	99
Table 4.2.	Name, number of hits, and frequency of analytes occurring above detection limits. All others were not detected. For screening table values, the three parameters with only one hit out of 46 samples were considered as non-detect by the NASA/KSC Remediation Program Team.	102
Table 4.3.	List of parameters that were below detection limits (non-detect) for all surface water locations.	105
Table 5.1.	List of parameters, the EPA methods used to analyze each, and detection limits, in sediment samples collected for the KSC Background Study.	125

List of Tables (continued)

Table	`	age
Table 5.2.	Name, number of hits, and frequency of analytes occurring above detection limits occuring in sediment samples. All parameters not listed were not detected. For screening table values, the four parameters with only one hit out of 46 samples were considered as non-detect by the NASA/KSC Remediation Program Team.	128
Table 5.3.	Parameters that were below detection limits (non-detect) in all sediment samples.	131
Table 5.4.	Parameters that were determined by the NASA/KSC Remediation team to be below detection limits (non-detect) for all sediment samples based on low frequency of detection.	132

List of Acronyms

Ag Silver Al Aluminum

ANOVA Analysis of variance

As Arsenic Ba Barium

BDDP Baseline Data Development Program

Be Beryllium Ca Calcium

CCAFS Cape Canaveral Air Force Station

CCF Component Cleaning Facility

Cd Cadmium

CEC Cation exchange capacity
CNS Canaveral National Seashore

Co Cobalt

COPC Chemical of Potential Concern

Cr Chromium Cu Copper

DO Dissolved oxygen DS Dune- Swale

EPA U. S. Environmental Protection Agency

F.A.C. Florida Administrative Code

FDEP Florida Department of Environmental Protection

Fe Iron

FWS Fish and Wildlife Service

GIS Geographic Information System
GPS Global Positioning System

Hg Mercury

IRL Indian River Lagoon

ITL Integrate, Transfer, and Launch area (on CCAFS)

K Potassium

KSC John F. Kennedy Space Center

Mg Magnesium

MINWR Merritt Island National Wildlife Refuge

Mn Manganese Na Sodium

NASA National Aeronautics and Space Administration

Ni Nickel

NPS National Park Service

PAPI Precision Approach Path Indicator

Pb Lead

PC Personal computer

PRL Potential Release Location

PS7 Pump Station 7

List of Acronyms (continued)

RCRA Resource Conservation and Recovery Act

SAP "NASA/KSC Sampling and Analysis Plan for RCRA Facility

Investigations at Kennedy Space Center, Florida and Cape Canaveral Air Station, Florida, Volume 4" of the Generic Work

Plans

Sb Antimony

Se Selenium

SLF Shuttle Landing Facility

SWMU Solid Waste Management Unit

TDS Total dissolved solids TOC Total Organic Carbon

TI Thallium V Vanadium

VAB Vertical Assembly Building

yr B P years before present

Zn Zinc

Acknowledgements

This study was conducted under NASA contract NAS10-12180 and supported by the NASA Environmental Program Office Remediation Project. We thank Mike Anders and Vickie Anders of ENCO for field sampling, Russ Erickson and Marcia Terlep of ENCO for chemistry and database information, and Venu Oddiraju and Billy Paine for database assistance. We thank Rosaly Santos-Ebaugh, Harold Williams, Burton Summerfield, and Dr. William Knott of the NASA Biomedical Office for their assistance. Trade names and commercial enterprises or products are mentioned only for information. No endorsement by NASA is implied.

Executive Summary

This study documents background chemical composition of soils, groundwater, surface water, and sediments of John F. Kennedy Space Center (KSC). Kennedy Space Center is a complex environment; the chemical composition of these media is a product of geologic history, climate, interactions with biota, and anthropogenic effects such as citrus agriculture and ranching.

KSC has been the primary launch complex for the United States Space program since the 1960's. In this time, many industrial practices have required use of hazardous materials; these uses have lead to potential contamination concerns center-wide. Currently, data collected during investigation of areas that may have been contaminated are compared to regulatory, risk-based screening values. The purpose of this study was to develop the baseline (background) values specific to KSC. Current screening values may be extremely conservative and may actually represent values that are lower than naturally exist within this landscape.

Soil

The soils of KSC are mapped in the soil surveys for Brevard County (Huckle et al. 1974) and Volusia County (Baldwin et al. 1980), and the resulting soil pattern is complex. Fifty-eight soil series and land types are represented. The primary source of parent material for KSC soils are sands of mixed terrestrial and biogenic origin. Soils differ in age across the KSC landscape. Differences in soil drainage affect the accumulation of organic matter, which influences many soil properties. In addition, proximity to the lagoon systems influences soil salinity.

We divided soils first into four groups: Upland, Wetland, Agricultural, and Disturbed. We divided Upland soils into well-drained and poorly drained categories. We divided well-drained, upland soils into three classes: 1) geologically recent, alkaline, sandy soils of coastal dunes where the vegetation is coastal dunes, coastal strand, or coastal scrub; 2) old, inland, leached, acid, sandy soils where the vegetation is oak-saw palmetto scrub or scrubby flatwoods; and 3) inland, circumneutral soils formed over coquina where the vegetation is oak-saw palmetto scrub or xeric hammock. We divided poorlydrained, upland soils into two classes: 1) acid, sandy soils with flatwoods vegetation; and 2) circumneutral to alkaline soils formed over coquina or limestone where the vegetation is mesic hammock. The primary division of wetland soils was between: 1) inland, freshwater wetlands where the vegetation was freshwater marshes or hardwood swamps; and 2) coastal, brackish to saline wetlands where the vegetation was salt marshes or mangroves. Agricultural soils were of two types: 1) active or abandoned citrus on scrub soils; and 2) active or abandoned citrus on hammock soils. Disturbed soils included various types modified by construction. This group could be heterogeneous, but there was no apparent division into homogeneous subgroups.

This classification resulted in ten groups. We selected twenty sample locations in each soil class, a total of 200 locations. A 0-30 cm layer composite sample was collected for analysis. Samples were also collected for bulk density and soil texture analyses. Standard sampling protocols were followed.

Samples were analyzed for organochlorine pesticides, aroclors, chlorinated herbicides, polycyclic aromatic hydrocarbons (PAH), total metals, pH, cation exchange capacity (CEC), bulk density, resistivity, and soil texture.

Sixteen organochlorine pesticides, all (6) aroclors, 15 chlorinated herbicides, one PAH, and four metals were below detection in all samples. Frequencies of organochlorine pesticides above detection limits in all soils were low; however, frequencies were higher in agricultural (citrus scrub, citrus hammock), disturbed, and saltwater wetland soil classes. This probably reflects past use for mosquito control and in citrus agriculture.

Frequencies of chlorinated herbicides above detection limits were low; frequencies were slightly higher in agricultural (citrus scrub, citrus hammock), disturbed, and saltwater wetland soil classes. The low frequency of detection of chlorinated herbicides indicates that contamination by them has not been widespread in the KSC environment.

PAHs were above detection limits in relatively few soil samples; frequencies were greater in agricultural (citrus scrub, citrus hammock) and disturbed soil classes. However, PAH values above detection occurred in other soil classes. PAH have both natural and anthropogenic origins. Natural sources include synthesis by microorganisms, phytoplankton, algae, and higher plants, wildfires, volcanic activity, and sediment diagenesis. PAH are produced by a variety of industrial combustion processes.

Among the metals, Ba, Cd, Se, Ag, and Tl occurred in few samples above detection limits. As, Cu, and Hg were above detection limits with an intermediate frequency, while the other metals were frequently above detection limits.

Kruskal-Wallis tests indicated that all soil parameters differed among soil classes (p< 0.001). However, given the low frequency of detection of organics and some metals, analysis of variance (ANOVA) was conducted only for metals where >25 % of values were above detection and for pH, bulk density, resistivity, and CEC. All metals analyzed (AI, As, Ca, Cr, Cu, Fe, Pb, Mg, Mn, Hg, K, Na, V, Zn) were significantly different among types (p < 0.001). Variances were not homogeneous (Levene test, p<0.05) except for Zn; thus, the Games-Howell test was appropriate for examining multiple comparisons.

Patterns of differences varied among metals. At had low values in flatwoods, acid scrub, and coastal soils and higher values in other types. As had higher

values in coastal and saltwater wetland soils and lower values elsewhere. Ca was high in coastal and disturbed soils, low in acid scrub and flatwoods, and intermediate in other types. Cr, Cu, Mn, and Zn had higher values in citrus scrub and citrus hammock soils and lower values elsewhere. Fe was low in acid scrub and flatwoods soils and higher in other types. Pb values were low in acid scrub, coastal, and flatwoods soils, high in saltwater wetlands, and intermediate in other types. Mg, K, and Na levels were lowest in acid scrub and highest in saltwater wetlands. Hg was low in acid scrub and coastal soils and high in hammocks and freshwater wetlands. V was low in acid scrub and flatwoods, high in saltwater wetlands, and intermediate in other types.

Metal concentrations in KSC soils reflect both the origins of the soils and subsequent changes from weathering and leaching. Concentrations of most metals were low compared to other regions. Cr, Cu, Mn, and Zn had higher values in agricultural soils probably from sources in fertilizers or pesticides.

Soil pH was highest in coastal soils and lowest in flatwoods. Calcium carbonate shell material is abundant in coastal soils making them alkaline, while other types are slightly to strongly acid. Leaching of basic cations from older soils has contributed to their acidity. Coquina scrub soils that have shell material deeper in the soil profile were less acid than acid scrub soils that lack this buffering. The agricultural types (citrus scrub, citrus hammock) were less acidic than the original soil types. This may be due to liming (Brady 1974) or to irrigation with groundwater high in carbonates.

Coastal, acid scrub, and coquina scrub soils had very low CEC values consistent with the low organic matter content of these soils; values in hammocks and wetlands, where organic matter accumulates, were substantially higher. Bulk density was low in flatwoods, hammock, freshwater wetland, and saltwater wetland soils where organic matter accumulates in surface soils and higher in the other types with little organic matter accumulation.

Resistivity was low in saltwater wetland soils that had high concentrations of Na and K as well as CI.

Soil texture analysis indicated that coastal and coquina scrub soil classes were medium sands in the Wentworth classification and the other soil classes were fine sands. Shell fragments in coastal and coquina scrub soils result in larger mean grain size than other soils that are composed mainly of quartz grains.

KSC Background screening values for soil are given in Appendix A, Table A-1. Screening values are calculated as twice the mean. Values are given for all soils and by soil classes (11 classes total). The NASA/KSC Remediation Team decision for screening allows only inorganic data to be used as a screening tool. Organic data should be discussed in the uncertainty section of reports. Combined values for pesticides (as noted on table) can only be used for the

following classes: saltwater wetland, citrus hammock, citrus scrub, and disturbed. To use the screening tables: 1) screen site data against the combined mean; 2) if the site value is higher than the combined mean, go to the soil class that corresponds to site data sampling locations. Figure 2.2 illustrates KSC background classification information. Individual site maps can be obtained, upon reqest, from NASA support contractor when additional site details are needed.

Groundwater

Several groundwater aquifers occur beneath the surface of KSC (Clark, 1987c). The aquifer of primary concern for contamination issues is the Surficial aquifer that occurs in the saturated part of the moderately permeable Pleistocene and Holocene deposits of fine to medium sand, shell, coquina, silts, and marl. The Surficial aquifer can be divided into four subsystems: Dune, Dune-Swale, West Plain, and Marsh. These subaquifers differ in area, amount of recharge, and degree of interaction with saline water of the Indian River Lagoon and Atlantic Ocean.

We located six sample sites in each subsystem, 24 total sites. The sampling plan designated that a shallow well (4.6 m) was to be installed at each site. Intermediate wells (10.7 m) were to be installed at four sites per subsystem (16 total); deep wells (15.2 m) were to be installed at three sites per subsystem (12 total). A total of 52 wells were planned. Due to the depth of the confining unit at one location, the deep well was not installed there. Therefore, a total of 51 wells were installed at varying depths.

Groundwater samples were collected using standard protocols. Groundwater samples were analyzed for organochlorine pesticides, aroclors, chlorinated herbicides, PAH, total metals, dissolved oxygen (DO), turbidity, pH, specific conductivity, temperature, total dissolved solids (TDS), and total organic carbon (TOC).

All organochlorine pesticides (25), all aroclors (6), and all chlorinated herbicides (18) were below detection in all samples. Ten PAHs were above detection in one to four samples. Four PAHs were above detection limits in only one sample each, and these values were between the method detection limit and the practical quantitation level; therefore, the Remediation team decided to treat these as below detection. Co and Hg were always below detection. Be, Ag, and Zn were above detection in just one sample each.

For some parameters, frequency of occurrence above detection varied with subaquifer or depth. PAHs were above detection only in shallow wells of the Dune, Dune-Swale, and West Plain subaquifers. As and Cd were most frequently above detection in the Dune subaquifer.

Among metals only AI, Ca, CI, Fe, Mg, Mn, K, and Na were above detection in sufficient samples to compare concentrations. In addition, total dissolved solids (TDS), total organic carbon (TOC), specific conductivity, pH, dissolved oxygen (DO), and field temperature could be compared.

Univariate, two-way ANOVA where subaquifer and depth were fixed effects was conducted using temperature and log-transformed chemistry data. Models were not significant for Al and DO; the significance level for the model (p=0.057) for Fe was marginal. Interactions were significant for pH. TOC differed across subaquifers but not depths. Ca, Cl, Mg, Mn, K, Na, TDS, and conductivity differed with subaquifer and depth.

For screening purposes, a simpler classification of groundwater was desired. The regulatory criteria (Florida Department of Environmental Protection [FDEP]) for dividing groundwater into classes G2, with TDS <10000 mg/L, and G3, with TDS >10000 mg/L, were adopted. Some parameters were more frequently above detection in class G2 (Cd, Fe, Tl) and some were more frequently above detection in class G3 (Sb, As, Pb). Ca, Cl, Mg, K, Na, TDS, and conductivity were substantially higher in class G3, as would be expected.

The baseline data suggest that widespread contamination of the Surficial aquifer on KSC has not occurred. No organochlorine pesticides, aroclors, or chlorinated herbicides occurred above detection limits. Although pesticide residues or degradation products and chlorinated herbicides occurred in some soils, those concentrations were low and migration into the aquifer either has not occurred or has not been widespread. Some PAHs occurred in the shallow wells. PAHs occur in a variety of KSC soils at relatively low concentrations. Some occurrence in shallow wells is not surprising. PAHs have both natural and anthropogenic sources (e.g., Suess 1976, Standley and Simoneit 1987, Jones et al. 1989b, c).

Most trace metals were in low concentrations in KSC groundwater if they occur above detection levels. This is consistent with the low concentrations of most trace metals in KSC soils and the primarily quartz composition of the terrigenous deposits comprising the surficial sediments of Merritt Island (Brown et al. 1962, Milliman 1972, Field and Duane 1974). Al, Fe, and Mn occurred above detection limits more frequently than other trace metals. Al and Fe are abundant crustal components and are present in KSC soils. Intense leaching, particularly in acid scrub and flatwoods soils, mobilizes Al and Fe (Paton et al. 1995). Iron is a typical constituent of groundwater in the Surficial aquifer in Florida (Miller 1997). Mn is one of the most abundant trace elements (Kabata-Pendias and Pendias 1984); it is present in KSC soils but the concentrations are relatively low. Solution and precipitation of Fe and Mn are affected by pH and oxidation-reduction conditions.

The chemical parameters varying most with subaquifer and depth were Ca, Cl, Mg, K, and Na, and conductivity and TDS that are related to these cations and

anions. The trends were generally consistent among these; the shallow wells in the Dune-Swale subaquifer had the lowest values. Concentrations increased with depth within a subaquifer. At a given depth, concentrations in the Dune-Swale and West Plain subaquifers were lower than in the Dune and Marsh subaquifers. These trends reflect increased mineralization with depth and differences between the freshwater Dune-Swale and West Plain subaquifers and the more saline Dune and Marsh systems. The Dune and Marsh subaquifers interact with saline water of the Atlantic Ocean and Indian River Lagoon system, respectively (Clark 1987c).

The screening table for groundwater is given in Appendix A, Table A-2. Screening values are calculated as twice the mean. The NASA/KSC Remediation Team decision on groundwater was to separate the groundwater into FDEP G2 and G3 groundwater classes. To use the screening table for groundwater: 1) collect TDS data at all groundwater sampling locations; 2) determine proper groundwater class for site data based on TDS data collected. FDEP G2 groundwater class is TDS less than 10,000 mg/L. FDEP G3 groundwater class is TDS greater than 10,000 mg/L.

Surface Water

KSC is surrounded by the Indian River Lagoon (IRL) System that extends along the East Coast of Florida from Ponce de Leon Inlet to St. Lucie Inlet near Stuart. This is system includes the Indian River, Banana River, and Mosquito Lagoon. These basins are shallow, aeolian, lagoons with depths averaging 1.5 m and maximums of 9 m, which are generally restricted to dredged basins and channels. The major freshwater body within KSC property is Banana Creek, which drains the numerous estuaries adjacent to the Space Shuttle launch pads.

Location of the surface water sampling stations was determined based on the watershed basins. Forty stations were selected to incorporate samples from open lagoonal water, rivers, creeks, ditches, borrow pits, and impoundments. Samples were collected using standard sampling protocols. Basins included Banana Creek, Banana River, Indian River Lagoon, Mosquito Lagoon, saline ditches (salinity > 6 ppt), and freshwater ditches (salinity < 6 ppt).

Surface water was analyzed for organochlorine pesticides, aroclors, chlorinated herbicides, PAH, and metals. Field parameters such as pH, temperature, turbidity, DO, and conductivity were also measured at each sampling location.

All of the aroclors (6) and chlorinated herbicides (18) were below detection. One of 25 organochlorine pesticides (Dieldrin) was above detection as were five of 17 PAHs. Sixteen of 24 metals were above detection limits; eight (Ba, Cd, Cr, Co, Hg, Ni, Vn, and Zn) were always below detection. Two PAHs (naphthalene, fluorene) and one metal (Cu) were above detection in only one sample. These were treated as always below detection by the Remediation team.

The occurrence of Dieldrin is probably related to past agricultural use. Concentrations of PAHs were low; these may result from natural sources or regional deposition.

Nine metals (Sb, As, Be, Cu, Pb, Mn, Se, Ag, and Tl) were above detection in too few samples to test for differences among watershed basins. Seven metals commonly above detection limits (Al, Ca, Cl, Mg, Fe, K, and Na) differed among basins (ANOVA, p<0.05). Patterns of differences varied among metals. For Al, Banana Creek was higher than the other basins. Fe was higher in Banana Creek, saline ditches, and freshwater ditches compared to Banana River, Indian River Lagoon, and Mosquito Lagoon. Values of Ca, Cl⁻, and Mg occurred in three classes with Banana Creek, Mosquito Lagoon, and Indian River Lagoon the highest, Banana River and saline ditches intermediate, and freshwater ditches low. K was highest in Mosquito Lagoon, intermediate in Banana Creek, Indian River Lagoon, Banana River, and saline ditches, and lowest in freshwater ditches. Na was highest in Mosquito Lagoon and the Indian River Lagoon, intermediate in Banana Creek, Banana River, and saline ditches, and lowest in freshwater ditches.

Screening tables for surface water are given in Appendix A, Table A-3. Screening values are calculated as twice the mean. The screening table for surface water has eight classes. To use the screening table for surface water: 1) collect salinity measurements at all surface water sampling locations; 2) compare site sample data to the surface water combined values based on the appropriate salinity ranges. If the site value is higher than the KSC Background combined salinity value, go to the appropriate surface water basin class for site data in the surface water table and compare site value to that.

Sediment

Approximately 90% of the IRL bottom area is believed to be flat, exposed substrate, and consist primarily of sand and shell fragments (Gilmore 1977). A thin layer of silt often overlays the bottom in the more open portions of the estuary while a surficial layer of organic detrital material is usually present in portions adjacent to shore areas (Woodward-Clyde Consultants 1994). However, fine-grained, organic-rich mucky sediment occurs in about 10% of the Lagoon in deposits ranging up to more than 2 m thick.

Sediments may have been affected by anthoropogenic modifications including dredging the Intercoastal Waterway and Haulover Canal, stabilization of inlets, construction of causeways, land filling, agricultural development, diking, impoundment for mosquito control, and modifications related to the space program.

Sediment sampling stations were co-located with surface water sampling locations. A total of 40 stations were selected to incorporate samples from open lagoonal water, rivers, creeks, ditches, burrow pits, and impoundments. Stations included 18 ditches with five located near KSC's industrial areas.

Sediments were analyzed for organochlorine pesticides, aroclors, chlorinated herbicides, PAH, and metals. Other parameters such as percent solids, pH, resistivity, texture, and TOC were also analyzed.

All of the aroclors (6), 16 of 18 chlorinated herbicides, 24 of 25 organochlorine pesticides, 6 of 18 PAHs, and 7 of 23 metals were always below detection. Chlorinated herbicides, organochlorine pesticides, and PAHs that were above detection were at low to very low frequencies. Ten metals (Al, Ca, Cr, Fe, Pb, Mg, Mn, K, Na, and V) were above detection in sufficient samples to analyze for difference among watershed basins; none were significant (ANOVA, p < 0.05).

The screening table for sediment is given in Appendix A, Table A-4. Screening values are calculated as twice the mean. There is only one class for sediment.

Database Evaluation and Validation

An extensive evaluation and validation of the database was conducted. Data collection efficiency was 99.99%. Field methods followed Florida Department of Environmental Protection (DEP) and NASA/KSC procedures. All analyses used methods approved by the Environmental Protection Agency (EPA). All sampleholding times were met. Chain of custody records were maintained with only minor errors. Results of equipment blank analyses indicated no significant procedural contamination of field samples. Results of method blank analyses indicated possible metal contamination in five sample runs. Low level sodium and calcium contamination did not meaningfully impact estimated means. Aluminum contamination in the laboratory may have produced a positive 4.25% bias in the estimated mean for coastal soils. Assessment of surrogate and spike recovery indicated good laboratory quality control. Metal analyses were of high quality with few problems noted. Matrix interferences were an issue in several coastal soil samples. Organic analyses were more variable than metals; this is expected given the more complex methodologies. However, the organic analyses results were of generally good quality, and most organics were below detection.

SECTION 1 Introduction

This report documents background chemical composition of pristine soils, groundwater, surface water, and sediments of John F. Kennedy Space Center (KSC). Kennedy Space Center is a complex environment; the chemical composition of these media is a product of geologic history, climate, interactions with biota, and anthropogenic effects (such as citrus, agriculture, and ranching).

The Landscape of Kennedy Space Center

1.1. Location

Kennedy Space Center is located on the east coast of central Florida in Brevard County, at approximately 28.5 degrees north latitude and 81.7 degrees west longitude (Figure 1.1). KSC is located on Merritt Island, a relict barrier island. Cape Canaveral Air Force Station (CCAFS) occupies the adjacent Cape Canaveral barrier island. Large water bodies surround these lands: the Atlantic Ocean, Banana River, Indian River, and Mosquito Lagoon.

1.2. Climate

Merritt Island experiences a warm, humid climate. Long-term precipitation records (1888-1987) give a mean annual precipitation of 136.6 cm (53.8 in) for Titusville, but annual rainfall varies widely (low: 84.8 cm – high: 207.5 cm) (Mailander, 1990). There is also seasonal variation in precipitation with a wet season from May to October and the rest of the year relatively dry. January is typically the coldest month with a mean minimum temperature of 9.6 C (49.2 F) and a mean maximum temperature of 22.3 C (72.2 F). July is typically the hottest month with a mean minimum temperature of 71.4 F (21.9 C) and a mean maximum temperature of 33.3 C (91.9 F) (Mailander 1990).

1.3. Geology

Florida has a complex geologic history with repeated periods of deposition when the Florida Plateau was submerged and erosion when the seas recessed (Randazzo 1997, Scott 1997). The oldest formation known to occur beneath Brevard County and KSC, the Avon Park limestone, was deposited in the early Eocene in an open ocean (Cooke 1945). This was followed by a withdrawal of the sea and a period of erosion. In the late Eocene, the seas advanced and limestones of the Ocala group were deposited (Cooke 1945). Following another period of recession of the sea and erosion of the land surface, the Hawthorn formation of calcareous clay, phosphatic limestone, phosphorite, and radiolarian clay was deposited in the late Miocene (Cooke 1945, Brown et al. 1962). Overlying this are unconsolidated beds of fine sand, shells, clay, and calcareous clay of late Miocene or Pliocene age (Brown et al. 1962). Surface strata in

Brevard County are primarily unconsolidated white to brown quartz sand containing beds of sandy coquina of Pleistocene and Holocene age (Brown et al. 1962). See Figure 1.2.

In addition to the sequences of sediments of varying age, the surface of Florida is marked by a series of terraces and former shorelines of varying ages. In Brevard County, two terraces have been recognized, the Silver Bluff Terrace on Cape Canaveral and Merritt Island and the Pamlico Terrace on the mainland (Healy 1975).

During the Pleistocene (ca. 1.6 million years before present [yr B.P.] to 13,000 yr B.P.), repeated glaciation of the northern hemisphere produced fluctuations in sea level (Bowen 1978). At the maximum of the Wisconsinan glaciation (ca. 18,000 yr B.P.), sea levels were on the order of 100 m lower than at present, and substantial additional areas were exposed along the Atlantic and Gulf coasts, including Florida (Field and Duane 1974, Delcourt and Delcourt 1981).

The alternating high and low sea stands of the Pleistocene and Holocene (since ca. 13,000 yr B.P.) shaped the surface of Brevard County. The outer barrier island and Cape Canaveral formed after sea levels rose when the Wisconsinan glaciers retreated (Davis 1997). Cape Canaveral is mapped as Holocene in age (Brooks 1981). Brooks (1972) suggested that the formation of the Cape Canaveral peninsula began about 7,000 years ago. Cape Canaveral is part of a prograding barrier island complex, the result of southward growth of an original cape at the site of the present False Cape (White 1958, 1970). Multiple dune ridges on Cape Canaveral suggest that periods of deposition and erosion alternated (Chaki 1974). The barrier island separating Mosquito Lagoon from the Atlantic Ocean also originated about 7,000 years ago (Mehta and Brooks 1973). However, its history has been marked by erosion, overwash, and landward migration rather than progradation; these processes continue today (Mehta and Brooks 1973). Some areas of the barrier island south of Cape Canaveral have a history of overwash, while others have been more stable (Bader and Parkinson 1990).

Merritt Island also formed as a prograding barrier island complex; the eastern edge of Merritt Island at its contact with the Mosquito Lagoon and the Banana River forms a relict cape aligned with False Cape (White 1958, 1970). Multiple dune ridges apparently represent successive stages in this growth. Brooks (1972) suggested that the geologic history of the Merritt Island-Cape Canaveral barrier island was complex. The western portion of Merritt Island is substantially older than the east (Brooks 1972, Clapp 1987). Erosion has reduced the western side to a nearly level plain (Brown et al. 1962).

1.4. Soils

Soils differ through the interaction of several factors: climate, parent material, topography, organisms, and time (Jenny 1941, 1980). The soils of KSC are mapped in the soil surveys for Brevard County (Huckle et al. 1974) and Volusia County (Baldwin et al. 1980), and the resulting soil pattern is complex. Numerous soil series and land types are represented, and these include representatives of many of the major soil groups (Schmalzer and Hinkle 1990). This is interesting since Merritt Island is a relatively young landscape and one formed from coastal plain deposits. Some differences in soil parent material do occur. In particular, soils that formed in deposits over limestone, coquina, or other alkaline material differ greatly in properties from those formed in sand. Textural differences in parent material such as that between loam or clay material and sand also influence soil properties.

The primary source of parent material for KSC soils are sands of mixed terrestrial and biogenic origin. The terrestrial material originated from southern rivers carrying sediments eroded from highly weathered Coastal Plain and Piedmont soils; these sediments are quartzose with low feldspar content (Milliman 1972). These sediments moved south through long-shore transport and may have been reworked repeatedly. The biogenic carbonate fraction of the sand is primarily of mollusk or barnacle origin with lesser contributions of coralline algae and lithoclasts; some may be reworked from offshore deposits of coquina and oolitic limestone (Milliman 1972).

The Cape Canaveral-Merritt Island complex is not all of the same age, as discussed earlier. Soils on Cape Canaveral, False Cape, and the barrier island section on the east side of Mosquito Lagoon are younger than those of Merritt Island and therefore have had less time to weather. Well drained soil series (e.g., Palm Beach, Canaveral) in these areas still retain shell fragments in the upper layers, while those inland on Merritt Island (e.g., Paola, Pomello) do not. The presence of shell fragments influences soil nutrient levels, particularly calcium and magnesium, and pH. The eastern and western sections of Merritt Island differ in age. The eastern section of Merritt Island inland to about State Route 3 has a marked ridge-swale topography presumably retained from its formation as a barrier island, while west of State Route 3, the island is flatter, without obvious ridges and swales probably due to the greater age of this topography.

Differences in age and parent material account for some soil differences, but on landscapes of Merritt Island with similar age, topography has a dramatic effect on soil formation. Relatively small elevation changes cause dramatic differences in the position of the water table that, in turn, affect leaching, accumulation of organic matter, and formation of soil horizons. In addition, proximity to the lagoon systems influences soil salinity.

1.5. Groundwater

The geologic structure and composition of the Merritt Island-Cape Canaveral barrier island complex together with climatic conditions form the basis for the hydrology of the system. Groundwater hydrology of KSC has been the subject of several studies (e.g., Edward E. Clark Engineers-Scientists, Inc., 1985; 1987a; b; c) [hereafter referenced as Clark]; the discussion that follows is based primarily on the areawide survey (Clark, 1987c). See Schmalzer and Hinkle (1990) for additional details.

The principal artesian aquifer beneath KSC is the Floridan aquifer, which occurs within the Ocala limestones (Figure 1.3). Recharge areas for this aquifer are the high ridges of central Florida. This is a large and productive aquifer; however, in the coastal areas, as beneath KSC, the water is highly mineralized. This aquifer is confined by the silts and clays of the Hawthorn formation in most places. Secondary artesian aquifers occur within the Hawthorn formation and the Caloosahatchee Marl Equivalent. The Hawthorn Limestone aquifer is associated with thin, discontinuous beds of limestone, sandstone, and sand within the silts and clays of the Hawthorn formation. It is recharged by upward leakage from the Floridan aquifer. The Shallow Rock aquifer is associated with beds of partially consolidated shelly quartz sand with silt and grey clay and some medium hard limestone of the Tamiami formation or Caloosahatchee Marl Equivalent. Recharge is by upward leakage from the Floridan aquifer. The Semi-artesian Shell and Sand Bed aquifer is associated with minor, discontinuous sand and shell beds within the Caloosahatchee Marl Equivalent. There is little freshwater recharge of this aquifer, and it may act as a conduit for seawater intrusion. Both the Shallow Rock and Sand and Shell Bed aquifers are confined by less permeable sediments of the Caloosahatchee Marl Equivalent. The artesian aquifers have little direct influence on surface vegetation; however, artesian wells have been used to irrigate orange groves and previously to maintain water levels in some mosquito impoundments on Merritt Island (Clark 1987c).

The Surficial aquifer occurs in the saturated part of the moderately permeable Pleistocene and Holocene deposits of fine to medium sand, shell, coquina, silts, and marl. Its upper boundary is the water table and the lower boundary is the confining unit at the base of the Pleistocene and Holocene deposits. Recharge is by direct infiltration of rainfall. The higher sand ridges in the center of the island are particularly important for recharge. These ridges are relatively high, are composed of permeable sands, and infiltration is less restricted by subsurface hardpans than in other areas. Two important areas of sand ridges have been distinguished: the Happy Creek Sand Ridges north of Banana Creek and the Schwartz Road Sand Ridges south of Banana Creek. From these prime recharge areas, groundwater flows east and west toward the lagoon systems and the ocean. Discharge from the surface aquifer is from evapotranspiration, seepage into canals and ditches, seepage into interior wetland swales, and seepage into impoundments, lagoons, and the ocean. Most of the seepage into

interior wetland swales is subsequently lost to evapotranspiration. Seasonal fluctations in the water table occur with changes in precipitation and evapotranspiration. The water table is highest late in the wet season (typically September-October) and drops as precipitation declines. In the winter, evapotranspiration is low as temperatures decline and some of the vegetation is dormant. In spring, evapotranspiration increases and the water table may decline during spring droughts. See Mailander (1990) for further discussion of precipitation and evapotranspiration patterns. The Surficial aquifer is extremely important since it supports the freshwater wetlands and provides fresh groundwater discharge to the surrounding subsaline lagoons (Clark 1987c).

The Surficial aquifer can be divided into several subsystems (Figure 1.4). The Dune (Barrier Island) subsystem has a lens of freshwater less than 3 m (10 ft) thick on top of intruded saline water. The primary dune acts as the prime recharge area. Shallow groundwater flows east of the ridge to the Atlantic Ocean and west to Banana River, Mosquito Lagoon, or swales; at depth (> 6.1 m [20 ft]) flow is to the Atlantic Ocean. The Dune-Swale subsystem includes the high ridges with permeable sand that favor recharge. This is the only area where the freshwater recharge of the deeper layers of the surficial aquifer occurs. During most of the year, shallow groundwater discharges to the swales. At the beginning of the rainy season after the spring drought, swales collect water and remain flooded; lateral and downward seepage from the swales helps to recharge the groundwater. In areas of pine flatwoods and swales, topography is lower and most soils have well-developed humic hardpans (spodic horizon, Bh layer) that restrict infiltration. During heavy rains, water perches above the hardpan and infiltrates slowly into the Surficial aquifer. This increases evapotranspiration and reduces recharge relative to the prime recharge areas. In the West Plain and Marsh (Lowland) subsystems, the water table is typically within 0.9 m (3 ft) of the land surface, evapotranspiration losses are high, and the dispersed saline water interface renders water quality variable. In the West Plain south of Banana Creek, a limerock "hardpan" replaces the humic hardpan of the Dune-Swale flatwoods. Along the coastlines, the Surficial aquifer contacts the saline water of the Atlantic Ocean and the brackish lagoons. Seawater intrusion occurs as a wedge at the base of the Surficial aquifer since seawater is denser than fresh. The position of the fresh-saline water interface fluctuates; when water levels are low saline water moves inland, and when water levels are high saline water is forced out, producing a dynamic system.

1.6. Surface Water

Kennedy Space Center is surrounded by the Indian River Lagoon System that extends along the East Coast of Florida from Ponce de Leon Inlet to St. Lucie Inlet near Stuart, Florida. This System was formed by changing sea levels, and its prominent features are the southern barrier islands, the Cape Canaveral foreland formation, the western mainland ridges, and the valleys and sloughs between the ridges (Head 1981).

This area is very biologically diverse as it includes the temperate Carolinian and the subtropical Caribbean zoogeographic Provinces. KSC is located in the transition zone between these two Provinces and is bordered by Mosquito Lagoon to the north, Banana River to the south, Indian River to the west, and the Atlantic Ocean to the east. These basins are shallow, aeolian, lagoons with depths averaging 1.5 m and maximums of 9 m generally restricted to dredged basins and channels.

Mosquito Lagoon and the Indian River are connected by Haulover Canal and the Intercoastal Waterway. Water flow between these two systems is primarily wind-driven. Because of the various anthoropogenic modifications related to the space program and mosquito control, circulation between Mosquito Lagoon and the Banana River was blocked in the earlier 1960s.

The Indian and Banana Rivers mix in the southern region near Eau Gallie, and through a man-made canal located just south of KSC. This navigation canal accesses the Atlantic Ocean through the Port Canaveral Locks, whose oceanic waters influence surface water quality in the northern Banana River. The northern-most Banana River is inside KSC property and closed to motorized boat traffic. It is part of the Merritt Island National Wildlife Refuge and its water quality is one of the best in the Indian River Lagoon System (Woodward-Clyde Consultants 1994). The region of the Banana River north of the NASA Causeway includes Pintail Creek and Max Hoeck Back Creek. Very little tidal fluctuation occurs, and the water movement in this location is influenced primarily by wind and evaporation.

The major fresh-water body within KSC property is Banana Creek, which drains the numerous estuaries adjacent to the Space Shuttle launch pads via a canal located northwest of the Vehicle Assembly Building to the Indian River. Salinity usually increases in a westward direction, but depending on wind direction, the Indian River system can have a greater or lesser affect on the Banana Creek water quality. Other freshwater inputs to the estuarine system surrounding KSC include direct precipitation, stormwater runoff, discharges from impoundments, and groundwater seepage.

1.7. Sediment

Numerous anthoropogenic modifications changed the natural ecosystem surrounding KSC. These included dredging the Intercoastal Waterway and Haulover Canal, stabilization of inlets, construction of causeways, land filling, agricultural development, diking, impoundment for mosquito control, and modifications primarily related to the space program.

These modifications led to the fragmentation of the native landscape that inherently changed sediment quality and characteristics. Because sediments are important for organic decomposition in both freshwater and marine systems

(Hargrave 1973, Nixon 1981), these landscape modifications inevitably affected sediment ecology surrounding KSC.

Within KSC property, the construction of roads, causeways, buildings, launchpads, and the Space Shuttle runway required additional soil. This land filling led to the formation of several burrow pits and relatively deep basins in the Banana River. The Banana River bottom topography was also modified to accomodate the Intercoastal Waterway and to create a navigable channel to access the Vehicle Assembly Building.

Numerous alteration to the marsh communities occurred in the late 1950 earlier 1960, the most prevalent being impoundment and diking as a control method for mosquitoes (Rey and Kain 1989). Most of them created numerous shallow pond-like features along the margins of the Indian River Lagoon System. These impoundments usually have a mucky substrate that is not directly exported to the lagoon.

Approximately 90% of the IRL bottom area is believed to be flat, exposed substrate, and consist primarily of sand and shell fragments (Gilmore 1977). A thin layer of silt often overlays the bottom in the more open portions of the estuary while a surficial layer of organic detrital material is usually present in portions adjacent to shore areas (Woodward-Clyde Consultants 1994). However, fine-grained, organic-rich mucky sediment occurs in about 10% of the Lagoon in deposits ranging up to more than 6 feet thick. These tend to be found in the mounths of tributary creeks, sheltered areas near harbors and causeways, and in the deeper parts of the Intercoastal Waterway (Woodward-Clyde Consultants 1994). These muck deposits tend to have high oxygen demand, particle-reactive metals and, synthetic organic compunds (Schropp et al. 1990).

1.8. Human History

Humans have occupied central Florida for thousands of years and interacted with the environment (Davison and Bratton 1986, Bense 1994). Activities that would effect background chemical composition of soil, groundwater, surface water, and sediment are relatively recent. Euro-American settlement in the 19th and early 20th centuries included logging, citrus agriculture, and free-range grazing of cattle (Davison and Bratton 1986). Use of chemical fertilizers and pesticides developed over time in agriculture, particularly citrus.

Organized efforts at mosquito control in the northern Indian River Lagoon (IRL) began in the 1940s and intensified after World War II (Provost 1959, 1977). Massive quantities of DDT were applied between 1946 and 1951; other insecticides were then adopted, because salt marsh mosquitos had become resistant to DDT. There was no documentation of the effects of insecticide applications on marsh ecosystems. Effects on non-target insects, arthropods, and birds must have been substantial (Sykes 1980). The population of the Dusky Seaside Sparrow (Ammodramus maritimus nigrescens) on Merritt Island

was estimated to have declined by 70% by 1957 due to aerial spraying of insecticides (Nicholson in Trost, 1968). DDT metabolites and other organochlorines were detected in fish (Thompson et al., 1977) and in the eggs (Blus et al. 1974, Thompson et al. 1977) and tissue samples (Nesbitt et al. 1981) of Brown Pelicans (*Pelecanus occidentalis carolinensis*) collected in the vicinity of Merritt Island in the late 1960s and early 1970s. DDT levels were generally low, and thinning of egg shells was slight (Blus et al. 1974).

Concerns over massive insecticide application and development of insecticide resistance by mosquitos spurred consideration of alternative control measures (Provost 1959, 1977). The life cycle of salt marsh mosquitos could be interrupted by maintaining flooded conditions in the irregularly flooded marshes that were primary breeding areas (Provost 1959, Clements and Rogers 1964). This was achieved by building perimeter dikes around salt marshes and flooding the marsh by either trapping rain water and runoff, pumping water from the lagoon, or tapping artesian wells (Provost 1959). Although flooding was necessary only in the mosquito breeding season, continuous flooding was common early in the use of impoundments (Provost 1959). The construction of higher, permanent dikes began in 1959, and most marshes were impounded by 1970 (Rey and Kain 1989). This reduced the widespread use of insecticides.

1.9. Reasons for Study

KSC has been the primary launch complex for the United States Space program since the 1960's. In this time, many industrial practices have required use of hazardous materials for day to day operation of this launch complex. The uses of these hazardous commodities have lead to potential contamination concerns center wide. The NASA/KSC Environmental Program Office Remediation Program Group is charged with investigating areas that may be contaminated. The remediation group is required by their Resource Conservation and Recovery Act (RCRA) permit to investigate potential release locations (PRLs). The group compiled a list of PRLs based on past industrial practices suspected to occur at each site. Each PRL is investigated to determine if contamination is present. If contamination is found, the horizontal and vertical extent and the source of the contamination are determined as well as any impacts to human health and the environment.

Data collected during the various stages of the process are screened against current regulatory risk-based screening values. These screening values include values for human health (residential and industrial) and ecological scenarios. CCAFS developed the Baseline Data Development Program (BDDP) in 1996 to develop background numbers which would be used in assisting the clean up of their remediation sites (O'Brien and Gere Engineers, Inc. 1996). At that time, KSC adopted those numbers. Due to the differences in the landscape (age, vegetation types, etc. discussed previously), KSC needed to develop baseline

(background) numbers specific to KSC. The purpose of this study was to develop the baseline (background) values specific to KSC.

When the subject of remediating a site is reached in the investigation process, the current screening values may be extremely conservative and may actually represent values that are lower than naturally exist within the landscape. In this study, areas of the landscape with little direct impact by industrial or launch operations were sampled. Sample locations were chosen away from known Solid Waste Management Units (SWMU). It should be noted that some soil, surface water, and sediment samples were collected within the industrial areas of KSC. Soil samples were collected as part of the disturbed soil class. Surface water and sediment samples were collected within the industrial area as requested by remediation team members to reflect background concentrations as part of the stormwater drainage systems as well as daily activities occurring on KSC. No sample was collected from any obvious release location. Potential impacts of site operations on these samples may include contamination by vehicles to include parking lot run-off and stormwater drainage. The data developed during this effort will be used to assist with clean up of remediation sites on KSC. The goal is to use the data as a screening tool, i.e. contaminant concentrations that exceeded risk-based screening values but are below background numbers would not force clean up at a given site.

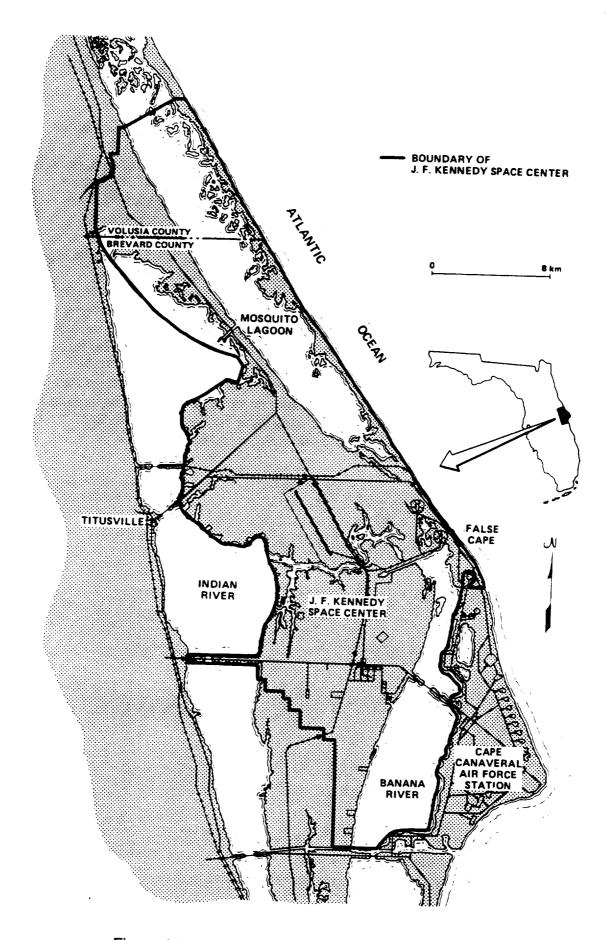
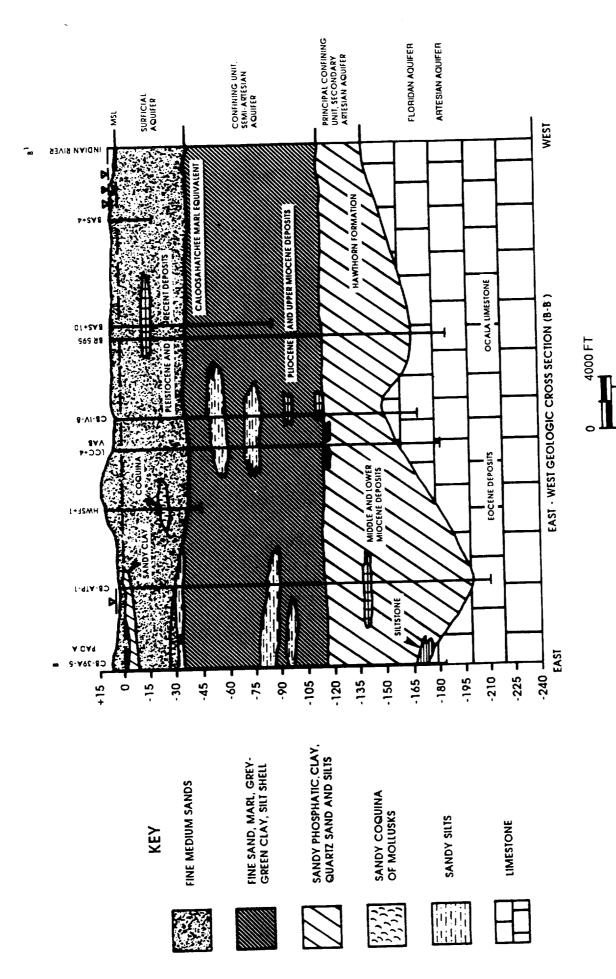


Figure 1.1. Location of Kennedy Space Center.



East-west geologic cross section, Kennedy Space center (redrafted from Clark 1987c). Veritcal scale is elevation in feet relative to mean sea level. Figure 1.2.

1000 M

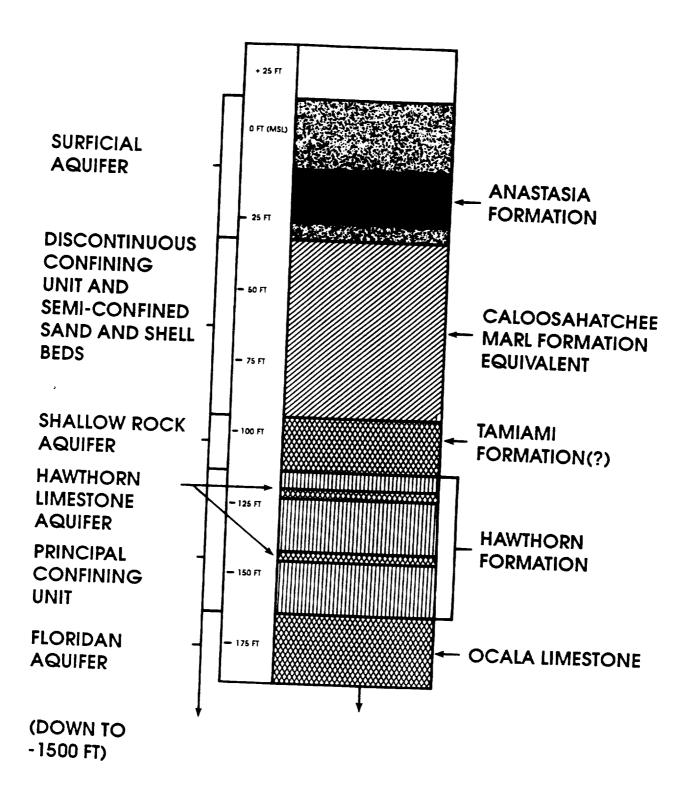


Figure 1.3. Geohydrological units on Kennedy Space Center (redrafted from Clark 1987c).

Groundwater Subaquifers

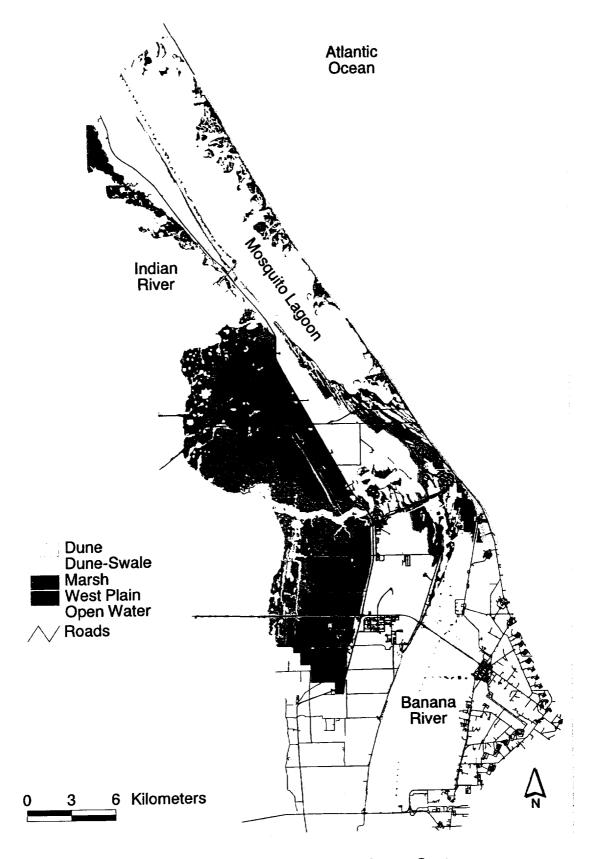


Figure 1.4. Groundwater subaquifers of Kennedy Space Center.

<u>.</u>	4

SECTION 2 Soil

2.1 Methods

2.1.1 Site selection

Numerous soil series occur on Kennedy Space Center (Huckle et al. 1974, Baldwin et al. 1980, Schmalzer and Hinkle 1990); 58 series or land types occur (Table 2.1). Chemical and physical characteristics of these soils vary (Madsen 1980, Schmalzer and Hinkle 1987, 1992a, 1992b). Sampling all these soil types adequately was impossible. Therefore, it was necessary to group soil series into similar classes before sampling.

We divided soils first into four groups: Upland, Wetland, Agricultural, and Disturbed. Upland soils are not flooded for substantial periods, while Wetland soils have standing water for substantial periods. Flooding affects organic matter accumulation, oxidation-reduction conditions, and other chemical properties of soils (Ponnamperuma 1972). We divided Upland soils into well-drained and poorly drained categories. Poorly drained soils accumulate more organic matter, which forms the cation exchange capacity in these soils retaining nutrients and metals (Schmalzer and Hinkle 1987, 1992b, 1996). We divided well-drained, upland soils into three classes: 1) geologically recent, alkaline, sandy soils of coastal dunes where the vegetation is coastal dunes, coastal strand, or coastal scrub; 2) old, inland, leached, acid, sandy soils where the vegetation is oak-saw palmetto scrub or scrubby flatwoods; and 3) inland, circumneutral soils formed over coquina where the vegetation is oak-saw palmetto scrub or xeric hammock. We divided poorly-drained, upland soils into two classes: 1) acid, sandy soils with flatwoods vegetation; and 2) circumneutral to alkaline soils formed over coquina or limestone where the vegetation is mesic hammock (Table 2.1).

The primary division of wetland soils was between: 1) inland, freshwater wetlands where the vegetation was freshwater marshes or hardwood swamps; and 2) coastal, brackish to saline wetlands where the vegetation was salt marshes or mangroves (Table 2.1).

Agricultural soils were of two types: 1) active or abandoned citrus on scrub soils; and 2) active or abandoned citrus on hammock soils (Table 2.1).

Disturbed soils included various types modified by construction (Table 2.1). This group could be heterogeneous, but there was no apparent division into homogeneous subgroups.

This classification resulted in ten groups. We selected twenty sample locations in each for a total of 200 locations. Sample number per group was determined

	-		
		_	
		, *	

based on variation in soil chemistry found in previous studies (Schmalzer and Hinkle 1992a, 1992b).

A Trimble Pathfinder Professional XL Geographic Positioning System (GPS) unit was used to collect soil sampling locations using real-time differentially-corrected methods (Trimble Navigation Limited 1994). Locations were exported to North American Datum 1927 (NAD27) State Plane coordinates for overlay in a Geographic Information System (GIS).

2.1.2 Sampling

Soil sampling was conducted at 200 locations (Figure 2.1). A 0-1ft (0-30.8 cm) layer composite sample was collected with a stainless steel scoop and bowl and placed in the appropriate containers for analysis. Preservatives for each analysis were used, as appropriate. Samples for bulk density were collected using a stainless steel ring, which was hammered into the ground. The bulk density-sampling ring was carefully collected to keep as much volume as possible inside the ring and was then placed in a plastic bag for analysis. Latex gloves were worn during sample collection and replaced for each sample collected. Proper decontamination procedures were conducted after each sample was collected. Environmental Conservation Laboratories (ENCO) was subcontracted for sample collection and analysis. ENCO followed sampling methods required by the "NASA/KSC Sampling and Analysis Plan for RCRA Facility Investigations at Kennedy Space Center, Florida and Cape Canaveral Air Station, Florida, Volume 4" of the Generic Work Plans (SAP) (NASA 1996). Quality assurance/quality control samples were collected as required by the SAP.

2.1.3 Chemical Analysis

Table 2.2 lists the analyses that were conducted on each soil sample: organochlorine pesticides (EPA method 8081), aroclors (EPA method 8082), chlorinated herbicides (EPA method 8151), and polycyclic aromatic hydrocarbons (PAH) (EPA method 8310) were analyzed. Methods for organochlorine pesticides, aroclors, and chlorinated herbicides use gas chromatography (USEPA, 1996). Polycyclic aromatic hydrocarbons were analyzed by high-pressure liquid chromatography (USEPA 1996). Metals were analyzed using one of the following EPA method numbers appropriate for the specific metal: 200.7, 204.2, 6010, 7210, 7471, 7520, 7610, 7770, or 7950 (Table 2). Methods 200.7 (aluminum [AI]) and 6010 (arsenic [As], barium [Ba], beryllium [Be], cadmium [Cd], calcium [Ca], chromium [Cr], cobalt [Co], copper [Cu], iron [Fe], lead [Pb], magnesium [Mg], manganese [Mn], nickel [Ni], potassium [K], selenium [Se], silver [Ag], thallium [TI], vanadium [V], and zinc [Zn] use inductively coupled plasma-atomic emission spectrometry (USEPA 1983, 1996).

Methods 7210 (Cu), 7520 (Ni), 7610 (K), 7770 (sodium [Na]), 7950 (Zn) use atomic absorption, direct aspiration techniques (USEPA 1996). Method 204.2

(antimony [Sb]) uses an atomic absorption, furnace technique (USEPA 1983). Mercury (Hg) was analyzed using method 7471, a cold-vapor, atomic absorption technique (USEPA 1996).

Soil samples were also analyzed for cation exchange capacity (CEC) using EPA method 9081 (USEPA 1996); this technique is based on Chapman (1965). Soil pH was determined on a 1:1 soil/water slurry by electrode using EPA method 9045 (USEPA 1996). Total organic carbon was determined by wet combustion (Nelson and Sommers 1982). Total percent solids was determined by drying to constant weight at 103-105 C [Standard Methods for the Examination of Water and Wastewate, 2540G(Clercerl et al. 1998)]. Bulk density was determined by the core method (Blake 1965). Soil texture was determined by sieving through a series of standard sieves (Day 1965, McLane 1995). Resitivity of a soil water extract was measured using a M.C. Miller Company Soil Box; resistivity is the reciprocal of conductivity (Bower and Wilcox 1965), which is often reported.

The documents "Test Methods for Evaluating Solid Waste" SW-846, (USEPA 1996) and "Standard Methods for the Examination of Water and Wastewater" (Clercerl et al. 1998) were used by ENCO for analysis protocols.

ENCO posses a Florida Department of Environmental Protection (FDEP) approved Comprehensive Quality Assurance Plan No. 960038. A data validation report was completed, see Appendix G.

2.1.4 Data Analysis

Data analyses was conducted using SPSS Version 9 (SPSS Inc., 1999) using the following approach.

- Frequency of parameters occurring above the detection limits were determined. Where most of the data are below detection, intensive data analyses cannot be justified.
- Box plots were used to scan for obvious outliers in the data. Outliers could arise from a contaminated sample or local, anomolous site conditions. Inclusion of outliers can skew the mean and standard deviation of a parameter.
- Tests for normality of distributions were done using normality plots and Kolmogorov-Smirnov statistics with Lilliefors significance levels. Normality of distribution determines whether parametric or nonparametric analyses should be used or whether transformations are required.
- Kruskal-Wallis nonparametric analog of one-way analysis of variance was used to test whether parameters differed among soil classes. Where the sample size of data above detection was sufficient, one-way analysis of variance (ANOVA) of transformed data (log₁₀) followed by Games-Howell post hoc tests was used to determined how parameters differed among soil classes. Games-Howell tests do not assume equal variances (Ray and Quinn 1989). These tests determine whether soil classes were the same for

a given parameter and the patterns of variation if differences were significant. If soil classes are not the same, then the use of a single mean (or screening value) to represent all soils may not be appropriate.

The occurrence of data below detection levels is a complication for data analysis and interpretation. Various approaches can be taken (e.g., Breckenridge and Crockett 1995); here one-half the detection level was substituted for values below detection (USEPA 1998). No summary statistics are reported where all values are below detection.

For screening purposes, data are given as twice the mean (USEPA 1998). The KSC Background Screening table for soil is located in Appendix A, Table A-1.

2.2 Results

2.2.1 Soil Distribution

Soil classes were distributed across the landscape in patterns related to topography, geology, landscape age, and human influences (Figure 2.2). Soil classes differ in extent (Table 2.3). Flatwoods, saltwater wetlands, and freshwater wetlands soils were the largest classes.

Sample sites were distributed within these types (Figure 2.1). See Appendix B, Table B-1 for sample site descriptions and Table B-2 for sample site coordinates.

2.2.2 Soil Chemistry

Sixteen organochlorine pesticides, all (six) aroclors, fifteen chlorinated herbicides, one polycyclic aromatic hydrocarbon, and four metals were below detection in all samples (Table 2.4). Frequencies of organochlorine pesticides above detection limits in all soils were low (Table 2.5); however, frequencies were higher in agricultural (citrus scrub, citrus hammock), disturbed, and saltwater wetland soil classes. Frequencies of chlorinated herbicides above detection limits were low (Table 2.5); frequencies were slightly higher in agricultural (citrus scrub, citrus hammock), disturbed, and saltwater wetland soil classes. PAHs were above detection limits in relatively few soil samples (Table 2.5); frequencies were greater in agricultural (citrus scrub, citrus hammock) and disturbed soil classes. However, PAH values above detection occurred in other soil classes. Among the metals, barium, cadmium, selenium, silver, and thallium occurred in few samples above detection limits (Table 2.5). Arsenic, copper, and mercury were above detection limits with an intermediate frequency, while the other metals were frequently above detection limits (Table 2.5).

The Remediation team decided to treat five organochlorine pesticides as below detection in all samples based on their low frequency of detection (Table 2.6).

There was considerable variation among many soil parameters indicated in box plots and other exploratory analyses. We considered only one metal value to be an outlier. One value of arsenic was 140 mg/kg, while the next highest value was 8.5 mg/kg. This value was deleted from further analyses.

Preliminary analysis of total organic carbon (TOC) data indicated that most values (186/220, 84.5%) were reported as >26,700 ppm (2.67%). These data provide little useful information on differences among KSC soils and were not analyzed further.

Preliminary analysis of cation exchange capacity (CEC) data indicated a bimodal distribution with most of the data <80 meq/100g and a second group between 500-1000 meq/100g. Mineral soils have CEC values typically <80 meq/100g; highly organic soils might have CEC = 180 meq/100g (Brady 1974). The 18 high values cannot be valid for KSC soils and were excluded from further analyses.

KSC Background Screening values for soil are given in Appendix A, Table A-1. Summary data are given in Table 2.7. Detailed summary statistics are given in Appendix B, Tables B-3 through B-13.

All soil parameters had distributions that differed from normality (Kolmogorov-Smirnov test, p < 0.001). Transformed values (\log_{10}) of Al, As, Ca, Cr, Cu, Fe, Hg, K, Mg, Mn, Na, V, and Zn still differed from normality (p < 0.05); however, normality plots showed that the distributions were much improved such that analysis of variance could be conducted. One-way ANOVA is robust to modest departures from normality (Ray and Quinn 1989).

Normality of distribution of resistivity and CEC was improved by transformation (log₁₀); however, normality plots indicated that bulk density was closer to normality than transformed values. Therefore, ANOVA was conducted on transformed values of resistivity and CEC but on untransformed bulk density data. pH is expressed on a log scale and was not transformed further.

Kruskal-Wallis tests indicated that all soil parameters differed among soil classes (p< 0.001). However, given the low frequency of detection of organics and some metals, ANOVA was conducted only for metals where >25 % of values were above detection and for pH, bulk density, resistivity, and CEC. All metals analyzed (Al, As, Ca, Cr, Cu, Fe, Pb, Mg, Mn, Hg, K, Na, V, Zn) were significantly different among types (p < 0.001) (Table 2.8). Variances were not homogeneous except for Zn (Levene test, p<0.05); thus, the Games-Howell test was appropriate for examining multiple comparisons (Table 2.8).

Patterns of differences varied among metals (Table 2.8). Aluminum had low values in flatwoods, acid scrub, and coastal soils and higher values in other types. Arsenic had higher values in coastal and saltwater wetland soils and lower values elsewhere. Calcium was high in coastal and disturbed soils, low in

acid scrub and flatwoods, and intermediate in other types. Chromium, copper, manganese, and zinc had higher values in citrus scrub and citrus hammock soils and lower values elsewhere. Iron was low in acid scrub and flatwoods soils and higher in other types. Lead values were low in acid scrub, coastal, and flatwoods soils, high in saltwater wetlands, and intermediate in other types. Magnesium, potassium, and sodium levels were lowest in acid scrub and highest in saltwater wetlands. Mercury was low in acid scrub and coastal soils and high in hammocks and freshwater wetlands. Vanadium was low in acid scrub and flatwoods, high in saltwater wetlands, and intermediate in other types.

Soil pH was highest in coastal soils and lowest in flatwoods (Table 2.9). CEC was low in coastal, acid scrub, and coquina scrub soils and high in freshwater wetlands and hammocks.

2.2.3 Soil Physical Parameters

Bulk density was low in flatwoods, hammock, freshwater wetland, and saltwater wetland soils and higher in the other types (Table 2.9). Resistivity was low in saltwater wetland soils.

Texture

The graphic mean grain size of each soil type was calculated according to the method defined by Folk and Ward (1957). This method involves the use of grain size frequency data obtained from sieve analysis. Grain size in mm is converted to phi units (ϕ) using the equation $\phi = -\log_2$ mm. (This conversion creates a logarithmic scale and allows a wide range of grain sizes to be graphically displayed on a smaller scale.) In order to obtain the graphic mean, cumulative percentage of grain size is plotted on a probability y-axis and phi size is plotted against the x-axis. (see Figure 2.3) The phi size of the 16, 50, and 84 percentage is marked and used in the following equation:

$$\frac{16 \phi +50 \phi +84 \phi}{3}$$
 = mean grain size in ϕ (Folk and Ward 1957)

Phi size can then be converted to millimeters and assigned to the appropriate size class in the Wentworth (1922) ratio scale preferred by most sedimentologists. Each sand grain size class in the Wentworth scale is one half the size of the next largest class in the scale. Therefore, the divisions are consecutive whole numbers on the phi scale. (The division between coarse sand and medium sand is 1 ϕ ; the division between medium sand and fine sand is 2 ϕ , etc.).

2.3 Discussion

2.3.1 Soil Distribution

Soil classes vary substantially in their extent across the KSC landscape. Flatwoods and wetlands predominate, reflecting the low, poorly drained topography of KSC. Current or past citrus agriculture has affected about 3% of the soil area and disturbed soils constitute about another 3.6%.

The differing extent of soil classes affects the interpretation of quantitative soil data. In this study, each class was sampled equally. Therefore, the mean of a given soil parameter will not be the same as an area-weighted mean.

2.3.2 Soil Chemistry

Organochlorine Pesticides.

Organochlorine pesticides have no natural sources and their presence in the environment comes from past human use. Due to their persistence in the environment, tendency to bioaccumulate, and deleterious effects on non-target species, their use in the U.S. has been discontinued. However, residual amounts persist from their former use (Kuhnt 1995, Kolpin et al. 1998). In KSC soils, DDT and its degradation products (DDD and DDE) and Dieldrin are most common in occurrence. DDT, DDD, DDE, and Dieldrin are among the most persistent organochlorine pesticides in the environment (Kuhnt 1995).

On KSC, these pesticides were detected most often in the agricultural types (citrus scrub and citrus hammock), saltwater wetlands, and disturbed areas. Application of DDT and other organochorine pesticides for mosquito control was common in the 1940s and 1950s (Provost 1959, 1977). Saltwater wetlands received high levels of pesticide application. High levels of organic matter enhance retention and persistence of these pesticides (Senesi and Chen 1989) and may enhance persistence in these marshes. Organochlorine pesticides were probably used in citrus agriculture during this period (1940s - 1950s) and perhaps into the 1970s when their widespread use was curtailed.

Parkinson et al. (1993) found DDE, Dieldrin, or Lindane in soils of 16 (out of 18) mosquito control impoundments in St. Lucie County where DDT, Lindane, Dieldrin, and Chlordane had been applied for mosquito control in the 1940s and 1950s (see also Wang et al. 1992). Minimum detectable concentrations in that study were DDE - 0.24 μ g/kg, Dieldrin - 0.32 μ g/kg, and Lindane - 0.18 μ g/kg. Detection limits in the current study were higher by a factor of ten (Table 2.2). Lower detection limits often increase frequency of detection (Kolpin et al. 1995).

Chlorinated Herbicides

The chlorinated herbicides surveyed belong to several different chemical classes; however, the only ones above detection limits (Silvex, 2,4-D, 2,4-DB) are all chlorinated phenoxy acids (Ashton and Craft 1981). Chlorinated herbicides are

less persistent in the environment than the organochlorine pesticides (Kuhnt 1995); however, they can persist long enough to move from the soil into the groundwater in some areas (Kolpin et al. 1995, Barbash and Resek 1996).

The low frequency of detection indicates that contamination by chlorinated herbicides has not been widespread in the KSC environment.

Polycyclic Aromatic Hydrocarbons

Polycyclic (or polynuclear) aromatic hydrocarbons (PAH) are a class of compounds containing two or more fused benzene rings that are of both natural and anthropogenic origin (Suess 1976). They include many mutagenic and some carcinogenic compounds. Natural sources include synthesis by microorganisms, phytoplankton, algae, and higher plants, wildfires, volcanic activity, and sediment diagenesis (Suess 1976, Wickstrom and Tolonen 1987). PAH are produced by a variety of industrial combustion processes including heating and power generation, internal combustion engines, coke production in the iron and steel industry, catalytic cracking in the petroleum industry, and refuse incineration (Suess 1976). Coal tar, carbon black, asphalt, and creosote contain high concentrations of PAH (Suess 1976).

Biomass burning, including wildfires and prescribed burning, produces PAH (Standley and Simoneit 1987, Balletine et al. 1996, Simoneit et al. 1996). Different conditions of fuels, combustion conditions, temperature, and oxygen levels produce different suites of PAH (Standley and Simoneit 1987, Balletine et al. 1996). Generation of PAH is greater during active combustion than smoldering combustion (Standley and Simoneit 1987). Levels of individual PAHs in smoke from wildland fires can be 10-1000 times greater than ambient rural conditions but still 10-1000 times less than urban atmospheres (Standley and Simoneit 1987).

In the atmosphere, PAH are adsorbed on aerosols. They are removed from the atmosphere by photochemical oxidation and by thermal reactions occurring in darkness (Suess 1976, Behymer and Hites 1985, Masclet et al. 1986). Wet and dry deposition transport PAH to soil and water surfaces (Suess 1976). Regional deposition occurs distant from point sources (Jones et al. 1989c).

PAH deposited to soil surfaces are additions to those produced from decayed plants and the activity of soil bacteria (Suess 1976). In soils, PAH are thermodynamically stable, have low aqueous solubilities, and adsorb to soil particles; these properties allow long-term persistence (Pothuluri and Ceniglia 1998). Microorganisms including bacteria, fungi, cyanobacteria, and green algae can degrade PAH (Pothuluri and Ceniglia 1998). Compounds with four or more fused rings degrade slowly, but lower molecular weight, more water soluble compounds such as naphthalene are more readily biodegraded (Pothuluri and Ceniglia 1998).

Increased combustion of fossil fuels since the middle of the 19^{th} century has caused increases in PAH levels in soil (Jones et al. 1989b, c) and sediments (Wickstrom and Tolonen 1987) distant from local sources. PAH levels in soils tend to decline along a gradient away from urban areas or point sources (Vogt et al. 1987, Jones et al. 1989d). Soils high in organic matter in rural areas have higher PAH levels than mineral soils in the same region (Vogt et al. 1987, Jones et al. 1989d). Not all PAHs follow the same pattern; Vogt et al. (1987) found naphthalene levels the same near and distant from point sources of PAH contamination suggesting either that it was primarily of natural origin or degraded quickly enough not to accumulate. In industrialized areas, levels of PAH in the soil can become quite high; Bodzek et al. (1998) reported Σ PAH > 4000 μ g/kg in an industrial region of Poland. Even in a semi-rural area of England Σ PAH = 1000 μ g/kg have been reported (Jones et al. 1989b).

Metals

Metal concentrations in KSC soils reflect both the origins of the soils and subsequent changes from weathering and leaching. Sands of coastal dunes are mixtures of terrigenous and biogenic deposits. The terrigenous deposits are primarily quartz with small amounts of feldspar, heavy minerals, and phosphorite, while the biogenic deposits are carbonates, primarily shell material (Field and Duane 1974). The terrigenous, siliciclastic deposits originated from rivers draining the Appalachians and Piedmont that moved south from longshore drift; however, they have been reworked repeatedly from older Quaternary sediments (Davis 1997, Scott 1997). Florida soils have low concentrations of many elements (Shacklette and Boerngen 1984).

Metals: Macronutrients

Six elements are required by plants in relatively large amounts and are termed macronutrients: nitrogen, phosphorus, potassium, calcium, magnesium, and sulfur (Brady 1974). Of these, Ca, Mg, and K are considered here.

Calcium. Ca comprises about 3.6 % of the earth's crust; however, Ca is mobile in the soil and readily leached (Paton et al. 1995). Calcium carbonate shell material is the initial source for most of the Ca in these soils. Thus, the young, coastal dune soils have the highest Ca levels (Tables 2.7, 2.8). Disturbed soils often include dredged material high in shell material accounting for their high levels. Soils associated with coquina (coquina scrub, hammocks) also have relatively high Ca levels. Those soils most thoroughly leached, acid scrub and flatwoods, have the lowest Ca levels (Tables 2.7, 2.8). Ca is abundant in seawater (Mitsch and Gosselink 1986) explaining the relatively high levels in saltwater wetland soils. Ca, an important structural element in plants, is involved in some enzyme systems, and is important to osmoregulation (Marschner 1986).

Magnesium. Mg makes up about 2.1 % of the earth's crust (Paton et al. 1995) and is abundant in seawater (Mitsch and Gosselink 1986). The highest levels of

Mg were in saltwater marshes reflecting the seawater source (Tables 2.7, 2.8). Magnesium carbonate shell material is another source for Mg in these soils. Mg is mobile in the soil and readily leached. The highly leached, acid scrub soils have the lowest Mg levels (Tables 2.7, 2.8). In plants, Mg is required for photosynthesis, protein synthesis, and many enzyme systems (Marschner 1986).

<u>Potassium</u>. K comprises about 2.6 % of the earth's crust (Paton et al. 1995) and is abundant in seawater (Mitsch and Gosselink 1986). The highest levels were in saltwater marshes, reflecting the seawater source (Tables 2.7, 2.8). K is highly mobile and readily leached. In plants, K is required for osmotic regulation, enzyme activation, membrane transport, and other physiological processes (Marschner 1986).

Metals: Trace Elements

Aluminum. Al is one of the main constituents of the earth's crust (Kabata-Pendias and Pendias 1984). Alumino-silicate clays are produced by weathering and are a major constituent of many soils (Brady 1974, Birkeland 1999). Although the clay fraction of KSC soils is small (Huckle et al. 1974, Baldwin et al. 1980), Al is present throughout. The lowest levels of total Al in KSC soils were in flatwoods and acid scrub (Tables 2.7, 2.8) perhaps due to intense weathering (Birkeland 1999). Low levels of Al in coastal soils may reflect low amounts in the original sediments. KSC soils are low in Al relative to other Florida soils (Shacklette and Boerngen 1984, Table 2.11). Availability of Al in soil is pH dependent, with its mobility increasing below a pH of 5.5 (Brady 1974). Al toxicity to crop plants is a problem in acid soils; however, low levels of Al may be beneficial for some species and genotypes (Marschner 1986).

Antimony. The crustal abundance of Sb is low (Kabata-Pendias and Pendias 1984) as are its levels in sandy soils (Table 2.12) and in Florida soils (Table 2.11); it is usually associated with sulfides in rocks, sediments, and soils (McBride 1994). Sb is not essential to plant growth but is taken up by plants if present in a soluble form; plant toxicity from Sb appears uncommon (Kabata-Pendias and Pendias 1984). Sources of Sb contamination include industrial pollution and coal combustion. Sb was never above detection limits in KSC soils (Table 2.4) reflecting low background levels and no widespread contamination.

Arsenic. As is widely distributed in rock types (Kabata-Pendias and Pendias 1984). The range of background concentrations in soils is fairly wide (Table 2.12). Levels in KSC soils (Table 2.7) are toward the low end of the range of As values for sandy soils but within typical Florida values (Table 2.11). Values in coastal and saltwater wetland soils are higher than the other types (Table 2.7, 2.8). As is associated with organic matter in soil (Kabata-Pendias and Pendias 1984) which may explain the higher values in saltwater wetlands. Significant contamination sources include industrial pollution, coal combustion, mining and metal processing, arsenic pesticides, and wood preservatives (Kabata-Pendias and Pendias 1984, Evangelou 1988); As also occurs in some phosphate fertilizers (Kabata-Pendias and Pendias 1984). As was once used in livestock

dips to control ticks, and this resulted in local contamination (Upchurch and Randazzo 1997). As is not essential to plants and little is known of its biochemical role, but plants take it up. It can be phytotoxic, but some plants accumulate large amounts with no apparent harm.

<u>Barium</u>. Ba is frequently present in soils at varying levels (Table 2.12); Florida soils are typically toward the low end of this range (Table 2.11). In rocks it is associated with feldspars and mica (McBride 1994). It is commonly reported in plants, but has no known function; there are only a few reports of toxicity (Kabata-Pendias and Pendias 1984). Ba is present in sewage sludge, fossil fuel, and peat (Kabata-Pendias and Pendias 1984, Evangelou 1988). Ba was above detection limits in only a few KSC soils (Table 2.5), reflecting low background levels and no widespread contamination.

Beryllium. Be is widely distributed but in low concentrations in rock types and soils (Kabata-Pendias and Pendias 1984, Table 2.12). Sources of Be contamination include smelters and coal combustion. Be has no clear physiological function in plants and is toxic in solution to plants (Kabata-Pendias and Pendias 1984). Be was never above detection limits in KSC soils (Table 2.4), reflecting low background levels and no widespread contamination.

Cadmium. Cd tends to be concentrated in clay and shale deposits (Kabata-Pendias and Pendias 1984), and it is associated with zinc in sulfide minerals (McBride 1994). Typical background values of Cd in soil are < 0.5 ppm (Kabata-Pendias and Pendias 1984, Table 2.12). Mining, smelting, phosphate fertilizers, automobile tires, fossil fuels, and sewage sludge are primary sources of Cd; atmospheric inputs exceed losses of Cd in many areas (Kabata-Pendias and Pendias 1984, Stevenson 1986, McBride 1994). Cd is not an essential plant nutrient but is taken up by plants and becomes phytotoxic at elevated levels. Cd was above detection limits in only a few soil samples on KSC (Table 2.5), reflecting low background levels and no widespread contamination.

<u>Chromium.</u> Cr occurs in a variety of rock types but is most abundant in serpentine (Kabata-Pendias and Pendias 1984); Cr has generally low values in sandy soils (Table 2.12) and in Florida soils (Table 2.11). Most KSC soils have low to very low values of Cr (Table 2.7, 2.8). Cr values in agricultural types (citrus scrub, citrus hammock) are higher; Cr occurs in phosphate fertilizers, and this may account for its greater values in agricultural soils here. Sources of Cr contamination include industrial wastes, mining and metal processing, wood preservatives, and sewage sludge (Kabata-Pendias and Pendias 1984, Evangelou 1988). Most Cr in soil is not readily available to plants. Cr has no established role in plant nutrition, and soluble Cr⁶⁺ is toxic to plants (Kabata-Pendias and Pendias 1984). However, Cr is an essential micronutrient for man and animals.

Cobalt. Co is most abundant in ultramafic rocks such as serpentine (Kabata-Pendias and Pendias 1984). Co levels in sandy soils are typically low (Table 2.12). Co levels in soils of the Atlantic Coastal Plain (Kabata-Pendias and Pendias 1984) and in Florida (Table 2.11) are particularly low. Co is an essential mineral element for ruminants and is required by nitrogen-fixing microorganisms (Marschner 1986). Sources of Co contamination include metal smelting, paint, and combustion of oil and coal (Kabata-Pendias and Pendias 1984, Evangelou 1988). Co was below detection limits in KSC soils (Table 2.4), reflecting low background values and no widespread contamination.

Copper. Cu is most abundant in mafic rocks. Worldwide mean background values of Cu in soil range between 6-60 ppm (Kabata-Pendias and Pendias 1984) with sandy soils at the low end of the range (Table 2.12). Values in Florida soils are typically low (Table 2.11). Cu is concentrated in surface soils due to bioaccumulation and recent anthropogenic sources; these sources include mining and smelting emissions, fertilizers, fungicide sprays, algicides, wood preservatives, and municipal and agricultural wastes (Kabata-Pendias and Pendias 1984, Evangelou 1988). Cu is an essential plant micronutrient, involved in several important enzyme systems, but also becomes phytotoxic above certain levels (Marschner 1986). Cu levels in KSC soils are low, but the agricultural types (citrus scrub, citrus hammock) have higher values probably due to past applications of Cu-containing fungicides or fertilizers (Tables 2.7, 2.8). Cu is relatively immobile in the soil (Kabata-Pendias and Pendias 1984, McBride 1994), so elevated levels in agricultural soils may persist.

Iron. Fe is a major constituent of the lithosphere; its abundance in rock types ranges from 0.4 % of limestones to 10.0 % of ultramafic rocks (Kabata-Pendias and Pendias 1984). Typical values in soils range from 0.5 % to 5.0 % (5,000-50,000 ppm) (Brady 1974). KSC soils have relatively low levels of Fe compared to this and compared to other Florida soils (Table 2.11). Acid scrub and flatwoods are much lower than other types (Tables 2.7, 2.8). Soils of these types are mainly podzols (spodosols) (Huckle et al. 1974). Formation of podzols involves intense leaching of the surface horizon removing iron and transporting it to deeper soil layers (Paton et al. 1995). Fe is an essential plant micronutrient, involved in critical enzyme systems (Marschner 1986). Soluble iron, available to plants, is controlled by pH and oxidation-reduction conditions. Iron deficiency usually occurs in alkaline, well-aerated soils, but iron toxicity can occur in acid, anaerobic soils (Kabata-Pendias and Pendias 1984, Marschner 1986).

Lead. Pb is concentrated in acid magmatic rocks and clay sediments. Pb content of U.S. soils averages 20 ppm with similar ranges in a variety of soils (Kabata-Pendias and Pendias 1984, Table 2.12). Typical Florida soils range lower than 10 ppm (Table 2.11). Pb is widely distributed in KSC soils, but Pb levels are relatively low (Table 2.7, 2.8). Levels in saltwater wetlands are higher than other types. In surface soils, Pb is associated with organic matter; saltwater wetlands tend to accumulate organic matter at the surface. Pb contamination is

widespread, particularly in surface soils; sources of contamination have included mining and smelting, fossil fuels, paint, automobile exhaust, sewage sludge, pesticides, and fertilizers (Kabata-Pendias and Pendias 1984, Evangelou 1988). Plants take up Pb, both from the soil and air, although it has no physiological function; it may become phytotoxic at higher levels (Kabata-Pendias and Pendias 1984).

Manganese. Mn is one of the most abundant trace elements but has its greatest concentrations in mafic rocks (Kabata-Pendias and Pendias 1984). Soil levels of Mn vary, although those reported for sandy soils (Table 2.12) are lower than in soils from mafic rocks (Kabata-Pendias and Pendias 1984). Tyical values for Florida soils are relatively low (Table 2.11). KSC soils are relatively low in Mn (Table 2.7, 2.8); however, agricultural soils (citrus scrub, citrus hammock) are higher than other types. Mn is found in sewage sludge and phosphate fertilizers (Kabata-Pendias and Pendias 1984), suggesting a fertilizer source of Mn for the agricultural soils. Other sources include mining and smelting (Evangelou 1988). Mn levels in coquina scrub soils are also relatively high. The chemistry of Mn in the soil is complex with cycles of oxidation and reduction, solution and reprecipitation under varying soil conditions (Kabata-Pendias and Pendias 1984, McBride 1994). Mn is often associated with Fe in soils, and Fe is also high in coquina scrub soils. Mn is an essential plant micronutrient, involved in a number of enzyme systems and physiological processes (Marschner 1986). Levels of Mn that are toxic to plants vary among plant species and with environmental conditions.

Mercury. Hg concentrations in rock types are greatest in organic-rich shales. In soils, concentrations are greatest in organic soils (Kabata-Pendias and Pendias 1984) with lower levels in acid, sandy soils (Table 2.12). Florida soils are generally low in Hg (Table 2.11). KSC soils are low in Hg (Tables 2.7, 2.8); types with more organic matter (hammocks, wetlands) tend to have higher Hg values, consistent with its affinity for organic matter. Sources of mercury contamination include mining and metal processing, some chemical industry, sewage sludge, coal combustion, and formerly the application of fungicides containing Hg (Kabata-Pendias and Pendias 1984, Stevenson 1986). The formation of organomercury compounds, particularly methyl mercury, is important to the distribution and cycling of Hg. Plants take up and translocate Hg, but it has no physiological function and is phytotoxic (Kabata-Pendias and Pendias 1984).

Nickel. Ni concentrations are greatest in ultramafic rocks (Kabata-Pendias and Pendias 1984). Ni levels in sandy soils are typically low (Table 2.12) as are levels in Florida soils (Table 2.11). Sources of Ni contamination include metal processing and combustion of oil and coal; some phosphate fertilizers and sewage sludge are also sources (Kabata-Pendias and Pendias 1984). Plants take up Ni; it has no established general function in plant metabolism but is involved in nitrogen metabolism of certain legumes (Marschner 1986). However,

it becomes toxic at higher levels. Ni was below detection limits in all KSC soils (Table 2.4), reflecting low background values and no widespread contamination.

<u>Selenium</u>. Se is present at low levels in many rock types but is most associated with the clay fraction of sedimentary rocks (Kabata-Pendias and Pendias 1984) and with sulfide minerals (McBride 1994). Levels in sandy soils are typically low (Table 2.12) as are levels in Florida soils (Table 2.11). KSC soils are consistent with this, with Se above detection limits in only a few samples (Table 2.5). Se is taken up by plants, and some species accumulate large quantities (Marschner 1986). Se is a required micronutrient for ruminants but is toxic at high levels.

<u>Silver</u>. The occurrence of Ag is similar to Cu but at much lower concentrations (Kabata-Pendias and Pendias 1984). Background values of Ag in soil can range from 0.03 to 3.2 ppm (Kabata-Pendias and Pendias 1984). Mining, metal processing, and film processing are the major sources of Ag contamination (Kabata-Pendias and Pendias 1984, Evangelou 1988). Plant toxicity appears uncommon. Ag was above detection limits in only a few KSC soil samples (Table 2.5), reflecting low background values and no widespread contamination.

Sodium. The Na content of the earth's crust is about 2.8% (Marschner 1986). The Na is the most abundant cation in seawater (Mitsch and Gosselink 1986). The highest levels in KSC soils (Tables 2.7, 2.8) were in saltwater marshes, reflecting that seawater source. The next highest, although much lower, levels were in coastal dune soils (Tables 2.7, 2.8) where salt spray provides a continuing input of Na (Ranwell 1972). Na is highly mobile and readily leached from surface soils where precipitation is abundant (Paton et al. 1995). The lowest levels in KSC soils were in the highly leached, acid scrub soils (Tables 2.7, 2.8). Na is not required by most plants, but it is by some halophytes (Marschner 1986). High concentrations of Na limit the growth of non-halophytic plants (Haines and Dunn 1985).

<u>Thallium</u>. Background levels of TI range between 0.02 to 2.8 ppm in surface soils of the U.S. (Kabata-Pendias and Pendias 1984). Anthropogenic sources are coal combustion and metal smelting. TI is toxic to plants and animals at high levels (Kabata-Pendias and Pendias 1984). TI was above detection limits in only a few KSC soil samples (Table 2.5), reflecting low background values and no widespread contamination.

Vanadium. V content of rock types varies widely but is most abundant in mafic rocks (Kabata-Pendias and Pendias 1984). Sandy soils contain relatively low amounts of V (Table 2.12, Kabata-Pendias and Pendias 1984) as do Florida soils (Table 2.11). V tends to be associated with organic matter in soil (McBride 1994). KSC soils are consistent with this with generally low values (Tables 2.7, 2.8), and somewhat higher ones in soils with more organic matter (wetlands, hammocks). Sources of V contamination include processing of mineral ores and combustion of coal and fuel oil; it also occurs in some phosphate fertilizers

(Kabata-Pendias and Pendias 1984). V is essential for the growth of freshwater algae, but its physiological role, if any, in higher plants is unclear (Marschner 1986).

Zinc. Zn is widely distributed in rock types and soils (Kabata-Pendias and Pendias 1984, Table 2.12) but is usually associated with sulfide minerals (McBride 1994). Concentrations in Florida soils are typically low (Table 2.11). Zn is an essential plant micronutrient involved in numerous enzyme systems but becomes toxic at high levels (Marschner 1986). Zn may be deficient in acid, leached soils (McBride 1994). Zinc contamination is related to mining and smelting, wood preservatives, and to agriculture where it is a component of fertilizers and some pesticides (Kabata-Pendias and Pendias 1984, Evangelou 1988). KSC soils have generally low levels of Zn (Tables 2.7, 2.8), but levels in the agricultural types (citrus scrub, citrus hammock) are higher suggesting sources in fertilizers or pesticides.

Physical and Other Soil Parameters

pH. Soil pH is a measure of the abundance of hydrogen (H⁺) ions in soil solution (Brady 1974). H⁺ ions are derived from rainfall, from carbonic acid formed by carbon dioxide and water in the soil environment, and from organic acids produced by decomposition (Brady 1974, Birkeland 1999). Soluble aluminum (Al³⁺) also affects soil pH, particularly under acidic conditions. Sources of hydroxyl ions (OH⁻) include reactions with cations, Ca²⁺, Mg²⁺, K⁺, and Na⁺ (Brady 1974, Birkeland 1999). Coastal soils of KSC, where calcium carbonate shell material is abundant, are alkaline, while other types are slightly to strongly acid (Tables 2.7, 2.9). Leaching of basic cations from older soils has contributed to their acidity. Coquina scrub soils that have shell material deeper in the soil profile are less acid than acid scrub soils that lack this buffering. The agricultural types (citrus scrub, citrus hammock) are less acidic than the original soil types. This may be due to liming (Brady 1974) or to irrigation with groundwater high in carbonates.

Cation Exchange Capacity. Colloidal material in soil is important to soil structure and ability to supply nutrients to plants. Soil colloids include clay minerals and organic (humic) matter (Brady 1974). Soil colloids carry negative electrical charges, and this negative charge is approximated by the cation exchange capacity (CEC) (Birkeland 1999). CEC varies with clay content and mineral species and organic matter content (Birkeland 1999). KSC soils have little clay content, so organic matter is the main determinant of CEC. Coastal, acid scrub, and coquina scrub soils have very low CEC values (Tables 2.7, 2.9) consistent with the low organic matter content of these soils, while values in hammocks and wetlands, where organic matter accumulation is greater, are substantially higher.

<u>Bulk Density</u>. Bulk density is a measure of the weight of soil per unit volume. It varies with the relative proportion of solid organic matter in the soil and with soil porosity. Typical values for soil surface (A) horizons range from 0.7-2.0

(Birkeland 1999). High organic matter decreases bulk density, but intense cultivation increases bulk density through loss of organic matter (Brady 1974). In KSC soils, bulk density is lowest in wetland soils, hammocks, and flatwoods where organic matter accumulates in surface soils and is highest in coastal and scrub soils with little organic matter accumulation (Tables 2.7, 2.9). Cultivation of citrus hammock soils appears to have increased bulk density, probably through loss of organic matter.

Resistivity. Electrical conductivity increases with the concentration of ionized constituents in soil solution. The predominant cations and anions in most soils are Ca²⁺, Mg²⁺, K⁺, Na⁺, CO₃²⁻, HCO³⁻, SO₄²⁻, Cl⁻, and boron (Bower and Wilcox 1965). Resistivity follows the reciprocal relationship of conductivity. Thus, resistivity is low in saltwater wetland soils (Table 2.7, 2.9) that have high concentrations of Na, K (Table 2.7) as well as Cl⁻ (not measured).

Soil Texture. Mean grain size was greatest in coastal and coquina scrub soils reflecting the shell fragments in these soils that are larger than the quartz grains that comprise much of the rest of the soil. The Wentworth classification reflects these differences better than the Unified Soil classification (Table 2.10). Grain size is less in acid scrub soils where weathering has removed shell fragments. Hammocks and wetlands have the smallest mean grain size, perhaps reflecting accumulation of finer grain material in these types.

2.3.3 Subsurface Soils

The data of this survey apply to surface soils (0-30 cm). Subsurface soil does not have background numbers for comparision and will follow the appropriate screening criteria found in the KSC Decision Process Document.

			. —	
			-	

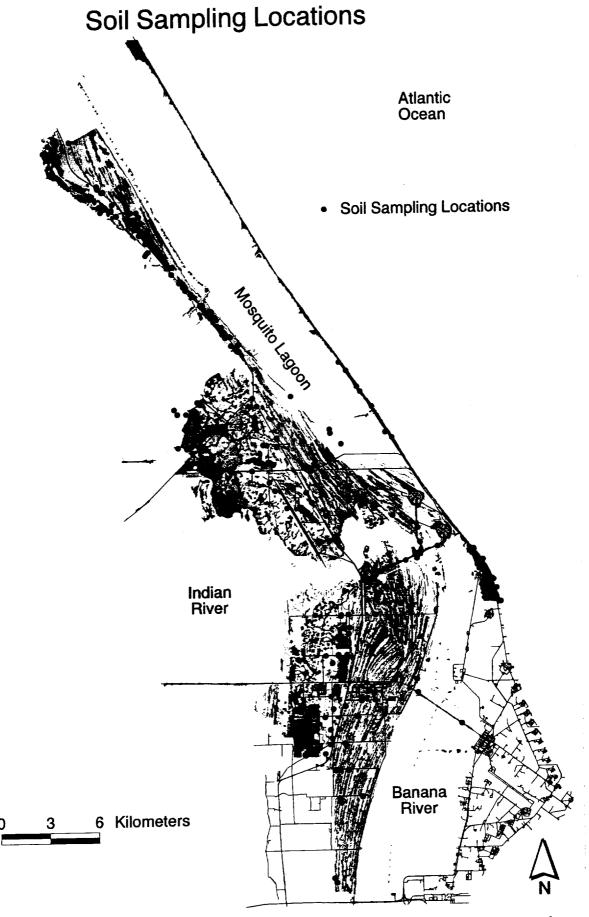


Figure 2.1. Soil sampling locations, Kennedy Space Center. Points may represent more than one sample location.

t	4

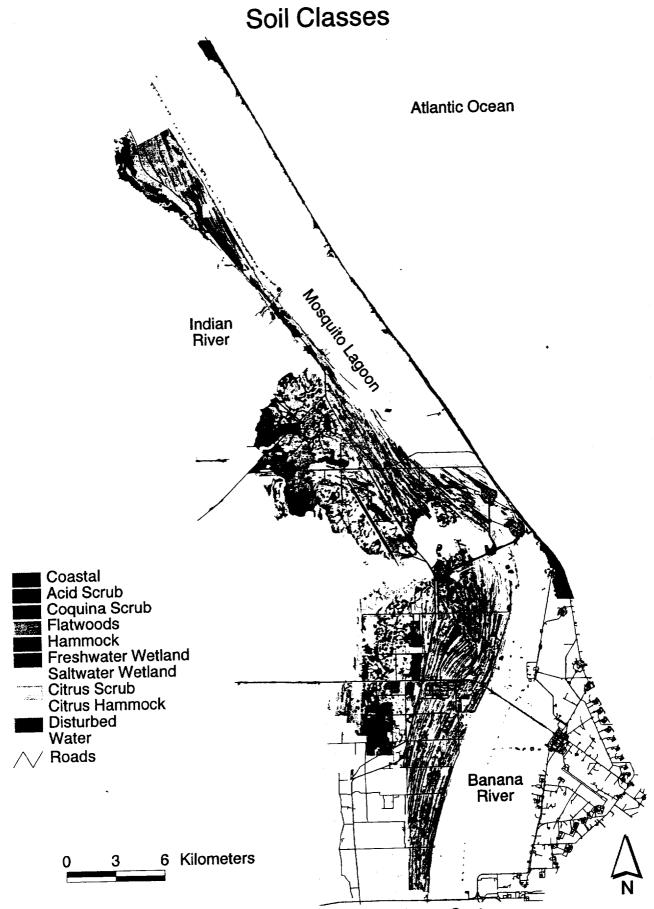
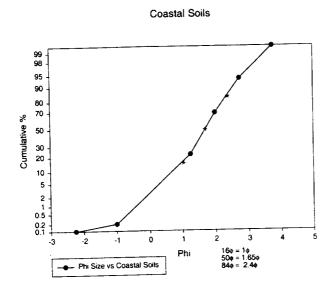


Figure 2.2. Distribution of soil classes on Kennedy Space Center.

Figure 2.3. Phi size distribution for coastal soils.



•

Table 2.1. Soil classification for Kennedy Space Center. Soil series are grouped into ten classes (indicated in bold) based on similarities.

1. Upland

1 1 Well-drained

1.1.1. Recent, coastal, alkaline; vegetation = coastal dunes, coastal strand, coastal scrub

Palm Beach

Canaveral

Welaka

1.1.2. Old, inland, acid; vegetation = scrub, scrubby flatwoods

St. Lucie

Paola

Astatula

Orsino

Pomello

Tavares

Daytona

Cassia

Satellite

1.1.3. Inland, circumneutral, over coquina; vegetation = scrub, xeric hammock

Cocoa

Bulow

1.2. Poorly drained

1.2.1. Acid, sandy; vegetation = flatwoods

Immokalee

Mvakka

Myakka variant

St. Johns

Pompano

Wabasso

Winder

Holopaw

Smyrna

1.2.2. Circumneutral to alkaline, over coquina or limestone;

vegetation = hammocks

Copeland

Parkwood

Bradenton, shallow variant

Tuscawilla

2. Wetland

2.1. Inland, freshwater; vegetation = fresh water marshes, hardwood swamps

Myakka, ponded

St. Johns, ponded

Felda & Winder, ponded

Immokalee, depressional

Placid, depressional

Floridana

Swamp

Anclote

Basinger

Canova

Chobee

Felda & Winder

Pineda

Riviera

Samsula muck

Tequesta muck

Valkaria

2.2. Coastal, brackish to saline; vegetation = saltmarshes, mangroves

Submerged marsh

Tidal marsh

Tidal swamp

Hydraquents

Turnbull muck

3. Agricultural

- 3.1. Active or abandoned citrus on scrub soils (1.1.2, 1.1.3)
- 3.2. Active or abandoned citrus on hammock soils (1.2.2)

4. Disturbed

Canaveral-urban land

Galveston-urban land

Urban land

Quartzipsamments

Arents

Spoil banks

Dikes

Made land

Turnbull variant

Table 2.2. List of parameters, the EPA methods used to analyze each, and detection limits, in soil samples collected for the KSC Background Study.

EPA Method 8081	Lab Reporting Limit for soil 3.3 µg/kg
	3.3 µa/ka
	3.3 ua/ka
8081	
0001	3.3 μg/kg
8081	3.3 μg/Kg
8081	3.3 μg/kg
8081	3.3 μ g/k g
8081	3.3 µg/kg
8081	1 μg/kg
8081	3.3 μ g/kg
8081	3.3 μg/kg
	3.3 μg/kg
	3.3 μg/kg
	3.3 μ g/kg
	3.3 μg/kg
	2 μg/kg
8081	Z pg/kg
8082	33 μg/kg
	33 μg/kg
6062	JO RG/NS
8151	10 μg/kg
0,5,	
8151	10 μg/kg
	8081 8081 8081 8081 8081 8081 8081

Table 2.2 (cont.).

	EPA	Lab Reporting
Chlorinated Harbididay (Method	Limit for soil
Chlorinated Herbicides (cont.) 2,4-Dichlorophenoxy acetic acid (2,4 - D)		
3,5-DCBA	8151	10 μg/kg
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	8151	10 μg/kg
4 - Nitrophenol	8151	10 μ g/k g
Acifluorfen	8151	10 μg/kg
Bentazon	8151	10 μ g/kg
Chloramben	8151	10 μg/kg
Dacthal	8151	10 μ g/k g
Dalapon	8151	10 μg/kg
Dicamba	8151	10 μg/kg
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic	8151	10 μg/kg
aciuj	8151	10 μg/kg
Dinoseb MCPA	8151	10 μ g /kg
MCPP	8151	100 μg/kg
	8151	100 μg/kg
Pentachlorophenol	8151	10 μg/kg
Picloram	8151	10 μg/kg
PAHs		то дужу
1 - Methylnaphthalene	8310	17 μg/kg
2 - Methylnaphthalene	8310	17 μg/kg
Acenaphthene	8310	17 μg/kg
Acenaphthylene	8310	3.3 μg/kg
Anthracene	8310	17 μ g/kg
Benzo(a)anthracene	8310	1.7 μg/kg
Benzo(a)pyrene	8310	
Benzo(b)fluoranthene	8310	2.0 μg/kg
Benzo(g,h,i)perylene	8310	3.0 μg/kg 3.3 μg/kg
Benzo(k)fluoranthene	8310	2.0 μg/kg
Chrysene	8310	2.0 μg/kg 1.7 μg/kg
Dibenzo(a,h)anthracene	8310	3.3 μg/kg
luoranthene	8310	
luorene	8310	3.3 μg/kg
ndeno(1,2,3-cd)pyrene	8310	3.3 μg/kg
laphthalene	8310	1.7 μg/kg
henanthrene	8310	17 μg/kg 1.7 μg/kg
yrene	8310	1.7 μg/kg 1.7 μg/kg

Table 2.2 (cont.).

	EPA	Lab Reporting
	Method	Limit for soil
Metals		
Aluminum	200.7	20 mg/kg
	204.2	2 mg/kg
Antimony	6010	0.5 mg/kg
Arsenic	6010	20 mg/kg
Barium	6010	1 mg/kg
Beryllium	6010	1 mg/kg
Cadmium	6010	25 mg/kg
Calcium	6010	1 mg/kg
Chromium (total)	6010	5 mg/kg
Cobalt	6010/7210	5 mg/kg
Copper	6010	10 mg/kg
Iron	6010	1 mg/kg
Lead	6010	25 mg/kg
Magnesium	6010	1 mg/kg
Manganese	7471	0.01 mg/kg
Mercury	6010/7520	5 mg/kg
Nickel	6010/7610	25 mg/kg
Potassium	6010	2 mg/kg
Selenium	6010	2 mg/kg
Silver	7770	25 mg/kg
Sodium	6010	1 mg/kg
Thallium	6010	1 mg/kg
Vanadium	6010/7950	5 mg/kg
Zinc	0010/7330	<u> </u>
Other Parameters	9045	NA
pH	SM2540G	NA
Percent Solids	Miller	NA
Resistivity	9081	NA
Cation Ion Exchange Capacity	Core Method	NA
Bulk Density	MSA Part2,	NA
Total Organic Carbon	29-3	1 97 1
	Sieve Method	NA
Texture (No. 4,10,40,60,100,200)	Sieve Wethou	

Table 2.3. Area of soil classes.

Soil Class	Area (acres)	Area (hectares)	Percent of Soil Area
			Aica
Coastal	2714.0	1098.3	3.30
Acid Scrub	3847.2	1556.9	
Coquina Scrub	668.2	270.4	4.76
Flatwoods	25779.5		0.81
Hammocks		10432.6	31.32
	4917.7	1990.1	5.97
Freshwater Wetlands	15207.5	6154.3	18.48
Saltwater Wetlands	23786.8	9626.2	
Citrus Scrub	863.1		28.90
Citrus Hammock		349.3	1.05
	1581.6	640.0	1.92
Disturbed	2946.4	1192.4	3.58

Table 2.4. Parameters that were below detection limits (non-detect) in all soil samples.

ORGANOCHLORINE PESTICIDES	CHLORINATED HERBICIDES (8151)
(8081)	
Aldrin	2,4,5-Trichlorophenoxy acetic acid
	(2,4,5-T)
Alpha-BHC	3,5-DCBA
Beta-BHC	4-Nitrophenol
Delta-BHC	Acifluorfen
Endosulfan I	Bentazon
Endosulfan II (beta)	Chloramben
Endosulfan Sulfate	Dacthal
Endrin	Dalapon
Endrin Aldehyde	Dicambia
Heptachlor	Dichloroprop [2-(2-Dichlorophenoxy)
	proponic acid]
Heptachlor Epoxide (a)	Dinoseb
Heptachlor Epoxide (b)	MCPA
Isodrin	MCPP
Methoxychlor	Pentachlorophenol
Mirex	Picloram
Toxaphene	
AROCLORS (8082)	PAHs (8310)
PCB-1016/1242	Acenaphthene
PCB-1221	METALS
PCB-1232	Antimony
PCB-1248	Beryllium
PCB-1254	Cobalt
PCB-1260	Nickel

Table 2.5. Frequency (percent) of values greater than detection limits for chemical parameters in soils. Only parameters greater than detection limits in at least one sample are shown.

Parameter	All Soils	All Soile Coactal	Pico								
	2	i alcono	Scrub	Scrib	riatwoods	наттоск	Freshwater	Saltwater	Citrus	Citrus	Disturbed
Sample Size	220	23	20	2000	,	8	Wetlands	Wetlands	Scrub	Hammock	
		3	77	77	7	0.7	55	21	23	23	20
Organochiorine Pesticides (8081)										3	3
4 4' - DDD	•	(,								
200	4.	5	5	0	4.8	0	0	48	c	0	
4,4 - DDE	10.5	0	0	0	4.8	0	C	0 5	56 E		4.3
4,4' - UDI	4.1	0	0	0	c	-		6.6	20.3	17.4	13.0
Chlordane, alpha or gamma	6.0	0	0		0			4.8	21.7	4.3	8.7
Chlordane, alpha	0.5	0	0	0				0	4.3	0	4.3
Chlordane, gamma	6.0	0	0) c				0	4.3	0	0
Chlordane (Total)	0.5	0	0	0				0	4.3	0	4.3
Dieldrin	4.1	4.3	0	0			o (0	4.3	0	0
Endrin Ketone	0.5	2	٥				0	4.8	21.7	4.3	4.3
Gamma - BHC (Lindane)	0.0	,			0.4	0	0	0	0	0	0
Chlorinated Herbicides			>			0	0	4.8	0	0	4.3
(8151)	-										
2-(2,4,5-	2.7	c	0	c		(
Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)		,)	>	>	0.00		0	0	8.7	13.0
2,4-Dichlorophenoxy acetic acid (2,4 - D)	2.7	0	4.5	0	0	0	0	14.3	0	8.7	0
4-(2,4-	27	4.3	15	-							
Dichlorophenoxy)butyric acid (2,4 - DB)	i	 F	r ř	>	>	0	0	0	13.0	0	4.3
											_

Table 2.5 (cont.).

Table 2.5 (cont.).

Disturbed 100 39.1 8.7 0 100 69.6 69.6 21.7 100 95.7 100 17.4 73.9 8.7 73.9 0 73.9
Citrus Hammock 100 21.7 0 100 91.3 95.7 100 100 52.2 69.6 0 91.3 8.7
Citrus Scrub 100 100 13.0 13.0 100 100 100 100 100 100 100 100 100 1
Saltwater Wetlands 100 100 100 100 100 100 0 0 0 0 0 0 0
Freshwater Wetlands 100 100 13.6 0 0 100 45.5 9.1 54.5 45.5 45.5 0 0 95.5 0 0 59.1 54.5 59.1
Hammocks 100 100 100 70.0 5.0 75.0 75.0 75.0 85.0 85.0 85.0
Flatwoods 100 4.8 0 0 100 76.2 95.2 19.0 19.0 19.0 0 0 0 0 19.0
Scrub Scrub 100 100 100 100 100 100 100 100 100 10
Acid Scrub 100 100 13.6 13.6 13.6 13.6 13.6 13.6 13.6 13.6
Coastal 100 100 100 100 100 100 100 100 100 10
All Soils 30.0 1.4 1.8 1.8 1.8 1.8 1.8 1.8 1.8 1.8 1.8 1.8
Parameter Metals Aluminum Arsenic (as carcinogen) Barium Cadmium Calcium Chromium (total) Copper Iron Lead Manganese Mercury (inorganic) Potassium Selenium Selenium Silver Sodium Thallium Vanadium

Table 2.6. Parameters that were determined by the Remediation team to be treated as below detection limits (non-detect) for all soil locations based on low frequency of detection.

ORGANOCHLORINE PESTICIDES (8081)
Chlordane alpha	_
Chlordane gamma	
Chlordane (total)	
Endrin Ketone	
Gamma -BHC (Lindane)	

Table 2.7. Means and standard deviations (in parenthesis) of chemical parameters in soils. Only parameters greater than detection limits in at least one sample are shown. ND = below detection limits.

rarameter	All Soils	All Soils Coastal	Arid	Soci in	F1-4						
			Scrib	Sodellia	Flatwoods	riatwoods Hammocks	<u> </u>	Saltwater	Citrus	Citrus	Disturbed
Sample Size	220	23	22	200	21	CC	Wetlands	Wetlands	Scrub	Hammock	
Organochlorine Pesticides	L			•	17	20	7.7	21	ຊ	23	23
(8081)											
4,4' - DDD	5.6	Q	Q	9	2.3	CN	2	100	1		
(µg/kg)	(6.4)				(4.1))	2	8.00	2	2	2.2
4,4 - DUE	4.5	9	QN	QN	2.1	2	Q	5.3	14.1	5.4	(1.7)
4 4' - DOT	6.	2			(0.7)			(6.9)	(22.9)	(13.6)	5.7 (9.0)
(ua/kg)	- 4.	2	 2 2	2	2	QN	QN	7.9	12.0	19	6.0
Chlordane (Total)	0.1.	2						(20.3)	(22.5)	(0,8)	(14.7)
(µa/ka)	, (c	2		2	2	2	QN	9	4.4	ND	S
Chlordane (alpha)	2.4	S	S	Ş	2				(12.6))
(нд/кд)	(3.2))	?	2		2	Q 2	2	2.0	2	Q
Chlordane (gamma)	2.4	2	2	2	S	CN		1	(1.2)		
(нд/кд)	(7.1)			!	2	2	2	- -	ص ز م ز	Q	2.0
Dieldrin	4.3	1.8	2	QN	CN	2			(0.7)		(0.8)
(µg/kg)	(17.4)	(0.3)		<u> </u>		2	2		13.9	10.0	5.0
Endrin Ketone	2.2	QN	2	S	00			(5.3)	(30.1)	(39.3)	(0.7)
(µg/kg)	(1.1)		!)	(0.9)		<u>_</u> _	Q	9	2	QN
Gamma - BHC (Lindane)	2.2	QN	2	9	ND	QN	CN	0 7			
(μg/kg)	(1.3))	2	5. č	2	O Z	2.0
Chlorinated Herbicides								(3.5)			(0.8)
(8151)											
2-(2,4,5-Trichlorophenoxy)	9.2	S	2	S	CZ	6.0	2				
propionic acid (2,4,5 - TP)	(22.4)			<u> </u>)	, (c	 2 2	 Q	O.	9.6	22.6
(Silvex) (µg/kg)						(5.6)				(12.3)	(0.99)
2,4-Dichlorophenoxy acetic	8.0	9	7.0	S S	S	S	2	,			
acid (2,4 - D) (μg/kg)	(8.6)		(7.4))	2	<u> </u>	ر - ر م م رو		1.6	Q
4-(2,4-Uchlorophenoxy) butyric acid (2,4 - DB)(un/kn)	0.7 6.5	5.8	6.7	Q	QN	QN	9	ND (S.S.)	4.6	() CN	0
(B) bLV	7/-//	(5.3)	(5.9)						(10.9))	(12.5)

35.1 (113.4) 13.2 (43.3) Disturbed 6.2 (12.5) 16.3 13.9 13.9 (45.4) 19.0 (40.2) 12.2 (30.7) 15.0 (41.4) 2 13.3 (13.9) 12.4 (10.2) 2 Hammock 2.1 (3.5) Q. 3.1 (3.4) 1.1 (0.6) ND Citrus S 1.4 (1.1) 10.4 (6.9) 9 2.1 3.5 14.2 (17.1) 2.8 (3.6) Citrus Scrub 2.5 (2.6) 2.9 (3.5) 3.5 (4.7) 20.6 (36.7) 1.9 (2.4) 3.6 (5.2) Q N Saltwater Wetlands 2 2 Q 9 S 皇 2 QN 皇 2 2 S 2 S Freshwater Wetlands 9 2 Q 2.4 (4.4) 2 g 皇 2 9 2 2 9 9 2 Flatwoods | Hammocks 10.9 (37.5) 13.2 (39.6) 1.9 (2.0) 3.1 (2.4) 4.8 (7.0) (3.0) 1.9 1.8 1.7 1.7 13.6 (7.8) QN 9 ND 2 3.2 ND ND 2 2 2.1 (3.2) 2 문 2 S S 2 2 Coquina Scrub 2.7 (2.6) ND QN 12.6 (10.5) QN (2.3) ND 14.8 (16.7) 2 S 2 QN 2 1.9 (6.19 2 Acid Scrub 2 21.8 (37.4) 1.8 (1.5) 2 2 17.9 (20.9) 2 Q 9 2 QN 1.1 (0.9) 2 1.1 (0.5) QN Coastal 4. E. 21.0 (58.6) 1.4 1.9 (1.2) 1.0 2 3.8 (9.9) 9 9 2.9 (14.3) All Soils 3.1 (13.9) 7.2 (39.2) 3.4 (11.6) 3.8 (15.1) 2.9 (10.4) 4.8 (14.1) 3.1 (4.9) 3.4 (17.2) 13.3 2.4 (9.0) 14.1 (15.8) 2.9 (4.3) Indeno(1,2,3-cd)pyrene Dibenz(a,h)anthracene 2 - Methylnaphthalene - Methylnaphthalene Benzo(b)fluoranthene Benzo(k)fluoranthene Benzo(g,h,i)perylene Benzo(a)anthracene PAHs (8310) Benzo(a)pyrene Acenaphthylene Fluoranthene Anthracene Chrysene Fluorene Parameter (µg/kg) ug/kg) ug/kg) ug/kg) ug/kg) μg/kg) ug/kg) (µg/kg) ug/kg) μg/kg) (µg/kg) (µg/kg)

Table 2.7 (cont).

Table 2.7 (cont.).

Parameter	All Soils	Coastal	Acid	Coquina	Flatwoods	Hammorke	Front	_ h			
PAHs (cont.)	1		Scrub	Scrub		200	Wetlands		Citrus	Citrus	Disturbed
Naphthalene	, 60						COLIBINO	Wellands	Screb	Hammock	
(µa/ka)	4.00	(15.2	2	74.2	13.2	CN				
Phenanthrene	(88.3)	(2.4)	(12.7)		(217.3)	(6.9)	2	ON .	107.9	18.6	QN
(µg/kg)	(10 5)	9.5	2.5	QN	9	5.6	S	2	(209.1)	(25.9)	
Pyrene	5.1	2 2	(5.2)			(19.4))	€	11.0	Q	12.2
(нд/кд)	(28.6)	2	2	4. 5	Q	21.7	QN	CZ	(4/./)		(31.2)
Metais	(0:00)			(1.3)		(71.4)	?	2	ے <u>۔</u> ت ز	O Z	18.8
Aluminum	1444 G	282 €	0000	1000							(26.2)
(mg/kg)	(2576.4)		(306.8)	6.1001	164.7	2395.0	2315.9	2683.8	11722	1001	
Arsenic (as carcinogen)	0.8	+	0.6	0.3	04.5	(2382.4)	(3148.0)	(1997.6)	(919.4)	(1768.2)	2491.3
Basisse	(1.1)	_	(1.8)	(0.5)	; ;	4. 6	9.0	1.3	0.4	0.5	3003.0)
(ma/ka)	13.1	-	2	2	CN	(V.4)	(0.6)	(1.1)	(0.3)	(0.7)	(0.5)
Cadmim	(6.7)	1)	 2 2	2	Q N	QN	QN	13.0
(ma/ka)	9 6	0.5	2	QN	QN N	CN	2				(6.8)
Calcium	(6.3)	(0.1)	\dashv			?	2		9.0	QN	S S
(mg/kg)	(36100)	50044		22057	383	11278	9600	14007	(0.4)		
Chromium (total)	106100	18497	135	(65351)	(211)	(20935)	(29489)	(1027)	2100	11099	43053
(mg/kg)	5 (g		8.0	2.3	QN	3.5	3.1	103/3)	3060)	(24043)	(60942)
Copper	230	0.0	(1.2)	(1.3)		(3.2)	(4.3)	4. မ ပ မ	12.2	1.8	3.9
(mg/kg)	(27.0)		ري اي زو	 Q	2.8	4.3	47	\dagger	0 5	(9.5)	(6.1)
Iron	1283 5	+	+		(0.7)	(3.8)	(4.8)		23.5	54.0	2.6
(mg/kg)			40.4	1511.8	144.1	1717.5	1326.1	1504 B	1500.4	(40.6)	(0.8)
ead	+	_	+	003.9)	(81.3)	၈	(2361.4)		(and A)	1984.8	1783.4
(mg/kg)	(11.3)	_	ر. د و	0.6	2.4		5.1	+	4.000	(2306.4)	(4059.3)
Magnesium	+	+	+	(9:10	(2.5)		(4.9)	33.00) i	9.0 9.0 9.0	5.1
(mg/kg)	_			87.6	112.3	534.1	十	3201.0	0.56	(3.8)	(4.6)
Manganese	+-	7	10.6)	37.7)	(60.5)	(836.0)	(468.5)	(3151.2)	(189.3)	3/7.4	288.9
(mg/kg)	\dashv		\dashv	44.3)	6.0 8.0	14.7	9.6	1_	82.0	58.7	7.9
							(50.4)	(10.5)	(52.1)	(51.1)	(6.4)

Table 2.7 (cont.).

Parameter	All Soils	Coastal	Acid	Coquina	Flatwoods	Hammocks	Freshwater	Saltwater	Citrus	Citrus	Disturbed
			Scrub	Scrub			Wetlands	Wetlands	Scrub	Hammock	
Metals (cont.)											
Mercury (inorganic)	0.157	QN	QN	900.0	0.007	9660.0	0.0375	0.0179	0.012	0.0161	0.0108
(mg/kg)	(0.262)			(0.002)	(0.003)	(0.0523)	(0.0471)	(0.0162)	(0.0116)	(0.0158)	(0.0166)
Potassium	229.0	58.4	42.6	74.0	57.0	114.6	52.5	1631.0	94.6	121.7	72.7
(mg/kg)	(673.7)	(37.8)	(33.1)	(39.3)	(41.8)	(113.8)	(65.6)	(1620.5)	(110.7)	(121.0)	(54.6)
Selenium	1.2	QN	ON	QN	QN	1.5	1.7	ΩN	ON	QN	1.2
(mg/kg)	(0.8)					(1.0)	(1.1)				(0.0)
Silver	1.3	1.0	QN	1.1	QN	an	QN	QN	QN	QN	1.1
(mg/kg)	(0.7)	(0.1)		(0.5)							(0.4)
Sodium	1459.3	461.3	12.6	174.0	48.7	168.5	169.4	13771	17.0	38.3	336.3
(mg/kg)	(6852.2)	(169.3)	(5.1)	(471.8)	(34.6)	(227.8)	(311.2)	(18366)	(21.6)	(67.2)	(444.0)
Thallium	0.7	9.0	QN	QN	QN	QN	QN	QN	9.0	6.0	QN
(mg/kg)	(0.5)	(0.8)							(0.4)	(1.0)	
Vanadium	3.2	2.4	2.0	2.2	8.0	5.2	3.7	7.4	2.0	4.5	3.4
(mg/kg)	(4.5)	(0.2)	(0.5)	(1.0)	(0.6)	(5.5)	(5.6)	(6.3)	(1.1)	(5.1)	(6.3)
Zinc	15.1	9.9	5.2	5.0	7.7	12.8	13.6	2.7	49.2	29.2	11.8
(mg/kg)	(21.2)	(5.1)	(3.8)	(3.1)	(5.9)	(23.7)	(11.1)	(4.8)	(36.3)	(23.1)	(11.4)
Other Parameters											
Hd	5.1	8.1	5.2	5.8	4.4	4.9	4.8	6.3	6.5	6.1	5.8
Bulk density	96.0	1.19	0.99	1.05	0.72	0.74	0.72	0.71	1.13	1.18	1.11
(g/cm3)	(0.27)	(0.13)	(0.14)	(0.16)	(0.15)	(0.24)	(0.26)	(0.33)	(0.15)	(0.09)	(0.21)
Resistivity	614762	586261	681467	1712941	473600	440645	151027	173	870130	749750	722345
(ohm-cm)	(792326)	(660162)	(806088)	(599591)	(368519)	(866900)	(265841)	(184)	(855419)	(910291)	(891528)
Cation exchange	14.0	2.5	2.9	3.8	17.8	23.5	28.7	18.3	19.4	11.3	10.6
capacity	(14.2)	(1.5)	(1.6)	(2.3)	(7.4)	(18.6)	(19.7)	(0.6)	(17.2)	(8.1)	(9.6)
(meq/100g)	[n=202]	[n=16]	[n=22]	[n=22]	[n=21]	[n=20]	[n=22]	[n=21]	[n=17]	[n=23]	[n=18]
(med/100g)	=	111=10	111=661	111=661	1112=11	וו≔בען	111=44	=4	[1]	_	[11=43]

Table 2.8. Analysis of variance of soil metals among soil classes. Only metals above detection limits in sufficient samples were analyzed. Data were transformed (log₁₀) before analysis. All ANOVAs were significant at p < 0.001. Table shows means (mg/kg) and multiple comparisons made using Games-Howell test. Soil classes not significantly different (p >0.05) are indicated by the same letter following the mean.

	\neg			_		т-		Т		_		_		_		_		-				_						
Disturbed		2491.3	ည္အ	0.70	ရှင	43053.5	Se	3.85	₹	2.55	AB	1783.4	B	5 06	ရှိသူ	0 000	8.00.0 BD	7.0		0.0108	AD	727	AB	336.3		3.44	e E	11.79 ARFF
Citrus	nammock 1991 6	1821.3	١	0.53	В	11099.1	8	11.79	2	54.04	۵	1984.8	a	5.84	8	377.4	S	58 B		0.01613	BCDE	121.7	AB	38.3		4 46	BC 2	29.20 CF
Citrus	3C(UD	7.7		0.36	8	2100.9	۵	12.18	۵	53.55	۵	1502.2	ω	9.70	۵	232 57	CEG	82.0	ď	0.01204	ACD	94.57	AB	17.0	AF	202	8	49.23 C
Saltwater Wetlands	2602 0	2003.0 B	3 ,	25.1	AC	11627.1	S	4.51	AC	5.48	ပ	1504.8	Δ	16.58	8	32910		10.10	ပ	0.01786	BCD	1631.0	۵	13771.4	ш	7.35	O	7.73 AE
Freshwater Wetlands	2315.0		930		מ	9609.5	CD	3.07	AE	4.70	ည္ထ	1326.1	В	5.06	B C	316.0	BDC	9.61	BC	0.0375	8	95.2	AB	169.5	BC	3.65	80	13.60 BDEF
Hammocks	2395.0	BC	77.0	† a	۵	11278.0	3	3.47	A	4.33	BC	1717.5	В	6.24	8	534.1	DE	14.7	ပ	0.0396	ပ	114.6	BC	168.5	O	5.16	BC	12.77 AB
Flatwoods	164.7	<	0.30	§ a	, 650	383.7	a	0.57	8	2.82	80	144.1	٨	2.44	AB	112.3	BC	0.93	ш	0.0065	AE	57.0	ABC	48.7	BCG	0.79	A	7.70 AD
Coquina Scrub	1001.6	ပ	0.31	2	0 03000	22020.8	BCD	2.26	V .	2.55	AB	1511.8	20	4.00	ပ	9.78	8	54.2	V	0.008	AB	74.0	8	174.0	ABCG	2.21	Ф	5.00 A
Acid Scrub	202.8	∢	0.63	a	150 4	4.00	<	ې پې تو	빏	3.25 C.Z.C	מ	240.4	∢ .	.53	۷	28.0	4	2.53	В	0.005	V.	45.6	٧	12.6	٨	0.72	∢	5.24 A
Coastal	282.6	∢	2.33	<	50042 E	30043.3			¥ ,	 9. <	<	1063.0	n i	2.16	AB	370.0	۵	11.34	ပ	0.0059	V	58.4	ABC	461.3	۵	2.36	m	6.60 AB
Parameter	Aluminum		Arsenic (as	(carcinoden)	Calcium			(total)	Copper	jaddoo		Lou	7 - 7	Lead		Magnesium		Manganese		Mercury	(Inorganic)	Potassium		Sodium		Vanadium	i	Zinc

Table 2.9. Analysis of variance among soil classes for pH, bulk density, resistivity, and CEC. Data for resistivity and CEC were transformed (log₁₀) before analysis. All ANOVAs were significant at p < 0.001. Table shows means and multiple comparisons made using Games-Howell test. Soil classes not significantly different (p >0.05) are indicated by the same letter following the mean.

								1	4.14.0	Diet irbod
Darameter	Coastal	Acid	Coguina	Flatwoods	Hammocks	Freshwater	Saltwater	CILLUS	Cillus	
		di ri	Scrub			Wetlands	Wetlands	Scrub	Hammock	
		25								
						,	6	2.0	L	8 4
I	œ	5.2	5.8	4.4	4. Q.	φ. Σ	5.0	0.0		3 6
	•	C	<u> </u>	æ	DE	빙	۵	DE	DE	a
	3	}	3			0.40	0.71	1 13		-
Bulk density	1.19	66.0	1.05	0.72	D./4	0.72	- -	<u>2</u> :		: <
(~/om ₃)	•	α	AB	C	ပ	ပ	ပ	AB		AB
(g/cm)	ς				21.001.	174007	470	070100	L	722345
Becietivity	586261	681467		473600	440645	120161	2	0/0130		25010
() ()		<	α	4	AD	_	ပ	∢		AD
(ma-muo)	AD	٢	2			- 3	3	, 0	L	40.6
Cation	2.5	5.9	3.8	17.8	23.5	7.87	18.3	j.		2.6
	; i	•	•	(٥	٢	Z.	<u>د</u>		08 -
exchange	∢	∢	₹	2	3	>	})		
capacity										
(2004/1000)										

Table 2.10. Mean grain size and classification of soil types.

Soil Class	Mean	Mean Grain	Wentworth	Unified Soil
	Grain Size (Phi)	Size (mm)	Classification	Classification
Coastal	1.7	0.31	Medium Sand	Fine Sand
Acid Scrub	2.6	0.17	Fine Sand	Fine Sand
Coquina Scrub	1.7	0.31	Medium Sand	Fine Sand
Flatwoods	2.2	0.22	Fine Sand	Fine Sand
Hammocks	2.8	0.14	Fine Sand	Fine Sand
Freshwater Wetlands	2.6	0.17	Fine Sand	Fine Sand
Saltwater Wetlands	2.85	0.14	Fine Sand	Fine Sand
Citrus Scrub	2.2	0.22	Fine Sand	Fine Sand
Citrus Hammock	2.65	0.16	Fine Sand	Fine Sand
Disturbed Soils	2.45	0.18	Fine Sand	Fine Sand

Table 2.11. Concentrations of selected elements in typical surface soils of Florida (Shacklette and Boerngen 1984). Samples were taken at 20 cm below the surface.

Element	Concentration
	(ppm)
Aluminum	<20,000
Antimony	<1
Arsenic	<3
Barium	<200
Beryllium	<1
Calcium	<3000
Chromium	<20
Cobalt	<3
Copper	<10
Iron	<10,000
Lead	<10
Magnesium	<1500
Manganese	<150
Mercury	<0.032
Nickel	<5
Potassium	<6800
Selenium	<0.2
Sodium	<2000
Vanadium	<20
Zinc	<17

Table 2.12. Background levels of trace elements in surface soils. Data are from Kabata-Pendias and Pendias (1984). Units are ppm/dry weight. — indicates no data reported.

Sandy Soils-United States

Podzols and Sands – Different Countries

Element	United State		Different C	Countries
Liement	Range	Mean	Range	Mean
Antimony				
Arsenic	<0.1-30.0		0.05-1.33	0.19
Barium		5.1	1.1-28.9	2.4-5.8
Beryllium	20-1500	400	180-780	-
	<1-3	1.9	-	•
Cadmium	-		0.01-1.80	0.07-0.43
Chromium	3-200	40	1.4-530	21-110
Cobalt	0.4-20	3.5	0.1-65	21-110
Copper	1-70	14	1-52	0.40
Lead	<10-70	17		8-16
Manganese	7-2000	345	2.3-47.5	10.4-57.0
Mercury	<0.01-0.54		9-1900	-
Nickel	<5-70	0.08	0.01-0.70	0.04-0.06
Selenium	 	13.0	1-52	7-11
Vanadium	0.005-3.5	0.5	0.05-1.32	0.14-0.27
	7-150	47	10-91	-
Zinc	<5-164	40.0	3.5-146	24-42

SECTION 3 Groundwater

3.1 Methods

3.1.1 Site selection

We used the four subsystems of the Surficial aquifer (Figure 1.4), originally delineated by Clark (1987c), as the basis for sample site locations. We located six sample sites in each subsystem, 24 total sites (Figure 3.1). We chose sites away from known contamination. One to three wells were installed at these sites. The sampling plan designated that a shallow well was to be installed at each site. Intermediate wells were to be installed at four sites per subsystem (16 total); deep wells were to be installed at three sites per subsystem (12 total). A total of 52 wells were planned.

A Trimble Pathfinder Professional XL Geographic Positioning System (GPS) unit was used to collect groundwater well locations using real-time differentially corrected methods (Trimble Navigation Limited 1994). Locations were exported to North American Datum 1927 (NAD27) State Plane coordinates for overlay in a Geographic Information System (GIS). Sample site descriptions of well locations are given in Appendix C, Table C-1, and coordinates of sites in Table C-2.

3.1.2 Sampling

The NASA/KSC Remediation Program Office contracted groundwater well installation to Universal Engineering Sciences, Inc. [Appendix D (Universal Engineering Sciences, Inc. 1998)]. Universal was contracted to install 52 wells, due to the depth of the confining unit at one location the deep well was not installed. Therefore, a total of 51 wells were installed at varying depths.

Groundwater samples were collected from 51 wells (Figure 3.1). Twenty-four wells were shallow (15 ft.), 16 wells were intermediate (35 ft.), and 11 wells were deep (50 ft.). Shallow and intermediate wells were screened for 10 ft (3 m) and deep wells for 5 ft. (1.5 m) (Universal Engineering Sciences, Inc. 1998). ENCO was subcontracted for sample collection and chemical analysis since they have the required certifications. ENCO followed the requirements of the "NASA/KSC Sampling and Analysis Plan for RCRA Facility Investigations at Kennedy Space Center, Florida and Cape Canaveral Air Station, Florida, Volume 4" of the Generic Work Plans (SAP) (NASA 1996) and ENCO's FLDEP Comprehensive Quality Assurance Plan No. 960038. Based on requirements in the SAP, organic vapor and explosimeter readings were collected from every well. All wells had a reading of zero for both measurements.

Well construction information (Universal Engineering Sciences, Inc. 1998) was consistent with well placement in the Surficial Aquifer; no confining layers were encountered in completed wells.

All well sampling was done using a low flow peristaltic pump (with Teflon check valves) and silicon and Tigon tubing. All sampling protocols in the SAP were followed. Turbidity was measured in the field throughout purging, and sample collection occurred when turbidity was <10 NTU). Quality assurance/quality control samples were collected in accordance with the requirements of the SAP. A data validation report was completed, see Appendix G.

3.1.3 Chemical Analysis

Table 3.1 lists the analyses that were conducted on each groundwater sample. Organochlorine pesticides (EPA method 8081), aroclors (EPA method 8082), chlorinated herbicides (EPA method 8151), and polycyclic aromatic hydrocarbons (EPA method 8310) were analyzed. Methods for organochlorine pesticides, aroclors, and chlorinated herbicides use gas chromatography (USEPA, 1996). Polycyclic aromatic hydrocarbons were analyzed by highpressure liquid chromatography (USEPA 1996). Target analyte list (TAL) metals were analyzed using one of the following EPA method numbers appropriate for the specific metal: 200.7, 204.2, 220.1, 245.1, 258.1, 273.1, 279.2 (Table 3.1). Method 200.7 uses inductively coupled plasma-atomic emission spectrometry (USEPA 1983). Method 204.2 (Sb) uses an atomic absorption, furnace technique (USEPA 1983). Methods 220.1, 258.1, and 273.1 use atomic absorption, direct aspiration techniques (USEPA 1983). Mercury (Hg) was analyzed using method 245.1, a cold-vapor, atomic absorption technique (USÉPA 1983). Chloride was analyzed using EPA method number 325.3, a titrimetric mercuric nitrate technique (USEPA 1983).

Field analysis of groundwater samples for dissolved oxygen, turbidity, pH, specific conductivity, and temperature were conducted with calibrated instruments. Total dissolved solids was determined by method 160.1 using filtration followed by drying at 180 C (USEPA 1983). Total organic carbon was determined by method 415.1 (USEPA 1983) using catalytic combustion or wet chemical oxidation.

3.1.4 Data Analysis

Data analyses was conducted using SPSS Version 9 (SPSS Inc., 1999) using the following approach.

- Frequency of parameters occurring above the detection limits were determined. Where most of the data are below detection, intensive data analyses cannot be justified.
- Boxplots were used to scan for obvious outliers in the data. Outliers could arise from a contaminated sample or a local, anomolous site conditions. Inclusion of outliers can skew the mean and standard deviation of a parameter.
- Normality of distributions were tested for using normality plots and the Kolmogorov-Smirnov statistics with Lilliefors significance levels. Normality of

- distribution determines whether parametric or nonparametric analyses should be used or whether transformations are required.
- Where the sample size of data above detection was sufficient, two-way analysis of variance (ANOVA) of transformed data (log₁₀) were used to test for differences related to aquifer subsystem and depth. Replicate samples were excluded from ANOVA. These tests determined whether parameters differed among subaquifers and depths. If such differences exist, then use of a single mean (or screening value) to represent all groundwater may not be appropriate.

For screening purposes, data are given as twice the mean (USEPA 1998). The KSC Background Screening table for groundwater is located in Appendix A, Table A-2.

3.2 Results

3.2.1 Groundwater Chemistry

Frequency of groundwater parameters above detection is given in Table 14. All organochlorine pesticides (25), all aroclors (6), and all chlorinated herbicides (18) were below detection in all samples (Table 3.3). Ten PAHs were above detection in one to four samples. Four PAHs were above detection limits in only one sample each, and these values were between the method detection limit and the practical quantitation level; therefore, the Remediation team decided to treat these as below detection (Table 3.4). Cobalt and mercury were always below detection (Table 3.3). Beryllium, silver, and zinc were above detection in just one sample each (Table 3.2).

For some parameters, frequency of occurrence above detection varied with subaquifer or depth (Table 3.5). PAHs were above detection only in shallow wells of the dune, dune-swale, and west plain subaquifers (Table 3.5). Arsenic and cadmium were most frequently above detection in the dune subaquifer.

Summary data by aquifer and depth are given in Table 3.6. Among metals only AI, Ca, CI, Fe, Mg, Mn, K, and Na were above detection in sufficient samples to compare concentrations. In addition, total dissolved solids (TDS), total organic carbon (TOC), specific conductivity, pH, dissolved oxygen (DO), and field temperature could be compared. The distributions of all parameters except temperature were not normal (Kolmogorov-Smirnov tests, p<0.001). Transformations (log₁₀) improved normality of distribution. Transformed values of Mg, K, Na, TOC, TDS, and conductivity did not differ from normal (p>0.05). Although distributions of AI, Ca, CI, Fe, Mn, and DO still differed from normal, normality plots showed that transformation improved the distributions such that ANOVA could be conducted.

Univariate, two-way ANOVA where subaquifer and depth were fixed effects was conducted using temperature and log-transformed chemistry data. Models were not significant for Al and DO; the significance level for the model (p=0.057) for Fe was marginally above the usual decision point. Interactions were significant for pH. TOC differed across subaquifers but not depths. Ca, Cl, Mg, Mn, K, Na, TDS, and conductivity differed with subaquifer and depth (Table 3.7).

For screening purposes, a simpler classification of groundwater was desired. The NASA/KSC Remediation team adopted Rule 62-520.410 Florida Administrative Code (F.A.C) groundwater classifications Class G-II and Class G-III. Class G-II is defined as groundwater with total dissolved solids <10000 mg/L and Class G-III is defined as groundwater with total dissolved solids >10000 mg/L. Some parameters (Table 3.8) were more frequently above detection in Class G-II (Cd, Fe, Tl) and some were more frequently above detection in Class G-III (Sb, As, Pb). Ca, Cl, Mg, K, Na, total dissolved solids, and conductivity were substantially higher in Class G-III (Table 3.9), as would be expected. The KSC Background screening table for groundwater is given in Appendix A, Table A-2. Detailed data for Class G-II and Class G-III are given in Appendix C, Table C-3 and C-4.

3.3 Discussion

Unconsolidated, surficial aquifers are subject to contamination from point sources and from general land use, and contaminants may include trace elements, pesticides, herbicides, and other organics (Page 1981, Burkart and Kolpin 1993, Kolpin et al. 1995, 1998, Barbash et al. 1999). Urban and agricultural land uses have affected some Florida aquifers (Rutledge 1987, Barbash and Resek 1996). Point source contamination to the KSC aquifer has occurred at certain facilities (Clark 1985, 1987a, b).

However, the baseline data presented here suggest that widespread contamination of the Surficial aquifer on KSC has not occurred. No organochlorine pesticides, aroclors, or chlorinated herbicides occurred above detection limits. Although pesticide residues or degradation products and chlorinated herbicides occurred in some soils, those concentrations were low and migration into the aquifer either has not occurred or has not been widespread.

Some PAHs occurred in the shallow wells. Earlier we showed that PAHs occur in a variety of KSC soils at relatively low concentrations. Some occurrence in shallow wells is not surprising. PAHs have both natural and anthropogenic sources (e.g., Suess 1976, Standley and Simoneit 1987, Jones et al. 1989b, c).

Clark (1987c) tested background groundwater wells for a smaller number of organochlorine pesticides and chlorinated herbicides including 2,4-D, 2,4,5-TP, Endrin, Lindane, Methoxychlor, and Toxaphene and found none above detection limits.

Most trace metals are in low concentrations in KSC groundwater if they occur above detection levels. This is consistent with the low concentrations of most trace metals in KSC soils and the primarily quartz composition of the terrigenous deposits comprising the surficial sediments of Merritt Island (Brown et al. 1962, Milliman 1972, Field and Duane 1974). Al, Fe, and Mn occur above detection limits more frequently than other trace metals. Al and Fe are abundant crustal components and are present in KSC soils. Intense leaching, particularly in acid scrub and flatwoods soils, mobilizes Al and Fe (Paton et al. 1995). Iron is a typical constituent of groundwater in the Surficial aquifer in Florida (Miller 1997). Mn is one of the most abundant trace elements (Kabata-Pendias and Pendias 1984); it is present in KSC soils but the concentrations are relatively low. Solution and precipitation of Fe and Mn are affected by pH and oxidation-reduction conditions.

The chemical parameters varying most with subaquifer and depth are Ca, Cl, Mg, K, and Na, and conductivity and total dissolved solids that are related to these cations and anions. The trends are generally consistent among these; the shallow wells in the Dune-Swale subaquifer have the lowest values. Concentrations increase with depth within a subaquifer. At a given depth, concentrations in the Dune-Swale and West Plain subaquifers are lower than in the Dune and Marsh subaquifers. These trends reflect increased mineralization with depth and differences between the fresh water Dune-Swale and West Plain subaquifers and the more saline Dune and Marsh systems. The Dune and Marsh subaquifers interact with saline water of the Atlantic Ocean and Indian River Lagoon system, respectively (Clark 1987c).

These data are generally consistent with the previous study of Clark (1987c) (Table 3.10), although there were differences in sampling and analysis.

			•
•			
	•		
		• •	

Groundwater Well Locations

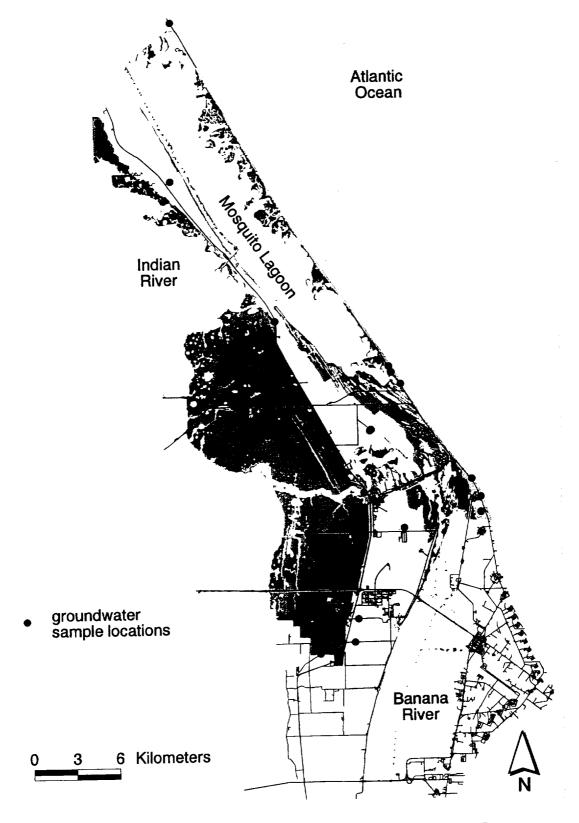


Figure 3.1. Groundwater well sampling locations, Kennedy Space Center. Points may represent more than one well.

ŧ	÷

Table 3.1. List of parameters, EPA method number, and lab reporting limit for parameters analyzed for all groundwater samples. Lab reporting limit values are based on pure samples. * = measurement made with a calibrated field instrument (YSI).

	EPA	Lab Reporting
	Method	Limit for groundwater
Organochlorine Pesticides		
4,4' – DDD	8081	0.05 μg/L
4,4' – DDE	8081	0.05 μg/L
4,4' – DDT	8081	0.05 μg/L
Aldrin	8081	0.05 μg/L
Alpha – BHC	8081	0.05 μg/L
Beta – BHC	8081	0.05 μg/L
Chlordane (alpha)	8081	0.05 μg/L
Chlordane (gamma)	8081	0.05 μg/L
Chlordane (Total)	8081	1 μg/L
Delta - BHC	8081	0.05 μg/L
Dieldrin	8081	0.05 μg/L
Endosulfan I	8081	0.05 μg/L
Endosulfan II (beta)	8081	0.05 μg/L
Endosulfan Sulfate	8081	0.05 μg/L
Endrin	8081	0.05 μg/L
Endrin Aldehyde	8081	0.05 μg/L
Endrin Ketone	8081	0.05 μg/L
Gamma – BHC (Lindane)	8081	0.05 μg/L
Heptachlor	8081	0.05 μg/L
Heptachlor Epoxide(a)	8081	0.05 μg/L
Heptachlor Epoxide(b)	8081	0.05 μg/L
Isodrin	8081	0.05 μg/L
Methoxychlor	8081	0.05 μg/L
Mirex	8081	0.05 μg/L
Toxaphene	8081	2 μg/L
Aroclors		
PCB-1016/1242	8082	1 μg/L
PCB-1010 12-42	8082	1 μg/L
PCB-1232	8082	1 μg/L
PCB-1248	8082	1 μg/L
PCB-1254 (noncarcinogenic)	8082	1 μg/L
PCB-1260	8082	1 μg/L

Table 3.1 (cont.).

	EPA	Lab Reporting
Chlorinated Herbicides	Method	Limit for groundwate
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP)		
(Gilvex)	8151	0.5 μg/L
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	8151	0.5 μg/L
2,4-Dichlorophenoxy acetic acid (2,4 - D)	8151	0.5 μg/L
3,5-DCBA	8151	0.5 μg/L
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	8151	0.5 μg/L
4 - Nitrophenol	8151	0.5 μg/L
Acifluorfen	8151	0.5 μg/L
Bentazon	8151	0.5 μg/L
Chloramben	8151	0.5 μg/L
Dacthal	8151	0.5 μg/L
Dalapon	8151	0.5 μg/L 0.5 μg/L
Dicamba	8151	
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	8151	0.5 μg/L 0.5 μg/L
Dinoseb	8151	
MCPA	8151	0.5 μg/L
MCPP	8151	5 μg/L
Pentachlorophenol	8151	5 μg/L 0.5 μg/L
Picloram	8151	
PAHs		0.5 μg/L
- Methylnaphthalene	8310	0.5
. – Methylnaphthalene	8310	0.5 μg/L 0.5 μg/L
cenaphthene	8310	
cenaphthylene	8310	0.5 μg/L
inthracene	8310	0.1 μg/L
enzo(a)anthracene	8310	0.5 μg/L
enzo(a)pyrene	8310	0.05 μg/L
enzo(b)fluoranthene	8310	0.05 μg/L
enzo(g,h,i)perylene	8310	0.1 μg/L
enzo(k)fluoranthene	8310	0.1 μg/L
hrysene	8310	0.05 μg/L
ibenz(a,h)anthracene	8310	0.05 μg/L
uoranthene	8310	0.1 μg/L
uorene	8310	0.1 μg/L
deno(1,2,3-cd)pyrene	8310	0.1 μg/L
aphthalene	8310	0.05 μg/L
	0310	0.5 μg/L

Table 3.1 (cont.).

	EPA	Lab Reporting
	Method	Limit for
	Metriod	groundwater
		groundwater
PAHs (cont.)		0.05 ~//
Phenanthrene	8310	0.05 μg/L
Pyrene	8310	0.05 μg/L
Metals		
Aluminum	200.7	0.05 mg/L
Antimony	204.2	0.006 mg/L
Arsenic	200.7	0.01 mg/L
Barium	200.7	0.1 mg/L
Beryllium	200.7	0.001 mg/L
Cadmium	200.7	0.001 mg/L
Calcium	200.7	0.5 mg/L
Chloride (total)	325.3	1 mg/L
Chromium (total)	200.7	0.01 mg/L
Cobalt	200.7	0.05 mg/L
	200.7 /	0.05 mg/L
Copper	220.1	
Iron	200.7	0.05 mg/L
Lead	200.7	0.005 mg/L
	200.7	0.5 mg/L
Magnesium	200.7	0.01 mg/L
Manganese	245.1	0.0002 mg/L
Mercury Nickel	200.7	0.01 mg/L
	258.1	0.5 mg/L
Potassium	200.7	0.01 mg/L
Selenium	200.7	0.01 mg/L
Silver	273.1	0.5 mg/L
Sodium	279.2	0.004 mg/L
Thallium	200.7	0.01 mg/L
Vanadium	200.7	0.1 mg/L
Zinc Ottor Parameters	200	
Other Parameters	*	0.4 mg/L
Dissolved oxygen	*	na
pH (converted to hydrogen ion for analysis)	*	na
Specific conductivity	*	na
Temperature	160.1	1 mg/L
Total dissolved solids		1 mg/L
Total organic carbon	415.1	I my c

Table 3.2. Frequency (percent) of values greater than detection limits for chemical parameters in groundwater.

Parameter	Frequency (%)
Sample Size	
Organochlorine Pesticides (8081)	57
4,4' - DDD	
4,4' - DDE	0
4,4' - DDT	0
Aldrin	0
Alpha - BHC	0
Beta - BHC	0
Chlordane (Total)	0
Chlordane alpha	0
Chlordane gamma	0
Delta - BHC	0
Dieldrin	0
Endosulfan i	0
Endosulfan ii (beta)	0
Endosulfan sulfate	0
Endrin	0
ndrin aldehyde	0
Indrin Ketone	0
Samma - BHC (Lindane)	0
leptachlor	0
leptachlor Epoxide (a)	0
leptachlor Epoxide (b)	0
sodrin	0
lethoxychlor	0
lirex	0
oxaphene	0
	0
Aroclors (8082)	
CB-1254 (noncarcinogenic)	0
CB-1016/1242	0
CB-1221	0
CB-1232	0
CB-1248	0
CB-1260	0
Chlorinated Hoshields (04.74)	
Chlorinated Herbicides (8151)	
(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 P) (Silvex)	0
1,5-Trichlorophenoxy acetic acid (2,4,5 -T)	0
-Dichlorophenoxy acetic acid (2,4 - D)	0
D-DCBA	0
2,4-Dichlorophenoxy)butyric acid (2,4 -DB)	0

Table 3.2. (cont.).

Parameter	Frequency (%)
Chlorinated Herbicides (cont.)	0
4 - Nitrophenol	0
Acifluorien	0
Bentazon	0
Chloramben	0
Dacthal	0
Dalapon	0
Dicamba	0
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic	0
acid)	
Dinoseb	0
MCPA	0
Pentachlorophenol	0
Picloram	0
PAHs (8310)	
1 - Methylnaphthalene	0
2 - Methylnaphthalene	0
Acenaphthene	0
Acenaphthylene	0
Anthracene	0
Benzo(a)anthracene	7.0
Benzo(a)pyrene	7.0
Benzo(b)fluoranthene	1.8
Benzo(g,h,i)perylene	0
Benzo(k)fluoranthene	5.3
Chrysene	5.3
Dibenz(a,h)anthracene	0
Fluoranthene	3.5
Fluorene	1.8
Indeno(1,2,3-cd)pyrene	3.5
Naphthalene	0
Phenanthrene	1.8
Pyrene	1.8
Metals	
Aluminum	66.7
Antimony	8.8
Arsenic (as carcinogen)	22.8
Barium	5.3
Beryllium	1.8
Cadmium	7.0
Calcium	98.2
	100.0
	5.3
Chloride Chromium (total)	

Table 3.2. (cont.).

Parameter	Frequency (%)
Metals (cont.)	
Cobalt	0
Copper	7.0
Iron	77.2
Lead	14.0
Magnesium	96.5
Manganese	70.2
Mercury (inorganic)	0
Nickel	3.5
Potassium	87.7
Selenium	7.0
Silver	1.8
Sodium	100.0
Thallium	
/anadium	12.3
Zinc	7.0
Other Parameters	1.8
otal Dissolved Solids	100.6
otal Organic Carbon	100.0
	93.0

Table 3.3. List of parameters that were below detection limits (non-detect) for all groundwater wells.

Organochlorine Pesticides (8081)	Chlorinated Herbicides (8151)
4.4' - DDD	2-(2,4,5-Trichlorophenoxy)propionic
14,1 222	acid (2,4,5 - TP) (Silvex)
4.4' - DDE	2,4,5-Tricholorophenoxy acetic acid
,,, ,,,	(2,4,5 –T)
4,4' - DDT	2,4-Dichlorophenoxy acetic acid (2,4-D)
Aldrin	3,5-DCBA
Alpha - BHC	4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)
Beta - BHC	4 - Nitrophenol
Chlordane (Total)	Acifluorfen
Chlordane, alpha	Bentazon
Chlordane, gamma	Chloramben
Delta- BHC	Dacthal
Dieldrin	Dalapon
Endosulfan I	Dicamba
Endosulfan II (beta)	Dichloroprop [2-(2,4-
	Dichlorophenoxy)proponic acid]
Endosulfan Sulfate	Dinoseb
Endrin	MCPA
Endrin Aldehyde	MCPP
Endrin Ketone	Pentachlorophenol
Gamma - BHC (Lindane)	Picloram
Heptachlor	PAHs (8310)
Heptachlor Epoxide(a)	1 - Methylnaphthalene
Heptachlor Epoxide(b)	2 - Methylnaphthalene
Isodrin	Acenaphthene
Methoxychlor	Acenaphthylene
Mirex	Anthracene
Toxaphene	Benzo(g,h,i)perylene
Aroclors (8082)	Dibenz(a,h)anthracene
PCB-1016/1242	Naphthalene
PCB-1221	Metals
PCB-1232	Cobalt
PCB-1248	Mercury
PCB-1254	
PCB-1260	

Table 3.4. List of parameters that were determined by Remediation team consensus to be treated as below detection limits (non-detect) for all groundwater locations based on low frequency of detection. These parameters had one value between the method detection level and the practical quantitation level.

	PAHs (8310)	_
Fluorene		
Phenant	nrene	
Pyrene		

Table 3.5. Frequency (percent) of values greater than detection limits for chemical parameters in groundwater by subaquifer and depth. Only parameters greater than detection limits in at least one sample are shown.

Parameter	All Ground- water	Dune Shallow	Dune Intermedi ate	Dune	Dune- Swale Shallow	Dune- Swale Intermedi ate	Dune- Swale Deep	West Shallow	West Intermedi ate	West Deep	Marsh Shallow	Marsh Intermedi ate	Marsh Deep
Sample Size	57	9	5	3	7	4	3	7	5	3	7	2	2
200													
PAHs (8310)											,		ľ
Benzo(a)anthracene	7.0	16.7	0	0	14.3	0	0	28.6	0	0	0	0	٥
Benzo(a)pvrene	7.0	16.7	0	0	14.3	0	0	28.6	0	0	0	0	0
Benzo(h)fluoranthene	1.8	0	0	0	0	0	0	14.3	0	0	0	0	
Benzo/k)fluoranthene	5.3	33.3	0	0	0	0	0	14.3	0	0	0	0	0
Chrysone	53	16.7	0	0	14.3	0	0	14.3	0	0	0	0	0
Flioranthone	3.5	c	0	0	14.3	0	0	14.3	0	0	0	0	0
Indeno(1,2,3-	3.5	16.7	0	0	0	0	0	14.3	0	0	0	0	0
cd)pyrene													
Metals								,		ļ	1000	000	000
Aluminum	66.7	2.99	100.0	33.3	57.1	75.0	33.3	71.4	40.0		85.7	100.0	200
Antimony	8.8	0	40.0	0	0	0	33.3	0	0	0	28.6	0	٥
Arsenic (as	22.8	20.0	0.09	2.99	0	0	0	14.3	20.0	0	42.9	o 	o
carcinogen)		ļ	ļ				c	c	c	c	28.6	20.0	0
Barium	5.3	5						٥	c		c	0	0
Beryllium	1.8	o i	20.0					0		٥	٥	C	c
Cadmium	7.0	16.7	20.0	33.3	14.3	0					2	200	2
Calcium	98.2	100.0	100.0	100.0	85.7	100.0	100.0	200.0	100.0	0.00	2.8	200.0	3 5
Chloride	100.0	100.0	100.0	100.0	100.0	100.0	100.0	100.0	100.0	100	100.0	0.00	100.0
Chromium (total)	5.3	0	20.0	0	0	0	0	0	0	0	28.6	0	0
Conner	7.0	0	20.0	0	0	0	0	0	0	33.3	14.3	20.0	0
lron	77.2	16.7	0.09	100.0	1000	100.0	2.99	85.7	100.0	100.0	71.4	80.0	20.0
HOIL	į				┨								

Table 3.5. (cont.).

Marsh Deep	0 100.0 100.0 0 0 0 0 0 0 0 0 0 0 0 0 0	100.0
Marsh Interme diate	╶┼┼┼┼┼┼┼┼ ┼┼	100.0
Marsh Shallow	14.3 100.0 71.4 0 0 100.0 14.3 28.6 0	100.0
West Deep	33.3 100.0 100.0 100.0 0 0 0 33.3 33.3	100.0
West Intermedi ate	0 100.0 100.0 0 0 0 0 0 0 40.0 0 0 0 0 0 0 0 0 0	100.0
West Shallow	0 100.0 100.0 0 14.3 0 0	100.0
Dune- Swale Deep	0 66.7 0 33.3 0 0 100.0 33.3 0	100.0
Dune- Swale Intermedi ate	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	100.0
Dune- Swale Shallow	0 71.4 14.3 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	100.0
Dune Deep	33.3 100.0 66.7 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	100.0
Dune Intermed iate	60.0 100.0 80.0 20.0 100.0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	100.0
Dune Shallow	0 100.0 33.3 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	100.0
All Ground- water	14.0 96.5 70.2 3.5 87.7 7.0 12.3 7.0 12.3 7.0	93.0
Parameter Metals (cont.)	Lead Magnesium Manganese Nickel Potassium Selenium Silver Sodium Thallium Vanadium Zinc Other Parameters	Total Dissolved Solids Total Organic Carbon

Table 3.6. Chemical parameters in groundwater by subaquifer and depth. Data are means with standard deviations in parentheses. Field parameters were not measured on replicate samples. ND indicates all samples below detection limits.

Darameter	Ali	Dine	Dune	Dune	Dune-	Dune-	Dune-	West	West	West	Marsh	Marsh	Marsh
	Ground-	Shallow	Intermed	Deep	Swale	Swale	Swale	Shallow	Intermedi	Deep	Shallow	Intermedi	Deep
	water		iate	•	Shallow	Intermedi ate	Deep		ate			ate	
Sample Size	25	9	5	3	7	4	3	7	5	3	2	5	2
PAHs (8310)													
Benzo(a)anthracene	0.035	0.047	Q	QN	0.036	QN	Q	0.051	Q	2	2	2	2
(mg/L)	(0.02)	(0.041)			(0.015)			(0.048)					
Benzo(a)pyrene	0.029	0.031	Q	Q	0.031	QN	QN	0.048	QN	2	Q	2	2
(ma/L)	(0.017)	(0.014)			(0.013)			(0.044)					
Benzo(b)fluoranthene	0.05	Q.	2	QN	Q.	QN	Q	0.067	QN	QN	Q	Q	2
(110V)	(0.02)							(0.045)					
Benzo(k)fluoranthene	0.028	0.037	Q	QN	Q	Q	QN	0.036	QN	ND	Q	2	9
(na/L)	(0.019)	(0.019)						(0.028)					
Chrysene	0.03	0.05	9	QΝ	0.031	QN	2	0.046	2	9	2	2	S
(uo/L)	(0.03)	(0.06)			(0.013)			(0.055)					
Fluoranthene	90.0	QN	Q	Q.	90.0	QN	Q	0.14	QN	Q	2	2	2
(uo/L)	(0.08)				(0.03)			(0.23)					
Indeno(1.2.3-	0.03	0.04	<u>Q</u>	2	2	QN	Q	0.034	2	2	2	2	2
cd)pyrene (µg/L)	(0.01)	(0.03)						(0.025)					

Table 3.6. (cont.).

Parameter	A	Dune	Dune	Dime	- Dring-	3							
	Ground-	Shallow	Intermed	Deep	Swale	Swale	Swale	West	West	West	Marsh	Marsh	Marsh
	water 		iate		Shallow	Intermed	Deep		ate	debo	Silallow	Intermed iate	Овер
Metais						Idle							
Aluminum	0.16	0.083	0 105	0.05	0000								
(mg/L)	(0.27)	(0.098)	(0.025)	0.03	0.298	0.117	0.049	0.143	0.057	0.033	0.44	0.15	0.066
Antimony	0.003	CN	0.00	L CIV		00:130	(0.041)	(0.175)	(0.054)	(0.014)	(0.50)	(0.08)	(0.020)
(mg/L)	(0.002)		(0.004)	2	2	 Z	0.0053	Q	2	ND	0.0038	QN	S
Arsenic (as	0.011	0.015	0.028	0.021	SN	C V	0.0043)				(0.0025)		
carcinogen) (mg/L)	(0.016)	(0.05)	(0.039)	(0.014)	2	2		0.000	0.008	2	0.025	Q.	S
Barium	90.0	QN	CN	S	2			(0.002)	(0.007)		(0.029)		
(mg/L)	(0.02)		?	2	2	2	2	2	Q	QN	0.11	90.0	9
Beryllium	0.0005	S	100		2						(0.13)	(0.03)	
(mg/L)	(0.0003)	?	0000	 2 2	2	2	2	2	QN	Q.	2	2	S
Cadmium	0 0007	8000	2000	2000	10000							:	-
(mg/L)	(0.0011)	(9000)	0.002	70000	0.000	2	<u>Q</u>	2	QN	2	2	Q.	S
Calcinm	242 4	140.0	1	2000	10.0002)
(mod)	(204.5)	(75.5)	322.4	336.7	56.1	9.76	254.0	144.3	192.0	246 7	2627	504.0	0 000
Chlorido	1501.61	(7.0.0)	(188.2)	(200.3)	(43.6)	(74.4)	(265.7)	(51.3)	(47.6)	(7.8.7)	(238.9)	0.4.00	0.020
9010000	4040	2995	12340	7433	27	102	3707	404	1000	1205	405.3	30.4)	17.07
(IIII)	(/2/2)	(4114)	(8322)	(7420)	(33)	(139)	(6316)	(689)	(049)	121	1624	14860	14800
Chromium (total)	900.0	2	900.0	CN	CN	2		1	(010)	(9101)	(3293)	(11870)	(15839)
(mg/L)	(0.003)		(0.002)	<u> </u>	<u> </u>	<u> </u>	2	<u> </u>	 Q	2	0.00	Q	2
Copper	0.031	Q.	00	CZ	CIA	2					(0.007)		
(mg/L)	(0.035)		(0.03)	2	}	2		 Q	2	0.110	0.022	0.028	S
Iron	1.12	0.058	0 21	30.6	90.0	60,				(0.147)	(0.00)	(900:0)	
(mg/L)	(1.76)	(0.08)	(66.0)	(3.24)	0.30	87.0	 	0.81	9.	5.00	1.60	2.31	121
				1	76:20	10:00	(00:1-	(0.94)	(0.20)	(0.97)	(3.71)	(2.38)	(1.68)

Table 3.6. (cont.).

												1000	March
Parameter	W	Dune	Dune	Dune	Dune-	Dane-	Dune-	West	West	West	Marsh	Marsn	Marsi
	Ground-	Shallow	Intermed	Deep	Swale	Swale	Swale	Shallow	Intermedi	Deep	Shallow	Interme	deen
	water		iate		Shallow	Intermedi	Deep		ate			diate	_
						ate							
Motels (cont.)													
l oad	0.004	QN	0.00	0.004	Q	QN	QN	Q	Q	0.011	0.003	900.0	 응
(mo/l)	(0.005)	!	(0.10)	(0.003)						(0.015)	(0.001)	(0.002)	
Magnesium	307.4	201.1	847.6	1036.7	2.2	10.0	244.9	35.6	73.0	98.7	248.6	796.8	782.5
	(493.8)	(267.6)	(571.1)	(845.6)	(5.6)	(13.5)	(420.1)	(31.7)	(19.4)	(28.7)	(211.6)	(734.0)	(1014.7)
Mongoogo	0.068	000	0.075	0.114	0.015	0.022	0.057	0.024	0.046	0.070	0.062	0.284	0.141
(maligaliese	0000	(0.03)	(0.072)	(0.162)	(0.026)	(0.05)	(0.08)	(0.095)	(0.019)	(0.002)	(0.02)	(0.146)	(0.112)
	9000	CIN	0.006	2000	S	QN	QN	QN	ΩN	0.015	Q	2	2
Nickei	0.000	2	0000	(0.003)) :	<u> </u>	!			(0.014)			
(mg/L)	00.01	0.89	274.2	316.7		1.2	31.5	8.1	17.0	13.3	74.8	215.6	239.4
Potassium	(150.6)	9.6	(177.6)	(211.3)	(9.6)	(1.7)	(54.1)	(8.1)	(13.2)	(0.0)	(63.7)	(241.7)	(326.2)
(mg/L)	130.0		200	CN	200	S	CZ	QN	QN	QN	QN	Q	2
Selenium	0.00	<u>}</u>	900	2	(0.003))	1						
(mg/L)	20.00			7000	CN	CZ	CN	CN	QN	QΝ	QN	Ş	2
Silver	0.000	€	}	0.00	2	}	2	2	!	!			
(mg/c)	0230	1510	6720	8167	13.1	53.6	1875	240	260	883	3121	7360	6650
	(4011)	2013	(4342)	(6526)	(11.9)	(59.5)	(3226)	(318)	(399)	(196)	(3030)	(5280)	(296)
T	1000	CN	CN	S	2	QN	0.001	QN	0.001	2	0.001	0.00	0.005
	0000	<u>}</u>) :	<u>:</u>	<u>!</u>	!	(0.0006)		(0.0005)		(0.0008)	(0.002)	(0.001)
(mg/L)	0.005	S	0000	2	2	QN	2	900.0	Q	QN	0.007	2	2
(mod)	(0.00)	<u>-</u>	(0.004)					(0.002)			(0.003)		
J. Silly C.	0.053	S	CN	S	Q	QN	QN	QN	Q	0.11	Q N	2	2
Zinc	0.033	<u></u>	<u>}</u>	?	<u>}</u>	}	!			(0.10)			
(mg/L)	10.04-1												

Table 3.6. (cont.).

rarameter	=	2	C										
	Ground- water	Shallow	Dune Intermedi ate	Dane Deep	Dune- Swale Shallow	Dune- Swale Intermedi	Dune- Swale Deep	West Shallow	West Intermedi	West Deep	Marsh Shallow	= E	Marsh Deep
						ate	•		}	***		late	
Other Parameters													
Total Dissolved	9908	5455	21564	22132	150	300							
Solids (mg/L)	(11275)	(6845)	(13441)	(19535)	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	808	6987	1164	2760	3900	8214	19020	21050
Total Organic	18.9	1 83	4 70	41 83	1,00	(20+	(1/2/0)	(1298)	(1228)	(1375)	(5227)	(13951)	(22557)
Carbon (mg/L)	(23.4)	(1.29)	(7.46)	(15 11)	19.14	6.5	12.3	31.4	9.5	7.3	51.3	26.4	15.5
					1,6:,	(4.1)	(2.5)	(30.0)	(8.9)	(3.5)	(35.5)	(18.9)	(6.4)
Sample Size (field)	2	ď											
PION STATE	5	٥	4	က	ဖ	4	۳.	ď		í			
Hydrogen Ion	8.80E-6	3.49E-8	4.41E-8	7.34F-8	7 40E-5	1 255 7	1 100	0 100	4	3	9	4	2
	(5.46E-5)	(5.46E-5) (2.46E-8) (4.27E-8)	(4.27E-8)	(8.35E-8)	/1 55E	7-305-7	1.00E-7	1.09E-7	6.16E-8	1.04E-7	1.04E-7 2.13E-7	1.55E-7	5.20F-R
				(0.300.0)	(1:30E-	(4.91E-0)	(7.9/E-8)	(4.29E-	(4.93E-8)	(1.38E-	(2.62E-	(4.00E-	(2.89E-
Ha	5.06	7.46	7 36	7 40				8)		6	~	6	1 6
Dissolved Oxvoen	1 82	2 70	1.00	51.7	4.13	6.90	6.97	96.9	7.21	6.98	6.67	680	7 20
(mg/L)	(1.44)	(1.24)	(1.5)	7.00 2.00 9.00 9.00	75.7	2.23	3.27	1.00	0.51	1.18	221	0.76	7 20
Temperature	25.7	26.8	26.9	26.0	0.83	(1.42)	(2.55)	(0.64)	(0.37)	(0.16)	(1.48)	(0.50)	0 40
(0)	(1.3)	(0.8)	20.5	0.0) (20.5	24.8	26.1	25.7	24.1	23.1	26.9	24.9	24.9
Specific	10012	6607	24875	22507	1:1	0.0	(1.4)	=======================================	(0.0)	(0.2)	(0.5)	(0.5)	8
Conductivity	(13156)	(7368)	(18001)	(18314)	(171)	8/5	7037	2242	3715	2220	11897	27210	25955
(nmhos/cm)				(F. 222.)		(020)	(10880)	(2119)	(1482)	(1440)	(7147)	(18546)	(23257)
								_	-		_	-	

Table 3.7. Summary of two-way analysis of variance of selected groundwater parameters. Shown are the significance levels for the overall model, effects due to subaquifer, effects due to depth, and subaquifer * depth interactions. NS = not significant.

Interactions (Subaquifer * Depth)	•	NS	SN	SN	SN	SN	SN	SN	SN		<u>S</u>	NIC	2	1000	0.037	SN	•
Depth	1	900.0	<0.001	0.029	0.003	0.009	0.032	0.003	SN		0.001	1000	0.00		NS	<0.001	•
Subaquifer	•	<0.001	<0.001	0.047	<0.001	0.002	<0.001	<0.001	<0.001		<0.001		<0.001		0.008	<0.001	1
Model	NS	0.02	<0.001	0.057	<0.001	9000	<0.001	<0.001	<0.001		<0.001		<0.001		0.001	<0.001	NS
Parameter	Aluminum	Calcium	Chloride	Lon	Magnesium	Mandanace	Potaccium	Codium	Total Organic	Carbon	Total Dissolved	Solids	Specific	Conductivity	Ho	Temperature	Dissolved Oxygen

Table 3.8. Frequency (percent) of values greater than detection limits for chemical parameters in groundwater with one or more samples above detection limits by groundwater class. Values are for all samples (N=57), samples with total dissolved solids <10000 mg/L (Class G-II, N=41), and samples with total dissolved solids >10000 mg/L (Class G-III, N=16).

Parameter	Groundwater (%)	Class G-II (%)	Class G-III (%)
Sample Size	57		
PAHs (8310)	5/	41	16
Benzo(a)anthracene			
Benzo(a)pyrene	7.0	7.3	6.3
Benzo(b)fluoranthene	7.0	7.3	6.3
Benzo(k)fluoranthene	1.8	2.4	0
Chrysene	5.3	2.4	12.5
Fluoranthene	5.3	4.9	6.3
Fluorene*	3.5	4.9	0
Indeno(1,2,3-cd)pyrene	1.8	2.4	0
Phenanthrene*	3.5	2.4	6.3
Pyrene*	1.8	2.4	0
	1.8	2.4	0
Metals Aluminum			
	66.7	61.0	81.2
Antimony	8.8	2.4	25.0
Arsenic (as carcinogen) Barium	22.8	17.1	37.5
	5.3	0	18.7
Beryllium	1.8	2.4	0
Cadmium	7.0	9.8	0
Calcium	98.2	97.6	100.0
Chloride	100.0	100.0	100.0
Chromium (total)	5.3	7.3	0
Copper	7.0	2.4	18.7
ron	77.2	85.4	56.2
ead	14.0	4.9	37.5
Magnesium	96.5	95.1	100
Manganese	70.2	68.3	75.0
lickel	3.5	4.9	0
Potassium	87.7	82.9	100.0
Selenium	7.0	7.3	6.3
ilver	1.8	0	6.3
odium	100.0	100.0	100.0
hallium	12.3	17.1	0
anadium	7.0	9.8	0
inc	1.8	2.4	0
Other Parameters			<u>_</u>
otal Dissolved Solids	100.0	100.0	100.0
otal Organic Carbon	93.0	92.7	93.7

^{*}These parameters had one value above the method detection level but below the practical quantitation level.

Table 3.9. Means and standard deviations (in parenthesis) for chemical parameters in groundwater by groundwater class. Values are for all samples (N=57), samples with total dissolved solids <10000 mg/L (Class G-II, N=41), and samples with total dissolved solids >10000 mg/L (Class G-III, N=16). ND = below detection.

Parameter	Groundwater	Class G-II	Class G-III
		44	16
Sample Size	57	41	10
PAHs (8310)		0.005	0.036
Benzo(a)anthracene	0.035	0.035	(0.025)
(μg/L)	(0.02)	(0.02)	
Benzo(a)pyrene	0.029	0.03	0.027
(μg/L)	(0.017)	(0.02)	(0.009)
Benzo(b)fluoranthene	0.05	0.05	ND
(μg/L)	(0.02)	(0.02)	
Benzo(k)fluoranthene	0.028	0.027	0.03
(μg/L)	(0.019)	(0.012)	(0.013)
Chrysene	0.03	0.03	0.03
(μg/L)	(0.03)	(0.02)	(0.04)
Fluoranthene	0.06	0.07	ND
(μg/L)	(0.08)	(0.10)	
Indeno(1,2,3-cd)pyrene	0.03	0.03	0.03
(μg/L)	(0.01)	(0.01)	(0.02)
(μg/L) Metals	· · · · · · · · · · · · · · · · · · ·		
Aluminum	0.16	0.18	0.11
(mg/L)	(0.27)	(0.32)	(0.07)
Antimony	0.003	0.003	0.006
(mg/L)	(0.002)	(0.0004)	(0.0004)
Arsenic (as carcinogen)	0.011	0.013	0.0097
(mg/L)	(0.016)	(0.02)	(0.008)
Barium	0.06	ND	0.08
(mg/L)	(0.05)		(0.09)
Beryllium	0.0005	0.0006	ND
(mg/L)	(0.0003)	(0.0004)	410
Cadmium	0.0007	0.0008	ND
(mg/L)	(0.0011)	(0.0013)	431.3
Calcium	242.4	168.8	(161.9)
(mg/L)	(201.2)	(164.3) 911	13856
Chloride	4545	(1331)	(8011)
(mg/L)	(7272)	0.006	ND
Chromium (total)	0.006 (0.003)	(0.003)	
(mg/L)	0.031	0.031	0.03
Copper	(0.035)	(0.04)	(0.02)
(mg/L)	1.12	0.84	2.00
iron	(1.76)	(0.84)	(2.96)
(mg/L)	0.004	0.004	0.005
Lead	(0.005)	(0.005)	(00.3)
(mg/L)	307.4	47.6	973.1
Magnesium (mg/L)	(493.8)	(53.8)	(495.5)

Table 3.9. (cont.).

Parameter	Groundwater	Class G-II	Class G-III
Metals (cont.)			0.000 0 111
Manganese	0.068	0.045	0.130
(mg/L)	(0.098)	(0.062)	(0.142)
Nickel	0.006	0.006	ND
(mg/L)	(0.004)	(0.004)	"
Potassium	89.2	12.5	285.9
(mg/L)	(150.6)	(20.9)	(161.6)
Selenium	0.006	0.006	0.008
(mg/L)	(0.007)	(0.004)	(0.011)
Silver	0.005	ND	0.005
(mg/L)	(0.007)	1,10	(0.001)
Sodium	2670	509	8206
(mg/L)	(4011)	(713)	(3638)
Thallium	0.001	0.001	ND
(mg/L)	(0.0005)	(0.0006)	טא
Vanadium	0.005	0.006	ND
(mg/L)	(0.002)	(0.002)	ND
Zinc	0.053	0.054	ND
(mg/L)	(0.024)	(0.028)	IND
Other Parameters		1 10.000	···
Total Dissolved Solids	8066	2019	23563
mg/L)	(11275)	(2202)	(10276)
Total Organic Carbon	18.9	19.5	17.5
mg/L)	(23.4)	(26.0)	(15.6)
		(33.0)	(10.0)
Sample Size (field)	51	36	15
lydrogen Ion	8.80E-6	1.24E-5	1.47E-7
	(5.46E-5)	(6.49E-5)	(1.74E-7)
Dissolved Oxygen	1.8	1.7	2.2
mg/L)	(1.4)	(1.2)	(1.9)
emperature	25.7	25.5	26.3
C)	(1.3)	(1.4)	(1.0)
specific Conductivity	10012	2866	27162
umhos/cm)	(13156)	(2935)	(12253)

Table 3.10. Groundwater chemistry data for the Surficial Aquifer on KSC from Clark (1987c). Upper includes wells from 3-20 ft (0.9-6.1 m), Intermediate- wells between 21-30 ft (6.4-9.1 m) and Lower- wells between 31-45 ft (9.4-13.7 m).

	Coastal Subaquifer	baquifer		Prime Recharge Area	ırge Area		Dune-Swale Subaquifer	Subaquifer	
	Upper	Inter- mediate	Lower	Upper	Inter- mediate	Lower	Upper	Inter- mediate	Lower
Sample Size	40	5	10	3	9	11	25	12	21
Total Dissolved	4426	1820	70868	167	868	559	550	1361	6046
Hd	7.50	7.50	7.40	6.70	7.57	7.49	7.26	7.62	7.60
Sodium (ma/L)	1059	183	10971	14	128	39.6	79	265	1936
Chloride (mg/L)	1854	307	24417	28	232	80	102.8	317	3734
Arsenic (ma/L)	<0.08	<0.05*	<0.05*	<0.05*	<0.05*	<0.05*	<0.64	<0.05*	<0.05
Barium (mo/L)	<0.03*	<0.28*	<0.10*	<0.70*	<0.07*	<0.70*	<1.0*	<1.0*	<0.7*
Cadmium (mo/L)	<0.013	<0.01	0.03	<0.012*	<0.012*	<0.17*	<0.01*	<0.01*	<0.017*
Chromium (ma/L)	<0.037*	<0.03*	<0.045	<0.04*	<0.04*	<0.02*	<0.05	<0.04*	<0.02*
Copper	>0.06	<0.03*	0.07	<0.05*	<0.02*	<0.03*	<0.10*	<0.04*	<0.03*
Iron (ma/L)	4.38	1.84	0.86	<0.62	6.10	1.58	3.35	4.50	1.50
Lead (mo/L)	<0.06	<0.06	0.17	<0.05*	<0.05*	<0.05*	<0.05*	<0.05*	<0.05*
Manganese (mg/L)	<0.09	0.15	0.09	<0.03*	<0.042	<0.037*	<0.03	<0.04	<0.04
Selenium (mg/L)	0.08	<0.05*	<0.006*	<0.02	<0.03	0.02	<0.05	<0.044	<0.04
Zinc (mg/L)	<0.07	<0.06	0.22	<0.14	<0.032*	<0.03*	<0.04	<0.03*	0.036

Table 3.10. (cont.)

	Lower		5	3812		7.43	1145	2136	<0.05*	<0.045	<0.014*	<0.03*	<0.04	0.65	<0.05	90.0	<0.01	<0.05	
aguifer	Inter-	- Indulate	٥	8208		7.50	3095	3628	<0.05*	<1.0*	<0.01	<0.04*	<0.047*	1.90	<0.06	0.11	<0.04	<0.04*	
Marsh Subaquifer	Upper	o	2000	1066		7.50	1469	2563	<0.07	<1.0*	<0.014*	<0.05*	<0.02*	0.30	<0.056	<0.05*	<0.12	<0.046	
)r	Lower	+	2544	1		7.55	407	1020	<0.05*	<0.7*	<0.014*	<0.04*	<0.05*	5.20	<0.05*	0.08	0.07	<0.03*	
West Plain Subaquifer	Inter- mediate	13	1071	-	7.00	Ç <u>0</u> ./	104.6	307	90.0	<1.0*	<0.01	<0.05*	<0.02*	3.20	<0.05*	<0.05*	<0.01	<0.05*	ı limits
West Pla	Opper	10	403	}	7 10	7.10	31.1	41.5	<0.056	<0.01*	<0.01*	<0.05	<0.10*	5.88	<0.05*	<0.05*	0.04	<0.05	v detection
		Sample Size	Total	Dissolved Solide (mod.)	HO HO	<u>.</u>	Sodium (mg/L)	Chloride (mg/L)	Arsenic (mg/L)	Barium (mg/L)	Cadmium (mg/L)	Chromium (mg/L)	Copper (mg/L)	Iron (mg/L)	Lead (mg/L)	Manganese (mg/L)	Selenium (mg/L)	Zinc (mg/L)	*all values below detection limits

SECTION 4 Surface water

4.1 Methods

4.1.1 Sampling Site Selection

Location of the surface water sampling stations was determined based on the watershed basins surrounding Kennedy Space Center. A total of 40 stations were selected to incorporate samples from open lagoonal water, rivers, creeks, ditches, burrow pits, and impoundments (Figure 4.1). Sampling sites were chosen away from known SWMU sites. Because 1998 was a year of the "El Nino" weather phenomenon, the climate in Florida was very hot and dry. The resulting drought reduced the number of ditches available for sampling. A total of 18 ditches were sampled, with 5 located near KSC's industrial areas. Location and description of each sampling station are given in Appendix E, Table E-1.

The geographical location of all sampling stations was recorded using a Trimble Pathfinder Professional XL Geographic Positioning System (GPS) unit using real time differentially corrected methods (Trimble Navigation Limited 1994). Data were exported to North American Datum 1927 (NAD27) State Plane coordinates and overlayed in a Geographic Information System (GIS). Coordinate data of each sampling station are given in Appendix E, Table E-2.

4.1.2 Sampling

Surface water sampling and analysis were subcontracted by Dynamac to Environmental Conservation Laboratories (ENCO), in Orlando, Florida. ENCO technicians followed the requirements of the "NASA/KSC Sampling and Analysis Plan for RCRA Facility Investigations at Kennedy Space Center, Florida and Cape Canaveral Air Station, Florida, Volume 4" of the Generic Work Plans (SAP) (NASA 1996) and ENCO's FLDEP Comprehensive Quality Assurance Plan No. 960038.

Sample collection was done from shore, by wading, or in deeper water, by boat. Care was taken not to disturb sediments in the immediate area, and collection was done upstream of the sampler. A total of 46 samples (includes 6 duplicates) were collected with decontaminated equipment, using latex gloves, and in the appropriate order as not to cause cross-contamination.

Surface water was collected directly into sample bottles that contained the proper preservative. Sample bottles were kept on ice and transported to the lab for analysis. Field parameters such as pH, temperature, turbidity, dissolved oxygen, and conductivity were also measured at each sampling location.

4.1.3 Chemical Analysis

Surface water was analyzed for 94 parameters that included organochlorine pesticides (EPA method 8081), aroclors (EPA method 8082), chlorinated herbicides (EPA method 8151), and polyaromatic hydrocarbons (EPA method 8310), and metals. The EPA methods used for metal analyses varied according to the appropriate analyte and included EPA methods: 200.7, 204.2, 258.1, 279.2, 7211, 7470, 7761, and 7770. Other miscellaneous parameters such as total chloride was analyzed using EPA method number 325.3, total dissolved solids was determined by method 160.1, and total organic carbon was determined by method 415.1 (Table 4.1). Field analysis of surface water samples for specific conductivity, dissolved oxygen, pH, turbidity, and temperature were conducted with calibrated instruments.

4.1.4 Data Analysis

Analyses were conducted using SPSS Version 9 (SPSS Inc., 1999), and SAS Version 6.02 (SAS Institute, 1998).

The frequency of analytes occurring above the detection limits was determined (Table 4.2). The NASA/KSC Remediation Program Team agreed that only those parameters with more than one detection would be used for analysis. Three analytes (naphthalene, fluorene, and copper) had only one detection and were, therefore, treated as non-detect (Appendix E, Table E-3). These results, however, are given in Appendix E, Table E-10. Those with more than one detection were analyzed, but their non-detect values were substituted by one-half the reported level.

Basic statistic analyses, such as normal probability plots, Shapiro-Wilks, and the Lilliefors tests, were conducted using SPSS to check for normality of distribution. The presence of significant outliers was also tested in SPSS by using histograms, stem-and-leaf plots, and boxplots. Extreme outliers were removed, however, in situations where a particular parameter had very few detection hits, these hits were erroneously flagged as outliers and were therefore, not removed. These will be discussed in detail on an individual basis.

For those parameters with sufficient sample size (n>30), multiple statistical comparisons were done using SAS. Non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α = 0.05) between the different basin groups. This analysis was able to subdivide the basins and ditches into groups that were statistically similar. These data were then combined and an overall mean reported. For the screening background values table, data are given as twice the mean (Appendix A, Table A-3).

4.2 Results and Discussion

Of the 94 surface water parameters sampled, only 27 were detected. One organochlorine pesticide (Dieldrin) out of the 25 sampled, and none of the aroclors (6), or chlorinated herbicides (18) were detected. Polyaromatic hydrocarbons (17) had 5 parameters that were detected in at least one station, and the 24 metals tested were detected in high frequency except for eight (Ba, Cd, Cr, Co, Hg, Ni, Vn, and Zn), these were always below detection (Table 4.2). Screening background tables are given in Appendix A, Table A-3.

Organochlorine Pesticides

Dieldrin is a chlorinated cyclodiene that was used as a pesticide in crops such as corn and cotton and as an insecticide to control root worms, beetles, and termites in the 1950s-1970. The Environmental Protection Agency banned all uses of Dieldrin in 1974 with the exception of termite control, although it is no longer registered for general use (Parrish et al., 1979).

Dieldrin is extremely persistent, but it slowly photorearranges to photodieldrin (water half-life is 4 months), and photodegrades. When released in soil, it persists for longer periods (> 7 yrs.), and reaches the air either through slow evaporation or adsorption on dust particles (Parrish et al., 1979). Dieldrin does not leach, and enters surface water from runoff. Once dieldrin reaches surface waters it adsorbs strongly to sediments, bioconcentrates in fish, other aquatic organisms, wildlife, foods, and humans (Chadwick and Brockson, 1969; Epifanio, 1973; Lane and Livingston, 1970; Murphy and Korschgen, 1970; Reinert, 1972; Tarzwell and Henderson, 1956). Human exposure to dieldrin appears to come mostly from food (Parrish et al., 1979; Walker et al., 1973).

The two surface water quality sampling stations with dieldrin were IRL001, located in the northern Indian River Lagoon, and SR3-D2, a saline-water ditch located on Route 3 just south of Haulover Canal (Figure 4.2). Measured concentrations were 0.39 μ g/L at IRL001 and 0.07 μ g/L at SR3-D2. Station IRL001 is located in the north of Haulover Canal and on the east coast of the Indian River Lagoon, while station SR3-D2 is on the east side of Route 3 south of Haulover Canal. This region has numerous citrus groves, and these levels of dieldrin might be related to past usage.

When analyzing these data, statistical analyses were restricted to those stations that were part of the same watershed basins as IRL001 and SR3-D2. Station IRL001 is part of the Indian River Lagoon basin, and SR3-D2 is a saline (sal ≥6 ppt) ditch. These two values, along with one-half the reported non-detection level of the other stations were used to compute the mean, minimum, maximum, and standard deviation of the corresponding watershed basins (Appendix E, Tables E-8 and E-9). Screening Table "Combined Water (sal >6 ppt)", values were calculated by averaging these two data points with those stations with the appropriate salinity (Appendix A, Table A-3).

Polycyclic Aromatic Hydrocarbons

Benzo(a)anthracene, benzo(k)fluoranthene, benzo(a)pyrene, chrysene, and indeno(1,2,3-cd)pyrene are all polycyclic aromatic hydrocarbons (PAHs). The name PAH generally refers to hydrocarbons containing two or more fused benzene rings that form as a result of incomplete combustion of organic compounds, by diagenesis, or by biosynthesis (Varanasi, 1989).

PAHs occur as needles, plates, crystals, leaflets, or prisms ranging from colorless to pale yellow to golden yellow. Benzo(a)anthracene and indeno(1,2,3-cd)pyrene show fluorescence ranging from greenish yellow to brilliant bluish violet to brown (Varanasi, 1989). PAHs are found in gasoline and diesel motor vehicle exhaust, by-products of open fires or refuse burning, coal tar, coal tar pitch, coke tars or coke oven emissions, creosote, mineral oils, bitumens, industrial smoke and soot, cigarette and cigar tobacco, tar, or smoke condensates, and charcoal-broiled foods. (ATSDR, 1989; Cerniglia et al., 1980; Farrington et al., 1983; Healy and Young, 1979 NAS, 1983; Varanasi, 1989).

Levels of individual PAH have been measured in aquatic organisms worldwide. Reported values range from undetectable quantities (approximately 0.01 μ g/kg dry weight) to values in excess of 5000 μ g/kg dry weight for individual PAH in tissues of aquatic organisms (Pancirov and Brown, 1977; Varanasi, 1989).

Generally elevated concentrations of PAH can be correlated with the proximity of the organism to areas receiving chronic hydrocarbon discharge (Cerniglia et al., 1980; Clark et al., 1978; Farrington et al., 1983; Healy and Young, 1979; Varanasi, 1989). The majority of PAH measurements have been made on bivalve mollusks. Bivalves are sessile, can rapidly accumulate PAHs, and have little capacity for PAH metabolism (Clark et al., 1978; Cross et al., 1978; Pancirov and Brown, 1977). PAH concentrations in fish have been less well surveyed, but are usually low relative to invertebrates inhabiting the same environment. The low body burdens of PAH to fish are believed to be due to their ability to rapidly metabolize PAH. Although PAHs are metabolized and eliminated in fish, it is not known how they interact with birds, mammals, or humans. Dermal contact with high concentrations of PAHs may cause irritation, burning, inflammatory spots on the skin, eruption of blood vessels and sensitivity to sunlight. However, there are a number of epidemiologic and mortality studies that show increased incidences of cancer in humans exposed to mixtures of PAHs (ATSDR, 1989; Varanasi, 1989).

Benzo(a)anthracene. Benzo(a)anthracene was detected at two surface water sampling stations, both located in the southern Mosquito Lagoon. The first station was MLS03 (duplicate sample) with 0.12 μ g/L, and the second was SJML02 with 0.13 μ g/L. The former station is located in the southern portion of Mosquito Lagoon, east of Gallinipper Point, and the latter southeast of Haulover Canal, north of Cucumber Island. Both stations are located approximately

halfway across the lagoon. MLS03 is in the Middle Banks, a relatively shallow (0.60m) region with abundant seagrass beds. SJML02 is in approximately 2 meters of water with sparse seagrass. Both stations are located away from the intercoastal navigation channel in regions used mainly by recreational fishermen (Figure 4.2).

Data analyses for benzo(a)anthracene were restricted to those stations that were part of the Mosquito Lagoon basin group. These two values, along with one-half the reported non-detection level of the other stations were used to compute the mean, minimum, maximum, and standard deviation of the corresponding watershed basin (Appendix E, Table E-9). Screening Table "Combined Water (sal >6 ppt)" values, were calculated by averaging these two data points with those stations with the appropriate salinity (Appendix A, Table A-3).

Benzo(k)fluoranthene. Benzo(k)fluoranthene was detected at two surface water sampling stations, located in the Banana River and in a saline ditch (sal \geq 6 ppt). The first station was PEF20 with 0.13 μ g/L, and the second was MAX01 with 0.21 μ g/L. Station PEF20 is located in the Pepper Flats region of the northern Banana River, east of the Turn Basin and near a security boat dock. This region is relatively shallow and has abundant seagrass. MAX01 is an impoundment located on the south side of Beach Road approximately 0.5 Km west of the ocean. This impoundment has a soft muddy bottom with very sparse seagrass (Figure 4.2).

Data analyses for benzo(k)fluoranthene were restricted to those stations that were part of the Mosquito Lagoon and Banana River basins. These two values, along with one-half the reported non-detection level of the other stations were used to compute the mean, minimum, maximum, and standard deviation for each corresponding basin (Appendix E, Tables E-5 and E-9). Screening Table "Combined Water (sal \geq 6 ppt)" values, were calculated by averaging these two data points with those stations with the appropriate salinity (Appendix A, Table A-3).

Benzo(a)pyrene. Benzo(a)pyrene was detected at three surface water sampling stations, located in the Banana River and in Mosquito Lagoon. The first station was PEF20 with 0.18 μ g/L, the second was MLS03 (duplicate sample) with 0.13 μ g/L, and the third was SJML02 with 0.22 μ g/L. Station PEF20 is located in the Pepper Flats region of the northern Banana River, east of the Turn Basin and near a security boat dock. This region is relatively shallow and has abundant seagrass. MLS03 is in the southern portion of Mosquito Lagoon, east of Gallinipper Point, and SJML02 is southeast of Haulover Canal, north of Cucumber Island. MLS03 is in the Middle Banks, a relatively shallow (0.60m) region with abundant seagrass beds, and SJML02 is in approximately 2 meters of water with sparse seagrass (Figure 4.2).

Data analyses for benzo(a)pyrene were restricted to those stations that were part of the Mosquito Lagoon and Banana River basins. These three values, along with one-half the reported non-detection level of the other stations were used to compute the mean, minimum, maximum, and standard deviation for each corresponding basin (Appendix E, Tables E-5 and E-9). Screening Table "Combined Water (sal ≥6 ppt)" values, were calculated by averaging these three data points with those stations with the appropriate (Appendix A, Table A-3).

<u>Chrysene</u>. Chrysene was detected at two surface water sampling stations located in the southern Mosquito Lagoon and in a saline ditch (sal \geq 6 ppt). The first station was SJML02 with 0.21 μ g/L, and the second was MAX01 with 0.07 μ g/L. Station SJML02 is southeast of Haulover Canal, north of Cucumber Island, and MAX01 is an impoundment located on the south side of Beach Road approximately 0.5 Km west of the ocean. SJML02 is in approximately 2 meters of water with sparse seagrass, and MAX01 is an impoundment with a soft muddy bottom and very sparse seagrass (Figure 4.2).

Data analyses for chrysene were restricted to those stations that were part of the Mosquito Lagoon basin. These two values, along with one-half the reported non-detection level of the other stations were used to compute the mean, minimum, maximum, and standard deviation of the corresponding watershed basin (Appendix E, Table E-9). Screening Table "Combined Water (sal >6 ppt)" values, were calculated by averaging these two data points with those stations with the appropriate salinity (Appendix A, Table A-3).

Indeno(1,2,3-cd)pyrene. Indeno(1,2,3-cd)pyrene was detected at two surface water sampling stations located in southern Mosquito Lagoon. The first station was SJML02 with 0.13 μ g/L, and the second was MLS03 (duplicate sample) with 0.12 μ g/L. MLS03 is in the southern portion of Mosquito Lagoon, east of Gallinipper Point, and SJML02 is southeast of Haulover Canal, north of Cucumber Island. MLS03 is in the Middle Banks, a relatively shallow (0.60m) region with abundant seagrass beds, and SJML02 is in approximately 2 meters of water with sparse seagrass (Figure 4.2).

Data analyses for indeno(1,2,3-cd)pyrene were restricted to those stations that were part of the Mosquito Lagoon basin. These two values, along with one-half the reported non-detection level of the other stations were used to compute the mean, minimum, maximum, and standard deviation of this watershed basin (Appendix E, Table E-9). Screening Table "Combined Water (sal >6 ppt)" values, were calculated by averaging these two data points with those with the appropriate salinity (Appendix A, Table A-3).

Fluorene and Naphthalene. Fluorene and naphthalene were each detected in only one surface water sampling station. Fluorene was detected at BPD002 with 0.23 μ g/L and naphthalene was detected at SR3-D2 with 1.6 μ g/L. Station BPD002 is a fresh-water impoundment located in a wildlife observation area

known as Black Point Drive, and SR3-D2 is a saline-water ditch located on Route 3 just south of Haulover Canal (Figure 4.2). Because these stations only had one detection (out of 46 samples), the NASA/KSC Remediation Program Team agreed to treat these parameters as non-detect.

Metals

Aluminum. Aluminum is the third most abundant element in the earth's crust, exceeded by oxygen (47%) and silicon (28%). Because of its strong affinity to oxygen, aluminum never occurs as a metal in nature, and is found in the form of alumina. Aluminum is commonly distributed throughout rocks and soils such as silicates and clays. It is ductile, nonmagnetic, and an excellent conductor of heat and electricity (Brown et al., 1962; Thurston et al., 1979).

Aluminum is widely used in many kinds of products because it's properties give it special advantages over other materials. Aluminum is a major factor in the human diet, and is found in buffered aspirin, antacids, antidiarrheal drugs, cake mixes, self-rising flour, processed cheese, baking powder, food starch modifiers, pickling salts, and anti-caking agents. It can also be acquired from beverage cans, aluminum foil, and aluminum pots and pans. Aluminosilicates are found in the dust from talcum powder, baby powder, cat-box litter, cement, asphalt mixes, tobacco smoke and ash (Brown et al., 1962; Faust and Aly, 1981; Moore and Ramamoorthy, 1947).

Aluminum was detected at 36 surface water sampling stations that included representatives from all six watershed basin groups. This sample size allowed more intricate statistical analysis. One outlier was detected at station BCE004 (1.9 mg/L), but it was not removed because it wasn't extreme.

Data values, along with one-half the detection level of the other stations, were used to compute the mean, minimum, maximum, and standard deviation of each watershed basin (Appendix E, Tables E-4 through E-9). Multiple statistical comparisons were done using a non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for Post Hoc multiple comparisons (∞ = 0.05). This analysis compared the mean of each watershed and ditch and was able to subdivide and combine them into groups that were statistically similar.

Results showed that aluminum data from the Banana Creek basin were significantly different than all others (p< 0.05). Aluminum concentrations in the Banana Creek stations were higher than at other basins (1.1 mg/L vs. 0.3 μ g/L). Therefore, in order to calculate the screening table values for aluminum, data were separated into two groups. The first group included the average of only the Banana Creek basin stations, while the second included the combined average of all other remaining surface water stations. Screening table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

Antimony. Antimony is a brittle, hard metal that is obtained as a byproduct or coproduct of base metals and ores. It is found primarily in stibnite, silver and lead ores, but it is also associated with mercury and gold-bearing ores (Thurston et al., 1979).

It is present in mining and smelting wastewaters, in the manufacturing of bearings, storage batteries, pewter, lead electrodes, tanks, pipes, pumps, glass, and paint industries. It is used in semiconductors and thermoelectric devices, in dyes, glass, metalware and enamels (Thurston et al., 1979). It gets into water from corrosion of lead pipes and fittings, but even then it is rarely detectable. More antimony is found in food than water (Faust and Aly, 1981). Antimony also has a therapeutic use in the treatment of tropical diseases such as schistosomiasis, bilharziasis, and leishmaniasis. Antimony and many of its compounds are toxic (Brown et al., 1962; Faust and Aly, 1981).

Antimony was detected in 19 surface water sampling stations, that included representatives from all six watershed basin groups. This sample size did not allow for intricate statistical analysis. Data were separated into their corresponding watershed and ditch groups, and along with one-half the reported non-detection level of the other stations were used to compute the mean, minimum, maximum, and standard deviation (Appendix E, Tables E4 through E-9). Screening Table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

Arsenic. Arsenic is a shiny, gray, brittle element and one of the most widely distributed in the earth's crust and in the biosphere. It possesses both metallic and non-metallic properties. Its compounds may be organic or inorganic, are present everywhere in nature, are insoluble in water, and occur mostly as arsenides and arsenopyrites (Thurston et al., 1979).

Arsenic is very poisonous and can kill in relatively small amounts. It is absorbed in the gastrointestinal tract, lungs, or skin. The systemic effects of arsenic poisoning include pain, nausea, diarrhea, abnormal heart function, impaired nerve function, blood-vessel damage, liver or kidney damage, and a pattern of skin abnormalities (Faust and Aly, 1981; Thurston et al., 1979). It distributes itself throughout the body and is excreted via urine, feces, sweat, or skin epithelium (Faust and Aly, 1981).

Arsenic, like mercury, undergoes transformation in the environment through the metabolic activities of microorganisms, especially bacteria and fungi. This indicates that there is a biological cycle for arsenic in the aquatic environment. (Moore and Ramamoorthy, 1947).

Arsenic was detected in 9 surface water sampling stations and did not include any samples from Mosquito Lagoon or fresh water ditches. This sample size did not allow for intricate statistical analysis. Data were separated into their

corresponding watershed and ditch groups, and along with one-half the reported non-detection level of the other stations were used to calculate the mean, minimum, maximum, and standard deviation (Appendix E, Tables E-4, E-5, E-7, and E-8). Screening Table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

Beryllium. Beryllium is present in a variety of compounds such as chlorides, nitrate salts, sulfates, carbonates, and hydroxides. The solubility of these compounds varies depending on such factors as hardness and pH (Thurston et al., 1979).

Beryllium and its alloys are extensively used in the manufacturing of springs, gyroscopes, computer parts, electrical contacts, spot-welding electrodes, and nonsparking tools. As a structural material, beryllium is used for communication satellites, missiles, high-speed aircraft, and spacecraft (Thurston et al., 1979).

Beryllium enters the water supply mainly from effluents of certain metallurgical industries such as atomic reactors, aircraft, rockets and missile fuels. The major toxic potential of beryllium to humans is by inhaling beryllium-containing fumes. Beryllium metal dust can cause major lung damage and beryllium salts are very toxic. Once exposed to the skin, beryllium causes dermatitis, conjunctivitis, acute pneumonitis, chronic pulmonary berylliosis, and burns (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Beryllium was detected in 5 surface water sampling stations that were all located in the Banana River basin. This sample size did not allow for intricate statistical analysis, and these data, along with one-half the reported non-detection level of the other stations were used to calculate the mean, minimum, maximum, and standard deviation (Appendix E, Table E-5). Screening Table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

<u>Calcium</u>. Calcium is the fifth most common element in order of abundance in the earth's crust. It is present in water supplies as a result of its passage through or over deposits of limestone, marble, chalk, calcite, alabaster, selenite, dolomite, gypsum, and gypsiferous shale. It can also be found in natural brines, salt beds, and as a byproduct of chemical industries (Thurston et al., 1979).

Calcium is the most abundant mineral (90%) in the body, and is found mostly in bones and teeth. A diet rich in calcium is very important when bones are growing and developing. Even after full bone development, an adequate calcium intake is necessary to keep bones strong and healthy. When calcium is low, the blood tends to become overly acidic. This promotes disease conditions such as: cancer, arthritis, heart disease, Alzheimer's, Parkinson's, colitis, asthma, muscle

pains, bursitis, sinusitis, diabetes, migraine and even depression (Brown et al., 1962; Thurston et al., 1979).

Calcium was detected in all 46 surface water sampling stations, which allowed more intricate statistical analysis. Data were separated into their corresponding watershed basins and ditch groups, and their mean, minimum, maximum, and standard deviation computed (Appendix E, Tables E-4 through E-9). Multiple statistical comparisons were done using a non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for Post Hoc multiple comparisons ($\alpha = 0.05$). This analysis was able to subdivide and combine the different watershed basins and ditches into groups that were statistically similar.

Calcium data showed that there were significant differences (p< 0.05), and three distinct groups existed. The first, with the highest calcium average concentration (345 mg/L) included those stations from Banana Creek, Indian River Lagoon, and Mosquito Lagoon basins. The second, with a combined average of 242 mg/L, included those stations from the Banana River and saline (sal \geq 6 ppt) ditches. The third, had the lowest calcium average (98 mg/L), and included the stations from fresh water ditches. These three groups were, therefore, used to compute the Screening Table values for calcium. "Combined Background Water" values were calculated by averaging only those stations with the appropriate salinity (Appendix A, Table A-3).

<u>Chloride</u>. Chloride (Cl⁻), is found in nature in the combined state, chiefly with sodium as common salt (NaCl), carnallite, and sylvite (Thurston et al., 1979). Chlorides in solid form are salts containing chloride and a metal. In water, these salts dissolve forming chloride ions and metal ions. Some common chlorides are sodium chloride (table salt), calcium chloride and magnesium chloride (Brown et al., 1962; Thurston et al., 1979).

Chlorides are found in nearly all rivers, lakes and streams. They may get into the water from chloride-containing rocks, seawater, farm fertilizer runoff, halite, sea spray, brines, hot springs, papermaking and galvanizing processes, softening operations, oil wells and refineries (Brown et al., 1962; Thurston et al., 1979).

Chloride was detected in 45 surface water sampling stations, which allowed more intricate statistical analysis. The only station that did not have detectable concentrations of chloride was KPI001, located in an impoundment at Kars Park II recreational area (Figure 4.2).

Data were separated into their corresponding watershed basins and ditch groups, and their mean, minimum, maximum, and standard deviation computed (Appendix E, Tables E-4 through E-9). Multiple statistical comparisons were done using a non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for Post Hoc multiple comparisons (∞ = 0.05). This analysis was able

to subdivide and combine the different watershed basins and ditches into groups that were statistically similar.

Chloride data showed that there were significant differences (p< 0.05), and three distinct groups existed. The first, with the highest chloride average concentration (14295 mg/L) included those stations from Banana Creek, Indian River Lagoon, and Mosquito Lagoon basins. The second, with a combined average of 8715 mg/L, included those stations from the Banana River and saline (sal ≥6 ppt) ditches. The third, had the lowest chloride average (1527 mg/L), and included the stations from fresh water ditches. These three groups were, therefore, used to compute the Screening Table values for chloride. "Combined Background Water" values, were calculated by averaging only those stations with the appropriate salinity (Appendix A, Table A-3).

<u>Copper</u>. Copper is a trace element that is usually present in the divalent state as a natural or native metal. It is reddish, malleable, ductile, and a good conductor of heat and electricity. Primary copper ores are sulfides, oxides, and carbonates. Copper is often alloyed with other metals to form brasses and bronzes. Copper is used extensively in electrical wire, coinage metal, as an agricultural poison, and water purifier (Brown et al., 1962; Thurston et al., 1979).

Copper is an essential trace mineral that is present in all of the body tissues. Good sources of copper are oysters and other shellfish, whole grains, beans, nuts, potatoes, dark leafy greens, dried fruits such as prunes, cocoa, black pepper, and yeast (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Copper is essential for the synthesis of chlorophyll and the propagation of leaves in plants. In animals, it is necessary for metabolism, blood chemistry, and keeping the blood vessels, nerves, immune system, and bones healthy. The toxicity of copper to aquatic life depends strongly on pH. At lower alkalinity, copper toxicity is greater to aquatic life depending on species, size, and type of exposure. Waters with high copper concentration are tolerated by many species, depending on the species and amount of exposure (Brown et al., 1962; Tchobanoglous and Schroeder, 1985, Thurston et al., 1979).

Copper was detected in only one surface water sampling station (0.007 mg/L at ROB001). This station was a fresh-water ditch located on Robert's Road, west of Route 3 (Figure 4.2). Because this station only had one detection hit (out of 46 samples), the NASA/KSC Remediation Program Team agreed to treat it as non-detect (Appendix E, Table E-3).

Iron. The fourth most abundant metal in the earth's crust is common in many types of rocks and soils, particularly clays. Major sources of Fe pollution are industrial waste waters, mines, and iron-bearing groundwaters. Iron is present

as insoluble ferric salts and pyrite, in natural organometallic or humic compounds, and in colloidal forms (Thurston et al., 1979).

Ferrous Fe⁺² and ferric Fe⁺³ ions are the primary forms of concern in the aquatic environment. When found in high concentrations in the water, Fe produces a distinguishing color that serves as a good pollution indicator. Iron also forms hydroxide or oxide precipitates which, if they flocculate or gelatinize, can be detrimental to benthic species and fish eggs. Iron is a trace element that is a very important component of vertebrate erythrocytes and of the blood of some invertebrates. It is also a limiting factor in the growth of anaerobic bacteria, algae, and other plants (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Iron was detected in 35 surface water sampling stations, that included representatives from all six watershed basin groups. This sample size allowed more intricate statistical analysis. Two outliers were detected at stations TEL004 (2.3 mg/L), and at HPC001-duplicate (1.2 mg/L). Station TEL004 was considered a serious outlier and was removed (Figure 4.2).

Data were separated into their corresponding watershed basins and ditch groups, and their mean, minimum, maximum, and standard deviation computed (Appendix E, Tables E-4 through E-9). Multiple statistical comparisons were done using a non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for Post Hoc multiple comparisons ($\alpha = 0.05$). This analysis was able to subdivide and combine the different watershed basins and ditches into groups that were statistically similar.

Iron data showed that there were statistical differences (p< 0.05), and two distinct groups existed. The first, and with the highest iron average concentration (0.5 mg/L) included those stations from Banana Creek and both ditches. The second, with a combined average of 0.1 mg/L, included those stations from the Banana River, Indian River Lagoon, and Mosquito Lagoon. These two groups were used to compute the Screening Table values for iron. "Combined Background Water" values were calculated by averaging only those stations with the appropriate salinity (Appendix A, Table A-3).

Lead. Lead is widely distributed in the earth's crust, in the atmosphere and in the hydrosphere. Lead is present in natural waters as hydroxide, sulfide, chloride, carbonate complexes, complexed with organic molecules or adsorbed on particulate matter (Thurston et al., 1979). It enters water supplies at a multitude of stages such as: mining, smelting, precipitation, lead dust fallout, erosion and leaching of soil, industrial waste, engine exhaust, and the runoff of surfaces painted with lead-based products. Lead has been used in the manufacture of metal products for thousands of years, and a considerable amount of information is available regarding its harmful effects in living organisms (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Lead is not considered an essential element, and has no beneficial nutritional qualities. It is a serious cumulative body poison that tends to deposit in bone, but is also found in the brain, liver, kidney, aorta, and muscles. In humans, lead toxicity is known to cause brain damage, anemia, neurological and renal disorders, and even death. This biotoxicity results from long-term consumption rather than from occasional small doses (Faust and Aly, 1981; James and Evison, 1989; Thurston et al., 1979).

Lead was detected in 15 surface water sampling stations, that included representatives from Mosquito Lagoon, Banana River and fresh water ditches. This sample size did not allow for intricate statistical analysis. Data were separated into their corresponding watershed and ditch groups, and along with one-half the reported non-detection level of the other stations from the same group, were used to calculate the mean, minimum, maximum, and standard deviation (Appendix E, Tables E-5, E-6, and E-9). Screening Table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

Magnesium. Magnesium is the eighth most abundant element and occurs naturally in a variety of minerals as mica, chlorites, amphiboles, dolomites, augites, pyroxenes, granites, talc, silicates, and serpentines (Thurston et al., 1979).

Magnesium is a common constituent of natural waters and is an important contributor to its hardness. Its concentration can vary from zero to several hundred milligrams per liter, depending on the source and treatment of the water. Magnesium concentration has to be carefully monitored in drinking water because more than 125 mg/L can be cathartic or diuretic (Brown et al., 1962; Cowan, 1995).

Primary sources of magnesium for biological organisms are rocks, soils, air, and water. Plants and animals absorb magnesium from these primary sources and after dying, return it to the environment, creating a magnesium cycle in the food web (Thurston et al., 1979).

Magnesium was detected in all 46 surface water sampling stations, which allowed more intricate statistical analysis. Station NSC001, a fresh water ditch along NASA Causeway, was considered a serious outlier (5200 mg/L), and was removed from analysis (Figure 4.2).

Data were separated into their corresponding watershed basins and ditch groups, and their mean, minimum, maximum, and standard deviation computed (Appendix E, Tables E4 through E-9). Multiple statistical comparisons were done using a non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for Post Hoc multiple comparisons ($\alpha = 0.05$). This analysis was able

to subdivide and combine the different watershed basins and ditches into groups that were statistically similar.

Magnesium data showed that there were statistical differences (p< 0.05), and two distinct groups existed. The first, and with the highest magnesium average concentration (1147 mg/L) included those stations from Banana Creek, saline (sal \geq 6 ppt) ditches, Indian River Lagoon, and Mosquito Lagoon basins. The second, with a combined average of 317 mg/L, included those stations from the Banana River and fresh water ditches. These two groups were used to compute the Screening Table values for magnesium. "Combined Background Water" values were calculated by averaging only those stations with the appropriate salinity (Appendix A, Table A-3).

Manganese. Manganese is found in salts and minerals such as manganese dioxide, carbonate and silicate. It occurs in domestic water, industrial effluents and receiving streams. Its primary sources are soils, where it is present as manganese oxide, and acid discharges of industrial wastes and mines. It is harmlessly ingested by humans as a trace nutrient in food, but high concentrations can cause liver damage (Thurston et al., 1979).

In marine waters, manganese is known to rapidly assimilate and bioaccumulate in shellfish, leading to a potential vector for human contamination. It is not a problem in domestic waters and the primary concerns associated with its high concentrations are the brown color and bad taste of water (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

It is a vital micronutrient for flora and fauna, and its absence may cause leafing disorders in plants and reproductive problems in animals (Faust and Aly, 1981; Thurston et al., 1979).

Manganese was detected in 15 surface water sampling stations that included representatives from Banana Creek and both saline and fresh water ditches. This sample size did not allow for intricate statistical analysis. Data were separated into their corresponding watershed and ditch groups, and along with one-half the reported non-detection level of the other stations from the same group, were used to calculate the mean, minimum, maximum, and standard deviation (Appendix E, Tables E-4, E-6, and E-7). Screening Table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

Potassium. Potassium, a silvery white metal, is the seventh most abundant element on earth. It is present in natural waters, but usually less than magnesium, sodium or calcium. The principal sources of potassium are silicates, igneous, and sedimentary rocks. Potassium has no special significance in natural water quality and is generally used as a tracer element and an indicator of the geological origin of the water (Thurston et al., 1979).

Potassium metal may ignite spontaneously on contact with air at room temperature and reacts explosively with water to form potassium hydroxide and/or potassium oxide. It reacts with the moisture on skin and other tissues to form highly corrosive potassium hydroxide. Contact of metallic potassium with the skin, eyes, or mucous membranes causes severe burns, and thermal burns may occur due to ignition of the metal and liberated hydrogen. Potassium salts are essential for life. The potassium cation is the major cation in intracellular fluids, and is essential for nerve and heart function (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Potassium was detected in 45 surface water sampling stations, which allowed more intricate statistical analysis. The only station that did not have detectable concentrations of potassium was IRL03, located in the Indian River Lagoon (Figure 4.2).

Data were separated into their corresponding watershed basins and ditch groups, and their mean, minimum, maximum, and standard deviation computed (Appendix E, Tables E-4 through E-9). Multiple statistical comparisons were done using a non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for Post Hoc multiple comparisons ($\alpha = 0.05$). This analysis was able to subdivide and combine the different watershed basins and ditches into groups that were statistically similar.

Potassium data showed that there were statistical differences (p< 0.05), and three distinct groups existed. The first, and with the highest potassium average concentration (523 mg/L) included those stations from the Mosquito Lagoon basin. The second, with a combined average of 264 mg/L, included those stations from the Banana Creek, Indian River Lagoon, Banana River, and saline (sal ≥6 ppt) ditches. The third, had the lowest potassium average (33 mg/L), and included the stations from fresh water ditches. These three groups were used to compute the Screening Table values for potassium. "Combined Background Water" values, were calculated by averaging only those stations with the appropriate salinity (Appendix A, Table A-3).

Selenium. Chemical sources of selenium are rarely found in nature, but selenium is closely associated with sulfur, sulfides, and trace amounts are found in igneous rocks. Depending on its concentration, selenium can range from being an essential nutrient to being highly toxic to biological species. It is considered to be toxic to humans because its symptoms are similar to those of arsenic (Thurston et al., 1979).

Selenium is found in low concentrations in standing and marine waters, except for polluted or seleniferous soils where its concentration is higher. Pollution sources that commonly include selenium wastes are industries related to glass-making, electronics, ceramics, and xerography. Selenium is present in soils as

ferric selenite, calcium selenate, and in the elemental state which must be oxidized to be soluble in water (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Selenium enters the food chain almost exclusively via plants. It is passed through the food chain to fish, increasing its bioaccumulative toxicity to higher levels (Brown et al., 1962; Thurston et al., 1979).

Selenium was detected in 3 surface water sampling stations, that were all part of the Indian River Lagoon basin. These stations were IRL004 with 0.024 mg/L, IRL005 with 0.012 mg/L, and REF001 with 0.018 mg/L. This sample size did not allow for intricate statistical analysis. Data from this basin were separated, and along with one-half the reported non-detection level of the other stations from the same group, were used to calculate the mean, minimum, maximum, and standard deviation (Appendix E, Table E-8). Screening Table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

<u>Silver.</u> Silver is a brilliant white metal that is rare and expensive. It occurs in ores such as argentite, lead, lead-zinc, copper and gold. It is harder than gold and is very ductile and malleable. Pure silver has the highest electrical and thermal conductivity of all metals, and possesses the lowest contact resistance. Silver is stable in pure air and water, but tarnishes when exposed to ozone, hydrogen sulphide, or air containing sulfur (Thurston et al., 1979).

Silver has no biological role, and it is a non-essential, non-beneficial element, that is toxic to aquatic life. Silver is toxic because once it is absorbed, it accumulates indefinitely in tissues particularly the skin, eyes, and mucus membranes. Silver also has a bacteriological action and has been added to water as a disinfectant, at concentrations that do not harm human health. The toxicity and the degree of dissociation of silver compounds to aquatic life varies depending on species (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Silver was detected in 2 surface water sampling stations, that were all part of the Indian River Lagoon basin. These stations were IRL004 with 0.0014 mg/L and IRL005 with 0.0015 mg/L. This sample size did not allow for intricate statistical analysis. Data from this basin were separated, and along with one-half the reported non-detection level of the other stations from the same group, were used to calculate the mean, minimum, maximum, and standard deviation (Appendix E, Table E-8). Screening Table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

Sodium. Sodium, a soft, bright, silvery metal, is the fourth most abundant element on earth, comprising about 2.6% of the earth's crust. It is a very reactive element, and it is never found free in nature. (Thurston et al., 1979).

The most common compound is sodium chloride, but it occurs in many other minerals, such as soda ash (Na₂CO₃), baking soda (NaHCO₃), caustic soda (NaOH), Chile saltpeter (NaNO₃), di- and tri-sodium phosphates, sodium thiosulfate (hypo, Na₂S₂O₃ . $5H_2O$), and borax (Na₂B₄O₇ . $10H_2O$) (Faust and Aly, 1981; Thurston et al., 1979).

Sodium compounds are important to the paper, glass, soap, textile, petroleum, chemical, and metal industries. Soap is generally a sodium salt of certain fatty acids. The importance of common salt to animal nutrition has been recognized since prehistoric times (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Sodium was detected in all 46 surface water sampling stations, which allowed more intricate statistical analysis. Data were separated into their corresponding watershed basins and ditch groups, and their mean, minimum, maximum, and standard deviation computed (Appendix E, Tables E-4 through E-9). Multiple statistical comparisons were done using a non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for Post Hoc multiple comparisons ($\alpha = 0.05$). This analysis was able to subdivide and combine the different watershed basins and ditches into groups that were statistically similar.

Sodium data showed that there were statistical differences (p< 0.05), and three distinct groups existed. The first, and with the highest sodium average concentration (9031 mg/L) included those stations from the Mosquito Lagoon and the Indian River Lagoon basins. The second, with a combined average of 5690 mg/L, included those stations from the Banana Creek, Banana River, and saline (sal \geq 6 ppt) ditches. The third, had the lowest sodium average (895 mg/L), and included the stations from fresh water ditches. These three groups were used to compute the Screening Table values for sodium. "Combined Background Water" values were calculated by averaging only those stations with the appropriate salinity (Appendix A, Table A-3).

<u>Thallium</u>. Thallium is a soft, malleable, lustrous, silver-gray metal, that resembles aluminum chemically and lead physically. A heavy oxide builds up on thallium if exposed to air, and in the presence of water, the hydroxide is formed (Thurston et al., 1979).

Thallium occurs in crooksite, lorandite, and hutchinsonite. It is also present in pyrites and is recovered from the roasting of this ore in connection with the production of sulfuric acid (Thurston et al., 1979).

Natural thallium is a mixture of two isotopes. The element and its compounds are toxic and should be handled carefully. Contact of the metal with skin is dangerous, and when melting the metal adequate ventilation should be provided. Thallium is suspected of having a carcinogenic potential for humans. Thallium sulfate has been widely employed as a rodenticide and ant killer (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Thallium was detected in 2 surface water sampling stations, that were in the Mosquito Lagoon Indian River Lagoon basins. These stations were IRL003 with 0.003 mg/L and SJML02 with 0.002 mg/L. This sample size did not allow for intricate statistical analysis. Data from these basins were separated, and along with one-half the reported non-detection level of the other stations from the same group were used to calculate the mean, minimum, maximum, and standard deviation (Appendix E, Tables E-8 and E-9). Screening Table "Combined Background Water" values, were calculated by averaging those stations with the appropriate salinity (Appendix A, Table A-3).

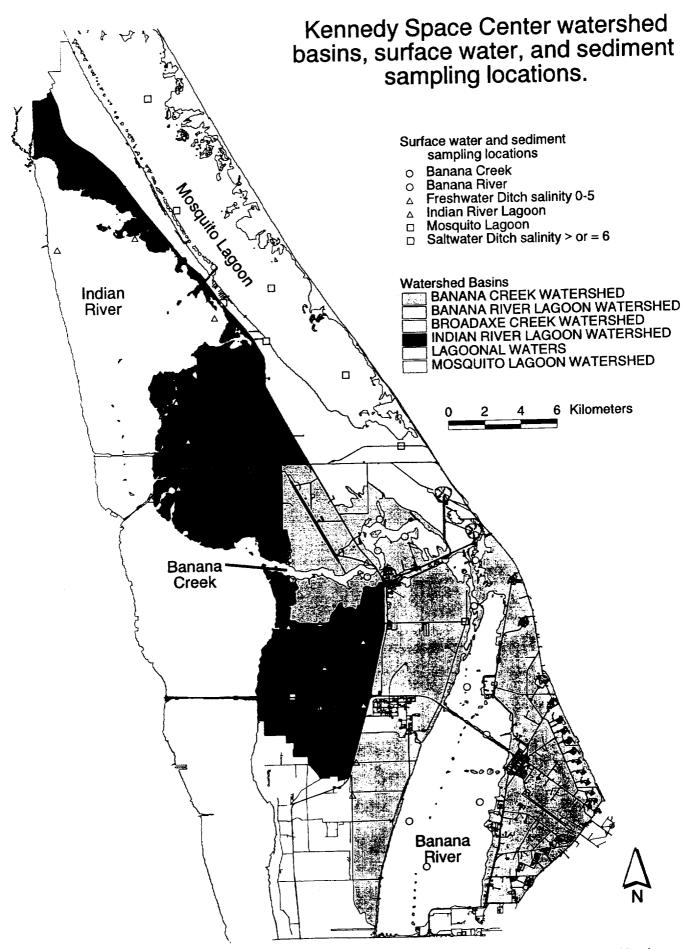


Figure 4.1. KSC Background surface water and sediment sampling locations in relation to watershed basins existing on Kennedy Space Center, FL. Symbols may represent more than one sample location.

			*	4
		·		

KSC Background Surface Water and Sediment Sampling Locations.

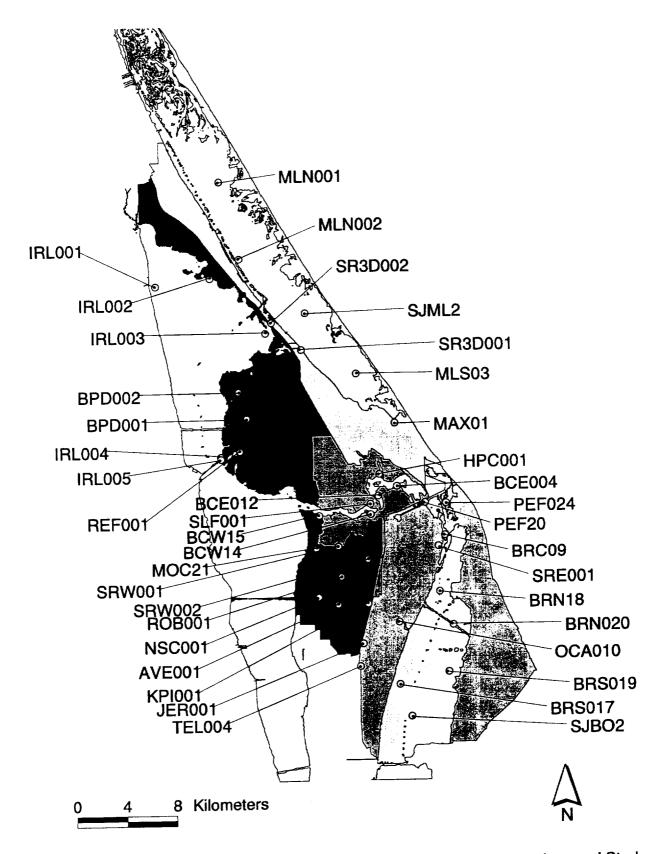


Figure 4.2. Surface water and sediment sampling locations for the KSC Background Study.

Table 4.1. List of parameters, EPA methods, and detection limits used to analyze surface water samples collected for the 1998 KSC Background Study. * = measurement made with a calibrated field instrument (YSI).

	EPA	Lab Reporting
	Method	Limit for surface water
Organochlorine pesticides		
	8081	0.05 μg/L
4,4' - DDD	8081	0.05 μg/L
4,4' - DDE	8081	0.05 μg/L
4,4' - DDT	8081	0.05 μg/L
Aldrin	8081	0.05 μg/L
Alpha - BHC	8081	0.05 μg/L
Beta - BHC	8081	0.05 μg/L 0.05 μg/L
Chlordane (alpha)		0.05 μg/L 0.05 μg/L
Chlordane (gamma)	8081	
Chlordane (Total)	8081	1 μg/L
Delta - BHC	8081	0.05 μg/L
Dieldrin	8081	0.05 μg/L
Endosulfan I	8081	0.05 μg/L
Endosulfan II (beta)	8081	0.05 μg/L
Endosulfan Sulfate	8081	0.05 μg/L
Endrin	8081	0.05 μg/L
Endrin Aldehyde	8081	0.05 μg/L
Endrin Ketone	8081	0.05 μg/L
Gamma - BHC (Lindane)	8081	0.05 μg/L
Heptachlor	8081	0.05 μg/L
Heptachlor Epoxide(a)	8081	0.05 μg/L
Heptachlor Epoxide(b)	8081	0.05 μg/L
Isodrin	8081	0.05 μg/L
Methoxychlor	8081	0.05 μg/L
Mirex	8081	0.05 μg/L
Toxaphene	8081	2 μg/L
Aroclors		
PCB-1016/1242	8082	1 μg/L
PCB-1221	8082	1 <i>µ</i> g/L
PCB-1232	8082	1 <i>µ</i> g/L
PCB-1248	8082	1 μg/L
PCB-1254	8082	1 <i>μ</i> g/L
PCB-1260	8082	1 μg/L
Chlorinated Herbicides		
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5-TP) (Silvex)	5 8151	0.5 μg/L
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	8151	0.5 <i>μ</i> g/L

Table 4.1. (cont.)

	EPA	Lab Reporting
	Method	Limit for surface water
Chlorinated Herbicides (cont.)		
2,4-Dichlorophenoxy acetic acid (2,4 - D)	8151	0.5 <i>μ</i> g/L
3,5-DCBA	8151	0.5 μg/L
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	8151	0.5 μg/L
4 - Nitrophenol	8151	0.5 μg/L
Acifluorfen	8151	0.5 μg/L
Bentazon	8151	0.5 μg/L
Chloramben	8151	0.5 μg/L
Dacthal	8151	0.5 μg/L
Dalapon	8151	0.5 μg/L
Dicamba	8151	0.5 μg/L
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic	8151	0.5 μg/L
acid		ο.ο μ <u>φ</u> Ε
Dinoseb	8151	0.5 μg/L
MCPA	8151	5 μg/L
MCPP	8151	5 μg/L
Pentachlorophenol	8151	0.5 μg/L
Picloram	8151	0.5 μg/L
Polyaromatic Hydrocarbons		0.0 µg L
1 - Methylnaphthalene	8310	0.5 μg/L
2 - Methylnaphthalene	8310	0.5 μg/L
Acenaphthene	8310	0.5 μg/L
Acenaphthylene	8310	0.1 μg/L
Anthracene	8310	0.5 μg/L
Benzo(a)anthracene	8310	0.05 μg/L
Benzo(a)pyrene	8310	0.05 μg/L
Benzo(b)fluoranthene	8310	0.1 μg/L
Benzo(g,h,i)perylene	8310	0.1 μg/L
Benzo(k)fluoranthene	8310	0.05 μg/L
Chrysene	8310	0.05 μg/L
Dibenzo(a,h)anthracene	8310	0.1 μg/L
luoranthene	8310	0.1 μg/L 0.1 μg/L
luorene	8310	0.1 μg/L 0.1 μg/L
ndeno(1,2,3-cd)pyrene	8310	0.05 μg/L
laphthalene	8310	0.5 μg/L
Phenanthrene	8310	0.05 μg/L
Pyrene	8310	0.05 μg/L

Table 4.1. (cont.)

	EPA	Lab Reporting
	Method	Limit for surface water
Metals		
Aluminum	200.7	0.05 mg/L
Antimony	200.7/	0.006 mg/L
Antimony	204.2	
Arsenic (as carcinogen)	200.7	0.01 mg/L
Barium	200.7	0.1 mg/L
Beryllium	200.7	0.001 mg/L
Cadmium	200.7	0.001 mg/L
Calcium	200.7	0.5 mg/L
Calciditi Chloride, Total	325.3	1 mg/L
Chromium (total)	200.7	0.01 mg/L
Cobalt	200.7	0.05 mg/L
	200.7/	0.05 mg/L
Copper	7211	
Iron	200.7	0.05 mg/L
Iron	200.7	0.005 mg/L
Lead	200.7	0.5 mg/L
Magnesium	200.7	0.01 mg/L
Manganese	7470	0.0002 mg/L
Mercury (inorganic)	200.7	0.01 mg/L
Nickel	200.7/	0.5 mg/L
Potassium	258.1	
Selenium	200.7	0.01 mg/L
Silver	200.7/	0.01 mg/L
Silver	7761	
Sodium	7770	0.5 mg/L
Thallium	279.2	0.004 mg/L
Vanadium	200.7	0.01 mg/L
Zinc	200.7	0.1 mg/L
Other Parameters		
Dissolved Oxygen	*	na
pH	*	na
Specific Conductivity	*	na
Temperature	*	na
Total dissolved solids	160.1	na
Total organic carbon	415.1	1 mg/L
Turbidity	180.1	na

Table 4.2. Name, number of hits, and frequency of analytes occurring above detection limits. For screening table values, the three parameters with only one hit out of 46 samples were considered as non-detect by the NASA/KSC Remediation Program Team.

Parameter	Number of Hits	Frequency %
Organochlorine pesticides		
4,4' - DDD	0	0
4,4' - DDE	0	0
4,4' - DDT	Ö	
Aldrin	0	0
Alpha - BHC	0	0
Beta - BHC	0	0
Chlordane (alpha)	0	0
Chlordane (gamma)	0	0
Chlordane (Total)	0	0
Delta – BHC	0	0
Dieldrin	2	0
Endosulfan I	0	4.35
Endosulfan II (beta)		0
Endosulfan Sulfate	0	0
Endrin	0	0
Endrin Aldehyde	0	0
Endrin Ketone	0	0
Gamma – BHC (Lindane)	0	0
Heptachlor	0	0
Heptachlor Epoxide(a)	0	0
Heptachlor Epoxide(b)	0	0
sodrin	0	0
Methoxychlor	0	0
Mirex	0	0
Toxaphene	0	0
Arociors	0	0
PCB-1016/1242		
PCB-1221	0	0
CB-1221	0	0
CB-1232	0	0
CB-1248	0	0
CB-1254 CB-1260	0	0
CB-120U	0	0

Table 4.2. (cont.).

Parameter	Number of Hits	Frequency %
Chlorinated Herbicides		
2-(2,4,5-	0	0
Trichlorophenoxy)propionic acid		
(2,4,5 - TP) (Silvex)		
2,4,5-Trichlorophenoxy acetic	0	0
acid (2,4,5 - T)		
2,4-Dichlorophenoxy acetic acid	0	0
(2,4 - D)		
3,5-DCBA	0	0
4-(2,4-Dichlorophenoxy)butyric	0	0
acid (2,4 - DB)		
4 - Nitrophenol	0	0
Acifluorfen	0	0
Bentazon	0	0
Chloramben	0	0
Dacthal	0	0
Dalapon	0	0
Dicamba	0	0
Dichloroprop [2-(2,4-	0	0
Dichlorophenoxy)proponic acid]		
Dinoseb	0	0
MCPA	0	0
MCPP	0	0
Pentachlorophenol	0	0
Picloram	0	0
Polyaromatic Hydrocarbons		
1 - Methylnaphthalene	0	0
2 - Methylnaphthalene	0	0
Acenaphthene	0	0
Acenaphthylene	0	0
Anthracene	0	0
Benzo(a)anthracene	2	4.35
Benzo(a)pyrene	3	6.52
Benzo(b)fluoranthene	0	0
Benzo(g,h,i)perylene	0	0
Benzo(k)fluoranthene	2	4.35
Chrysene	2	4.35
Dibenzo(a,h)anthracene	0	0
Fluoranthene	0	0
Fluorene	1	2.17

Table 4.2. (cont.).

Parameter	Number of Hits	Frequency %
Polyaromatic Hydrocarbon	ıs	, , , , , , , , , , , , , , , , , , ,
(cont.)		
Indeno(1,2,3-cd)pyrene	2	4.35
Naphthalene	1	2.17
Phenanthrene	0	0
Pyrene	0	0
Metals		
Aluminum	36	78.26
Antimony	19	41.30
Arsenic (as carcinogen)	9	19.57
Barium	0	0
Beryllium	5	10.87
Cadmium	0	0
Calcium	46	100.00
Chloride, Total	45	97.83
Chromium (total)	0	0
Cobalt	0	0
Copper	1	2.17
ron	35	76.09
_ead	15	32.61
Magnesium	46	100.00
Manganese	15	32.61
Mercury (inorganic)	0	0
Nickel	0	0
Potassium	45	97.83
Selenium	3	6.52
Silver	2	4.35
Godium	46	100
hallium	2	4.35
/anadium	0	0
linc	0	0
Other Parameters		
Dissolved Oxygen	41	89.13
otal Dissolved Solids	43	93.48
otal Organic Carbon	46	100

Table 4.3. List of parameters that were below detection limits (non-detect) for all surface water locations.

Owners ablasina masticidas (9091)
Organochlorine pesticides (8081)
4,4' - DDD
4,4' - DDE
4,4' - DDT
Aldrin
Alpha - BHC
Beta - BHC
Chlordane (alpha)
Chlordane (gamma)
Chlordane (Total)
Delta - BHC
Endosulfan I
Endosulfan II (beta)
Endosulfan Sulfate
Endrin
Endrin Aldehyde
Endrin Ketone
Gamma - BHC (Lindane)
Heptachlor
Heptachlor Epoxide(a)
Heptachlor Epoxide(b)
Isodrin
Methoxychlor
Mirex
Toxaphene
Aroclors (8082)
PCB-1016/1242
PCB-1221
PCB-1232
PCB-1248
PCB-1254
PCB-1260
Metals
Barium
Cadmium
Chromium (total)
Cobalt
Mercury (inorganic)
Nickel
Vanadium
Zinc

Chlorinated Herbicides (8151)
2-(2,4,5-Trichlorophenoxy)propionic
acid (2,4,5 - TP) (Silvex)
2,4,5-Trichlorophenoxy acetic acid
(2,4,5 - T)
4-(2,4-Dichlorophenoxy)butyric acid
(2,4 - DB)
2,4-Dichlorophenoxy acetic acid (2,4
- D)
3,5-DCBA
4 - Nitrophenol
Acifluorfen
Bentazon
Chloramben
Dacthal
Dalapon
Dicamba
Dichloroprop [2-(2,4-
Dichlorophenoxy)proponic acid]
Dinoseb
MCPA
MCPP
Pentachlorophenol
Picloram
Polyaromatic Hydrocarbons (8310)
1 - Methylnaphthalene
2 - Methylnaphthalene
Acenaphthene
Acenaphthylene
Anthracene
Benzo(b)fluoranthene
Benzo(g,h,i)perylene
Dibenzo(a,h)anthracene
Fluoranthene
Phenanthrene
Pyrene

SECTION 5 Sediment

5.1 Methods

5.1.1 Sample Site Selection

Location of the sediment sampling stations was determined based on the watershed basins surrounding Kennedy Space Center. A total of 40 stations were selected to incorporate samples from open lagoonal water, rivers, creeks, ditches, burrow pits, and impoundments (Figure 4.1). Sampling sites were chosen away from known SWMU sites. Because 1998 was a year of the "El Nino" weather phenomenon, the climate in Florida was very hot and dry. The resulting drought reduced the number of ditches available for sampling. A total of 18 ditches were sampled, with five located near KSC's industrial areas. Location and description of each sampling station are given in Appendix E, Table E-1.

The geographical location of all sampling stations was recorded using a Trimble Pathfinder Professional XL Geographic Positioning System (GPS) unit using real time differentially corrected methods (Trimble Navigation Limited 1994). Data were exported to North American Datum 1927 (NAD27) State Plane coordinates and overlayed in a Geographic Information System (GIS). Coordinate data of each sampling station are given in Appendix E, Table E-2.

5.1.2 Sampling

Sediment sampling and analysis were subcontracted by Dynamac to Environmental Conservation Laboratories (ENCO), in Orlando, Florida. ENCO technicians followed the requirements of the "NASA/KSC Sampling and Analysis Plan for RCRA Facility Investigations at Kennedy Space Center, Florida and Cape Canaveral Air Station, Florida, Volume 4" of the Generic Work Plans (SAP) (NASA 1996) and ENCO's FLDEP Comprehensive Quality Assurance Plan No. 960038.

Sample collection was done from shore, by wading, or in deeper water, by boat (dredging or coring). A total of 46 samples (includes 6 duplicates) were collected with decontaminated stainless steel equipment using latex gloves.

5.1.3 Chemical Analysis

Sediment was analyzed for 96 parameters that included organochlorine pesticides (EPA method 8081), aroclors (EPA method 8082), chlorinated herbicides (EPA method 8151), and polycyclic aromatic hydrocarbons (EPA method 8310), and metals. The EPA methods used for metal analyses varied according to the appropriate analyte and included EPA methods: 200.7, 204.2, 6010, 7210, 7471, 7520, 7610, 7770, and 7950. Other miscellaneous

parameters such as percent solids were analyzed using SM 2540G, pH by EPA method 9045, resistivity by Miller, texture by the sieve method, and total organic carbon was determined by MSA Part2, 29-3 (Table 5.1).

5.1.4 Data Analysis

Analysis were conducted using SPSS Version 9 (SPSS Inc., 1999), and SAS Version 6.02 (SAS Institute, 1998).

Frequency of analytes occurring above the detection limits was determined (Table 5.2). The NASA/KSC Remediation Program Team agreed that only those parameters with more than one detection would be used for analysis. Four analytes (dieldrin, benzo(g,h,i)perylene, indeno(1,2,3-cd)pyrene, and 2 - Methylnaphthalene) had only one detection and were, therefore, treated as non-detect (Table 5.4). These values, however, are given in Appendix F, Table F-5. Those with more than one detection were analyzed, but their non-detect values were substituted by one-half the reported level.

Basic statistic analyses, such as Normal Probability Plots, Shapiro-Wilks, and the Lilliefors tests, were used in SPSS to check for normality of distribution (α =0.05). The presence of significant outliers was also tested in SPSS using histograms, stem-and-leaf plots, and boxplots. Extreme outliers were removed, however, in situations where a particular parameter had very few hits, these hits were erroneously flagged as outliers and were therefore, not removed. These will be discussed in detail on an individual basis.

For those parameters with sufficient sample size (n>30), multiple statistical comparisons were done using SAS. Non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences between all watershed basins and data were combined. For the screening values table, data are given as twice the mean, and the "Combined Background Sediment" column includes the overall mean of all 46 sediment stations (Appendix A, Table A-4).

5.2 Results and Discussion

Of the 96 sediment parameters sampled, only 31 were detected. One organochlorine pesticide (Dieldrin) out of the 25 sampled, none of the six aroclors, and two chlorinated herbicides were detected (out of 18). Polyaromatic hydrocarbons (18) had 12 parameters that were detected in at least one station. The 23 metals tested were detected in high frequency except for seven (As, Ba, Be, Cd, Co, Ni, Se), that were always below detection (Table 5.3). Screening background tables are given in Appendix A, Table A-4.

Organochlorine Pesticides

Dieldrin was detected in only one sediment sampling station (7.9 μ g/kg at BRS017) located on the west side of the northern Banana River south of the NASA causeway (Figure 4.2). Because Dieldrin only had one detection (out of 46 samples), the NASA/KSC Remediation Program Team agreed to treat it as non-detect.

Chlorinated Herbicides

<u>2.4.5-Trichlorophenoxy acetic acid (2,4,5 - T)</u>. 2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T) was detected at two sediment sampling stations, both located in the Banana River. The first station was BRS019 with 17 μg/kg, and the second was PEF024 with 27 μg/kg. BRS019 is located in the east Banana River south of the NASA causeway, and PEF024 is in the northern Banana River in a region known as Pepper Flats. Both regions are relatively shallow with abundant seagrass beds (Figure 4.2).

Because 2,4,5-trichlorophenoxy acetic acid only had two detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these two 2,4,5-trichlorophenoxy acetic acid values, along with one-half the reported non-detection level of the other Banana River stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Table F-2). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

2,4-Dichlorophenoxy acetic acid (2,4 - D). 2,4-dichlorophenoxy acetic acid (2,4 - D) was detected at two sediment sampling stations, both located in the Banana River. The first station was BRS019 with 19 μ g/kg, and the second was PEF024 with 21 μ g/kg. BRS019 is located in the east Banana River south of the NASA causeway, and PEF024 is in the northern Banana River in a region known as Pepper Flats. Both regions are relatively shallow with abundant seagrass beds (Figure 4.2).

Because 2,4-dichlorophenoxy acetic acid (2,4 - D) only had two detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these two 2,4-dichlorophenoxy acetic acid (2,4 - D) values,

along with one-half the reported non-detection level of the other Banana River stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Table F-2). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Polycyclic Aromatic Hydrocarbons

2 –Methylnaphthalene, benzo(a)anthracene, benzo(a)pyrene, benzo(b)fluoranthene, benzo(g,h,i)perylene, benzo(k)fluoranthene, chrysene, fluoranthene, indeno(1,2,3-cd)pyrene, naphthalene, phenanthrene, and pyrene are all polycyclic aromatic hydrocarbons (PAHs). The name PAH generally refers to hydrocarbons containing two or more fused benzene rings that form as a result of incomplete combustion of organic compounds, by diagenesis, or by biosynthesis (Varanasi, 1989).

PAHs occur as needles, plates, crystals, leaflets, or prisms ranging from colorless to pale yellow to golden yellow. PAHs are found in gasoline and diesel motor vehicle exhaust, by-products of open fires or refuse burning, coal tar, coal tar pitch, coke tars or coke oven emissions, creosote, mineral oils, bitumens, industrial smoke and soot, cigarette and cigar tobacco and smoke, tar, or smoke condensates, and charcoal-broiled foods. (ATSDR, 1989; Cerniglia et al., 1980; Farrington et al., 1983; Healy and Young, 1979; NAS, 1983; Varanasi, 1989).

Levels of individual PAH have been measured in aquatic organisms worldwide. Reported values range from undetectable quantities (approximately 0.01 μ g/kg dry weight) to values in excess of 5000 μ g/kg dry weight for individual PAH in tissues of aquatic organisms (Pancirov and Brown, 1977; Varanasi, 1989).

Generally elevated concentrations of PAH can be correlated with the proximity of the organism to areas receiving chronic hydrocarbon discharge (Cerniglia et al., 1980; Clark et al., 1978; Farrington et al., 1983; Healy and Young, 1979; Varanasi, 1989). The majority of PAH measurements have been made on bivalve mollusks. Bivalves are sessile, can rapidly accumulate PAHs, and have little capacity for PAH metabolism (Clark et al., 1978; Cross et al., 1978; Pancirov and Brown, 1977). PAH concentrations in fish have been less well surveyed, but are usually low relative to invertebrates inhabiting the same environment. The low body burdens of PAH to fish are believed to be due to their ability to rapidly metabolize PAH. Although PAHs are metabolized and eliminated in fish, it is not known how they interact with birds, mammals, or humans. Dermal contact with high concentrations of PAHs may cause irritation, burning, inflammatory spots on the skin, eruption of blood vessels and sensitivity to sunlight. However, there are a number of epidemiologic and mortality studies that show increased incidences of cancer in humans exposed to mixtures of PAHs (ATSDR, 1989; Varanasi, 1989).

<u>2 –Methylnaphthalene</u>. 2 –Methylnaphthalene was detected in only one sediment sampling station (140 μg/kg at SR3-D2), a saline-water ditch located on State Route 3 just south of Haulover Canal (Figure 4.2). Because 2 – Methylnaphthalene only had one detection (out of 46 samples), the NASA/KSC Remediation Program Team agreed to treat it as non-detect.

Benzo(a)anthracene. Benzo(a)anthracene was detected at three sediment sampling stations, located in the Mosquito Lagoon, Indian River Lagoon, and Banana River watershed basins. The first station was SR3-D1 with 5.9 μg/kg, the second was PEF20 with 4.7 μg/kg, and the third was IRL004 with 12 μg/kg. SR3-D1 is a ditch located on State Route 3 approximately three miles north of the State Route 406 intersection, PEF20 is in the northern Banana River in a region known as Pepper Flats, and IRL004 is in the Indian River Lagoon just north of the State Route 406 causeway (Figure 4.2).

Because benzo(a)anthracene only had three detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these three benzo(a)anthracene values, along with one-half the reported non-detection level of the other stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-2 through F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

<u>Benzo(a)pyrene</u>. Benzo(a)pyrene was detected at four sediment sampling stations located in the Indian River Lagoon (three stations) and Banana River (one station) watershed basins. The Indian River Lagoon stations were IRL001, IRL002, and IRL 003 with 3 μ g/kg, 11 μ g/kg, and 5 μ g/kg respectively. Station BRN18 had 3 μ g/kg and is located north of the NASA causeway across from the Integrate, Transfer, and Launch (ITL) area on CCAFS(Figure 4.2).

Because benzo(a)pyrene only had four detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these four benzo(a)pyrene values, along with one-half the reported non-detection level of the other Banana River and Indian River Lagoon stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basins (Appendix F, Tables F-2 and F-3). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Benzo(b)fluoranthene. Benzo(b)fluoranthene was detected at five sediment sampling stations located in the Mosquito Lagoon, Indian River Lagoon, and Banana River watershed basins. The first station was IRL004 with 12 μ g/kg, the second was IRL005 with 13 μ g/kg, the third was OCA010 with 4.0 μ g/kg, the fourth was SLF001 with 4 μ g/kg, and the fifth was SRE001 with 14 μ g/kg. Stations IRL004 and IRL005 are located in the Indian River Lagoon north and

south of the State Route 406 causeway, respectively. OCA010 is a fresh-water ditch located on Avenue 10 south of NASA's Occupational & Checkout building, SLF001 is a fresh-water ditch southeast of NASA's Shuttle Landing Facility, and SRE001 is a ditch located on Schwartz Road east of State Route 3 (Figure 4.2).

Because benzo(b)fluoranthene only had five detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these five benzo(b)fluoranthene values, along with one-half the reported non-detection level of the other Mosquito Lagoon, Indian River Lagoon, and Banana River stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basins (Appendix F, Tables F-2 through F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Benzo(g,h,i)perylene. Benzo(g,h,i)perylene was detected at only one sediment sampling station (11 μ g/kg at IRL004). This station is located in the Indian River Lagoon just north of the State Route 406 causeway (Figure 4.2). Because benzo(g,h,i)perylene only had one detection (out of 46 samples), the NASA/KSC Remediation Program Team agreed to treat it as non-detect.

Benzo(k)fluoranthene. Benzo(k)fluoranthene was detected at four sediment sampling stations located in the Indian River Lagoon (two stations), and in two ditches near the KSC industrial areas, part of the Banana River watershed basin. The Indian River Lagoon stations were IRL004 and IRL005 with 10 $\mu g/kg$ and 2 $\mu g/kg$, respectively. Station OCA010 (duplicate) had 3 $\mu g/kg$ and SRE001 had 9 $\mu g/kg$. Indian River Lagoon stations are located north and south of the State Route 406 causeway, respectively. OCA010 is a fresh-water ditch located on Avenue 10 south of NASA's Occupational & Checkout building, and SRE001 is a ditch located on Schwartz Road east of State Route 3 (Figure 4.2).

Because benzo(k)fluoranthene only had four detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these four benzo(k)fluoranthene values, along with one-half the reported non-detection level of the other Banana River and Indian River Lagoon stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basins (Appendix F, Tables F-2 and F-3). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

<u>Chrysene</u>. Chrysene was detected at six sediment sampling stations located in the Indian River Lagoon (two stations), the Banana River, and in two ditches (plus duplicate) near the KSC industrial areas. The Indian River Lagoon stations were IRL004 and IRL005 with 42 μ g/kg and 6.9 μ g/kg, respectively. Station OCA010 had 5.3 μ g/kg and its duplicate sample had 3.6 μ g/kg, SRE001 had 13

 $\mu g/kg$, and PEF20 had 4.3 $\mu g/kg$. Indian River Lagoon stations are located north and south of the State Route 406 causeway, respectively. OCA010 is a fresh-water ditch located on Avenue 10 south of NASA's Occupational & Checkout building, and SRE001 is a ditch located on Schwartz Road east of State Route 3. These three stations are part of the Banana River watershed basin. PEF20 is in the northern Banana River, east of the Turn Basin in a region known as Pepper Flats (Figure 4.2).

Because chrysene only had six detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these six chrysene values, along with one-half the reported non-detection level of the other Banana River and Indian River Lagoon stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-2 and F-3). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

<u>Fluoranthene</u>. Fluoranthene was detected at three sediment sampling stations located in three ditches near the KSC industrial areas. Station SLF001 had 9.6 μg/kg, SRE001 had 67 μg/kg, and SRW001 had 4.5 μg/kg. SLF001 is a freshwater ditch located southeast of NASA's Shuttle Landing Facility, and SRE001 and SRW001 are ditches located on Schwartz Road. The former ditch is east of State Route 3, while the latter is west (Figure 4.2). Stations SLF001 and SRW001 are part of the Banana Creek watershed basin, and SRE001 part of the Banana River basin.

Because fluoranthene only had three detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these three fluoranthene values, along with one-half the reported non-detection level of the other Banana River and Banana Creek stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basins (Appendix F, Tables F-1 and F-2). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Indeno(1,2,3-cd)pyrene. Indeno(1,2,3-cd)pyrene was detected at only one sediment sampling station (9 μ g/kg at IRL004). This station is located in the Indian River Lagoon just north of the State Route 406 causeway (Figure 4.2). Because indeno(1,2,3-cd)pyrene only had one detection (out of 46 samples), the NASA/KSC Remediation Program Team agreed to treat it as non-detect.

Naphthalene. Naphthalene was detected at four sediment sampling stations located in four ditches near the KSC industrial areas. Station SR3-D1 had 54 μ g/kg, its duplicate had 24 μ g/kg, SR3-D2 had 47 μ g/kg, and SRW001 had 67 μ g/kg. SR3-D1 is a ditch located on State Route 3 approximately 3 miles north of the State Route 406 intersection, SR3-D2 is a saline-water ditch also

located on State Route 3 just south of Haulover Canal, and SLF01 is a freshwater ditch located southeast of NASA's Shuttle Landing Facility (Figure 4.2). Stations SR3-D1 and SR3-D2 are part of the Mosquito Lagoon watershed basin, and SLF01 is part of the Banana Creek watershed basin.

Because naphthalene only had four detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these four naphthalene values, along with one-half the reported non-detection level of the other Mosquito Lagoon and Banana Creek stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-1 and F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Phenanthrene. Phenanthrene was detected at three sediment sampling stations located in three ditches near the KSC industrial areas. Station BRN18 had 7.3 $\mu g/kg$, SRE001 had 7.5 $\mu g/kg$, and SJML02 had 10 $\mu g/kg \times 10^{-3}$ mg/kg. BRN18 is located north of the NASA causeway across from the ITL area, SRE001 is a ditch located on Schwartz Road, and SJML02 is southeast of Haulover Canal, north of Cucumber Island (Figure 4.2). Stations SRE001 and BRN18 are part of the Banana River basin, and SJML02 is part of the Mosquito Lagoon basin.

Because phenanthrene only had three detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these three phenanthrene values, along with one-half the reported non-detection level of the other Mosquito Lagoon and Banana River stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-2 and F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

<u>Pyrene</u>. Pyrene was detected at two sediment sampling stations located in a fresh-water ditch (SRE001) and in Mosquito Lagoon (SJML02). Station SRE001 had 34 μ g/kg, and SJML02 had 15 μ g/kg. SRE001 is a fresh-water ditch located on Schwartz Road east of State Route 3, and SJML02 is southeast of Haulover Canal, north of Cucumber Island (Figure 4.2). Station SRE001 is part of the Banana River basin, and SJML02 is part of the Mosquito Lagoon basin.

Because pyrene only had two detection hits, statistical comparisons between the different watershed basins could not be done. Therefore, these two pyrene values, along with one-half the reported non-detection level of the other Mosquito Lagoon and Banana River stations were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-2 and F-4). The Screening table "Combined Background Sediment"

values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Metals

Aluminum. Aluminum is the third most abundant element in the earth's crust, exceeded by oxygen (47%) and silicon (28%). Because of its strong affinity to oxygen, aluminum never occurs as a metal in nature, and is found in the form of alumina. Aluminum is commonly distributed throughout rocks and soils such as silicates and clays. It is ductile, nonmagnetic, and an excellent conductor of heat and electricity (Brown et al., 1962; Thurston et al., 1979).

Aluminum is widely used in many kinds of products because its properties give it special advantages over other materials. Aluminum is a major factor in the human diet, and is found in buffered aspirin, antacids, antidiarrheal drugs, cake mixes, self-rising flour, processed cheese, baking powder, food starch modifiers, pickling salts and anti-caking agents. It can also be acquired from beverage cans, aluminum foil, and aluminum pots and pans. Aluminosilicates are found in the dust from talcum powder, baby powder, cat-box litter, cement, asphalt mixes, tobacco smoke, and ashes (Brown et al., 1962; Faust and Aly, 1981; Moore and Ramamoorthy, 1947).

Aluminum was detected at all sediment sampling stations, which allowed more intricate statistical analysis. Two extreme outliers were detected at stations SRW002 (5300 mg/kg) and MLN002 (4500 mg/kg), and were removed from data analyses.

The mean, minimum, maximum, and standard deviation of each watershed basin was calculated and results are given in Appendix F, Tables F-1 through F-4. A Non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

Arsenic. Arsenic is a shiny, gray, brittle element and one of the most widely distributed in the earth's crust and in the biosphere. It possesses both metallic and non-metallic properties. Its compounds may be organic or inorganic, are present everywhere in nature, are insoluble in water, and occur mostly as arsenides and arsenopyrites (Thurston et al., 1979).

Arsenic is very poisonous and can kill in relatively small amounts. It is absorbed in the gastrointestinal tract, lungs, or skin. The systemic effects of arsenic poisoning include pain, nausea, diarrhea, abnormal heart function, impaired nerve function, blood-vessel damage, liver or kidney damage, and a pattern of

skin abnormalities (Faust and Aly, 1981; Thurston et al., 1979). It distributes itself throughout the body and is excreted via urine, feces, sweat, or skin epithelium (Faust and Aly, 1981).

Arsenic, like mercury, undergoes transformation in the environment through the metabolic activities of microorganisms, especially bacteria and fungi. This indicates that there is a biological cycle for arsenic within the aquatic environment (Moore and Ramamoorthy, 1947).

Arsenic was detected in 20 sediment sampling stations, which did not allow for more intricate statistical analysis. One extreme outlier was detected at station TEL004 (5.4 mg/kg), and was removed from data analyses.

Because arsenic had a relatively low number of hits, statistical comparisons between the different watershed basins could not be done. Therefore, these data values, along with one-half the reported non-detection levels were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-1 through F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

<u>Calcium</u>. Calcium is the fifth most common element in order of abundance in the earth's crust. It is present in water supplies as a result of its passage through or over deposits of limestone, marble, chalk, calcite, alabaster, selenite, dolomite, gypsum, and gypsiferous shale. It can also be found in natural brines, salt beds, and as a byproduct of chemical industries (Thurston et al., 1979).

Calcium is the most abundant mineral (90%) in the body, and is found mostly in bones and teeth. That's why a diet rich in calcium is so important, particularly when bones are growing and developing. Even after full bone development, an adequate calcium intake is necessary to keep bones strong and healthy. When calcium is low, the blood tends to become overly acidic. This promotes disease conditions such as: cancer, arthritis, heart disease, Alzheimer's, Parkinson's, colitis, asthma, muscle pains, bursitis, sinusitis, diabetes, migraine, and even depression (Brown et al., 1962; Thurston et al., 1979).

Calcium was detected in 45 sediment sampling stations, which allowed more intricate statistical analyses. Data were first separated into their corresponding watershed, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled

"Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

<u>Chromium</u>. Chromium, the seventeenth most abundant non-gaseous substance on earth, is found in air, soil, some foods, and biological systems, but is rare in natural waters (Brown et al., 1962; Faust and Aly, 1981).

It occurs in nature mainly in chromite or the red lead ore, crocoite. Chromium occurs in oxidation states 0, 2⁺, 3⁺, and 6⁺, but only the hexavalent is toxic. It is stable and soluble in sea waters by forming a complex with other compounds such as amino acids and polybasic organic acids. Complex formation with organic compounds is affected by chloride in acidic solutions, and by magnesium and calcium ions in alkaline solutions (Brown et al., 1962).

Chromium is widely used in industries such as metallurgy and the manufacture of leather tanning and pigments, as a coloring agent in ceramic glazes, in glass making, fungicides, wood preservatives, and as a rust inhibitor in cooling-tower recirculating water systems (Faust and Aly, 1981; Thurston et al., 1979).

Chromium plays an essential role in some metabolic processes, but it is toxic in high concentrations. This toxicity can affect the metabolization of insulin and glucagon, the mucous membranes, respiratory disorders, and lung cancer. This toxicity also affects fishes and aquatic invertebrates, depending on species, pH, temperature, and chromium oxidation state (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Chromium was detected in 37 sediment sampling stations, which allowed for more intricate statistical analysis. Two extreme outliers were detected at stations MLN002 (11 mg/kg) and SRW002 (13 mg/kg), and were removed from data analyses.

Data were first separated into their corresponding watershed, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

<u>Copper</u>. Copper is a trace element that is usually present in the divalent state as a natural or native metal. It is reddish, malleable, ductile, and a good conductor of heat and electricity. Primary copper ores are sulfides, oxides, and carbonates. Copper is often alloyed with other metals to form brasses and bronzes. Copper is

used extensively in electrical wire, coinage metal, as an agricultural poison, and water purifier (Brown et al., 1962; Thurston et al., 1979).

Copper is an essential trace mineral that is present in all of the body tissues. Good sources of copper are oysters and other shellfish, whole grains, beans, nuts, potatoes, dark leafy greens, dried fruits such as prunes, cocoa, black pepper, and yeast (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Copper is essential for the synthesis of chlorophyll and the propagation of leaves in plants. In animals, it is necessary for metabolism, blood chemistry, and keeping the blood vessels, nerves, immune system, and bones healthy. The toxicity of copper to aquatic life depends strongly on pH. At lower alkalinity, copper toxicity is greater to aquatic life depending on species, size, and type of exposure. Waters with high copper concentration are tolerated by many species, depending on the species and amount of exposure (Brown et al., 1962; Tchobanoglous and Schroeder, 1985, Thurston et al., 1979).

Copper was detected in only six sediment sampling stations, which did not allow for intricate statistical analysis. Two extreme outliers were detected at stations IRL005 (380 mg/kg) and TEL004 (18 mg/kg), and were removed from data analyses.

Because copper had a relatively low number of hits, statistical comparisons between the different watershed basins could not be done. Therefore, these data values, along with one-half the reported non-detection levels were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-1 through F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

<u>Iron</u>. Iron is the fourth most abundant metal in the earth's crust common in many types of rocks and soils, particularly clays. Major sources of iron pollution are industrial wastewaters, mines, and iron-bearing groundwaters. Iron is present as insoluble ferric salts and pyrite, in natural organometallic or humic compounds, and in colloidal forms (Thurston et al., 1979).

Ferrous Fe⁺² and ferric Fe⁺³ ions are the primary forms of concern in the aquatic environment. When found in high concentrations in the water, iron produces a distinguishing color that serves as a good pollution indicator. Iron also forms hydroxide or oxide precipitates which, if they flocculate or gelatinize, can be detrimental to benthic species and fish eggs. Iron is a trace element that is a very important component of vertebrate erythrocytes and of the blood of some invertebrates. It is also a limiting factor in the growth of anaerobic bacteria, algae, and other plants (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Iron was detected in all 46 sediment sampling stations, which allowed more intricate statistical analysis. Two outliers were detected at stations SRW002 (5600 mg/L), and at MLN002 (5400 mg/L), and were removed from data analyses.

Data were first separated into their corresponding watershed, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

<u>Lead</u>. Lead is widely distributed in the earth's crust, in the atmosphere and in the hydrosphere. Lead is present in natural waters as hydroxide, sulfide, chloride, carbonate complexes, complexed with organic molecules or adsorbed on particulate matter (Thurston et al., 1979). It enters water supplies at a multitude of stages such as: mining, smelting, precipitation, lead dust fallout, erosion and leaching of soil, industrial waste, engine exhaust, and the runoff of surfaces painted with lead-based products. Lead has been used in the manufacture of metal products for thousands of years, and a considerable amount of information is available regarding its harmful effects in living organisms (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Lead is not considered an essential element, and has no beneficial nutritional qualities. It is a serious cumulative body poison that tends to deposit in bone, but is also found in the brain, liver, kidney, aorta, and muscles. In humans, lead toxicity is known to cause brain damage, anemia, neurological and renal disorders, and even death. This biotoxicity results from long-term consumption rather than from occasional small doses (Faust and Aly, 1981; James and Evison, 1989; Thurston et al., 1979).

Lead was detected in 37 sediment sampling stations, which allowed for intricate statistical analysis. Data were first separated into their corresponding watershed, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

Magnesium. Magnesium is the eighth most abundant element and occurs naturally in a variety of minerals as mica, chlorites, amphiboles, dolomites, augites, pyroxenes, granites, talc, silicates, and serpentines (Thurston et al., 1979).

Magnesium is a common constituent of natural waters and is an important contributor to its hardness. Its concentration can vary from zero to several hundred milligrams per liter, depending on the source and treatment of the water. Magnesium concentration has to be carefully monitored because more than 125 mg/L can be cathartic or diuretic (Brown et al., 1962; Cowan, 1995).

Primary sources of magnesium for biological organisms are rocks, soils, air, and water. Plants and animals absorb magnesium from these primary sources and after dying, return it to the environment, creating a magnesium cycle in the food web (Thurston et al., 1979).

Magnesium was detected in all 46 sediment sampling stations, which allowed more intricate statistical analysis. Station MLN02 was considered a serious outlier (6500 mg/L), and was removed from analysis (Figure 4.2).

Data were first separated into their corresponding watershed, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

Manganese. Manganese is found in salts and minerals such as manganese dioxide, carbonate, and silicate. It occurs in domestic water, industrial effluents, and receiving streams. Its primary sources are soils, where it is present as manganese oxide, and acid discharges of industrial wastes and mines. It is harmlessly ingested by humans as a trace nutrient in food, but high concentrations can cause liver damage (Thurston et al., 1979).

In marine waters, manganese is known to rapidly assimilate and bioaccumulate in shellfish, leading to a potential vector for human contamination. It is not a problem in domestic waters and the primary concerns associated with its high concentrations are the brown color and bad taste of water (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

It is a vital micronutrient for flora and fauna, and its absence may cause leafing disorders in plants and reproductive problems in animals (Faust and Aly, 1981; Thurston et al., 1979).

Manganese was detected in 44 sediment sampling stations, allowed for intricate statistical analysis. Data were first separated into their corresponding watershed, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

Mercury. Mercury is a silver-white liquid metal that is non-essential and non-beneficial. There are several mercury-bearing minerals in nature, but sulfides, cinnabar, and metacinnabar, are the most common. Because of its tendency to vaporize, strong ligand affinity and ease of adsorption onto surfaces, mercury is widely distributed in rocks, soils, air and water. It enters aquatic systems through runoff of mining, agriculture, and waste discharges. Its concentration in unpolluted waters is less than 0.1 μ g/L, and in marine waters it ranges from 0.03 to 0.2 μ g/L. It is present in natural waters as simple chloride and hydroxide complexes, depending on pH and chloride ion concentrations (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

It may be methylated in aquatic environments by both biological and non-biological processes, and it is commonly present as monomethylmercury and dimethylmercury. Certain microorganisms, depending on pH and temperature, have the ability to convert inorganic forms of mercury to the highly toxic monomethyl or dimethyl compounds, thus making any mercury isotope potentially harmful to the environment (Faust and Aly, 1981; Thurston et al., 1979).

Because of its toxic nature, mercury has been extensively studied. Toxicity may be acute or chronic varying with the form of mercury and its mode of entry into the organism. The majority of mercury present in an aquatic system will be found in the sediment where it is easily bioconcentrated in shellfish and fish. Because of methylation and bioconcentration of methylmercury, its limits must take into consideration the food chain path from aquatic organisms to humans. For humans, the maximum daily intake levels from all sources (air, water, and food) should not exceed 30 μ g/day (Brown et al., 1962; Thurston et al., 1979).

Mercury was detected in only five sediment sampling stations, which did not allow for statistical comparisons between the different watershed basins. Therefore, these data values, along with one-half the reported non-detection levels were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-1 through F-4). The Screening table "Combined Background Sediment" values were computed by

averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Potassium. Potassium, a silvery white metal, is the seventh most abundant element on earth. It is present in natural waters, but usually less than magnesium, sodium, or calcium. The principal sources of potassium are silicates, igneous, and sedimentary rocks. Potassium has no special significance in natural water quality and is generally used as a tracer element and an indicator of the geological origin of the water (Thurston et al., 1979).

Potassium metal may ignite spontaneously on contact with air at room temperature, and reacts explosively with water to form potassium hydroxide and/or potassium oxide. It reacts with the moisture on skin and other tissues to form highly corrosive potassium hydroxide. Contact of metallic potassium with the skin, eyes, or mucous membranes causes severe burns, and thermal burns may occur due to ignition of the metal and liberated hydrogen. Potassium salts are essential for life. The potassium cation is the major cation in intracellular fluids, and is essential for nerve and heart function (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Potassium was detected in 42 sediment sampling stations, which allowed more intricate statistical analysis. One extreme outlier was detected at station MLN002 (2100 mg/kg), and were removed from data analyses.

Data were first separated into their corresponding watersheds, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

Silver. Silver is a brilliant white metal that is rare and expensive. It occurs in ores such as argentite, lead, lead-zinc, copper and gold. It is harder than gold and is very ductile and malleable. Pure silver has the highest electrical and thermal conductivity of all metals, and possesses the lowest contact resistance. Silver is stable in pure air and water, but tarnishes when exposed to ozone, hydrogen sulphide, or air containing sulfur (Thurston et al., 1979).

Silver has no biological role, and it is a non-essential, non-beneficial element, that is toxic to aquatic life. Silver is toxic because once it is absorbed, it accumulates indefinitely in tissues particularly the skin, eyes, and mucus membranes. Silver also has a bacteriological action and has been added to water as a disinfectant, at concentrations that do not harm human health. The

toxicity and the degree of dissociation of silver compounds to aquatic life varies depending on species (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Silver was detected in 5 sediment sampling stations. Because of this relatively low number of hits, statistical comparisons between the different watershed basins could not be done. Therefore, these data values, along with one-half the reported non-detection levels were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-1 through F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Sodium. Sodium, a soft, bright, silvery metal, is the fourth most abundant element on earth, comprising about 2.6% of the earth's crust. It is a very reactive element, and it is never found free in nature. (Thurston et al., 1979).

The most common compound is sodium chloride, but it occurs in many other minerals, such as soda ash (Na₂CO₃), baking soda (NaHCO₃), caustic soda (NaOH), Chile saltpeter (NaNO₃), di- and tri-sodium phosphates, sodium thiosulfate ((hypo, Na₂S₂O₃) . $5H_2O$), and borax (Na₂B₄O₇ . $10H_2O$) (Faust and Aly, 1981; Thurston et al., 1979).

Sodium compounds are important to the paper, glass, soap, textile, petroleum, chemical, and metal industries. Soap is generally a sodium salt of certain fatty acids. The importance of common salt to animal nutrition has been recognized since prehistoric times (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Sodium was detected in 42 sediment sampling stations, which allowed more intricate statistical analysis. One extreme outlier was detected at station MLN001 (12000 mg/kg), and was removed from data analyses.

Data were first separated into their corresponding watersheds, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

<u>Thallium</u>. Thallium is a soft, malleable, lustrous, silver-gray metal, that resembles aluminum chemically and lead physically. A heavy oxide builds up on

thallium if exposed to air, and in the presence of water, the hydroxide is formed (Thurston et al., 1979).

Thallium occurs in crooksite, lorandite, and hutchinsonite. It is also present in pyrites and is recovered from the roasting of this ore in connection with the production of sulfuric acid (Thurston et al., 1979).

Natural thallium is a mixture of two isotopes, and it is suspected of carcinogenic potential for humans. Thallium sulfate has been widely employed as a rodenticide and ant killer (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Thallium was detected in three sediment sampling stations, which did not allow for intricate statistical analysis. Two extreme "outliers" were detected at stations TEL004 (5.4 mg/kg) and NSC001 (2.9 mg/kg), but they were not removed from data analyses.

Because thallium had a relatively low number of hits, statistical comparisons between the different watershed basins could not be done. Therefore, these data values, along with one-half the reported non-detection levels were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-1 through F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

<u>Vanadium</u>. Pure vanadium is a silvery metal, and is soft and ductile. It has good corrosion resistance to alkalis, sulfuric and hydrochloric acid, and salt water. It is found in numerous minerals such as carnotite, roscoelite, vanadinite, and patronite. Vanadium is also found in phosphate rock, certain iron ores, meteorites, and in some crude oils in the form of organic complexes (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

Vanadium metal is important in a number of areas. Its structural strength and neutron cross section properties makes it useful in nuclear applications. The metal is used for producing rust-resistant springs and steels used for making tools. About 80% of the vanadium now produced is used as ferrovanadium or as a steel additive. The metal oxidizes readily above 660°C to form V_2O_5 , and this pentoxide is used in ceramics and as a chemical catalyst. Vanadium compounds are also used for dyeing and printing fabrics (Faust and Aly, 1981; Thurston et al., 1979).

Vanadium and its compounds are toxic and should be handled with care. However, vanadium is essential to sea squirts that can bioaccumulate concentrations a million times higher than that of sea water. Vanadium is also a necessary part of the diet of rats and chicks, but only in very small amounts.

Deficiencies cause reduced growth and impair reproduction (Thurston et al., 1979).

Vanadium was detected in 37 sediment sampling stations, which allowed more intricate statistical analyses. Two extreme outliers were detected at stations SRW002 (11 mg/kg) and MLN002 (12 mg/kg), and were removed from data analyses.

Data were first separated into their corresponding watershed, and their mean, minimum, maximum, and standard deviation computed (Appendix F, Tables F-1 through F-4). A non-parametric analysis of variance (ANOVA) with the Bonferroni procedure for multiple comparisons was used to check if there were statistically significant differences (α =0.05) between the different basin groups. Results did not provide enough evidence for statistically significant differences. Therefore, for screening background values (Appendix F, Table F-4), all sediment data were averaged into a group labeled "Combined Background Sediment", and data are given as twice the mean (USEPA, 1998).

Zinc. Zinc is present in natural waters mainly as hydroxide, chloride, and carbonate depending on pH and other factors. The hydroxides are usually in the form of colloids or adsorbed on particulate matter. There is also evidence to suggest that much of the zinc in natural waters is contained in organic complexes. Zinc is essential for human growth, but can be toxic to lower organisms. Its toxicity to aquatic organisms depends on various factors such as temperature, pH, and dissolved oxygen (Brown et al., 1962; Faust and Aly, 1981; Thurston et al., 1979).

The natural concentration of zinc in uncontaminated water is between 0.2-5.0 μg dm⁻³, the average concentration in US drinking waters is between 0.06 and 7.0 mg/L, and oceanic surface waters contain between 0.4 and 12.6 μg dm⁻³. Zinc enters the water supply from sources including industrial waste products, the deterioration of galvanized pipes, and from the dezincification of brass (Faust and Aly, 1981; Thurston et al., 1979).

Zinc was detected in only ten sediment sampling stations, and because of this relatively low number of hits, statistical comparisons between the different watershed basins could not be done. Therefore, these data values, along with one-half the reported non-detection levels were used to compute the mean, minimum, maximum, and standard deviation for the corresponding basin (Appendix F, Tables F-1 through F-4). The Screening table "Combined Background Sediment" values were computed by averaging all hits and one-half the reported non-detections (Appendix A, Table A-4).

Table 5.1. List of parameters, the EPA methods used to analyze each, and detection limits, in sediment samples collected for the KSC Background Study.

	EPA	Lab Reporting
	Method	Limit for sediment
Organochlorine Pesticides		
4,4' - DDD	8081	3.3 μg/kg
4,4' - DDE	8081	3.3 μg/kg
4,4' - DDT	8081	3.3 μg/kg
Aldrin	8081	3.3 μg/kg
Alpha - BHC	8081	3.3 μg/kg
Beta - BHC	8081	3.3 μg/kg
Chlordane, alpha or gamma	8081	1 μg/kg
Chlordane (Total)	8081	3.3 μg/kg
Delta - BHC	8081	3.3 μg/kg
Dieldrin	8081	3.3 μg/kg
Endosulfan I	8081	3.3 μg/kg
Endosulfan II (beta)	8081	3.3 μg/kg
Endosulfan Sulfate	8081	3.3 μg/kg
Endosulian Sunate	8081	3.3 μg/kg
	8081	3.3 μg/kg
Endrin Aldehyde Endrin Ketone	8081	3.3 μg/kg
Gamma - BHC (Lindane)	8081	3.3 μg/kg
	8081	3.3 μg/kg
Heptachlor Enovido(a)	8081	3.3 μg/kg
Heptachlor Epoxide(a)	8081	3.3 μg/kg
Heptachlor Epoxide(b)	8081	3.3 μg/kg
Isodrin	8081	3.3 μg/kg
Methoxychlor	8081	3.3 μg/kg
Mirex	8081	2 μg/kg
Toxaphene Aroclors		
	8082	33 μg/kg
PCB-1016/1242	8082	33 μg/kg
PCB-1221	8082	33 μg/kg
PCB-1232	8082	33 μg/kg
PCB-1248	8082	33 μg/kg
PCB-1254	8082	33 μg/kg
PCB-1260		
Chlorinated Herbicides 2-(2,4,5-Trichlorophenoxy) propionic acid (2,4	1,5 - 8151	10 μg/kg
TP) (Silvex) 2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	8151	10 μg/kg

Table 5.1. (cont.).

	EPA	Lab Reporting
Chlorinated Harbistal	Method	Limit for sediment
Chlorinated Herbicides (cont.)		
2,4-Dichlorophenoxy acetic acid (2,4 - D) 3,5-DCBA	8151	10 μ g/k g
	8151	10 μg/kg
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB) 4 - Nitrophenol	8151	10 μg/kg
Acifluorfen	8151	10 μg/kg
Bentazon	8151	10 μg/kg
Chloramben	8151	10 μ g/k g
Dacthal	8151	10 μg/kg
Dalapon	8151	10 μg/kg
Dicamba	8151	10 μ g/k g
	8151	10 μg/kg
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	8151	10 μg/kg
Dinoseb	8151	10 μg/kg
MCPA	8151	100 μg/kg
MCPP	8151	100 μg/kg
Pentachlorophenol	8151	10 μg/kg
Picloram	8151	10 μ g/kg
PAHs		το μαγκα
1 - Methylnaphthalene	8310	17 ug/kg
2 - Methylnaphthalene	8310	17 μg/kg
Acenaphthene	8310	17 μg/kg
Acenaphthylene	8310	17 μg/kg
Anthracene	8310	3.3 μg/kg
Benzo(a)anthracene	8310	17 μg/kg
Benzo(a)pyrene	8310	1.7 μg/kg
Benzo(b)fluoranthene	8310	2.0 μg/kg
Benzo(g,h,i)perylene	8310	3.0 μg/kg
Benzo(k)fluoranthene	8310	3.3 μg/kg
Chrysene	8310	2.0 μg/kg
Dibenzo(a,h)anthracene	8310	1.7 μg/kg
luoranthene	8310	3.3 μg/kg
luorene	8310	3.3 μg/kg
ndeno(1,2,3-cd)pyrene	8310	3.3 μg/kg
laphthalene	8310	1.7 μg/kg
henanthrene	8310	17 μg/kg
yrene	8310	1.7 μg/kg 1.7 μg/kg

Table 5.1. (cont.).

	EPA	Lab Reporting
	Method	Limit for sediment
Metals		
luminum	200.7	20 mg/kg
ntimony	204.2	2 mg/kg
Arsenic	6010	0.5 mg/kg
Barium	6010	20 mg/kg
Beryllium	6010	1 mg/kg
Cadmium	6010	1 mg/kg
Calcium	6010	25 mg/kg
Chromium (total)	6010	1 mg/kg
Cobalt	6010	5 mg/kg
	6010/7210	5 mg/kg
Copper	6010	10 mg/kg
ron	6010	1 mg/kg
ead	6010	25 mg/kg
Magnesium	6010	1 mg/kg
Manganese	7471	0.01 mg/kg
Mercury	6010/7520	5 mg/kg
Nickel	6010/7610	25 mg/kg
Potassium	6010	2 mg/kg
Selenium	6010	2 mg/kg
Silver	7770	25 mg/kg
Sodium	6010	1 mg/kg
Thallium	6010	1 mg/kg
Vanadium	6010/7950	5 mg/kg
Zinc		
Other Parameters	9045	NA
рН	SM2540G	NA
Percent Solids	Miller	NA
Resistivity	MSA Part2,	NA
Total Organic Carbon	29-3	
Texture (No. 4,10,40,60,100,200)	Sieve Method	NA

Table 5.2. Name, number of hits, and frequency of analytes occurring above detection limits occurring in sediment samples. For screening table values, the four parameters with only one hit out of 46 samples were considered as non-detect by the NASA/KSC Remediation Program Team.

Parameter	Number of Hits	Frequency %
Organochlorine pesticides		1 requericy %
4,4° - DDD	0	
4,4' - DDE	0	0
4,4' - DDT	0	0
Aldrin	0	0
Alpha - BHC	0	0
Beta - BHC	0	0
Chlordane (alpha)	0	0
Chlordane (gamma)	0	0
Chlordane (Total)	0	0
Delta – BHC	0	0
Dieldrin	1	0
Endosulfan I	0	2.17
Endosulfan II (beta)	0	0
Endosulfan Sulfate	0	0
Endrin	0	0
Endrin Aldehyde	0	0
Endrin Ketone		0
Gamma - BHC (Lindane)	0	0
Heptachlor	0	0
Heptachlor Epoxide(a)	0	0
Heptachlor Epoxide(b)	0	0
Isodrin	0	0
Methoxychlor	0	0
Mirex	0	0
Toxaphene	0	0
Arociors	0	0
PCB-1016/1242		
PCB-1221	0	0
PCB-1232	0	00
PCB-1248	0	0
PCB-1254	0	0
CB-1260	0	0
	0	0

Table 5.2. (cont.).

Parameter	Number of Hits	Frequency %
Chlorinated Herbicides		
2-(2,4,5-	0	0
Z-(2,4,5- Trichlorophenoxy)propionic acid		
(2,4,5 - TP) (Silvex)		
2,4,5-Trichlorophenoxy acetic	2	4.35
acid (2,4,5 - T)		
2,4-Dichlorophenoxy acetic acid	2	4.35
(2,4 - D)		
3.5-DCBA	0	0
4-(2,4-Dichlorophenoxy)butyric	0	0
acid (2,4 - DB)		
4 - Nitrophenol	0	0
Acifluorfen	0	0
Bentazon	0	0
Chloramben	0	0
Dacthal	0	0
Dalapon	0	0
Dicamba	0	0
Dichloroprop [2-(2,4-	0	0
Dichlorophenoxy)proponic acid]		
Dinoseb	0	0
MCPA	0	0
MCPP	0	0
Pentachlorophenol	0	0
Picloram	0	0
Polyaromatic Hydrocarbons		
1 - Methylnaphthalene	0	0
2 - Methylnaphthalene	1	2.17
Acenaphthene	0	0
Acenaphthylene	0	0
Anthracene	0	0
Benzo(a)anthracene	3	6.52
Benzo(a)pyrene	4	8.70
Benzo(b)fluoranthene	5	10.87
Benzo(g,h,i)perylene	1	2.17
Benzo(k)fluoranthene	4	8.70
Chrysene	6	13.04
Dibenzo(a,h)anthracene	0	0
Fluoranthene	3	6.52
Fluorene	0	0

Table 5.2. (cont.).

Parameter	Number of Hits	English
Polyaromatic Hydrocarbons (cont.)	The state of this	Frequency %
Indeno(1,2,3-cd)pyrene	1	
Naphthalene	4	2.17
Phenanthrene	3	8.70
Pyrene	2	6.52
Metals		4.35
Aluminum	46	
Antimony	46	100
Arsenic (as carcinogen)		0
Barium	20	43.48
Beryllium	0	0
Cadmium	0	0
Calcium	0	0
Chromium (total)	45	97.83
Cobalt	37	80.43
Copper	0	0
Iron	6	13.04
Lead	46	100
Magnesium	37	80.43
Manganese	46	100.00
Mercury (inorganic)	44	95.65
Nickel	5	10.87
Potassium	0	0
Selenium	42	91.30
Silver	0	0
Sodium	5	10.87
hallium	45	97.83
/anadium	3	6.52
Zinc	37	80.43
	10	21.74
Other Parameters Percent Solids		
	46	100
otal Organic Carbon	38	82.61

Table 5.3 Parameters that were below detection limits (non-detect) in all sediment samples.

Organochlorine pesticides (8081)
4,4' - DDD
4,4' - DDE
4,4' - DDT
Aldrin
Alpha - BHC
Beta - BHC
Chlordane (Total)
Chlordane, alpha
Chlordane,gamma
Delta - BHC
Endosulfan I
Endosulfan II (beta)
Endosulfan Sulfate
Endrin
Endrin Aldehyde
Endrin Ketone
Gamma - BHC (Lindane)
Heptachlor
Heptachlor Epoxide(a)
Heptachlor Epoxide(b)
Isodrin
Methoxychlor
Mirex
Toxaphene
Aroclors (8082)
PCB-1016/1242
PCB-1221
PCB-1232
PCB-1248
PCB-1254
PCB-1260

Chlorinated Herbicides (8151)
2-(2,4,5-Trichlorophenoxy)propionic
acid (2,4,5 - TP) (Silvex)
3 5-DCBA
4-(2,4-Dichlorophenoxy)butyrica acid
(2,4 - DB)
4 - Nitrophenol
Acifluorfen
Bentazon
Chloramben
Dacthal
Dalapon
Dicamba
Dichloroprop [2-(2,4-
Dichlorophenoxy)proponic acid]
Dinoseb
MCPA
MCPP
Pentachlorophenol
Picloram
Polyaromatic Hydrocarbons (8310)
1 - Methylnaphthalene
Acenaphthene
Acenaphthylene
Anthracene
Dibenzo(a,h)anthracene
Fluorene
Metals
Antimony
Barium
Beryllium
Cadmium
Cobalt
Nickel
Selenium

DRAFT

Table 5.4 Parameters that were determined by the NASA/KSC Remediation team to be below detection limits (non-detect) for all sediment samples based on low frequency of detection.

Organochio	rine pesticides (8081)
Dieldrin	
Polyaromatic	Hydrocarbons (8310)
2-Methylnaptha	lene
Benzo(g,h,l)per	ylene
Indeno(1,2,3-cd)pyrene
	/

SECTION 6 Literature Cited

- Ashton, F.M. and A.S. Crafts. 1981. Mode and action of herbicides. John Wiley & Sons, New York. 525 p.
- Bader, S.F. and R.W. Parkinson. 1990. Holocene evolution of Indian River Lagoon in Central Brevard County, Florida. Florida Scientist 53:204-215.
- Baldwin, R., C.L. Burch, R.B. Hinton, H.H. Huckle, P. Nichols, F.C. Watts, and J.A. Wolfe. 1980. Soil survey of Volusia County, Florida. USDA Soil Conservation Service, Washington, D.C. 207 p. and maps.
- Ballentine, D.C., S.A. Macko, V.C. Turekian, W.P. Gilhooly, and B. Martincigh. 1996. Chemical and isotopic characterization of aerosols collected during sugar cane burning in South Africa. p. 460-465. In: J.S. Levine (ed.). Biomass Buring and Global Change. Volume I. Remote Sensing, Modeling and Inventory Development, and Biomass Burining in Africa. The MIT Press, Cambridge, Massachusetts.
- Barbash, J.E. and E.A. Resek. 1996. Pesticides in groundwater: Distribution, trends, and governing factors. Ann Arbor Press, Chelsea, Michigan. 588 p.
- Barbash, J.E., G.P. Thelin, D.W. Kolpin, and R.J. Gilliom. 1999. Distribution of major herbicides in groundwater of the United States. U.S. Geological Survey Water-Resources Investigations Report 98-4245. Sacramento, California. 57p.
- Behymer, T.D. and R.A. Hites. 1985. Photolysis of polycyclic aromatic hydrocarbons adsorbed on simulated atmospheric particles. Environmental Science & Technology 19:1004-1006.
- Bense, J.A. 1994. Archaeology of the Southeastern United States. Academic Press, New York. 388p.
- Birkeland, P.W. 1999. Soils and geomorphology. 3rd edition. Oxford University Press, New York. 430 p.
- Blake, G.R. 1965. Bulk density. p. 374-390. In: C.A. Black (ed.). Methods of Soil Analysis, Part 1, Physical and Mineralogical Properties, including Statistics of Measurement and Sampling. Agronomy Monograph 9. American Society of Agronomy, Madison, Wisconsin.
- Blus, L.J., A.A. Belisle, and R.M. Prouty. 1974. Relations of the Brown Pelican to certain environmental pollutants. Pesticide Monitoring Journal 7:181-194.

- Bodzek, D., K. Luks-Betlej, and B. Janoszka. Occurrence of PAHs in various elements of environment in Zabrze (Upper Silesia, Poland). Water, Air, and Soil Pollution 103:91-100.
- Bowen, D.Q. 1978. Quaternary geology: a stratigraphic framework for multidisciplinary work. Pergamon Press. New York. 221p.
- Bower, C.A. and L.V. Wilcox. 1965. Soluble salts. p. 933-951. <u>In</u>: C.A. Black (ed.). Methods of Soil Analysis, Part 2, Chemical and Microbiological Properties. Agronomy Monograph 9. American Society of Agronomy, Madison, Wisconsin.
- Brady, N.C. 1974. The nature and properties of soils. 8th edition. MacMillan Publishing Company, New York. 639 p.
- Breckenridge, R.P. and A.B. Crockett. 1995. Determination of background concentrations of inorganics in soils and sediments at hazardous waste sites. EPA/540/S-96/500. Technology Innovation Office, Office of Solid Waste and Emergency Response, U. S. Environmental Protection Agency, Washington, D.C.
- Brooks, H.K. 1972. Geology of Cape Canaveral. p. 35-44. <u>In</u>: T.E. Garner (ed.). Space-age geology. Southeastern Geology Society. Tallahassee, Florida.
- Brooks, H.K. 1981a. Geologic map of Florida. Florida Cooperative Extension Service, Institute of Food and Agricultural Sciences, University of Florida. Gainesville.
- Brown, D.W., W.E. Kenner, J.W. Crooks, and J.B. Foster. 1962. Water resources of Brevard County, Florida. Report of Investigations No. 28. Florida Geological Survey. Tallahassee. 104p.
- Burkart, M.R. and D.W. Kolpin. 1993. Hydrologic and land-use factors associated with herbicides and nitrate in near-surface aquifers. Journal of Environmental Quality 22:646-656.
- Chaki, S. 1974. Study of beach ridges, Cape Canaveral, Florida. Coastal Research Notes 4(2):5-7.
- Chapman, H.D. 1965. Cation-exchange capacity. p. 891-901. <u>In</u>: C.A. Black (ed.). Methods of Soil Analysis, Part 2, Chemical and Microbiological Properties. Agronomy Monograph 9. American Society of Agronomy, Madison, Wisconsin.
- Clapp, D. 1987. Chapter 1. Overview of physiographic and surface drainage features. p. 1-1-1-27. In: J.S. Steward and J.A. VanArman (eds.). Indian

- River Lagoon Joint Reconnaissance Report. St. Johns River Water Management District and South Florida Water Management District. Palatka and West Palm Beach, Florida.
- Clements, B.W., Jr., and A.J. Rogers. 1964. Studies of impounding for the control of salt-marsh mosquitoes in Florida, 1958-1963. Mosquito News 24:265-276.
- Clescerl, L.S., A.E. Greeberg, and A.D. Eaton. (eds.). 1998. Standard methods for the examination of water and wastewater, 20th edition. American Public Health Association, American Water Works Association, and Water Environmental Federation, Washington, D.C.
- Cooke, C.W. 1945. Geology of Florida. Florida Geological Survey Bulletin 29:1-339.
- Davis, R.A., Jr. 1997. Geology of the Florida coast. p. 155-168. <u>In</u>: A.F. Randazzo and D.S. Jones (eds.). The Geology of Florida. University Press of Florida. Gainesville.
- Davison, K.L. and S.P. Bratton. 1986. The vegetation history of Canaveral National Seashore, Florida. National Park Service Cooperative Park Studies Unit Technical Report 22. Institute of Ecology, University of Georgia, Athens. 75pp.
- Day, P.R. 1965. Particle fractionation and particle-size analysis. p. 545-567. .

 <u>In:</u> C.A. Black (ed.). Methods of Soil Analysis, Part 1, Physical and Mineralogical Properties, including Statistics of Measurement and Sampling. Agronomy Monograph 9. American Society of Agronomy, Madison, Wisconsin.
- Delcourt, P.A. and H.R. Delcourt. 1981. Vegetation maps for eastern North America: 40,000 Yr. B.P. to the present. p.123-165. In: R.C. Romans (ed.). Geobotany II. Plenum Publishing Corporation.
- Edward E. Clark Engineers-Scientists, Inc. 1985. KSC subsurface hydrology and groundwater survey. Volume I. The site specific program. KSC-DF-3081. Edward E. Clark Engineers-Scientists, Inc., Miami, Florida. Contract No. NAS10-11161.
- Edward E. Clark Engineers-Scientists, Inc. 1987a. KSC subsurface hydrology and groundwater survey. Volume II. Continuation of the site specific program. KSC-DF-3081. Edward E. Clark Engineers-Scientists, Inc., Kennedy Space Center, Florida. Contract No. NAS10-11293.
- Edward E. Clark Engineers-Scientists, Inc. 1987b. KSC subsurface hydrology and groundwater survey. Volume III. CERCLA remedial investigations. KSC-

- DF-3081. Edward E. Clark Engineers-Scientists, Inc. Kennedy Space Center, Florida. Contract No. NAS10-11293.
- Edward E. Clark Engineers-Scientists, Inc. 1987c. KSC subsurface hydrology and groundwater survey. Volume IV. The areawide groundwater survey program. KSC-DF-3081. Edward E. Clark Engineers-Scientists, Inc., Kennedy Space Center, Florida. Contract No. NAS10-11293.
- Evangelou, V.P. 1988. Environmental soil and water chemistry: principles and applications. John Wiley & Sons. New York. 564 p.
- Field, M.E. and D.B. Duane. 1974. Geomorphology and sediments of the Inner Continental Shelf, Cape Canaveral, Florida. Technical Memorandum No. 42. U.S. Army Corps of Engineers Coastal Engineering Research Center, Fort Belvoir, Virginia. 88p.
- Folk, R.L. and Ward, W.C. 1957. Brazos River bar: a study in the significance of grain size parameters. Journal of Sedimentary Petrology. 27:3-26.
- Gilmore, R. G. 1977. Fishes of the Indian River Lagoon and adjacent waters, Florida. Bulletin of the Florida State Museum 22:101-147.
- Haines, B.L. and E.L. Dunn. 1985. Coastal marshes. p. 323-347. In: B.F. Chabot and H.A. Mooney (eds.). Physiological Ecology of North American Plant Communities. Chapman and Hall, New York.
- Hargrave, B.T. 1973. Coupling carbon flow through some pelagic and benthic communities. Journal of Fisheries Research Board of Canada 30:1317-1326.
- Head, C.M. 1981. Geological Perspective of Florida. Unpublished Report.
- Healy, H.G. 1975. Terraces and shorelines of Florida. USDI Geological Survey and Florida Department of Natural Resources Bureau of Geology. Tallahassee, Florida.
- Huckle, H.F., H.D. Dollar, and R.F. Pendleton. 1974. Soil survey of Brevard County, Florida. U.S. Dept. of Agricultural, Soil Conser. Ser., Washington, D.C., USA. 231 p.
- Jenny, H. 1941. Factors of soil formation. McGraw-Hill. New York.
- Jenny, H. 1980. The soil resource origin and behavior. Springer-Verlag. New York. 377b.
- Jones, K.C., G. Grimmer, J. Jacob, and A.E. Johnston. 1989a. Changes in the polynuclear aromatic hydrocarbon content of wheat grain and pasture

- grassland over the last century from one site in the U.K. The Science of the Total Environment 78:117-130.
- Jones, K.C., J.A. Stratford, P. Tidridge, and K.S. Waterhouse. 1989b.

 Polynuclear aromatic hydrocarbons in an agricultural soil: long-term changes in profile distribution. Environmental Pollution 56: 337-351.
- Jones, K.C., J.A. Stratford, K.S. Waterhouse, E.T. Furlong, W. Giger, R.A. Hites, C. Schaffner, and A.E. Johnston. 1989c. Increases in polynuclear aromatic hydrocarbon content of an agricultural soil over the last century. Environmental Science & Technology 23:95-101.
- Jones, K.C., J.A. Stratford, K.S. Waterhouse, and N.B. Vogt. 1989d. Organic contaminants in Welsh soils: polynuclear aromatic hydrocarbons. Environmental Science & Technology 23:540-550.
- Kabata-Pendias, A. and H. Pendias. 1984. Trace elements in soils and plants. CRC Press, Inc. Boca Raton, Florida. 315 p.
- Kolpin, D.W., J.E. Barbash, and R.J. Gilliom. 1998. Occurrence of pesticides in shallow groundwater of the United States: initial results from the National Water-Quality Assessment Program. Environmental Science & Technology 32:558-566.
- Kolpin, D.W., D.A. Goolsby, and E.M. Thurman. 1995. Pesticides in near-surface aquifers: an assessment using highly sensitive analytical methods and tritium. Journal of Environmental Quality 24:1125-1132.
- Kuhnt, G. 1995. Long-term fate of pesticides in soil. p. 123-133. In: W. Salomons and W. M. Stigliani (eds.). Biogeodynamics of pollutants in soils and sediments: Risk assessment of delayed and non-linear responses. Springer-Verlag, New York.
- Mailander, J.L. 1990. Climate of the Kennedy Space Center and Vicinity. NASA Technical Memorandum 103498. Kennedy Space Center, Florida. 62 p.
- Madsen, B.C.. 1980. A continuation of base-line studies for environmentally monitoring Space Transportation Systems (STS) at John F. Kennedy Space Center. Volume II. Chemical studies of rainfall and soil analysis. NASA Contract No. NAS 10-8986. NASA Contract Report 163122.
- Marschner, H. 1986. Mineral nutrition of higher plants. Academic Press, New york. 674 p.

- Masclet, P., G. Mouvier, and K. Nikolaou. 1986. Relative decay index and sources of polycyclic aromatic hydrocarbons. Atmospheric Environment 20:439-446.
- McBride, M.B. 1994. Environmental chemistry of soils. Oxford University Press, New York. 406 p.
- McLane, M. 1995. Sedimentology. Oxford University Press. New York. 423 p.
- Mehta, A.J. and H.K. Brooks. 1973. Mosquito Logoon barrier beach study. Shore and Beach 41(2):26-34.
- Miller, J.A. 1997. Hydrogeology of Florida. p. 69-88. In: A.F. Randazzo and D.S. Jones (eds.). The Geology of Florida. University Press of Florida. Gainesville.
- Milliman, J.D. 1972. Atlantic Continental Shelf and Slope of the United States petrology of the sand fraction of sediments, Northern New Jersey to Southern Florida. U.S. Geological Survey Professional Paper 529-J. Washington, D.C. 40 p. and maps.
- Mitsch, W.J. and J.G. Gosselink. 1986. Wetlands. Van Nostrand Reinhold, New York. 539 p.
- NASA. 1996. Sampling and Analysis Plan for RCRA Facility Investigations at Kennedy Space Center, Florida, and Cape Canaveral Air Station, Florida, Volume IV. Remediation Program Office, NASA, Kennedy Space Center, Florida.
- Nelson, D.W. and L.E. Sommers. 1982. Total carbon, organic carbon, and organic matter. p. 539-579. In: A.L. Page, R.H. Miller, and D.R. Keeney (eds.). Methods of Soil Analysis, Part 2. Chemical and Microbiological Properties. Agronomy Monograph No. 9 (2nd edition). American Society of Agronomy and Soil Science Society of America. Madison, Wisconsin.
- Nesbitt, S.A., P.E. Cowan, P.W. Rankin, N.P. Thompson, and L.E. Williams, Jr. 1981. Chlorinated hydrocarbon residues in Florida Brown Pelicans. Colonial Waterbirds 4:77-84.
- Nixon, S.W. 1981. Remineralization and nutrient cycling in coastal marine ecosystems, p. 111-138. In: B.J. Neilson and L.E. Cronin (eds.). Estuaries and Nutrients. Humana Press, Clifton, New Jersey.
- Page, G.W. 1981. Comparison of groundwater and surface water for patterns and levels of contamination by toxic substances. Environmental Science & Technology 15:1475-1481

- Parkinson, R.W., T.C. Wang, J.R. White, J.R. David, and M.E. Hoffman. 1993. Distribution and migration of pesticide residues in mosquito control impoundments St. Lucie County, Florida, USA. Environmental Geology 22:26-32.
- Paton, T.R., G.S. Humphreys, and P.B. Mitchell. 1995. Soils a new global view. Yale University Press, New Haven, Connecticut. 213 p.
- Paton, T.R., P.B. Mitchell, D. Adamson, R.A. Buchanan, M.D. Fox, and G. Bowman. 1976. Speed of podzolisation. Nature 260:601-602.
- Ponnamperuma, F.N. 1972. The chemistry of submerged soils. Advances in Agronomy 24:29-96.
- Pothuluri, J.V. and C.E. Cerniglia. 1998. Current aspects on polycyclic aromatic hydrocarbon biodegradation processes. p. 461-520. <a href="mailto:ln:specific substantial-sub
- Provost, M.W. 1959. Impounding salt marshes for mosquito control and its effects on bird life. Florida Naturalist 32:163-170.
- Provost, M.W. 1977. Source reduction in salt-marsh mosquito control: past and future. Mosquito News 37:689-698
- Randazzo, A.F. 1997. The sedimentary platform of Florida: Mesozoic to Cenozoic. p.39-56. <u>In</u>: A.F. Randazzo and D.S. Jones (eds.). The Geology of Florida. University Press of Florida. Gainesville.
- Ranwell, D.S. 1972. Ecology of salt marshes and sand dunes. Chapman and Hall, London. 258 p.
- Ray, R.W. and G.P. Quinn. 1989. Comparisons of treatments after an analysis of variance in ecology. Ecological Monographs 59:433-463.
- Rey, J.R. and T. Kain. 1989. A guide to the salt marsh impoundments of Florida. Florida Medical Entomology Laboratory. Vero Beach. 447p.
- Rutledge, A.T. 1987. Effects of land use on groundwater quality in Central Florida—Preliminary results: U.S. Geological Survey toxic waste—groundwater contamination program. U.S. Geological Survey Water-Resources Investigations Report 86-4163. Tallahassee, Florida. 49p.
- Schmalzer, P.A. and C.R. Hinkle. 1987. Effects of fire on composition, biomass and nutrients in oak scrub vegetation on John F. Kennedy Space Center, Florida. NASA Technical Memorandum 100305. John F. Kennedy Space Center, Florida. 146p.

- Schmalzer, P.A., and C.R. Hinkle. 1990. Geology, geohydrology and soils of Kennedy Space Center: a review. NASA Technical Memorandum 100313, Kennedy Space Center, Florida. 46p.
- Schmalzer, P.A. and C.R. Hinkle. 1992a. Soil dynamics following fire in <u>Juncus</u> and <u>Spartina</u> marshes. Wetlands 12:8-21.
- Schmalzer, P.A. and C.R. Hinkle. 1992b. Species composition and structure of oak-saw palmetto scrub vegetation. Castanea 57:220-251.
- Schmalzer, P.A. and C.R. Hinkle. 1996. Biomass and nutrients in aboveground vegetation and soils of Florida oak-saw palmetto scrub. Castanea 61:168-193.
- Schropp, S.J., F. G. Lewis, H.L. Windom, J.D. Ryan, F.D. Calder, and L. C. Burney. 1990. Interpretaion of metal concentrations in estuarine sediments of Florida using aluminum as a reference element. Estuaries 13:227-235.
- Scott, T.M. 1997. Miocene to Holocene history of Florida. p. 57-68. <u>In</u>: A.F. Randazzo and D.S. Jones (eds.). The Geology of Florida. University Press of Florida. Gainesville.
- Senesi, N. and Y. Chen. 1989. Interactions of toxic organic chemicals with humic substances. p. 37-90. ln: Z. Gerstl, Y. Chen, U. Mingelgrin, and B. Yaron (eds.). Toxic Organic Chemicals in Porous Media. Springer-Verlag, New York.
- Shacklette, H.T. and J.G. Boerngen. 1984. Elemental concentrations in soils and other surficial material of the conterminous United States. U.S. Geological Survey Professional Paper 1270. Washinton, D.C. 105 p.
- Simoneit, B.R.T., M. R. Abas, G.R. Cass, W.F. Rogge, M.A. Mazurek, L.J. Standley, and L.M. Hindemann. Natural organic compounds as tracers for biomass combustion in aerosols. p. 509-517. In: J.S. Levine (ed.). Biomass Buring and Global Change. Volume I. Remote Sensing, Modeling and Inventory Development, and Biomass Burining in Africa. The MIT Press, Cambridge, Massachusetts.
- SPSS Inc. 1999. SPSS Base 9.0 User's Guide. SPSS Inc. Chicago, Illinois. 740 p.
- Standley, L.J. and B.R.T. Simoneit. 1987. Characterization of extractable plant wax, resin, and thermally matured components in smoke particles from prescribed burns. Environmental Science & Technology 21:163-169.
- Stevenson, F.J. 1986. Cycles of soil: carbon, nitrogen, phosphorus, sulfur, micronutrients. John Wiley & Sons, New York. 380 p.

- Stout, I.J. 1980. A continuation of base-line studies for environmentally monitoring Space Transportation Systems (STS) at John F. Kennedy Space Center. Volume I. Terrestrial Community Ecology. NASA Contract No. NAS 10-8986. NASA Contract Report 163122.
- Suess, M.J. 1976. The environmental load and cycle of polycyclic aromatic hydrocarbons. The Science of the Total Environment 6:239-250.
- Sykes, P.W., Jr. 1980. Decline and disappearance of the Dusky Seaside Sparrow from Merritt Island, Florida. American Birds 34:728-737.
- Thompson, N.P., P.W. Rankin, P.E. Cowan, L.E. Williams, Jr., and S.A. Nesbitt. 1977. Chlorinated hydrocarbon residues in the diet and eggs of the Florida Brown Pelican. Bulletin of Environmental Contamination and Toxicology 18:331-339.
- Trimble Naviagation Limited. 1994. Pro XL System Operations Manual. Sunnyvale, California.
- Trost, C.H. 1968. Dusky Seaside Sparrow. p. 849-859. <u>In</u>: A.C. Bent and O.L. Austin, Jr. (eds.). Life histories of North American cardinals, grosbeaks, buntings, towhees, finches, sparrows, and allies. United States National Museum Bulletin 237.
- Upchurch, S.B. and A.F. Randazzo. 1997. Environmental geology of Florida. p. 217-249. <u>In</u>: A.F. Randazzo and D.S. Jones (eds.). The Geology of Florida. University Press of Florida. Gainesville.
- O'Brien and Gere Engineers, Inc. 1996. Remediation Investigation/Feasiblility Study for 45th Space Wing Facilities at Cape Canaveral Air Station and Patrick Air Force Base, Baseline Data Development Program, Volume 2B. Air Force Space Command, 45th Space Wing, 45th CES/CEVR, Patrick Air Force Base, Florida. Contract No. F33615-90-D-4012, Delivery Order No. 004.
- United States Environmental Protection Agency (USEPA). 1983. Methods for chemical analysis of water and wastes. EPA-600/4-79-020. Environmental Monitoring and Support Laboratory, Cincinnati, Ohio.
- United States Environmental Protection Agency (USEPA). 1996. Test methods for evaluating solid waste. SW-846. Office of Solid Waste, Washington, D.C.
- United States Environmental Protection Agency (USEPA). 1998. Statistical tests for background comparison at hazardous waste sites, interim draft. Supplemental guidance to RAGS: Region 4 Bulletins-Addition #1. Waste Management Division, Office of Technical Services, Atlanta, Georgia.

- Universal Engineering Sciences, Inc. 1998. Summary of field work activities Basic Order Agreement NAS10-12206(B)/DO-5 Installation of 52 permanent Monitoring Wells Kennedy Space Center, Florida. Universal Engineering Sciences, Inc. Orlando, Florida.
- Vogt, N.B., F. Brakstad, K. Thrane, S. Nordenson, J. Krane, E. Aamot, K. Kolset, K. Esbensen, and E. Steinnes. 1987. Polycyclic aromatic hydrocarbons in soil and air: statistical analysis and classification by the SIMCA method. Environmental Science & Technology 21:35-44.
- Wentworth, C.K. 1922, A scale of grade and class terms for clastic sediments. Journal of Geology 30:377-92.
- Wang, T.C., M.E. Hoffman, J. David, and R. Parkinson. 1992. Chlorinated pesticide residue occurrence and distribution in mosquito control impoundments along the Florida Indian River Lagoon. Bulletin of Environmental Contamination and Toxicology 49:217-223.
- White, W.A. 1958. Some geomorphic features of central peninsular Florida. Geological Bulletin No. 41. Florida Geological Survey. Tallahassee. 92p.
- White, W.A. 1970. The geomorphology of the Florida peninsula. Geological Bulletin No. 51. Bureau of Geology, Florida Department of Natural Resources. Tallahassee. 164p.
- Wickstrom, K. and K. Tolonen. 1987. The history of airborne polycyclic aromatic hydrocarbons (PAH) and perylene as recorded in dated lake sediments. Water, Air, and Soil Pollution 32:155-175.
- Woodward-Clyde Consultants. 1994. Biological Resources of the Indian River Lagoon. Project Number 92F274C. Indian River Lagoon National Estuary Program. Melbourne, Florida.
- Woodward-Clyde Consultants. 1994. Status and Trends Summary of the Indian River Lagoon. Project Number 92F274C. Indian River Lagoon National Estuary Program. Melbourne, Florida.

Appendix A

KSC Background Screening Tables for Soils, Groundwater, Surface Water, and Sediments of Kennedy Space Center

How to use the KSC Background Screening Tables provided in this Appendix.

Soil:

The KSC Background soil screening table has 11 classes. The NASA/KSC Remediation Team decision for screening purposes only allows inorganic data to be used as a screening tool. Organic data should be discussed in the uncertainity section of reports. Combined values for pesticides (as noted on table) can be used for the following classes: saltwater wetland, citrus hammock, citrus scrub, and disturbed.

Steps for screening your data against the KSC Background soil screening table (Appendix A, Table A-1) as determined by the NASA/KSC Remediation Team:

- 1. Screen site data against the combined mean.
- If the site value is higher than the combined mean, go to the soil class that
 corresponds to site data sampling locations. Figure 2.2 illustrates KSC
 Background classification information. Individual site maps can be obtained,
 upon reqest, from NASA support contractor when additional site details are
 needed.

Groundwater:

The NASA/KSC Remediation Team decision on groundwater was to separate the groundwater according to Rule 62-520.410 F.A.C. Class G-II and Class G-III. The NASA/KSC Remediation Team decision for screening purposes only allows inorganic data to be used as a screening tool. Organic data should be discussed in the uncertainity section of reports

Steps for screening your data against the KSC Background groundwater screening table (AppenidxA, Table A-2) as determined by the NASA/KSC Remediation Team:

- 1. Collect total dissolved solid data at all groundwater sampling locations.
- 2. Determine proper groundwater class for site data based on total dissolved solid data.
 - Class G-II defined as total dissolved solids less than 10,000 mg/L. Class G-III defined as total dissolved solids greater than 10,000 mg/L.

Surface Water:

The NASA/KSC Remediation Team decision for surface water resulted in eight classes. The NASA/KSC Remediation Team decision for screening purposes

only allows inorganic data to be used as a screening tool. Organic data should be discussed in the uncertainity section of reports

Steps for screening your data against the KSC Background surface water screening table (Appendix A, Table A-3) as determined by the NASA/KSC Remediation Team:

- 1. Collect salinity measurements at all surface water sampling locations.
- 2. Compare site sample data to the surface water combined values based on the appropriate salinity ranges. If the site value is higher than the KSC Background combined salinity value, go to the appropriate surface water basin class for site data in the KSC Background surface water table and compare site value to that. Refer to Figure 4.1, for basin classes.

Sediment:

The NASA/KSC Remediation Team decision was to have only one class for sediment (AppendixA, Table A-4). The NASA/KSC Remediation Team decision for screening purposes only allows inorganic data to be used as a screening tool. Organic data should be discussed in the uncertainity section of reports

Table A-1. KSC Background screening tables for soils. Only metals values can be used as a screening tool. Organic values may not be used as a screening tool. Organics should be discussed in the uncertainty section of reports. * Combined mean for these pesticides can only be used for the following classes: saltwater wetland, citrus hammock, citrus scrub, and disturbed. All other classes are non detect for these pesticides. ** Team consensus determined that these parameters are non-detect based on frequency of detection. ND = non-detect

		₹	Coastal	Acid	Conting	Flotting						
		Sign		4: 20		latwoods	nammocks	Freshwater	Saltwater	Citrus	Citrus	Disturbed
Parameter	1			Scrub	Scrub		-	Wetland	Wettand	Scrub	Hammock	
	Sills	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Moon V 2	No.	
Organochiorine Pesticides (8081)										Nicell A &	Mean A.2	Mean X 2
4,4'-DDD	maka	99000	CN	2								
4.4'-DDE •	a dyou	3000		2	2	0.0046	Q	2	0.016	ş	9	0.0044
4 4. DOT	a a	0.010	2	QN	2	0.0043	Q	£	0.011	0.028	0011	0.045
	mg/kg	0.014	ND	2	Q	S	Q	2	0.016	0.004	0000	0.0.0
4,4'-DDT *	mg/kg	0.037	Q.	Q	2	0.013	Q	Q	0.042	0.056	0.018	210.0
Aldrin	mg/kg	2	S	2	2	Q	Ş	2	9			
Alpha - BHC	mg/kg	9	2	QN	S	2	2 2	2 4	2	2	Q	Q
Beta - BHC	mg/kg	2	2	CN	2	2 2	2 2	2	2	2	QN	Q
Chlordane (alpha)**	mg/kg	0.0048	S	2	2 2	2 2	2	2	2	9	QN	2
Chlordane (gamma) * *	movko	0.0048	2		2 4	Q.	QN	2	2	0.004	Q.	S
Chlordana (total) • •		or or or	2	Q.	Q	Q	Q	Q	Q.	0.0038	QN	0000
Dole DHC	gKg.	0.0048	2	QN	2	Q	Q	2	2	0.0087	S	
Oelia - bric	mg/kg	Q	9	QN	S	Q	2	S	Ę	2		2
Dieligran -	mg/kg	0.015	0.0036	Q	Q	QN	S	2	02.00	2 6	2	Q
Endosulfan I	mg/kg	2	2	QN	2			2 4	6/00.0	0.028	0.05	0.004
Endosulfan II (beta)	mg/kg	Ş	2	QN	2	2 2	2 9	ا ج	2	2	Q	Q
Endosulfan Sulfate	morko	2	S	S	2	2 2	2	2	QN	Q	QN	Q
Endrin	morka	Ę	2			2	2	Q	2	g	Q	QN
Endrin Aldehyde	morks	2	2 2	2 5		2	2	Q	2	9	QN	QV
Endrin Ketone * *	a de la	2000	2 2	2 4	2	2	Q	Q	S	£	£	QV
Gamma -BHC (Lindane) • •	mo/kg	77000		2	2	0.0044	Q	QN	2	Q	QN	QV QV
Hentachlor	8 2	30.0	2	2	9	Q	Q	9	0.008	S	2	0.00
	By B	3	2	QN	9	Q	2	2	2	2	S	N. C.
riepudomor Epoxide (a)	тдука	9	2	QN	QN	QV	2	S	Ş	Ş		2
hlor Epoxide (b)	mg/kg	9	Ð	QV	2	QN	5	2		2	ON.	Q
Isodrin	mg/kg	₽	Q.	QN	Ç			2	2	2	Q	Ş
						2	2	2	Q	2	QN	Q

Table A-1 (cont.). KSC Background screening tables for soils on KSC. Only metals values can be used as a screening tool. Organic values may not be used as a screening tool. Organics should be discussed in the uncertainty section of reports. * Combined mean for these pesticides can only be used for the following classes: saltwater wetland, citrus hammock, citrus scrub, and disturbed. All other classes are non detect for these pesticides. * * Team consensus determined that these parameters are non-detect based on frequency of detection. ND = non-detect

		All All	Coastal	Acid	Coquina	Flatwoods	Hammocks	Freshwater	Saltwater	Citrus	Citrus	Disturbed
		Soils		Scrub	Scrub			Wetland	Wetland	Scrub	Hammock	
Parameter	Units	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2				
Organochlorine Pesticides (8081) (cont.)												
Methoxychior	mg/kg	2	9	Q	QN	2	Ş	Q	QN	2	Q	QN
Mirex	то⁄ка	QN	QN	QN	QN	QN	Q	Q	9	ę	Q	Q
Toxaphene	mg/kg	QN	QN	Q	QN	QN	QN	9	Q	9	Q	Q
Aroclors (8082)												
PCB - 1016/1242	mg/kg	QN	QN	QN	QN	QN	QN	QN	Q	S	QN	QN
PCB - 1221	mg/kg	QN	Q	Q	QN	Q	Q	QN	Q	Ð	Q	QN
PCB - 1232	mg/kg	QN	QN	S	QΝ	Q	2	Q	QN	S	S	Q
PCB - 1248	mg/kg	QN	QN	QN	QN	QN	2	QV	QN	2	Q.	Q
PCB - 1254	mg/kg	QN	S	2	QN	Q	Q	Q	Q	Q	Q	Q
PCB - 1260	mg/kg	QN	Q	QN	QN	Q	9	Q	Q	S	Q	ND
Chlorinated Herbicides (8151)												
2-(2,4,5 - Trichlorophenoxy) propionic acid (2,4,5 - TP) (Silvex)	тд/кд	0.018	QN	QN	QN	QN	0.016	QN	QN	2	0.019	0.045
2,4,5 - Trichlorophenoxy acetic acid (2,4,5 - T)	mg/kg	QN	QN	Q	QN	QN	QN	QN	Q	Q	<u>Q</u>	QN
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	то/ка	0.016	QN	0.014	Q	2	QN	Q	0.035	Q	0.018	QN
3,5 - DCBA	mg/kg	QN	QN	QN	QN	QN	ΩN	QN	Q	Q.	QN	Q
4-(2,4 - Dichlorophenoxy) butyric acid (2,4 - DB)	mg/kg	0.016	0.012	0.013	Q	ON	QN	QN	Q	0.019	Q	0.018
4 - Nitrophenol	mg/kg	Q	Q	QN	QN	QN	QV	9	Q	Q	Q	Q
Acifluorfen	mg/kg	QN	QN	QN	QN	QN	QN	QN	QN	Q	Q	Q
Bentazon	mg/kg	QN	QN	ND	QN	QN	QN	QN	QN	Q	Q	QN
Chloramben	mg/kg	QN	QN	QN	QN	QN	QN	QN	Q	QN	QN	ND

Table A-1 (cont.). KSC Background screening tables for soils. Only metals values can be used as a screening tool. Organic values may not be used as a screening tool. Organics should be discussed in the uncertainty section of reports. *Combined mean for these pesticides can only be used for the following classes: saltwater wetland, citrus hammock, citrus scrub, and disturbed. All other classes are non detect for these perameters are non-detect based on frequency of detection. ND = non-detect

		¥	Coastal	Acid	9011100	1						
		Soils		Sor di		riatwoods	Hammocks Freshwater	Freshwater	Saltwater	Citrus	Citrus	Disturbed
Parameter	15.00	200	:	On Inc	ocuno			Wettand	Wetland	Scrub	Hammock	
	SIIIS	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Moon V 2	2007	W	
Chiorinated Herbicides (8151) (cont.)	•								7	Medii A 2	Mean X 2	Mean X 2
Dacthal	mg/kg	Ð	QN	Ş	2							
Dalapon	morka	Ş	Ş		2 5	2	Q	Q	2	QN	Q	9
Dicambia		2	2	2	ND.	QN	2	QN	QV	2	S	S
Dichloropros (2 /2 4	DYD.	2	Q	Q	ð	QN	QN	₽	Q	S	2	2 4
Dichlorophenoxy) proponic	6 6 6 6 7 7 8	2	2	Q	Q	Q	9	QN	Q	S S	2 2	2 2
acid												
Diriosed	mg/kg	2	Q	Q	Q	2	S	2	2			
MCPA	шg/kg	2	2	QN	S	Ş		2	⊋ .	ON I	Q	Q
MCPP	mg/kg	Ş	2	C	i Z		2	⊋	Q	Q	Q	ę
Pentachlorophenol	mo/ka	Q	S	2 2	۽ ڇ	2	Q	2	QN	9	ð	Q
Picloram	power	Ş	2		2	2	Q	2	Q	£	£	QN
Sum of Chlorineted	A P		2	2	Q	Q N	Q	S	Ş	QV	S	2
Herbicides	D NO	S.	0.032	0.038	0.03	0.036	0.046	0.072	0.093	0.048	0.052	200
Polyaromatic												
nydrocarbons (8310)							_		-			
I-Memyinaphmalene	mg/kg	0.028	Q	0.036	0.030	S	7600	Ç				
2-Methylnaphthalene	mg/kg	0.033	0.042	0.044	0.005		0.027	2	2	9	0.019	0.025
Acenaphthene	morka	2	Ş	Ş	Call City	2 5	2	2	QN	0.041	0.021	0.027
Acenaphthylene	mo/ka	0.0058	0.0078	2 2	9 9	2	2	2	Q	2	ð	Q
Anthracene	movice	7600	2		2	2	QN	Q	Q	0.0072	2	Q
Benzo(a)anthracene	no Ven	100		2	2	QN	QN	Q	ş	0.028	Ş.	CZ
Benzo(a)pyrene	Sugar.	0.0048	2000	Q	0.0034	0.0042	0.0038	Q.	Q	0.0039	0.0022	0100
Out (d/m)	Dug.	0.000/	0.0028	0.0036	2	9	0.0036	0.0084	S	0,000	2000	6.019
Denzo(b)nuoranthene	mg/kg	0.0076	0.0037	ð	2	₽ Q	0.0061	Ş	2 2	0.0043	0.0029	0.033
Benzo(g,h,i)perylene	mg/kg	9600.0	QN ON	QN	S	Ş	10000	2 !	2	9500.0	0.004	0.028
Benzo(k)fluoranthene	mg/kg	0.0058	0.0027	CZ	2 2	2 2	0.0030	2	2	0.0069	0.0061	0.038
Chrysene	mo⁄ka	0.0062	00000	90000		2 2	0.0055	Q.	Q	0.0056	0.0022	0.024
				3000	0.000	0.0045	0.0038	2	Q	0.0057	0.0021	0.030

Table A-1 (cont.). KSC Background screening tables for soils. Only metals values can be used as a screening tool. Organic values may not be used as a screening tool. Organics should be discussed in the uncertainty section of reports. * Combined mean for these pesticides can only be used for the following classes: saltwater wetland, citrus hammock, citrus scrub, and disturbed. All other classes are non detect for these pesticides. ** Team consensus determined that these parameters are non-detect based on frequency of detection. ND = non-detect

	_	-	1000	Pick	Cognition	Flatwoods	Hammocks	Freshwater	Saltwater	Citrus	Citrus	Disturbed
		₹ ;	Coasta	200	4::00			Wotland	Wetland	Scrib	Hammock	
		_		Scrub	ocuno			A COURTING	O A GOOM	Moon V.	Moon V 2	Moon Y 2
Parameter	Units	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X Z	Mean A 2	Mean A 2	Mean	MODILOC
Polyaromatic Hydrocarbons (8310)												
Dibenzo(a,h)anthracene	mg/kg	0.0061	Ş	S	Q.	9	2	QN	QN	QN	Q	0.012
Fluoranthene	mg/kg	0.014	S	2	0.0055	Q	0.026	ND	QN	6900:0	S	0.070
Fluorene	mg/kg	0.0068	Q	QN	QN	Q	0.022	ND	QN	Q	Q	Q
Indeno(1,2,3-cd)pyrene	mg/kg	0.0058	0.0022	QN	Q	Q	0.0038	ND	QN	0.0042	0.004	0.026
Naphthalene	тожа	0.059	0.019	0.03	Q	0.15	0.026	ND	QN	0.22	0.037	Q
Phenanthrene	mg/kg	0.0082	0.0032	0.0051	Q	2	0.011	ON	Q	0.022	Q	0.024
Pyrene	mg/kg	0.01	QN	Q.	0.0028	QN	0.043	ND	QN	0.003	Q	0.038
Sum of PAHs	mg/kg	0.24	0.14	0.20	0.15	0.34	0.25	0.69	0.22	0.40	0.14	0.45
Metals												
Aluminum	mg/kg	2889.23	565.22	405.6	2003.2	329.4	4790	628.9	9.7983	2344.3	3642.6	4982.5
Antimony	mo/ka	9	S	9	Q	2	Q	Q	QN	QN	QN	QN
Areanic	mo/ka	1.51	4.67	1.26	0.63	0.61	0.87	1.22	2.64	0.72	1.05	1,41
Rarism	mo/ka	26.14	19.94	2	QV	2	2	Q	2	QN	QN	26
Bendlim	mo/ko	2	Q	Q	2	2	2	2	2	Q	QN	QN
Cadmim	mo/ka	1.24	0.97	Ş	2	9	2	2	Q	1.28	QN	QN
Calcium	mo/kg	32932.47	100087	300.7	44113.5	766.2	22556	1532.4	23254.3	4201.7	22198.3	86107
Chromium	mg/kg	60.6	4.22	1.67	4.51	Q	6.93	2.28	9.01	24.37	23.57	7.7
Cobalt	mg/kg	Ş	S	Ş	Q.	2	₽	Ş	QN	QN	Q	<u>Q</u>
Comer	mo/ka	27.88	3.69	6.5	2	5.6	8.65	11.3	QN	107.1	108.07	5.1
lino	mo/kg	2567	2126.1	480.8	3023.6	288.3	3435	576.6	3009.5	3004.3	3969.6	3566.8
Load	moKo.	13.62	4.31	3.06	8	4.88	12.48	9.75	33.2	19.41	11.68	10.12
Magnachim	mo/ka	1098.43	740	55.91	175.23	224.62	1068.2	449.2	6281.9	465.1	754.8	577.7
Mandanase	mo⁄ka	51.5	22.7	5.05	108.37	1.87	29.5	3.73	29.5	<u>1</u> 20	117.48	15.79

Table A-1 (cont.). KSC Background screening tables for soils. Only metals values can be used as a screening tool. Organic values may not be used as a screening tool. Organics should be discussed in the uncertainty section of reports. * Combined mean for these pesticides can only be used for the following classes: saltwater wetland, citrus hammock, citrus scrub, and disturbed. All other classes are non detect for these pesticides. ** Team consensus determined that these parameters are non-detect based on frequency of detection. ND = non-detect

											1	
		₹	Coastal	Acid	Cocinina	Flohannda						
		Soils		Court		raimodas	REFINEDCKS	_	Saltwater	Citrus	Citrus	Disturbed
Dozen				20100	SCIUD			Wetland	Wettend	Consider		
rarameter	Chrits	Mean X 2	Mean X 2	Mean X 2	Magn Y 2	Moon V			Dimo	SCIUD	Hammock	
Metals (cont.)					7 V DOI!!	Z Y I IRAIN	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2
11000												
Mercury	mg/kg	0.31	Q	Q	0.018	6,00	0.00					
Nickel	marks.	9			200	0.013	8/0.0	0.026	0.036	0.024	0.032	5000
	ב ב ב	2	2	2	Q	Ę	Ş	2			200.0	0.022
Potassium	24/00	450 00	35.00				2	₹	2	2	Q	Ş
	Run.	436.06	116.78	85.23	148.05	114	229.2	228	0000			2
Selenium	mo⁄ka	2.5	Ş	Ç	9			250	8.1026	189.13	243.39	145.43
	,)	2	₹	2	3.05	0 7	Ş			
Silver	mg/kg	2.54	204	Ş	200	١		2.	2	2	2	2.4
Codium				2	6.43	2	2	2	Ş	٤	2	
	mg/kg	2918.57	922.6	25.28	348	07 38	200				2	2.28
Thallium	moden	1 27	55,			3: (5	કે	9.4.0	27542.9	34.03	76.5	6727
	A.A.);	3	2	2	2	Ş	٩				
Vanadium	mo/ka	6.37	471	1 15				2	2	 8	<u>~</u>	2
2500	,		. ,	÷.	4. 5.	1.57	10.31	3.14	147	70,7	100	
Z. I.C.	mg/kg	30.28	13.2	10.47	800						- O	6.87
					200	4.0.4	20.53	30.8	15.5	98.45	58.4	22 58
											-	2

Table A-2. KSC Background screening tables for groundwater. Groundwater has been divided based on F.A.C. groundwater classifications Class G-II (total dissolved solids <10,000) and Class G-III (total dissolved solids <10,000) and Class G-III (total dissolved solids <10,000).

		Class G-II	Class G-III
		N = 41	N = 16
Parameter	Units	Mean X 2	Mean X 2
Organochlorine pesticides (8081)			
4,4' -DDD	µg/L	9	9
4,4' -DDE	hg/L	ND	QN
4,4' -DDT	hg/L	QN	QN
Aldrin	µg/L	QN	QN
Alpha - BHC	T/Bri	QN	QN
Beta - BHC	rg/L	QN	QN
Chlordane (alpha)	hg/L	QN	QN
Chlordane (gamma)	T/Brl	ND	ND
Chlordane (total)	μg/L	ND	QN
Detta BHC	ng/L	QN	QN
Dieldrin	μg/L	QN	QN
Endosulfan I	µg/L	ND	ND
Endosulfan II (beta)	μg/L	QN	QN
Endosulfan Sulfate	µg/L	ND	QN
Endrin	μg/L	DN	QN
Endrin Aldehyde	µg/L	QN	QN
Endrin Ketone	μg/L	QN	QN
Gamma - BHC (Lindane)	µg/L	QN	QN
Heptachlor	μg/L	QN	ON
Heptachlor Epoxide (a)	ηgγ	QN	QN
Heptachlor Epoxide (b)	μg/L	QN	QN
Isodrin	hg/L	QN	ND
Methoxychlor	μg/L	QN	QN
Mirex	μg/L	ND	DN
Toxaphene	μg/L	QN	QN

Table A-2 (cont.). KSC Background screening tables for groundwater. Groundwater has been divided based on F.A.C. groundwater classifications Class G-II (total dissolved solids <10,000) and Class G-III (total dissolved solids <10,000) and Class G-III (total dissolved solids >10,000). ND = non-detect

		Class G-II	Clace G.III
		N=41	N = 16
Parameter	Units	Mean X 2	Mean X 2
Aroclors (8082)			
۱. ا	na/L	Q	S
PCB - 1221	ma/L	2	S
PCB - 1232	rg/L	₽	2
PCB - 1248	Hg/L	QN	Q
11	µg/L	QN	9
PCB - 1260	µg/L	QN	S
Chlorinated Herbicides (8151)			
2-(2,4,5 - Trichlorophenoxy) propionic acid (2,4,5 - TP) Silvex	µg/L	QN	Q
2,4,5-Triclorophenoxy acetic acid (2,4,5-T)	ng/L	2	QN
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	нgЛ	9	2
3,5 – DCBA	μg/L	Q	QN
4-(2,4-Dichlorophenoxy) butyric acid (2,4-DB)	ндЛ	Q	QN N
4 - Nitrophenol	1/011	S	S
Acifluorfen	no/	2	2
Bentazon	T/OII	2	2
Chloramben	µg/L	2	2
Dacthal	ид∕Г	9	QN
Ualapon	μg/L	QN	2
Dicambia	μg/L	QN	2
Dichlorprop [2-(2,4-Dichlorophenoxy) proponic acid]	μg/L	2	Q
Dinoseb	μg/L	9	2
MCPA	µg/L	Q	2
MCPP	µg/L	Q	2
	Hg/L	2	

Table A-2 (cont.). KSC Background screening tables for groundwater. Groundwater has been divided based on F.A.C. groundwater classifications Class G-II (total dissolved solids <10,000) and Class G-III (total dissolved solids <10,000) and Class G-III (total dissolved solids <10,000).

			111 0 00010
		Class G-II	Class G-III
		N = 41	N = 16
Parameter	Units	Mean X 2	Mean X 2
Chlorinated Herbicides (8151) (cont.)			
Pentachlorophenol	ng/L	QN	QN
Picloram	µg/L	QN	QN
Polyaromatics Hydrocarbons (8310)			
1 - Methylnapthalene	пg/L	QN	QN
2 - Methylnapthalene	ng/L	QN	QN
Acenaphthene	ng/L	QN	QN
Acenapththylene	ng/L	QN	QN
Anthracene	µg/L	QN	QN
Benzo(a)anthracene	пgЛ	0.07	0.02
Benzo(a)pyrene	hg/L	90:0	0.05
Benzo(b)fluoranthene	пgЛ	0.11	QN
Benzo(g,h,i)perylene	μg/L	QN	ON
Benzo(k)fluoranthene	ηđγ	0.05	90.0
Chrysene	μg/L	90.0	0.02
Dibenz(a,h)anthracene	hg/L	QN	Q
Fluoranthene	hg/L	0.13	QN
Fluorene	μg/L	ND	QN
Indeno(1,2,3-cd)pyrene	hg/L	0.05	90.0
Napthalene	μg⁄L	QN	2
Phenanthrene	T/Brl	QN	QN
Pyrene	µg/L	QN	QN
Metals			
Aluminum	µg/L	370	210
Antimony	μg/L	10	10
Arsenic (as carcinogen)	µg/L	30	20
Barium	µg/L	Q	160

Table A-2 (cont.). KSC Background screening tables for groundwater. Groundwater has been divided based on F.A.C. groundwater classifications Class G-II (total dissolved solids <10,000) and Class G-III (total dissolved solids <10,000) and Class G-III (total dissolved solids >10,000). ND = non-detect

		Class G-II	Class G-III
		N = 41	N = 16
Parameter	Units	Mean X 2	Mean X 2
Metals (cont.)			
Beryllium	ua/L	-	S
Cadmium	mg/L	2	2
Calcium	μg/L	337530	862500
Chromium	µg/L	10	2
Chloride	µg/L	1822240	27712500
Cobait	ид∕Г	QN	2
Copper	μg/L	09	8
Iron	hg/L	1680	4000
Lead	ng/L	7	6
Magnesium	ng/L	95290	1946250
Manganese	µg/L	8	260
Mercury	µg/L	Q	S
Nickel	µg/L	12	9
Potassium	μg/L	24930	571750
Selenium	μg/L	11.6	8
Silver	μg/L	ON	9
Sodium	μg/L	1017890	16412500
	µg/L	2	QN
vanadium	μg/L	10	QN
Zinc	ησγ	110	ND

Table A-3. KSC Background Screening Tables for surface water. Certain values have combined means based on the frequency of detection which was <20°, <10°°, or <4*** hits. ND = non-detect.

		Combine	Combined Water	Banana	Banana	Ditches	hes	Indian	Mosquito
		Salinity	Salinity	Creek	River	Salinity	Salinity	River	Lagoon
		0-5 ppt	≥ 6 ppt			0-5 ppt	> 6 ppt	Lagoon	
Parameter	Units	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2
Organochlorine Pesticides (8081)									
4,4'-DDD	µg/L	QN	ND	Q	Q	Q	QN	ND	9
4,4'-DDE	/ba/L	ND	ND	QN	2	2	Q	9	QN
4,4'-DDT	µg/L	ND	ND	QN	Q	Q.	Q	9	2
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	µg/L	ND	ND	ΟN	QN	QN	Q	QN	QN ON
Aldrin	µg/L	ON	ND	QN	QN	QN	QN	2	QN
Alpha - BHC	µg∕L	ND	ND	QN	QN	QN	QN	QN	Q.
Beta - BHC	µg∕L	QN	ND	ON	QN	QN	QN	QN	QN
Chlordane (alpha)	μg/L	QN	ND	QN	QN	QN	QN	QN	N _O
Chlordane (gamma)	µg/L	ON	ND	ON	ON	QN	ND	QN	QN
Chlordane (total)	ηg/L	QN	ND	ND	QN	ON	QN	QN	QN
Delta - BHC	/g/r	ON	ND	ND	QN	ON	QN	ND	ND
Dieldrin * * *	ug/L	QN	0.074	QN	QN	ON	0.068	0.154	QN
Endosulfan I	ηg/L	ON	ND	QN	ND	QN	ON	ND	QN
Endosulfan II (beta)	µg/L	QN	ND	QN	QN	QN	QN	ND	QN
Endosulfan Sulfate	µg∕L	ND	ND	QN	ND	QN	QN	QN	QN
Endrin	µg∕L	QN	ND	QN	QN	QN	QN	ND	QN
Endrin Aldehyde	μg/L	ND	ND	QN	ON	QN	QN	QN	QN
Endrin Ketone	µg/L	Q	ND	QN	QN	QN	QN	ND	ND
Gamma -BHC (Lindane)	µg/L	ND	ND	Q	QN	ND	ND	QN	QN
Heptachlor	μg/L	QN	ND	QN	QN	ON	ND	ND	ΩN
Heptachlor Epoxide (a)	μg/L	ND	ND	QN	QN	QN	ND	QN	QN
Heptachlor Epoxide (b)	µg/L	QN	ND	QN	QN	QN	ND	ND	QN
Isodrin	µg∕L	ND	ND	QN	ND	ON	ND	QN	QN
Methoxychlor	µg/L	Q	QN	QN	QN	QN	ND	QN	QN
Mirex	µg/L	Q	Q	QN	QN	QN	QN	QN	ND

Table A-3 (cont.). KSC Background Screening Tables for surface water. Certain values have combined means based on the frequency of detection which was <20*, <10**, or <4*** hits. ND = non-detect.

		Complete	24 14/242						
			Compiled Water	banana	Banana	Ditc	Ditches	Indian	Mosquito
		Salinity	Salinity	Creek	River	Salinity	Salinity	River	Lagoon
Daramatar	11.		> 6 ppt			0-5 ppt	> 6 ppt	Lagoon	
i alailietei	Onits	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Maan Y o
Organochlorine Pesticides (8081) (cont.)									No and No.
Toxaphene	1/0/I	QN	CN	CZ	CZ		2		
Aroclors (8082)	<u>, </u>				2	3	2	2	Q
PCB - 1016/1242	/\0/	CZ	CN	CZ			2		
PCB - 1221	707	QN	S			2 2	2 2	2	Q
PCB - 1232	10,			2 5	2	2	2	QN	2
PCB - 1948	7,67	2 5	2	QN	QN ND	Q N	ND	QN	Q.
127 1240	#9/L	Q	QN	QN	ND	Q.	QN	QN	QN
1204	1/6/	Q	ND	QN	QN	Q.	QN	QN	S
PCB - 1260	/√g/r	Q	Q	QN	QN	QN	S	S	
Chlorinated Herbicides (8151)								2	
Silvex (2,4,5 - TP) (2-(2,4,5 - Trichlorophenoxy)) propionic acid	Mg/L	QN	QN	QN	QN	QN	QN	QN	QN
						-			
2,4,5 - 1 (2,4,5 - I riclorophenoxy acetic acid)	Mg/L	Q N	Q	QN	Q	QN	QN	QN	QN
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	76 <i>n</i>	Q	Q	QN	Q.	QN	Q	Q.	QN
3,5 - DCBA	1/g/r	S	9	2	QN	CN	S	2	
2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid)	7/6#	Q	Q	QN	Q	2	2	Q.	202
4 - Nitrophenol	1/g/r	QN ON	Q	S	CN	Ş	2	2	1
Acifluorfen	700	S	S	2		2 2	2 5	2	2
Bentazon	1/0//	S	2 2	2 2	2 2	2 2	2	QN.	Q N
Chloramben	1/0//	2			2 5	2	2	QN	QN
Dacthal	180		2 9	2	2	Q	QN	QN	ND
	//g/L	S	ON.	2	2	2	Q	QN	S

Table A-3 (cont.). KSC Background Screening Tables for surface water. Certain values have combined means based on the frequency of detection which was <20*, <10**, or <4*** hits. ND = non-detect.

		Combined Water	d Water	Banana	Banana	Ditches	hes	Indian	Mosquito
		Velicity	Colinity	7005	Divor	Calinity	Salinity	Rive	מסטער
		0-5 ppt	> 6 ppt	S S S S S S S S S S S S S S S S S S S	2	0-5 ppt	> 6 ppt	Lagoon	
Parameter	Units	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2
Chlorinated Herbicides (8151) (cont.)									
Dalabon	ηð/L	QN	QN	QN	QN	ND	ND	QN	QN
Dicambia	ng/L	QN	Q	QN	QN	QN	QN	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	1/6/r	QN	QN N	QN	QN	QN	QN	ND	ND
Dinoseb	MO/L	QN	QN	QN	QN	QN	QN	ND	ND
MCPA	Ng/L	QN	QN	Q	QN	QN	QN	QN	QN
MCPP	ηg/L	QN	QN	Q	QN	QN	QN	ND	S
Pentachlorophenol	1/g/r	QN	QN	용	QN	QN	ON	QN	QN
Picloram	/g/r	QN	QN	QN	QN	QN	ON	ND	NO
Sum of Chlorinated Herbicides	Mg/L	ND	ON	QN	Q	QN	QN	Q	9
Polyaromatic Hydrocarbons (8310)									
1-Methylnaphthalene	Mg/L	QN	QN	QN	QN	QN	QN	QN	QN
2-Methylnaphthalene	76/	QN	Q	QN	QN	QN	ND	QN	QN
Acenaphthylene	1/6/	QN	QN	Q	Q	QN	QN	QN	Q
Anthracene	1/g/	QN	Q	QN	QN	QN	Q	Q	QN
Benzo(a)anthracene * * *	1/g/r	QN	0.101	QN	Q	9	Q	9	0.14
Benzo(a)pyrene * * *	1/g/r	QN	0.115	ON	0.132	QN	Q	QN	0.174
Benzo(b)fluoranthene	7/6//	QN	QN	QN	QN	QN	QN	Q	QN
Benzo(q,h,i)perylene	Mg/L	QN	QN	QN	QN	QN	2	Q	Q
Benzo(k)fluoranthene* * *	767	QN	0.108	QN	0.14	QN	ΩN	Q	0.126
Chrysene * * *	76/r	Q	0.104	QN	QN	QN	2	2	0.16
Dibenzo(a,h)anthracene	761	Q	Q.	QN	QN	QN	Q	Q	Q
Fluoranthene	1/g//	9	QN	QN	QN	QN	2	Q	Q
Fluorene	ηgη.	Q	QN	QN	QN	Q.	2	2	2

Table A-3 (cont.). KSC Background Screening Tables for surface water. Certain values have combined means based on the frequency of detection which was <20°, <10°*, or <4*** hits. ND = non-detect.

		-							
		Compin	Combined Water	Banana	Banana	Ť	Ditches	Indian	Mosquito
	-	Salinity 0-5 pot	Salinity	Creek	River	Salinity	Salinity	River	Lagoon
Parameter	1 Inite	Mosn V 2	o A Do			0-5 ppt	> 6 ppt	Lagoon	
Polvaromatic Hydrocates (9310)	2115	Meall	Mean A 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2	Mean X 2
(cont.)									
Indeno(1,2,3-cd)pyrene * * *	ייט/ן	S	0 101	Q.	2	4			
Naphthalene	101			2 4	QN :	Q N	Q	QN	0.14
Phenanthrana	700	Q.	ON.	QN	QN	QN	QN	QN	QN
Pyrana	//g/L	QN:	2	Q	QN	QN	QN	QN	QN
Sim of Delices	#g/L	QN	Q	QN	ON	Q	QN	QN	S
Sum of Polyaromatic Hydrocarbons	76/	Q	QN	QN	9	QN	QN	CZ	2
Metals									2
Aluminum	7/0//	642	895	2200	640	640	0.40	9	
Antimony *	707	15	10	41.2	45	045	042	642	642
Arsenic * *	/2,	2	1	5	0	6	33	24	16
Barium	707	3 5	<u> </u>	9	29	QN	14.8	13	QN QN
Don't *	18r	2	ND	QN	NO	2	QN	QN	QN
Codoni	rg/L	QN	1.3	ND	2.3	QN	QN	QN	QN
Caumur	ng/r	QN	ND	QN	ΩN	2	QN	CN	S
Carcium	√g/r	196760	680484	000069	483060	196760	483060	690000	00000
Chloride	7∕6⁄⁄	3054300	24193940	28590000	17430760	3054300	17430760	2859000	030000
Chromium	ηg/L	QN	QN	QN	9	QN	CN	NO CN	20330000
Cobait	µg∕L	ND	QN	QN	2	QN	S	S	
copper	//g/r	QN	QN	2	9	QN	CN	2	2 2
Iron	<i>√</i> 8⁄	966	520	966	242	966	900	240	200
Lead *	/g/r	8.6	7.7	QN	8.5	8	2	NO NO	245
Magnesium	מער	634276	2031515	2203600	634076	504030	20000	2	2
Manganese *	707	78	23	20000	204570	0342/0	2293600	2293600	2293600
Mercury		2	3 2	9 5	QN.	8/	2	48	QN
Nickel	ייסיו מייסיו	2 2	2 2	2 9	Q !	Q	QN	QN	ND
Potassium	1 0	65404	20,000	QN	2	QN	QN	ND	Q
	WWL	92164	022182	527852	527852	65184	527852	528	1046667

Table A-3 (cont.). KSC Background Screening Tables for surface water. Certain values have combined means based on the frequency of detection which was <20*, <10**, or <4*** hits. ND = non-detect.

		Combined Water	d Water	Banana	Banana	Ditches	hes	Indian	Mosquito
		Salinity	Salinity	Creek	River	Salinity	Salinity	River	Lagoon
		0-5 ppt	> 6 ppt			0-5 ppt	≥ 6 ppt	Lagoon	
Parameter	Units	Mean X 2	Mean X 2	Mean X 2 Mean X 2 Mean X 2 Mean X 2	Mean X 2				
Metals (cont.)								-	
Selenium * * *	/⁄g/L	QN	20	QN	Q	ND	QΝ	25	ND
Silver * *	µg∕L	QN	2.8	QN	QN	QN	QN	2.6	ND
Sodium	/Jg/r	1789353	14012121	11380000	11380000	1789353	11380000	18061540	18061540
Thallium * * *	Mg/L	QN	2.9	QN	QN	ND	QN	4	2.6
Vanadium	₩g/L	QN	QN	QN	ND	ND	QN	ND	ND
Zinc	Mg/L	QN	QN	QN	QN	ND	QN	ND	QN

Table A-4. KSC Background Screening Tables for sediment. ND = non-detect.

		Sediment
Parameter	Units	Mean X 2
Organochlorine Pesticides (8081)		
4,4'-DDD	mg/kg	QN
4,4'-DDE	mg/kg	QN
4,4'-DDT	mg/kg	QN
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	mg/kg	QN
Aldrin	mg/kg	QN
Alpha - BHC	mg/kg	Q
Beta - BHC	mg/kg	QN
Chlordane (alpha)	mg/kg	Q
Chlordane (gamma)	mg/kg	QN
Chlordane (total)	mg/kg	Q.
Delta - BHC	mg/kg	Q
Dieldrin	mg/kg	QN N
Endosulfan I	mg/kg	QN
	mg/kg	QN
Endosulfan Sulfate	mg/kg	Q
Endrin	mg/kg	QV
Endrin Aldehyde	mg/kg	9
Endrin Ketone	mg/kg	N
Gamma -BHC (Lindane)	mg/kg	9
Heptachlor	mg/kg	N N
Heptachlor Epoxide (a)	mg/kg	Q
Heptachlor Epoxide (b)	mg/kg	2
Isodrin	mg/kg	Q
Methoxychlor	mg/kg	QN N
Mirex	mg/kg	Q.
Toxaphene	mg/kg	2
1		
•	mg/kg	ND
- 1221	mg/kg	QN
PCB - 1232	mg/kg	Q

Table A-4(cont.). KSC Background Screening Tables for sediment. ND = non-detect.

		Combined
		Sediment
Parameter	Units	Mean X 2
Aroclors (8082) (cont.)		
PCB - 1248	mg/kg	ND
PCB - 1254	mg/kg	ON
PCB - 1260	mg/kg	QN
Chlorinated Herbicides (8151)		
Silvex (2,4,5 - TP) (2-(2,4,5 -	mg/kg	Q
Trichlorophenoxy)) propionic acid		
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	mg/kg	0.017
2,4 - D (2,4 - Dichlorophenoxy acetic	mg/kg	0.017
acid)		
3,5 - DCBA	mg/kg	2
2,4 - DB (4 - (2,4 - Dichlorophenoxy)	mg/kg	<u>Q</u>
butyric acid)		
4 - Nitrophenol	mg/kg	Q
Acifluorfen	mg/kg	2
Bentazon	mg/kg	2
Chloramben	mg/kg	2
Dacthal	mg/kg	Q
Dalapon	mg/kg	2
Dicambia	mg/kg	Q
Dichlorprop [2-(2,4-Dichlorophenoxy)	mg/kg	2
Dinoseb	ma/ka	QN
MCPA	mg/kg	QN
MCPP	mg/kg	QN
Pentachlorophenol	mg/kg	ND
Picloram	mg/kg	QN

Table A-4(cont.). KSC Background Screening Tables for sediment. ND = non-detect.

		Combined
		Sediment
Parameter	Units	Mean X 2
Polyaromatic Hydrocarbons (8310)		
1-Methylnaphthalene	mg/kg	QN
2-Methylnaphthalene	mg/kg	Q
Acenaphthene	mg/kg	Q
Acenaphthylene	mg/kg	Q
Anthracene	mg/kg	2
Benzo(a)anthracene	mg/kg	0.003
Benzo(a)pyrene	mg/kg	0.004
Benzo(b)fluoranthene	mg/kg	9000
Benzo(g,h,i)perylene	mg/kg	QN
Benzo(k)fluoranthene	mg/kg	0.004
Chrysene	mg/kg	0.005
Dibenzo(a,h)anthracene	mg/kg	QN
Fluoranthene	mg/kg	0.008
Fluorene	mg/kg	QN
Indeno(1,2,3-cd)pyrene	mg/kg	Q
Naphthalene	mg/kg	0.031
Phenanthrene	mg/kg	0.003
Pyrene	mg/kg	0.004
Metais		
Aluminum	mg/kg	1953.6
Antimony	mg/kg	QN
Arsenic	mg/kg	1.63
Barium	mg/kg	QN
Beryllium	mg/kg	QN
Cadmium	mg/kg	QN
Calcium	mg/kg	104416
Chloride, total	mg/kg	ND
Chromium	mg/kg	4.95
Cobalt	mg/kg	ND
Copper	mg/kg	6.5
	2 2	,

Table A-4(cont.). KSC Background Screening Tables for sediment. ND = non-detect.

	<u> </u>	Combined
		Sediment
Parameter	Units	Mean X 2
Metals (cont.)		
Iron	mg/kg	1934
Lead	mg/kg	4.7
Magnesium	mg/kg	1949
Manganese	mg/kg	22.6
Mercury	mg/kg	0.05
Nickel	mg/kg	QN
Potassium	mg/kg	208
Selenium	mg/kg	QN
Silver	mg/kg	3.2
Sodium	mg/kg	5730
Thallium	mg/kg	1.33
Vanadium	mg/kg	5.2
Zinc	mg/kg	14.5

Appendix B

Soil Data for Kennedy Space Center

Table B-1. List of sample site descriptions for KSC Background Study soil sampling locations.

Soil Class	Class Number	Site id	Site Description
Coastal	-	SSC001	Coastal scrub south of Pump Station 7 (PS7)
Coastal	-	SSC002	Coastal scrub south of PS7
Coastal	-	SSC003	Coastal scrub north of PS7
Coastal	-	SSC004	Coastal scrub west side of Beach Road, north of LC41, Fish and Wildlife Service (FWS) harricade, nossible fire break
Coastal	-	SSC005	Coastal scrub west side of Beach Road, helicopter pad go in to the left
Coastal	-	SSC006	Coastal strand just north of fence line on CCAFS
Coastal	-	SSC007	Coastal strand north of Center Director's Conference Room (Beach House)
Coastal		SSC008	Coastal strand east of LC41
Coastal	1	600DSS	Coastal strand east of helicopter pad walk into palmettos from dune
Coastal	-	SSC010	Coastal strand north of LC41, east of Beach Road
Coastal	1	SSC011	Coastal dune just north of fence line on CCAFS
Coastal	_	SSC012	Coastal dune north of Center Director's Conference Room (Beach House)
Coastal	1	SSC013	Coastal dune east of LC41
Coastal	-	SSC014	Coastal dune east of helicopter pad walk in to the north
Coastal	1	SSC015	Coastal dune north of LC41, east of Beach Road
Coastal	-	SSC016	Playalinda parking 9
Coastal	_	SSC017	Playalinda parking 10
Coasta	_	SSC018	Playalinda walk up boardwalk to sea oat zone
Coastal	-	SSC019	Playalinda walk up boardwalk to sea oat zone
Coastal		SSC020	Playalinda walk up boardwalk to sea oat zone
Acid Scrub	2	SSC021	North of National Park Service (NPS) pay station at Playalinda entrance
Acid Scrub	2	SSC022	Shiloh 1 west of well location
Acid Scrub	2	SSC023	Shiloh 2 (south of Shiloh 1) firebreak
Acid Scrub		SSC024	KARS park, Hall Road, 1st Road to south just inside gate (head toward borrow pit area),
			stay in oaks
Acid Scrub	2	SSC025	Tel-IV use trail at Apiary 6
Acid Scrub		SSC026	Jerome Road east, Apiary 67
Acid Scrub	2	SSC027	Schwartz Road east, fire break to north
Acid Scrub		SSC028	Schwartz Road east, fire break to north
Acid Scrub		SSC029	Schwartz Road east, fire break turns to the west, on south side

Table B-1. (cont.).

2 SSC034 2 SSC034 2 SSC035 2 SSC036 2 SSC036 2 SSC036 3 SSC041 3 SSC041 3 SSC047 (9) 3 SSC049 (9) 3 SSC049 (9) 3 SSC049 (9) 3 SSC049 (9) 3 SSC049 (9) 3 SSC049 (9)	Acid Scrub	2 6	SSC030	Happy Creek Road east of railroad tracks burned area (higher area), south side
2 SSC033 2 SSC034 2 SSC036 2 SSC036 2 SSC038 2 SSC040 3 SSC041 3 SSC044 3 SSC044 3 SSC048 B 3 SSC049 S	Acid Scrub		SSC032	Hanov Creek Road at V recently human
2 SSC034 2 SSC035 2 SSC036 2 SSC037 2 SSC040 3 SSC041 3 SSC044 3 SSC048 I	Acid Scrub		SSC033	Happy Creek south of camera nad restoration sit
2 SSC035 2 SSC036 2 SSC037 2 SSC039 2 SSC040 3 SSC041 3 SSC042 3 SSC048 E	Acid Scrub		SSC034	Happy Creek sand road, restoration site
2 SSC036 2 SSC037 2 SSC038 2 SSC040 3 SSC041 3 SSC042 3 SSC044 3 SSC045 3 SSC046 13 SSC046 13 SSC046 13 SSC048 13 SSC049 13 SSC049 13 SSC050 14 SSC050 15 SSC050 16 SSC050 17 SSC050 18 SSC050 18 SSC050 19 SSC050 10 SS	Acid Scrub		SSC035	Happy Creek sand road, north side recent hum
2 SSC037 2 SSC038 2 SSC038 3 SSC041 3 SSC042 3 SSC043 3 SSC045 3 SSC045 3 SSC046 3 SSC046 3 SSC046 3 SSC046 3 SSC046 3 SSC046 3 SSC046	Acid Scrub		SSC036	Shiloh 1 south of sand road, east
10	Acid Scrub		SSC037	Scrubby flatwoods south of Shiloh 1, west side SB3
SSC039	Acid Scrub	2	SSC038	Farther in on same sand road as C37
3 SSC040 3 SSC041 3 SSC042 3 SSC044 3 SSC046 3 SSC046 3 SSC046 3 SSC046 3 SSC046 3 SSC048 3 SSC048 3 SSC048	Acid Scrub	2	SSC039	Shiloh 2, North side of sand road, recent hum
3 SSC041 3 SSC042 3 SSC044 3 SSC045 3 SSC046 3 SSC046 3 SSC046 3 SSC046 3 SSC048 3 SSC048 3 SSC050 13 SSC050	Acid Scrub	2	SSC040	Happy Creek sand road near sand pine
3 SSC042 3 SSC043 3 SSC044 3 SSC045 3 SSC046 3 SSC046 3 SSC048 3 SSC049 3 SSC050 13 SSC050	Coquina Scrub	င	SSC041	Haulover boat ramp just west of SR3
3 SSC044 3 SSC046 3 SSC046 3 SSC046 3 SSC047 3 SSC049 3 SSC069 3 SSC050	Coquina Scrub	င	SSC042	Haulover boat ramp just west of SR3
3 SSC045 3 SSC046 3 SSC046 3 SSC047 3 SSC048 3 SSC049 3 SSC050	Coquina Scrub	င	SSC043	Haulover boat ramp east of SR3
3 SSC045 3 SSC047 3 SSC048 3 SSC049 3 SSC049 3 SSC050	Coquina Scrub	င	SSC044	Shiloh 4 site, xeric hammock area don't go into scrub
3 SSC046 3 SSC048 3 SSC049 3 SSC050	Coquina Scrub	3	SSC045	Shiloh 4 scrub just past xeric hammock
3 SSC048 3 SSC049 3 SSC050	Coquina Scrub	8	SSC046	Dummit Grove, south end, xeric hammock
3 SSC048 3 SSC050 3 SSC050	Coquina Scrub	3	SSC047	South of Shiloh 1, on west side of SR3
3 SSC049 3 SSC050 3 SSC051	Coquina Scrub	င	SSC048	East of SR3, FWS sign and road south of C47
3 SSC050	Coquina Scrub	3	SSC049	South of C47, along west side of SR3
3 SSC051	Coquina Scrub	6	SSC050	North of Haulover, west side of SR3, south of C49
	Coquina Scrub	ဧ	SSC051	South of Haulover, west side of SR3, north of guard st

Table B-1. (cont.).

Coquina Scrub	6	SSC052	South of Haulover, north of guard shack, road to east turn left (north)
Coquina	8	SSC053	Continue north of C52, road turns to east, 2nd dirt road on the south side
Coquina Scrub	8	SSC054	Continue east of C53, turn left (north) on west side
Coquina Scrub	8	SSC055	South of Haulover, north of guard shack, road to east turn right (south)
Coquina Scrub	E	SSC056	South of Haulover, directly east of guard shack, south side of road
Coquina Scrub	8	SSC057	South of Haulover, south of guard shack, east side SR3, east of C58
Coquina Scrub	3	SSC058	South of Haulover, south of guard shack, west side SR3, west of C57
Coquina Scrub	8	SSC059	South of Haulover, south of guard shack, west side SR3, south of C58
Coquina Scrub	8	090088	South of Haulover, south of guard shack, east side SR3, south of C58, west of C59
Flatwoods	7	SSC061	KARS park Hall Road - south of powerline clearings, burned area
Flatwoods	7	SSC062	TEL-IV east of Apiary 6, (see flagging), gallberry, holly. & palmettos
Flatwoods	7	£902SS	Ransom Road, east, go in along pines and head south into small swale, go into palmettos from there
Flatwoods	7	SSC064	Schwartz Road East, fire break to the north off fire break to east across from C27
Flatwoods	4	SSC065	Happy Creek heading north just south of hammock
Flatwoods	4	SSC066	Apiary site 47, PAPI Lights Road, east of Apiary 47
Flatwoods	4	SSC067	Tel-IV Road, east, firebreak, south, go south east side go east to dead pine
Flatwoods	4	8902SS	Tel-IV Road, east, firebreak, south, go south west side go west 5 to 10 meters, palmetto- Ivonia
Flatwoods	4	690 2 SS	Tel-IV Road, east "pavement ends" sign, south side go south, palmetto-Ivonia
Flatwoods	7	SSC070	SR3 east side, north of Jerome Road new culvert, go east, palmetto-lyonia
Flatwoods	4	SSC071	SR3 east side, north of Ransom Road, go east
Flatwoods	4	SSC072	North end of SLF south side of entrance road, go south, palmetto-lyonia
Flatwoods	4	SSC073	West side of SLF, go to tower and north along SLF fence line, palmetto-lyonia
Flatwoods	4	SSC074	West side of SLF, go to tower and south along SLF fence line, palmetto-lyonia

Table B-1. (cont.).

Ī			
Fiatwoods	4	SSC075	West side of SLF, east side of road, palmetto-Ivonia
Flatwoods	4	SSC076	SR406/SR3 south along powerline, west palmetto-Ivonia recent burn
Flatwoods	4	SSC077	South of C76 west of powerline
Flatwoods	7	SSC078	PAPI Lights Road, NW of Aniary 47 adda Cut
Flatwoods	4	SSC079	West side SR3. north of Met Dome (east side of SLE) west of nounciling
Flatwoods	7	SSC080	North of 79, west of powerline
Hammocks	5	SSC081	Off of SR3 (north), this is north of SR3 & 402 intersection
Hammocks	5	SSC082	Jerome Road west, west of citrus groves (north side of road, see flagging)
Hammocks	5	SSC083	Jerome Road west, west of C83 on south side of road
Hammocks	5	SSC084	Range Road (connects Ransom Road and Jerome Road on west), west of road into
			hammock
Hammocks	5	SSC085	Range Road, north of Ransom Road, narrow strip between road and grove towards and
Hammocks	5	SSC086	Roberts Road on south side across from grove no in a wave from disturbed area
Hammocks	5	SSC087	Same as C86, just farther up the road
Hammocks	5	SSC088	Happy Creek Road east side of hammock
Hammocks	5	SSC089	West Jerome Road, east of well site
Hammocks	5	060DSS	West Jerome Road, west of well site
Hammocks	5	SSC091	West Jerome north (2nd) pump station furn anto powerline
Hammocks	5	SSC092	West of C91 along powerline north side by the hos tran
Hammocks	5	SSC093	Oak hammock trail- MINWR north of railroad tracks. Oak Cathodo aging
Hammocks	5	SSC094	West along railroad north of 402 south side on past wetter area 1 in Oct.
Hammocks	2	SSC095	East along railroad north of 402 south side near turn to the south cathering
Hammocks	5	960288	East along railroad north of 402 north side Oak-Cahbade Dalm past of "pc" sign
Hammocks	5	SSC097	East along railroad north of 402 south side Cabhane Palm west of "RE" sign
Hammocks	ည	8602SS	West of SR3, north of Shiloh 1 restoration site, west through old grove, 1st turn to left go straight at fork
Hammocks	5	SSC099	West of SR3, north of Shiloh 1 restoration site, to west through old grove, 1st turn to left go
			right at fork, west of hardwood swamp
Hammocks	5	SSC100	West of SR3 Turnbull drainage
Freshwater Wetlands	9	SSC101	Tel-IV, east of Apiary 6, see also C62
Freshwater Wetlands	9	SSC102	Tel-IV, east of Apiary 6

Table B-1. (cont.).

Freshwater	9	SSC103	Happy Creek Road, hammock on west side walk in to the south, mucky bog area
Wetlands			
Freshwater	9	SSC104	Happy Creek sand road, marsh at turn
Freshwater	9	SSC105	Happy Creek north of camera pad ignition strip swale
Wetlands			
Freshwater	9	SSC106	Tel-IV Road, east, firebreaks south, southwest side W Calamovilfa swale C68
Wetlands			
Freshwater	9	SSC107	Tel-IV Road, east, "pavement ends sign", go south past C69 Calamoviifa swale
Wetlands			
Freshwater	9	SSC108	Happy Creek sand road, hardwood swamp to southwest
Freshwater	9	SSC109	North end of SLF entrance road, north side swale go north Spartina bakeri
Wetlands			
Freshwater	9	SSC110	North end of SLF entrance road north side swale Calamovilfa
Wetlands			
Freshwater	9	SSC111	PAPI Lights Road northwest Apiary 47, Calamovilla swale
Wetlands			
Freshwater	9	SSC112	CNS entrance, KSC gate along railroad tracks, go west along powerline
Wetlands			
Freshwater	9	SSC113	MINWR, Oak Hammock Trail boardwalk
Wetlands			
Freshwater	9	SSC114	West along railroad tracks north of 402 south side, Cabbage palm
Wetlands			
Freshwater	9	SSC115	Firebreak north of NASA Causeway, Willow swamp
Wetlands			
Freshwater	9	SSC116	Firebreak north of NASA Causeway, west of Willow swamp (C115), Spartina
Wetlands			
Freshwater	9	SSC117	West of SR3, north of Shiloh 1 restoration, 1st turn to the left (west) turn right, barricade,
Wetlands	:		either side of road
Freshwater	9	SSC118	West SR3 Turnbull drainage west end, south side
Wetlands			
Freshwater	ဖ	SSC119	West SR3 Turnbull drainage Spartina marsh, near well site
Wetlands			

Table B-1. (cont.).

Table B-1. (cont.).

Saltwater Wetlands	7	SSC137	Dike road east of Bio-lab, Spartina bakeri on west (this is the same dike road as C123, CNS entrance road)
Saltwater Wetlands	7	SSC138	Dike Road east of Bio-lab, mangrove area on east of dike road (this is the same dike road as C123, CNS entrance road)
Saltwater Wetlands	7	SSC139	Dike road east of Bio-lab, <u>Distichlis</u> south of C138, west side of dike road (this is the same dike road as C123, CNS entrance road)
Saltwater Wetlands	7	SSC140	T-16 near well site
Citrus Scrub	8	SSC141	Shiloh 1 restoration, east of well site
Citrus Scrub	8	SSC142	North of Shiloh 1 restoration, 1st road east of SR3, pine plantation
Citrus Scrub	8	SSC143	North of Shiloh 1 restoration, west of SR3 across from C142, oaks planted
Citrus Scrub	8	SSC144	West side of SR3, north of Shiloh 1 restoration
Citrus Scrub	8	SSC145	West of SR3 south of Shiloh 2 restoration
Citrus Scrub	8	SSC146	North of C145, pine plantation on west, south of Shiloh 2 restoration
Citrus Scrub	8	SSC147	South end of C145 grove, entrance to Shiloh 1 impoundment
Citrus Scrub	8	SSC148	Entrance to Shiloh 3 impoundment
Citrus Scrub	8	SSC149	Heading south towards Haulover, Road cuts off to west side parallel with SR3
Citrus Scrub	8	SSC150	Just north of Haulover on west side of SR3
Citrus Scrub	8	SSC151	South of Shiloh 4 restoration, south of Haulover, east side of SR3
Citrus Scrub	æ	SSC152	Shiloh 1 restoration, east, abandoned grove, south of sand road, east of well site
Citrus Scrub	8	SSC153	Shiloh 1 restoration, northwest side of SR3 planted oaks
Citrus Scrub	8	SSC154	North of Shiloh 1 restoration, east side of SR3 old field, north of C142
Citrus Scrub	8	SSC155	South of Shiloh 2 restoration, east SR3, fishing access road
Citrus Scrub	8	SSC156	Young grove near radar site (empty) west of SR3
Citrus Scrub	8	SSC157	South of Shiloh 4, south of Haulover west of SR3
Citrus Scrub	8	SSC158	West of SR3 Griffis Cemetery, north of Shiloh 1 restoration, oak planted in rows
Citrus Scrub	8	SSC159	1st grove heading north towards Haulover on west side of road
Citrus Scrub	8	SSC160	North of C160 on east, same grove as C159
Citrus	6	SSC161	Jerome Road west, north of road
Hammock			
Citrus	6	SSC162	Jerome Road west, south of road
Hammock			

Table B-1. (cont.).

9 SSC163 Range Road, 1/2 way down on west side, south of Ransom Road	9 SSC164 Ransom Road on south side of road	9 SSC165 Range Road, north of Ransom Road	9 SSC166 Roberts Road, north end	9 SSC167 Roberts Road, grove on north side of road	9 SSC168 Same as C167	9 SSC169 End of dirt road south of Robert's Road (1st dirt road)	9 SSC170 Grove before second dirt road on Roberts Road, south side	9 SSC171 Robert's Road heading north, young grove on east side of the road	9 SSC172 1st grove north of Schwartz Road, west	9 SSC173 2nd grove north of Schwartz Road, west	9 SSC174 3rd grove north if Schwartz Road, west, grove after security training area	9 SSC175 SR3 west side at Tel-IV Road	9 SSC176 Jerome Road west of hammock sites, guinea grass	9 SSC177 Jerome Road west, northeast well site	9 SSC178 Schwartz Road west, east of tower	9 SSC179 Fire break north of the NASA Causeway
Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus	Citrus
Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock	Hammock

Table B-1. (cont.).

Citrus	6	SSC180	Fire break north of the NASA Causeway
Hammock			
Disturbed	10	SSC181	Road side, south side of road, west of Oak Hammock Trail, Gate12, 4 roads east of FWS service area
Disturbed	10	SSC182	PAPI Lights Road, east of Apiary 47, borrow pond edge
Disturbed	10	SSC183	Schwartz Road west, borrow pit to west of road, just south of tower
Disturbed	10	SSC184	Saturn Causeway, south side of road, just east of Vertical Assembly Building (VAB),
			vegetation begins southeast of VAB, west of Component Clean Facility (CCF)
Disturbed	10	SSC185	Saturn Causeway, south side of road, east of CCF, west of East Crawlerway Park Site
Disturbed	10	SSC186	East of C186, west of boat dock, grassy area near river
Disturbed	10	SSC187	Road to Pad B, north of Saturn Causeway, east side of road
Disturbed	10	SSC188	Road to Pad B, north of Saturn Causeway, east side of road, north of C187
Disturbed	10	SSC189	Saturn Causeway heading east towards pad, past guard shack, south side of road
Disturbed	10	SSC190	Road side corner of C avenue and NASA Causeway
Disturbed	10	SSC191	Fifth street southeast, south side of road between B and C Avenue
Disturbed	10	SSC192	Corner of D Avenue and Fifth Street
Disturbed	10	SSC193	3rd Street and D Avenue southwest corner, field
Disturbed	10	SSC194	10th Street southeast on corner, near borrow pond
Disturbed	10	SSC195	North side of NASA Causeway, (Juliet parking area)
Disturbed	10	SSC196	South side of NASA Causeway, (5th pole west of CCAFS entrance gate)
Disturbed	10	SSC197	Static Test Road
Disturbed	10	SSC198	Schwartz Road east, north side of road
Disturbed	10	SSC199	Saturn Causeway, east of VAB, west of CCF, east of C184
Disturbed	10	SSC200	Grassy area behind headquarters building, clump of pines, south side of parking lot, west
			side of 3rd street entrance

Table B-2. List of X and Y coordinates for soil sampling locations. Coordinates were collected with a Global Positioning System (GPS) in Florida Sate Plane NAD27. SSC = surficial soil chemistry.

Soil ID	X	Υ
SSC001	635,559.888	1,540,287.250
SSC002	636,876.886	
SSC003	635,618.720	
SSC004	634,292.390	
SSC005	632,467.913	
SSC006	637,724.871	1,539,773.513
SSC007	636,981.726	1,542,315.272
SSC008	636,323.234	1,544,442.330
SSC009	632,802.003	1,550,316.155
SSC010	634,396.327	1,548,079.570
SSC011	638,083.057	1,539,719.494
SSC012	637,196.803	1,542,455.578
SSC013	636,621.469	1,544,212.823
SSC014	632,787.067	1,550,456.732
SSC015	634,589.436	1,548,212.906
SSC016	610,558.575	1,582,221.149
SSC017	609,852.836	1,583,184.654
SSC018	607,789.661	1,586,402.850
SSC019	612,881.777	1,579,165.089
SSC020	616,828.370	1,572,917.021
SSC021	601,368.381	1,567,387.936
SSC022	565,059.649	1,618,388.874
SSC023	568,479.954	1,614,922.015
SSC024	604,238.632	1,484,851.910
SSC025	605,699.056	1,506,419.486
SSC026	606,666.112	1,511,719.630
SSC027	618,193.048	1,539,631.836
SSC028	617,638.209	1,543,048.246
SSC029	615,283.086	1,542,965.808
SSC030	603,460.023	1,557,327.261
SSC031	607,915.851	1,557,475.573
SSC032	607,553.829	1,563,242.767
SSC033	610,368.120	1,559,388.205
SSC034	613,195.871	1,558,821.731
SSC035	611,508.920	1,560,306.478
SSC036	566,740.056	1,617,585.098
SSC037		1,616,290.888
SSC038		1,616,347.737

Table B-2. (cont.).

Soil ID	X	Υ
SSC039	566,282.627	1,615,158.454
SSC040	612,883.376	1,559,086.328
SSC041	578,778.740	1,599,758.815
SSC042	579,056.975	1,599,406.156
SSC043	579,222.770	1,599,789.408
SSC044	582,763.044	1,595,339.026
SSC045	582,911.112	1,595,346.487
SSC046	586,437.095	1,590,474.710
SSC047	574,652.265	1,604,299.356
SSC048	575,360.744	1,604,152.301
SSC049	575,422.411	1,603,457.248
SSC050	576,972.837	1,602,035.648
SSC051	580,813.829	1,597,173.634
SSC052	580,610.703	1,598,640.819
SSC053	580,985.758	1,598,537.794
SSC054	581,014.356	1,597,789.456
SSC055	581,406.979	1,597,061.068
SSC056	581,955.095	1,596,864.730
SSC057	582,692.023	1,595,145.655
SSC058	582,702.399	1,594,991.333
SSC059	583,193.730	1,594,316.131
SSC060	583,318.923	1,594,452.114
SSC061	604,750.886	1,483,915.221
SSC062	607,385.434	1,506,413.834
SSC063	609,264.988	1,517,079.898
SSC064	618,362.420	1,539,643.279
SSC065	608,016.025	1,560,315.807
SSC066	591,732.561	1,571,679.694
SSC067	605,845.344	1,499,896.502
SSC068	605,619.471	1,500,788.579
SSC069	610,125.060	1,501,152.663
SSC070	605,620.770	1,514,549.607
SSC071	606,894.556	1,519,117.401
SSC072	596,862.420	1,563,078.181
SSC073	593,221.534	1,562,494.984
SSC074	595,598.519	1,558,966.230
SSC075	592,177.277	
SSC076	588,970.577	
SSC077	589,808.804	
SSC078	591,034.142	
SSC079	600,588.594	1,560,098.305

Table B-2. (cont.).

Soil ID	X	Y
SSC080	599,502.216	1,562,058.338
SSC081	593,680.417	1,573,104.440
SSC082	601,068.864	1,510,630.436
SSC083	599,688.833	1,510,321.281
SSC084	604,149.007	1,512,750.057
SSC085	603,990.407	1,521,138.583
SSC086	607,422.567	1,529,211.500
SSC087	605,600.359	1,529,259.588
SSC088	607,872.768	1,561,257.939
SSC089_	598,791.746	1,509,395.713
SSC090	598,611.669	1,509,591.517
SSC091	600,856.516	1,512,034.520
SSC092	600,368.755	1,512,004.113
SSC093	591,073.976	1,567,465.836
SSC094	584,310.646	1,567,110.967
SSC095	600,105.060	1,567,052.326
SSC096	599,650.813	1,567,242.093
SSC097	599,278.110	1,567,068.650
SSC098	562,335.619	1,617,813.418
SSC099	560,921.545	1,618,913.702
SSC100	547,797.338	1,630,366.314
SSC101	604,894.394	1,506,461.919
SSC102	604,845.132	1,506,310.755
SSC103	607,677.702	1,561,224.127
SSC104	616,782.497	1,557,346.864
SSC105	611,041.211	1,561,405.453
SSC106	605,554.265	1,500,773.887
SSC107	610,127.351	1,501,084.396
SSC108	611,363.006	1,559,995.106
SSC109	596,196.063	1,563,274.825
SSC110	596,270.592	1,563,189.280
SSC111	590,970.604	1,571,764.278
SSC112	600,484.174	1,566,396.276
SSC113	591,047.635	1,567,606.717
SSC114	584,289.438	1,566,721.891
SSC115	618,236.472	1,529,500.136
SSC116	618,186.283	1,529,483.522
SSC117	561,300.869	1,618,930.196
SSC118	550,593.438	1,632,365.086
SSC119	548,771.920	1,631,900.516
SSC120	553,015.327	1,632,565.757

Table B-2. (cont.).

Soil ID	Х	Υ
SSC121	615,473.842	1,573,996.947
SSC122	606,350.744	1,588,066.131
SSC123	606,596.614	1,571,720.061
SSC124	578,644.619	1,571,949.019
SSC125	578,116.633	1,571,672.979
SSC126	576,638.310	1,571,612.072
SSC127	575,847.309	1,571,212.803
SSC128	574,043.276	1,571,466.773
SSC129	573,332.913	1,579,103.081
SSC130	573,949.314	1,578,180.237
SSC131	575,688.091	1,578,153.860
SSC132	577,963.280	1,580,516.083
SSC133	567,963.565	1,609,866.779
SSC134	568,245.734	1,610,263.829
SSC135	568,476.199	1,610,470.354
SSC136	559,138.645	1,618,610.925
SSC137	596,635.658	1,581,364.153
SSC138	604,332.458	1,574,680.610
SSC139	604,477.377	1,573,920.625
SSC140	587,132.686	1,559,638.542
SSC141	565,184.173	1,618,420.660
SSC142	559,696.649	1,622,635.199
SSC143	559,381.024	1,622,517.272
SSC144	564,588.777	1,616,827.427
SSC145	566,765.694	1,613,843.567
SSC146	565,869.274	1,615,200.591
SSC147	568,077.297	1,612,188.361
SSC148	573,148.326	1,606,033.114
SSC149	575,879.833	1,602,809.141
SSC150		1,601,490.712
SSC151	584,713.075	1,593,348.901
SSC152	566,258.056	1,618,370.849
SSC153	558,547.745	1,623,263.768
SSC154	558,428.031	1,623,785.919
SSC155	573,121.840	1,607,411.617
SSC156	572,122.503	1,607,287.421
SSC157	583,696.670	1,593,307.977
SSC158	561,511.845	1,620,052.610
SSC159	584,828.988	1,591,680.751
SSC160	586,140.030	1,591,840.985
SSC161	603,727.823	1,511,525.195

Table B-2. (cont.).

	Soil ID	X	Υ
	SSC162	601,603.380	1,510,858.552
	SSC163	603,811.961	1,514,485.058
	SSC164	602,715.129	
	SSC165	604,426.939	1,518,643.132
	SSC166	607,690.082	1,533,694.904
	SSC167	607,323.682	1,529,460.857
	SSC168	605,579.040	1,529,466.689
	SSC169	604,422.300	1,527,810.271
	SSC170	603,023.851	1,529,254.024
	SSC171	607,870.187	1,534,888.055
	SSC172	607,634.377	1,537,475.273
	SSC173	606,258.752	1,537,453.038
Į	SSC174	601,022.518	1,537,456.719
١	SSC175	602,762.057	1,509,678.480
ļ	SSC176	600,270.364	1,510,578.584
l	SSC177	598,790.596	1,509,524.316
l	SSC178	606,199.740	1,537,201.713
L	SSC179	616,899.285	1,524,303.096
L	SSC180	616,977.207	1,524,556.431
L	SSC181	588,701.406	1,566,578.973
L	SSC182	593,695.436	1,571,302.690
L	SSC183	606,527.860	1,534,758.207
L	SSC184	614,405.086	1,546,039.397
_	SSC185	619,335.805	1,548,233.123
_	SSC186	621,532.653	1,548,640.011
	SSC187	622,180.375	1,551,858.476
	SSC188	621,731.107	1,554,776.027
	SSC189	625,601.316	1,551,004.880
	SSC190	611,195.005	1,523,787.431
_	SSC191	610,121.372	1,520,246.468
	SSC192	612,851.045	1,520,360.862
	SSC193	612,740.912	1,521,691.770
_	SSC194	614,627.624	1,517,427.239
_	SSC195	621,461.603	1,521,772.762
	SSC196	630,075.048	1,515,425.024
	SSC197	621,269.415	1,524,456.541
	SSC198	613,171.689	1,537,425.035
	SSC199	615,430.579	1,546,485.010
	SSC200	611,873.442	1,522,200.218

Table B-3. Minimum, maximum, mean, and standard deviation for each parameter analyzed in all soil samples.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)						
4,4'-DDD	μ g/k g	220	1.7	96	2.61	6.44
4,4'-DDE	μ g/kg	220	1.7	110	4.54	11.88
4,4'-DDT	μg/kg	220	1.7	96	4.06	11.03
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/kg	220	5.1	222	11.20	25.08
Aldrin	μg/kg	220	ND	ND	ND	ND
Alpha – BHC	μg/kg	220	ND	ND	ND	ND
Beta – BHC	μg/kg	220	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	220	1.7	41.0	2.4	3.2
Chlordane (gamma)	μg/kg	220	1.4	41.0	2.4	3.2
Chlordane (total)	μg/kg	220	1.7	62	2.41	4.17
Delta – BHC	μg/kg	220	ND	ND	ND	ND
Dieldrin	μg/kg	220	1.7	190	4.31	17.41
Endosulfan I	μg/kg	220	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	220	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	220	ND	ND	ND	ND
Endrin	μg/kg	220	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	220	ND	ND	ND	ND
Endrin Ketone	μg/kg	220	1.7	10.5	2.15	1.07
Gamma –BHC (Lindane)	μg/kg	220	1.7	14	2.18	1.29
Heptachlor	μg/kg	220	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	220	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	220	ND	ND	ND	ND
Isodrin	μg/kg	220	ND	ND	ND	ND
Methoxychlor	μg/kg	220	ND	ND	ND	ND
Mirex	μg/kg	220	ND	ND	ND	ND
Toxaphene	μg/kg	220	ND	ND	ND	ND
Aroclors (8082)	Party					
PCB - 1016/1242	μg/kg	220	ND	ND	ND	ND
PCB - 1221	μg/kg	220	ND	ND	ND	ND
PCB - 1232	μg/kg	220	ND	ND	ND	ND
PCB - 1248	μg/kg	220	ND	ND	ND	ND
PCB - 1254	μg/kg	220	ND	ND	ND	ND
PCB - 1260	μg/kg	220	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
(2-(2,4,5 - Trichlorophenoxy)) propionic acid (2,4,5 - TP) (Silvex)	μ g/kg	220	5	320	9.2	22.4
2,4,5 - Trichlorophenoxy acetic acid (2,4,5-T) μg/kg	220	ND	ND	ND	ND
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	220	5	62	8.0	8.6
3,5 – DCBA	μg/kg	220		ND	ND	ND
4-(2,4 - Dichlorophenoxy)butyric acid (2,4-DE		220	5	60	7.9	7.7
4 – Nitrophenol	μg/kg			ND	ND	ND
Acifluorien	μg/kg	220		ND	ND	ND
Bentazon	μg/kg	220		ND	ND	ND
Chloramben	μg/kg			ND	ND	ND
Dacthal	μg/kg			ND	ND	ND

Table B-3. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)		 		·		Deviation
Dalapon	μg/kg	220	ND	ND	ND	ND
Dicambia	μg/kg	+	ND	ND	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy)	μg/kg		ND	ND	ND	ND
proponic acid]	-33					''
Dinoseb	μg/kg	220	ND	ND	ND	ND
MCPA	μg/kg	220	ND	ND	ND	ND
MCPP	μg/kg	+	ND	ND	ND	ND
Pentachlorophenol	μg/kg	220	ND	ND	ND	ND
Picloram	μg/kg	220	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/kg	220	15	440	25.1	34.7
PAHs (8310)	<u> </u>	220	,,,	440	25.1	34.7
1-Methylnaphthalene	μ g/kg	220	8.5	115	14.08	15.22
2-Methylnaphthalene	μg/kg	220	8.5	290	16.36	27.86
Acenaphthene	μg/kg	220	ND ND	ND	ND	ND
Acenaphthylene	μg/kg	220	1.7	49	2.91	4.29
Anthracene	μg/kg	220	8.5	115	13.28	13.43
Benzo(a)anthracene	μ g/kg μ g/kg	220	0.9	130	2.44	
Benzo(a)pyrene	μg/kg μg/kg	220	1	250	3.4	9.04
Benzo(b)fluoranthene		220	1.5	220		17.2
Benzo(g,h,i)perylene	μg/kg	220	1.7		3.82	15.06
Benzo(k)fluoranthene	μg/kg	220	1.7	190	4.80	14.11
Chrysene	μg/kg	220	0.9	120	2.9	10.4
Dibenzo(a,h)anthracene	μg/kg	220	1.7	200	3.12	13.87
Fluoranthene	μg/kg	220		59	3.05	4.87
Fluorene	μg/kg	+	1.7	540	7.16	39.18
Indeno(1,2,3-cd)pyrene	μg/kg	220		170	3.39	11.60
Naphthalene	μg/kg	220	0.9	210	2.88	14.33
Phenanthrene	μg/kg	220	8.5	930	29.36	99.33
Pyrene	μg/kg	220	0.9	230	4.08	19.51
Sum of PAHs	μg/kg	220	0.9	320	5.07	28.59
Metals	μg/kg	220	50.5	2397.5	122.06	230.11
Aluminum	mg/kg	220	26	22000	14440	0570.4
Antimony	mg/kg	220	ND ND	ND	1444.6	2576.4
Arsenic	mg/kg	219	0.2	8.5	ND 0.76	ND 1.06
Barium	mg/kg	220	1.8	60	13.07	1.06 6.68
Beryllium	mg/kg	220	ND ND	ND	ND	ND
Cadmium	mg/kg	220	0.1	3	0.62	0.34
Calcium	mg/kg	220	38	260000	16466.2	36189.7
Chromium	mg/kg	220	0.5	34	4.55	6.25
Cobalt	mg/kg	220	ND ND	ND	ND	ND
Copper	mg/kg	220	0.5	130	13.94	26.97
Iron	mg/kg	220	24	16000	1283.5	1909.3
Lead	mg/kg	220	0.5	150	5.82	11.53
Magnesium	mg/kg	220	12.5	11000	549.22	1354.42
Manganese	mg/kg	220	0.5	200	25.75	39.52
Mercury	mg/kg	220	0.005	0.19	0.1570	0.2620
Nickel	mg/kg	220	ND	ND	ND	ND

Table B-3. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Potassium	mg/kg	220	12.5	6200	229.04	673.68
Selenium	mg/kg	220	0.02	6	1.248	0.774
Silver	mg/kg	220	0.95	6	1.270	0.658
Sodium	mg/kg	220	3.3	67000	1459.28	6852.24
Thallium	mg/kg	220	0.5	3.6	0.69	0.52
Vanadium	mg/kg	220	0.5	28	3.19	4.45
Zinc	mg/kg	220	1.2	140	15.14	21.28
Other Parameters						
Bulk density	g/cm ³	219	0.2	1.4	0.96	0.27
Cation exchange capacity (as Na)	meq/	202	0.5	85.0	14.0	14.2
, can be a second of the secon	100g	İ				
Hydrogen ion		219	1.58E-08	2.51189E-04	7.29E-06	2.22E-05
Percent solids	%	220	16	100	85.5	17.6
Resistivity	ohm-cm	199	26	3100000	614762.4	792326.6
Texture (No. 4)	%	220	73.9	100	99.61	1.98
Texture (no. 10)	%	220	68.3	100	98.47	4.40
Texture (No. 40)	%	220	36.4	99.7	85.73	15.09
Texture (no. 60)	%	220	7.4	98.5	71.17	24.86
Texture (No. 100)	%	220	0.9	80.2	35.83	20.20
Texture (No. 200)	%	220	0	77	8.0	9.8

Table B-4. Minimum, maximum, mean, and standard deviation for each parameter analyzed in coastal soils.

Parameter	Unit	s N	Minimum	Maximum	Mean	Std.
Organochlorine Pesticides (8081)						Deviation
4,4'-DDD	μg/kg	23	ND	1	1.5	
4,4'-DDE		_		ND	ND	ND
4,4'-DDT	μg/kg			ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4'4-DDT	μg/kg			ND	ND	ND
Aldrin	μg/kg			ND	ND	ND
Alpha - BHC	μg/kg			ND	ND	ND
Beta - BHC	μg/kg			ND	ND	ND
Chlordane (alpha)	μ g/k g		ND	ND	ND	ND
Chlordane (gamma)	μg/kg		ND	ND	ND	ND
Chlordane (total)	μg/kg		ND	ND	ND	ND
Delta – BHC	μg/kg	_	ND	ND	ND	ND
Dieldrin	μg/kg	_	ND	ND	ND	ND
Endosulfan I	μ g/kg		1.7	3.3	1.82	0.34
Endosulfan II (beta)	μg/kg		ND	ND	ND	ND
Endosulfan Sulfate	μg/kg		ND	ND	ND	ND
Endrin	μg/kg		ND	ND	ND	ND
Endrin Aldehyde	μg/kg	_	ND	ND	ND	ND
Endrin Ketone	μ g/kg		ND	ND	ND	ND
	μg/kg		ND	ND	ND	ND
Gamma-BHC (Lindane) Heptachlor	μ g/kg	23	ND	ND	ND	ND
	μg/kg	23	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	23	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	23	ND	ND	ND	ND
	μg/kg	23	ND	ND	ND	ND
Methoxychlor Mirex	μg/kg	23	ND	ND	ND	ND
	μg/kg	23	ND	ND	ND	ND
Toxaphene	μ g/k g	23	ND	ND	ND	ND
Aroclors (8082)						
PCB - 1016/1242	μg/kg	23	ND	ND	ND	ND
PCB - 1221	μg/kg	23	ND	ND	ND	ND
PCB - 1232	μ g/kg	23	ND	ND	ND	ND
PCB - 1248	μg/kg	23	ND	ND	ND	ND
PCB - 1254	μ g/k g	23	ND	ND	ND	ND
PCB - 1260	μ g/k g	23	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
-(2,4,5 - Trichlorophenoxy) propionic acid 2,4,5 - TP) (Silvex)	μg/kg	23	ND	ND	ND	ND
,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	23	ND	ND	ND	115
,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	23	ND	ND		ND
5 – DCBA	μg/kg	23	ND	ND	ND	ND
-(2,4-Dichlorophenoxy)butyric acid (2,4-DB)	μg/kg μg/kg	23	5	19	ND F.O.	ND
- Nitrophenol	μg/kg	23	ND	ND ND	5.8	2.9
cifluorfen		23	ND	ND	ND	ND
entazon		23	ND	ND	ND	ND
hioramben		23	ND		ND	ND
	μy/Ny	20	ND	ND	ND	ND

Table B-4. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)						
Dacthal	μ g/kg	23	ND	ND	ND	ND
Dalapon	μg/kg	23	ND	ND	ND	ND
Dicambia	μg/kg	23	ND	ND	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	μg/kg	23	ND	ND	ND	ND
Dinoseb	μ g/kg	23	ND	ND	ND	ND
MCPA	μg/kg	23	ND	ND	ND	ND
MCPP	μg/kg	23	ND	ND	ND	ND
Pentachlorophenol	μg/kg	23	ND	ND	ND	ND
Picloram	μg/kg	23	ND	ND	ND	ND
Sum Chlorinated Herbicides	μg/kg	23	15	29	16.1	3.0
PAHs (8310)	μg/ng	-				
1-Methylnaphthalene	μg/kg	23	ND	ND	ND	ND
2-Methylnaphthalene	μg/kg	23	8.5	290	21.00	58.64
Acenaphthene	μg/kg	23	ND	ND	ND	ND
	μg/kg	23	1.7	49	3.81	9.85
Acenaphthylene	μg/kg μg/kg	23	ND	ND	ND	ND
Anthracene	μg/kg	23	0.9	3.2	1.02	0.48
Benzo(a)anthracene	μg/kg μg/kg	23	1	3	1.4	0.7
Benzo(a)pyrene		23	1.5	7	1.87	1.24
Benzo(b)fluoranthene	μg/kg	23	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/kg	23	1	6	1.4	1.1
Benzo(k)fluoranthene	μg/kg	23	0.9	3.1	1.11	0.54
Chrysene	μg/kg	23	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/kg	23	ND	ND	ND	ND
Fluoranthene	μg/kg	23	ND	ND	ND	ND
Fluorene	μg/kg	23	0.9	5.4	1.12	0.94
Indeno(1,2,3-cd)pyrene	μg/kg		8.5	20	9.26	2.42
Naphthalene	μg/kg	23	0.9	16	1.58	3.14
Phenanthrene	μg/kg	23	ND	ND	ND	ND
Pyrene	μg/kg	23	50.5	347.8	69.04	62.22
Sum of PAHs	μ g/kg	23	50.5	347.0	05.04	02.22
Metals		100	120	1400	282.6	262.7
Aluminum	mg/kg		ND	ND	ND	ND
Antimony	mg/kg mg/kg		0.2	6.4	2.33	1.28
Arsenic	mg/kg		1.8	13.5	9.97	1.94
Barium	mg/kg	_	ND	ND	ND	ND
Beryllium	mg/kg		0.1	0.5	0.48	0.08
Cadmium	mg/kg			92000	50043.5	18496.9
Calcium Chromium	mg/kg			3.3	2.11	0.57
Cobalt	mg/kg			ND	ND	ND
Copper	mg/kg	_		3.5	1.84	1.01
Iron	mg/kg	_		1500	1063.0	209.6
Lead	mg/kg	_		3.5	2.16	0.90
Magnesium	mg/kg		140	640	370.0	131.6
Manganese	mg/kg		6	15	11.3	2.3

Table B-4 (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
Metals (cont.)		+	 			Deviation
Mercury	maka	100		 		
Nickel	mg/kg	23		ND	ND	ND
Potassium	mg/kg	23		ND	ND	ND
Selenium	mg/kg	23	+	200	58.39	37.77
Silver	mg/kg	23	+	ND	ND	ND
Sodium	mg/kg	23		1.5	1.020	0.105
Thallium	mg/kg	23	180	920	461.3	169.3
Vanadium	mg/kg	23	0.5	3.6	0.77	0.75
Zinc	mg/kg	23	1.4	3.2	2.36	0.46
Other Parameters	mg/kg	23	1.2	19	6.6	5.1
Bulk density		L_				
Cation exchange capacity	g/cm ³	23	0.9	1.4	1.19	0.13
Oation exchange capacity	meq/	16	1	6.0	2.5	1.5
Hydrogen ion	100g	22	0.004075.00	4.00====		
Percent solids	%		3.98107E-09	1.99526E-08	8.03E-09	3.07E-09
Resistivity		23	75	99	96.3	5.7
Texture (No. 4)	ohm-cm	23	13000	2000000	586260.9	660162.2
Texture (no. 10)	%	23	99	100	99.9	0.2
Texture (No. 40)	%	23	97.9	100	99.82	0.44
Texture (no. 60)	%	23	50	97.9	78.15	12.97
	%	23	7.4	70.4	31.49	16.50
Texture (No. 100)	%	23	0.9	28.4	6.44	6.25
Fexture (No. 200)	%	23	0	4	0.6	0.8

Table B-5. Minimum, maximum, mean, and standard deviation for each parameter analyzed in acid scrub soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)	· · · · · · · · · · · · · · · · · · ·					
4,4'-DDD	μg/kg	22	ND	ND	ND	ND
4,4'-DDE	μg/kg	22	ND	ND	ND	ND
4,4'-DDT	μg/kg	22	ND	ND	ND	ND
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μ g/kg	22	ND	ND	ND	ND
Aldrin	μg/kg	22	ND	ND	ND	ND
Alpha – BHC	μg/kg	22	ND	ND	ND	ND
Beta - BHC	μg/kg	22	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	22	ND	ND	ND	ND
Chlordane (gamma)	μg/kg	22	ND	ND	ND	ND
Chlordane (total)	μg/kg	22	ND	ND	ND	ND
Delta – BHC	μg/kg	22	ND	ND	ND	ND
Dieldrin	μg/kg	22	ND	ND	ND	ND
Endosulfan I	μg/kg	22	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	22	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	22	ND	ND	ND	ND
Endrin	μg/kg	22	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	22	ND	ND	ND	ND
Endrin Ketone	μg/kg	22	ND	ND	ND	ND
Gamma-BHC (Lindane)	μg/kg	22	ND	ND	ND	ND
Heptachlor	μg/kg	22	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	22	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	22	ND	ND	ND	ND
Isodrin	μg/kg	22	ND	ND	ND	ND
Methoxychlor	μg/kg	22	ND	ND	ND	ND
Mirex	μg/kg	22	ND	ND	ND	ND
Toxaphene	μg/kg	22	ND	ND	ND	ND
Aroclors (8082)	FSINS					
PCB - 1016/1242	μg/kg	22	ND	ND	ND	ND
PCB - 1221	μg/kg	22	ND	ND	ND	ND
PCB - 1232	μg/kg	22	ND	ND	ND	ND
PCB - 1248	μg/kg	22	ND	ND	ND	ND
PCB - 1254	μg/kg	22	ND	ND	ND	ND
PCB - 1260	μg/kg	22	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
2-(2,4,5 - Trichlorophenoxy) propionic acid (2,4,5 - TP) (Silvex)	μg/kg	22	ND	ND	ND	ND
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	μg/kg	22				
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	22	5	40	7.0	7.4
3,5 - DCBA	μg/kg	22	ND	ND	ND	ND
4-(2,4- Dichlorophenoxy)butyric acid (2,4 -DB)		22	5	33	6.7	5.9
4 - Nitrophenol	μg/kg	22	ND	ND	ND	ND
Acifluorfen	μg/kg	22	ND	ND	ND	ND
Bentazon	μ g/kg	22	ND	ND	ND_	ND
Chloramben	μg/kg	22	ND	ND	ND	ND
Dacthal	μg/kg	22	ND	ND	ND	ND

Table B-5. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
Chlorinated Herbicides (8151) (cont.)		-				Deviation
Dalapon		22	ND	1	 	
Dicambia	μg/kg		ND	ND	ND	ND ND
Dichloroprop [2-(2,4-Dichlorophenoxy)	μg/kg		ND	ND	ND	ND
proponic acid]	μg/kg	22	ND	ND	ND	ND
Dinoseb	μ g/kg	22	ND	ND	ND	ND
MCPA	μg/kg	22	ND	ND	ND	ND
MCPP	μg/kg	22	ND	ND	ND	ND
Pentachiorophenol	μ g/k g		ND	ND	ND	ND
Picloram	μ g/kg	22	ND	ND	ND	ND
Sum Chlorinated Herbicides	μg/kg	22	15	50	19.1	9.3
PAHs (8310)	Far.i.g	1=-		30	13.1	9.3
1-Methylnaphthalene	μg/kg	22	8.5	94	17.86	20.90
2-Methylnaphthalene	μg/kg	22	8.5	180	21.77	37.37
Acenaphthene	μg/kg	22	ND	ND	ND	ND
Acenaphthylene	μg/kg	22	ND	ND	ND	
Anthracene	μg/kg	22	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	22	ND	ND	ND	ND
Benzo(a)pyrene	μg/kg	22	1	6		ND
Benzo(b)fluoranthene	μg/kg	22	ND	ND	1.8	1.5
Benzo(g,h,i)perylene	μg/kg μg/kg	22	ND	ND	ND	ND
Benzo(k)fluoranthene		22	ND		ND	ND
Chrysene	μg/kg	22		ND	ND	ND
Dibenzo(a,h)anthracene	μg/kg		0.9	7.6	1.90	1.88
Fluoranthene	μg/kg	22	ND	ND	ND	ND
Fluorene	μg/kg	22	ND	ND	ND	ND
ndeno(1,2,3-cd)pyrene	μg/kg	22	ND	ND	ND	ND
Naphthalene	μg/kg	22	ND	ND	ND	ND
Phenanthrene	μg/kg	22	8.5	46	15.18	12.67
Pyrene	μg/kg	22	0.9	25	2.53	5.16
Sum of PAHs	μg/kg	22	ND	ND	ND	ND
Metals	μg/kg	22	50.5	343.3	97.89	89.50
Aluminum						
Antimony	mg/kg		26	1200	202.8	306.8
Arsenic	mg/kg	22	ND	ND ND	ND	ND ND
Barium			0.2	8.5	0.63	1.76
Beryllium			ND ND	ND	ND	ND
Cadmium		22	ND ND	ND ND	ND ND	ND
Calcium			ND 20	ND 500	ND	ND
Chromium		22	38	580	150.4	135.1
Cobalt		22	0.5 ND	6 ND	0.84	1.19
Copper			2.5	ND 11	ND 205	ND 0.40
on	mg/kg		2.5	14	3.25	2.43
ead	mg/kg		0.5	1500	240.4	421.5
fagnesium		22	12.5	3.5 65	1.53	0.83
langanese	mg/kg		0.5	7.8	27.96	18.79
lercury	mg/kg		ND	ND ND	2.53 ND	1.83
lickel	mg/kg		ND	ND	ND	ND ND

Table B-5. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						Deviation
Potassium	mg/kg	22	12.5	90	42.61	33.06
Selenium	mg/kg	22	ND	ND	ND	ND
Silver	mg/kg	22	ND	ND	ND	ND
Sodium	mg/kg	22	3.3	22	12.64	5.05
Thallium	mg/kg	22	ND	ND	ND	ND
Vanadium	mg/kg	22	0.5	2	0.72	0.47
Zinc	mg/kg	22	2.5	16	5.24	3.77
Other Parameters						
Bulk density	g/cm ³	22	0.7	1.2	0.99	0.14
Cation Exchange Capacity	meq/ 100g	22	0.5	6	2.93	1.60
Hydrogen ion		22	1.58489E-07	1.99526E-05	6.92E-06	6.65E-06
Percent solids	%	22	71	100	91.0	10.6
Resistivity	ohm-	15	45000	2400000	681466.7	806087.6
•	cm					
Texture (No. 4)	%	22	99.9	100	99.99	0.04
Texture (No. 10)	%	22	99.2	100	99.80	0.20
Texture (No. 40)	%	22	72.9	99.4	96.20	5.69
Texture (No. 60)	%	22	17.3	98 .5	84.9	20.39
Texture (No. 100)	%	22	4.5	64.6	39.12	19.79
Texture (No. 200)	%	22	1.3	6.8	2.87	1.31

Table B-6. Minimum, maximum, mean, and standard deviation for each parameter analyzed in coquina scrub soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)		1		 	 	Deviation
4,4'-DDD	μg/kg	22	ND	ND	ND	ND
4,4'-DDE	μg/kg			ND	ND	ND
4,4'-DDT	μg/kg			ND	ND	ND
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μg/kg		ND	ND	ND	
Aldrin	μg/kg		ND	ND	ND	ND ND
Alpha – BHC	μg/kg		ND	ND	ND	ND
Beta - BHC	μg/kg		ND	ND	ND	
Chlordane (alpha)	μg/kg		ND	ND	ND	ND
Chlordane (gamma)	μg/kg		ND	ND	ND	ND
Chlordane (total)	μg/kg		ND	ND	ND	ND
Delta - BHC	μg/kg		ND	ND	ND	ND
Dieldrin	μg/kg	_	ND	ND ND	ND	ND
Endosulfan I	μg/kg		ND	ND	ND	ND
Endosulfan II (beta)	μg/kg		ND	ND	ND	ND ND
Endosulfan Sulfate	μg/kg	22	ND	ND	ND	ND
Endrin	μg/kg	22	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	22	ND	ND	ND	ND
Endrin Ketone	μg/kg	22	ND	ND	ND	ND
Gamma-BHC (Lindane)	μg/kg	22	ND	ND	ND	ND
Heptachlor	μg/kg	22	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	22	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	22	ND	ND	ND	ND
Isodrin	μg/kg	22	ND	ND	ND ND	ND
Methoxychlor	μg/kg	22	ND	ND ND	ND	ND
Mirex	μg/kg	22	ND	ND	ND	ND
Toxaphene	μg/kg	22	ND	ND	ND	ND
Aroclors (8082)	Paring		110	ND	ND	ND
PCB - 1016/1242	μg/kg	22	ND	ND	ND	AID
PCB - 1221	μg/kg	22	ND	ND	ND	ND ND
PCB - 1232	μg/kg	22	ND	ND	ND	ND ND
PCB - 1248	μg/kg	22	ND	ND	ND	ND ND
PCB - 1254	μg/kg	22	ND	ND	ND	ND ND
PCB - 1260	μg/kg	22	ND	ND	ND	ND
Chlorinated Herbicides (8151)	F 9 · · 9	-	110	- ND	IND	טא
2-(2,4,5 - Trichlorophenoxy) propionic acid	μg/kg	22	ND	ND	ND	ND
2,4,5 - TP) (Silvex)			110	IND	טאו	ND
2,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	22	ND	ND		ND
,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	22	ND ND	ND ND	ND ND	ND
,5 DCBA	μg/kg	22	ND ND		ND ND	ND
-(2,4-Dichlorophenoxy) butyric acid (2,4-DB)	μg/kg	22	ND	ND ND	ND ND	ND
- Nitrophenol	μg/kg	22	ND	ND ND	ND	ND
cifluorfen	μg/kg μg/kg	22	ND ND	ND	ND ND	ND ND
entazon	μg/kg μg/kg	22	ND ND	ND ND	ND ND	ND
Chloramben		22		ND ND	ND ND	ND
Pacthal	μg/kg μα/kg	22	ND ND	ND ND	ND ND	ND
alapon	μg/kg		ND ND	ND ND	ND ND	ND
	μg/kg	22	ND	ND	ND	ND

Table B-6. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)						
Dicambia	μg/kg	22	ND	ND	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	μg/kg	22	ND	ND	ND	ND
Dinoseb	μg/kg	22	ND	ND	ND	ND
MCPA	μg/kg	22	ND	ND	ND	ND
MCPP	μg/kg	22	ND	ND	ND	ND
Pentachiorophenol	μg/kg	22	ND	ND	ND	ND
Picloram	μg/kg	22	ND	ND	ND	ND
Sum Chlorinated Herbicides	μg/kg	22	ND	ND	ND	ND
PAHs (8310)						
1-Methylnaphthalene	µg/kg	22	8.5	75	14.77	16.73
2-Methylnaphthalene	μg/kg	22	8.5	42.5	12.64	10.52
Acenaphthene	μg/kg	22	ND	ND	ND	ND
Acenaphthylene	μg/kg	22	ND	ND	ND	ND
Anthracene	μg/kg	22	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	22	0.9	11	1.68	2.31
Benzo(a)pyrene	μg/kg	22	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/kg	22	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/kg	22	ND	ND	ND	ND
Benzo(k)fluoranthene	µg/kg	22	ND	ND	ND	ND
Chrysene	μg/kg	22	0.9	4.4	1.38	1.20
Dibenzo(a,h)anthracene	μg/kg	22	ND	ND	ND	ND
Fluoranthene	μg/kg	22	1.7	10	2.73	2.57
Fluorene	μg/kg	22	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	22	ND	ND	ND	ND
Naphthalene	μg/kg	22	ND	ND	ND	ND
Phenanthrene	μg/kg	22	ND	ND	ND	ND
Pyrene	μg/kg	22	0.9	4.8	1.40	1.25
Sum of PAHs	μg/kg	22	50.5	250	74.76	61.64
Metals					ļ	
Aluminum	mg/kg	22	75	2500	1001.6	678.2
Antimony	mg/kg	22	ND	ND	ND	ND
Arsenic	mg/kg	21	0.2	2.5	0.31	0.50
Barium	mg/kg		ND	ND	ND	ND
Beryllium	mg/kg	22	ND	ND	ND	ND
Cadmium	mg/kg	22	ND	ND	ND	ND CEOE1 1
Calcium	mg/kg	22	59	260000	22056.8	65351.1
Chromium	mg/kg	22	0.5	5	2.26	1.31
Cobalt	mg/kg	22	ND	ND_	ND	ND
Copper	mg/kg	22	ND	ND	ND 4544.0	ND 000.0
Iron	mg/kg	22	100	3200	1511.8	883.9
Lead	mg/kg	22	1.5	8	4.00	1.56
Magnesium	mg/kg	22	12.5	180	87.61	37.74
Manganese	mg/kg	22	3.4	150	54.19	44.27
Mercury	mg/kg	22	0.005	0.019	0.0080	0.0047
Nickel	mg/kg	22	ND	ND	ND	ND
Potassium	mg/kg	22	12.5	200	74.02	39.26

Table B-6. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
Metals (cont.)		+		+	 	Deviation
Selenium	mg/kg	22	ND	110		
Silver	mg/kg	22	ND	ND	ND	ND
Sodium			<u> </u>	3.2	1.12	0.48
Thallium	mg/kg	22	11	1900	174.0	471.8
Vanadium	mg/kg	22	ND_	ND	ND	ND
Zinc	mg/kg	22	0.5	4.2	2.21	1.04
Other Parameters	mg/kg	22	2.5	12	5.00	3.06
Bulk density	g/cm ³	22	0.9	1.4	1.05	0.16
Cation exchange capacity	meq/	22	1	8	3.8	2.3
	100g			1	0.0	2.5
Hydrogen ion		22	5.01E-09	2E-05	1.77E-06	4.57E-06
Percent solids	%	22	73	99		
Resistivity	ohm-cm	17	730000		96.6	5.5
Texture (No. 4)	%	22		2500000	1712941.2	599591.2
Texture (no. 10)	%		92.4	100	99.29	1.94
Texture (No. 40)		22	68.3	100	97.01	8.09
Texture (no. 60)	%	22	36.4	97.3	63.01	16.50
Fexture (No. 100)	<u>%</u>	22	28.9	91.6	50.90	15.70
Fexture (No. 200)	%	22	13.9	49.7	20.64	7.52
GAIGIE (140. 200)	%	22	1.3	5.8	3.80	1.19

Table B-7. Minimum, maximum, mean, and standard deviation for each parameter analyzed in flatwoods soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)						1 40
4,4'-DDD	µg/kg	21	1.7	8.3	2.31	1.40
4,4'-DDE	μg/kg	21	1.7	5	2.14	0.70
4,4'-DDT	μg/kg	21	ND	ND	ND	ND
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μg/kg	21	5.1	12.3	6.47	1.73
Aldrin	μg/kg	21	ND	ND	ND	ND
Alpha BHC	μg/kg	21	ND	ND	ND	ND
Beta – BHC	μg/kg	21	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	21	ND	ND	ND	ND
Chlordane (gamma)	μg/kg	21	ND	ND	ND	ND
Chlordane (total)	μg/kg	21	ND	ND	ND	ND
Delta – BHC	μg/kg	21	ND	ND_	ND	ND_
Dieldrin	μg/kg	21	ND	ND	ND	ND
Endosulfan I	μg/kg	21	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	21	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	21	ND	ND	ND	ND
Endrin	μg/kg	21	ND_	ND	ND	ND
Endrin Aldehyde	μg/kg	21	ND	ND	ND	ND
Endrin Ketone	μg/kg	21	1.7	6	2.21	0.91
Gamma-BHC (Lindane)	μg/kg	21	ND	ND_	ND	ND
Heptachlor	μg/kg	21	ND	ND	ND	ND_
Heptachlor Epoxide (a)	μg/kg	21	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	21	ND	ND	ND	ND
Isodrin	μg/kg	21	ND	ND	ND	ND
Methoxychlor	μg/kg	21	ND	ND	ND	ND
Mirex	μg/kg	21	ND	ND	ND	ND
Toxaphene	μg/kg	21	ND	ND	ND	ND_
Aroclors (8082)	F 3 3					
PCB - 1016/1242	μg/kg	21	ND	ND	ND	ND
PCB - 1010/12-42	μg/kg	21	ND	ND	ND	ND
PCB - 1232	μg/kg	21	ND	ND	ND	ND
PCB - 1248	μg/kg	21	ND	ND	ND	ND
PCB - 1254	μg/kg	21	ND	ND	ND	ND
PCB - 1260	μg/kg	21	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
2-(2,4,5-Trichlorophenoxy propionic acid (2,4,5 - TP) Silvex	μg/kg	21	ND	ND	ND	ND
2,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	21	ND	ND	ND	ND
2,4-Dichlorophenoxy acetic acid (2,4-D)	μg/kg	21	ND	ND	ND	ND
3,5 DCBA	μg/kg	21	ND	ND	ND	ND_
4-(2,4-Dichlorophenoxy) butyric acid (2,4-DB		21	ND	ND	ND	ND
4 - Nitrophenol	μg/kg	21	ND	ND	ND	ND
Acifluorien	μg/kg	21	ND	ND	ND	ND
Bentazon	μg/kg	21	ND	ND	ND	ND
Chloramben	μg/kg	21	ND	ND	ND	ND
Dacthal	μg/kg	21	ND	ND	ND	ND
Dalapon	μg/kg	21		ND	ND	ND

Table B-7. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
Chlorinated Herbicides (8151) (cont.)					+	Deviation
Dicambia	μg/kg	21	ND	ND	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	μg/kg	21	ND	ND	ND	ND
Dinoseb	μg/kg	21	ND	ND	ND	H NID
MCPA	μg/kg	21	ND	ND	ND	ND
MCPP	μg/kg	21	ND	ND	ND	ND
Pentachlorophenol	μg/kg	21	ND	ND	ND	ND
Picloram	μg/kg	21	ND	ND	ND	ND ND
Sum Chlorinated Herbicides	μg/kg	21	ND	ND	ND	ND ND
PAHs (8310)	73.3		.,,,,	140	i i i i i i i i i i i i i i i i i i i	ND
1-Methylnaphthalene	μg/kg	21	ND	ND	ND	ND
2-Methylnaphthalene	μg/kg	21	ND	ND	ND	ND
Acenaphthene	μg/kg	21	ND	ND	ND	ND
Acenaphthylene	μg/kg	21	ND	ND	ND	ND
Anthracene	μg/kg	21	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	21	0.9	11.5	2.08	3.15
Benzo(a)pyrene	μg/kg	21	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/kg	21	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/kg	21	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/kg	21	ND	ND	ND	ND
Chrysene	μg/kg	21	0.9	11.5	2.22	3.24
Dibenzo(a,h)anthracene	μg/kg	21	ND	ND	ND	ND
Fluoranthene	μg/kg	21	ND	ND	ND	ND
Fluorene	μg/kg	21	ND	ND	ND	ND
ndeno(1,2,3-cd)pyrene	μg/kg	21	ND	ND	ND	ND
Naphthalene	μg/kg	21	8.5	930	74.21	217.34
Phenanthrene	μg/kg	21	ND	ND	ND	ND
Pyrene	μg/kg	21	ND	ND	ND	ND
Sum of PAHs	μg/kg	21	50.5	1494.5	172.24	366.25
Metals						_ 000.20
Aluminum	mg/kg	21	43	360	164.7	84.6
Antimony	mg/kg	21	ND	ND	ND	ND
Arsenic	mg/kg	21	0.2	0.8	0.30	0.14
Barium	mg/kg	21	ND	ND	ND	ND
Beryllium	mg/kg	21	ND	ND	ND	ND
Cadmium	mg/kg	21	ND	ND	ND	ND
Calcium	mg/kg	21	53	780	383.1	211.3
Chromium	mg/kg	21	ND	ND	ND	ND
Cobalt	mg/kg	21	ND	ND	ND	ND
opper	mg/kg	21	1	4	2.8	0.7
on	mg/kg	21	36	320	144.1	81.3
ead	mg/kg	21	0.5	12	2.44	2.45
lagnesium	mg/kg	21	12.5	270	112.31	60.51
langanese	mg/kg	21	0.5	3	0.93	0.80
lercury	mg/kg	21	0.005	0.017	0.0065	0.0031
ickel	mg/kg	21	ND	ND	ND	ND
otassium	mg/kg	21	12.5	140	57.00	41.80

Table B-7. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
						Deviation
Metals (cont.)						
Selenium	mg/kg	21	ND	ND	ND	ND
Silver	mg/kg	21	ND	ND	ND	ND
Sodium	mg/kg	21	12.5	140	48.69	34.63
Thallium	mg/kg	21	ND	ND	ND	ND
Vanadium	mg/kg	21	0.5	2.9	0.79	0.60
Zinc	mg/kg	21	2.5	28	7.70	5.90
Other Parameters						
Bulk density	g/cm ³	21	0.4	1	0.72	0.15
Cation exchange capacity	meq/	21	4	28	17.8	7.4
,	100g	1				
Hydrogen ion		21	2.51189E-08	7.94328E-05	3.57E-05	2.48E-05
Percent solids	%	21	59	98	84.4	9.9
Resistivity	ohm-cm	20	27000	1500000	473600.0	368518.7
Texture (No. 4)	%	21	99.4	100	99.94	0.14
Texture (no. 10)	%	21	98.4	99.9	99.41	0.46
Texture (No. 40)	%	21	67.2	98.7	89.68	9.42
Texture (no. 60)	%	21	12.2	97.1	73.16	26.71
Texture (No. 100)	%	21	3.2	72.4	33.74	19.82
Texture (No. 200)	%	21	2.4	8.9	5.48	1.74

Table B-8. Minimum, maximum, mean, and standard deviation for each parameter analyzed in hammock soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)				 		Deviation
4,4'-DDD	μg/kg	20	ND	ND	ND	ND
4,4'-DDE	μg/kg	20	ND	ND	ND	
4,4'-DDT	μg/kg	20	ND	ND	ND	ND
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μg/kg	20	ND	ND	ND	ND
Aldrin	μg/kg	20	ND	ND		ND
Alpha - BHC	μg/kg	20	ND	ND	ND	ND
Beta – BHC	μg/kg	20	ND	ND	ND ND	ND
Chlordane (alpha)	μg/kg	20	ND	ND		ND
Chlordane (gamma)	μg/kg	20	ND	ND	ND	ND
Chlordane (total)	μg/kg	20	ND	ND	ND	ND
Delta – BHC	μg/kg	20	ND	ND	ND	ND
Dieldrin	μg/kg	20	ND	ND	ND	ND
Endosulfan I	μg/kg	20	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	20	ND		ND	ND ND
Endosulfan Sulfate	μg/kg	20	ND	ND	ND	ND
Endrin	μg/kg	20	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	20	ND	ND ND	ND	ND ND
Endrin Ketone	μg/kg	20	ND	ND	ND	ND
Gamma-BHC (Lindane)	μg/kg	20	ND	ND ND	ND	ND
Heptachlor				ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	20	ND ND	ND NO	ND	ND
Heptachlor Epoxide (b)	μg/kg	20	ND ND	ND	ND ND	ND
Isodrin	μg/kg	20	ND	ND	ND	ND
Methoxychlor	μg/kg	20	ND ND	ND ND	ND	ND
Mirex	μg/kg	20	ND ND	ND ND	ND	ND
Toxaphene	μg/kg	20	ND ND	ND I	ND	ND
Aroclors (8082)	μg/kg	20	ND	ND	ND	ND
PCB - 1016/1242	ua/ka	100	- ND	- 110		
PCB - 1221	μg/kg	20	ND ND	ND ND	ND	ND
PCB - 1232	μg/kg	20	ND	ND ND	ND	ND
PCB - 1248	μg/kg	20	ND	ND	ND	ND
PCB - 1254	μg/kg	20	ND	ND NO	ND	ND
PCB - 1260	μg/kg	20	ND	ND	ND	ND
Chlorinated Herbicides (8151)	μg/kg	20	ND	ND	ND	ND
2-(2,4,5–Trichlorophenoxy)propionic acid	ug/kg	20				
(2,4,5 – TP) Silvex	μg/kg	20	5	23	8.2	5.2
	 					
2,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	20	ND	ND	ND	ND
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	20	ND	ND	ND	ND
3,5 DCBA	μg/kg	20	ND	ND	ND	ND
I-(2,4-Dichlorophenoxy) butyric acid (2,4-DB)	μg/kg	20	ND	ND	ND	ND
I – Nitrophenol	μg/kg	20	ND	ND	ND	ND
Acifluorfen	μg/kg	20	ND	ND	ND	ND
Bentazon	μg/kg	20	ND	ND	ND	ND
Chloramben	μg/kg	20	ND	ND	ND	ND
Dacthal	μg/kg	20	ND	ND	ND	ND
Dalapon	μg/kg	20	ND	ND	ND	ND

Table B-8. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)	 	1				Doviduori
Dicamba	μg/kg	20	ND	ND	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy)	μg/kg	20	ND	ND	ND	ND
proponic acid]	75.5					
Dinoseb	μg/kg	20	ND	ND	ND	ND
MCPA	μg/kg	20	ND	ND	ND	ND
MCPP	μg/kg	20	ND	ND	ND	ND
Pentachlorophenol	μg/kg	20	ND	ND	ND	ND
Picloram	μg/kg	20	ND	ND	ND	ND
Sum Chlorinated Herbicides	μg/kg	20	15	67.5	22.90	11.83
PAHs (8310)						
1-Methylnaphthalene	μg/kg	20	8.5	38	13.60	7.77
2-Methylnaphthalene	μg/kg	20	ND	ND	ND	ND
Acenaphthene	μg/kg	20	ND	ND	ND	ND
Acenaphthylene	μg/kg	20	ND	ND	ND	ND
Anthracene	μg/kg	20	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	20	0.9	12	1.92	2.50
Benzo(a)pyrene	μg/kg	20	1	8	1.8	1.7
Benzo(b)fluoranthene	μg/kg	20	1.5	10	3.07	2.39
Benzo(g,h,i)perylene	μg/kg	20	1.8	26	4.80	7.03
Benzo(k)fluoranthene	μg/kg	20	1	11	2.7	3.0
Chrysene	μg/kg	20	0.9	8.1	1.88	1.95
Dibenzo(a,h)anthracene	μg/kg	20	ND	ND	ND	ND
Fluoranthene	μg/kg	20	1.8	180	13.23	39.56
Fluorene	μg/kg	20	1.8	170	10.91	37.47
Indeno(1,2,3-cd)pyrene	μg/kg	20	0.9	9	1.88	2.03
Naphthalene	μg/kg	20	8.5	38	13.20	6.94
Phenanthrene	μg/kg	20	0.9	88	5.58	19.41
Pyrene	μg/kg	20	0.9	320	21.67	71.40
Sum of PAHs	<i>µ</i> g/kg	20	51.4	457	126.36	112.71
Metals						
Aluminum	mg/kg	20	220	9900	2395.0	2382.4
Antimony	mg/kg	20	ND	ND	ND	ND
Arsenic	mg/kg	20	0.2	1.6	0.44	0.37
Barium	mg/kg	20	ND	ND	ND	ND
Beryllium	mg/kg	20	ND	ND	ND	ND
Cadmium	mg/kg	20	ND	ND	ND	ND
Calcium	mg/kg	20	320	88000	11278.0	20934.6
Chromium	mg/kg	20	0.5	13	3.47	3.22
Cobalt	mg/kg	20	ND	ND	ND	ND
Copper	mg/kg	20	2.5	18	4.33	3.77
Iron	mg/kg	20	200	6400	1717.5	1716.3
Lead	mg/kg	20	2.5	17	6.24	4.52
Magnesium	mg/kg	20	69	3800	534.1	836.0
Manganese	mg/kg	20	1	79	14.7	17.5
Mercury	mg/kg	20	0.005	0.19	0.0396	0.0523
Nickel	mg/kg	20	ND	ND	ND	ND
Potassium	mg/kg	20	15	530	114.6	113.8

Table B-8. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
84-4-1- (Deviation
Metals (cont.)						
Selenium	mg/kg	20	1	4.5	1.53	0.98
Silver	mg/kg	20	ND	ND	ND	ND ND
Sodium	mg/kg	20	15	880	168.5	227.8
Thallium	mg/kg	20	ND	ND	ND	ND
Vanadium	mg/kg	20	0.5	24	5.16	5.49
Zinc	mg/kg	20	2.5	110	12.77	23.67
Other Parameters				110	12.77	23.07
Bulk density	g/cm ³	20	0.3	1.1	0.74	0.24
Cation exchange capacity (as Na)	meg/	20	4	71	23.5	18.6
	100g	_	·	, ,	23.3	10.0
Hydrogen ion		20	2.51189F-08	2.51189E-04	1.31E-05	5.60E-05
Percent solids	%	20	22	96	75.3	
Resistivity	ohm-cm	20	4600	3100000	440645.0	18.8
Texture (No. 4)	%	20	98.5	100		866899.7
Texture (no. 10)	%	20	97.7	99.9	99.83	0.40
Texture (No. 40)	%	20	85.2		99.45	0.64
Texture (no. 60)	%	20	63.3	98.8	93.92	4.49
Texture (No. 100)	%	20	21	95.7	84.95	10.36
Texture (No. 200)	%	20		65.3	45.79	12.68
	/6	20	6.6	28.8	12.45	7.07

Table B-9. Minimum, maximum, mean, and standard deviation for each parameter analyzed in freshwater wetland soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)						
4,4'-DDD	μg/kg	22	ND	ND	ND	ND
4,4'-DDE	μg/kg	22	ND	ND	ND	ND
4,4'-DDT	μg/kg	22	ND	ND	ND	ND
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μg/kg	22	ND	ND	ND	ND
Aldrin	μg/kg	22	ND	ND	ND	ND
Alpha - BHC	μg/kg	22	ND	ND	ND	ND
Beta - BHC	μg/kg	22	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	22	ND	ND	ND	ND
Chlordane (gamma)	μg/kg	22	ND	ND	ND	ND
Chlordane (total)	μg/kg	22	ND	ND	ND	ND
Delta – BHC	μg/kg	22	ND	ND	ND	ND
Dieldrin	μg/kg	22	ND	ND	ND	ND
Endosulfan I	μg/kg	22	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	22	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	22	ND	ND	ND	ND_
Endrin	μg/kg	22	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	22	ND	ND	ND	ND
Endrin Ketone	μg/kg	22	ND	ND	ND	ND
Gamma-BHC (Lindane)	μg/kg	22	ND	ND	ND	ND
Heptachlor	μg/kg	22	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	22	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	22	ND	ND	ND	ND
Isodrin	μg/kg	22	ND	ND	ND	ND
Methoxychlor	μg/kg	22	ND	ND	ND	ND
Mirex	μg/kg	22	ND	ND	ND	ND
Toxaphene	μg/kg	22	ND	ND	ND	ND
Aroclors (8082)	Fana					
PCB - 1016/1242	μg/kg	22	ND	ND	ND	ND
PCB - 1221	μg/kg	22	ND	ND	ND	ND
PCB - 1232	μg/kg	22	ND	ND	ND	ND
PCB - 1248	μg/kg	22	ND	ND	ND	ND
PCB - 1254	μg/kg	22	ND	ND	ND	ND
PCB - 1260	μg/kg	22	ND	ND	ND	ND
Chlorinated Herbicides (8151)	7.3.3					
2-(2,4,5 - Trichlorophenoxy) propionic acid (2,4,5 - TP) (Silvex)	µg/kg	22	ND	ND	ND	ND
2,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	22	ND	ND	ND	ND
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	22	ND	ND	ND	ND
3.5 DCBA	μg/kg	22	ND	ND	ND	ND
4-(2,4-Dichlorophenoxy) butyric acid (2,4-DB)	μg/kg	22	ND	ND	ND	ND
4 – Nitrophenol	μg/kg	22	ND	ND	ND	ND
Acifluorfen	μg/kg	22	ND	ND	ND	ND
Bentazon	μg/kg	22	ND	ND	ND	ND
Chloramben	μg/kg	22	ND :	ND	ND	ND
Dacthal	μg/kg	22	ND	ND	ND	ND
Dalapon	μg/kg	22	ND	ND	ND	ND

Table B-9. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)	- 			†	 	Deviation
Dicamba	μg/kg	22	ND	ND	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	μg/kg		ND	ND	ND	ND
Dinoseb	μg/kg	22	ND	ND	ND	ND
MCPA	μg/kg		ND	ND	ND	ND
MCPP	μg/kg		ND	ND	ND	ND
Pentachlorophenol	μg/kg	22	ND	ND	ND	ND
Picloram	μg/kg	22	ND	ND	ND	ND
Sum Chlorinated Herbicides	μg/kg	22	ND	ND	ND	ND
PAHs (8310)	Fang	+	,,,,,	100	IND	ND
1-Methylnaphthalene	μg/kg	22	ND	ND	ND	ND
2-Methylnaphthalene	μg/kg	22	ND	ND	ND	ND
Acenaphthene	μg/kg	22	ND	ND	ND	ND
Acenaphthylene	μg/kg	22	ND	ND	ND	ND
Anthracene	μg/kg	22	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	22	ND	ND	ND	ND
Benzo(a)pyrene	μg/kg	22	1	22	2.4	4.4
Benzo(b)fluoranthene	μg/kg	22	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/kg	22	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/kg	22	ND	ND	ND	ND
Chrysene	μg/kg	22	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/kg	22	ND	ND	ND	ND
Fluoranthene	μg/kg	22	ND	ND	ND	ND
Fluorene	μg/kg	22	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	22	ND	ND	ND	ND
Naphthalene	μg/kg	22	ND	ND	ND	ND ND
Phenanthrene	μg/kg	22	ND	ND	ND	ND ND
Pyrene	μg/kg	22	ND	ND	ND	ND
Sum of PAHs	μg/kg	22	50.5	172	78.93	32.00
Metals	F 3 3			17,2	70.33	32.00
Aluminum	mg/kg	22	110	12000	2315.9	3148.0
Antimony	mg/kg	22	ND	ND	ND	ND
Arsenic	mg/kg	22	0.2	3	0.55	0.62
3arium	ma/ka		ND	ND	ND	ND
Beryllium	mg/kg	22	ND	ND	ND	ND
Cadmium	mg/kg	22	ND	ND	ND	ND
Calcium	mg/kg	22	150	140000	9609.5	29489.2
Chromium	mg/kg	22	0.5	18	3.07	4.34
Cobalt	mg/kg	22	ND	ND	ND	ND
Copper	mg/kg	22	0.5	25	4.70	4.84
ron	mg/kg	22	70	8700	1326.1	2361.4
ead	mg/kg	22	0.5	23	5.06	4.91
Magnesium	mg/kg	22	52	2100	316.0	468.5
Manganese	mg/kg	22	0.5	94	9.61	20.41
Mercury	mg/kg	22	0.005	0.16	0.0375	0.0471
lickel	mg/kg	22	ND	ND ND	ND	ND
Potassium	mg/kg	22	14	250	95.2	65.6

Table B-9. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Selenium	mg/kg	22	1	6	1.7	1.1
Silver	mg/kg	22	ND	ND	ND	ND
Sodium	mg/kg	22	21	1400	169.4	311.2
Thallium	mg/kg	22	ND	ND	ND	ND
Vanadium	mg/kg	22	0.5	26	3.65	5.64
Zinc	mg/kg	22	3	48	13.6	11.1
Other Parameters						
Bulk density	g/cm ³	22	0.2	1.1	0.72	0.26
Cation exchange capacity	meq/	22	0.7	8 5	28.7	19.7
3 ,	100g					
Hydrogen ion		22	1.99526E-08	6.30957E-05	1.47E-05	2.16E-05
Percent solids	%	22	31	98	69.9	19.5
Resistivity	ohm-cm	22	2000	870000	151027.3	265840.7
Texture (No. 4)	%	22	98.7	100	99.86	0.33
Texture (no. 10)	%	22	87.8	100	98.43	2.74
Texture (No. 40)	%	22	63.3	99.1	87.31	11.85
Texture (no. 60)	%	22	19.6	97.1	74.12	23.72
Texture (No. 100)	%	22	7.2	74	42.91	21.72
Texture (No. 200)	%	22	1.2	37.6	11.70	9.52

Table B-10. Minimum, maximum, mean, and standard deviation for each parameter analyzed in saltwater wetland soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std.
Organochlorine Pesticides (8081)		+	 -	+		Deviation
4,4'-DDD	μg/kg	21	1.9	96	7.00	
4,4'-DDE	μg/kg		1.9	30	7.89	20.31
4,4'-DDT	μg/kg		1.9	96	5.30	6.91
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μg/kg		5.7	222	7.89	20.31
Aldrin	μg/kg		ND	ND	21.08	46.72
Alpha - BHC	μg/kg		ND	ND	ND	ND
Beta - BHC	μg/kg	_	ND		ND	ND
Chlordane (alpha)	μg/kg		ND	ND	ND	ND
Chlordane (gamma)	μg/kg		ND ND	ND	ND	ND
Chlordane (total)	μg/kg		ND	ND	ND	- ND
Delta - BHC	μg/kg			ND	ND	ND
Dieldrin	μg/kg		ND	ND	ND	ND
Endosulfan I	μg/kg	21	1.9	10.5	3.94	2.53
Endosulfan II (beta)		21	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	_	ND	ND	ND	ND
Endrin	μg/kg	21	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	21	ND	ND	ND	ND
Endrin Ketone	μg/kg	21	ND	ND	ND	ND
Gamma-BHC (Lindane)	μg/kg	21	ND	ND	ND	ND
Heptachlor	μg/kg	21	1.9	14	3.99	3.17
Heptachlor Epoxide (a)	μg/kg	21	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	21	ND	ND	ND	ND
Isodrin	μg/kg	21	ND	ND	ND	ND
Methoxychlor	μg/kg	21	ND	ND	ND	ND
Mirex	μg/kg	21	ND	ND	ND	ND
Toxaphene	µg/kg	21	ND	ND	ND	ND
Aroclors (8082)	µg/kg	21	ND	ND	ND	ND
PCB - 1016/1242						
PCB - 1221	μg/kg	21	ND	ND	ND	ND
PCB - 1232	μg/kg	21	ND	ND	ND	ND
PCB - 1248	μg/kg	21	ND	ND	ND	ND
PCB - 1254	μg/kg	21	ND	ND	ND	ND
PCB - 1260	µg/kg	21	ND	ND	ND	ND
	μg/kg	21	ND ND	ND	ND	ND
Chlorinated Herbicides (8151)						
2-(2,4,5 - Trichlorophenoxy) propionic acid 2,4,5 - TP) (Silvex)	µg/kg	21	ND	ND	ND	ND
				1		
2,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	21	ND	ND	ND	ND
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	21	6	62	17.6	15.3
3,5 DCBA	µg/kg	21	ND	ND	ND	ND
-(2,4-Dichlorophenoxy) butyric acid (2,4-DB)	μg/kg	21	ND	ND	ND	ND
Nitrophenol	µg/kg	21	ND	ND	ND	ND
cifluorfen	μg/kg	21	ND	ND	ND	ND
Bentazon	μg/kg	21	ND	ND	ND	ND
Chloramben	μg/kg	21	ND	ND	ND	ND
Pacthal	μg/kg	21	ND	ND	ND	ND
alapon	μg/kg	21	ND	ND	ND	ND

Table B-10. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)						
Dicamba	µg/kg	21	ND	ND	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	µg/kg	21	ND	ND	ND	ND
Dinoseb	μg/kg	21	ND	ND	ND	ND
MCPA	μg/kg	21	ND	ND	ND	ND
MCPP	μg/kg	21	ND	ND	ND	ND
Pentachlorophenol	μg/kg	21	ND	ND	ND	ND
Picloram	μg/kg	21	ND	ND	ND	ND
Sum Chlorinated Herbicides	μg/kg	21	18	155	46.3	35.1
PAHs (8310)						
1-Methylnaphthalene	μg/kg	21	ND	ND	ND	ND
2-Methylnaphthalene	μg/kg	21	ND	ND	ND	ND
Acenaphthene	μg/kg	21	ND	ND	ND	ND
Acenaphthylene	μg/kg	21	ND	ND	ND	ND
Anthracene	μg/kg	21	ND	ND	ND	ND_
Benzo(a)anthracene	μg/kg	21	ND	ND	ND	ND
Benzo(a)pyrene	μg/kg	21	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/kg	21	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/kg	21	ND	ND	ND	ND_
Benzo(k)fluoranthene	μg/kg	21	ND	ND	ND	ND
Chrysene	μg/kg	21	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/kg	21	ND	ND	ND	ND_
Fluoranthene	μg/kg	21	ND	ND	ND	ND
Fluorene	μg/kg	21	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	21	ND	ND	ND	ND
Naphthalene	μg/kg	21	ND	ND	ND	ND
Phenanthrene	μg/kg	21	ND	ND	ND	ND
Pyrene	μg/kg	21	ND	ND	ND	ND
Sum of PAHs	μg/kg	21	ND	ND	ND	ND
Metals						
Aluminum	mg/kg	21	660	9100	2683.8	1997.6
Antimony	mg/kg	21	ND	ND	ND	ND
Arsenic	mg/kg	21	0.3	4.1	1.32	1.08
Barium	mg/kg	21	ND	ND	ND	ND
Beryllium	mg/kg		ND	ND	ND	ND
Cadmium	mg/kg	21	ND	ND	ND	ND
Calcium	mg/kg	21	380	67000	11627.1	18373.1
Chromium	mg/kg	21	0.5	15	4.51	3.28
Cobalt	mg/kg		ND	ND	ND	ND ND
Copper	mg/kg		ND	ND	ND	ND 1001.0
Iron	mg/kg		250	3800	1504.8	1091.3
Lead	mg/kg		1	150	16.6	33.8
Magnesium	mg/kg		490	11000	3291.0	3151.2
Manganese	mg/kg		2	42	10.1	10.5
Mercury	mg/kg	_	0.005	0.068	0.0179	0.0162
Nickel	mg/kg		ND	ND	ND	ND 1000 F
Potassium	mg/kg	21	110	6200_	1631.0	1620.5

Table B-10. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
Metals (cont.)		—				Deviation
Selenium		 				
Silver	mg/kg	21	ND	ND	ND	ND
Sodium	mg/kg	21	ND	ND	ND	ND
Thallium	mg/kg	21	1400	67000	13771.4	18366.1
	mg/kg	21	ND	ND	ND	ND
Vanadium	mg/kg	21	1.6	28	7.35	6.30
Zinc	mg/kg	21	3	19	7.7	
Other Parameters					/./	4.8
Bulk density	g/cm ³	21	0.2	1.3	0.74	
Cation exchange capacity	meq/	21	6	35	0.71	0.33
	100g	١-٠	J	33	18.3	9.0
Hydrogen ion	1.008	20	1.58489E-09	0.511005.00	1.000	
Percent solids	%	21		2.51189E-06	4.65E-07	7.60E-07
Resistivity	ohm-cm	_	16	86	56.9	20.7
Texture (No. 4)		21	26	790	172.7	183.6
Texture (no. 10)	%	21	93.1	100	99.45	1.50
Texture (No. 40)	%	21	71.9	100	96.55	6.25
Texture (no. 60)	%	21	53.9	98.7	88.08	11.53
Texture (No. 100)	%	21	50.2	95.1	81.22	12.26
	%	21	35.1	80.2	51.94	12.65
Fexture (No. 200)	%	21	5.8	77	22.72	19.16

Table B-11. Minimum, maximum, mean, and standard deviation for each parameter analyzed in citrus scrub soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)						
4,4'-DDD	μg/kg	23	ND	ND	ND	ND
4,4'-DDE	μg/kg	23	1.7	87	14.10	22.86
4,4'-DDT	μg/kg	23_	1.7	86	11.96	22.47
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μg/kg	23	5.1	174.7	27.82	45.08
Aldrin	μg/kg	23	ND	ND	ND	ND
Alpha - BHC	μg/kg	23	ND	ND	ND	ND
Beta - BHC	μg/kg	23	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	23	1.7	7.4	2.0	1.2
Chlordane (gamma)	μg/kg	23	1.7	5.0	1.9	0.7
Chlordane (total)	μg/kg	23	1.7	62	4.37	12.56
Delta - BHC	μg/kg	23	ND	ND	ND	ND
Dieldrin	μg/kg	23	1.7	160	13.90	36.10
Endosulfan I	μg/kg	23	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	23	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	23	ND	ND	ND	ND
Endrin	μg/kg	23	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	23	ND	ND	ND	ND
Endrin Ketone	μg/kg	23	ND	ND	ND	ND
Gamma-BHC (Lindane)	μg/kg	23	ND	ND	ND	ND
Heptachlor	μg/kg	23	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	23	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	23	ND	ND	ND	ND
Isodrin	μg/kg	23	ND	ND	ND	ND
Methoxychlor	μg/kg	23	ND	ND	ND	ND
Mirex	μg/kg	23	ND	ND	ND	ND
	μg/kg	23	ND	ND	ND	ND
Toxaphene Aroclors (8082)	Pg/Ng	20	145	1		
	μg/kg	23	ND	ND	ND	ND
PCB - 1016/1242	μg/kg μg/kg	23	ND	ND	ND	ND
PCB - 1221	μg/kg	23	ND	ND	ND	ND
PCB - 1232	μg/kg	23	ND	ND	ND	ND
PCB - 1248	μg/kg	23	ND	ND	ND	ND
PCB - 1254	μg/kg	23	ND	ND	ND	ND
PCB – 1260 Chlorinated Herbicides (8151)	μ _g /κg	23	140			
2-(2,4,5 - Trichlorophenoxy) propionic acid (2,4,5 - TP) (Silvex)	μg/kg	23	ND	ND	ND	ND
2,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	23	ND	ND	ND	ND
2,4,5-1 ichlorophenoxy acetic acid (2,4 - D)	μg/kg	23	ND	ND	ND	ND
3,5 DCBA	μg/kg	23	ND	ND	ND	ND
4-(2,4-Dichlorophenoxy) butyric acid (2,4-DB		23	5	50	9.4	10.9
4 - Nitrophenol	μg/kg	23	ND	ND	ND	ND
Acifluorfen	μg/kg	23	ND	ND	ND	ND
Bentazon	μg/kg		ND	ND	ND	ND
Chloramben	μg/kg		ND	ND	ND	ND
	μg/kg	_		ND	ND	ND
Dacthal Dalapon	μg/kg			ND	ND	ND

Table B-11. (cont.).

Parameter	Unit	is N	Minimun	n Maximum	Mean	Std.
Chlorinated Herbicides (8151) (cont.)					 -	Deviation
Dicamba	μg/k	g 2:	3 ND	ND	ND	
Dichloroprop [2-(2,4-Dichlorophenoxy)	μg/k			ND	ND ND	ND ND
proponic acid	1.0			110	שא	ND
Dinoseb	μg/k	g 23	3 ND	ND		
MCPA	μg/k			ND	ND ND	ND
MCPP	μg/kg			ND	ND	ND
Pentachlorophenol	μg/kg			ND	ND	ND
Picloram	μg/kg			ND	ND	ND
Sum Chlorinated Herbicides	μg/kg			ND 150	ND	ND
PAHs (8310)	- Paring	1 20	15	150	23.8	28.3
1-Methylnaphthalene	μg/kg	23	ND	ND.	<u> </u>	
2-Methylnaphthalene	μg/kg			ND	ND	ND
Acenaphthene	μg/kg			170	20.61	36.74
Acenaphthylene	μg/kg			ND	ND ND	ND
Anthracene	μg/kg		8.5	22	3.58	5.21
Benzo(a)anthracene	μg/kg			85	14.24	17.05
Benzo(a)pyrene	μg/kg		0.9	9.3	1.93	2.36
Benzo(b)fluoranthene	μg/kg			9	2.5	2.6
Benzo(g,h,i)perylene	μg/kg		1.5	16.5	2.94	3.45
Benzo(k)fluoranthene	μg/kg		1.7	18	3.45	4.65
Chrysene	μg/kg		1	14	2.80	3.59
Dibenzo(a,h)anthracene	μg/kg		0.9	14	2.85	3.23
luoranthene	μg/kg		ND	ND	ND	ND
luorene	μg/kg	23	1.7	18	3.45	4.65
ndeno(1,2,3-cd)pyrene	μg/kg	23	ND 0.0	ND	ND	ND
laphthalene	μg/kg	23	0.9	17	2.10	3.66
henanthrene	μg/kg	23	8.5	830	107.89	209.08
yrene	μg/kg	23	0.9	230	11.03	47.74
um of PAHs	μg/kg	23	0.9	8.5	1.50	1.73
Metals	Purky	23	50.5	1467	199.90	310.91
luminum	maka	22	000	1000		
ntimony	mg/kg mg/kg	23	200	4200	1172.2	919.4
rsenic	mg/kg	23	ND	ND	ND	ND
arium		23	0.2	1	0.36	0.27
eryllium	mg/kg		ND ND	ND	ND_	ND
admium	mg/kg	23	ND OF	ND	ND	ND
alcium	mg/kg	23	0.5	2.2	0.64	0.41
nromium	mg/kg	23	170	15000	2100.9	3060.3
balt	mg/kg	23	3.7	34	12.18	7.81
pper	mg/kg	23	ND	ND	ND_	ND
n	mg/kg	23	7	130	53.5	36.4
ad	mg/kg	23	270	3500	1502.2	900.4
agnesium	mg/kg	23	2.4	19	9.70	4.82
anganese	mg/kg	23	68	760	232.6	189.3
proury		23	10	200	82.0	52.1
kal		23	0.005	0.046	0.0120	0.0116
taccium		23	ND	ND	ND	ND
	mg/kg	23	12.5	560	94.57	110.71

Table B-11. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Selenium	mg/kg	23	ND	ND	ND	ND
Silver	mg/kg	23	ND	ND	ND	ND
Sodium	mg/kg	23	3.8	110	17.01	21.62
Thallium	mg/kg	23	0.5	2.6	0.59	0.44
Vanadium	mg/kg	23	0.5	4.8	2.02	1.10
Zinc	mg/kg	23	7.2	140	49.23	36.29
Other Parameters						
Bulk density	g/cm ³	23	0.8	1.4	1.13	0.15
Cation exchange capacity	meq/	17	4.0	55.0	19.4	17.2
	100g					
Hydrogen ion		23	5.01187E-08	1.58489E-06	3.42E-07	3.93E-07
Percent solids	%	23	78	99	95.4	6.9
Resistivity	ohm-cm	23	10000	2500000	870130.4	855419.3
Texture (No. 4)	%	23	99.4	100	99.96	0.13
Texture (no. 10)	%	23	97.8	100	99.78	0.49
Texture (No. 40)	%	23	52.8	99.3	83.14	16.24
Texture (no. 60)	%	23	43.1	97.8	73.95	20.44
Texture (No. 100)	%	23	14.8	46.9	29.13	11.50
Texture (No. 200)	%	23	2	8.4	3.70	1.60

Table B-12. Minimum, maximum, mean, and standard deviation for each parameter analyzed in citrus hammock soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std.
Organochlorine Pesticides (8081)						Deviation
4,4'-DDD	μg/kg	23	ND	ND -		
4,4'-DDE	μg/kg	23	1.7	ND	ND	ND
4,4'-DDT	μg/kg	23	1.7	67	5.39	13.63
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μg/kg	23	5.1	5.4	1.94	0.77
Aldrin	μg/kg	23	ND	70.4	9.12	13.64
Alpha - BHC	μg/kg	23	ND	ND	ND	ND
Beta - BHC	μg/kg	23	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	23	ND	ND ND	ND	ND
Chlordane (gamma)	μg/kg	23		ND ND	ND	ND
Chlordane (total)	μg/kg μg/kg	23	ND	ND	ND	ND
Delta - BHC	μg/kg	23	ND	ND	ND	ND
Dieldrin			ND 17	ND	ND	ND
Endosulfan i	μg/kg	23	1.7	190	9.97	39.25
Endosulfan II (beta)	μg/kg	23	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	23	ND	ND	ND	ND
Endrin	μg/kg	23	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	23	ND	ND	ND	ND
Endrin Ketone	μg/kg	23	ND	ND	ND	ND
Gamma-BHC (Lindane)	μg/kg	23	ND	ND	ND	ND
Heptachlor	μg/kg	23	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	23	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	23	ND	ND	ND	ND
Isodrin	μg/kg	23	ND	ND	ND	ND
Methoxychlor	μg/kg	23	ND	ND	ND	ND
Mirex	_µg/kg	23	ND	ND	ND	ND
Toxaphene	μg/kg	23	ND	ND	ND	ND
Aroclors (8082)	µg/kg	23	ND	ND	ND	ND
PCB - 1016/1242	 -	1				
PCB - 1221	μg/kg	23	ND	ND	ND	ND
PCB - 1232	μg/kg	23	ND	ND	ND	ND
PCB - 1248	μg/kg	23	ND	ND	ND	ND
PCB - 1254	μg/kg	23	ND	ND	ND	ND
PCB - 1260	µg/kg	23	ND	ND	ND	ND
		lacksquare				
Chlorinated Herbicides (8151)	<u> </u>					
2-(2,4,5-Trichlorophenoxy) propionic acid 2,4,5 - TP) (Silvex)	μg/kg	23	5	50	9.6	12.3
2,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	23	ND	ND	ND	ND
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	23	5	50	9.1	11.1
3,5 DCBA	μg/kg	23	ND	ND	ND	ND
-(2,4-Dichlorophenoxy) butyric acid (2,4-DB)	μg/kg	23	ND	ND	ND	ND
- Nitrophenol	µg/kg	23	ND	ND	ND	ND
cifluorfen	μg/kg	23	ND	ND	ND	ND
Bentazon	µg/kg	23	ND	ND	ND	ND
Chloramben	μg/kg	23	ND	ND	ND	ND
Pacthal	µg/kg	23	ND	ND	ND	ND
Palapon	μg/kg	23	ND	ND	ND	ND

Table B-12. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)						
Dicamba	μg/kg	23	ND	ND	ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	μg/kg	23	ND	ND	ND	ND
Dinoseb	μg/kg	23	ND	ND	ND	ND
MCPA	μg/kg	23	ND	ND	ND	ND
MCPP	μg/kg	23	ND	ND	ND	ND
Pentachlorophenol	μg/kg	23	ND	ND	ND	ND
Picloram	μg/kg	23	ND	ND	ND	ND
Sum Chlorinated Herbicides	μg/kg	23	15	150	26.0	31.0
PAHs (8310)						
1-Methylnaphthalene	μg/kg	23	8.5	20	9.39	2.44
2-Methylnaphthalene	μg/kg	23	8.5	42	10.35	6.94
Acenaphthene	μg/kg	23	ND	ND	ND	ND
Acenaphthylene	μg/kg	23	ND	ND	ND	ND
Anthracene	μg/kg	23	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	23	0.9	2.8	1.08	0.49
Benzo(a)pyrene	μg/kg	23	1	5	1.4	1.1
Benzo(b)fluoranthene	μg/kg	23	1.5	7	2.00	1.31
Benzo(g,h,i)perylene	μg/kg	23	1.7	13	3.05	3.42
Benzo(k)fluoranthene	µg/kg	23	1	3	1.1	0.4
Chrysene	μg/kg	23	0.9	3.7	1.06	0.58
Dibenzo(a,h)anthracene	μg/kg	23	ND	ND_	ND_	ND
Fluoranthene	µg/kg	23	ND	ND	ND	ND
Fluorene	µg/kg	23	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	23	0.9	15	2.14	3.45
Naphthalene	μg/kg	23	8.5	100	18.61	25.90
Phenanthrene	μg/kg	23	ND	ND	ND	ND
Pyrene	μg/kg	23	ND	ND	ND	ND
Sum of PAHs	μg/kg	23	50.5	164	68.33	32.10
Metais						
Aluminum	mg/kg	23	130	7200	1821.3	1768.2
Antimony	mg/kg	23	ND	ND	ND	ND
Arsenic	mg/kg	23	0.2	2.7	0.53	0.74
Barium	mg/kg	23	ND	ND	ND	ND
Beryllium	mg/kg	23	ND	ND	ND	ND_
Cadmium	mg/kg	23	ND	ND	ND	ND
Calcium	mg/kg	23	210	99000	11099.1	24043.0
Chromium	mg/kg	23	0.5	30	11.79	9.50
Cobalt	mg/kg	23	ND	ND	ND	ND
Copper	mg/kg	23	2.5	130	54.04	40.63
Iron	mg/kg	23	100	8000	1984.8	2368.4
Lead	mg/kg	23	0.5	17	5.84	3.83
Magnesium	mg/kg	23	52	1200	377.4	328.3
Manganese	mg/kg	23	2.6	160	58.74	51.13
Mercury	mg/kg	23	0.005	0.06	0.0161	0.0158
Nickel	mg/kg	23	ND	ND	ND	ND
Potassium	mg/kg	23	12.5	490	121.70	120.98

Table B-12. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
						Deviation
Metals (cont.)						
Selenium	mg/kg	23	ND	ND	ND	ND
Silver	mg/kg	23	ND	ND	ND	ND
Sodium	mg/kg	23	10	330	38.3	67.2
Thallium	mg/kg	23	0.5	3.5	0.92	+
Vanadium	mg/kg	23	0.5	19		0.96
Zinc	mg/kg	23	2.5	76	4.46	5.09
Other Parameters		+ = -	2.5	/6	29.20	23.14
Bulk density	g/cm ³	23	1	1.3	1.18	0.09
Cation exchange capacity	meq/ 100g	23	4	31	11.3	8.1
Hydrogen ion	1	23	1.99526E-08	6.30957E-06	7.83E-07	1 775 00
Percent solids	%	23	74	99		1.77E-06
Resistivity	ohm-cm	16	29000	2900000	93.9	7.1
Texture (No. 4)	%	23	98.2		749750.0	910290.8
Texture (no. 10)	%	23	93.8	100	99.76	0.51
Texture (No. 40)	%	23		100	99.41	1.39
Texture (no. 60)	%		83.5	99.7	94.63	5.60
Texture (No. 100)		23	51.4	98.1	83.57	14.50
	%	23	15.3	60.6	44.61	14.91
Texture (No. 200)	%	23	3.8	20.7	9.79	4.93

Table B-13. Minimum, maximum, mean, and standard deviation for each parameter analyzed in disturbed soils.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)						
4,4'-DDD	μg/kg	23	1.7	10	2.21	1.71
4,4'-DDE	μg/kg	23	1.7	110	7.27	22.57
4,4'-DDT	μg/kg	23	1.7	67	6.00	14.71
Sum 4,4' DDD + 4,4' DDE + 4'4 DDT	μg/kg	23	5.1	187	15.48	38.46
Aldrin	μg/kg	23	ND	ND	ND	ND
Alpha - BHC	μg/kg	23	ND	ND	ND	ND
Beta - BHC	μg/kg	23	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	23	ND	ND	ND	ND
Chlordane (gamma)	μg/kg	23	1.7	5.8	2.0	8.0
Chlordane (total)	μg/kg	23	ND	ND	ND	ND
Delta - BHC	µg/kg	23	ND	ND	ND	ND_
Dieldrin	µg/kg	23	1.7	5	1.99	0.68
Endosulfan I	μg/kg	23	ND_	ND	ND	ND
Endosulfan II (beta)	μg/kg	23	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	23	ND	ND	ND	ND
Endrin	μg/kg	23	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	23	ND	ND	ND	ND
Endrin Ketone	μg/kg	23	ND	ND	ND	ND_
Gamma-BHC (Lindane)	μg/kg	23	1.7	5.4	2.00	0.76
Heptachlor	μg/kg	23	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	23	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	23	ND	ND	ND	ND
Isodrin	μg/kg	23	ND	ND	ND	ND
Methoxychlor	μg/kg	23	ND	ND	ND	ND
Mirex	μg/kg	23	ND	ND	ND	ND
Toxaphene	μg/kg	23	ND	ND	ND	ND
Aroclors (8082)						
PCB - 1016/1242	μg/kg	23	ND	ND	ND_	ND
PCB - 1221	μg/kg	23	ND	ND	ND	ND
PCB - 1232	μg/kg	23	ND	ND	ND	ND
PCB - 1248	μg/kg	23	ND	ND	ND	ND
PCB - 1254	μg/kg	23	ND	ND	ND	ND
PCB - 1260	μg/kg	23	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
2-(2,4,5-Trichlorophenoxy) propionic acid (2,4,5 - TP) (Silvex)	µg/kg	23	5	320	22.6	66.0
2,4,5-Trichlorophenoxy acetic acid (2,4,5-T)	μg/kg	23	ND	ND	ND	ND
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	μg/kg	23		ND	ND	ND
3,5 DCBA	μg/kg	23	ND	ND	ND	ND
4-(2,4-Dichlorophenoxy) butyric acid (2,4-DB)	μg/kg	23		60	9.0	12.5
4 - Nitrophenol	μg/kg	23		ND	ND	ND
Acifluorfen	μg/kg	23		ND	ND	ND_
Bentazon	μg/kg	23		ND	ND	ND
Chloramben	μg/kg	23	ND	ND	ND	ND
Dacthal	μg/kg	23		ND	ND	ND_
Dalapon	µg/kg	23		ND	ND	ND

Table B-13. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std.
Chlorinated Herbicides (8151) (cont.)		\top		 		Deviation
Dicamba	μg/kg	23	ND	ND	ND	110
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	μg/kg	23	ND	ND	ND	ND ND
Dinoseb	μg/kg	23	ND	ND	100	-
MCPA	μg/kg	23	ND	ND	ND	ND
MCPP	μg/kg	23	ND	ND	ND	ND
Pentachlorophenol	μg/kg	23	ND	ND	ND	ND
Picloram	μg/kg	23	ND	ND	ND	ND
Sum Chlorinated Herbicides	μg/kg	23	15	440	ND 20.4	ND
PAHs (8310)	F33	1-5		440	39.4	88.8
1-Methylnaphthalene	μg/kg	23	8.5	46	10.07	10.10
2-Methylnaphthalene	μg/kg	23	8.5	68	12.37	10.19
Acenaphthene	μg/kg	23	ND	ND	13.33	13.86
Acenaphthylene	μg/kg	23	ND	ND	ND	ND ND
Anthracene	μg/kg	23	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	23	0.9	130	ND	ND
Benzo(a)pyrene	μg/kg	23	1	250	9.74	26.86
Benzo(b)fluoranthene	μg/kg	23	1.5	220	16.3	51.9
Benzo(g,h,i)perylene	μg/kg	23	1.7	190	13.87	45.44
Benzo(k)fluoranthene	μg/kg	23	1	120	18.99	40.15
Chrysene	μg/kg	23	0.9	200	12.2	30.7
Dibenzo(a,h)anthracene	μg/kg	23	1.7	59	15.02	41.42
Fluoranthene	μg/kg	23	1.7	540	6.22	12.46
Fluorene	μg/kg	23	ND	ND	35.11	113.35
ndeno(1,2,3-cd)pyrene	μg/kg	23	0.9	210	ND 13.00	ND 10.00
Naphthalene	μg/kg	23	ND	ND ND	13.22	43.32
Phenanthrene	μg/kg	23	0.9	120	ND 10.10	ND Of 10
Pyrene Pyrene	μg/kg	23	0.9	240	12.19	31.16
Sum of PAHs	μg/kg	23	50.5	2397.5	18.80	56.21
Metals	FSTIS		30.5	2397.3	223.17	495.95
Numinum	mg/kg	23	54	22000	2491.3	E000.0
Antimony	mg/kg	23	ND	ND	ND	5863.6
Arsenic	mg/kg	23	0.2	2.4	0.70	ND 0.70
Barium	mg/kg	23	10	36	13.0	0.70
Beryllium	mg/kg	23	ND	ND	ND	6.8
Cadmium	mg/kg	23	ND	ND	ND	ND ND
Calcium	mg/kg	23	130	200000	43053.5	ND COOLO O
Chromium		23	0.5	24	3.85	60942.3
Cobalt		23	ND	ND ND	ND	6.08
opper	mg/kg	23	0.5	4.9	2.55	ND 0.75
on		23	37	16000	1783.4	0.75
ead		23	0.5	18	5.06	4059.3 4.61
lagnesium		23	12.5	1300	288.85	343.32
langanese		23	1.2	20	7.90	6.43
lercury		23	0.005	0.07	0.0108	0.0166
ickel		23	ND	ND ND	ND ND	ND
otassium		23	12.5	180	72.72	54.62

Table B-13. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						Bottation
Selenium	mg/kg	23	1	3	1.2	0.6
Silver	mg/kg	23	1	2.4	1.14	0.40
Sodium	mg/kg	23	10	1400	336.3	444.0
Thallium	mg/kg	23	ND	ND	ND	ND
Vanadium	mg/kg	23	0.5	25	3.44	6.26
Zinc	mg/kg	23	2.5	50	11.79	11.41
Other Parameters						
Bulk density	g/cm ³	22	0.6	1.4	1.11	0.21
Cation exchange capacity	meq/	18	1.0	39.0	10.6	9.6
, and the same of	100g	1				
Hydrogen ion		23	5.01187E-09	1.25893E-05	1.50E-06	3.34E-06
Percent solids	%	23	73	100	91.4	8.2
Resistivity	ohm-cm	22	3800	2900000	722345.5	891527.5
Texture (No. 4)	%	23	73.9	100	98.11	5.46
Texture (no. 10)	%	23	69.8	99.9	95.12	7.68
Texture (No. 40)	%	23	46.7	99.3	84.39	17.27
Texture (no. 60)	%	23	28	98.3	76.17	21.94
Texture (No. 100)	%	23	10.7	71.5	46.24	19.23
Texture (No. 200)	%	23	1.9	37.8	7.98	9.40

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC001	SSC002	SSC003	SSC004	SC005 S	90005	SSC007	SCOOR	Scono	0000	ļ	10000				
		2,98	06/12/98	06/12/98	96/18/29	06/18/98 0	06/15/98 0	08/12/98 0	96/18/98	06/18/98	8/28/38	- 6 0	821% 821%	6/12/98	SSC012 06/12/98	SSC013	SSC014
Organochlorine Pesticides (8081)			T		İ	1	1	1			0	duplicate	ľ	Suplicate			9
4.4 - 500	5,7,000	Ī				1	1									Ì	
4,4 · DDE	3		3	200	34	33	٦	<3.3	<3.4	H	<3.3	<3.3	<33	Ę	Ę	1	Ī
4,4' · DOT	3		3	3	7	7	┪	633		H	<3.3	633	633	37	1	1	36
Aldrin				3	2	1	7	33	H	┝	633	<3.3	33	5	:		3
Alpha - BHC	3		3		+	┪	┪	633	Н	Н	<3.3	333	633	597	5		3 6
Beta - BHC	No.		36		\dagger	+	┪	633	Н	Н	<3.3	<3.3	t	\$37	37	t	36
Chlordene (alpha)	, oliko		3 5	3	+	5	┪	<u>ئ</u>	20	Н	<3.3	<3.3	¢3.3	5.5	5	t	
Chlordene (germma)	, ov			3 5	+	\dagger	+	┪	1	Ħ	<3.3	<3.3	H	5	437	t	
Chlordene (total)	2000		36	3	†	+	┪	┪			<3.3	<3.3	t	5	5	t	
Detta - BHC	, ovik		3 5	315	3	63	333	63.3	43.4	<3.4	<3.3	<3.3	43.3	5	3,7	100	
Dieldrin	io/Ko			?;;	†	+	┪	┪	-	٦	<3.3	<3.3	H	ر د د د	37	t	
Endosultan i	, orko		3	3	†	+	+	┥	+	Н	<3.3	<3.3	┢	37	237	t	
Endosultan II (beta)	V)/Or		36	3 6	\dagger	+	+	┥	+	Н	<3.3	<3.3	r	437	37	t	
Endosultan Sultate	oy/or				\dagger	+	┪	┪	┪	\dashv	<3.3	<3.3	H	3.7	437	t	5
Endrin	O)VOV		3 5	3 5	\dagger	+	+	+	┪	Н	<3.3	<3.3	H	43.7	37	t	33.5
Endrin Aldehyde	i V			3	+	+	+	┥	┥	H	<3.3	<3.3	H	5	37	t	
Endrin Ketone			3	5	+	+	┪	4	Н	-	<3.3	633	╁	3,	100	t	36
Germma - BHC (Lindana)	2 2	•		3	+	┪	+	_	H	┝	33	683	633	Ė		t	3 6
Heptachior	200		3	5	34	┥	<3.3	-	-	H	683	╀	t		†		3
Heptachlor Epoxide (a)	37.00	:	3	+	+	┨	۲	_	H	H	<3.3	╁	633	727		\dagger	3
hectachic Frontie (h)	No.	*	633	┥	┨	Н	-	ļ.,	┢	╁	633	╁	t			+	3
Isodin	MO/Kg	3	§	Н	Н	-	H	┞	╁	╀	5 6		+		3 :	+	
Methosophos	MO/Kg	**	65	\dashv	Н	Н	H	╀	╁	╁	25.5	 - -	t	3 5	2	†	633
Minex	WW	7	63.3	<3.3	<3.4	H	33	╀	╁	╀	56.		3 5	2 5	;;;	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	33
Tovenhere	MO/Kg	×4.4	693	Н	Н	H	┝	╀	╀	╁	1		$^{+}$	31.		1	69
	MVK9	777	<3.3	H	Н	┝	H	╀	╀	1		3 5	+	3		(3)	당 당
DCD (018/13/2)				-	┝	┝	┝	╀	╀	+	+	+	╁		è	(§)	633
PCB-1231	0 Vov	â	85	<ع	λ\$.	8	┝	╁	╁	25	t	+	+	ļ.			
100 to 1000	No.	<u>\$</u>	چ چ	<33	\vdash	\vdash	┞	H	t	۶	╁	+	+	3 6		j	8
DCB-13.48	M/G	\$	8	<33	\vdash	-	┝	╀	╁	5	+	+	$^{+}$	3 5		Ę)	8
0+21-00-1	₩o/Ko	<44	SS SS	<33	\vdash	H	╁	╀	t	\$ 3	╁	+	+	Ì	÷	ş	8
FCB-1254	₽9/Kg	<44	ξ	జీ	H	╀	╀	+	+	3 3	+	+	+	(37	Ę	<37	33
1	μο⁄Kg	7 7 >	8	8	\vdash	╁	Ę	3 8	ş ₹	5 3	3 8	3 8	8	(§)	Ę	<37	ج ج
Chorinated Herbicides (8151)			-	-	╀	t	+	╀	+	5	+	+	+	Ę	Ş	3 3	33
Z-(Z,4,5-1 inchiorophenoxy)propionic acid (2,4,5 · TP) (Silvex)	40/Kg	<13	<10		┝	╁	╁	+	+	+	+	+	+	+		+	
2.4. Dichlomobacous acets acid (2.4.5 - 1)	W/Ko	₽ F	ş	۶ و	<10	<10	┝	₽ P	╁	+	: 5	+			₹	+	ę
3.5-DCBA	D'A	₽ 	ê e	\dashv	+	Н	Н	\vdash	9	9	╀	9	+	+	† -	╡	9
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	5/2/	2 5	-	+	+	+	ę	Н	Н	H	\vdash	┝	9	\dagger	 -	+	
4 - Nitrophenol	Z Z	2 5		2 5	2 5	\$	+	+	ş	9	Н	<10	H	┝	Ę	Ę	: 8
Acithorien	Š	 - -	9 8	+	+	+	\dagger		+	+	\dashv	\dashv	<10	H	41	-	9
Bertlazon	10VKg	<13	ê	+	╁	╁	+	+	2 5	Ş ;	+	\dashv	\dashv		Н	H	ę
Chloramben	MO/Kg	£ € 5	100	+	t		+	+	+	+	+	+	+	\dashv	Н	<11	9
Decthal	MO/Kg	£13	┝	t	╁	+	+	\dagger	+	+	+	+	휭	Ę	Н	-	ş
Dalapon	W/Kp	5	900	 - -	5	+	+	\dagger	+	 -	\$:	┥	\dashv		<11		9
Dicamba	₩0/Kg	¢13	H	╁	╁	╁	+	\dagger	+	+	+	+	+	1			ę
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	MVK0	<13	¢10	┝	╁	Ş				\dagger	+	020	+	+	₹	\dashv	ê
Circoseo	₩o⁄Kg	<13	-	¢10	┞	H	╁	t	+	+	+	+	+	\dagger	+	₹ V	ş
MCPA MCDD	Н	Н	<100	H	┝	╁	╁	t	t	+	╁	+	2 5	\dagger	+	+	ခ
W.C.T.	Н	Н	H	┝	×100	×100	100	100	+		3 8	+	+	+	۲ چ	┪	8
Periodicipanol	MVK9	Н	Н	<10	H	H	t	╀	$^{+}$	+	+	+	+	+	+	+	8
ricorem	70/Kg	<13	<10	Н	¢10	\$10 \$	8	9		╁	╁	+	۲ إ	+	\dagger	÷	읡
							ł	1	\mathbf{I}	┨	+	┨	\forall	Į Į	El	┥	9

I = value is between the Method Detection Limit and the Practical Quantitation Level
J = value is below the established limit for accuracy
* = 8310s were dikited (1:5)
** =8310s were dikited (1:10)

Table B-15. KSC Background raw data for soil locations. " = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		10000	SSCOO	50003	SCOOL	SC0051	90000	SC007 1	20008	SC009	SSC010	_	SSCO		SSC0121	SC013 .	2007
		06/12/98	06/12/98	06/12/98	06/18/98	06/18/98	06/12/98	06/12/98	06/18/98	06/18/98	8/28/98	90	6/12/98		06/12/98	06/18/98	06/18/98
										_	Ĭ	duplicate	Ĭ	duplicate			
PAH (8310)																	
1 - Methylnachthalene	MO/Kg	2 3	45	Ę.	<u> </u>	 	 - -	412	/ >	<17	<u> </u>	<17	<u> </u>	<19	<19	<19	<17
2 - Methylnaphthalene	pa/Kg	Z>	417	417	8	<17	<17	<17	<17	<17	<17	<17	<17	<19	<19	<19	دار
Acenaphthene	MQ/Kg	ß	<17	<17	<17	<17	<17	<17	<17	<17	<17	<17	<17	<19	<19	<19	<17
Acenaphthylene	0X/6/	7 4 7	<3.3	<3.3	<3.4	49	<3.3	<3.3	<3.4	<3.4	<3.3	<3.3	<3.3	<3.7	<3.7	<3.7	<3.3
Anthracene	MQ/Kg	<23	<17	<17	<17	<17	<17	<17	<17	<18	~17	<17	<17	<19	<19	<19	<u>1</u>
Benzo(a)anthracene	DA/Kg	<2.2>	<1.7	<1.7	3.2	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.9	<1.9	<1.9	<1.7
Benzo(a)pyrene	MO/Kg	63	2	2	21	2	<2	<2	<2	<2	3 1	2	<2	<2	<2	31	3-
Benzo(b)fluoranthene	MO/Kg	4>	Ş	ŝ	63	<3	63	ç	دع	<3	1 1 4	-	دع	<3	ç	<3	೮
Benzo(a.h.)benviene	νο⁄Κα	4,45	<3.4	43.4	434	43.4	43.4	43.4	43.4	43.4	43.4	\$34	43.4	43.7	43.7	3.7	34
Renzolkilioranthena	₩/V	\$3	~	8	8	\$	2	8	ѷ	2	3-1-2	2 -	<2	\$ \$	2	•	2
Chosene	wo/Ko	200	417	15	6.	- - -	7.5	712	7	, V	1-0	7		61×	615	-	2.11
Diversor's Nanthream	0)/0/		75	7	2	165	180	180	785	75	100	165	150	Ę	Ę	(5)	3
e la comita de la comita del la comita del la comita de la comita de la comita del la comita de la comita del la comita	1						+							,	15	1	
Fluoranthene	OV.		3				3	3	3		3						
Filorene	DV/OV	•	3		•	*		*	;	;							
Indeno(1,2,3-cd)pyrene	MO/Kg	42.2	Ì	Ş	2	\ \ \ \ \	5.4		V	Į.		7		8.0	812	Ŝ	\$
Naphthalene	//0/Kg	¢23	<17	<17	201	÷	4	<12 <12	Ş	ş	 	 	4	¢19	√19	ę	ş
Phenanthrene	MO/Kg	<2.2	<1.7	<1.7	<1.7	16	<1.7	<1.7	<1.7	<1.7	د17	<17	<1.7	<u>د1.9</u>	41.9	<1.9	<1.7
Pyrene	W/Kg	<2.2	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.9	<1.9	<1.9	<1.7
Surrogates																	
2.4-DCAA	፠	124	8	æ	8	100	112	132	8	28	132	8	136	88	108	72	8
2.4.5.6-TCMX	8	8	S	89	ž	88	58	38	29	88	72	75	8	88	7.5	53	45
	8	F	78	8	115	133	105	8	68	118	126	136	89	16	88	72	8
Pulsary	36	24	1	8	ē	3	28	38	2	88	8	113	8	83	6	74	92
Metele																	
Abinin	mo/Ka	QQ.	260	170	240	240	240	280	210	1400	0001×	-1000	961	210	8	240	22
Action	S)	Š	S	2	8	0	ş	ş	ç	65	65	<2	?	62	62	6	Ş
Annual	3300	, ,	,	,	, =			, ,	,	150	120	180	:	, 3	,;	37	, =
- Raeuc	T T	6.5	6:5	1		2	3		٤	3		5	; [5	Ş	: [5	3 5	۶
Barium	morks	Ì	3	3	Ş	3	3	8	ş,	Į,	3	8	3	3	3	3	3
Beryllium	mo/Kp	Ÿ	v	V	₹	⊽	V	V	₹	+	₹	⊽	v	₹	V	┙	Į.
Cadmium	mo/Ko	v	₹	v	Ţ	ī	╗	٠	v	V	V	v	v	⊽	V	v	v
Calcium	mg/Kg	51000	67000	900	1000	42000	900	90	2800	┪	2 08	9000	9000	9006	900	8	2600
Chromium (total)	mp/Kg	2.8	2.3	~	5.6	2.3	27	22	22	-+	2.5	2.4	2.3	5.6	2	24	1.3
Cobatt	ωο/Ko	>	Ş	Ş	ş	\$	ş	Ş	Ş	7	ŝ	Ş	Ş	ş	٧	ş	ş
Copper	mo/Ko	<7	< 5	~ 5	V	V	\$	<\$	V	┪	ŝ	ç	ç	ş	ŷ Ŷ	₹	v
Iron	пр/Ко	1200	1200	8	8	82	8	8	1200	1200	8	200	8	8	8	8	8
Lead	mo/Kg	3.2	6	2.4	6	1.7	<u>.</u>	2.3	2	2	2	52	22	55		7	Ţ
Magnesium	mo/Kp	240	320	310	5	90	욹	280	ş	8	ş	<u>Ş</u>	8	ŝ	29	£	280
Manganese	Oγ/Oω	13	13	10	11	10	=	5	7	8.8	12	2	15	5	F	2	_
Mercury	mo/Kg	¢0.01	-0°0	6 0.0	6 0.0	*0.0	¢0.01	<0.01	<0.01	<0.01	<0.01	<0.01	¢0.01	¢0.01	*0.0	<0.01	<0.01
Nickel	mo/Ko	7	Ŷ	ş	ş	Ş	\$	\$	ŝ	ŝ	ŝ	ŝ	\$	ê	99	ê	ŝ
Detection	mo/Ko	8	28	\$	2	46	22	28	ş	3	23	\$3	33	8	8	25	8
Character	mo/Ko	40.05	v	8 9	8	2	\$0.0v	<0.2	8	8	v	Ÿ	v	40.2	402	2	ð
O Charles	ω. Vom	\$3	8	ş	8	\$	2	8	S	8	8	ş	Ş	v	8	?	Ş
	Ş	944	460	43	8	S	370	97	8	210	9	057	570	2	9	929	270
SOUGH.	3	1	-	-	-	1	ŀ	ŀ	-	1	v	V	V	3,6	V	V	ī
I Damum	3		,			,,				:		2.9	9,0	-	-	-	-
A STATE CHILL				-	: ::	3	:	3	3	3	=	95	2	۽	=	. 6	
7117							+										

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy = 8310s were diluted (1:5) = 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** value is an outlier and not used in the analysis to develop the KSC Background screening values.

		06/12/98	06/12/96 06/12/96 06/16/96	6/12/98	06/12/96 06/16/98 06/18/98 (55C005 06/18/98	SSC006 06/12/98	SSC006 SSC007 SSC008 SSC009 OS/12/96 OS/12/96 OS/18/96 SSC005 SSC006 SSC007 SSC006 SSC009 O6/18/98 O6/12/98 O6/12/98 O6/12/98 O6/18/98	SSC009	SSC010	2 g	SSC011		SSC012	SSC013	SSC014	
					_				2	}	3	2	5		<u> </u>		5
Other Becausedans	Ī	1	1	1								duplicate	۲	duplicate			
										ľ		l	l	l	l		I
Bulk Density	, 0,000	<u>:</u>	<u> </u>	ļ	Į.	ļ	Ī										
090				7	-	7	1.2	1.3	-	_	?	1.3	1.2	11	98.0	-	ŀ
	med I	,	7	7	8	8	7	-	1000	8	,	,	ŀ	l	3	?	-
Percent Solids	*	75	8	8	8	8	8	8	3 8		,	7	1	-	-	220	8
PH (96)	-	:	•		;		Ca	S	R	ř	ŝ	88	8	28	8	68	8
Reciptivity		70	•	┪		-	8.2	8.2	.	8.2	8	1.8	82	82	ŝ		:
	5	8000	2000	0 000	×590000	>1700001	39000	47000	20000		50000	20000			,	٩	9.4
Texture (No. 4)	%	٤	٤	٤	٤	1			3	3	_	SWW0	3000	14000	88	100000 00000	15000
Terdura (No. 10)			3	3	3	3	8	8	8	8	8	5	901	9 66	٤	٤	٤
	¢	8	8	8	8	8	100	400	٤	0.70	8	8	1			3	3
Texture (No. 40)	8	A A A	78.2	70 5	6,30	5			3		93.0	ŝ	8786	89.5	8	8	100
Texture (No. 60)	8	293	200		3	8	8	33.5	4/4	90.3	79.8	75.9	8	89	84.3	2	2
Texture (No. 100)	֓֡֓֜֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֡֓֓֓֓֓֓֓֡֓֓֓֓֡֓֡	3	23	c g	R. RS	29.3	46.9	51.6	35.4	70.4	26.8	25.5	15.8	15.6	27.5	98.6	į
Taking (IV. 1W)	ę	17.5	5.9	5.9	7.5	4.2	12.1	10.3	,	7 86	3	ļ			1		7.05
Texture (No. 200)	8	80	90	3	-	1	,				2.5	?	ş	2	3.4	'n	4
Total Organic Carbon	2000	2000	200	+	-	20	3	¥ï)	5	7	1.4	1.3	0.3	0.2	0.2	0	٦
		300	2000	3000	>20/00	15000	2100	800	2300	>26700 >26700		>26700	133	17000	5	1	

I = value is between the Method Detection Limit and the Practical Quanitation Level J = value is below the established limit for accuracy
= 8310s were diluted (1:5)
= 8310s were diluted (1:19)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

						I			10000	000000	000000	70000	0000	١	COCOC		7000
		SSC015	SSC016 06/09/98	SSC017 06/09/98	SSC018 6/22/98		SSC019 S	SSC020 06/22/98	07/08/98	06/16/98	06/16/98	06/26/98	7/27/98	. 60	8/20/98		07/07/98
	_				Đ	duplicate					1		٦	duplicate	٦	duplicate	
Organochlorine Pesticides (8061)						Н								1	1	1	
. A V.	0//Ko	<3.3	<3.3	<3.3	<3.5	<3.4	<3.3	<3.3	<3.6	<4.2	633	\$. 6.6	5	+	+	+	
4 Y. DDF	Š	433	<3.3	<3.3	Н	<3.4	<3.3	<3.3	3.6	25	<3.3	\$. 6.4 6.6	34	7	+	+	
44'.DOT	Š	683	<3.3	<3.3	Н	<3.4	<3.3	<3.3	38	42	69	8.6	34	85	9.4	\$ 6	
Action	No.Ko	<3.3	<3.3	<3.3	Н	<3.4	<3.3	633	38	4 2	33	\$	¥.	†	+	†	÷
Apra - BHC	νο/Κα	<3.3	<3.3	<3.3	<3.5	<3.4	<3.3	\$33	3.6	24.5	60	9,0	53.4	80.0	†	+	
Beta - BHC	MO/Kg	<3.3	<3.3	633	<3.5	\$	333	43.3	43.6	242	5	\$	*	500	0 4		
Chlordane (alpha)	MO/Kg	<3.3	<3.3	<3.3	\$35	70	633	69	36	25		2 4	\$ 5	8.5	\dagger	\dagger	
Chlordane (gamma)	MO/Kg	Ş	<3.3	5	43.5	3		3		 	3				+	t	,
Chlordane (total)	νο/Kg	33	\$33	£	500		3			*	3 5				97	187	, e
Detta - BHC	MO/Kg	333	333	5	250	*	36	3			36				t	347	2
Dieldrin	2	Ş	533	5			35	35		;		7 9 9	34	386	990	t	437
Endosultan I	2			36			3 6	565	3.6	242	43.3	×4.6	465	<3.8	H	H	<3.7
Endosulian II (beta)	Z S	36			1	2	6	\$33	43.6	42	63.3	<4.6	<3.4	<3.8	4.6	H	<3.7
Endosultan Suffate	No.		333	433	\$35	34	33	<3.3	43.6	4.2	<3.3	<4.6	<3.4	<3.8	<4.6	<4.6	<3.7
Engin	No.	633	<33	<33	<35	460	633	<3.3	<3.6	<4.2	<3.3	<4.6	<3.4	<3.8	<4.6	<4.6	<3.7
Engral Address	, OKO	<33	<33	<333	<3.5	43.4	<3.3	<3.3	<3.6	<4.2	<3.3	<4.6	<3.4	<3.8	4.6	4.6	3.7
Character (BLC (Lindens)	wo/Ka	<3.3	<3.3	<3.3	<3.5	<3.4	<3.3	<3.3	<3.6	<4.2	<3.3	<4.6	3.4	<3.8	979	4.6	(S)
Linear Carol Carolina	, o.Ko	633	<3.3	<3.3	<3.5	<3.4	<3.3	<3.3	<3.6	<4.2	<3.3	4.6	\$ 5	3.8	\$ \$ P	4.6	37
Hondachter Enough (a)	DW/O/	433	<3.3	<3.3	<3.5	<3.4	<3.3	<3.3	<3.6	<4.2	633	\$ 6	\$ 5	\$38 8.5	\$- 6.6	4.6	37
hard-ortice Enough (h)	40/Ko	<3.3	<3.3	<3.3	<3.5	<3.4	<3.3	<3.3	<3.6	42	<33	4.6	\$3	85	¢.6	×4.6	Ş
lendrin	DX/QV	<3.3	<3.3	<3.3	<3.5	<3.4	<3.3	<3.3	<36	¢4.2	633	×4.6	5	43.8	\$.	9	63.7
Methorychior	W/K0	<3.3	<3.3	<3.3	<3.5	43.4	633	<3.3	9.0	c4.2	£5.	ş	8	\$3.8 6	\$.	9	
X G-TA	0V/V0	<3.3	<3.3	<3.3	<3.5	5	633	£	98	24.5	3	\$.			3 8	2 6	2 5
hene	W.Ko	43.3	63.3	63.3	335	\$ P	53.3	5	*	7	3	000	Ì			3	T
Aroclors (8082)	1	8	ş	8,	ļ	7,	٤	Ę	ş	9	(33	\$ 4 6	20	85	\$\$ \$	448	437
PCB-1016/1242	DVA.	3 8	3 8	3 8	3 8	5	8	8	8	45	ŝ	94,	\$3	<38	<46	<46	<37
PCB-1221		3 2	Ę	Ę	435	8	ŝ	ŝ	86,	¢42	68 >	<46	<34	<38	<46	<46	<37
PCB-1232	N CONT	Ę	ş	8	8	8	ફ	ક્ષ્	85	<42	<33	<46	×34	<38	<46	<46	33
FCB-1248	VO/KG	8	S,	ŝ	85	ਲ੍ਹ	ફ્	ŝ	×36	<42	<33	<46	ξŞ.	88	9	\$	37
V30-1234	, oy	ŝ	ŝ	8	8,	रु	633	\$	<36	<42	<33	<46	ह	88	\$	99	Ç\$
Chloringted Herbicides (8151)																1	1
2-72 4 5-Trichlorophenoxyloropionic acid (2,4,5 - TP) (Silvex)	WO/Kg	¢10	⁴ 10	<10	<u>.</u>	¢10	ş	ş	₹	¢13	ş	ş	?	Ę :	*	*	Ę
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	D)/G/	<10	ot >	ę	Ę	ş	Ş	ę	Į.	¢13	Ş Ş	0 5	2 5	Į.	**	÷ =	=
2,4-Dichlorophenoxy acetic acid (2,4 · D)	Š.	9	۶	۶		2 5	2 5	2 5	 -	2 5	ę	9	2	Ī	Ę	15	Ę
3,5-DCBA	DY/O/		2 5	2		2 8	2 2	ę	Ę	¢13	¢10	64.6	ę	Ę	¢14	<14	1
4-(2,4-Dichlorophenoxy)burync acid (4,4 - US)	2	1	Ę	Ş	₹	ę	012	ę	<u>^</u>	<13	<10	<4.6	<10	<11	<14	<14	Ę
4 - Miropherio	No.	ę	95	5	₹	9	د†0	¢10	<11	<13	<10	<10	\$ 10 10	41	7	₹	Ę
Conteston	VQ/KG	ş	ş	ę,	حاا	<10	<10	<10	<11	<13	<u>م</u> 10	ş	ş	Ę	7	7	Į:
Chlorenhen	ξ	ş	9	¢10	<11	<10	<10	¢10	11	€ 13	9	ę	ş	ŧ	7	*	Į.
Darthal	Š	ڊ 9	×10	<10	<11	¢10	د 10	ę	÷	¢13	ę.	9	Ş	Į.	*	₹ :	\
Dalanon	νογκα	c10	¢10	<10	Ę	ę,	ş	ş	÷	Ę	ş	ęļ	۶	Ş	*	₹ ;	1
Dicamba	M0/Kg	¢10	¢10	ę	Ŧ	ş	ę	9	ē	£	9	200	2 5	į		:	7
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	0A/07	¢10	ş	ş	Ş	ę	ęş	₽ •	₹:	¢13	9		9 5	Į	*	:	1
Dinoseb	MVK9	ę	Ş	ş	ŧ	Ş \$	9	200	- 65	2 5	2 5	26	200	5	280	200	9150
MCPA	WKa	8	8	8 5	0170	38	3 5	38	200	900	8 5	4.6	88,5	\$30	280	280	410
MCPP	2 2	3 5	3 5	3 5		3 5	9	9	V	€13	ę	5	9	Ę	414	¢14	ij
Pentachlorophenol	2 2	 	1	1	1	Ę	95	9	Ę	Ę ₹	₽ V	د <u>1</u> 0	٠ <u>10</u>	11>	<14	<14	11
Pictoram																	

^{1 =} value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy • = 8310s were diluted (1:5) • =8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC015	SSC016	SSC0171	SSCO			000000	10000	1					İ		
		06/18/96	96/60/90	96/60/90	6/22/96		08/22/98	06/22/96	07/08/88	06/16/98	98/16/98	35CU24 06/26/98	SSC025	8 8	SSC026		SSC027
PAH (8310)	ŀ			1		OUDIFICATIO						_	ľ	duplicate		S S S S S S S S S S S S S S S S S S S	96/2/
ı	3,000	ļ	ŀ	ŀ								ľ	Ī		1	Branch	I
2 - Methymachthalene		:		₹	8	<17	<17	41>	1812	Ş	÷		ļ	ļ	1		
Acenaphthene	2		Ì	Ì	√18 √18	<17	<17	412	85 85	2	Ę	2			†	\$	8
Acenachthylene	T C		è	È	<18	<17	<17	<12 1	¢18	100	1	1 5	 		†	787	6
Anthracene	a di	3	69	633	<3.5	<3.4	<3.3		989	\$2	565				3	8	\$ 20 20 20 20 20 20 20 20 20 20 20 20 20
Benzo(a)enthracene	2 2	Ì	È	Ş	<18	<17	41>	H	8	Ş	1	 		3	†	┥	37
Benzolalovrene	S C	V	<u>~</u>	₹	<1.8	H	<1.7	H	8.5	100		† ?!?		*	3	┪	419
Benzofbifluorenthene	SV.	9	ş	\$	<2	ح	3	H	2	t	ļ,	;	†	2,	6.3	3	<1.9
Benzo(ch imendene	WOYKO	Ş	ç	ç	<3	H	S	8	2	t	, , ,	†	٦ŀ	»	5	•	3
Berryklikerenthen	WW.	34	<3.4	<3.4	<3.5	\vdash	t	t	t	†	?	:	2	8	¥	٥,	೮
Cont. Cont. M. Marie Cont. Con	MO/Kg	7	2	8	0	t	 	;;;		†	\$	7	34	<3.8	64.6	<4.7	5,
	W/Ko	<u></u>	712	Î		t	\dagger	†	†	7	3		~	8	ŝ	t	0
Utbenzo(a,h)anthracene	10Xo	\$34	5	1	200	+	+	†	41.8	2.1	<1.7	<23	<1.7	6.5	4.7	2	٩
Fluoranthene	MO/Kg	38	187		\dagger	\dagger	+		1	٦	<3.4	H	43.4	85	t	t	
Fluorene	νον γον	1	*	;	+	5	┥	7	٦		43.4	H	100	85	t	\dagger	
Indeno(1,2,3-cd)pyrene	3		;	3	\dagger	┪	<3.4	<3.4	-	┝	150	H	Į		t	+	
Naphthalene					+	_		-	-	H	<u> </u>	t			†	+	ş
Phenanthrene		*	Ì	È	┥	_	<17	<17	H	t	S	t		2	†	+	Ş
Pyrene			Ş	⋛	4	Н	<1.7	 - -	41.8	2	12	ž	†		†	424	ę
Surronatee	2	į	\ \{\bar{\bar{\bar{\bar{\bar{\bar{\ba	<1.7	<1.8	<1.7	<1.7	\ \ \ \ \	H	t	-	t	†	8	6.3	┪	6
2 4. DCAA						-		t		1		1	┨	<1.9 ⊥	1	-	<1.9
24 & & TOUY	8	28	132	35	8	Ͱ	127	E	2	-	Į	-					
SAC CARA	8	- 67	78	8	2	╀	3	3 3	3;	5 8	3	2	9	52 2	132	135	2
	*	83	121	22	25	25	3 5	5 5		3	3	8	62	91	18	28	57
	*	28	8	8	2	+	3 2	2 2	_ [≥	3	5	2	16	126	135	=3	3
Wetals						╀	3	3	S	3	2	2	75	16	112	8	=
Auminum	mo/Ko	8	3	982	25	+	1	-	ł	ł							
Ammony	mo/Ko	8	3	8	╀	╀	\dagger	+	2	+	┥	z	31	H	42	-	2
Arsenc	mo/Ko	- -	-			╀	+	+	+	\dashv	-	ç	<	┢	H	t	
Barken	morka	8	ķ		+	•	7	+	+	Н	_	┝	┝	86	t	t	, ,
Beryffum	morka	V	十 小	+	+	+	+	4	1		ر20 ح	<28	8	t	28	800	3 8
Cedmium	┿	ļ.	 	t	+	4	†	6.88	\dashv	□	₽	\vdash	Ī	t	t	+	7
Celcium	٠	240m	╀	2000	+	╬	┪	4	⊽		٦	\vdash	Į	V	ł	t	Ţ
Chromium (total)	┿		+	+	4	2 00019	76000	4	Н	270	F	₹ 25	29],
Cobet	ao/Kom		; ; ; ;	t	+	+	4	+	┪	Н	H	┝	ī	-	+	+	,,
Copper	mo/Ko	, -	† ***	\dagger	+	+	+	┪	7	Н	Н	├	\$	9	 	 	- 4
tron	morka	82	Ş	+	4	+	+	┪	7	Н	Н	\vdash	\$	*	+	+	7
read	mg/Kg	0.5	2.8	2.6	1	3 :	R :		3	959	8	82	28	24	t	+	72
Magnesum	mo/Ka	88	370	t	ļ	+	+	+	+	┨	┥		Н	r	H	-	2
Manganese	mo/Ka	7.3	2	╁	1	+	+	+	7	4	┥		_	8	88	89	5
Mercury	┪	t	╀	t	Ţ	†	†	+	7	┪	٦	Н	H	H	t	╁	-
Nicke	t	ű	t	+	+	+	†	20.02	_	┪		<0.01	10.0>	<0.01 <	t	t	Ę
Potassium	۰	٤		+	╁	+	0	┪	1	-		۲	r	t	t	t	Ţ.
Selenium	┿		3 5	╁	+	+	┪	┪		_	H	H	H	t	╁	7,	,
Silver	DOM(C)	+	, ,	╁	»/«	2	9	4		H	H	ç	t	t	t	+	9 9
Sodium		, 5		+	+	+	+	_		\vdash	H	H	H	t	\dagger	\dagger	,,
Thalkum	200	3	-	+	┥	┩	\dashv	Ц	131	===	=	Z	***	18	+	2 5	9
Vanadium	2000		;	+	+	\dashv	-	Н	٠ ١٧	┝	┞	+	+	\dagger	\dagger	+	5
Zinc	200	+	•	77	2.2	2.1	2.1	1.7	L		H	+	\dagger	\dagger	\dagger	+	Ţ.
		,	70	4	\dashv	4	٦	L	L	\vdash	\vdash		1	t	+		Ţ
								l		\mathbf{I}	1	1			_	~ ~	_

I = value is between the Method Detection Limit and the Practical Quantitation Level
J = value is below the established limit for accuracy
* = 8310s were diluted (1:5)
** = 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC015	SSC016	SSC017	SSC018	ľ	161005	SSC020	SSC019 SSC020 SSC021 SSC022 SSC023	SSC022	SSC023	SSC024	SSC025	23	\$SC026	9	SSC027
		06/18/98 06/09/98	96/60/90	96/60/90	6/22/98		06/25/98	06/22/98	07/08/98 06/16/98	06/16/98	06/16/98	96/52/90	7/27/98	*	8/20/98		96/20/20
				-	б	ouplicate							j	duplicate		duplicate	
Other Parameters																	
Bulk Density	,wa/d	1.2	1.3	1.3	1.4	1.3	1.3	1.3	1.1	1	0.93	-	1.1	1.1	0.92	0.91	1.1
CEC	meo/100g	1000	_	~	•	-	•	9	-	4.4	4.4	•	2	1	0.47	0.88	3
Percent Solids	×	83	8	8	3	96	66	66	91	78	66	72	8	88	72	71	90
(del) Ho	S.U.	8.2	8.13	8.2	60	8.2	8	8.1	2	5.8	5.7	5.1	5.1	5.2	5.9	5.9	4.7
Resistivity	Chin-Chi	930000	1900	17000	280000	19000	100000	1700000	320000	92000	140000	2400000	1500000	1600000	45000	290000	390000
Texture (No. 4)	8	8	8	100	100	100	100	100	100	6.66	100	100	901	100	100	100	100
Texture (No. 10)	×	8	8	9.86	8.66	8.66	100	100	8.66	98.8	6.66	8.66	6.66	6.66	8.66	8.66	99.8
Texture (No. 40)	%	6.76	8	57.8	8	58.7	78.4	84.2	96.4	96.2	66	72.9	97.9	90.3	8	- 28	99.1
Texture (No. 60)	×	57.2	11.8	14.6	7.4	10.2	21.4	28.9	97.1	9.68	8.96	17.3	68.2	88	55.3	55.4	98.2
Texture (No. 100)	×	10.6	1.6	2	6.0	1.3	2.2	3.5	46	31.1	43.3	4.5	14.2	14.2	8.5	9.6	63.8
Texture (No. 200)	%	5	0.5	0.4	0	0	0.1	0.5	3.2	3.1	3.5	1.9	1.3	1.5	1.5	1.6	1.7
Total Organic Carbon	тожо	4700	>26700	26700	>26700	>26700	23300	17000	>26700	>26700	>26700	>26700	>26700	>26700	>26700	>26700	>26700

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy
* = 8310s were diluted (1:5)
** =8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

4.4' - DDD Organochlorine Pesticides (8081) 40/40 4.4' - DDE 4.4' - DDE 4.4' - DDE 4.4' - DDT 4.4' - DDT 40/40 Adha - BHC 40/40 40/40 Beta - BHC 40/40 40/40 Chloridane (samma) 40/40 40/40 Chloridane (samma) 40/40 40/40 Delta - BHC 40/40 40/40 Endosulfan II (beta) 40/40 40/40		8	07/07/98 (<4.6	06/29/96 <3.4 <3.4	33,4 <3.4 <3.4	06/29/98	06/29/98 (6/29/98 (4/3)	28	S 88	SSC036 -	SSC037 06/16/98	SSC036 •	. SSC039 06/16/98		<u> </u>	10,0	SSC043 06/30/98
anochiorhe Pesitcides (8081)	▐▐ ▝ ▐▕▐ ▐	11 	11 +	2 69	111	683	633									Bancon	200
	▋┤┟┧┨╏╒ ┰┼┼┤	HH	\mathbf{H}	5 5	1	c3.3	<3.3	ļ									
	╅╏╏┩╏┩┩┩┩	++	+	5 6	\top	<3.3 3.3	<3.3	۱									
	╂╂╃╂╃╃╃╃	Н		43.4	_			<3.3	434	<3.6	43.4	100	-133	133	, , ,		
	╏┩╏╏ ┼┼┼┼┼		t		I	<3.3	<3.3	<3.3	<3.4	<3.6	43.4	434	5	,		3	3
	┪╏╏┪╏ ┪	ŀ	7	c3.4	-	<3.3	<3.3	<3.3	<3.4	<36	187	14.	1		3	3	3
0	┤┼┼┞ ┼┼	43.4		<3.4	-	<3.3	<3.3	633	434	45,				3	55.5	53.3	34
	╶╏╸╏╸╏╸╏		<4.6	<3.4	H	<3.3	<33	667	100	96			3	63.3	<3.3	<3.3	<34
	┞╺╂╌╂╌┞ ╶╂	<3.4	H	43.4	434	66	t	;				2	200	<3.3	<3.3	<3.3	<3.4
	╏╏╏╏	H	t	t	t		†	3	•	63.0	\$	<3.4	<3.3	<3.3	<3.3	<3.3	460
	┞╴╽ ╶╽	43.4	446	t	t		; ;;;	3	450	43.6	43.4	<3.4	<3.3	<3.3	<3.3	<3.3	125
	╅╅╁┪	+	t	t	+	3	†	433	43.4	<3.6	<3.4	<3.4	<3.3	<3.3	<33	55.	1
	╁┼┤	t	†	+	†	63.3	7	<3.3	<3.4	<3.6	<3.4	43.4	433	233	,		
	╁┤	* 2	\$.6 \$	┪	٦	<3.3	-	<3.3	43.4	<3.6	434	7	,33			3	3
	┥	٦		_	_	<3.3	H	<3.3	785	36/				3	200	Ş	34
	l	H	H	┢	t	223	t				3	43.4	633	<3.3	<3.3	<3.3	<3.4
	┝	34	t	t	t	1	+	3	*	43.6	<3.4	<3.4	<3.3	<3.3	<3.3	<3.3	36
	╁	t	t	+	+	3	┪	S.S.	<34	<3.6	<3.4	43.4	<3.3	<3.3	633	5	ķ
	+	†	†	+	1	<3.3	-	<3.3	<3.4	<3.6	<3.4	43.4	233	123	,		
Aldeburde	SV/SV	43.4	¢4.6	-	-	<3.3	<3.3	<3.3	43.4	636	16/	,			200	2	Ş
R	-	_		_	H	<3.3	H	66,	ļ	300			3	5	63.3	<3.3	<3.4
	-	H	H	┝	H	682	t			900	2	3	63.3	<3.3	<3.3	<3.3	<3.4
+C (Lindane)	┝	t	t	╁	t		200	3	3	436	<3.4	<3.4	<3.3	<3.3	<3.3	<3.3	\$34
	┝	t	t	t	t		†	55	3.4	<3.6	<3.4	<3.4	<3.3	<3.3	<3.3	633	5
Heptachlor Epoxide (a)	╀	t	t	+	3	31	+	<3.3	434	<3.6	<3.4	<3.4	<3.3	<33	<3.3	66.	
	+	t	†	+	\$3.4	63.3	<3.3	<3.3	<3.4	<3.6	<3.4	43.4	<333	633	193		
	+	• 2	9	+	┪	<3.3	٦	<3.3	<3.4	<3.6	<3.4	<3.4	533	56	,		
ychlor	+	+	+	┪	5	63	<3.3	<3.3	634	3.6	\$60	534	293			300	2
	200	†	40	+	+	<3.3	۲	<3.3	43.4	436	<3.4	<34	667			200	3
hene	+	\dagger	+	+	┥	ဗွ ဗ		<3.3	43.4	3.6	465	34	633		36	555	3
Arrelive (1999)	SV.	8	3	+	┥	633	<3.3	<3.3	34	636	3.5	34	633	,	3 6	200	0
	+	+	+	+												8	è
	> 0 // O/	+	9	ठु	ठु	SS SS	<33		ठु	8	Š	15.3	Ę	5,	100	1	
	+	3	+	\forall	लु	8	8	H	9	85	\$	5	2	۶	3 8	3 8	3
	2	†	┪	+	₹ -	ଞ୍ଚ ଞ୍ଚ	8	H	ಕ್ರ	80	ŝ	25	25	۶	3 6	3 8	3
	+	1	1	-	×34	8	رغ ج	H	50	85	75	12/		3 8	3 8	3	3
PCB 1260	4	, 26,		Н	34	8	8	t	25	8	3 3	3	3 8	3 5	3	Š.	ন্ত
	4	H		\vdash	25,	85	ŝ	8	5	3 8	5 3	3	3 5	3	8	ଞ୍ଚ	हु
+				-	_		1	\dagger	†	1	5	3	33	8	8	ફ	ŝ
2.4 F. Trichhomehanner and Land Carlot (Sivex) 49/	┪	+	<14 -	¢10	Н	ر د ا 0	ş	+	Ę	Ę	Ę	\$	۶	١	1		
71.	+	1	-	Н		H	┞	╁	Ę	ŧ	: 1			₹		00	ş
1 S. D.C.D.A.	┥	1	4	Н	<10	H	┝	\$	9	Ę	:	2		₹	₽	9	9
	, py/Kg	1	4		Н	H	H	╁	e e	=	1	2 5		₹	₹	\$	휭
A Missessian Management and (2,4 - UB)	ᅱ	4	Н	Щ	-	┝	H	t	Ę	5		?		3	₽	Ş	위
	┥	1	Н		<10	\vdash	┞	ş		1		2 5		₹	₹	9	ş
	4	<10	_		┡	\vdash	H	╀	15					₹!	2	95	ş
	┥	Н	<14	<10	H	┝	ş	18		† 非			2 5	₽;	₽	ş	ş
	-	<10		-	┝	\vdash	╁	╁	: 5				2	₹	DI.V	9	ş
	_	<10 <	H	H	╁	H	╁	+		:	₹	₽	9	ş	ş	<10	<10
	╀	H	t	╀	\dagger	+	+	2 5	2	Į.	ş	ş	ş	¢10	<10	<u>چ</u>	ê
	VO/K0	012	╁	9	Ę	Ę	+	+	→	Į.	₽	₽	윙	ş	<10	<10	¢10
prop [2-(2,4-Dichlorophenoxy)proponic acid]	Н	H	H	╀	t	+		+		 -	₽!	₽!	ş	ş	ş	<10	<10
0	┝	H	┢	t	╁	╁	\dagger	\dagger		†	9	₽	95	ę,	왕	<10	<10
	┢	H	₹140 ×	8	8	812	+	2 5	+	†	8	ş	ş	ę	<10	H	<10
	Н	×100	H	\vdash	t	╁	() ()	\dagger	+	†	3	8	2 2 8	ş 8	\$ 5		د100 د
Pertachlorophenol yo/Kg	Н	H	┢	t	t	+	+	+	3 5	₹	8	8	8	8	<100	Н	د100
Prooram Mg/Kg	H	<10	┝	H	t	╁		+	$^{+}$	†		₽ -	ş	ş	ş		<10
					ł	1	\mathbf{I}	1	\forall	1	200	2	\$10 \$10	~10 ~	<10	ę	٠ <u>٠</u>

| = value is between the Method Detection Limit and the Practical Quanitation Level
J = value is below the established limit for accuracy
** = 8310s were diluted (1:5)
** = 8310s were diluted (1:10)

		SSC028 07/07/98	SSC029 07/07/98	SSC030 06/29/98	SSC031 06/29/98	SSC032 06/29/98	SSC033 (SSC034 8	SSC035 8	SSC036 -	SSC037 SSC038 • 06/16/98		SSC039 06/16/98	SSC040 06/29/98	SSC041 8	SSC042 8	SSC043 06/30/98
				1	1	1	1	1	1	1	1	1		1			
PAH (8310)				I											٦		
1 - Methylnaphthalene	10/Kg	¢	Ş	8	5	Ş	÷	(-	- 26 - 26	(B)	<85	41 2	41 >	Н	/ >	Ę
2 - Methymaphthalene	DVK0	Ş	623	×18	Ş	<1 <u>/</u>	<17	Ş	÷	<92	8	85	~1	<17	<17	<17	<17
Acenaphthene	58 84	Ş	Ş	ş	Ş	Ş	÷	 	5	ŝ	86	685	<17	<17	Н	<17	<17
Acenaphthylene	WK9	43.4	4.6	34	43.4	<33	633	<u>ئ</u>	434	<18	<17	<17	<3.3	<3.3	_	<3.3	<3.4
Arithracene	wo/Kg	\$	\$	8	8	5	<17	<17	<17	-91	-86 -	<85	<17	<17	Н	<17	417
Benzo(a)anthracene	<i>μ</i> ο⁄Κα	<u>^-</u>	<2.3	41.8	<1.7	<1.7	<1.7	<1.7	<1.7	<9.2	<8.6	<8.5	<1.7	<1.7	<1.7	<1.7	<1.7
Benzo(a)pyrene	My/Kg	<2	63	<2	<2 -	<2	<2	<2	<2	6>	<8	<8	<2	~	\$	2	8
Benzo(b)fluoranthene	νο⁄Κα	લ	64	ç	<3	جع	جع	<3	<3	<19	<18	<17	<3	Ş	8	6	8
Benzo(g.h,i)perylene	1/0/Kg	<3.4	<4.6	<3.5	<3.4	<3.4	<3.3	<3.3	<3.4	¢18	412	<17	43.4	43.4	t	43.4	34
Benzo(k)fluoranthene	, VO.	ş	ಽ	3	2	?	2	2	ş	60	80	\$	2	2	~	2	9
Chrysene	W/Ko	<17	<2.3	<1.8	<1.7	<1.7	<1.7	4.15	<u></u>	<92	68.6	<8.5	<1.7	7.12	t	712	÷
Dibenzo(a,h)anthracene	MQ/Kg	<3.4	<4.6	<3.5	<3.4	<3.4	<3.3	<3.3	34	\$ 8 5	41>	47	<3.4	<3.4	434	t	34
Fluoranthene	₽Ø/Kg	43.4	<4.6	<3.5	43.4	<3.4	<3.3	<3.3	43.4	ê	45	417	43.4	434	H	T	34
Fluorene	MO/Kg	43.4	4.6	<3.5	43.4	43.4	433	43.3	<3.4	₽	Ę	15	434	434	t	t	7
Indeno(1,2,3-cd)pvrene	οχο _ν	12	623	87	212	1=	- 112	/÷	12	<92	68.6	85	\ \ \	Ę	$^{+}$	t	1
Nachthalene	ρχο	Ę	\$3	800	Ę	\ \{\bar{\pi}	15	115	15	8	\$	£	Ę	Ę	\dagger	\dagger	1
Phenanthrene	io/Ko	47	<23	¢1.8	1-2	Ę	÷	Ę	Ę	ê	88	28.5	1	 -	t	 	 -
Pyrene	io/Ko	47	<23	418	47	7	7	7	1	Ş	86	285	1	1	t	t	T.
Surrocates											3	3			1	┨]
9.4.DCAA	ş	5	7,6	5	2	2	S	5	-	2	5	a	961	8	150	5	Ş
24 K TOUY	8	2	2 6	3 2	*	: 5	3 2	3 2		2		3 2	3 6	8 2	3 8	,,,	
CHO LOGO	e	۶	i c	5 6	: 8	2 2	3 8	3 5	٦	5 6	\$	3 3	١	8	K	3 :	8
Damped	اة اع	2 2	2 8	2 8	8 5	8 8	3 8	2 %	8 5	9 6	3 4	2 4	S a	9	8	2 5	2 5
Mainte		3	3	ŝ		1	8	3	2	í	3	2	8	23	*	3	3
	1	ļ		,	,		1				ļ						
Aumoum	DVA	3	3 €	8	g ·	3	ş ,	8,	\$	2	=	8	3	2	2	8	2
Antimony	MO/KO	ð	Ş	3	V	2	2	8	8	7	ş	2	3	\$	^ 2	2	7
Arsenic	mo/Kg	\$ \$	607	ŝ	Ş	c 0.5	\$0.5	<0.5	c05	\$0.5	6.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Barum	mo/Kg	\$	8	Ş	Ş	Ş	ş	۶Ş	S S	22	<u>چ</u>	<20	<20	<20 >	<20	<20	<20
Beryflium	mo/Kg	v	v	v	ē	v	Ψ	Ÿ	Ţ	Ţ	Ţ	V	ī	۲,	Н	دا	ŗ
Cadmium	mg/Kg	Ÿ	v	v	₹	٦	ē	Ţ	V	ē	v	v	Ţ		-	ī	Ş
Calcium	шФЖа	20	8	ş	8	230	2	5	47	130	8	8	2	110	Н	160	980
Chromium (total)	mo/Kg	v	⊽	v	⊽	Ţ	╗	₹	⊽	-5	Ţ	V	V	v	Н	4.2	~
Cobalt	mo/Ko	ç	70	ş	\$	Ş	ş	\$	ş	\$	Ş	٥ د	v	╗	┥	ŝ	ŝ
Copper	mo/Ko	Ş	\^	\$	Ş	Ş	Ŝ.	\$	Ş	ç	\$	\$	ŝ	┪	┥	ŝ	ŝ
Iron	MOK O	3	24	5	2	23	8	220	23 ;	8	8	21.0	3	22	+	2800	80
D#91	S S	•	V	?	;	2	3	,	2	2	=	+	Į.	7	+	*	9
Magnetin	MOM	\$	3	\$	2	2	87	8	ş	3	ş	+	ş	7	+	8	2
Manganese	morka	22	6.	2	5.5	2	5	2	~	-	V	7	2		┥	┥	3
Mercury	πOΚG	٥٥ 10	¢0.01	6 0 0	<u>\$</u>	Ş 5	6 0.0	<u>6</u>	<u>0</u>	6 0.0	60 20	7	6.0 20.0		Н	\dashv	<0.01
Nickel	mo/Ka	<5	<7	Ş	ŝ	ş	Ş,	\$	ç	\$	< 5	٦	<\$		-	جۇ	Ş
Potassium	mo/Kg	₹ 5	3	د170 د	<170	¢170	<160	<160	<170	30	<26	Н	<25		-	<170	<170
Selenium	m o /Kg	ح2	<3	<2	<2	42	42	<2	<2	<2	<2	<2	<2	Г	⊢	2	3
Silver	оууош	<2	<3	<2	<2	<2	<2	<2	42	7	<2	<2	2	Г	3	2	3
Sodium	5y/6w	141	161	151	111	16	121	9.51	101	<7.1	1.7.1	7.01	9.9>	ā	Z	171	9
Thailium	ōy∕6w	۷,	~ 1	<1	<١	<۱	<1>	<1	<1	<1	<1	⊽	₽	Г	ē	⊽	⊽
Vanadium	0У/бш	دا	1	Ş	į	دا	1	1	<1	2	<1	۲	<1	⊽	2.8	4.2	2.5
Zinc	mg/Kg	<5	د/	10	16	6.9	6.7	8	<5	<5	<\$	\$>	Ş.	5.7	2	6.3	.

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy = = 8310s were diluted (1:5) = ***-8310s were diluted (1:10)

219

Table 8-15. KSC Background raw data for soil focations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC028 07/07/98	SSC029 07/07/98	SSC030	SSC031 06/29/98	SSC032 06/29/98	SSC033 06/29/98	SSC034 06/29/98	SSC035 06/29/98	\$\$C036 •	SSC037 06/16/98	SSC038 * SSC039 06/16/96 06/16/98		SSC040 SSC041 06/29/98 06/30/98	SSC041 8	SSC042 06/30/98	SSC043
Other Bergmeters								1									
							_							ŀ	ŀ	l	I
Bulk Density	o/cm²	1.1	1.2	0.93	0.75	98.0	-	-	-	ŀ	150	2	į			1	
CEC	meo/100a	•	-	·	-	,	ļ	·) O	ŝ	5/3	8	0.94	1.2	
Persent Solice	,	į		1	•	7	,	,	2	S	တ	•	7.	e	9	-	,
	*	à	ž	88	8	8	8	8	8	91	1 26	8	8	8	8	1 8	, [
(OB)	o Si	5.4	5.6	5.3	8.8	8.2	ļ	,	3	3		†		3	8	S	8
Resistivity	ohm-cm 21000	210000	- COUNTY	;	5		,	,	3	9	•	•	2	2	6.5	6.2	6.3
Taxture (No. 4)					,	Į	ž	715	<12	130000	130000	130000	150000	<12	<12	2500000	5000
	,	3	8	9	8	6. 66	2	5	100	5 66	2	٤	٤	٤	t		3
I EXTURE (NO. 10)	×	8	666	8 66	600	g	000	8	5	2		3	3	3	3	3	/88
Texture (No. 40)	3	8	8	1	1			88.9	63.5	ŝ	ŝ	99.68	100	606	6.00	8	1.00
Taxture (No. Ro)			200	Ř	,,	B	4.8	99.3	89.4	97.7	- 8	93.6	9.96	94.6	187	5	£3 £
	ŗ	2.86	97.9	97.6	28	828	97.3	286.5	- 28	936	87.0	- 2	7 00	8			3
Pexture (No. 100)	*	50.3	657	765	£1.4	·	1	9			;	5	8	Š	-	20.2	51.3
Texture (No. 200)	,	33					7	2 .	Š	8	31.5	26.6	27.2	62.7	5	20.2	23.4
Total Omanic Carbon	_					8.5	6	6	2.9	*	3.5	2.8	2	6.8	5.5	8.5	3.7
	MONG	30/02×	20/92	>26/00	×26700	>26700	>26700 >26700		>26700	100736	2000	00726	500	ķ			;
				-							•						

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy = = 8310s were dikined (1:5) = 8310s were dikined (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC044	SSC045	SSC046	SSC047	SSC048		SSC049	SSC050	SSC051	SSC052-	SSC053	SSC054	SSC055	SSC056		SSC057
		200	200	200	as as a	5	Cate				96/00/30				0	olicate	800
Organochlorine Pesticides (8081)					Ī										Ī	İ	
4.4. DDD	ø√Ka	<3.3	<3.4	<3.5	34	<3.5	┢	43.4	4.63	465	<3.4	<4.5	\$. \$. \$.	1 60	<333	434	5
4,4' · DDE	W/Ko	<3.3	<3.4	<3.5	43.4	H	43.4	43.4	<3.4	<3.4	43.4	4.5	5.5	43.4	633	34	3.4
4,4' - DDT	0X/0/	<3.3	<3.4	<3.5	43.4	┝		<3.4	43.4	43.4	43.4	<4.5	43.4	460	63.3	465	4.65
Adnin	DA/W	<3.3	<3.4	<3.5	<3.4	H	H	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	4.65	<3.3	43.4	3.4
Apha - BHC	63/6 <i>1</i>	<3.3	<3.4	<3.5	<3.4	<3.5	H	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	<3.4	<3.3	<3.4	<3.4
Beta - BHC	W/Kp	633	43.4	<3.5	43.4	┪	1	50	ş	<u>د</u>	3	<4.5	Ş.	<3.4	<3.3	<3.4	<3.4
Chlordane (alpha)	νο⁄K ₀	<3.3	<3.4	<3.5	<3.4		_	43.4	<3.4	<3.4	<3.4	<4.5	<3.4	<3.4	<3.3	<3.4	<3.4
Chlordane (gamma)	₽9/Kg	<3.3	<34	<3.5	<34	Н	\forall	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	<3.4	<3.3	<3.4	<3.4
Chlordene (total)	po/Kg	<3.3	<3.4	<3.5	<3.4		Н	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	<3.4	<3.3	<3.4	<3.4
Detta - BHC	υσ/Kg	<3.3	<3.4	<3.5	<3.4	-	-	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	<3.4	<3.3	434	465
Dieldrin	DW/W	<3.3	<3.4	<3.5	43.4	┝	H	43.4	<3.4	434	460	¢4.5	4.65	43.4	<3.3	<3.4	<3.4
Endosultan I	₩o/Ka	433	<3.4	<3.5	<3.4	-	H	3.4	43.4	43.4	3.65	<4.5 4.5	34	43.4	\$33	43.4	3.4
Endosultan II (beta)	MO/Kg	<3.3	43.4	<3.5	3.4	-	H	43.4	43.4	43.4	434	<4.5	434	43.4	<3.3	43.4	460
Endosultan Sultate	W/Ko	433	43.4	43.5	460	H	H	34	43.4	43.4	<3.4	۸4.5 د	34	465	<3.3	43.4	₩ 9.6
Endrin	VO/Ko	<33	43.4	<3.5	43.4	H	H	43.4	43.4	<3.4	434	c4.5	34	4.65	43.3	4.65	43.4
Endrin Aldehyde	νο. Vo.	<3.3	<3.4	<3.5	43.4	-	H	<3.4	<3.4	43.4	43.4	¢4.5	434	465	c3.3	465	4.65
Endrin Ketone	₩o⁄kg	43.3	465	435	34	H	H	48,	43.4	43.4	43.4	4.5	43.4	434	<3.3	460	3.4
Garnma - BHC (Lindane)	0X/dv	ć33	43.4	<3.5	43.4	┞	H	43.4	<3.4	434	34	<u>د4.5</u>	434	43.4	<3.3	34	43.4
Heptachlor	MO/Ko	<3.3	<3.4	<3.5	<3.4	H	-	<3.4	<3.4	<3.4	<3.4	<4.5	43.4	43.4	<33	4.0	43.4
Heptachlor Epoxide (a)	py/Kg	<3.3	<3.4	<3.5	<3.4	j	-	<3.4	<3.4	<3.4	<3.4	<4.5	43.4	<3.4	<3.3	<3.4	43.4
heptachior Epoxide (b)	₽V/Q/	<3.3	<3.4	<3.5	<3.4	\vdash	<3.4	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	43.4	<3.3	43.4	43.4
Isodrin	8X/S/	<3.3	<3.4	<3.5	<3.4	-	-	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	<3.4	<3.3	<3.4	43.4
Methoxychior	07/Q/	<3.3	<3.4	<3.5	<3.4	\vdash	H	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	43.4	<3.3	<3.4	43.4
Mirex	νον γ	<3.3	<3.4	<3.5	<3.4	<3.5	<3.4	<3.4	<3.4	<3.4	<3.4	<4.5	<3.4	<3.4	<3.3	<3.4	<3.4
Toxaphene	μο/Kg	~ 67	<67	د/2	89			8	8	<67	<67	œ,	<67	×68	<67	<67	<67
Aroctors (8082)						┪											
PCB-1016/1242	MO/Kg	ફ	Ş	ફ	ठु	\dashv	<u>ئ</u>	ह	ह	ह	रु	¢45	ج ع	\$	جي جي	<34	\$
PCB-1221	WW.Kg	<u>န</u> ်	Ş	şŝ	ਨ੍ਹ	\dashv	रु	ठु	हु	ફ	हु	\$\$	ु ह	ङ्	လ ို	रु	ŝ
PCB-1232	₩o/Kg	8	रु	8	इ	+	रु	ङ्	₹	ह	रु	\$	ਨ੍ਹ	ह		हु	9
PCB-1248	W/Kg	8	ह	ş	₹	<u>چې</u>	हु	ह	ह	Ş	हु	\$	ह	रु	ဆို	रु	\$
PCB-1254	SX.	8	Ş	8	हुं।	+	5	ह्य	8	8	Ş	\$	ह	₹ 	₩ ₩	₹ -	9
	MO/Kg	8	Ş	ş	हु	\dagger	इ	\$	ş	\$	ह	\$	ङ्	ह	ŝ	8	Ş
Chlorinated Herbicides (8151)					,		1	1	1								
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	MO/KG	₽	Ş	į	Ş	₹ Į		₽!	ş	Ş	₽!	*	ęļ	₽ P	Ş	₽!	Ş
2,4,5- Inchlorophenoxy acetic acid (2,4,5 - 1)	W.V.	Ş		₹	2 5	Ę ;	2 5	2 5	2 5		۶	*	2 5		₽ 5		2
2.4-United by the roxy about about (2,4 · U)	200			1		 -	 	***	*		2 5	**	*	*	3 8) 	
4-(2.4-Dichlorophenoxy)butyric acid (2.4 - DB)	, OXO	ę	92	Ę	ę	Ę	ę	Ş	ę	9	ę	*	ę	100	Ş	₽	Į Į
4 - Nitrophenol	W/Kg	ę	9	Ę	ę	Ę	ş	ş	9	ę	ę	=	ę	ş	ę	ş	ş
Acfluorien	00/Kg	<10	<10	<11	<10	<11	<10	<10	<10	<10	<10	<14	<10	¢10	¢10	<10	<10
Bentazon	₽/Kg	<10	<10	<11	<10	<11	<10	<10	<10	<10	<10	<14	<10	<10	<10	<10	<10
Chloramben	W/Kg	<10	<10	<11	<10	<11	<10	¢10	<10	<10	<10	<14	<10	<10	<10	<10	<10
Decthal	NO/Kg	<10	<10	<11	<10	<11	<10	¢10	<10	<10	<10	c14	<10	<10	<10	<10	<10
Delapon	MO/KQ	<10	<10	<11	<10		<10	<10	<10	<10	<10	<14	<10	<10	<10	<10	<10
Dicamba	1/Q/Kg	√10	ot>	Ų	¢10	داا	¢10	ę,	¢10	¢10	9	<14	¢10	د د	د10 د	¢10	<10
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	1/2/Kg	ç 9	95	Ę	¢10		ę	ê	ę	ęş	¢10	41	¢10	¢10	\$10	¢10	<10
Dinoseb	νο. V	ş	ş	Ę	ş	┪	ş	ş	ş	ş	ş	Ş	ş	ş	ş	ş	<10
MCPA	νς.	<u>\$</u>	8	<u>د</u> اه	8	†	8	8	8	8	8	2 1 60	8	8	8	8	8
MCPP	S Q	8 ?	<u>8</u>	<u>و</u>	8	1	ş,	<u></u>	8	8	8 V	9;	8	<u>ş</u>	<u>\$</u>	<u>ş</u>	8
Pentachlorophenol	ξ. (2)	ş	ę	₹	ş	₹ :	ę;	ę;	e e	ę,	ş	7	ş	el:	ş	ş	ę;
Pictoram	MA.	2	Q.V	ē	OLV	1	QIQ T	γıΩ T	410	010 T	OL>	<14 L	410	010	012	010	9

1= value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy * = 8310s were diluted (1:5) * = 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil focations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC044		SSCOAR	COCONT.	66004		9,000		r							
		06/30/98	96/06/90	96/06/90	96/30/36	6/30/98		200000	06026	SSC051	SSC052-	55053		SSC055	SSC056	5	SSC057
						Ď	dicate	}		_			0 96/90//0	18690/	7,6798	_	96/90/
PAH (8310)							1	I	t	I	1	ı			4	duplicate	
	WO/Ko	ļ	ķ	ļ	ļ	t	1	1	1		_					ŀ	Ī
2 - Methylnaphthalene	٤		3 4		 	†	+	1	H	41	<85	<23	Į	Į	ŀ	t	Į
Acenaphthene	S)/ci		3			†	+	+	1	\		<23	<u> </u>	4	t	150	
Acanaphthylene	1				į	1	+	1	┪	<17		423	÷	-	t	t	
Anthracene						0.53	43.4	\$34	\dashv	<3.4 -	<17	<4.5	4.65	100	t	t	
Benzo(a)anthracene			Į,	₽:	8	+	+	1	<17	<17	H	\$3	H	8	Ę	t	
Benzo(a)pyrene			600	= -	Ì	+				<1.7	<6.5	23		t	ŀ	t	
Berzofbillucranthana)	ş	2	%	-	_	7	<2	7	f	t	t	t	\ 	†	
Benny b Incorpora	100 KG	2	<17	<3	ç	_	┝	H	H	S	1		,,	, , ,	†	+	9
owickly, it, spory years	40/Kg	<3.4	<17	<3.5	<3.4	43.6	43.4	t				t	†	+	5	ç	ç
Derzokymuorantnene	MO/Kg	?	85	ş	Ş	t	t	t	t	;	†	0.5	+	1	7	_	34
Chrysene	0X/04	<1.7	<8.5	7		t	\dagger	†	†	y i	V	1	-			H	8
Dibenzo(a,h)anthracene	MO/Ko	23.6	1	35,			†	+	Ž	\ \?	1			-	41.7	H	-
Fluoranthene	SWC.	,			; ;	+	†	1	1	<3.4	_	<4.6	Н	┝	t	t	1
Fluorene	١			2	3	+	486	<3.4 <	٦	<3.4	<17	\$ 6.6	<3.4	43.4	t	t	
Indeno(1.2.3-edhwana	2	3	V	43.5 43.5	3.4	-	-		<3.4	<3.4	H	t	t	t			
Nachthalana	WAY	ì	<8.5	41.8	<1.7	<1.8	<1.7	<1.7 ×	┢	Ę		t	t	\dagger	†	†	
Division in the state of the st	W.Ko	<17	<82	<18	<17	\vdash	ŀ	f	t	1	t	t	†	†	1	1	Ş
	MO/Kg	<1.7	<8.5	×1.8	<1.7	t		t	t		t	3	+	1	1	1	417
Pyrene	MONKO	<1.7	<8.5	18.4	12	8	t	t	t	+	†	+	+	7	<1.7	Н	<1.7
Surrogates		L				1	1	1) 	90	\dashv	<1.7	<1.7	<1.7	<1.7	1. 1√
	×	98	09	9	2	925	ŀ	,	}								
2,4,5,6-TCMX	¥	ž	5		*		+	ę	3	ž	2	88	76	96	2	8	æ
280		3 2	3 4	B	₹;	ž i	5	2	8	5	74	48	22	22	82	7	5
p-terphenyl	8		3	8 2	•	B	+	8	8	2	78	76	2	8	8	2	£
Metais		2	,			Se	\dashv	E	8	88	0	38	8	2	2	1 2	٤
Alminum	Sylven	989	1	389		ŀ	ł	ł	ļ			ı					3
Antimony				3	3			-	100	9 <u>9</u>	-	┝	H	H	۲	ŀ	5
Arranic	S I	ď	3	2	\$	٦	_	H	┝	┞	H	╀	╀	╀	3	+	₹,
The state of the s	S S S S S S S S S S S S S S S S S S S	60.5	<0.5	<0.5	<0.5	<0.5	<0.5	┞	\$0.5	205	1	, ,	*****	,	†	y	V
	₽ Q¥	8	<20	ح2ا	2	H	H	╁	t	╁	\dagger	+	4	+	Q.5	+	9.5
Denymum	mo/Kg	<1	<1	⊽	⊽	H	╁	╀	+	\dagger	+	\dagger	4	╅	1	┥	8
Cadmium	mg/Kg	1>	ī	H	ī	t٠	╀	+	+	 	1	- 1		4	٦	4	⊽
Cacum	mg/Kg	190	510	┢	2002	099	┾	50	+	1	T	- 1	⊽	4	Į	4	v
Chromium (1008)	morKg	3.2	1>	H	=	t	۲	+	╁		†	ı	. 1	4	7	\dashv	1000
Cooper	mo/Kg	ŝ	<5	Н	₩	t	┢	+	╀	+	+		- 1	-	7	4	÷
laction .	mg/Kg	\$	\$	-	_	H	┢	۲	╁	╀	+	┿	+	+	┪	┥	v
TOIL	mo/Ko	28	38	Н	щ	1	┢	┿	╁	╀	$^{+}$	┿	+	e E	┪	+	5
Warnesim	SXO.	53	2	\$	1	3.4	4	3.9	9	5.5	9	7.	十	4-	3,	4	8
Mercener	morko M	22	28	_	_		⊢	⊢	┞	╀	t	+	╀	†	+	4	
Mary	S YOU	ş		-	Ц	H	┝	╁	╁	╀	t	┿	+	<u> </u>	+	4	8
Mercury	E X	<0.01	<0.01	┢	L	E	┢	╁	Ë	╀	╁	+	+	†	+	4	2
NCKE	mg/Kg	\$>	ŝ	⊢	┺	۲	╁	╁	+	\dagger	+	+	4	+	7	┥	<0.01
Polessum	по/Ко	<170	8 1.0	Ļ	↓_	8	+		t	+	†	+	+	┥		_	55
Selenium	⊒O/Ka	2	\$	t	┸	+	3 3	\dagger	B	\$	+	1	4	\dashv		Н	2
Silver	Ş	2		ļ.	,,	\dagger	+	+	+	\dashv	+	٦	_	_	Г	L	8
Sodium	S C	-		, ,	y ;	Y	٧ ارد	V	┨	-	+	H	L	H	T	L	
Thefficm	S S		,	 		+	+	_	4	Н	141	S	1900	300	151	3	8
Vanadium	J. Day		\ \	,	 	⊽\ ₹ :	▽	+	Į	-	Н	Н		╀	T	Ļ	Į
Zinc	Ž	, 2	*	3 5		+	5.	+	\dashv	3.4	V	2.5	\vdash	┞	H	╀	2
				?	2	+	+	ě	\dashv	\dashv	\dashv	Н	\$	ŝ	\$	\$	Ş.

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy = 8310s were dikited (1:5) = 8310s were dikited (1:10)

Table B-15. KSC Background raw data for soil locations. *** x value is an outlier and not used in the analysis to develop the KSC Background screening values.

	100000	-	CECCOAE CCCOAR CCCOA7	550007	SCOMB		670055	SSC049 SSC050 SSC051		SSC052-	SSC053	SSC053 SSC054 SSC055	SSC055	SSC056		SSC057
	90000	_	90000	06/30/98	86/30/88		96/06/30/36	96/06/9		96/30/90	96/90/20	07/06/98	96/90/20	7/6/98	Ĭ	07/06/98
	<u> </u>	_	_			Olicate	_							٥	duplicate	
					I			T						_		
Other Parameters							I	ļ	ļ	33,0	į	ŀ	:	30,0	-	12
O. t. Ossethy	66.0 cm2/o	=	66.0	=	98	0.82	0.93	-	ŝ	3	ŝ	•		3	†	
DUIK Lienary	L	•	6	ľ	,	5	2	8	6	-	က	2	3	<u> </u>	-	-
CEC	1			8	a	8	6	8	8	8	73	86	- 26	8	8	8
Percent Solids	+	+	S.				ļ		9,	ļ	2.3	~	7 9	٠	1 95	2
AH (lah)	S.U. 6.5	9	6.5	9 0	6.5	٥	;	ŝ	•				2000	,	2000	200
M (1997)	23000	9	12300000	<12	220000	2300000	0	2300000	13000001	0	800000	1400000	/30000	300000	3	3
Hesistivity			٤	٤	8 00	٤	8 88	6 66	1 66	100	9	4.28	ī	8	8	8
Texture (No. 4)	9.00 0.00 0.00	4	+	3			1	1	ç	8	8	689	78.7	8	2	97.2
Texture (No. 10)	8.68 - %	9.66 — 8	8	8.66	28.7	ŝ	È	Š	Š	99.0	99.9	3				
Tektura (190. ro)		70.5	85.4	7 79	74.6	74.4	78.9	6.98	46.7	ક્ક	57.9	36.4	43	75.3	2	ķ
Texture (No. 40)	†	+	╀		2 63	R2 R	A. S.	55.3	38.2	83.8	8.64	28.9	35.4	56.4	55.8	60.0
Texture (No. 60)	8 47.5	-	+		3					9 90	10.5	-	-	22.7	101	187
Takes (No. 400)	17.1	16.6	7.0	21.4	24.6	21.4	0.0	51.5	•	8						
I BANUTA (190. 100)	t	,	33	8.4	3.6	3.6	3.9	8.4	4.5	2.5	3.8	1.3	2.7	2.8	2.5	٦
Texture (No. 200)		t	ľ	ľ	00/26	0029C		\$26700 \$26700		>26700	16600	>26700	>26700	>26700 >26700 >26700	>26700	×26700
Total Omanic Carbon	MO/KG - >20/02	M/974 M		200100	300	3										

Table B-15. KSC Beckground raw data for soil locations *** = value is an outlier and not used in the analysis to develop the KSC Beckground screening values.

		SSC058	SSC059	SSC060	SSC061 S	s zecoes	SSC063**	SSCOGA		SC065 S		CD67 CC	0000	00000		
		3	8	9600/0	0 86/92/9		96/20/38	96///		90/0/20	06/22/98 07	07/27/98 07/2	07/27/98 07/27/98	38,000	2000	SSC072
Organochlorine Pesticides (8081)	L		ľ	T	\dagger	1	1		duplicate	1	_					96/90/0
000 · */*	WO/Ka	Γ	T	1	t	╁	t	1	T							Ī
4.4 · DDE	Sylver		T	?	+	+		Н	H	H	H	<36	t	ļ		
4,4' · DDT		T	T	3	7 7	\dashv		-	H	H	t	t	t	3	\$	6.3
Adrin	2	550	7	┪	1	<3.6	<4.6	H	43.6	t	3.5	t	t	80	<4.5	*
Alpha - BHC	NA.	T	7	┪			H	t	t	t	t	t	†	43.9	<4.5	4
Beta - BHC	o vo	7	┪	┥	H	H	H	865	436	1	\dagger	\$30 \$30 \$30 \$30 \$30 \$30 \$30 \$30 \$30 \$30	29	<3.9	<4.5	\$
Chlordane (alpha)	NO.	٦	٦	-		H	┢	t	t	t	\dagger	†	+	<3.9	<4.5	*
Chlordana (gamma)	WO/KG	<3.3	┪		Г	┝	t	t	+	t	2	†	+	<3.9	<4.5	*
Chlordene (total)	70/KG	┪		Н	T	┝	t	t	t	t	1	+	+	<3.9	<4.5	\$
Cele But	νς. Vo.	<3.3		┝	T	╀	T	t	+	†	7	+	\dashv	<3.9	<4.5	3
The Park	MO/Kg	<3.3	H	H	44	t	t	+	+	7	┪	1	Н	<3.9	<4.5	
Chargen	VO/Ko	T	۲	t	t	+	†	1	┥			Н	┢	43.9	377	
Endosurian	ν. V.	t	t	+	†	┪	7	┪	٦		Г	t	t	1		;
Endosultan II (beta)	νο/Kα	5	+	+	v V	┪	7	Н	Н		T	t	t	200		ş .
Endosultan Suffate	, o, Ko	t	†	+	†	+	7			Г	t	H	t			•
Endrin	03/0/		$^{+}$	+	V	┪	┪	Н	Н	T	t	╆	t	600	\$	\$
Endrin Aldehyde		†	+	+	7	\dashv		H	┝	T	t	t	t		<4.5	Ž,
Endrin Ketone	2	†	+	┪	1	_	٢	┝	⊦	T	✝	$^{+}$	†	655	<4.5	4
Gamma - BHC (Lindane)		+	+	┪	+	Н	┢	╁╴	╁	T	+	╁	†	23.9	<4.5	×4
Heptachlor	2	+	+	┥	+	Н	H	┢	╁	T	+	╁	†	<3.9	<4.5	*
Heptachlor Epoxide (a)	O L	+	200	<33	<4.4	<3.6	<4.6	<3.9	<3.6	165	36	400	437	<3.9	<4.5	7
heotachlor Enoxida (h)	MAN	<3.3	┥	\dashv	_	┝	H	╁	t	T	+	+	+	<3.9	<4.5	4
sodrin	MO/Kg	┪	_	-	H	┝	╁	╁	t	t	+	+	1	<3.9	<4.5	*
Methorophoc	00X0	<3.3	Н	Н	<4.4 <	⊦	t	╀		†	+	╅	+	<3.9	<4.5	7
Miss	MVK0	٦	Н	⊢	┢	┝	╁	╀	+	†	+	+	1	<3.9	<4.5	\$
Toxantana	WK0	٦		<3.3	┝	<3.6	╁	30	\dagger	†	+	<3.4	+	<3.9	<4.5	2
Aroctore (Bres)	0X/07	\$	\dashv		<4.4	├	t	t		t	+	+	+	<3.9	<4.5	\$
PCB-1016/1242		1			-	H	H	t	t	†	2/2	†	+	<78	06>	80
PCB-1221	DX/O/	នុ	ठ	<33	Н	×36	t	t	\dagger	†	+	\dagger				
PCB-1232	OV.	+	1	┪	-	_	H	H	t	t	t	\dagger	è	S S V	45	<40
PCB-1248	//O/Kg	+	+		× 444	\vdash	┝	╁	t	\dagger	\dagger	\dagger	è	889	<45	**
PCB-1254	DVK9	1	-		-	\vdash	H	t	t	\dagger	t	+	έŞ	×38	<45	·40
PCB-1260	10Kg	ဂ ္ဂ	Н	Н	\vdash	\vdash	╁	\dagger	t	t	+	+	^{<37}	<39	<45	6
Chlorinated Marbicides (885)	69/Kg	+	+		H	┝	×46	80	3 8	3 4	8 8	8	637	<u>ئ</u>	<45	0 *>
2-(2,4,5-Trichlorophenoxylpropionic soid (2,4,5-TD) (Silver)	ļ	+	+	H	Н	\vdash	┝	╁	+	\dagger	+	\dagger	(3)	639	<45	<40
2.4.5-Trichlorophenoxy acetic acid (2.4.5 - T)	Day of	†	+	Ş	13 <11	Н		12 <11	+	-	-	\dagger				
2,4-Dichlorophanoxy acetic acid (2,4 - D)	2000	₹ 5	\dagger	+	+	+	<14		-	t	\dagger		=		÷	~
3,5-DCBA	O NOW	\dagger	\dagger	+	+	+	1		Н	-	H	+	ŧ	***	Ť	
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	P) Vo	t	2 8	 		\dagger	†	<12 <11	₹	1		ş	₹	100	十 非	712
4 - Nitrophenol	MO/Kg	H	t	t	2 5	+	\dagger	+	+	\dashv	Н	ڊ 9	41	<12	T	
Aciniconen	0W01	H	t	t		\dagger	†	\dagger	+	+		<10	Ę	<12	¥1.	1
Bernazon	W/Ka	T	t	\dagger	╁	+	†	╅	+	+			Ę	<12	t	1
Chloramban	0X/0/	t	t	t	\dagger	+	v	<12 <12	₹	7	<u>-</u>	Н	₹	<12	t	12
	W/K0	┢	T	t	\$13 \$13	+	\dagger	2	\dagger	₹	1		<11	<12	T	T _e
Composi	DX/6/	H	T	t	t	\dagger		\dagger	₹	+	+	Н	<11	<12	T	1
Control	53V6v/	ş	t	t	t	\dagger	t	\dagger	+	+	7	Н	<11	<12	t	1
Discount of (2,4-Dichlorophenoxy)proponic ecid)	ρν⁄κα	<10	t	┝	¢13	t	t	2 2	₹ `	+	1	┪	<11	<12		12
LA LOSSE	63/6 <i>H</i>	¢10	H	H	+	t	t	†	+	₹	7		<11	<12	t	2
MCPA 1	⊢	t	t	t	t	╁	†	†	+	7	Н	Н	Ę	<12	t	
W.C.P.	Н	H	t	\$100 \$1	0000	+	2000	25.5	8	₽ ₽	0220	Н	¢220	<240	t	92
Distriction	₽9/Kg	\vdash	-	┢	t	╁	\dagger	+	+	†	+	+	<220	<240	╁	9
r con em	μς/Kg	<10 ,	9	¢10 <13	t	╁	t	₹ ÷	₹ ·	+	1	ę	<11	<12	=	2
			ĺ	l		1		1	┨	₹	₹	ş	<11	<12	H	_≃
															ĺ	

1 = value is between the Method Defection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy * = 8310s were diluted (1:5) ** = 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

	•	CCCASA	CCCORO	Cecogn	Second		690055	SSCO		SSC065	88088	SSC067	SSC068	880069		SSC071**	SSC072
		07/06/98	96/90/20	96/90/20		07/27/98	08/20/98	7/7/98		07/07/98	06/22/98	07/27/98	07/27/98	07/27/98	08/20/38		96/90//0
							۰		plicate								
100000000000000000000000000000000000000		Ī	Ì	I	ľ	İ		Ī	I			l	l				Ī
PARIENT		ļ	ļ	ļ	667	ţ	ORC Y	8	Ę	٩	ļ	Ę	ţ	۶	ķ	08.62	02°
1 - Meinymagninavene	2						36		=		Ę	Ę	11,	710	200	2330	5
2 - Methytraphtmatene	No.				3 6		3 8			2	2 15	2 2	1	200	1	٤	8
Acenaphthene	Nov	*	Į.	÷	3		8097	3 6				36,	,	26		†	
Acenaphthylene	WVK0	3	3	3	* 6	3	\$ 15	3 5		3 8	2	200			3 5		, 8
Arthracene	W/W	è	į	₹	*	₽ ;	3 2	3	2	֭֭֚֓֓֓֓֓֝֓֓֓֟֝֓֓֓֓֓֟֜֓֓֓֓֓֓֓֟		֓֜֜֜֜֜֜֓֓֓֓֓֓֓֓֓֓֜֜֜֜֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓			3	3 5	
Benzo(a)anthracene	10 Kg	Ş	Ì	į	775	*	3	\\	• •	Ŷ,	2,	٥,	1	<u>"</u>		3 8	,
Benzo(a)pyrene	μο⁄Kg	\$	ç>	~	ç	8	\$2	ѷ	V	2	8	۶	2	25	25	ęş,	3
Benzo(b)fluoranthene	03V01	<3	<3	دع	*	လ	<46	v	ş	ş	Ş	ş	ş	2	₹	45	\$
Panzolo h imerviene	MO/Ko	43.4	<3.4	43.4	4.4	<3.6	9*>	45	<3.6	<3.7	<3.6	<3.6	<3.4	<3.7	<3.9	<46	*
December of the Company of the Compa	o)/or	ç	Ş	8	63	2	<24	ş	ş	2	2	2	2	42	8	423	6 2
Charles and a second a second and a second and a second and a second and a second and a second and a second and a second and a second and a second a	0)/0.7	1	1	12	33	87	53	2	×1.8	612	8	8.1×	7	8	5.6	<23	\$
Chrystere	2000	1				436	9	7	636	537	436	436	434	37	43.9	94	ş
Utoenzo(a,n)ammacene	NA COL	3					1		3	137	38	36.	Ę	160	969	446	*
Fluoranthene	2 2	3				3 4	2 4	 			3 5	386	18.	150	2	446	
Fluorene	WO'KO	*	43.4	3	•	3	2 2	,	3		3		;			,	,
Indeno(1,2,3-cd)pyrane	40/Kg	è		÷	7 7	2	8	y	2	a i	2	٩	:	2	y	3	× 5
Naphthalene	wo/Kg	<17	<17	Ş	<22	ş	930	ş	8	ę.	2	\$	Ì	812	S)	0	Ş
Phenanthrene	μο/Kg	<1.7	4.7	-	\$55 \$75	8	<23	_₹	æ. ₹	6.5	8.	8	<u>-</u>	615	3	ć23	9
Pyrene	μο⁄Kg	<1.7	<1.7	<1.7	42.2	41.B	<23	۶	₹	41.9 □	4.8 6.1	\$. 8.	<1.7	6.5	~	\$23	Ş
Surrogetes																	
2 4.DCAA	8	8	100	Ş	8	8	108	8	112	3	Z	3	98	89	901	104	3
2 4 E TOWY	8	69	88	2	88	52	26	2	8	8	35	79	89	108	25	51	85
Carcination	ď	*	100	5	8	2	118	۶	2	35	ھ	129	28	120	74	98	128
Campand		78	2	7.	2	3		28	5	115	11	29	8	3	20	0	8
Matale																	
A business	mo/Ko	100	2200	1100	18	88	65	26	120	22	82	210	53	90	120	360	260
Adimoni	S V	60	ç	ç	63	ç	<230	\$	\$	د2	\$	42	<2	Z>	<2	<3	<2
MILITALIA		,	30,	Š	707	20,	.07	80,	,0,0	40 k	505	<0.5	<0.5	902	40 g	18.0	¢0.6
Arsenic	NA P	200		3 5	300	3	,20		3	3	1	5	۶	200	767	100	100
Berlum	D N	Ş	₹.	Ş -	ŞĮ.	3	27,	,	,	,		1	1	,			Ī
Beryllium	₩.O	v.	Į.	₹	₹	•	<u>.</u>	7	7	,	,	7	7		1	\\ \{\}	Ţ
Cadmium	₩Q.K	v	⊽	V	V	V	v	v	V	- S	٤	7	7	,	, [
Calcium	mo/Kg	8	882	240	D.Z.	3	8	3	3	3KU	3	₃,	2	ş.	3,	8	200
Chromium (total)	шожо	5.8	c	2.9	v	₹ -	Į,	~ ·	- 	₹	۲	<u>-</u>	<u>-</u>	Į,	,	,	,
Cobatt	D WOE	ç	Ş	?	ν̈́	Ŷ.	۱ ۲	٠	₽.	۶	Į:	?	۶۱۰	9	9	1	
Copper	mo/Ko	Ŷ	Ş	0	V	0	ì	8	0	9	?	0	0			Ì	2
Iron	шоЖа	8	3200	2000	5	ş	2	0	230	B	387	3	g,	8	2	Ş	3
Lead	mp/Kg	~	2	=	=	v	1.5	7	1.5	7	ð	5.6	V	2	,	2	7 3
Magnesium	mp/Kg	2	2	2	2	8	51	\$	3	3	3	В	ş	•	2	3	2
Manganese	mp/Kg	2	- 22	23	v	⊽	ī	⊽	⊽	v	3	v	v	v	-	2.9	V
Mercury	шожо	<0.01	0.0151	<0.01	<0.01	¢0.01	40.01	6 0 0	ê. 9	Б Б	ê.	- - -	Q.0	\$0.0 \$0.0	¢0.01	0.0	0.0
	mo/Ko	\$	\$	<5>	<200	<5	<7	ç	<5	9	41	çŞ	ŝ	ş	ę	۲,	ŷ
Protessium	moAko	25	2	47	ဆွ	28	<35	22	40	23	.	<27	<25	ζ9	3	110	200 ~
Colonism	mo/Ko	3	9	2	ŝ	8	ŝ	? >	<2	?>	42	<2	<2	ح	~	<3	42
Civar	mo/Ko	3	ø	?	S	2	ಽ	7>	<2	<i>?</i> >	y>	ح2	<2	<2	<2	جع	\$
Softin	mo/Ka	=	Ŋ	141	112	427	<35	59	30	48	æ	19	<25	.9	41	130	\$
Theilien	a Ye	v	ē	P	V	v	⊽	Þ	Þ	۲۷	۱>	<1	1>	<u>دا</u>	1>	۷.	۲,
Vacadum	mo/Ka	2.6	33	2.4	₽	₹	۲	1.3	Þ	1>	7	احا	1>	V	1	2.9	1.8
7:00	Š	Ş	8 6	3	/>	2	¢22	\$	2	9	Ę	ş	ŝ	ş	<19	ζŞ	11
ZIIIZ																	

1 = value is between the Method Detection Limit and the Practical Quanitation Level
J.s. value is below the established limit for accuracy
= 8310s were diluted (1:5)
= 8310s were diluted (1:10)

Table B-15. KSC Beckground raw data for soil locations. *** value is an outlier and not used in the analysis to develop the KSC Beckground screening values.

	•	H															
	"	90 (SSC060		SSC062	SSC062 SSC063**	SSC064		SSC065	SSC66 S	SC067 S	89005	0800	SSC067 SSC068 SSC069 SSC070 SSC021:::	0000	0.000
	<u> </u>	07/06/98	02/06/98	96/90//0	06/26/98	07/27/98	08/20/38	7/7/98		0 86/20/20	_	07/27/98 0	07/27/98	07/07/08	2000	1/0000	2200/2
								9	Curbicate		_	-		2		DSA SA	DEVOY/
Other Parameters		-					Ī			t	I	1					
Bulk Density		ļ	l	I									_			ľ	
	E C	0.9	1.2	7.	0.72	0.79	0.72	0.65	0.68	90	2000	٤	ŀ	ļ			
200	meo/100a	2	 	,	å	ľ	1				3	3	-	99.0	0.75	98	8
Percent Solids	,	8	2	. [8		•		3	2	16	19	22	•	12	-	5	38
Man Man	֓֡֓֜֜֜֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֡֓֜֓֓֓֡֓֜֓֓֓֓֓֡֓֜֓֓֡֓֡֓֡֓֡֓֡֓֓֡֓		R	ß	9	23	22	2	8	88	83	26	8	a	Ag		
	'n	?	7.0	2.5	87	7.6	1 2 7	67	•	ļ				1	3	?	3
Header with	2	210000	S	2	ļ			┪	┪	?	6.5	4.2	4.0	-	2.5	4.6	
Tarking (No. 4)	+	3	3	338	717	440000	9999	320000	6 0000 000 000 000 000 000 000 000 000	940000	200000	750000	Somme	2000	20000	225	
	»e	99	8	8	- 2	100	100	٤	٤	t	╁	+	4	4	333	3	
Texture (No. 10)	<i>y</i> e	8	8	800	8	,	3 8	3 3	+	3	3	8	90	9	8	001	8.66
Texture (No. 40)	•			2	02.3	'n	2.5	39.5	99.2	7 00	- 8.06 8.06	9.66	8 66	7 66	2 2 8	8	8
Total 100	2	2	40.4	97.3	70.5	~ 8	97.2	4.8	26	1 96	07.3	9	, 5				8
BAKIUTE (INC. OU)	×	9.28	37.6	91.6	12.2	8	2	693	 			2 2	R	Š	7/0	82.8	8.5
Taxture (No. 100)	3	0 65	-					300	┨	23.5	77.	63.6	86.3	97.1	16.5	25.3	49.5
Texture (No 200)					7	*	25.5	93.6	59.5	61.2	33.8	32.3	31.5	12.4	-	;	3
Towns (110) FAO	ę	3.9	2.5	3.	2.4	8.2	1 75	ļ	2,5	١		ļ		+	3	*	200
Total Organic Carbon	mo/Kg >	>26700	>26700 >26700		28700	28700	Ş	200	200				, 	6.4	3.9	2	3.8
		1		_		30/00	7	×60/00	>26/00 >26/00 >26700	\$6700		>26700 >	>26700 >26700	_	>26700	>26700	>26700
																1	

Table B-15, KSC Background raw data for soil locations. *** - value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC073	\$SC074	SSC075	SSC076	SSC077	SSC78 S	SSC079 S	SSC080 S	SSC081 S	SSC082 S	SSC083 S	SSC084 S	SSC085 S	SSC086 S	SSC087 S	SSC088
		96/90//0	96/90//0					96,90//	98,90	88/41/	96020	96/624	96,624	86/21/6	96/21/6	88.2	96/67/
Organochlorine Pesticides (8081)																<u> </u>	
4.4' - DDD	po/Kg	<3.8	<4.7	<3.9	<3.5	Н	H	H	H	<5.1	Н	<3.4	<3.4	<3.9	\vdash	\vdash	5.7
4.4 · DDE	MO/Kg	<3.8	<4.7	<3.9	<3.5		H	Н	H		H	Н	Н	43.9	Н	Н	<5.7
4,4 · DDT	₽9/Kg	<3.8	<4.7	<3.9	<3.5	\forall	<3.7	<5.6	Н		\dashv		H	<3.9	Н	H	5.7
Adrin	MVK0	3.8	\$7	639	3.5	43.6		+	1	- -	┪	28	+	<3.9	1	1	\$
Alpha - BHC	MO/Kg	38	¢4.7	<3.9	<3.5	1	1	┪	+	1	┥	┪	+	39	1	1	5.7
Beta - BHC	MO/Kg	<3.8	¢4.7	<3.9	<3.5	<3.6		\dashv	\dashv	_	\dashv	\dashv	+	<3.9	1	1	25.
Chlordane (alpha)	MO/Kg	63.8	47	39	335	3.6	57	\$5.6	+		35	20	34	63.9	7	+	657
Chlordene (gamma)	60 KG	88	\ \ \ \		35	23.6	1	┪	+	1	+	+	+	63.9	1	+	è
Chlordane (total)	WYK9	3.8	\$	<3.9	43.5	+	1	┪	+	1	+	+	+	65	1	1	2.5
Delta - BHC	MO/Kg	38	<u>\$</u>	<3.9	3.5	336	1	+	+	- -	+	+	+	<3.9	1	1	22
Diekfrin	FO/Ko	63.8	47	439	<3.5	+	43.7	+	┪	1	<3.5	43.4	+	<3.9	+	+	2,7
Endosulfan I	MO/Kg	<3.8	,	<3.9	<3.5	┪	1	+	+	_	\dashv	┪	+	<3.9	7	+	Ş,
Endosulfan II (beta)	MVK0	8	Ş	6. 5	3.5	98	1	+	\dagger	5	┪	┪	+	33	1	÷	(5.7
Endosulfan Sulfate	00/Kg	\$3.8	2	639	3.5	+	1	+	+	+	┪	+	+	63.9	1	+	Ç;
Endrin	69/Kg	\$3 8.0	Ş	69	3.5	+	1	+	×4.3	\$	+	+	+	63.9	\$	+	- -
Endrin Aldehyde	W/Kg	8	~	\$ \$ \$	<3.5	<3.6	1	+	┪	_	┪	+	7	ş	1		5.
Endrin Ketone	νο⁄Kg	ي 8	\$	-	3.5	85 85		+	1	_	┪	┥	┪	<39	1	1	5,7
Gamma - BHC (Lindane)	MD/Kg	89	\$7	339	<3.5	<3.6	<3.7	<5.6	+	_	<3.5	43.4	43.4	<3.9	į	- - -	<5.7
Heptachlor	MO/Kg	<3.8	<4.7	<3.9	<3.5	<3.6		1	7	_	┪	+	┪	<3.9	+	+	(5.7
Heptachlor Epoxide (a)	₩o/Kg	<3.8	\$	63.9	335	3.6	1	┪	+	1	\dashv	+	\dashv	<3.9	1	+	57
heptachtor Epoxide (b)	MO/Kg	3.8	\$	86 86	35	<u>ي</u> 8	1	┪	+	1	+	┥	7	639	1	+	\$27
Isodrin	₩,	<3.8	¢4.7	<3.9	<3.5	<3.6		\dashv	+	\dashv	+	<3.4	43.4	3.9		+	5.7
Methoxychior	WW.Kg	<3.8	<4.7	<3.9	<3.5	<3.6		+	+	\exists	\dashv	\dashv	+	<3.9	1	c4.1	57
Mirex	₩/Ka	<3.8	47	65	35	3.6	37	5.6	£	5.	\dashv	\$ 8	3.6	<3.9	4	+	<5.7
Toxaphene	WO/Kg	\$	â	8	ę,	\$	7	┪	┪	+	+	\dashv	+	\$78	1	ٷ	5.7
Araciors (8082)						1		†	+	1		+	+	\dagger	+	+	
PCB-1016/1242	WOYKO	3	Ì	3	3	8	ò	+	3	5	3	+	+	+	0	ş	è
PCB-1221	70/Kg	8	747	8	<u>چې</u>	8	3,	+	+	5	ફ્	\dashv	┪	+	+	-	Ġ
PCB-1232	₩o.	85	c47	8	<u>چ</u>	8	<u>ئ</u>	┪	Ş	<u></u>	8	+	1	+	+	Ş	(%)
PCB-1248	₩.	8	47	88	ş	8	3 3	┪	£	<u>\$</u>	8	\dashv	+	┨	\dashv	₹	\$
PCB-1254	MVK0	8	ş	8	ફ	8	ئغ م	\$	\$	5	ફ	ठु	ڻ ئ	89	\$	Ę	ξ <u>ς</u>
	WK0	88	¢2	8	څ څ	8	<u>ئ</u>	8	<u>\$</u>	<u></u>	8	+	+	+	+	₹	- (§)
2.0.4 £ Technological (2.4 £ . TD) (Char)	U)JJOH	Ę	1	Ę	Ę	+	+	†	-	4	ŧ	-	4	5	- 65,	2.6	1
2.4 f. Trichlorunbanouv acatic acid (2.4.5. T)	, OKO	; -	7	100	1	1	F	100	£	t	=	+	; ; ; ;	100	\dagger	3 6	1
(2.4-Dichlorophenoxy acetic acid (2.4 - D)	VO/Kg	Ē	414	¢12	Ę	Ę	Ę	15	£13	<15	=	+	9	- 2 2 7	H	<12 <12	417
3,5-DCBA	DA/G/	<11	<14	<12	<11	<11	<11	<17	<13	<15	<11	<10	<10	<12	<12	<12	<17
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	₩ø/	<11	<14	<12	Ę	Ę	Į	41	H	\dashv	1	\dashv	Н	\dashv	\dashv	<12	41
4 - Nitrophenol	NO/Kg	Ş	\$	¢12	₹	₹	₹	Ş	S)	\$ \$	₹	읭	응	Ç	- 25	 	÷
Acifluorien	VO.	₹	*	21>	ӛ	;	⇟	;	\dagger	\dagger	÷	+	\dagger	\dagger	\dagger	2 5	÷:
Bernazon		- -	:	į	┆	;	†	† ₹ ;	2 5	\dagger	 	2 5	2 5	¥ 5	\dagger	 	;
Chicamoen	0 10	į	•	7	;				2 5					¥ 5	†	7 5	
	01/07	Į	·	ž	₹ :		;	;	2 5	\dagger	 	+ }!;		*	†	¥ !	
Delepon	100KG	5	*	2 5	₹	 	† 	***	2 5	\dagger		\dagger	2 5	\dagger	t	<u> </u>	;
District of the second of the	DV.O.			***	†	 	† 	†	1	\dagger	 	→		3 5	215	\dagger	1
Discussion of the Commence of	2 5	1	;		 -	 -	† 	t	2 5			$^{+}$		t	t	***	T
MCPA	60/Kg	4230	65°	977	¢210	83	\dagger	\dagger	987	t	t	╁	H	t	t	t	25
deC)#	No.	<230	×290	4240	<210	823	01.5	t	H	6310	ot 5	H	H	<240	H	F	25
Pentachlorophenol	W/Kg	<11	<14	<12	411	<11	H	H	Н	Н	Н	<10	H	Н	<12	Н	<17
Pictoram	6A/Ø/	11>	<14	<12	<11	<11	<11	Н	<13	<15	<11	Н	<10	H	Н	<12	<17
														۱			

I = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established firmt for accuracy = ± 8310s were diluted (1:5) = ±8310s were diluted (1:10)

Table B-15. KSC Beckground raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC073	SSC074	SSC075	SSC076 .	250077	SCC78 1		Coccoo	100000	000000	-	1.0000	10000			
		07/08/98			07/07/98	07/0/79	06/22/96	07/08/96		07/14/96	987999	06/23/98	55C084 06/29/98	SSC085 08/12/98	SSC086 06/12/98	SSC087 06/12/98	SSC088 06/29/98
DAM (2710)			J	1	1	1	1	1									
1 - Methybrachthaton	ŀ				1		-										
9. Methylandralan	2	615	424	Ş	+	8	H	<28	422	<26	æ	41>	41>	Ŗ	Ş	Ş	Ŗ
A constitution	D. CO	V.	727	83	9,0	√18	<19	88	Ŋ	4 56	<18	412	412	ş	82	5	8
According	MONG	8	78	Ş	1	+		<28	22	<26	<18	/₽	4	8	ş	5	8
	9	63.8	Ì	<3.9	<3.5	٦	-	<5.6	<4.3	<5.1	<3.5	<3.4	<3.4	<3.9	70		1,4
Managere	Š	Ş	75	8	\forall	<19	<19	<28	<22	9 2	\$ 8 5	418	8 ₽	8	85	5	8
	WW.	<1.9	<2.4	<2		_	_	<2.8	<2.2	<2.6	87	417	417		Ş	,21	,
Benzo(a)byrene	ρ.χο./	<2	دع	ح2	H	┝	t	5	3	2		1		3	,	,	Ş,
Benzo(b)fluoranthene	W/Kp	ខ	*	*	┝	┝	H	5	1	ľ	,	1	,	٩	,	*	3
Benzo(g,h,i)perytene	0X/04	<3.8	4.8	<3.9	t	t	737	, ,		, 	,	7	3	=	ě	ē	ş
Benzofkifuoranthene	5)/0/	5	\	,	\dagger	t	t	t	3	• •	200	ŝ	35	Z	Ţ	<4.1	<5.7
Chysene	3			,	\dagger	†	t	+	»	3	2	8	8	•	<2	د2	ç
Dibenzyle hlanthranene		2		7	+	1	<1.9	┪	\$5 \$5	\$.6 5.6	<1.8	<1.7	<1.7	6.3	2>	<u>~</u>	429
Fluctorhere	2	0.50	2	6.55	+	+	+	1	£3	\$1	<3.6	<3.5	<3.5	\$	â	<u>*</u>	557
Flicense	SV.Co	3	2 P	63.9	<3.5	\$ 2	37	\$.6	<4.3	-6.	<3.6	<3.5	<3.5	5	3	2	457
Samoning and the same of the s	WO.	<3.8	4.8	43.9	┪	\dashv		Н	<4.3	<5.1	<3.6	<3.5	<3.5	¥	52	41	5.5
moeno (, z, 3-ca)pyrene	ο γογο	<1.9	<2.4	<2	_		┝	H	22	\$5 \$6	8	Ę	150		·	,	,
Nachthalene	W/KD	<19	<24	0Z>	<18	┝	419	H	200	*	176	ŀ		١	, [,	ş
Phenanthrane	₩o/Ko	<1.9	42.4	2	H	H	t	t	100					3	3	3	3
Pyrene	W/Kp	4.9 €.19	42.4	8	t	t	612	800	16					y 5	»Į:	3	53
Surrogates					1	1	1	1						B	3	9	62.9
	30	8	42	23	55	63	63	2	9	٤	100	Ş	ļ	ļ			
2.4,5,6-TCMX	86	2	8	5		RE	1 2	; ; ;	5 8	*	2 3	¥	\$	*	2	25	R
280	×	2	318	3 2	3 3	3 8	3 3	֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓	!	ا الا	e	8	3	8	=	8	5
p-terohenvi	3	ž	F		3 8	2 8	1	\$		3 2	2	8	8	=======================================	119	5	8
Metals					36	8	8		2	3	2	8	5	6	82	~	75
Akminum	a Xon	91	280	220	3	120	98	27.0	1000	5007	***						
Antimony	mo/Ka	0	Ş	5	1	1		•	t	3		3	8	3	BZZ	2300	ŝ
Arsenic	ωο/Ko	90%	20,2	80,	╁	t	t	┿	t	2	*	»	3	8	7	8	ş
Berkm	S S	50	128		\dagger		t	+	9 8	9	ş,	ę.	ŝ		√0.6	۷0°6	<0.8
Benyffum	D X	1	1	†	t	\dagger	; ; ;	3 9	8	ē (ş .	8	Ş	2	₹	ŞŞ	ठ
Cadmium	S)Vou	1		† 	 	t	Ť	,	₹	y.	- 	₽	V	V	⊽	v	۲5
Celcium	o Work	929	162	5	98	, 52	†	*	- -	»	Į.	⊽	v	v	⊽	v	Ÿ
Chromium (total)	morKo	ī	-	†	†	+	+	† 	3	3	8 .	R.	2	8	8	8	000
Cobalt	ωoγKo	ę	1/2		t	╁	t	+	 	3	┆	- -	Į.	8;	2	2	3
Copper	JO Wo	ę	-	, 4	t	╀	†	╁		*	٥.	٥,	<u>وا</u>	ęļ	ę,	ę	8
Non	mo/Ko	2	2	8	t	✝	+	+		9 5	e s	0	0	8	ę	ę	8
P #9]	mo/Ko	2.5	~	v	-	2.6	+	╅		3 :	8:	3	3	3	8	800	8
Magnesium	mo/Ko	8	5	22	t	十	+	+	240	3 2	3 8	,		2 2	7	2	-
Manganasa	mg/Kg	v	2.3	Ī	t	t	╁	+		į	:	3 :	8 .	3	3 ;	2	B
Mercury	mo/Ko	¢0.01	0.0171		t	۲	t	╁					3 3		=	3	2
Nickel	ω V	ę	₽	t	t	╁	=	+	3 4		5 5	200	5	3,	5,0	0.015	5
Potassium	mo/Ko	×190	052	190	25	t	╁	t		,	3 8			6	8	9	8
Selenium	mo/Ko	Ş	5	1	+	,	t	3	; ; ; ;	3	3	\$	†	3	8	3/	88
Silver	o de	,		 	\dagger	$^{+}$	\dagger	\dagger	» («	3	۶ŀ	»	۶ ا	9	3	8	ŝ
Sogn	S S S S S S S S S S S S S S S S S S S	, 2	?	,	+	y	†	+	9 P	0	8	9	7	9	8	ح2	<3
Thallism	ALL DE	5	ş -	8	ę .	+	+	+	2	2	2	2	=	8	<30	æ	660
Vanadium	200	1	7	† • •	 	\dagger	9 9	» ·	Į.	2	⊽	Ţ	Ţ	v	V	Į	<2
Zirc	DVO.	•	₹	₹,	₹ -	- -	+	+	ē	99	- 8:	1.9	ī	•	8	3.2	2.9
	W				0	1	\dashv	\dashv	22	- =	Ş	8.8	6	6.9	9>	1.	15

I = value is between the Method Detection Lirrit and the Practical Quantitation Level J = value is below the established firrit for accuracy = 8310s were diluted (1:5) = 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** value is an outlier and not used in the analysis to develop the KSC Background screening values.

		Second Second Second Second Second Second	17000	200	200		SSC		SSC080 SSC081	-	28000	3333	3500	2500	/ACCUSO ののこのの ののこのの	_	3
		07/06/96	07/08/98	96/80/20	96/20/20			07/08/98	96/90/20				96/62/90	08/12/98	06/12/96 08/12/98	06/12/96	06/29/98
Other Parameters		Ī	İ	T	Ī												
	ovem	0.59	0.79	28.0	6.0	0.82	0.72	0.58	0.42	99.0	11	96.0	0.92	0.91	1.1	0.7	0.48
Con Contain	mea/100a	22	12	22	8	6	೩	22	22	31	6	-	- 9	22	14	27	47
Densey Courte	*	28	2	88	g	5	8.	S	1	88	ぁ	98	96	25	83	19	28
out deb)	n's	5	6.7	:	4.7	8.4	9.4	4.2	4.2	7.6	5.8	5.9	6.2	9.9	6.7	6.7	7.4
Decleticity	ohm-cm 220	8	320000	27000	100000	730000	370000	200002	260000	130000	2300000	1700000	3100000	62000	63000	46000	4600
Technology Al	×	-	8	8	68	2	88	666	6.66	8	20	801	100	6.86	100	100	100
Technic (No. 4)	*	8	98	666	966	68	85	7 98	98	6.66	66	6.66	6.66	97.7	8.66	2.66	99.4
Technol (No. 40)	,	5	88	2	8	75.1	2	91.2	91.4	85.2	67.7	88.4	89.2	86.2	89.4	7.96	92.8
Technology (No. 60)	,		2	533	8	2	989	4.2	2	78.4	89.3	88	64.5	63.7	63.3	93.9	82.9
Texture (No. 100)	3	٥	287	15.7	=	25.5	18.8	31.8	- 97	34.2	41.8	46.9	35	27.2	28.3	54.9	60.2
Texture (No. 200)	8	Ļ	53	9	82	3.4	4.9	5.6	5.6	11.5	7.9	7.5	7.2	7.2	9.6	15.1	25.7
Total Omanin Carbon	١,	18	700 >26700 >26700		>26700	>26700	>26700	>26700	>26700	>26700	3000	>26700	>26700	>26700	>26700	>26700	>26700

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC089	000055	50000	Course		10000	2000	0000							
		06/29/98		08/12/98	08/12/98	08/13/96		07/14/98 07/	07/14/96 07/	SSC097 SS 07/14/96 06	SSC098 SSC 08/21/96 06/2	SSC099 SSC100 08/21/98 08/21/98	00 SSC101 98 07/27/98	31 SSC102 36 07/27/98	SSC103	SSC104
Organochlorine Pesticides (8081)	L			1	1	1	1		1	1		_	_	_		
4,4' - 000	2,000			ļ	†	1	1	1	1	\dashv						
4.4' - DDE	2 2	900		Ş	+	9	2.1	<3.9	٦	<3.6	<6.7 <1	H	H	<3.8	897	ķ
4.4 - 001	2 2	200	•	3	+	+	_	┪	<4.4		-	5 <4.3	H	<3.8	889	
Adrin	2			\$	+	6 7	+	+		H	\vdash	┝	<3.8	<3.8	689	
Alpha - BHC	2	200	•		+	+		<3.9	<4.4		H	5 <4.3	┢	<3.8	<8.8	
Beta - BHC	2 2	200	*	.	+	+	1	┪		<3.6	6.7 <15		╁	38	9	
Chlordana (alpha)		965	•	:	+	2	+	┪	<4.4 <0.4	Н		5 <4.3	H	<3.8	668	
Chlordane (gamme)	200	53.0		ş	+	+	_	<3.9	\dashv	Н	<6.7 <15	H	H	<3.8	\$ 9°	3
Chlordane (total)	200	000		<u>.</u>	+	1	1	┪	<4.4	<3.6	<6.7 <1	H	<3.8	<3.8	698	
Delta - BHC	2	500	•	3	+	1	¢4.1	٦	Н	Н	-	H	H	438	84	
Dieldrin	2	25.6	*	<u>.</u>	+	, (ê)	1	1	۲	H	c6.7 <1	-	88	<3.8	668	
Fodosida		200	**	3	+	1	1		H	H	H	┝	H	43.8	8 9/	
Endosulan II (heta)	WO'K	9	Ş	\$	\$	¢ 7.	<4.1	<3.9	H	\vdash	<6.7 <1	┝	H	5	3 5	
Forbeitten Culture	NO.	9.6	4.4	÷	1			H		\vdash	17 <15	H	t	800	9	
Frides	MVK9	<3.6	4.4	ź	<4.4 4.4		_		H	\vdash	H	H	╁			
English Aldebude	MA/KO	<36	4.4	ş			<4.1	H	<4.4	┝	67 <15	t	\dagger			
Codio Kasa	WAKE WAKE	<3.6	<4.4	¢4.1	<4.4		-	۲	┝	╁	╁	1	200			
CHOIR NAIGH	WKg	<3.6	<4.4	<4.1	<4.4	H	c4.1	t	<4.4	H	t	t	\dagger		000	0
Garrina - Britc (Lindane)	MO/Kg	<3.6	4.4	<4.1	H	H	l	۲	117	+	\dagger	\dagger		\$ 1	8	ç
Heptachior	μο⁄Κg	<3.6	4.4	1.4	H	¢6.7	×4	t	\dagger	t		$^{+}$		43.8	89>	\$
Heptachlor Epoxide (a)	ρο/Kg	<3.6	4.4		H	t	t	3.0	t	\dagger	+	\dagger	23.8	<3.8	¢6.8	\$
heptechior Epoxide (b)	//O/Kg	<3.6	4.40		<4.4	t	ł	t			+	+	43.B	<3.8	6.8	\$5
Bodrin	pa/Kg	<3.6	4.4	L	t	t	t	╁	\dagger	\dagger	+	\$	3	<3.8	899	\$5
Methoxychkor	₽Ø/Kg	<3.6	444		H	t	\dagger	$^{+}$		+	†	+	63.8	<3.8	6.8	5.4
Wirex	W/Kg	<3.6	444	t	777	¢67	t	439	\dagger	9 4		+	800	89	99	\$
loxephene	μο/Kg	<3.6	<4.4	ŝ	H	L	t	+	\dagger	t	+	+	23.5	8.50 0.00	6.8	\$
Aroclors (8082)					t	t	t	t	\dagger	2	+		\$	ş,	89	\$5
PCB-1016/1242	μογΚα	<36	, 44	190	\$	H	145	\dagger	t	+	t	15		ļ		
PCB1221	W/Kg	96>	44>	140	H	29>	H	t	\dagger	\dagger	t	+	3	8	8	ŝ
PCB-1232	₩0/K0	×36	4	2	+	t	t	150 140 140 140 140 140 140 140 140 140 14	3 8	9 4	8 5	+	3	800	\$	20
PCB-1248	ολ/αν	% %	<u>\$</u>	₹	43	19	ł	t	\dagger	+	†	+	8	88	88	Ŷ
PCB-1254	DA/Q/	85	\$	t	t	t		\dagger	\dagger	+	†	Ş	8	8	ŝ	Ŷ
	0A∕Q∕I	8,	4	t	170	Ę	\dagger	\dagger	\dagger	+	t	+	8	°38	×68	20
Chlorinated Herbicides (8151)				t	t	\dagger	t	t	+	₹	+	<u>\$</u>	8	86	89	Ŷ
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	ρλ/gα	<11	<13	<12	£53	╁	<12	t	\dagger	t	,	+				
2.4.5- Inchlorophenoxy acetic acid (2,4,5 - T)	0√Kg	(<13	<12		-	H	<12	3	t	+	2 5		7 5	Ş .	
2.4-Uchlorophenoxy acetic acid (2,4 - D)	W.Ko	Ę	<13	Н	Н	Н	-	2 <13	H	t	╁	+	=	755	5	9 9
4-(2 4-Dichlemphanoxylly flyic soid (2.4 - DB)	W.Y.	Į.	₽ V	+	+	+	\dashv	Н	Н		H	\vdash	₹	- F	1	2 5
4 - Ntrophanol	NA C	₹ 	2		+	+	+	\dashv	3 <11		Н	H	₹	<12	8	95
Acifluorien	WOYKO		2 5	\dagger	2 5	\dagger	2 5	\$\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	+	1	Н	Н	11 >	<12	Ş	9
Bentazon	WO/Kg	1		t	\dagger	\dagger	\dagger	\dagger	+	+	+	┥	ţ	<12	ح2ا	9
Chloramben	wo/Ko	ē	1		†	\dagger	\dagger	\dagger	\dagger	†	+	+	Ę	<12	<21	<16
Dacthal	worKo	₹	- E	t		t	215	200	\dagger	+	╅	+	Ę	<12	<21	۰16 مراو
Dalapon	oy.ov	Į	1	†	t	†	\dagger	+	+	1	┪	+	Ę	<12	<21	ş.
Dicambe	D/O/Kg	Ę	Ş	1 2	1	t	+	†	\dagger	†	+	+	₹	<12	<21	<16
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic scid]	0X04	ē	£ 5:5	╁	t	\dagger	\dagger	\dagger	+	†	+	╅	₹	Ş	د21	<16
Dinceeb	μg/Kg	Ę	┢	t	\dagger	t	12	t	t	\dagger	\$ 1	+	₹ :	2	2	99
MCPA	<i>ν</i> ο/Κα	<110	Н	H	<270 <	┢	H	220	t	\dagger	+	┿	Ş	2 5	ζ <u>ζ</u>	9 5
Pentechiomoteroi	\$ 2	o110	+	+	Н	Н	50 <240	H	t	t	╀	+		360	2000	3 5
Pictoran	δ. Vo.	₹ V	£	<12	<13	° 02>	<12 <1	Н	H	ŝ	╀	Ş	3 =	÷		3 4
	MONTO	Ę	┪	┨	┨		12 <12	Н	H	┝	\$45	¢13	Ę	197		
											ł]

I = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy = 8310s were diluted (1:5) = 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** - value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC089 06/29/98	SSC090	SSC091 08/12/98	SSC092 8	SSC093 08/13/98	SSC94 8	SSC095 8	SSC096 07/14/98	SSC097 07/14/98	SSC098 3	SSC099 :	SSC100	SSC101	SSC102 (07/27/98 (0	SSC103 (06/29/98 (0	SSC104 06/29/98
					_		_	_			_	_		_	_	1	
PAH (8310)															_		
1 - Methylnaphthalene	po/Ko	<18	<22>	<21	<22	<34	-51	<20	<22	<18	₩,	9/≥	ZŞ	<19	<19	\$E>	<21
2 - Methylnaphthalene	MO/Ko	<18	~ 22	<21	<22	×34	ح21	<20 >	<22	<18	×34	<76	ζŞ	<19	<19	\$34 34	<27
Acenaphthene	µ9∕Kg	<18	<22	<21	<22	×34	<21	<20	<22	<18	\$\$	<76	<22	<19	<19	×34	<27
Acenaphthylene	MQ/Kg	<3.6	474	<4.1	<4.4	<6.7	<4.1	<3.9	<4.4	<3.6	<6.7	<15	<4.3	<3.8	<3.8	€.8	\$5.4
Arithracene	MO/Kg	<18	çŞ	2	Ş	3 3	2	Ş	ş	ş	ર્સ	ķ	Ş	ę	Ş	SS,	8
Benzo(a)anthracene	₩ø/Kg	<1.8	<2.2	42.1	<2.2	<34	<u>دي</u>	Ÿ	2	8.15	₹ 8	\$7.6	27	<u>د</u> و	615	<3.4	27
Benzo(a)pyrene	μο/Κα	د 2	ç	Ÿ	ç	¥	ş	\$	Ş	8	\$	Ş	Ş	8	Ÿ	Ş	ဗ
Benzo(b)fluoranthene	PO/Kg	<3	44	1	*	8	*	П	7.9	دع	9>	<14	*	<3	<3	9>	<5
Benzo(g.h.i)pen/lene	MO/Kg	<3.6	<4.5	56	<4.4	€6.8	c4.1	H	4.4	<3.7	c6.8	<15	<4.4	<3.8	<3.8	c <u>6.9</u>	5.4
Benzo(k)fluoranthene	₽/Vo/	3	8	=	Ŧ	\$	25	r	8.3	د2	*	69	ဗွ	ر ₂	8	\$	8
Chrysene	₩0/Kg	<1.8	<2.2	<2.1	<2.2	<3.4	<2.1	Н	8.1	<1.8	<3.4	<7.6	<2.2	<1.9	<1.9	<3.4	<2.7
Dibenzo(a,h)anthracene	PO/Kg	<3.6	<4.5	<4.2	<4.4	<6.8	<4.1	*	<4.4	<3.7	<6.8	<15	<4.4	<3.8	<3.8	6.9 >	<5.4
Fluoranthene	100/Kg	<3.6	<4.5	<4.2	<4.4	9	<4.1	\$	4.4	<3.7	€.8	¢ \$	4.4	<3.8	<3.8	6.9	<5.4
Fluorene	₩Q/Kg	<3.6	<4.5	<4.2	4.4	€6.8	64.1	*	¢4.4	<3.7	<6.8	<15	444	<3.8	<3.8	<6.9	4.5
Indeno(1,2,3-cd)pyrene	WO/Kg	¢1.8	<2.2	5.5	25	ş	2	Ş	22	2 8 7	34	¢7.6	25°	6,5	6 5	46	25/
Nachthalene	WONKO	8 V	ŝ	2	3	3	₹	Ş,	3	*	\$	\$	3	Ę,	615	₹ -	3
Phenanthrene	W/Kg	¥.	27	2	2	28	Ş	8 ·	2	- -	Ş.	9,0	22	S .	Ş	Ş.	<u></u>
Pyrene	WO/Ko	8.5	42.2	22	220	ē	2	8	42.2	₩ ₩	34	\$7.6	422	5.5	6.5	34	\$27
Surrogates								ļ						ļ		-	
2,4-DCAA	%	2	8	2	38	<u> </u>	8	25	ğ	ş	2	8	8	8	8	23	8
2,4,5,6-TCMX	8	8	S	~	7	ş	5	2	8	5	27	S	45	\$	8	3	~
DBC	×	8	8	2	£	3	28	2	2	3	8	<u> </u>	-	8	22	28	22
p-terphenyl	8	ę,	8	8	 88	7	8	88	B	8	2	8	8	2	2	2	9
Metals	- 7		2000	,	200	2	30,	2000	9000	200	400,	9000	993	989	3	2000	
Aumanum	MOUL	36,	3	3	3	3	3	3		3,	3	3	3	3		3	3
Antimony	MOTO	ď	?	y	3	Į.	*	*	2 3	۶	\$		3	۷	y	•	3
Arsenic	mo/Kg	<0.5	ÇĞ.	ş	è	٠ ا	9 6	900	è	÷	9,		9 8	900	S G	⊽ ;	9
Barlum	mork ₀	3	ğ	ş,	3	₹ ·	şļ.	\$	ş,	3	\$	₹	ę,	ş	3	÷ (١
Вегуйит	mo/Kg	v	Į.	1	₹	» °	₹	<u>.</u>	₽	₽.	9,	Į.	Į.	₹	₹	ÿ,	3
Cedmum	щОКО	v	V	V	V	7	~	v	V	V .	7	* 5	V	v	V	۶ ا	7
Calcum	D NO	3	3	3	3	3	3,	3	3	3	3	3		ş ,	β.	3 ;	3
Chromium (total)	D S	2 4	,	- K	-	۶ ۲	,	. 4	2	3	3	. 6	7	 	7	= 5	,
Constit	No.	, ,	,	, (†	 			 	, 5	=	3 8	, 5	, 4	۴	2 2	,
ima	mo/Ka	1000	1300	22	3	8	1700	2500	6190	1900	178	3	3	8	3	870	3
Per	mo/Ka	3.4	•	5:	-	3.5	15	4.5	6.3	3.8	1	9	2.5	2	v	•	55.
Manesium	mo/Ko	230	170	225	35	8	28	98	8	230	1200	1100	280	110	8	989	230
Manganese	πο/Kα		12	2	5.	2	12	2	8.4	2	8.6	77	9.	⊽	v	18	4.4
Mercury	ay/on	0.012	<0.01	9700	0.02	\$ 89.09	0.028	80	0.08	0.0161	0.16	0.19	0.021	0.011	<0.01	0.12	~0.0 2
Nickel	шУKо	\$\$	<i>\</i> 2	ş	<i>\\</i> 2) (10	ş	8	2>	\$	<10	<23	\$	9>	9>	<10	8
Potessium	щXо	×180	\$20	75	\$	23	8	8	130	3	530	300	55	æ	¢2>	<340	<270
Selenium	оу/бш	<2	6>	<2	<3	<4	7	<2	<3	<2	•	6>	ç	<i>چ</i>	42	<4	ડ્ર
Silver	mo/Ka	8	6>	₹	دع	*	<2	ح2	<3	7	64	6>	دع	<2	42	64	<3
Sodium	mp/Kg	25	88	35	<33	<51	210	130	320	8	240	130	130	\$	51	190	2
Theillum	mo/Ko	<1	<1	<1	~1	ح	1>	V	۷.	۷.	ح	*	۲	ţ	دا	<2	42
Vanadium	mo/Kg	3.6	3.4	2.1	9	7	4.3	4.4	6	4.9	22	=	v	-	V	9.1	ç>
Zinc	mo/Kg	6.7	12	<6	<۲>	5	6.6	9	ç	ŝ	82	2	ş	ę	ş	22	15
				:													

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

	<u> </u>	00000	00000	0 .00000													
	3 8				08/12/96	08/13/98 (55C94 07/14/98	SSC095 07/14/98	SSC94 SSC095 SSC096 SSC097 07/14/96 07/14/96 07/14/96 07/14/96 07/14/96	SSC097 07/14/98	SSC096 SSC097 SSC098 SSC099 SSC100 SSC101 07/14/98 07/14/98 07/14/98 08/21/98 08/21/98 08/21/98	SSC099 SSC100	SSC100	SSC101 SSC102	SSC102	SSC103	SSC104
				_		-					-	1	3			B6/67/80	86288
Other Parameters			1	-	t	İ	Ì	1	1	1	1						
Bulk Density			1			1	1									l	
	- LD	9/		20	3	0.74	90	A O	0.5	99 0	į	ţ	ļ	l	ı		
	төө/1000	٥	1	2	Š	8				8	eş,	è	0.43	0.7	-	0.35	0.79
Percent Solids	L	8		1			1	2	8	9	7	42	8	22	18	2	٥
(Qu) HO	\dagger			3	?	ŝ	18	ž	75	5	49	22	78	8	6		
17.17.17.1	9.0	3.6	0.0	7.	œ.	3.6	- 6	11/	1	5		;	†	3		?	ß
Messavily	ohm-cm 1600	8	4000	24000	5	20000				*	6.0	9.2	6.5	7.	4	9.4	7.1
Texture (No. 4)	1		7	+	+	3300	300	300)	130000	360000	2300	2000	12000	93000	16000	690	2
Texture (No. 10)		+	+	4	3	B	6.66	60.00	28	8	001	9	8	٤	٤		3
	8		9.66	- 66	- 29	88	7 00	8	9,6	٤	ļ			3	3	3	8
lexture (No. 40)	»	9 A AQ	2	ł	8	2 2	+		9	8	ŝ	26.	- 8	80.08	8.66	95.9	92
Texture (No. 60)	H	ł	╀	╀	2 2	2 2		ž	8	88	91.1	97.9	6.96	74.7	72.6	833	8
Texture (No. 100)	3	404	+		֓֞֜֜֜֜֜֝֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֡֓֓֓֓֓֓֡֜֜֜֡֓֡֓֡֓֡֓֜֜֡֡֓֡֓֡֡֡֡֓֡֡֡֡֓֡֡֡֡֡֡	723	3	/88	83.7	60.9	83.7	95.7	2	20.6	962	3	6
Texture (No. 200)	\dagger	7	+	٥	8	2	53.4	45	45.2	32.8	54.2	653	888	1	1		
	2	_	920	10.5	8.1	99	o	7.3	,	Ļ	0 00				5	7.7	8.5
Total Organic Carbon	mo/Kg >267	8	>26700 >26700	6700	Ì,	Ī	Ę	_	-		20.0	45	9.6	4.6	6.4	22	3.6
	1	1					250/00	3000	250/00	>26700	>26700 >26700	>26700	>26700	>26700	>28700	>26700	×28700
																4	

Table B-15. KSC Background raw data for soil locations. *** value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC1		SSC106	SSC107	SSC108	SC109	SSC11		SC111 5	SC112 5	SC113	SC114		_		SSC118
		6/29/98		07/27/98		96/5290	07/08/98	7/8/98		06/22/98 0	06/22/98 0	08/13/98 0	07/14/98	08/20/38	08/20/38 0	08/21/98 0	96/92/90
			duplicate					٦	duplicate	1		1	1	1	1	1	
Organochiorine Pesticides (8081)									╡						1	1	
4.4 DDD	DA/A	<5.6	<5.3	<3.6	<3.7	<3.4	<3.5	<3.9	<3.6	<4.9	<4.8	<10	*	<5.6	ş	<6.3	<5.2
4.4. · DDE	DW/K0	<5.6	<5.3	<3.6	<3.7	<3.4	_		٦	_	c4.8	ر د 10	*	<5.6	+	┪	\$25
4.4' - DDT	MO/Kg	<5.6	<5.3	<3.6	<3.7	<3.4	Н	Н	<3.6	\dashv	<4.8	¢10	7	<5.6	ę	<6.3	\$52
Adrin	pg/Kg	<5.6	<5.3	<3.6	<3.7	<3.4	35	┪	3.6	1	4.8	ę	¥	<5.6	ê	6.3	\$52
Apha - BHC	MO/Kg	<5.6	<5.3	<3.6	537	\$ 5	1	┪	43.6	¢4.9	8.4	9	¥	\$56	ę	Ş	\$2
Beta - BHC	₩.dv	\$5.6	<5.3	3.6	<u>ئ</u>	3.4	+	65	38	د4.9 د	\$ 8 8	ş	₹	\$	*		\$25
Chiordane (alpha)	MO/Kg	<5.6	<5.3	43.6	43.	43.4	35	+	38	6.45 6.45	\$ \$	윙	¥	5.6	ę	63	\$52
Chlordane (gamma)	MO/Kg	<5.6	<5.3	<3.6	43.7	43.4	┪	┪	3,6	\$ 69	\$ \$	윙	Ÿ	\$5.6	9	ê Ş	25
Chlordane (total)	Dyl/Gr/	<5.6	<5.3	<3.6	<3.7	<3.4	Н	Н	<3.6	<4.9	<4.8	\$	₹	\$5.6	ş	6.3	\$25
Detta - BHC	5 _M on	65.6	<5.3	<3.6	<3.7	<3.4	H	Н	<3.6	<4.9	<4.8	<10	*	<5.6	ş	<6.3	<5.2
Dietdrin	WO/Kg	65.6	<5.3	43.6	<3.7	<3.4	┢	<3.9	<3.6	<4.9	<4.8	<10	*	<5.6	9>	<6.3	<5.2
Endosultan	ολ/ον	45.6	<5.3	<3.6	<3.7	<3.4	Н	<3.9	<3.6	٦	<4.8	c10	45	<5.6	\$	<6.3	<5.2
Endosurtan II (beta)	νο/Κα	65.6	653	43.6	<3.7	<3.4	Н	<3.9	<3.6	<4.9	<4.8	<10	<4	<5.6	9>	<6.3	<5.2
Endosultan Sultate	Mo/Kg	<5.6	<5.3	<3.6	<3.7	<3.4	H	<3.9	<3.6	Н	c4.8	<10	*	<5.6	9	<6.3	<5.2
Endrin	DW6	\$. \$.	<5.3	<3.6	<3.7	<3.4	<3.5	<3.9	<3.6	Н	<4.8	<10	44	<5.6	ę,	<6.3	<5.2
Endrin Aldehyde	MO/Kg	\$56	<5.3	<3.6	<3.7	<3.4	<3.5	<3.9	<3.6		<4.8	<10	*	<5.6	9>	<6.3	<5.2
Endrin Ketone	DW/Q/	<5.6	<5.3	<3.6	<3.7	<3.4	<3.5	<3.9	<3.6	<4.9	<4.8	<10	*	<5.6	g>	<6.3	<5.2
Gamma - BHC (Lindane)	νο⁄Κα	<5.6	<5.3	<3.6	43.7	<3.4	<3.5	<3.9	<3.6	¢4.9	¢4.8	ę	3	\$ 26	8	<6.3	\$ 2
Heptachlor	<i>μ</i> ο/Κα	<5.6	<5.3	<3.6	્ટ 2	\$ \$	\$ 5	659	43.6	6.9	8.8	ę ę	Ş	45.6	g g	83	\$5
Heptachtor Epoxide (a)	<i>μ</i> ς/Κο	<5.6	<5.3	<3.6	<3.7	<3.4	35	<3.9	43.6	64.9	8.8	ę,	*	<5.6	ę	63	\$2
heptachlor Epoxide (b)	DAVRO	<5.6	<5.3	<3.6	<3.7	<3.4	<3.5	<3.9	<3.6	64.9	4.8	Ş	₹	\$56	ę	69	c 5.2
Isodnin	5y/dr/	<5.6	<5.3	<3.6	<3.7	<3.4	<3.5	<3.9	\$36 8	¢4.9	¢4.8	ş	₹	\$5.	ę	69	65.2
Methoxychlor	PQ/Kg	<5.6	<5.3	<3.6	<3.7	<3.4	<3.5	3.9	\$ 8 8 8	\$ 6	8,48	윙	\$	\$5.6	Ç.	63	\$25
Mirex	MO/Kg	<5.6	5.3	3.6	43.7	\$3	35	85	\$36	6.5	8.4.8	ę	V	\$28	ç	Ş	\$52
Toxaphene	MO/Kg	<5.8	5.3	Ş	\$	\$ \$	Ş	\$ \$	2%	6.9	8.48	230	8	2	2 2	\$	S S
Aroctors (8062)				1				1	1		†		١	ļ	1	:	Ę
PCB-1016/1242	WW.	Ş	ş	8	è	3	S	8	8	*	†	3 3	3	B S	3 3	3 8	, is
PCB-1221	WW.	8	ŝ	\$	è	3	8	3	8	₹	†	3 5	3 5	8 :	3 5	3 5	Į,
PCB-1232	WYK9	ŝ	ŝ	8	ş	Ş	ŝ	8	8	3	†	3 5	\$	8	3	3	3
PCB-1248	W/Kg	8	ŝ	8,	⁽³⁷	8	8	8	8	Ş :	†	38	3	8	8 8	3 5	
PCB-1254	o. Voi	\$ \$	3	8	(3)	3	3 4	3 8	\$ \$	2 5	\$ 8	3 5	2 5	8 á	8 8	3 8	ş Ş
PCB-1260	OX.	8	3	ş	Ì	\$	3	3	8		t	3	•	8	3	3	,
Chiormated Merbicioes (6131)	1	,	94.	1	-	5	•	ļ	ŧ	į	1	Ę	55,	5	418	9	418
2-(2,4,5-1nchlorophenoxy)propionic acid (2,4,5-1P) (Silvex)	N S) (9	1	1	2 5			 	1	55	5	9	Ę	=	65	¥16
2.4.5- Incincipling toxy scentic acid (2.4.5) - 1.	W.	ŧ	416	1	į	9	ī	×12	ī	\$5	\$15	5	<12	412	85	615	÷16
4 C.D.CRA	νο/Κο	412	×16	Ę	Ę	ę	Ę	<12 <12	ē	\$15	<15	ê	42	45	×18	<19	<16
4-(2.4-Dichlorophenoxylbutyric acid (2.4 - DB)	λ δ	÷	~16	Ę	Ę	9>	Ę	<12	<11	<15	<15	د31	<12	<17	<18	<19	¢16
4 - Nitrohand	SX O	41	~16	<11	11 >	<10	<11	<12	<11	<15	<15	<31	<12	<17	¢18	¢15	ş
Acituorien	0X/QV	<17	<16	<11	<11	<10	<11	<12	<11	<15	<15	31	¢12	45	8	ę	\$
Bentazon	JO/Kg	-	91>	<11	<11	<10	<11	<12	<11	<15	<15	31	<12	¢17	8	6	ş
Chloramben	₽VQ/	<17	<16	<11	<11	<10	· 11	<12	411	<15	<15	é	25	÷	\$	ş	\$
Dachel	09/Kg	41>	<16	<11	<11	<10	<u>-11</u>	<12	41	¢15	\$	é	\$	Ş	ê	ş	48
Dalapon	PO/Kg	<17	<16	<11	<11	<10	411	¢15	Ę	\$5	ş	é	25	Ę	ê	ę	ş
Dicamba	07/6/	<17	<16	<11	<11	¢10	Ę	Ş	Ę	\$ 5	2	<u>ق</u>	Ç 	ş	ê ê	65	9
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	MO/Kg	<17	<16	<11	<11	¢10	Ę	Ç	÷	Ş	\$	<u>Ş</u>	2	ş	8	65	9
Dinoseb	W/Kg	<17	<16	<11	41	ę	₹	2	Ŧ	\$	\$	<u>ئ</u>	2	Ę	8	eş	9
MCPA	W/Ka	۶ ک	8 \$	ρŞ	¢230	8	230	\$50 \$10 \$10 \$10 \$10 \$10 \$10 \$10 \$10 \$10 \$1	8	S S	\$	8	240 250	8	88	8	8
MCPP	WW.	۶ ک	8 V	Ş	829	3	200	3	3	3	3	3 3	3	3	3	3	
Pentachlorophenol	S S S	ş	ş	ē	Į.	٥	=	2	₹	€ ÷	£ .	5 5	215	Ì	200		۶
Pictoram	8 X	\$	98	ē	Ş	9	ē	3	Ī	Ş	C12	3	22		•	ŝ	•

¹⁼ value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy = = 8310s were dikited (1:5) = 8310s were dikited (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SS	8	SSC106	201055	801.000	001,000	7,033	T								
		6/29/98	8	07/27/98	_	96/29/98	07/08/08	2/8/08		200111		SSC113	SSC114	_		SSC117	SSC118
			duplicate						S. Contraction	06/22/8	0 0000000	98.5	96/4//	96/20/280	06/20/38	821/38	8/28/98
PAH (6310)	L			I	İ	I	I	5	AKCERE								
1 - Methymaphthalene	1	*	Ī					٦	Н		-				l	l	I
2 - Methylnachthalene		Ş	/2/	SI V	<19	<17	<18	H	┝	ŀ	t	Ę,	ļ	ķ		I	
Acanachthana	SV.	5	/Z>	<18	<19	<17	<18	H	H	t	t				3	ķ	ç28
Assessment Assessment	Q Q	<28	<27	<18	<19	41×	618	t	97	\dagger	t		3	8	8	38	8
	MVK0	<5.6	<5.3	3.6	Į Į	34	1		+	9	9	3	Ş	8	ŝ	85	89
	WO/Kg	¢29	427	61.0	61.0	1	1	\dagger	†	+	+	210	\$	<5.6	9	<6.3	295
Benzo(a)anthracene	₩.	<2.8	100	Ę	100	†		\dagger	1	+	1	1	<u>ئ</u>	¢29	431	8 8	8
Benzo(a)pyrene	WO/Ko	5	,	,	?		<u>.</u>	+	7	\dashv	-	-	8	42.8	63	Ş	3
Benzo(b)ffuoranthene	Ş	,		7	*	* ·	3	1		Η	Н	H	8	2	1	ļ,	٦
Benzo(g,h,i)perylene			7	2	2	ç	Ş	_		H	\vdash	H	2	ķ	+	,	3
Benzo(k Muoranthene	2	\$	\$0.	Ş	<3.8	_	<3.5	┝	H	┝	t	t			;	8	e
Chrysena	ST.	8	ę	3	<2		<2	H	t	t	t	t	,	֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓	8		\$2
Olivernia Heathmann	D.W.	42.8	<2.7	€1.8	<1.9	H	<1.8	H	1812	t	1	t	, , ,	2	3	\$	Ş
Fluoradhana	WO'K	<5.6	<5.4	<3.7	<3.8	H	<3.5	t	t	+	t	t	,	\$:	V V	┪	5.8
Flacens	No.	<5.6	5.4	<3.7	<3.8	43.4	<3.5	2	t		\dagger			8	8	┪	<5.2
	MOVKG	<5.6	<5.4	<3.7	<3.8	H	63.5	t	1	╁	\dagger	†	\$	\$5.6	ş V	-	<5.2
moeno(1,2,3-cd)pyrene	₩0/Kg	<2.8	<2.7	- -	017	t		†	+	╅	†	+	÷	<5.6	9>	H	\$25
Neormalene	νο/Kα	\$2,5	165	=	100	t		\dagger	+	┨	1	-	?	<2.8	ŝ	H	90
Phenanthrene	₩o/Ka	867	127			†	91,	+	+	┥	٦		82	8	065	╁	*
Pyrene	SVO.				2	+	8.5	×	┪	-	<2.5	<5.2	2	800	Ş	190	
Surrogates					۲۱.۷	÷	\$18 \$18	1	<1.8	-	H	┝	8	50	•	+	
2.4-DCAA	į										1	1			7	┨	ş
2.4.5.6-TCMX	,e	23	Z	Z	99	98	\$	┝	┝	8	2	2	5				
DBC	ę	23	78	28	61	19	2	\vdash	╁	-	<u> </u>	5 2	? \$	8	87	2	8
o.terbres.d	ę	8	និ	148	જ	200	8	5	8	<u>ا</u>	 - -	 	¥ 5	- -	3	R	38
	æ	73	78	2	74	73	80	H	╀		2 6	5 8	3 2	e e	8	21	116
Metals								+	d	2	3	8	à	8	ē	8	101
Action	mo/Kg	1600	1900	110	8	1900	\vdash	r	H	ŀ	H	ŀ					
Ammony	morkg	دع	Ş	2	┝	\dagger	╁	t	+	+	,	4	+	2		_	3000
ABOUTC	ОУ/ОШ	40.8	8.05	98	╀	t	+	,	+	+	┪	ş	-	٦	Н	H	బ
Bertum	DAYOU.	80	89	8	╁	3 8		+		+	+	4		Н	┝	L	ê
Beryflium	mo/Ko	2	0	-	t	+	+	1	┪	+		4	Н	-	┝	L	ē
Cadmium	mo/Ko	2	10	 -	+	╁	+	+	┪	+	┪	Ц	<1	H	┝	╀	\ \?
Calcium	₽ V	2	8	259	9	23	5	- S	V	v	v	4	⊽	<2	H	┝	8
Chromium (total)	mg/Kg	\$	89	7	╀	╁	╀	+	\dagger	┥	┪	4	Н		H	Ľ	900
Cobert	mg/Kg	8		ę	+	+	+	+	+	+	7	_		Н	┝	┞	-
Copper	mp/Kg	8	80	ê	9	╀	+	+	+	╅	┪	4	~	Н	Н	┡	8
uo.i	mo/Ko	88	8	2	╁	╀	╀	+	+	+	+	4	4	Н	Ι	L	22
Ceac	mg/Kg	4.1	-	1.5	╁	7	+	3 6	3	3 6	B:	7800	┥	\dashv	1200	2500	2200
TI NEW TOWN	тожа	71	65	22	110	2	2	╀	+	+	+	4	┪	1	-	_	2
Manganese	mg/Kg	2	~	v	t	┿	+	+	+	+	+	4	3	2	Щ	Ц	8
Mercury	moʻKa	\$0.0°	t	t	1	┿		†	* °	2 3	+	4	4	Н	Ц	L	೩
Nickel	mo/Ko	r	t	t	t	┿	\dagger	+	†	┪	┥	_1	-	Н	L	L	=
Potassium	morka	t	270	,	t	+		Ç	†	+	\dashv			┝	L	L	-
Setenium	o June	†		3 ,	\dagger	+	+	+	┪	-	_		⊦	â		Ļ	Š
Siver		t	7	7	» ·	4	7	°	<2 <	H	┝	L	8	2	╀	╀	Ţ
Sodium		†	3	3	1	_	٦			┞	┞	┞	╁	ł	+	+	Ţ,
Prelium	S :	5	3	4		Ц	Н	┞	十	┞	╀	1400	+	+	╁	5 1	ə
Variation	E S	31	3	┨	÷	دا دا	L	├	5	╀	╀	╀	+	<u></u>	8 9	2 0	8 .
302	Ž	2/5	8	+	-	Ĥ	V	ľ	V	┝	H	2		+	╁	y :	, l
	BVA	62	B	÷	\dashv	-	1	-	4.	H	12	ł	+	+	+	+	:
									ł	l	$\frac{1}{1}$	4	1		_	_	-

1 = value is between the Method Detection Limit and the Practical Quantitation Level
J = value is below the established limit for accuracy
* * 8310s were diluted (1:5)
** 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC106		SSC106 SSC107		SSC108		SSC110		SSC111	SSC112	SSC113	SSC111 SSC112 SSC113 SSC114 SSC115	SSC115	SSC116 SSC117	SSC117	SSC118
		6/29/98		07/27/98	96/22/20	06/23/38	07/08/98	7/8/98		98/22/98	06/22/98 06/22/98 08/13/98		07/14/98 08/20/98	08/20/38	06/20/36 06/21/36	08/21/98	08/28/98
			duplicate						duplicate								
Other Perameters																	
Pulk Depairs	o/cm ³	98.0	0.85	96.0	0.71	0.38	98'0	0.8	0.74	96.0	0.69	0.41	0.83		0.91	0.18	9
	meo/100g	53	≂	=	2	\$5	13	33	22	18	82	85	20	0.65	21	9	2
Demant Collde	8	88	8	8	8	8	ā	2	8	- 69	8	æ	85	- 69	55	52	3
recent conce	S U	4.9	5.8	9.7	6.6	6.5	4.2	6.4	4.5	5.1	6.4	9.9	6.6	6.5	9	7.7	7.6
Decision	opun cu	10000	12000	260000	34000	2900	620000	830000	870000	170000	180000	3600	120000	36000	13000	2000	2800
Tachica (Mo. 4)	*	100	100	8	8	8	2.68	8.66	99.4	100	901	99.3	901	100	100	98.7	5
Ted.::4 (No. 10)	*	5 06	7 66	88	8	8	99.5	89.3	7.8	6.66	2.86	676	99.5	6.66	9.66	87.8	98.3
Todaire (No. 10)	*	- 8	7.98	6	7.78	67.5	85.8	628	77.6	88.2	7.98	88	9.96	1.66	1.88	67.1	91.3
Techno (No. 60)	*	93.1	35	78.8	8	58.5	59.8	57.7	53.7	73.3	91.3	77.3	93.3	96.7	95.5	60.3	83.8
Texture (No. 100)	,	65.9	7.99	282	7	43.4	13.2	12.8	15.1	21.3	46.7	50.9	97.5	44.7	56.4	44.1	60.2
Techno (No 200)	*	7	122	3.8	10.2	10.6	3.9	4.5	4.8	7.7	4.6	37.6	8.6	1.2	10.9	20.4	31.3
Total Ownerin Carbon	C)	>26700	>26700	>26700	>26700 >26700	>26700	>26700 >26700 >26700	>26700	>26700	>26700	>26700	>26700	>26700	>26700	>26700	>26700	>26700

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established first for accuracy = = 8310s were diluted (1:5)

Table B-15. KSC Background raw data for soil focations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		011000	0000		10000												
		08/21/98	08/21/98	96/60/90	06/09/98	55C123 06/09/98	SSC124 06/06/98	SSC125 8	SSC126 06/08/98 0	SSC127 06/08/98	SSC128 9000000000000000000000000000000000000	SSC129 (06/08/98 0	SSC130 06/06/98	SSC131 08/08/98	SSC132	SSC133	SSC134
Organochiorine Pesticides (8061)				Ī		1	1	1	1	1	1						0000
4,4' - DOD	W/Ka	ŀ	ļ	ļ	ļ	ļ		┪									
4.4' · DOE	2		₹	0	Ş	3	<5.6	Н	<4.6	<4.6	<3.8	<21	Ş	5	Ę	1	
4.4' - DOT	2			٠	3	¢4.2	<5.6	5.6	<4.6	<4.6	<3.8	5	2	478	848	: 2	?
Aidrin	2		ē	Ŷ	<5.3	<4.2	<5.6		<4.6	<4.6	<3.8	ī	12	870	877	3 8	•
Alcha - BHC	2	*	₹	ş	65.3	<4.2	<5.6		<4.6	\$.6 6.6	43.8	2	1	27.8			2
Bets - BHC	DY G	3	ē	\$	\$33	4.2	<5.6	<5.6	<4.6	¢4.6	98	12	120			***	2
Chlordena (eloke)	W.KO	*	Ę	ç	<5.3	<4.2	<5.6	<5.6	9.4.6	970	85	Ę	Ę				€ :
	W/Ko	4.4	c 11	Ş	<5.3	<4.2	-	65.6	448	1	2		; -		•	•	Ç
Chlordane (partma)	40/Kg	<4.4	11	\$	\$3	42	45.8	45.6	187		900	, ,	:	\$	9	₹	<15
Chlordane (total)	DA/Or	4.4	₽	\$	<5.3	645	t	+		;	9	ş	•	8	×4.8	~14	<15
Detta - BHC	MO/Ka	7	Į				000	+	\$	\$	3.8	2	Ş	<7.8	<4.8	\$1.V	¢15
Dieldrin	WO/Ko	7	1			 	†	+	+	4.6	8	<u>دی</u>	<14	<7.8	<4.8	=	\$55
Endosultan I	2000		1			Š	+	+	┨	4.6	<3.8	<21	<14	47.8	<4.8	15	1
Endosultan II (beta)				٥.	5.65	?	<5.6	┨	<4.6	<4.6	<3.8	15>	₹.	8/>	×4.8	1	1
Fodosulan Olikata		•	₹	0	<5.3	42	٦	_	<4.6	×4.6	438	150	=	8/2		 	
Endin	DV/V	7	₹	ÿ	<5.3	<4.2	H	┝	4.6	949	860	ē	1	2,0		;	2
Court Ald Land	MyKa	<4.4	- -	<\$	<5.3	<4.2	┝	┝	949	446		\dagger	†		\$		Ş
Endrin Augenyde	MO/Kg	4.4	11>	ŝ	<5.3	<4.2	t	╁	╀		900	+	•	\$	\$4.8	₹	\$
Endrin Ketone	₩⁄W	**	Ę	\$	65%	143	t	+	+		9	Ş	4	¢7.8	<4.8	~14	1 5
Gamma - BHC (Lindane)	wa/Ka	175				*	\dagger	+	4.6	9.40	<3.8	Ş	<14	<7.8	<4.8	1	\$5
Heptachlor	2005			†	?!	Š	+	+	\dashv	┪	<3.8	~	<14	<7.8	4.8	=	į
Heptachlor Epoxide (a)				0	50	275	<5.6	┥	-	<4.6	<3.8	<21	\vdash	8/2	<4.8	15	*
hertachlor Frontide (h)			Į	0	533	<4.2	_	-	_	<4.6	<3.8	┝	×14	7.8	877		
Isodon	SV OV	•	ē	ΨĮ	<5.3	<4.2	_	┝	<4.6	H	H	\dagger	t	2/2		;	2
Methowother	70/K0	\$	₹	ş	53	<4.2	H	H	┝	H	╀	+	t			:	9
Mino	MVK0	\$ 4	Ę	ŝ	<5.3	<4.2	H	┝	╁	H	╀	\dagger	t		0,0	•	3
Tovanhana	DY.O.	\$	Ę	ŝ	653	<4.2	<5.6	<5.6	\vdash	979	38	1	t			:	2
	10VK0	8	2 2 3	Ş	<5.3	<4.2	H	┝	╁	╁	╀	+				:	ç :
PCB. 1018/1949							┝	├	┝	╁	╁	+	\dagger	;		•	ę
DCD:1994	W/Ko	\$	2	ŝ	જુ	<42	-	H	H	┝	t	╁	╀	,	\dagger	+	
120 000	WXK9	₹	<u>د</u>	ŝ	63	<42	┝	┝	H	H	t	210	1	2		+	3
POPIESSE BOBISAS	MVK9	ş	<110	-\$0 -\$0	<\$3	290	┝	H	t	╁	t	+	╁	; ; ;	†	+	3
FCB-1248	νο/Κο	×44	<110	ŝ	ŝ	3	╁	╁	\dagger	╁	+	†	+	8,	\$	9	2 25
PUBLICATION	₽9/Kg	â	<110	°20	<53	4 2	85	ŝ	9	9	3 8	\dagger	+	\$	†	+	8
ı	W/Kg	â	<110	-96 -	<53	<42	┞	\vdash	╀	╁	t	t		<u>ا</u>	\dagger	†	3
Critical Nerbicides (8151)					-	-	H	╀	╁	+	+	t	+	\$	8	\$	8
2.12.4.5 The Inchine Control of Silver)	MVK9	₹	89	<15	<16	<13	H	t	t	t	╁	+	+	+	+	†	
2.5.5.1 Inches Cymercally Science Science (2,4,5 · 1)	νο. Κο	₹	8	<15	<16	Н	H	┢	r	t	t	110	\dagger	\dagger	+	†	ş
3.5-DCBA	No.W	÷	8)		9	Н	Н	H	t	┞	t	╀	15	\dagger		\dagger	\$ 1
4.0 4. Dichlossopherson/Pertrain and 10.4 Pick	70.Kg	Ş	ŝ	\$	ş	<13	Н	┝	t	╁	t	╁	+	t	+	t	3
4 - Mirochand	MO/KG	₹ :	8	\$	ş	+	, 82	, 28	ر23 ب	¢3	H	╁	38	8	+	1 2	3 4
Acifluorien	2,200	†	3 8	<u>.</u>	2	+	+	1	H	Н	Н	<110 •	H	╁	t	t	5
Bentazon	N. Control	:	3 5	₽		+	+	1	٦	Н	H	H	H	H	┝	t	â
Chloremben	DVAN.	:		₽	+	+	+	\dashv	H	Н	┝	×100	H	t	+	t	¥
Dacthal	OV.	*	8	+	+	┪	\dashv	Н	Н	\vdash	┢	┝	H	t	t	†	į
Dalabon	DAYON S	;	8	\$; }	 	+	+	Н	Н	\vdash	H	┝	t	ŝ	5 5	†	3 4
Dicamba	D. A.		3 8	\dagger	+	+	\dashv	Н	Н	Н	H	\vdash	H	╁	+	t	į
Dichloroprop (2-(2,4-Dichlorophenoxyloropropic acid)	200	:	şķ	\dagger	\dagger	\dagger	+	┥	\dashv	Н	Н	<110	┝	┝	╀╌	✝	ŝ
Dinoseb		:	t	\dagger	†	+	+	┪	\dashv	\dashv	П	Н		H	<15 <15	t	45
YdDM	╄	28	t	2 5	†	2 5	+	+	\dashv	+	<19	<100	Н	Н	H	â	3
MCPP	WO/Ko	1	t	†	8 5	+	+	+	+	+		۲	Н		┝	H	95
Pentachlorophenol	┿	7	t	t	t	+	+	+	+	┪	7	\dashv	Н	H	<290	H	55
Pictoram	wo/Ka	1	18	\dagger	2 5	2 5	+	+	70	┪	+	<110 <	د/ا	· 0#>	\vdash		\$\$
				1	┨	┨	┥	┨	┥	┥	┨	-		H	<15	┝	ŝŝ

I = value is between the Method Defection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy = 8310s were diluted (1:5) = 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** a value is an outlier and not used in the analysis to develop the KSC Background screening values.

	<u></u>	SSC119	SSC120	SSC121	SSC122	SSC123	SSC124	SSC125 S	SSC126	SSC127	SSC128	SSC129	SSC130	SSC131	SSC132 (SSC133	SSC134
		Jan. 1, 30	VOX 1/30		as as as as	2000				2	www.						
PAH (6310)					Н			H									
1 - Methylnaphthalene	MO/Ko	<23	\$5	<25	<27	<21	د <u>28</u>	\$Z>	72>	د23	<19	×100	0</th <th>0#></th> <th><25</th> <th><!--0</th--><th>ş/9</th></th>	0#>	<25	0</th <th>ş/9</th>	ş/9
2 - Methylnaphthalene	₩.	<23	\$	<25	427	₽	88	8	\$	ş	<u>د</u>	8	گ گ	ş	\$	ę Ş	8
Acenaphthene	MO/Kg	623	ŝ	şŞ	\$	Ş	8	8	₹	Ş	ş	8	675	ş	\$	ş	ę;
Acenaphthylene	W/Kg	*	Ş	ŝ	65.3	4.2	65.6	65.6	\$ 6.6	\$.	89	Ş	₹	87.5	8 8	÷	÷
Anthracene	W/Ka	នូ	\$	8	<27	²²	89	83	8	\$	Ş	2	5	ş	\$	Ę	Ę
Benzo(a)anthracene	₩0/Kg	<2.3	45.4	42.5	<2.7	2.7	8.5	82	3	\$3 23	<u>د</u> 5.5	ş	ş	₹	\$25	⊽	9,
Benzo(a)pyrene	₽9/Kg	د3	Ŷ	જ	6 3	<2	دې	ខ	င့	ç	7	ş	8	\$	Ş	8	8
Benzo(b)fluoranthene	03/Joh	44	<10	4>	<5	*	<5	<5	*	*	<3	<19	<12	د/>	٥4	<12	¢14
Benzo(g,h,i)penylene	03/61	<4.5	<u>دا</u> ۱	ŝ	<5.4	<4.2	<5.6	<5.6	<4.7	<4.6	<3.9	<21	<14	6.7>	6.4>	<14	<15
Benzo(killuoranthene	₽V/V	ន	ş	Ş	Ş	2	ဗ	ç	ç	ŝ	2	<12	80	ŝ	ŝ	80	65
Chrysene	Joy.Ko	<2.3	65.4	<2.5	42.7	42.1	<2.8	42.8	<2.4	<2.3	<1.9	410	' >	*	<2.5	\	47.6
Dibenzo(a,h)anthracene	₽VQ/K	<4.5	Ę	ŝ	45.4	<4.2	6.5.6	<5.6	<4.7	<4.6	<3.9	<21	<14	6.7>	<4.9	√14	<15
Fluoranthene	₽/Ko	<4.5	Ę	ŝ	<5.4	<4.2	<5.6	<5.6	4.7	<4.6	<3.9	421	¢1¢	6.75	6.4.9	41.	<15
Fluorene	₩o/Ka	¢4.5	₹	ş	\$54	<4.2	65.6	656	47	×4.6	<3.9	₽	₹.	6.75	6.49	Ę	45
Indeno(1.2.3-cd)ovrene	W/Ka	<2.3	554	42.5	427	42.1	<2.8	<2.8	42.4	<2.3	41.9	95	\\$	¥	42.5	\ '\	9.7>
Nachthalene	₩.	<23	ŝ	\$	427	120	82	85	\$	83	61>	2 8	ر درم	\$ \$	\$\$	ر م	د <u>7</u> 6
Phenanthrane	₩0/K0	623	<5.4	42.5	75	421	428	<2.8	<24	<2.3	<1.9	95	2>	\$	<2.5	7>	47.6
Pyrene	DX/O/	<2.3	<5.4	<2.5	<2.7	42.1	<2.8	<2.8	<2.4	<2.3	<1.9	ot >	<i>t></i>	45	<2.5	د/	47.6
Surrocates																	
2 4.DCAA	8	128	148	136	35	120	88	8	2	2	121	8	33	8	26	8	22
2.4.5.6.TCMX	×	57	51	8	28	2	122	9	83	9	128	=	83	7.	53	88	82
	8	7.	٤	8	1	6	145	122	6	115	ž	111	116	16	25	8	8
o-famboout	* *	2	117	8	a	8	8	8		è	à	=	 	92	8	2	8
Metals																	
Auminum	morka	1200	3500	1300	1500	2300	1200	1900	1200	1300	610	2800	3600	2400	069	3700	9100
Antimony	mo/Ko	دع	9>	63	دع	42	<3	લ્૩	ę	ç	7	<12	8>	<5	<3	8	6
Asenic	mo/Ko	¢0.7	<1.6	1:-	9.0	=	40.8	18.0	40 2	<0.7	90>	3.81	2.1	412	¢ 0,	15.2	1.4
	mo/Ka	702	39	8	8	425	\$	9	85	85	\$3	2120	8	8	85	ŝ	<u>6</u>
Bendium	mo/Ko	⊽	Ş	2	2	v	2	2	⊽	v	⊽	ę	ş	2	V	2	2
Cadmium	mo/Ko	v	င့	2	8	⊽	8	2	⊽	⊽	ē	ş	\$	2	ē	\$	\$
Calcium	mo/Kg	3100	12000	18	35000	380	8	1500	9	3	929	7500	12000	3300	910	5400	2
Chromium (total)	mo/Kg	1.9	£	2	2	4.9	<2	2.5	2.2	2.1	ŀ	9>	6.7	3.6	۲,	6.2	15
Cobatt	mg/Kg	L>	<16	8	8>	ŷ,	8	8	<i>\\</i> 2	<2	Ŷ	دع	421	<12	د/	5	នុ
Copper	mg/Kg	<i>(</i> 2	<16	<8	8>	ş	8 9	8	ر ₂	~	ę	ŝ	Ş	~12	۲>	25	Ş
lron	mg/Kg	920	1100	1700	1400	1800	0 23	510	\$ \$	380	8	969	2	ş	220	ş	3300
pee_1	то/Ко	4.3	10	150	3.1	7.8	3.9	5.4	2.2	2.4	-	62	35	3.6	£.	2	24
Magnesium	то/ко	380	360	2100	2400	1100	1200	1600	089	620	490	11000	7900	2400	\$20	0096	7300
Manganese	mg/Kg	6.9	•	6.4	7.9	5.6	2	5.8	2.2	2.2	2.6	<و ا	56	10	2.2	**	9.5
Mercury	тожо	0.026	0.16	40.02	\$0.00 0.00	60.01	\$0.0°	0.0241	<0.01	<0.01	<0.01	<0.06	<0.04	<0.02	<0.01	40.0 ×	0.0681
200	тоЖо	2	ر 16	\$	ê	ê	\$	8	<i>\\</i>	<i>L</i> >	9>	<31	<21	<12	- 12	421	Ş
Potessium	mo/Ko	\$	\$	1200	1600	98	919	8	370	250	170	4100	2600	260	110	6200	4300
Selenium	mg/Kg	6>	9>	6>	ေ	7	લ્ગ	çş	<3	<3	د2		و	<5	<3	- 8>	69
- SAINS	mg/Kg	65	9>	6>	જ	ح2	ç	دع	<3	<3	<2		8 >	< 5	<3	8>	6>
Sodium	οχ/ο _ш	270	130	7300	12000	3000	6400	9300	3900	2400	2600	0	35000	8300	1800	00029	32000
Thellium	mp/Kg	Þ	ေ	Z>	ح	-l>	د 2	<2	V	دا	<1		<4	<2	۲,	<4	44
Vanadium	mg/Kg	2.4	6.1	7	4.4	4.3	2.9	3.6	2.5	2.2	1.6	18	12	6.9	2.4	13	28
Zinc	mg/Kg	13	<16	11	13	ę,	8>	Ş	Ş	۲,	ş		421	¢12	12	421	¢23
															!		

1 = value is between the Method Detection Limit and the Practical Quanitation Level J = value is below the established limit for accuracy * = 8310s were diluted (1:5) * = 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		•															
		08/21/98	08/21/36	06/09/98	SSC122 06/09/98	SSC123 SSC124 SSC125 SSC126 SSC127 O6/06/96	SSC124 06/08/98	SSC125 06/08/98	SSC126 06/08/98	SSC127 06/06/98	SSC128 06/08/98 (SSC129 SSC130 06/08/96 06/08/98		SSC131 8	SSC132 S	SSC133 9	SSC134
Other Parameters		T	Ī	1	1	1											3
								_			ľ			ŀ	ł	ı	
DUM Density	p/cm,	98.0	0.18	9.0	0.57	-		2	ŀ	į		ļ		1			
AC.	med/1000	2	9	٤	=					9	=	2	0.21	0.66	0.94	78.0	0.24
Percent Solids						•	ñ	2	°	8	9	-	5	24	12	S	8
H (db)	,	†	7	8	K	ę,	88	\$	7	72	98	92	24	2	8	7	3
	ó	-	ص ص	- 43	.	8		6.3	<u> </u>	١					3		7
(estativity	opundo	12000	7400	٤	ā	Ş	1			2	9	, i	9./	6.8	¥	5.6	6.9
exture (No. 4)	×	٤	٤	1		3 3	3	8	3	S R	220	8	48	74	200	26	3
Birture (No. 10)	36	8	3 6	3 8	2 2	3 5	200	88	8	2	8	93.1	100	8	8	7 66	9
exture (No. 40)	,	38	20.70	7 6	2	3 3	8	8	88	2.6	60.6	71.9	92.4	88	6.66	8	2
exture (No. 60)	,	3 6	3 8	8 6	G G	R	5 6	628	88.5	ន	8	53.9	65.8	7.2	26	85.8	87.4
exture (No. 100)	*	0 0 0	3 8	3 6	8 5	S	200	ŝ	833	2.0	2 88	20.2	57.9	7.7	242	83.7	85.3
exture (No. 200)	8	12.8	325	3 8		֚֚֓֞֝֟֝֟֝֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֟	*	2	37.6	8	8.8	46.7	48.2	52.6	72.9	75.2	205
otal Organic Carbon	Ī,	Ţε	28700	Т.	00.00	2		872	//	e 2	8.2	9	26.8	16.7	5.8	282	-
	1	7		7	3000	-	W/62	256/00	22000	2800	2000	>26700 >26700	26700	>26700	>26700	>26700	×26700
																	1

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		3	90000		000000	900000	000000	900000	000000	00/01/00	00/41/90	6/47/00		00/17/00	00/17/00	80/1/20	06/17/08
		96.650.000	3	cate	98/80/0		96/60/0	26.00	8	96.80	200	8	plicate	86/199		200	200
Organochiorine Pesticides (8081)											П	П					
4.4 DDD	0X/00	<4.7	6.7>	<\$	<4.1	<5	<4.9	<6.2	<4.2	<3.3	<3.3	<3.8	*	<3.3	<3.3	ÿ	<3.8
4,4' · DDE	Mo/Kg	<4.7	<7.8	¢\$	<4.1	<5	<4.9	<6.2	5.4	2.3	22	2	7	<3.3	2	4	43.8
4,4' - DDT	MO/Kg	<4.7	<7.8	\$	2	\$	6.49	<6.2	¢4.2	<3.3	ß	63.8	70	ć3.3	<3.3	3	Ş
Adrin	S S	¢,	¢7.8	ç	\frac{1}{2}	Ş,	649	6 26	242	63.3	633	338	v	53.5	63.3	\$ 1	6.5.4 6.5.4 6.5.4
Apha - BHC	NO.	.	9,7				2 3	3 6	֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓	36	36		1	3	3		200
Beta - BHC	2 2	3	9,5	8 4			200		;	315	3 5	9 6	7	3 6		7	98
Chlordane (arpha)	2000			 	†	†					5		1	66	5	7	865
Chordene (garring)	200			*	†	*			, 2			85	7	293	683	7	38
Chickers (Cours)	NAKO VAKO	247	27.0	3 50	1	; 5	65	\$	\$ 25	633	333	865	\$	333	633	2	<3.8
Caldia	, ovik	1,90	67.8	ş	-	\$	6.49	6.2	4.2	433	_	<3.8	Ŷ	<3.3	633	\$	43.8
Endowline	, o,	4	8%	Ş	1.45	\$	649	<6.2	<4.2	633	¢3.3	43.8	\$	<3.3	<3.3	2	<3.8
Endosultan II (beta)	₩.ov	7.40	6.7.9	\$	\$ 1.4	\$	<4.9	<6.2	<4.2	<3.3	<3.3	<3.8	<4	<3.3	<3.3	*	<3.8
Endosultan Suffate	10/Kg	c4.7	47.8	\$	<4.1	Ş	6.4>	<6.2	<4.2	<3.3	<3.3	<3.8	*	<3.3	<3.3	<4	<3.8
Endrin	0λ/Q/I	<4.7	<7.8	<5	<4.1	\$	<4.9	<6.2	<4.2	<3.3	<3.3	<3.8	*	<3.3	<3.3	<4	<3.8
Endrin Aldehyde	00/Kg	<4.7	<7.8	ŝ	<4.1	¢\$	<4.9	<6.2	<4.2	<3.3	<3.3	<3.8	<4	<3.3	<3.3	4>	<3.8
Endrin Ketone	W/Kg	<4.7	6.7>	جو	<4.1	<5	<4.9	<6.2	<4.2	<3.3	<3.3	<3.8	4	<3.3	<33	\$	<3.8
Garmma - BHC (Lindane)	<i>μ</i> 9/Κ9	<4.7	6.7>	\$	<4.1	\$	4.9	<6.2	<4.2	<3.3	<3.3	<3.8	*	<3.3	<3.3	ţ	<3.8
Heptachlor	₽9/Kg	<4.7	<7.8	Ş	<4.1	ç	¢4.9	2 9	4 2	<3.3	<3.3	<3.8	*	<3.3	633	Ŷ	<3.8
Heptachlor Epoxide (a)	MO/Kg	<4.7	<7.8	\$	<4.1	\$	<4.9	<62	<4.2	<3.3	<3.3	86	Ÿ	<333	<3.3	\$	3.8
heptachlor Epoxide (b)	μο/Κο	<4.7	<7.8	\$	<4.1	\$	<4.9	<6.2	<4.2	43.3	<3.3	<3.8	*	<3.3	69	₹	43.8 8.6
Isodrin	WO/Kg	<4.7	6.7>	<5	<4.1	<5	<4.9	<6.2	<4.2	<3.3	<3.3	<3.8	v	633	633	\$	<38
Methoxychior	μο/Kg	<4.7	<7.8	\$	41	Ŷ	67.9	\$2	242	633	693	43.8	₹	633	63.3	₹	<3.8
Mirex	₩ø.	\$7	<7.8	\$	ş	ç	\$	\$ \$	\$2	533	\$33	43.8	ş	<3.3	63.3	3	<3.8
Toxaphene	₽X/Q/	c4.7	د ا	ş	2	\$	8,9	\$2	4 2	333	633	43.8	\$	63.3	<3.3	Ŷ	3.8
Aroclors (8082)	200	,	,	ş	,	ş	97	697	677	133	/33	8	40	733	233	240	85
PCP-1010/1242	O O	197	2,78	8	145	3	₹	9	3	88	\$33	860	·40	\$33	ŝ	9	85
DCB 1939	, ook		, A	8	ī	8	65	39	3	\$33	છુ	85	9	ŝ	ŝ	9	8
DCB-1248	wo/Ko		8/2	8	3	8	6	ŝ	3	8		80	°40	8	జీ	9	88,
PCB-1254	, Vo	\$	8/\$	ŝ	2	ક્	649	â	3	8	ŝ	×38	<40	<33	<33	×40	<38
PCB-1260	oy.Vo	(Å)	\$2	ફ	142	85	649	â	<42	લ્લ	88	×38	<40	<33	ج33	·*	<38
Chlorinated Herbicides (8151)																	
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvax)	MO/Kg	<14	<24	<15	<12	<15	<15	Ę	<13	ę,	ş	<12	<12	ş	ş	2 7	₹ ₹
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	₩.	\$	70	<15	2	\$	\$	ŝ	¢3	ę	Ş	?	ş		2	ç ₁ 5	2 5
2,4-Dichlorophenoxy acetic acid (2,4 - D)	M/Kg	Ş	Ş	Ş	2	2	\$	3 8	2	\$		۶	֚֭֚֚֚֭֚֚֚֚֓֞֝֜֝֝֟֝֟֝֝֟֝֟֝֟֝֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֡֡֝	•	۶	3	3 5
3,5-DCBA	W.Kg	*	8	=	2	2 4		Š	2	₹ 	\$ {	3 5	3 5	2 5	2 5	3 5	3 5
4-(2,4-Uchiorophenoxy)outync acid (2,4 - Ub)	S S		3	*	1	**	**	5 8	1	2 5	950	43	1	9	2 2	2	- - - - - - - - - - - - - - - - - - -
4 - INITIONIERO	W/Q		3	2 2	12	÷	\$	Ş	¢13	95	410	<12	<12	9	ę	2 2 7	~12
Registro	0/V0/	2	7	¢15	2 1 2	\$5	\$5	63	c13	ę	¢10	<12	<12	ş	ş	~12	<12
Chloranian	λογ	41.	*	¢15	25	\$	¢15	31	c13	¢10	¢10	<12	<12	<10 <	د <u>ا</u> 0	<12	<12
Dacthal	₽VVV	7.	757	<15	<12	<15	¢15	<31	<13	<10	<10	<12	<12	1 <10	<10	<12	<12
Datacon	VQV	414	424	<15	<12	<15	<15	æ>	<13	<10	<10	<12	<12	¢10	<10	<12	<12
Dicamba	MO/Kg	<14	<24	<15	<12	<15	<15	œ,	<13	<10	¢10	<12	<12	ę	¢10	<12	<12
Dictrioroprop (2-(2,4-Dictriorophenoxy)proponic acid)	MVK0	<14	<24	<15	<12	~15	\$	Ŗ	<u>6</u> 3	ş	ę	<12	<15	ę	ş	¢12	<12
Dinoseb	MVK0	Ş	₹	¢ 1 5	<12	\$ \$	÷	ē	ĘĮ.	ş	9	27	200			2	27
WCPA	No.	240	98	88	R	8 5	35	355	85	800		\$ 5	\$ 5	3 5	3 5	\$ 5	2 5
MCPP			Š	3	3 5		3 4	S S	3	3 5	9	3	1	3 5		1	
Pertachlorophenol	S S		\$ 100	Ş	1	2 4	*	18	2 5	ę	9	÷	1	Ę	1	9	3
Ficoram	MAN		5														

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		06/06/98	98C136		26,000	2000	880138	SSC140	SSC141	SSC142	SSC143		¥	SSC145	SSC146"	SSC147	SSC14
				18	96,60		Bevenion	assan or	8679129	06/16/98	06/16/98		6/17/98	06/17/98	06/17/98	06/11/96	06/17/98
PAH (8310)			I		t	1	İ	T	Ī	1			duplicate				
- Methynaphthalene	oy/cor	165	07>	3	ķ	ķ	ļ	Ť	ŀ	ŀ							
2 - Mothyhaphthalene	νο/Ko	424	4	36,	+	***	3	1	ÿ	Ì	Ì	S	Ş	<17	<170	0Z)	Ę
Acenephthene	oy/or	160	4	į,	+			†	ş	Ì	11	<19	<20	<17	<170	83	6 <u>0</u>
Acenephthylene	οχ/o/r	177		3 5	;	şļ.	ş	7	2	Ş	Ş	<19	د20	<17	4170	8	615
Arithracene	Sylve	1					\$	7	\$ \$ \$	<3.3	<3.3	<3.8	7	z	88	\$	465
Benzo(a)anthracene				ş	ş	ş	ş	ŝ	ŝ	<17	<17	<20	421	412	27.5	100	8
Benzo(alowane		, ,	*	\$ \$	Ş	55	\$25		42.1	<1.7	<1.7	<1.9	Ş	417	417	281	
Benzothikusanhana	2	2	ç,	Ş	ş	Ş	_{اگ}		<2	8	3	2	8	0	111	1	9
Person h Institute	WON	Ÿ	¢	2	₹	*	*	┢	2	2	Ş	Ş	7	, 5		,	•
A CANAL STATE OF THE STATE OF T	VO.	×4.8	<7.9	Ş	¢.1	<\$	\$	┢	5.5	465	34	63.9	-		3 6	:	3
	rowe o	Ş	< 5	<3	<2	çş	<u>ئ</u>	T	8	8	5	1	,		3	•	3
Cripsene	₩,0//	<2.4	*>	<2.5	<2.1	42.5	25	t	1		一	,	۶ ۰	y	è	۶	ð
Dibenzo(a,h)anthracene	MV/Kg	<4.8	6.7.5	\$	1.45			t				<u>.</u>	9	÷	Ŷ	42	6.2
Fluoranthene	DX/Q/	<4.8	6/2	5	1	100	 					63.9	V	34	ह	¢4.1	<3.9
Fluorene	WO/Ko	848	675	2	-			t		•	\$ 5	<3.8	\$	<3.4	,34 ,34	c4.1	<3.9
Indeno(1,2,3-cd)pyrene	100/Kg	224		36,) ? *		†	3	50	5	<3.9	c4.1	<3.4	<34	2	<3.9
Naphthalene	νο/Kα	763	97			7 1		†	5	Ž	÷	د <u>ا</u> 6	3	<1.7	<17	2	6 12
Phenanthrene	ω/νον/	100	1		,	; ;;	Ş	†	3	ş	22	¢19	\$	99	630	ş	61×
Pyrene	SVO!	100	1		7	3	ş	+	2	Ş	÷	<1.9	د2	<1.7	230	Ø	615
Surrogates				5.5	7	6.3	8	┥	- - - -	5	₹	<1.9	42	<1.7	45	3	e1>
	3	8	77.	133	1	5											
2,4,5,6-TCMX	ļ	į	ye.	3 5	5 7	<u> </u>	;	*	8	۶	8	22	æ	7.5	901	=	8
280	,	ā	3 2	8	=	8 2	ē	2	3	88	8	69	20	29	88	2	8
p-terphenyl	,	8 8	× 5	=	8 8	5 8	3 2	8	ž	=	8	107	115	8	=	1	8
Metals			3	3		£	28	88	ž	-	ā	88	88	61	•	28	8
Auminum	mo/Ko	4100	2005	Som.	H	250	H	ı									
Antimony	mo/Ko	ç	ŀ		+	3 ,	3	+	8	2	2	ş	210	929	230	8	4200
Arsenic	Syou				+	3	+	✝	2	V	8	8	\$	~	8	3	2
Berum	Ş	86,		<u> </u>		7 8	\dagger	┪	9	20.5	=	¢0.6	<0.6	<0.5	6.5	99	900
Benyfilm	S C		,	3 5	╅	3 4	3	╅	8	8	Ş	<23	<24	<20	8	\$24	423
Cadmium	E C		┿	,,	+	y	†	┪	V	V	⊽	v	_	~ 1	Ī	ī	Ī
Calcium	mo/Ko	67000	┿	, 607	+	y §	- I	+	Į.	V	=	Ÿ	⊽	~ 1	Ų	⊽	⊽
Chromium (total)	mo/Ka	4	┿	2	┿		+	+	₹	8	8	280	7	320	630	2300	5300
Cobett	mo/Kα	ç	╀		╁	3 4	†	┰	• •	•	뭐.	ş,	7	2	4.8	11	19
Copper	mo/Ko	1	╀	80	╁		† 	+	,	;	0 1	e	7	٧	\$	ş	9>
lion	mp/Kg	2800	╂	8000	٠	2100	╁	┿	+	5	* 5	2	7	9.	٦	2	8
0880	mp/Kg	5.4	⊢	92	┝	27	╁	٠	+		3	3	†	3	8	8	88
Magnesium	mo/Kg	4900	Н	28	18	875	1400	┿	+	¥	٤	3 5		7	-	3	2
Manganese	mg/Kg	23	┝	2	⊦	2	╁	┿	+	3 2	3 8		†	3	3	8	8
Mercury	mg/Kg	0.0231	⊢	⊢	15	8	╁	┿	: 6	3 5	+	8 5	✝	3 3	\$	2	8
NICKE	mo/Kg	د/	Н	H	⊢	┖	╁	┿	┿	1	+		t	3	5,0	5,	8
- Consistent	mo/Kg	3400	Н	ş	┝	2300	╁	8	t	8		? (?		0	• •	8	ę
Constitution	mg/Kg	Ş	Н	<3	Н	8	H	2	t	0	10		3	• (9 9	2	8
CAVE	mp/Kg	<3	Н	<3	H	8	┝	2	t	,	1	, 	,	y ·	31.	3	3
Sodium	mp/K ₀	11000	Н	1400	3400	4000	1500	000	153	, -	, =	,,,,	**	y	» ·	2	۶ ا
Veneral	ω V	v	3	\$	Н	~	H	8	Ţ	-	Į	 -	十 引。		1	₹ ,	<u>.</u>
Zinc	E S	<u>.</u>	┪	22	•	8.2	\vdash	2.8	2.3	ī	~	V		;	十 小		
	E S		۲	*	\dashv	8.6	Н	6>	14	58	75	2	=	2	, E	3 2	; 5
														1		5	3

I = value is between the Method Detection Limit and the Practical Quanitation Level J = value is below the established limit for accuracy * * 8310s were dikited (1:5) * * 8310s were dikited (1:10)

Table B-15. KSC Background raw data for soil locations. *** a value is an outlier and not used in the analysis to develop the KSC Background screening values.

0.81 0.26 20 35 20 35 6.4 5.7 300 240 100 992.2 99.8 92.4 68.5 97.4 68.5 97.4 68.5 97.7 35.1 21.8 11.4	SSC135	SSC136	SSC137	SSC137 SSC138 SSC139 SSC140 SSC141 SSC142 SSC143	SSC139	SSC140	SSC141	SSC142	SSC143	SSC144		SSC145	SSC145 SSC146** SSC147 SSC148	SSC147	SSC148
Other Parameters Op/Em/lem 0.31 0.93 0.91 0.81	86/60/90	08/21/98	86/60/90		86/60/90		96/16/98	96/11/90	06/16/98	6/17/98		06/11/98	06/17/98	06/11/98	06/11/98
Other Parameters gykm² 0.31 0.93 0.91 0.84 1.2 0.81 med/100g 24 28 20 13 12 0.81 xmed/100g 24 26 81 66 87 87 S. U. 8.5 8.6 8.6 6.8 7.9 6.4 ohm-cm A4 790 8.90 170 85 100 xme 86.7 89.8 89.8 100 89.5 100 xme 87 86.7 89.8 87 89.8 100 89.8 100 xme 87 81.2 86.7 82.9 80.9 87 89.8 87 89.8 87 89.8 87 89.8 87 89.8 87 89.8 89.8 87 89.8 87 89.8 87 89.7 89.9 87 87 87 87 87 87 87 87 87 87 87		duplic	ate						-	Ť	duplicate				
appen 0.31 0.93 0.91 0.84 1.2 0.81 % 70 26 20 13 12 20 % 70 42 66 68 79 64 S.U. 8.5 8.8 8.6 6.8 79 64 Ammorn 44 790 530 170 50 300 % 86.7 89.6 99.8 170 59.8 100 % 87 84 85 89.9 87.5 89.8 % 81.2 89.6 85.2 89.9 87.5 89.8 % 81.2 84 85.7 80.9 87.4 87.4 87.4 % 86.7 50.8 53.5 42.4 53.1 82.7 % 86.7 50.8 53.5 42.4 53.1 82.7 % 86.7 50.8 53.5 42.4 53.1 82.7 % 86.7 50.8 53.5 42.4 53.1 82.7 % 86.7 50.8 53.5 42.4 53.1 82.7 % 86.7 50.8 53.5 42.4 53.1 82.7 <	ameters	_													
meq/100g 24 28 20 13 12 20 % 70 42 66 61 66 67 S.U. 8.5 8.6 8.6 6.8 7.9 64 Mm-cm 44 790 530 170 50 300 % 96.7 96.8 99.8 100 99.5 100 % 87 94.6 95 99.9 97.5 99.8 % 81.2 94.6 95 99.9 97.5 99.8 % 87 94.6 95 99.9 97.5 99.8 % 87 96.7 94.6 95 99.9 97.5 99.8 % 87 56.7 50.8 53.4 90.4 97.4 % 56.7 50.8 53.5 42.4 53.1 82.7 % 56.7 50.8 53.5 42.4 53.1 82.7 % <td>H</td> <td>Н</td> <td>H</td> <td>1.2</td> <td>0.81</td> <td>0.26</td> <td>98.0</td> <td>1.1</td> <td>ı</td> <td>1.1</td> <td>1.1</td> <td>1.1</td> <td>1.1</td> <td>1.4</td> <td>1.1</td>	H	Н	H	1.2	0.81	0.26	98.0	1.1	ı	1.1	1.1	1.1	1.1	1.4	1.1
% 70 42 66 81 66 67 S.U. 8.5 8.8 8.6 6.8 7.9 6.4 ohm-cm 44 790 8.9 170 8.9 100 % 96.7 99.8 99.8 100 89.5 100 % 87 94.6 85 92.4 90.4 97.4 % 81.2 84.7 89.7 80.9 97.5 99.8 % 50.7 50.8 53.5 42.4 80.4 87.4 % 56.7 50.8 53.5 42.4 80.1 82.7 % 56.7 50.8 53.5 42.4 53.1 82.7 % 56.7 50.8 53.5 42.4 53.1 21.6	meq/100g 24		13	12	20	35	•	4.2	8	5	5	31	38	37	જ
S.U. 8.5 8.6 6.8 7.9 6.4 ohm-cm ohm-cm 44 790 870 170 50 30 % 84.7 89.8 89.8 100 89.5 100 % 66.3 89.5 99.2 99.9 97.5 99.8 % 87 84.6 85 82.4 80.4 97.4 % 81.2 84.6 85 82.9 80.9 97.4 % 56.7 50.8 53.5 42.4 53.1 82.7 % 38.4 24.3 24.6 14.3 16.1 21.6] 02 %	Н		99	67	53	78	86	66	98	85	8	66	83	8
ohm-cm 44 790 530 170 50 300 % 96.7 99.8 99.2 99.9 97.5 100 % 87 94.6 95 99.9 97.5 99.8 % 87 94.6 95 99.9 97.5 99.8 % 87 94.6 95 82.4 80.4 97.4 % 56.7 50.8 53.5 42.4 53.1 62.7 % 38.4 24.3 24.6 14.3 16.1 21.8	-	8.8 8.6		1.8	6.4	5.7	5.8	9.4	9.9	6.3	6.1	6.3	5.9	^	7.3
% 96.7 99.8 99.8 100 99.5 100 % 86.9 98.5 99.2 99.9 97.5 99.8 % 87 94.6 95 92.4 90.4 97.4 % 81.7 88.7 89.7 80.9 94.7 % 56.7 50.8 53.5 42.4 53.1 82.7 % 38.4 24.3 24.6 14.3 16.1 21.8	ohm-cm 44	Н		જ	300	240	93000	170000	270000	>880000 >880000 >270000	>880000	>270000	>120000	>10000	2500000
% 96.5 99.5 99.9 97.5 99.8 % 87 94.6 95 92.4 90.4 97.4 % 81.2 88.7 89.7 80.9 97.4 % 56.7 50.8 53.5 42.4 90.4 % 38.4 24.3 24.6 14.3 16.1 21.8	Н	Ц	Н	99.5	100	99.5	100	100	100	100	100	100	100	8.66	99.4
% 87 94.6 85 82.4 80.4 97.4 % 81.2 86.7 82.9 80.9 14.7 % 56.7 50.8 53.5 42.4 53.1 82.7 % 38.4 24.3 24.6 14.3 16.1 21.6		4	Н	97.5	866	92.4	100	6.66	6.66	100	9.66	100	100	2.66	6.86
% 81.2 88.7 89.7 82.9 90.9 94.7 % 56.7 50.8 53.5 42.4 53.1 82.7 % 38.4 24.3 24.6 14.3 16.1 21.8		_	Н	90.4	97.4	68.5	87.9	88.9	99.5	92.9	93.2	87.5	9.68	75.3	- 29
% 56.7 50.8 53.5 42.4 53.1 82.7 % 38.4 24.3 24.6 14.3 16.1 21.8		Н	Н	80.9	2.7	59.2	ぁ	96.4	97.3	62.6	62.2	72	75.7	54.8	52.8
% 384 24.3 24.5 14.3 16.1 21.8	_	Н		53.1	62.7	35.1	33.8	45.8	46.9	29.5	26.2	21.9	22.8	19.4	24.6
	Н	Н		16.1	21.8	11.4	2.4	2.1	2.2	3	3	3.2	3.4	4.1	8.4
_	>26700	1		>26700	23700	>26700		>25000	>25000	>26700	>26700	>26700	>26700	>26700	>26700

1 = value is between the Method Defection Limit and the Practical Quantitation Level J. = value is below the established limit for accuracy = = 8310s were district (1:5) = 8310s were district (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC149		660161	1000												
		06/17/98	06/17/98	06/17/98	SSC152 6/16/98		SSC153 S	SSC154***	SSC155-	SSC156	SSC157	SSC158****	SSC159	Γ	SSC160	550161	660,163
Ordenochlorine Preticides (8081)					٦	dicate		_	98/100	B8/21/98	96/17/98	06/17/98	6/17/98		06/17/98	06/26/38	80,80
4.4' · DOD		1					l	Ì	Ì	1	1			duplicate			?
4.4' · DOE	S S	┪	<3.4	<3.3	H	t	5	:	į	į	1						
4.4 001	DX/GI	7	<3.4	<3.3	=	-	77	3 2	3	3		<3.3	<3.3	<3.4	633	Ę	200
Aldrin	100 Kg	<3.3	<3.4	<3.3	t	$^{+}$: 2	B	2	87	<3.4	44	<3.3	43.4	633		3
Aloha Blan	WO/Kg	Н	43.4	333	t		3 5	8 3	2	633	43.4	8	433	43.4	Ę		
	MO/Kg	<3.3	50	ê	t	\dagger	†	3	633	633	<3.4	<3.3	693	185		Ť	
260	W/Ko	┝	33	5	t	†	555	5	633	<3.3	43.4	<3.3	¢33			1	5
	AO/Ko	<33	34	:	\dagger	+	†	6	<3.3	<3.3	┢	<3.3	693			7	333
Chordene (germa)	Ş.	<333	2		36		<3.3	<3.3	<3.3	<3.3	\vdash	<33			3	÷	ဗွ
Chlordane (total)	JW/Ko	23		3	+	7	┪	<3.3	<3.3	H	H	23			200	٦	<3.3
Delta - BHC		3	200	63.3	7		-	<3.3	683	t			1	77	<3.3	_	<3.3
Dieldrin	WAY	433	\$3.4 \$3.4	33	<3.3	┝	t	Ę		\dagger	\dagger	7	7	<3.4	H	✝	633
Endostation	S Ka	<3.3	<3.4	<3.3	H	t	t	3	1	1	+		_	<3.4	r	t	66,
Crafton II de	MO/Ko	<3.3	<3.4	665	t	t	†	+	7	٦		Г	۲	T	t	t	
CINCUSURANT II (DOTR)	₽Q/Ka	<33	2		+	†	+	\dashv		Г	H	T	t	Ť	+	7	63.3
Endosultan Suffate	, m/Kr	,			+	1	٦	-	r	r	t	T	t	†	+	٦	<3.3
Endrin		3	•	2	┪	╗	۲.	┝	t	t	+	Ť	†	7	٦	_	<3.3
Endrin Aldehyde	O L	\$3.3	43.4	<3.3	<3.3	H	┢	╁	100	†	+	1	7		H	Н	633
Endrin Ketone	No.	63.3	43.4	<3.3	-	┢	╁	╁	†	t	+	7	┪		-	H	693
Gamma - BHC (Lindana)	WO.K	<3.3	3.4	<3.3	┝	<3.3	╁	╁	t	+	\$	7	٦	Т	┝	╁	6.6
Heotachior	70/Kg	<3.3	<3.4	<3.3	┝	t	t	+	†	+	+	٦	-	Н	┝	t	:
Hampohlov Enovide (=)	WO/Kg	<3.3	<3.4	<3.3	t	t	+	+	7	┪	٦	-	H	t	╁	+	3 6
herischice Equida (b)	₩o⁄Ko	<3.3	43.4	<33	t	t	+	+	7	٦		H	۲	t	╁	+	3
(a) Bourout choruse (b)	W/Kg	<3.3	434	13.9		†	╁	7	7	۲	H	T	╁	t	+	+	63.3
isodrini A-A-	νον Κο	<3.3	7	╁	†	+	╁	┥		H	├	t	╁	+	+	+	33
Metroxycanor	₩o/Ka	633	75	╁	†	+	┥	4		Н	╀	t	╁	+	+	┥	633
	40/Ka	633		╁	+	+	┥	닉	۲	┝	┝	t	┿		+	3	6
loxalgrene	MO/Ko	<33		36	200	200	63.3	┥	<3.3	<3.3	43.4	t	193	+	500	+	69
Aroclors (8062)				+	+	+	4	Н	┝	⊦	┞	t	+	†	+	7	633
PCB-1016/1242	WO/Ko	5,3	Ę	\dagger	†	+	+			╀	+	t	╀	†	+	34	္ပ
PCB-1221	νο/Ko	Į.	 	†	†	+	\dashv		H	H	-	t	†	†	+	+	
PCB-1232	φ/χον,	1	1	+	+	+	+	_	┝	r	╀	t	†	+	+	┥	8
PCB-1248	WO/Ko	Ę	5	+	+	+	-		H	H	╀	t	\dagger	+	+	+	<u>용</u>
PCB-1254	2	3 5	3	+	+	\dashv		\vdash	H	t	+	\dagger	+	+	4	۲	
	1	3 8	3	3	چ ج	\ \ \ \	<33	\$33	33	8	100	3 6	3 8	3	8	\$	ęş S
Chlorinated Herbicides (8151)		3	3	+	\dashv	Н	H	\vdash	+	t	+	+	+	+	\dashv	Н	8
2-(2,4,5-Trichlorophenory)propionic acid (2,4,5 - TP) (Silvay)	100	,		+	+	Н		╀	╁	t	+	\dagger	+	1		Н	8
2.4.5-Trichlorophenoxy acetic acid (2.4.5 - T)		Ť		9	<10 <	<10	П	╁	+	5	+	†	+	+	\dashv	Н	
2,4-Dichlorophenoxy acetic acid (2,4 - D)	2		;	┪	+	٦		┝	t	t	+		+	+	\dashv	Н	9
3,5-DCBA	, OK	Ť	→	2	+	+	<10	×100	¢10	t	012	\dagger			ş	9	¢10
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	00/Ko	T	 	\dagger	+	\$ P	1	H	-	┝	╀	95	\dagger	+	+	7	ş
4 - Mirophenol	W/Ko	Ť	†		+	+	7	Н	L	H	H	\dagger	t	\dagger	+	+	9
Actiuorien	wo/Ko	T		\dagger	۲ ا	Ş	7	Н	<u> </u>	1	01×	\dagger	\dagger		+	7	9
Bertazon	no/Ko	T	+	\dagger	\dagger	+	\$ \$	Н	-	┝	╁	t	\dagger	+	+	; }	₽
Chloramben	DX/O/I	T	+		+	+	7	Н		¢10	+	01	\dagger	~	+	+	ş
Decthe	S Vov	t	+	\dagger	\dagger	+	7	Н	H	┝	0.5 0.5	t	t	\dagger	+	+	ءِ
Delapon	y X	t	+	\dagger	200	\$ P	٦	Н		H	+	\dagger		+	\$	۲ ا≎	١ء
Осатов	, X	t	+	+	+	+	٦	Н	┞	9	95	\dagger	\dagger	\dagger	+	┪	<u></u>
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	νογ γογ	T	+		9	+	٦	Н	┝	t	+	T	+	\dagger)	┪	2
Dinoseb	, OX	t	╁	†	000	ę 2	7	Н	<10	\$10 \$10	╀	Ť	2 5	\dagger	+	+	وا
MCPA	200	t	+	2 .	†	+	٦	Н	t	t	+	T	+	†	+	+	2
MCPP	2/0/	t	+	+	00 - 100	+	Н	H	t	0017	╁	Ť	+	+	ر داه	┥	2
Pentachiorophenol	, S	t	+	۲ 3	7	8 5 8	۲	H	<100 <100	t	╁		+	38	+	× 200	8
Pictoram	98/08	t	+	\dagger	₽ -	+	7	Н		┝	╀	T	十	†	8	╁	8
		1	1	1	┨	┥	0 100	Н	<10 <1	┝	-	t	+	v '	+	┥]
											1	1	┨	┨	┨	\dashv]

I = value is between the Method Detection Limit and the Practical Quantitation Level
J = value is below the established limit for accuracy
* = 8310s were dituted (1:5)
** =8310s were dituted (1:10)

Table B-15. KSC Background raw data for soil locations. *** x value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC149	SSC150	SSC151	SSC1		SSC153	SSC154***	SSC155*	SSC156		SSC158	SSC159		SSC 180	550161	SSC182
		06/17/98	06/1/98	96/11/90	6/16/98			06/16/98	06/17/98	06/17/98	06/17/98		6/17/98		06/17/98	06/26/98	06/26/98
						duplicate								duplicate			
PAH (8310)																	
1 - Methyingphthalene	1/0/Kg	<17	<17	<17	Н	<17	<17	<17	18 >	4 1>	<12	4 12	/ 1>	41>	41>	412	<i>(</i> 1)
2 - Methylnaphthalene	₩.	¢1>	÷	<17	<17	<17	<17	<17	×84	170	<17	21>	<17	417	412	<17	412
Acenaphthene	<i>μ</i> ο/Κα	~1	۲۱>	¢12	41	<17	<17	<17	√84	<17	<17	21>	<17	41 2	<17	<17	417
Acenaphthylene	ν γ γ	ş	\$	<3.3	<3.3	<3.3	<3.3	<3.3	<17	<3.3	<3.4	<3.3	<3.3	<3.4	<3.3	<3.4	<3.3
Anthracene	WVK9	ş	Ş	ş	Ş	Ş	4	<17	ŝ	<17	<17	<17	<17	<17	21	<17	<17
Benzo(a)anthracene	10VK0	<1.7	6	(-	4.7	<1.7	<1.7	<1.7	<8.4	<1.7	<1.7	6.3	<1.7	<1.7	<1.7	<1.7	<1.7
Benzo(a)pyrane	₩.	\$	7	چ	•	Н	ج	<2	8 >	•	9	Z>	<2	<2	ş	\$	%
Benzo(b)fluoranthene	ωσ/Kα	<3	<3	31	<3		ç	લ	41 >	ç	19	-9	ŝ	ę	ŝ	2	ŝ
Benzo(g,h,i)perylene	00/K0	<3.4	<3.4	<3.4	<3.4	<3.4	<3.4	<3.4	417	43.4	43.4	18	43.4	43.4	434	43.4	460
Benzo(k)fluoranthene	6N/M	6 7	<2	*	2	H	8	2	8	2	2	=	5	5	S	3	8
Chrysene	₽0/Kg	2.41	2.61	4.7	<1.7	H	417	41.7	×8×	2.1	•	14	7	-	17.1	417	417
(Dibenzo(a,h)anthracene	MO/Kg	43.4	<3.4	<3.4	<3.4	H	43.4	43.4	417	<3.4	<3.4	<3.4	43.4	434	43.4	43.4	<3.4
Fluoranthene	MO/Ko	<3.4	<3.4	43.4	<3.4	43.4	434	43.4	417	43.	18	<3.4	43.4	43.4	34	434	43.4
Fluorene	10/Kg	43.4	43.4	4.65	43.4	H	43.4	43.4	45	465	43.4	<3.4	43.4	34	34	434	43.4
Indeno(1,2,3-cd)pyrene	νογ.	<1.7	417	<1.7	<1.7	H	47	<1.7	48	1	<1.7	17	417	<u>-</u>	110	12	\ √-1,2
Naphthalene	/0/Kg	<17	<17	41	5	8	230	<17	2	120	417	<17	₽	Ę	4	92	8
Phenanthrene	MO/Kg	<1.7	<1.7	<1.7	<1.7	Η	<1.7	<1.7	48.4	<1.7	<1.7	<1.7	<1.7	<1.7	47	7	417
Pyrene	µ0∕K0	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	<1.7	484	<1.7	<1.7	3.21	<1.7	4.12	7.12	417	1.5
Surrogates																	
	8	28	8	96	120	136	78	86	2	88	8	116	20	88	8	88	88
2,4,5,6-TCMX	ж	67	47	20	84	86	9/	3	61	99	7	85	83	74	65	86	ន
D8 C	æ	98	85	102	28	69	131	103	114	109	74	150	8	82	ş	137	8
p-terphenyl	æ	æ	28	8	28	ē	87	15	8	93	Ş	102	75	87	63	106	85
Metais																	
Auminum	mo/Ko	8	8	S S	8	8	8	000	8	- 20 20 20 20 20 20 20 20 20 20 20 20 20	1800	1100	99	1500	87.0	980	450
Antimony	mo/Kg	8	8	8	≈	8	8	\$	۶	2	Ş	7	۶	?	<2	<2	< 2
Arsenic	mo/Kg	60.5	<0.5	60.5	800	=	9.0	0.71	60.5	6 0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	1.6
Serium	mo/Kp	ş	Ş	ş	Ş	ş	8	<20	ر 30	~ 50	<20 20	<20	<20	<20	<20	<20	۶Ş
Beryflium	mo/Ko	v	Ÿ	v	v	V	ē	v	Ţ	v	V	۲,	<1	۲,	۱۷	<1	.
Свотист	⊒Q¥0	v	v	⊽	V	⊽	22	=	v	v	v	٦	Ţ	v	v	V	ē
Celcium	₩Q.	300	2000	99	8	8	8	8	2	8	180 081	780	2500	500	88	8	210
Chromium (total)	mo/Ko	٩	2	_	5	3.9	3	2	7	=	2	11	12	22	3.7	=	3.5
Cobait	E S	ç	Ç.	0	* -	e i	ç	8	ا ا	9	Ş	çş	Ş	1 0	Ş	\$	\$
in the second se	S N	è	3 5	2 8	2	2	2	5 2		3	3	02	2	3	-	8	z
ID:	DVA DVA DVA DVA DVA DVA DVA DVA DVA DVA	200	36	2 4	3 ;	3 :	3	3		3	3	1900	877	322	<u>g</u> .	8	P.
Memorium		3 5	2 2	2 5	2 2	2 5	9 5	2 2	48	: :	5	120	2 5	= \$	4,7		• [
Manager	S S	2	3		67		5	3	5	=	3	2	Ş		3	3 6	3
Marrier	Z YOW	Ş	100	100	0 00.21	0.0151	350	1000	200	è		15100	200		2 5	. 6	3 6
	ν V	ç	Ş	ç	╁╌	ŝ	ç	ç	\$	5	v	\$2	9		5	8	5
Potassium	mo/Ka	2	ន្ត	27	5	8	2	3	Z	2	9	4	9	5	\$	8	240
Selenica	mo/Kg	ş	3	8	8	3	8	2	8	ş	8	7	2	ş	2	2	8
Silver	mo/Kg	ş	8	3	3	3	9	2	8	2	2	77	2	3	8	8	3
Sodium	mg/Kg	121	110	=	9.6	-6	122	161	-	12	23	111	211	151	5	101	100
Thallium	шо/Ко	<1	<ا	دا ا	- ↓>	<1	۲>	þ	2.6	V	•	۱,	⊽	⊽	⊽	₽	67
Vanadium	mo/Kg	1.2	3.9	1.3	2	2.2	1.0	1.4	2.2	2.3	3.8	9.1	2.5	2.3	2.0	-5	_
Zinc	mo/Kg	35	53	31	7.4	70	140	120	88	25	92	7	3	45	7.2	15	6.9

^{1 =} value is between the Method Detection Limit and the Practical Quantitation Level
J = value is below the established limit for accuracy
= 8310s were diluted (1:5)
= 8310s were diluted (1:10)

Table B-15. KSC Background raw date for soil locations. *** value is an outlier and not used in the analysis to develop the KSC Background screening values.

	Č.	07100	S	121.000	CCCIES		100	1000									
	8	06/11/98		06/17/98	6/16/98		06/16/98	06/16/96	06/17/8	SSC156	SSC157	08/16/98 06/16/98 06/17/98 06/17/09 06/17/09 06/17/09	SSC159				SSC162
			_	_		dichie	-		2	3	2	26/1/20	98/10		86/1/86	06/26/98	06/26/98
Other Parameters		l	Ī	T	Ì		I			1	1			duplicate			
Pulk Deneity	ł			Ī													
	D/Cm.	0.82	د .		1.2		1.2	[-	•	ŀ		ļ	ļ	İ		I
	mea/100a	33	101	500	٥	ŗ	0					-	7	5.1	3	-	7
Percent Solids	,	1	1		3	<u> </u>	0.0	â	8	37	1000	90	- 8 -	100	1000	,	,
ALL (Ink)	┨	S	ß	ŝ	8	8	8	8	8	8	88	8	8	8	1	,	•
Mr. (lab.)	S.U.	- 69	7.5	7.1	9.9	6.8	2	8.8	-	3 9	Į.			R	B	8	8
Resistivity	The state of	5	5000	3	+	+		Т		6.0	9.0	6.3	7.2	7,5	9.9	9	6.3
Taxture (No. 4)	•	3	-1	300	5	20000	40000	460000	>2000000 >990000	<u> 1000064</u>	×170000	>2200000	>2000001 > 2100000	210000	A Second	300000	1
	ę	8	8	5. 6.	8	8	100	100	٤	٤	٤	T	33,		_	3	
lexture (No. 10)	%	6 66	8 76	000	08	8	٤	2			3	3	3	3	8	8	8
Texture (No. 40)	ŀ	2113	693				3	ŝ	ŝ	8	8	666	8	6.66	666	901	100
Texture (No. 60)	\dagger		3	200	2	3	6.98	88	97.9	59.1	52.8	99.3	68.6	693	202	2 08	0,00
		29.60	25.7	47.5	ä	2	-	97.3	0 50	47.3	Ę	97.0	1			3	9/19
lexture (No. 100)	*	14.8	243	12.1	37.0	37.0	5					9,79	2,2	78	51.1	4.8	80 80
Texture (No. 200)	+		†	1	1	2	3.5	5	33	19.3	17.9	46.3	17.7	17.2	18.5	22.9	7
Total Omenin Carbon	1	0.7		•	2.6	2.3	2.9	3.7	3.5	4.6	6.2	,	5		4.	ŀ	
	- mp/Kg >2	>56700	-56700 -26700	>26700	>26700	>26700	>28700	287mn	28700	2670	20000	3330		-	-	²	,
						4		3	-	250/W	mocz<	2000	>26/00	>26700	>26700	>26700	0006

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy
= 8310s were diluted (1:5)
= 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC163	SSC164	SSC165		SSC166	SSC167	SSC168	SSC169	Γ	SSC170	SSC171	SSC172	SSC173	SSC174	SSC175	SSC176
	•	06/29/98	96/5290	06/12	qtac	96/30/98	96/30/36	98/06/90	83	icate	86/06/96	86/30/38	86/06/90	86/30/88	86/06/90	6/29/98	6/26/98
A Property of the Property of	l	Ī	Ī		BURNING	Ť	Ī	I	Ì		T	T	T	T	T	T	
Organochiorne Pesticides (SUG1)	1	1.6	ļ	ļ	ļ	ķ	7.67	ķ	ķ	ķ	ķ	23.3	1	1	,	43	465
4.4 · D00	2	5	250	3 5		2 3	3		3	3 6		26	18	180	; =	200	34
4,4 · UDE	2000	127		1		1	2	200	333	65	34	333	434	465	¥	42	434
4.4 - DOI	200	3	3,5			78	437	34	333	333	434	6333	3.4	465	\$	42	3.4
Alche and	, V	437	<35	<3.7	*	43.4	43,7	460	<33	633	46	<3.3	434	434	₹	<4.2	43.4
Bots - BHC	OX/O/	<3.7	<35	437	4.4	150	53.7	43.4	<3.3	<3.3	43.4	<3.3	43.4	4.63	₹	<4.2	<3.4
Objection (alcha)	₩o/Ka	43.7	<3.5	43.7	77	43.4	43.7	<3.4	<3.3	<3.3	<3.4	<3.3	<3.4	<3.4	*	<4.2	<3.4
Chlordene (nemma)	DVO/	<3.7	35	43.7	4.4	43.4	43.7	43.4	433	<33	43.4	<3.3	<3.4	<3.4	*	<4.2	<3.4
Chloridana (total)	₩o⁄Ko	43.7	\$35	43.7	4.4	48,	43.7	\$34	<3.3	<3.3	<3.4	<3.3	<3.4	<3.4	*	<4.2	<3.4
Data - BHC	Š	43.7	43.5	3,7	4.45	<3.4	<3.7	43.4	<33	<3.3	<3.4	<3.3	<3.4	<3.4	<4	<4.2	<3.4
Dieldric	₩.	437	<3.5	<u>ئ</u>	4.4	<3.4	437	450	<3.3	<3.3	48	<33	<3.4	<3.4	ş	<4.2	<3.4
Codos Han I	W/Kg	43.7	<3.5	<3.7	444	43.4	43.7	5.5	633	<3.3	43.4	<3.3	<3.4	43.4	₹	<4.2	43.4
Endoes Man II (heta)	νον.	<3.7	<3.5	43.7	444	434	43.7	43.4	<33	<3.3	43.4	43.3	43.4	<3.4	\$ >	<4.2	<3.4
Thoops And Original	νο/Kα	43.7	<35	537	444	465	37	43.4	<3.3	<3.3	43.4	<3.3	<3.4	<3.4	\$	<4.2	<3.4
Encourage Consists	, o/Ko	437	<3.5	¢3.7	*	5	597	434	<3.3	633	43.4	<33	434	434	₹	<4.2	434
Endrin Aldebude	, QX	43.7	<3.5	43.7	3	50	43.7	43.4	633	c33	43.4	<3.3	434	43.4	*	<4.2	<3.4
Endrin Kalona	, ov	<3.7	<3.5	5	1	465	<3.7	43.4	<33	<3.3	43.4	<3.3	<3.4	43.4	*	<4.2	<3.4
Gamma - BMC (Lindana)	oy/ov	<37	<35	43.7	4.4	43.4	<3.7	434	<3.3	<3.3	<3.4	<3.3	<3.4	<3.4	*	<4.2	<3.4
Hartachior	vo/Ka	43.7	<3.5	437	4.4	434	<3.7	<3.4	<3.3	<3.3	<3.4	<3.3	<3.4	<3.4	*	<4.2	<3.4
Hantschlor Frontida (a)	WO/Kg	<3.7	<3.5	43.7	444	43.4	<3.7	<34	<3.3	<3.3	<3.4	<3.3	<3.4	<3.4	<4	<4.2	<3.4
hartachlor Frontida (h)	₩.VKG	<3.7	<3.5	<3.7	44.4	\$34	<3.7	<3.4	<3.3	<3.3	<3.4	<3.3	<3.4	<3.4	40	<4.2	<3.4
Isodin	WO/Ko	43.7	<3.5	43.7	444	<3.4	<3.7	<3.4	<3.3	<3.3	<3.4	<3.3	<3.4	<3.4	64	<4.2	<3.4
Methorychior	W/Kg	<3.7	<3.5	<3.7	<4.4	<3.4	<3.7	<3.4	<3.3	<3.3	<3.4	<3.3	<3.4	<3.4	3	<4.2	46
Mirex	νο. Vo	<3.7	<3.5	<3.7	<4.4	<3.4	<3.7	<3.4	<3.3	<3.3	<3.4	<3.3	<3.4	7 69	Ţ	42	43.4
hene	MQ/Kg	<3.7	<3.5	<74	68 9	68	<73	88	Ç	6 9	Ç	\$	\$	Ş Ş	8	<4.2	43.4
Aroclors (8082)																	
PCB-1016/1242	₽9/Kg	<37	<35	<37	ş	रु	\$	\$	چ چ	श्	हु	8	हु	8,	9	3	8
PCB-1221	W/Kg	<37	ŝ	¢3	4	ਨ੍ਹ	,37	Ş	8	8	3	83	\$	3	3	3	3
PCB-1232	MO/Kg	437	, 8	ÇŞ	¥,	रु	69	ह	8	8	3	3	3	3	3	¥	3
PCB-1248	MO/Kg	437	ફુ	637	3	ह	(§	Ş	83	3	3	3	3	3	3 5	Ş	3
PCB-1254	VOY.	ŝ	<u>ئ</u>	Ę	\$	3	÷	3	3	3 8	\$ 8	Ş	ş	\$ 5	3 5	3 5	5 5
PCB-1260	/OVG	Ş	ş	Ì	\$	5	Ì	5	3	3	3	3	1	3			3
Chlorinated Herbicides (8151)	,		۶	;	ļ	5	1	9,	9,	Ę	5	5	91,7	950	513	613	8
2-(2,4,5-1nchorophenoxy)propione acid (2,4,5-1P) (Silvex)	DVA.		2 5	1		2 2	1	9	٤	2 5	95	9	ę	92	¢15	£	2 0 0 1
2.4.5 Inchorphenoxy scene and (4.4.5 · 1)	W/Ko	Ę	2 0	Į	7	9	Ę	9\$	95	\$10	95	92	¢10	<10	<12	<13	<100
3 F.DCBA	SY.	Ę	ş	ΙĮ	7	د10 د	11	<10	<10	<10	<10	<10	<10	<10	<12	<13	د 100
4-(2.4-Dichlorophenoxy)butyric acid (2,4 - DB)	Š	₹	د10	411	<14	¢10	<11	<10	<10	<10	<10	<10	<10	<10	¢15	£5	ş
4 - Nitrophenol	MO/Kg	11	<10	<11	<14	<10	<11	<10	ę	ę	ş	ş	ş	ş	<12 <12	£	Ş
Aciliuorien	0X0	Į,	<10	<11	<14	<10	<11	<10	√ 10	¢10	ę	ę	ş	ş	- - -	ş	8
Bentazon	δ.	₹	¢10	<11	*1>	<10	<11	¢10	<10	<10	¢10	ę	ę	ę	ş	ş	ŝ
Choramban	0W04	112	<10	<11	<14	<10	<11	<10	<10	¢10	¢10	ę	ę	ş	₹	£ €	ş
Decthal	9W6	دا ا	c10	11>	<14	<10	<11	<10	¢10	ç 9	ę	ş	ę	ę	2	ĘŞ	<u>\$</u>
Delapon	W/Kg	11 >	ot >	<11	<14	<10	<11	<10	د10 د	ę	ş	ę	\$	ę	<12	Ş	دِ 1
Dicamba	M/Kg	Ę	ot>	11	<14	ę	Ę	ę,	ş	ę	ę	9	ę,	ę	212	Ş	812
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	10/Kg	41	٠ 10	<u>.</u>	\$	ş	1	÷	ş	ę	ş	ę	Ş	ş	2 2 2	E I	8
Dinceeb	WK0	₹	ş	Ę	41	ę	Ę	ş	9	₽	9	00,0	₽.	چ اچ	21>	£[\$	3
MCPA	W/Kg	\$ \$	ş	823	¢270	8	2 7	8 V	8	8	8	8	8 5	8	2 2	8	300
WCPP	Ş.	2 :	8	Ş;	275	3 5	2 -	3 5	3 5	3 5	3 5	3 5	3 5	3 5	3 5,	3 5	3 5
Pentachiorophenol	OX.	\	2	Ę	*	2 5	į	2 5	2 5			2	2 5	2 5	1	2 5	3 8
Picloram	MAN	V	3	5													

1 = value is between the Method Detection Limit and the Practical Quantitation Level J. = value is below the established limit for accuracy = x 8310s were diluted (1:5) = x8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		601.00	100000		I				ĺ								
		06/29/98	06/29/98	08/12/98		SSC166	SSC167	SSC168	SSC	SSC169	SSC170	SSC171	SSC172	SSC173	SSC174		200
					Cate	200	2000	980500	8	8	96/30/38	86/06/90	96/30/98	96/30/36	96/30/30	06/29/98	08/26/20
PAH (6310)						Ī	Ī		1	Ouproate							
7 - Methymachthalene	MO/Kg	8F.V	Ę	ķ	1	ļ	ŀ									Ī	l
2 - Methythaphthalene	40/Ko	9	2		3 8		Ŝ	È	Ŷ	<17	412	(1)	412	į	ķ	Ŕ	ŀ
Acanaphthane	oy/wi	9			3	₹ V	8	ş	دا ک	<17	41 >	4	Ę	Ė	2		ì
Acenaphitylene	27//001				3	Ì	<19 	<17	<17	<17	1	412	1	:	! [7	Ì
Anthracene			650	è	×4.4	<3.4	<3.7	<3.4	<3.3	633	180	133			3	Į.	ŧ
Benzo(a)enthracene		â	618	¢19	623	<17	61>	Ę	Ę	Ę	1		, ,	3	*	2	<34
Renaviation	28	<1.9	2.8	2.4	<2.3	41.7	419	1	,		•	1	į	È	& %	\$	412
	MO/Kg	3	9	21	-	0	5	1	,		Ì	Ì	11.	<1.7	~	<2.1	\ <u>`</u>
	νον/Kα	ŝ	-	7	-	†	,	,	y	ÿ	2	8	<2	2	8	0	
Benzo(g,h,i)perylene	w/Kn	ŝ				2	?	Ş	ç	<3	6>	\$3	ç	Ş	1	†	·ŀ
Benzo(k)fluoranthene	2			2	121	Ş	ج 37	<3.4	<3.4	460	98	95	R		;	•	2
Chysene	1	y	7	8	<3.0	~	\$	2>	Ş	8	1	1	3	* 1	*	×.3	\$3 2.
Directoria Mantheoreas	WO'KO	<1.9	3.7	<1.9	<2.3	112	612	1	1		 	*	3	8	?	<2	7
	40/Kg	<3.7	<3.5	<3.7	4.5	43.4	1						/:\>	<1.7	<2	<2.1	1.
	νο⁄Κο	<3.7	<35	637	377			3		Š	434	<3.4	<3.4	<3.4	*	643	1
Flucrene	W/Ka	15,					3	ş	34	3.4	<3.4	260	50	1987	k		
Indeno(1,2,3-cd)pyrene	Š				C \$	43.4	<3.7	<3.4	<3.4	<3.4	<3.4	434	1	,	,		Š
Naphthalene			3	A.	2	<1.7	<1.9	<1.7	<1.7	<1.7	417	<u> </u>	† 		;	2	3
Phenanthrane		Ŝ	8	<19	<23	<17	<19	12	Ę	15	k				×	9	-
Pyrene	MO/KG	6. F	¥.6	<1.9	<2.3	<1.7 <1.7	6.5	 -	1	t	† 	 		Ì	2	₹	~1
	W/Kg	<1.9	<1.8	<1.9	<2.3	112	100	ļ	 	†		į	7. .v	<1.7	3	42.1	<u>دا</u> ،
Surrogales										Ì	÷ i	÷	<1.7	<1.7	25	2	1
K,4-UC/W	8	8	4	130	133	99	5	Į									
2,4,5,6-1CMX	ď	57	75		1	8 8	X S	8	₽	2	8	78	2	2	02	4	2
080	×	Ş			+	2	8	Z,	જ	93	8	29	23	8	3		8 8
p-terphenyl	36	305	<u> </u>	1	=	2	22	8	8	<i>"</i>	88	2	8	1	3 4	6	8
Metals		3		3	3	5	8	22	91	3	8	8	ğ	٤	+ = E	3 6	8
Akminum	mo/Ko	7200	2400	Wo.	-	ł	t										
Antimony	movKo	2	,		╁	+	7	8	2 2 2	140	1400	210	200	r	H	- Servi	5
Arsenic	D W	, ,	, ,	ا ا		» :	2	Ş	<2	7	8	8	9	100	╁	3 5	3
Berium	S. W.	3 8		919	4	ŝ	┪	<0.5	<0.5	<0.5	0.51	\$05	é	t	╁	,	٧
Beryffium		*	¥	3	4	┥	٦	89	8	8	8	8	! 5	†	+	8	ê
Cadmium	DVA	<u> </u>	₹ -	┪	┪	Н	Н	⊽	Ī	Ī	V	 -	 -	+	┿	ę	ŝ
Catclum		₹	v	┥	ī	Н		┢	⊽	Ī	t	+	Т	┿	+	Į.	ē
Chromium (total)	S I	B	8	4	⊣	Н	┝	⊢	82	S	╁	+	Т	+	+	v	v
Cobalt	2	₽,	=	+	┪	-	Н	⊢	⊽	ī	╁	+	7	┿	+	8	230
Copper		0 5	₽	e	7	<\$	9>	<\$	\$	\$	\$	2	+	+	2	P	اع
l'ou	2 2	3 8	3 2	+	┪	4	Н	Н	H	=	╁	┿	+	┿	+	e	0
Lead	DOM:	3	3 :	+	+	4	┪	-	Н	8	╀	┝	1	┿	+	3 5	8
Magnesium	S S	1 5	2 5	+	+	4	┪	Н	Н	Į	┝	╁	Т	┿	╀	3 :	3
Manganese	S C	3 8	3 5	4	╅	4	┪	-4		28	┝	H	1	┿	╀	+	9
Mercury	2000	1000	3 8	+	+	4	┥	_	Н	7.5	H	╁	✝	+	+	+	2
Nickel		9 4		4	+	4	4	Н	H	10.0	╀╴	╁	+	+	+	+	2
Potassium		9 8	0 5	4	+	4	Н	Н	┝	\$	╀	╀	+	+	+	4	Ž.
Selenium		3	2,	4	┪	4	Н	Ļ	25	Ş	8	t	+	1	+	4	3
Silver	S C)	9	4	┥	Ц	Н	┝-	2	2		t	+	+	+	4	2
Sodium	OV.	3	2	-	_	Ļ	H	╄	0	,	,,	t	+	31	+	7	ş
The Man	D .	7	=	Ц	Н	L	┞	↓_	=	1 2	, ,	y =	+	9 8	+	7	3
magam	DAGE	Į	⊽	Ц	Н	L	⊢	↓.	Ī		 -	\dagger	+	3	+	2	ē
Znc	morkg	2	2	•	4.6	L		2.5	Ī	 -		+	┿	•	5.5	V	Ţ
	S S S	38	52	Н	Ц	7	28	18	-	 	: 3	-	31		2	8.6	-
												,	1	Z Z	8	31	12

| = value is between the Method Detection Lirrit and the Practical Quantitation Level J = value is below the established lirrit for accuracy * = 8310s were diluted (1:5) * *8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** value is an outlier and not used in the analysis to develop the KSC Background screening values.

		S	2	CVV			2012012012012012012012012012012012012012	200	2000		2	2	2001/01 2001/1 2001/2 2001/2 2001/2 2001/01	2	3	2	3
		2 9	06/29/98	08/12/98		96/06/90	96/30/90	96/36/98	6/30/98		86/06/90	96/30/90	96/06/90	96/30/38 06/30/38	96/06/90	06/29/98	06/26/98
			-	آ	duplicate			•		duplicate							
Other Parameters																	١
	, m.yo	٤	F	-	-	13	Ξ	1.2	1.2	1.3	1.2	1.3	1.1	1.2	1.1	1.2	1.1
DUR COUSNY	men/100n	:	٥	33	82	9	R	_	-	6	8	9	6	17	19	13	8
	,	2	8	22	7	8	8	26	83	86	8	66	96	88	83	78	88
Tercent School	= 5	-	9	7	9/	72	6.9	6.8	7.2	6.8	7.7	7.2	7.4	9.7	6.8	9.7	6.1
(OB) Ld		6000	110000	32000	29000	¢12	140000	2 1>	21×	<12	1600000	<12	<12	350000	64000	450000	<12
Texas Ale Ale	ď	8	8	666	ē	8	8	88.	8	8	83	901	, 001	98.2	100	100	5
1 externe (No. 4)	3	8	8	82 28	8	966	7.86	98.3	8	6.66	5.86	8	9	93.8	6.66	6.66	100
Texture (NO. 10)	, 4	ā	8	8	26	85.4	85.3	696	7.86	7.98	97.5	87.4	83.5	84.7	9.86	8	98.3
Texture (No. 40)	,	٤	7 16	7 88	668	51.4	57.3	82.3	91.6	92.1	7.3	28	28.7	78.3	1.78	91	9.68
Texture (No. out)	2 3	ŝ	57 A	8	57.2	15.3	28	57.8	51.9	33	85	17.6	21.9	51.4	22.7	59.6	48.2
To done (No. 100)	8	=	12.3	20.7	62	4.2	10.5	8	66	9.6	9.8	4.7	9	14.2	11.4	11.7	9.5
Tarris (No. 200)	Ī	226700	00 >26700 >26700 >26700	>26700		>26700	>26700	>26700 >26700	>26700	>26700	>26700	>26700	>26700	>26700	>26700	>26700	9800

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy = = 8310s were district (1:5) = 8310s were district (1:10)

Table B-15. KSC Background raw data for soil focations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		SSC177	4	SSC178	SSC179	SSC180	SSC11	Γ	SSC182		SSC183	SSCIRA	Section	SCOTER	660103	90,000
		823	8	07/14/98	06/13/96	08/13/98	8/13/98		08/22/98		7/14/98	07/28/98	07/28/98	06/18/98	06/18/98	06/18/08
			duplicate				1	dicate			duplicate				3	8
Copanocracine Pesticioes (8051)																
14 - DOU	W/Kg	ج 33	43.4	<3.6	<3.4	<3.4	۲۷	<4.3	<3.4	<3.7	<3.9	<33	6.63	5	ķ	ļ
4,4 · OUE	6WG	633	<3.4	<3.6	<3.4	<3.4	**	<4.3	<3.4	43.7	439	<3.3	633	633	1	
1,4	S S	<u>دي</u>	5	38	\$6	<3.4	*	<4.3	<3.4	<3.7	<3.9	<3.3	633	633	3.6	4.5
Alcha - BHC	2	2	3	\$36	8	53.	₹	<u>\$</u>	<3.4	<3.7	<3.9	<3.3	<3.3	<3.3	43.4	4.5
Beta - BHC				200	3	Į.	V	Ş	\$34	3.7	<3.9	<3.3	<3.3	<3.3	43.4	4.5
Chlordana (aloha)	3	3 6			700	Ş.	₹ V	6.43	\$3.4	3.7	<3.9	<3.3	<3.3	<3.3	<3.4	\$4.5
Chlordana (gamma)	2			ş	3	70	¥	6.43	\$3.4	43.7	<3.9	<3.3	<3.3	<3.3	4.63	\$4.5
Chlordene (Intel)	1	3	3	3	5	634	ş	Ş	3.4	43.7	<3.9	<3.3	<3.3	<3.3	5.6	4.5
Date - PMC	DVQ.	5	200	88	35	36	₹	\$3	<3.4	<3.7	<3.9	<3.3	<3.3	<3.3	43.4	465
Dieldrix	NO.	250	700	336	Ş	\$	¥	6.43	<3.4	<3.7	<3.9	<3.3	<3.3	<3.3	460	4,5
Codos Men	/OK	633	34	43.6	\$	43.4	Ş	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	633	34	24.5
Codes, Man 1 Act -	40/Kg	633	33	43.6	Ş	3.4	70	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	<33	34	3
Endowster Cutters	S X	633	34	3.6	\$3	ş	3	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	<33	43.4	5
Codin	MONKO	5	¥	38	5	Ş	3	<4.3	<3.4	<3.7	<3.9	<33	<3.3	<33	434	645
Codin Aldebids	WO'KG	33	34	436	8	5	\$	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	<33	38	4.5
Entrin Augusty	MO/Kg	33	3.4	36	46	<3.4	**	<4.3	43.4	37	<3.9	633	<3.3	<33	434	,
Camma DLV I indeed	W.Ko	Ç.	43.4	<3.6	\$	34	*	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	<33	34	645
Herischio.	WO'K0		50	437	\$ P	5	ş	<4.3	<3.4	<3.7	<3.9	<33	<3.3	633	34	24.5
Handachice Conside (a)	NO.	5	ş	43.6	5	ş	Ÿ	<4.3	<3.4	<3.7	<3.9	<33	<3.3	633	60	64.5
heriother Econids (a)	W/Kg	\$33	434	\$3 6.6	43.4	34	*	<4.3	<3.4	43.7	<3.9	<3.3	633	<3.3	5	
Inspection Spource (U)	0X/80	Ş	34	99	Ş	3.4	Ÿ	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	<33	34	44.5
Welthousehor	O X	5	7	436	Ş	8	7	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	<33	43.4	2.5
Misso	NO.KG		8	3.6	8	8	₹	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	<3.3	460	45
Toxenbane	DY.	3	3	9	70	34	₹	Ş	3.4	<3.7	<3.9	<3.3	<3.3	<3.3	34	4.5
Araches (BOS)	S N	2	3	25	8	Ş)	8	8	3.4	¥/>	<78	99>	8 9	<3.3	\$34	4.5
PCB-1016/1242	27/01	,	ļ	,	1											
PCB-1221	2 2	3 5	3	3 2	÷	हुं	\$	<u>Ş</u>	₹	63	8	8	ج ج	<33	<u>ئ</u>	¢45
PCB-1232	2 2	3 5	3 3	8 8	3	3	ş	ŝ	ş	رغ ع	89	8	<u>چ</u>	<33	₹ -	45
PCB-1248	D'A CO	3 5	3	8 8	3	3	ş	<u>ş</u>	ठु	<37	œ,	<33	<33	85	₹	\$
PCB-1254	2 2	3 8	3	8 8	3	Ş	ş;	<u>\$</u>	रु	\$	85	ć33	<33	ر عع	\$	¢45
PCB-1260	SVA.	3 8	3 3	3 8	3	3	3	ŝ	रु	Ę	88	ŝ	ج ج	κ,	8	¢45
Chorinated Herbicides (8151)	2	3	3	8	\$	3	ş	<u>\$</u>	Ş	Ş	89	8	ŝ	န်	\$\$	<45
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	wo/Ka	ę	9	Ę	7	†	33	:	ļ		ļ	1				
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	Jo.Ko	ę	9	Ę	1 8	t	\$ 5	8 =	2 5	+	*	2 5		₽	ş	7
2,4-Dichlorophenoxy acetic acid (2,4 · D)	0X/6√	¢10	<10	₽	2	38	\$ \$ \$	- - - - -	2 2	1	100	2 5		2 5	2 5	÷
35-DCBA	WK ₀	¢10	ę	41	ر10 د	Н	<120	<13	9	÷	¢15	ę	9	9		
4-14.4 UKI MONIQUING MONIQUING BCIG (2,4 - DB)	S S	ş	ş	₹	ş	H	<120	<13	<10	11 >	<12	9	ę	9	ę	7
Acidinofen		\$	2	Ę	2	ş	2 2	ş	ę	41	<12	<10	<10	ot.>	ę	4.
Rentazon	DVA.	\$		₹	₽;	+	8	Ę	휭	₹	<12	<10	<10	<10	ę,	414
Choramban	3,00		2 5	┇	2 5	\dagger	2 2	₽ 	ş	₹	÷	ş	ę	<10 10	<10	~14
Decthal	, VOY	2 2	1	 -	2 5	\dagger	2 5	;	ęļ:	₹	₹	ş	ş	ş	<10	41
Delapon	WO/Ko		1	十 末	2 5	\dagger	3 5	2	2	Į.	₹	ę	ş	ş	ş	<14
Dicamba	, S	1	†	 		t	2 5	3	Ş :	ē	2	운	ş	ş	ڊ و	<14
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	oy/co	۱		 	? {	\dagger	+	3	2	Į.	ř	ş	ş	ş	ş	<14
Dinoseb	, ov	ę	: =			 	3 5	 	2 9	Ę	₹	ş	ş	ę	ş	÷
MCPA	D)/Q	8	8	83	0.50	t	╁	2 8	2 5	- 6	2 5	2 8	9	₽ .	ę	7
MCPP	1/0/Kg	<100	20 00 100	0ZŞ	0270	╁	╀	 8 8	8 2	8	980	3 8	3 8	3 8	3 8	3 4
Pentachiorophenol	6X/6/	<10	<10	<11	65	┝	⊢	5	ê	Į	ê	9	3 5	3 5	3 5	
Pictoram	19/Kg	<10	<10	411	ş	┝	╀	€	ş	₹	1	15				:
						l	ł						,		2	:

^{1 =} value is between the Method Detection Limit and the Practical Quanitation Level J = value is below the established limit for accuracy = 8310s were diluted (1:5) = 8310s were diluted (1:5)

Table B-15. KSC Background raw data for soil locations. *** value is an outlier and not used in the analysis to develop the KSC Background screening values.

		25017		SSC178	8/17/00	250130	930181		55C15Z		22/18/2	02/20/20	02/28/98	08/18/08 08/18/08	05/18/08/	06/18/9R
		878		06/*1//0		200	8	900	000000		1	200	00000		3	3
			OUDINCARE	1	1	Î	1	OUDING		J	OUDICATE			1	I	
PAH (\$310)																
1 - Methymaphthalene	MO/Kg	<17	<17	<18	<17	<17	02>	ÇŞ,	<17	<19	Ŗ	<17	<17	¢12	9	ş
2 - Methylnaphthalene	MO/Kg	<17	<17	<18	<17	<17	ر ح	ζŞ	<17	<19	ر م	<17	<17	<17	3	Ş
Acenaphthene	03/0v	<17	<17	<18	<17	<17	<20	Z2>	<17	<19	<20	<17	<17	<17	<17	Ş
Acenephthylene	MO/Kg	<3.3	<3.4	<3.6	<3.4	<3.4	<4	<4.3	<3.4	<3.7	<3.9	<3.3	<3.3	<3.3	<3.4	45
Anthracene	DA/KD	<17	<18	<19	<18	<17	<20	ŝ	<18	<19	Ş	<17	<17	<17	<17	ş
Benzo(a)anthracene	M/Kg	<1.7	<1.7	<1.8	<1.7	<1.7	2.21	<2.2	<1.7	<1.9	<2	<1.7	3.8	4.8	21	<2.3
Benzolalowene	ρ.γ.ν.	8	2	\$	2	\$	2	31	۲۶	?>	<2	ح2	6	9	22	دع
Benzolbilluoramhene	vo/Ko	8	ဗ	8	ŝ	ဗ	-	15	ဗ	\$	*	ဗ	14	6>	15	*
Benzo(a,h.)iperviene	00/Kg	434	43.4	43.7	43.4	4.65	5	16.7	<3.4	<3.7	<3.9	<3.3	8	9	28	4 6
Benzofkilltoranthene	wo/Ka	2	ş	8	3	8	21	-2	Ŋ	<2	8	د 2		8	17	ç
Chysene	vo/Ka	7.5	15	41.8	41.7	<1.7	4.6	<2.2	<1.7	<1.9	Ÿ	<1.7	9.9	6.7	26	<23
Dibenzola hianthracane	₽V/Kg	434	43.4	<3.7	<3.4	43.4	v	64.3	<3.4	<3.7	<3.9	<33	43.4	<34	29	4.6
Fluoranthene	no/Ko	43.4	434	<3.6	<3.4	<3.4	*	<4.3	<3.4	<3.7	43.9	<3.3	4	2	\$34	\$ 4. 6
Fluorene	Mo/Ko	<3.4	43.4	<3.7	43.4	43.4	*	<4.3	<3.4	43.7	<3.9	<3.3	<3.4	<3.4	43.4	¢4.6
Indeno/1.2.3-cd)ovrene	νο/Κα	41.7	417	×1.8	417	417	3.	:	<1.7	<1.9	ζ5	<1.7	5.2	4.8	14	<2.3
Nachthalane	Mo/Ko	-€	412	₽ ₽	15	417	۶Ş	ŝ	<17	<19	ر 20	<17	41 2	<17	<17	423
Phenanthrene	vo/Ko	41.7	41.7	41.8	417	<1.7	7.8	Ξ	<1.7	<1.9	<2	<1.7	5.4	1 6.4	<1.7	<2.3
Pyrane	νο/Κο	<1.7	41.7	4.8	<17	417	?	<2.2	<1.7	<1.9	<2	<1.7	15	5.8	<1.7	<2.3
Surrogates																
2,4-DCAA	%	64	99	29	90	156	100	148	76	35	98	88	2	82	ಪ	8
2,4,5,6-TCMX	%	53	71	108	69	74	8	901	126	125	118	<u>5</u>	138 88	112	Ę	8
080	%	78	73	143	108	119	65	87	88	97	ē	-	8	æ	29	£
p-terphenyl	%	88	62	8	73	61	-87	11	88	86	ğ	8	æ	88	8	8
Metals																
Auminum	руусы	8	28	82	200	8	Ş	8	2 <u>5</u>	22000	2000	8	8	026 826	8	8
Antimony	mo/Kg	7	V	2	ş	V	?	7	ş	<2.2	9	v	2	V	9	ç
Arsenic	m0/Kg	40.5	\$05	\$95	\$65	60.5	¢0.6	99	=	<0.6	¢0.6	40.5	9.0	2:	12	1.2
Barium	mg/Kg	ر د20	ح21	ŝ	421	ςŞ	<24	8	ج21	36	B	ş	Ş	ģ	Ş	\$
Beryflium	mg/Kg	<1	۲	<1	۲,	1>	۲	~1	~ 1	V	V		⊽	v	v	⊽
Cadmium	mg/Kg	۲	٧	1 >	۲۷	<1	· ·	Þ	۲,	¥	·	~	V	<1	V	⊽
Calcium	mg/Kg	1300	670	00099	1200	006	1600	1900	130000	7600	10000	Ш	1900	00006	28000	49000
Chromium (total)	тожо	5	3.6	8	3.4	1.4	V	Ÿ	6	70	~		-	34	⊽	9.
Cobatt	по/Ко	çŞ	ŝ	ŝ	â	ŝ	ę	ę	8	ş	ę		ŝ	Ş	Ş	\$
Copper	mo/Ko	ž	ឌ	8	Ş	2	ę	ę	2.5	ę	ę	Ŷ	Ŷ	8	v	-
lron	mo/Kg	ŝ	5	28	\$	20	120	0 22 20	200	909	13000	22	8	200	200	229
Lead	mg/Kg	9.7	3.6	2	22	2.	7	3.0	~	22	8	3	7.7	2	2	2
Magnesium	mo/Kg	8	180	1200	120	150	8	110	2	1300	069	3	8	9	3	8
Manganese	mg/Kg	42	\$	Ę	2	2	~	2.5	-	2	2	~	7	2	9.0	7
Mercury	mg/Kg	<0.01	<0.01	0.019	*0.01	-0.0v	900	6 0 0	6 0 0	0.07	900	<u></u>	<u>6</u>	ê 0	6 6 9	§
Nickel	mg/Kg	< 5	<>	<5	<5	\$	9>	çę	ŝ	ę	ş	ŝ	ş	Ş	ş	۶
Potassium	mg/Kg	<170	<170	97	8	9 2 3	8	42	8	ž	ş	\$	2	2	\$	8
Setenium	mo/Kg	<2	<2	ح2	8	7	v	ş	ş	•	-	Ş	8	7	ş	Ş
Silver	mg/Kg	ح2	<2	<2	7	8	ح2	٧	2.4	Ş	?	7	3	2	Ÿ	Ş
Sodium	тожа	171	141	110	% %	8 3	80	ŝ	-	2	9	\$	2	8	စ္တ	Ş
Thallium	mg/Kg	<1	۲,	٥	v	⊽	V	V	v	ē	⊽	v	⊽	₹	v	⊽
Venedium	mo/Kg	3	3.7	4.7	7	V	ī	÷	22	22	~	V	75	2.8	ē	-
7,00	ww/ka	20	76	2	9	#	ç	+2	Υ,	4		¥	<u> </u>	5		,

I = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy
* = 8310s were district (1:5)
** =8310s were district (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

		6000	Γ	1000					į							
		3		35C18 SSC179 SSC180	SSC129	SSC180	SSCIBI		CG1700	1000	ľ					
		06/29/98		7/14/96	307178R	07/14/96 08/13/98 08/13/98	00/13/00		200	3		<u> </u>	55.85	SC186	SSC184 SSC185 SSC186 SSC187 SSC188	SSC188
			o solicita	-		3	2		98/22/00	/14/98		28/38 07	0 8686	RV18/QR	07/28/98 07/28/98 06/18/98 06/18/98 06/18/98	90,01,90
Other Decembers							٥	duplicate		T OF	displicate		-	}	8	R
								l	l			1	l			
DOM DENEMY	, moyo	1.2	- 2	۲	:	:		İ	1				_	-		
	100/100	ı		:		?	=	-	1.2		ş	1.1	1.3	12	280	:
Percent Solids	,	ľ	.	2	•	-	٥	12	6	24	8	~	5	5	1	1 8
DH (lab)		S	R	16	6	8	3	77	- 26	2	3	1	, 2		3	8
	S. U.	6.7	7.9	7.4	5.2	5	3,5	ç		1	3	3	ŝ	8	8	2
LIBERTALIA	ohm-cm	510000	(COO)	Conne	S	2000	+,	2	2	5.	7.3	6.9	9	8.2		1,1
Texture (No. 4)	¥	٤	╀	3 2	ℷ	3	회	910000	30000	9800	3800	60000 29	2900000 > 1	>19000n	40000	200000
Texture (No. 10)	ļ	3 8	3 3	93.	3	8	100	8	7.66	8	8	٤	6			
Tarture (No. 40)	٩	A.R.	3	97.5	6.06	\$	6.66	6 66	S	, 8	╀	+			Ŝ	3
Tet - (1)	æ	5. 6. 6.	4.66	93	9 66	282	7 08	9		+	+	1	9/.1	61.0	97.2	286.2
Texture (No. 60)	8	93.4	94.3	ğ	8	è		0.00	8	+	96.5	8	92.4	59.2	716	3
(exture (No. 100)	ď	45.7	1.67	1 5		8	3	2	88	8	92.4	98.3	89.5	2	200	633
Texture (No. 200)	ď	9		3		Š	13.2	107	33.2	69.5	8.07	6.09	888	9		
Total Organic Carbon	Ŀ	Ī	٦,	2 2	1		4.5	3.8	3.8	37.8	24.8	9	2			200
	7	3000		7.00/co	>26700 >26700		>26700 >26700	26700	>26700	>28700	APRILL SETON SECOND	200	┸	_	3	35
					!							3	J	300	>28/00	23000

I = value is between the Method Detection Limit and the Practical Quantitation Level
J = value is below the established limit for accuracy
** = 8310s were diluted (1:5)
** =8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** x value is an outlier and not used in the analysis to develop the KSC Background screening values.

		1000	Γ	1001	101,000	667100	CCC 103	201722	201000	80100	250197	86108		20000
		6/18/98		96/62/90	07/28/98	07/28/98	08/12/98	07/28/98	07/15/98	07/15/98	07/28/98	08/13/98	07/28/98	08/12/98
Control of the second		1	OUDSCREE	1	Ī				Ī	Ī	Ī	Ī	Ī	
Organochiorne Pasticioss (ede.)		I	ŀ	ŀ	,	•	ķ	ļ	36,	3,45	Ķ	867	682	3.6
4.4' - 000	100/K0	:	*	2 5	3		Ç.		3 4	3	386	8	333	36
4.4' · DDE	MOVE	•	;	2	,	3	2 8				3		5	136
4,4 · DOT	200	*	*		3 5		3 4		35.	3	38	38	633	3.6
Adrin	DVA.		,				36.	180	5	35	436	43.8	<3.3	38
April 5:10	200			100	293	34	43.5	160	435	435	3.6	38	<3.3	43.6
Optional (state)	wo/Ko	1	\$ 2	434	<3.3	465	<3.5	434	<3.5	43.5	<3.6	<3.8	<3.3	<3.6
Chicadese (segme)	, ov	1.40	442	334	<333	465	5.8	434	<3.5	<3.5	<3.6	43.8	<3.3	<3.6
Chloriese (viet)	, ovika	•	24.2	434	<33	434	<3.5	43.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Castle and Castle and	ŽÝ	3	<4.2	434	433	43.4	<3.5	43.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Sie Sie	wa/Ka	44.1	<4.2	5	\$33	43.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Endowiden I	₩o/Kg	**	442	465	<3.3	<3.4	<3.5	43.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Codon: Han II (hate)	oy/c	145	442	43.4	433	<3.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Endoes Hen Californ	,/o/Ka	44.1	<4.2	434	<33	43.4	3.8	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Firedox	₩o/Ka	44	<4.2	434	433	43.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Endin Aldebude	oy.or	4	4.2	\$34	<3.3	43.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Endin Katon	√o/Ka	- V	c4.2	<3.4	<3.3	<3.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Garma - PMC (Indana)	₩,0/Kg	-42	<4.2	43.4	<3.3	<3.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	5.4
Harden	wo/Ka	¢4.1	<4.2	43.4	<3.3	<3.4	<3.5	₹3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Handacker Epovide (s)	₩/V	1.40	<4.2	43.4	<3.3	<3.4	<3.5	434	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
hartection Frontide (b)	₩o/Ko	4.1	<4.2	43.4	<3.3	<3.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
feeders	wo/Ko	<4.1	<4.2	434	<3.3	<3.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Lethoughton	₩o/Ka	÷	<4.2	<3.4	<3.3	<3.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Minex	10/Ko	¢4.1	<4.2	<3.4	<3.3	<3.4	<3.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	<3.6
Toxachene	0Wor	<4.1	<4.2	<3.4	29>	99>	<20	89 °	88	د/ا	ڊ <u>۲</u>	<75	8	Ç,
Aroctors (8062)														
PCB-1016/1242	MONG	1	<42	ठु	ŝ	Ş	\$	रु	ફ્	8	8	8	8	8
PCB-1221	W/Kg	ş	¢5	ठ्	8	हु	ŝ	रु	ş	ŝ	8	8	3	8
PCB-1232	MO/Kg	ş	2	Ş	ŝ	3	3	3	ş	3	8	3	3 5	3 8
PCB-1248	WO/Kg	ş	ş	ह	8	ج د	8	3	ş	3	ş	3	3 8	3 8
PCB-1254	MO/Kg	₹	ŝ	Ş	8	3	8	3	3	ş	8	38	3 8	g
	W.Ka	ş	ş	Ş	<u>\$</u>	ş	3	3	8	3	ş	8	3	3
- 1					!	•	ļ	ķ	٩	ļ		·	Ş	:
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	S Vov	2 V	ĘŞ,	ş	9	9	Ş :	\$	2 5	•	•	7	2 5	3 =
2,4,5-Trichlorophenoxy acetic acid (2,4,5 · T)	DX S	Ş	2	\$	2 5	Ş	1	2 5	2 5	1	1	Ę	9	Į
2,4-Dichlorophenoxy acetic acid (4,4 · D)	NOW Y	3	7	2 2	9	ę	₹	9	ę.	₹	₹	Ę	ر د10	ŧ
4.0 4. Dioblomphenomylanturic soid (2.4 - DR)	, Se	252	√ 13	ę	ş	ę	₹	ę	ot.	₽	داا	11 >	<10	33
4 - Nitrophanol	yo.	<12	c13	<10	<10	¢10	د11	<10	<10	41	Ę	۱۱>	°10	÷
Acilionen	M/Kg	<12	<13	<10	<10	<10	Ę	ę	ş	Ş	₹	₹	ę	Ş
Bentazon	νο⁄Kg	<12	<13	<10	ç10	ę	₹	ş	ş	₹	₹	ŧ	ę	Ş
Chloramben	₩S/Kg	<12	<13	¢10	ę	ş	₹	ş		₹	₹	₹	ş	Į
Dacthal	MO/Kg	<12	<13	ş	ş	ş	₹	ş	ş	ē	ē	₹	9	Ş
Dalapon	₽9/Kg	<15	ct3	ę	ş	ş	₹	Ş	ş	₹	ē	ē	9	₹:
Dicamba	WO/Kg	25	Ş	ę	ş	ę	₹	ş	\$	į	5	Į:	2	į
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	WO/Kg	Ç]	ş	ş	ş	Ş	Ş	Ş		Į.	₹	•	2 5	į
Dinoseb	M/W/	Ç	ĘŞ	ę	ş	₽ ;	Ş		2		Ş	- 6	٤	۶
MCPA	W.Kg	2	S .	8	8 8	255	200		3 5		3 6	386	3 8	36
MCPP	W.Kg	R V	3	3 5	3 5	2 5	3 5			7	1	1	9	Ī
Pertachlorophenol	S S	**	:		? =	1	1	Ę	9	Ī	₹	Ę	ę	Ę
Proorem														

I = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy * * 8310s were diluted (1:5) * * 8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

DALLAGRAM		6/18/96	6/18/96	98/5-190 06/29/98	55C191	SSC192 07/28/98	SSC193 08/12/98	SSC194 07/26/96	SSC195 07/15/98	SSC196 07/15/98	SSC197 07/28/98	SSC198	SSC199 07/28/98	SSC200
DALI (***			A spligate		200		8	807/2	96/51//0	86500	0//28/98		07/28/98	
10110			3			•					_		_	
ı													I	
1 - Methymaphthelene	0X/0//	160	ķ	Ė	ŀ	ļ	9,	•		ľ				
2 - Methyfnaphthalene	o)//on	2	Ş	1		3 8	9		۶	\$18	912	<19	ŧ	ş
Acenaphthene	w.VKo	į,	į			8 2	2	Ì	812	e c	×18	<19	<17	8
Aceneorithylene					Ž,	ĝ	91	è	8	<18	<18	<19	<17	€18
Arthresan			Ň	\$ 63.4	<3.3	È	43.5	<3.4	<3.5	<3.5	<3.6	<3.8	<3.3	436
Description of the second of t	OV.	5	ŝ	41	41	98	<18	<17	<18	<18	<18	61>	412	957
Certol and an active	6W/07	•	•	20	<1.7	130	2.01	<1.7	6.1 ∧	×1.8	418	9	1	2
Benzo a joynane	WO/Kg	Ξ	9 1	7	₹>	95. 25.	s	ş	V	8	2	Ş	R	: 2
Benzo(b)fluoranthene	MO/Ko	9	¥	63	63	220		Ş	,				,	\$
Benzo(g.h.)perylene	vo/Ko	26	5	2		٤	12.0	3	3	3	2	63	₽	8
Benzofkifuoranthane	o y	=	,	3	,	2		*	63.5	<3.6	43.6	<3.8	<3.4	3
Chrysene			y.	- -	» :	2	7	ð	ş	۲	~	<2	\$	X
Dihentole Manthracens	2	١	?	2	÷	8	6.3	√	<1.8	<1.8	<1.8	8	417	3
Charles and Control of the Control o	DV/OV	2	-	7.2	43.4	<17	<3.5	<3.4	<3.5	<3.6	<3.6	<3.6	<3.4	597
TILL OF ICHER IS	10/Kg	22	=	8	3	3	21	<3.4	<3.5	<3.6	<3.6	<3.8	34	3
riudiene	40/Kg	<4.2	<4.3	<3.4	<3.4	<17	<3.5	5	335	<36	63.6	865	1	1
Indeno(1,2,3-cd)pyrene	ω/Kα	9.2	<2.1	5.3	417	210	9.6	÷	Ę	5			十)
Naphthelene	₽9/Kg	122	12>	5	Ę	88	=	4	Ę	2				
Phenanthrene	ολ/ον	9.5	19.4	V.12	417	2		•	?	,	9	*	\	₽.
Pyrene	0√Ka	221	100	-	117	240				,			V	2
Surrogates						3	2		2	8.15	41.B	<1.9	Ş	2
2,4-DCAA	ļ,	a	a	9	S	95	925	5						
2,4,5,6-TCMX	8	133	2	2 5	3 2	3 2	8 5	3 5		ָצ	128	8	8	2
DBC	3	ş	4	2 8	3 5	B	8 8	3		8	28	ă	23	87
D-temphenyl		2 8	3 8	3	3	5 8	3	2	2	à	R.	8	g	ន
Metals		1	3			R		•	8	6	æ	=	8	13
Akminan	Tooks	8	9	*	27.0	1000	500	4000						
Antimony	SVOE SVOE	2	1	3	,	3	3	3	۶	B	978	1400	200	Z
Arsenic	mo/Ka	17	100	36	, ,		,	, I	۷	ď	2	₽	8	ð
Barium	mower	, ×,	*	3 8	3			Ş	77	2.4	\$0.5	900	Ĉ.	<0.5
Berylium	200		9	ş -	₹.	ş .	Ş	Ş	Ş	\$	Ş	<23	8ş	22
Cadmium		1	7	<u>.</u>	1	7	₹	Ÿ	Ÿ	÷	V	⊽	<1	۲
Calcium	2	7	- 1	*	Į.	7	ē	v	V	v	Į,	ī	دا	₽
Chromism flotel)	OVA E	3		REZ.	2	8	200	8	20000	190000	39000	7300	12000	350
Cobalt		,	7	₹	Į.	3	*	~	97	4.1	2.6	2	1.7	Ī⊽
Corner	DWA.	8	٠	9	0	ş	Ŷ	Ş	.6	Ş	\$	9>	ŝ	ç
and the second s	D C	7	7	0	0	ş	Ÿ	Ş	ç	Ş	\$	9>	£	ŝ
l and	2	₹:	3	3	6	210	8	8	8	9001	980	290	ş	3
Warnesim		3	3 5	-	7	=	?	V	25	-	=	2.5	2.4	4.9
Ипперава			3	9	5	3	2	2	ž	180	150	140	140	8
Mercus	D L	٩	2	92	2	7	-	=	9.5	17	6.3	5.4	92	5:
Nickel	MONG.	Ş,	5	9	ê.	969	99	9 0 0	<0.01	<0.01	<0.01	<0.05	\$0.0	60.0 1
Dobace	mo/Kg	ę	ę	Ş	\$	v	ŝ	Ş	\$	çŞ	9	ę	2	Ş
Salastina	morkg	3	8	23	ŝ	۶	ş	%	2	3	23	Į	2	12
	mo/Kg	7	2	ş	3	Ÿ	Ÿ	~	<2	2	3	2	2	3
Sales and Sales	₩Q.W	9	9	8	Ø	7	Ÿ	Ø	2.3	\$	3	2	8	3
DOCUMENT.	mo/Kg	ş	8	₽	\$	\$\$	2	110	1400	1300	82	3	25	18
	mo/Kg	Ţ	Ÿ	V	v	1>	دا ا	⊽	⊽	⊽	⊽	v	V	
Variation	mo/Kg	2	25	Ÿ	ē	2	1.4	1.5	8:	2.5	2	2.5	-	1.5
JUIC	mo/Ko	12	12	15	7	95	92	ŞŞ	ş	8.6		26	**	

^{1 =} value is between the Method Defection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy
** = 8310s were diluted (1:5)
** =8310s were diluted (1:10)

Table B-15. KSC Background raw data for soil locations. *** = value is an outlier and not used in the analysis to develop the KSC Background screening values.

						2000	2000	200	200	2000	-	200	SSC:190 SSC:191 SSC:192 SSC:193 SSC:194 SSC:195 SSC:196 SSC:196 SSC:196 SSC:196 SSC:197	3
		6/18/98		96/5790	07/28/98	07/28/98	07/28/98 07/28/98 06/12/98	07/28/98	07/15/98	07/15/98	07/28/98 08/13/98	08/13/98	02/28/98	08/12/98
			duplicate									1		
Other Parameters														
	, myo	0 65	0.61	1.2	0.91	1.1	1.2	1.3	1.4	1.4	1.2	1.2	1.2	-
Bur Density	1001	805	2	-	-	=	=	6	6	2	8	11	7	22
	3	8	5	8	8	97	3	26	8	83	26	88	66	91
Percent Sonos	, ,	3	2	4	-	7.9	1.7	2	8.7	7.4	5	5.9	7.2	7
(DH (lab)		1	130	, 5	140000	100000	410000	230000	130000	44000	72000	280000	1000000	00006
Hesistivity	3		8	٤	+-	8	8	8	4.76	3	2.69	8	9.86	9.66
Texture (No. 4)	٩	909	a	8	g	8	8	9 66	853	1 28	288	86	296	888
Texture (No. 10)	۹	3 8	,	200	8	07.7	7 38	7 80	46.7	667	88	77.8	206	9.20
Texture (No. 40)	, ,	8		e de	200			8	27 €	86	AS A	30.1	24.0	6
Texture (No. 60)	%	52.7	2.5	206	20.0	72.1	8	B	3		3		100	1
Texting (No. 100)	*	42.9	89	46.1	52.6	52.6	45.7	56.3	21.8	2.5	? ?	2	3	20.5
Takinia (No. 100)	8	10.9	14.8	1.9	7.7	7.7	6.4	3.7	2.6	4.6	4.6	5	5.7	4.9
Texture (No. 200)	modko	26700	To	>26700	26700	>26700	>26700 >26700	19000	>26700	>26700	>26700 >26700	>26700	>26700 >26700	>26700

1 = value is between the Method Detection Limit and the Practical Quantitation Level J = value is below the established limit for accuracy • = 8310s were diluted (1:5) • = 8310s were diluted (1:10)

APPENDIX C

Groundwater Data for Kennedy Space Center

Table C-1. Descriptions of groundwater well sampling locations for the KSC Background Study.

Well id	Well depths	Description of location
Dune Subsystem		
Dune 1	Shallow Intermediate Deep	North end of Canaveral National Seashore Parking Area 4-5
Dune 2	Shallow Intermediate Deep	North of KSC Security Zone/Playalinda Beach: all three - in disturbed area near Parking Lot 2 behind bathroom at south end of lot.
Dune 3	Shallow	North of KSC Security Zone/Playalinda Beach: shallow - in disturbed area near Parking Lot 5 behind bathroom on north end of lot.
Dune 4	Shallow	Dunes East of LC41: shallow – 1st Southeast of LC41 entrance
Dune 5	Shallow Intermediate	North of Corrosion Test (Solar Panel) site East of Road along dune ridge
Dune 6	Shallow Intermediate Deep	North of Pump Station 7 Road west of Cape Road
Dune-Swale Subsystem		
DS 1	Shallow Intermediate Deep	North of Haulover/Shiloh Shiloh Scrub Restoration Site
DS 2	Shallow Intermediate Deep	Happy Creek: north of Camera Pad: all three near old Clark study well
DS 3	Shallow	Happy Creek: south of Camera Pad
DS 4	Shallow Intermediate Deep	Southern Ridge Jerome Road- E of SR3 Apiary Site #67
DS 5	Shallow Intermediate	Ransom Road: E of SR3 Apiary Site #10
DS 6	Shallow	Schwartz Road north along firebreak

Table C-1. (cont.).

Mall donthe	Description (1)
vveii deptris	Description of location
 	
	West of SR3 road to PAPI Lights
	Apiary Site #47
Shallow	Schwartz Road West of SR3 South
	side road along ditch
Shallow	Roberts Road West of SR3 South side
Intermediate	The state troop of Still South Side
Deep	
Shallow	Jerome Road West of SR3 South side
Intermediate	The state was to be on the country side
Deep	
	West side of SLF near tower
1 - 1 - 1 - 1 - 1	West side of SET Tieal (Owe)
	West of SR3 North of SR406 near
O. I.a.iiovi	Dummit tower
	Dummit tower
Shallow	Obilet Tark # O 1 D 1
	Shiloh Turnbull Creek Drainage Road
	West of SR3
	Shiloh 1 Impoundment
	T-10K area near tower
	Playalinda - Mosquito Lagoon by 1st
Intermediate	Camera Pad
Shallow	Banana Creek West of SR3
Intermediate	
Deep	
Shallow	T-16 dike road
	Intermediate Deep Shallow Intermediate Deep Shallow Intermediate Shallow Intermediate Deep Shallow Intermediate Deep Shallow Intermediate Shallow Intermediate Shallow Intermediate Shallow Intermediate Deep

Table C-2. List of X and Y coordinates for groundwater well locations. Coordinates were collected with a Global Positioning System (GPS) in Florida State Plane NAD27. Wells are labeled according to the sub-aquifer they fall into, i.e. DS means that it was within the dune-swale sub-aquifer system. Well depth is represented by the final letter in the well id (D = deep, I = intermediate, S = shallow).

Well-id	X	Υ
DS-1D	565163.592	1618299.238
DS-1I	565161.961	1618306.34
DS-1S	565161.395	1618314.069
DS-2D	610770.013	1560857.645
DS-2I	610776.552	1560850.899
DS-2S	610784.743	1560845.373
DS-3S	610483.337	1560561.588
DS-4D	606674.751	1511926.727
DS-4I	606681.555	1511928.325
DS-4S	606689.565	1511927.143
DS-5I	607478.034	1517300.319
DS-5S	607472.622	1517301.149
DS-6S	618187.188	1538267.209
DUNE-1S	565597.22	1654529.305
DUNE-1I	565600.288	1654524.604
DUNE-1D	565603.645	1654518.148
DUNE-2D	617651.842	1571357.347
DUNE-2I	617647.958	1571362.708
DUNE-2S	617644.022	1571367.488
DUNE-3S	615039.142	1575440.361
DUNE-4S	635674.225	1545203.455
DUNE-5S	633642.311	1549318.58
DUNE-5I	633646.749	1549311.608
DUNE-6S	635731.302	1541647.686
DUNE-6I	635730.751	1541642.813
DUNE-6D	635729.264	1541634.675
MARSH-1D	548909.542	1632066.595
MARSH-1I	548915.961	1632071.242
MARSH-1S	548919.566	1632073.57
MARSH-2S	567796.577	1609694.81
MARSH-3I	573244.011	1578966.49
MARSH-3S	573239.764	1578963.367
MARSH-4I	615540.472	1574061.608
MARSH-4S	615544.648	1574058.185

Table C-2. (cont.).

Mall id	1	T
Well-id	X	<u>Y</u>
MARSH-5D	609001.289	1546419.353
MARSH-5I	608997.001	1546412.271
MARSH-5S	608991.484	1546405.833
MARSH-6S	587027.914	1559758.35
WEST-1S	591237.926	1571748.326
WEST-1I	591237.757	1571756.899
WEST-1D	591237.878	1571767.006
WEST-2S	596021.685	1537228.254
WEST-3D	604369.208	1529239.382
WEST-3I	604362.077	1529240.914
WEST-3S	604355.962	1529242.8
WEST-4D	598699.436	1509308.247
WEST-4I	598701.648	1509316.207
WEST-4S	598708.437	1509320.18
WEST-5I	594113.653	1559022.89
WEST-5S	594106.243	1559021.928
WEST-6S	588926.112	1585998.041

Table C-3. Minimum, maximum, mean, and standard deviation for each parameter analyzed for groundwater samples designated as Class G-II. Class G-II groundwater classification is defined as groundwater with total dissolved solids <10,000 mg/L. ND = non-detect.

			Class G-II		
			N=41		
Parameter	Units	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)					
4,4' -DDD	η⁄β/L	QN	ND	ON	QN
4,4' –DDE	/b//	QN	QN	QN	ND
4,4' -DDT	ηg/L	QN	QN	QN	QN
Aldrin	η⁄β/L	QN	ND	QN	QN
Alpha – BHC	µg/L	QN	QN	ON	QN
Beta - BHC	μg/L	QN	ND	QN	QN
Chlordane (alpha)	η⁄β/L	QN	ND	QN	QN
Chlordane (gamma)	μg/L	QN	QN	QN	QN
Chlordane (total)	μg/L	QN	ND	QN	QN
Delta – BHC	μg/L	QN	ND	QN	Q
Dieldrin	µg/L	QN	ND	ND	NO
Endosulfan I	µg/L	QN	ND	ND	QN
Endosulfan II (beta)	ng/L	QN	ND	ON	QN
Endosulfan Sulfate	µg/L	QN	ND	QN	Q
Endrin	1/6//	QN	ND	QN	ND
Endrin Aldehyde	ng∕L	QN	ND	QN	QN
Endrin Ketone	η ₀ η.	QN	ND	QN	ON
Gamma - BHC (Lindane)	ηg/L	QN	ND	QN	Q
Heptachlor	hg/L	QN	QN	QN	ND
Heptachlor Epoxide (a)	ηg/L	QN	QN	ON	2
Heptachlor Epoxide (b)	µg/L	Q	Q	2	Q
Isodrin	μg/L	ND	QN	QN	QN

Table C-3. (cont.).

Organochlorine Pesticides (8081) (cont.) Units Minimum M				Class G-II N=41		
Organochlorine Pesticides (8081) (cont.) μg/L ND ND ND chlor Aroclors (8082) μg/L ND ND ND 156/1242 Aroclors (8082) μg/L ND ND ND 223 μg/L ND ND ND ND 223 μg/L ND ND ND ND 228 μg/L ND ND ND ND 228 μg/L ND ND ND ND 228 μg/L ND ND ND ND 228 μg/L ND ND ND ND 258 Lichlorrophenoxy) propionic acid (2,4,5-TP) Silvex μg/L ND ND ND 260 Lichlorrophenoxy acetic acid (2,4,5-TP) Silvex μg/L ND ND ND 27 Lichlorrophenoxy acetic acid (2,4,5-TP) Silvex μg/L ND ND ND 28 Nn ND ND ND ND ND	Parameter	l Inite	Minimum	AA		
chlor reference (8082) 16/1242 221 224 2248 2249 2249 2249 2249 2249 2240	Organochlorine Pesticides (8081) (conf.)			Maximum	Mean	Std. Deviation
High ND	Methoxychlor					
He	Miro	#g/L	QN	Q	Q	CN
16/12/42	T	μg/L	QN	QN	GN	S
- TP) Silvex	loxaphene	ηση γ	Q	CN		
1961242)			2	QN
221 222 223 224 224 225 224 226 226 227 228 228 229 228 229 229 229 229 229 229	PCB - 1016/1242	na/L	CN	CN	2	
232 248 249 249 254 254 254 257 260 260 260 260 261 262 263 264 261 260 260 260 260 260 260 261 262 261 262 263 264 261 261 262 263 264 261 262 263 264 261 262 263 264 261 262 263 264 264 264 264 264 264 264 264 264 264	PCB - 1221	/0//	2 2			QN
248 ND ND ND 254 Lg/L ND ND ND 254 Lg/L ND ND ND 260 Lg/L ND ND ND 260 Chlortnated Herbicides (8151) Lg/L ND ND ND - Trichlorophenoxy) propionic acid (24,5 - T) Lg/L ND ND ND ND - Inichlorophenoxy acetic acid (24,5 - T) Lg/L ND ND ND ND ND Nichlorophenoxy acetic acid (2,4 - D) Lg/L ND	PCB - 1232	l'en		2	QN.	QN
254 NB ND	PCB - 1248	My/L	ON.	QN	QN	QN
Pug/L	PCB - 1254	Mg/L	Q	QN	QN	Q
Chlorinated Herbicides (8151) µg/L ND ND ND - Trichlorophenoxy acetic acid (2,4,5 - T) µg/L ND ND ND - Indrophenoxy acetic acid (2,4,5 - T) µg/L ND ND ND NIchlorophenoxy acetic acid (2,4 - D) µg/L ND ND ND Dichlorophenoxy acetic acid (2,4 - D) µg/L ND ND ND Dichlorophenoxy butyric acid (2,4 - DB) µg/L ND ND ND Dichlorophenoxy butyric acid (2,4 - DB) µg/L ND ND ND No µg/L ND ND ND ND No µg/L ND ND ND ND No µg/L ND ND ND ND No µg/L ND ND ND ND No µg/L ND ND ND ND No µg/L ND ND ND ND No µg/L ND ND N	PCB _ 1960	µg/L	ND	QN	QN	QN
Chilorinated Herbicides (8151) μg/L ND ND - Trichlorophenoxy) propionic acid (2,4,5 - Tr) μg/L ND ND ND Nidorophenoxy acetic acid (2,4,5 - Tr) μg/L ND ND ND BA μg/L ND ND ND Dichlorophenoxy acetic acid (2,4 - D) μg/L ND ND ND Dichlorophenoxy) butyric acid (2,4 - DB) μg/L ND ND ND Dichlorophenoxy) butyric acid (2,4 - DB) μg/L ND ND ND Dichlorophenoxy) butyric acid (2,4 - DB) μg/L ND ND ND Dichlorophenoxy) butyric acid (2,4 - DB) μg/L ND ND ND Dichlorophenoxy) butyric acid (2,4 - DB) μg/L ND ND ND Den μg/L ND ND ND ND Dichlorophenoxy) proponic acid] μg/L ND ND ND μg/L ND ND ND ND μg/L ND ND ND ND	l	µg/L	Q	QN	QN	S
- Trichlorophenoxy) propionic acid (2,4,5 - TP) Silvex μg/L ND ND ND ND ND ND ND ND ND ND ND ND ND	Chlorinated Herbicides (8151)					3
iclorophenoxy acetic acid (2,4,5 – T) Indrophenoxy acetic acid (2,4 – D) BA Indrophenoxy acetic acid (2,4 – D) BA Dichlorophenoxy acetic acid (2,4 – D) Indrophenoxy acetic acid (2,4 – D) Indrophenoxy acetic acid (2,4 – D) Indrophenoxy butyric acid (2,4 – D) Indrophenoxy butyric acid (2,4 – D) Indrophenoxy butyric acid (2,4 – D) Indrophenoxy butyric acid (2,4 – D) Indrophenoxy butyric acid (2,4 – D) Indrophenoxy acetic acid (2,4 – D) Indrophenoxy acid (2,4 – D) Indrophenoxy acid (2,4 – D) Indrophenoxy acid (2,4 – D) Indrophenoxy acid (2,4 – D) Indrophenoxy acid (2,4 – D) Indrophenoxy acid (2,4 – D) Indrophenoxy acid (2,4 – D) Indrophenoxy acid (2,4 – D) Indrophenoxy acid (2,4 – D) In	4,5	ua/L	S	CZ		
Dichlorophenoxy acetic acid (2,4 – D)		lo,	2 2	2 5	QN.	QN
BA Vg/L ND ND ND Dichlorophenoxy) butyric acid (2,4 – DB) µg/L ND ND ND shenol µg/L ND ND ND in µg/L ND ND ND sen µg/L ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND	2,4 - Dichlorophenoxy acetic acid (2.4 - D)	מאלין ליי	2	ON I	QN	Q
Dichlorophenoxy) butyric acid (2,4 – DB) µg/L ND ND ND shenol µg/L ND ND ND in µg/L ND ND ND in µg/L ND ND ND ben µg/L ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND	3.5 – DCBA	//g/L	QN	QN	QN	Q
Definition of the control of	4 - (2 4 - Dichlorophonomy britain and 10 4 - Par	#g/L	Q	QN	QN	QN
In the control of the contro	4 - Nitronhand	μg/L	QN	QN	QN	QN
ug/L ND ND ND Den µg/L ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND pp [2-(2,4-Dichlorophenoxy) proponic acid] µg/L ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND µg/L ND ND ND ND	Aciflinden	₩g/L	Q	ON	Q	QN
Den Mg/L ND ND ND μg/L ND ND ND ND μg/L ND ND ND ND ορ [2-(2,4-Dichlorophenoxy) proponic acid] μg/L ND ND ND μg/L ND ND ND ND μg/L ND ND ND μg/L ND ND ND μg/L ND ND ND μg/L ND ND ND	Bentazon	μg/L	Q	QN	QN	QV
μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND ND ND	Chloramben	μg/L	QN	ND	QN	Q
μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND μg/L ND ND ND ND ND	Dacthol	√g/L	QN	9	Q	Q
μg/L ND ND op [2-(2,4-Dichlorophenoxy) proponic acid] μg/L ND ND μg/L ND ND ND μg/L ND ND ND μg/L ND ND ND μg/L ND ND ND	Dalamon	µg/L	ON	QN	ND	QN
prop [2-(2,4-Dichlorophenoxy) proponic acid] b \triangle \triang	Dicambia	√g/L	QN	QN	ND	QN
b C-(-2,4-Dichlorophenoxy) proponic acid]	Dioplement (2 / 2 / P. L.)	μg/L	QN	QN ON	ND	CN
MD ND ND ND ND ND ND ND ND ND ND ND ND ND	up [2-(2,4-Dichlorophenoxy) proponic ac	μg/L	QN	QN	QN	QN
ON ON TOWN	MCBA	//g/r	QN	Q.	QN	QN
		Ng/L	9	S	2	

Table C-3. (cont.).

			Class G-II		
			N=41		
Parameter	Units	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)					
MCPP	ηg/L	QN	ND	QN	ND
Pentachlorophenol	Mg/L	QN	QN	QN	ON
Pictoram	μg/L	QN	QN	QN	QN
PAHs (8310)					
1 - Methylnapthalene	ηg/L	QN	ON	ON	QN
2 - Methylnapthalene	/⁄g/L	QN	QN	QN	QN
Acenaphthene	ng/L	QN	QN	QN	QN
Acenapththylene	ng∕L	QN	QN	ON	QN
Anthracene	µg/L	QN	QN	ON	QN
Benzo(a)anthracene	µg/L	0.03	0.16	0.035	0.021
Benzo(a)pyrene	/b//	0.025	0.14	0.0301	0.0197
Benzo(b)fluoranthene	η/δη	0.05	0.17	0.054	0.019
Benzo(g,h,i)perylene	η _σ η.	ON	QN	QN	2
Benzo(k)fluoranthene	ηg/L	0.025	0.1	0.0272	0.0117
Chrysene	η⁄gη/	0.025	0.17	0.0298	0.0231
Dibenz(a,h)anthracene	µg/L	QN	ON	Q	Q
Fluoranthene	√6n/	0.05	99.0	0.067	0.096
Fluorene	μg/L	ND	ND	Q	Q
Indeno(1,2,3-cd)pyrene	μg/L	0.025	0.09	0.0270	0.0102
Napthalene	hg/L	ND	ON	QN	2
Phenanthrene	η⁄δη/	ND	QN	QN	Q
Pyrene	ng/L	ND	QN	Q	Q
Metals					
Aluminum	mg/L	0.025	1.3	0.1841	0.3172
Antimony	mg/L	0.0025	0.005	0.00256	0.00039
Arsenic	mg/L	0.005	0.096	0.0125	0.0205

Table C-3. (cont.).

			Class G-II		
Daramotor			N=41		
raidiletei	Units	Minimum	Maximum	Mean	Cottoivior Pto
Metals (cont.)					ord. Deviation
Barium	l/om	0.05	30.0	100	
Beryllium	l/om	5000	0.00	0.05	0
Cadmium		0.0000	0.003	0.00056	0.00039
Calcium	mg/L	0.0005	0.00	0.00077	0.00134
Chromina	mg/L	0.3	670	168.77	164.43
Chlorida	mg/L	0.005	0.019	0.0058	0.0030
Shirt of the shirt	mg/L	5	4700	911.1	1330.7
Coball	mg/L	Q	Q	QN	S
Copper	mg/L	0.025	0.28	0.0312	00300
Iron	mg/L	0.025	2.7	0.8418	0.0330
Lead	mg/L	0.0025	0.028	0.000	0.0370
Magnesium	l/om	0.25	240	710001	0.00343
Manganese	/ou	2000	012	47.047	53.811
Mercury	IIIg/L	c00.0	0.26	0.0454	0.0620
Nickel	Ingr	ON S	QN	Q	QN
Potassium	mg/L	0.005	0.031	0.0059	0.0042
Selenium	mg/L	0.25	120	12.467	20.865
	mg/L	0.005	0.027	0.0058	0.0036
OIIVE	mg/L	0.005	0.005	0.005	C
Sodium	mg/L	4.1	3400	508 95	713 10
nallium	mg/L	0.001	0.003	0.0012	2000
Vanadium	mg/L	0.005	0.013	0.005	0.0000
Zinc	l/ma/	0.05	000	0.0030	0.0020
	1	3.5	0.23	6.05 4	0.028

Table C-3. (cont.).

			Class G-II		
			N=41		
Parameter	Units	Minimum	Maximum	Mean	Std. Deviation
Other Parameters					
Dissolved oxygen (N=36)	mg/L	0.04	4.33	1.65	1.20
Hydrogen ion (N=36)		8.91251E-09	3.89045E-4	1.24032E-05	6.49072E-05
Specific conductivity (N=36)	nmhos/cm	71	10010	2866.3	2934.7
Temperature (N=36)	Celcius	22.9	27.9	25.5	1.4
Total dissolved solids	mg/L	10	8400	2019.1	2201.8
Total organic carbon	mg/L	9.5	95	19.45	25.98

Table C-4. Minimum, maximum, mean, and standard deviations for each parameter analyzed for groundwater samples designated as Class G-III. Class G-III groundwater classification is defined as groundwater with total dissolved solids >10,000. ND = non-detect

			Cla	Class G-III	
Parameter			2	N=16	
	Units	Minimum	Maximum	Mean	O Pro
Organochlorine Pesticides (8081)					ord. Deviation
4,4 -UUU	1/0/1	2			
4,4' -DDE	7 %	2	QN	Q	2
4.4' -DDT	µg/L	QN	QN	2	QN
Aldrin	ng/L	Q	QN	Q	CZ
Alpha - BHC	ηg/L	QN	Q	QN	2 2
Beta - BHC	µg/L	QN	QV	QN	S
Chlordane (alnha)	1/6/L	QN	QN	QN	QN
Chlordane (gamma)	µg/L	Q	QN	QN	NO.
Chlordane (total)	µg/L	Q	ON	QN	QN
Delta - BHC	µg/L	Q	QN	QN	QN
Dieldrin	/JØ/L	Q	QN	QN	QN
Endosulfan I	ng/L	QN	QN	QN	QN
Endosulfan II (heta)	ng/∟	QN	QN	QN	QN
Endosulfan Sulfate	mg/L	NO	QN	QN	QN
Endrin	#g/L	Q	QN	QN	QN
Endrin Aldehyde	#g/L	QN	QN	QN	QN
Endrin Ketone	µg/L	QN	QN	QV	QN.
Gamma - BHC (Lindane)	#g/L	Q	ND	QN	N
Heptachlor	//d/	Q :	QQ	QN	QN
Heptachior Epoxide (a)	#g/L	QN	ND	QN	QN
Heptachlor Enoxide (h)	µg/L	2	QN	QN	QN
Isodrin	₩g/L	2	QN	QN	QN
Methoxychior	#g/L	Q	QN	QV	QN
Mirex	#g/L	Q	Q	QN	Q
Toxaphene	#g/L	Q	QN	QN	2
	Mg/L	QN	QN	QN	2

Table C-4. (cont.).

			Clas	Class G-III N=16	
Parameter	Units	Minimum	Maximum	Mean	Std. Deviation
Aroclors (8082)					
PCB - 1016/1242	ng/L	QN	QN	ND	ON
PCB - 1221	ng∕L	QN	QN	QN	QN
PCB - 1232	ηg/L	QN	QN	QN	QN
PCB - 1248	ng∕L	QN	QN	QN	QN
PCB - 1254	µg∕L	QN	QN	QN	QN
PCB - 1260	μg/L	QN	QN	QN	QN
Chlorinated Herbicides (8151)					
2-(2,4,5 - Trichlorophenoxy) propionic acid (2,4,5 - TP) Silvex	7∕6⁄/	QN	an	ND	ND
2,4,5 - Triclorophenoxy acetic acid (2,4,5 - T)	ng∕L	QN	QN	QN	ON
2,4 - Dichlorophenoxy acetic acid (2,4 - D)	µg/L	QN	an	QN	QN
3,5 - DCBA	µg/L	ND	QN	QN	QN
4 - (2,4 - Dichlorophenoxy) butyric acid (2,4 - DB)	ηd/L	QN	QN	QN	QN
4 - Nitrophenol	ng/L	QN	QN	QN	QN
Acifluorfen	/bg//	ON	ND	QN	QN
Bentazon	μg/L	QN	ND	ND	QN
Chloramben	µg/L	QN	ND	QN	QN
Dacthal	μg/L	QN	QN	Q	Q
Dalapon	η _φ /Γ	QN	ND	QN	ND
Dicambia	7/6 <i>n</i>	ON	ON	QN	ON
Dichlorprop [2-(2,4-Dichlorophenoxy) proponic acid]	Hg/L	QN	QN	an	QN
Dinoseb	ng/L	QN	QN	QN	QN
MCPA	ng/L	QN	QN	QN	ND
MCPP	ηg/L	QN	ON	QN	QN
Pentachlorophenol	√6/r	QN	ND	QN	QN
Picloram	ηδ/L	QN	QN	Q	QN

Table C-4. (cont.).

			Clas	Class G-III	
Parameter	Units	Minimum	Maximum	Mean	Oto Dowingtion
PAHs (8310)					Old. Deviation
1 - Methylnapthalene	no/L	CN	QN	CIA	24
2 - Methylnapthalene	Von	S	S		ON.
Acenaphthene	1 VO	2 2		QN S	ON.
Acenapththylene	Mg/L	2 5	ON S	QN	Q
Anthracene	µg/L	Q !	QN	QN	ND
Ronan/anthropos	µg/L	QN	S	Q	QN
Denizo(a) annuacente	µg/L	0.03	0.13	0.036	0.025
Denizo(a)pyrene	μĝγ	0.025	90.0	0.0272	0.0088
benzo(b)filuorantnene	ng∕L	0.05	0.05	0.05	0
Benzo(g,n,I)perylene	/\g/r	QN	Q	QN	QN
Benzo(k)fluoranthene	√l/6/r	0.025	0.07	0.0294	0.0125
Cnrysene	η⁄β <i>ι</i> /	0.025	0.18	0.0347	0.0388
Dibenz(a,h)anthracene	μg/L	2	QN	ND	CN
Fluoranthene	µg/L	0.05	0.05	0.05	0
Fiuorene	ηgηΓ	Q	QN	S	CN
Indeno(1,2,3-cd)pyrene	η _β /Γ	0.025	0.1	0.0297	0.0188
Napthalene	µg/L	QN	QN	CN	00 ON
Phenanthrene	µg/L	2	QN	CN	
Pyrene	ηση.	Q	GN	S	
Metals					2
Aluminum	mg/L	0.025	0.28	0.1053	0.0686
Antimony	mg/L	0.0025	0.012	0.00566	0.00364
Arsenic	mg/L	0.005	0.028	0.0097	0.0076
Barium	mg/L	0.05	0.41	0.082	5000
Beryllium	mg/L	0.0005	0.0005	0.0005	800
Cadmium	mg/L	0.0005	0.0005	0 0005	
Calcium	mg/L	170	760	431.3	161.0
				2	00

Table C-4. (cont.).

Metals (cont.) Units M Metals (cont.) mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (14) mg/L mg/L (14) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L (15) mg/L mg/L </th <th></th> <th></th> <th></th> <th></th>				
Metals (cont.) mg/L Jae mg/L f mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L ium mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L		Maximum	Mean	Std. Deviation
Jean mg/L Jean mg/L In mg/L mg/L In mg/L mg/L In mg/L mg/L In mg/L mg/L In mg/L mg/L In mg/L mg/L In mg/L mg/L In mg/L mg/L In mg/L mg/L In mg/L mg/L Je or oxygen (N=15) mg/L Jen ion (N=15) mg/L Jec conductivity (N=15) mg/L Jec conductivity (N=15) umhos/cm Jec conductivity (N=15) colcius Jec conductivity (N=15) colcius				
sium mg/L r mg/L r mg/L mg/L mg/L rium mg/L n mg/L n mg/L n mg/L ium mg/L ium mg/L ium mg/L ium mg/L jem ion (N=15) mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L jec conductivity (N=15) umhos/cm srature (N=15) umhos/cm srature (N=15) colcius		0.005	0.005	0
r. mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L		31000	13856.3	8010.9
rt mg/L mg/L mg/L mg/L mg/L mg/L mg/L mg/L		QN	Q	Q
mg/L ssium mg/L ry mg/L ry mg/L sium mg/L ium mg/L in mg/L in mg/L in mg/L ium other Parameters mg/L yed oxygen (N=15) mg/L gen ion (N=15) mg/L gen ion (N=15) umhos/cm gratture (N=15) umhos/cm gratture (N=15) coelcius		0.1	0.0295	0.0195
nsium mg/L Inese mg/L ry mg/L ry mg/L n mg/L n mg/L ium mg/L ium mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L ic conductivity (N=15) mg/L ic conductivity (N=15) unthos/cm stature (N=15) colcius		10	1.9978	2.9587
sium mg/L ry mg/L ry mg/L sium mg/L um mg/L n mg/L ium mg/L ium mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L ic conductivity (N=15) mg/L ic conductivity (N=15) umhos/cm srature (N=15) celcius		0.013	0.00456	0.00320
ry mg/L ry mg/L sium mg/L um mg/L n mg/L ium mg/L ium mg/L ium mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L ic conductivity (N=15) umhos/cm stature (N=15) umhos/cm stature (N=15) colcius		1900	973.1	495.5
ry mg/L sium mg/L um mg/L n mg/L ium mg/L ium mg/L ium mg/L gen ion (N=15) mg/L gen ion (N=15) mg/L ic conductivity (N=15) unhos/cm srature (N=15) celcius		0.46	0.1299	0.1422
ium um um um mg/L mg/L mg/L ium Other Parameters ved oxygen (N=15) ic conductivity (N=15) ic conductivity (N=15) ium Other Parameters mg/L mg/L mg/L ium Other Parameters mg/L ic conductivity (N=15)		QN	QN	QN
sium mg/L um mg/L m mg/L m mg/L lium mg/L slium mg/L ved oxygen (N=15) mg/L gen ion (N=15) mg/L ic conductivity (N=15) umhos/cm erature (N=15) celcius		0.01	0.0056	0.0017
um mg/L m mg/L lium mg/L lium mg/L lium mg/L ved oxygen (N=15) mg/L gen ion (N=15) mg/L ic conductivity (N=15) umhos/cm erature (N=15) celcius		540	285.9	161.6
mg/L um mg/L lium mg/L Mg/L Other Parameters mg/L ved oxygen (N=15) mg/L gen ion (N=15) mg/L ic conductivity (N=15) umhos/cm erature (N=15) celcius		0.05	0.0081	0.0112
um mg/L ium mg/L idium mg/L Other Parameters mg/L olved oxygen (N=15) mg/L ogen ion (N=15) mg/L iffic conductivity (N=15) umhos/cm perature (N=15) Celcius		0.01	0.0053	0.0013
ium mg/L Idium mg/L Other Parameters mg/L olved oxygen (N=15) mg/L ogen ion (N=15) umhos/cm perature (N=15) celcius		15000	8206.3	3638.4
Idium mg/L Other Parameters mg/L olved oxygen (N=15) mg/L ogen ion (N=15) mg/L iffic conductivity (N=15) umhos/cm perature (N=15) Celcius		0.002	0.0011	0.0003
Other Parameters mg/L olved oxygen (N=15) mg/L ogen ion (N=15) 1 iffic conductivity (N=15) umhos/cm perature (N=15) Celcius		0.005	0.005	0
er Parameters mg/L 1		0.05	0.05	0
mg/L 1 1 umhos/cm Celcius				
) umhos/cm Celcius		6.08	2.22	1.88
umhos/cm Celcius	1.47911E-08	7.24436E-07	1.41E-07	1.76E-07
Celcius		44700	27162	12253
		27.5	26.3	1.0
s mg/L	mg/L 11000	44000	23562.5	10276.0
Total organic carbon 0.5		55	17.47	15.64

	_	DS-1S	DS-11	08-10	DS-2S	DS-21	08:20	200	9	3, 33	
		09/27/98	09/27/98	09/27/98	10/11/98	10/11/98	10/11/98	10/11/98	2 86	10/10/98	10/10/00
	Chris								duplicate		8
Organochiorine Pesticides (8061)											
4,4 - 000	ZQ.	<0.05	<0.05	90:0>	90:0>	\$0.0°	<0.05	\$0.05	90.02	800	300
4,4 · UUE	W	<0.05	<0.05	\$0.0×	\$ 80 80 80	\$0.0°	40 05	5	36	3 5	9
4,4°- DOT	W	<0.05	\$0:0>	\$0.0¢	\$0.0×	900	\$0.05	866	3 5	3 5	8 8
Adrin	M	<0.05	<0.05	\$0.05	\$0.0×	\$0.05	,005 S	200	3 5	3 5	3
April 64.0	W.	<0.05	<0.05	<0.05	\$0.0°	¢0.08	8	900	38	8 8	200
Chicago and Chicago	707	60.05 85	<0.05	<0.05	<0.05	800	¢0.05	8,0	\$0.05 80.05	5	3 8
Chouse (epite)	700	900	ده 90%	٥.05 د0.05	<0.05	<0.05	\$0.0°	\$0.0\$	<0.05	50.05	3 6
Chocker (see)	NA.	9000	د0 05 60 05	<0.05	<0.05	<0.05	\$0.0\$	\$0.0¢	\$0.05 0.05	90 00	36
Crixouarie (total)	100	ī	Ų	·	<1	⊽	Ţ	⊽	v		3
	No.	<0.05	<0.05	<0.05	90:0>	800	868	40 05 40 05	5	30.00	30,0
Dietorn	707	<0.05	<0.05	<0.05	\$0.0°	800	800	50.05	300	3 5	3 5
Endosuran	101	<0.05	<0.05	90:0>	\$0.05	8	900	50,00	300	3 6	300
Endosullan II (Deta)	70/	<0.05	90.0>	90.0×	\$0.0°	998	\$0.09 80.09	50.05	800	3 5	9 8
Endosulan Sulfate	78/r	<0.09	60:0>	80 O>	60.05	80.00	80.00	800	3 5	8 8	98
Endrin	1/0//	<0.05	<0.05	\$0.0¢	6 8	\$0.09 90.09	800	200	3 5	800	8 8
Endrin Aldehyde	MOL	<0.05	\$0.0×	\$0.0\$	8	898	50.05	8 6	3 5	9.5	88
Endrin Ketone	NO/L	<0.05	90.0>	88	\$0.0\$	99	8 8	3 6	90.00	98	40.05 60.05
Gamma - BHC (Undane)	MOL	<0.05	\$0.0×	\$0.00	800	\$0.05	38	3 6	8 6	8 8	8 5
Heptachlor	NOV	<0.05	90.0×	\$0.0¢	800	\$0.0V	900	8	3 5	8 8	8
Heplachlor Epoxide(a)	7/6//	<0.05	90:0>	\$0.0¢	80%	60.05 0.05	50.00	3 5	3 6	9 9	8 8
Heptachlor Epoxide(b)	W	<0.05	90.0>	90.05	800	90.09	9000	8	3 5	9 9	8 8
l sodnin	NOV	<0.05	<0.05	<0.05	\$0.0¢	90.0	900	800	3 6	8 6	9 9
Methorychior	ZQ.	<0.05	<0.05	<0.05	\$0.0°	\$0.0¢	800	900	38	3 5	8 8
MEX	Į,	8	¢0.05	<0.05	<0.05	<0.05	<0.05	\$0.08	\$0.05	8 6	36
District Annual Control of the Contr	Š	2	\$	7	د2	62	<2	<2	8	2	8
OCB 1018/12/2											
26210101207	Į,	۶	v	⊽	۲,	<1	1>	₽	ī	ī	7
1221-021	100	v	v	v	<1	<1	⊽	⊽	V	1	,
PUB-12/2	100	v	V	۲	<1	ţ	₹	⊽	v	1	T
1909 1907	707	v	V	v	۲,	<1	۲	v	⊽	ē	
PCB-1234	, A	v	v	v	v	5	V	ţ>	⊽	⊽	Į
Change of Market Assets		Ÿ	V	v	Ÿ	۲,	•	~ 1	۲>	⊽	₽
2 2 4 E Table Charles of the Colors (8131)											
2.4 E. Trichtomothernous speaks society 2.4 E. T.	100	\$00	\$0.5	92	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	505
2.4.Dichlomorphanom scale scid (2.4.0)		\$02	\$02	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<05	605
3.5-DCBA		000	000	000	<0.5	<05	<0.5	<0.5	<0.5	<0.5	<0.5
4-(2,4-Dichloraphenoxy)butyric acid (2.4 - DB)	1 2 2	2	200	000	800	40.5	<0.5	<0.5	<0.5	<0.5	<0.5
4 - Nitrophenol	100	200	000	000	000	\$00	<0.5	405	<0.5	<0.5	<0.5
Acifluorien	100	\$05	505	200	600	603	000	\$02	<0.5	\$0.5	<0.5
Bentazon	ğ	40.5	88	200	3,4	9	000	900	\$65	<0.5	605
Chloramben	ķ	<0.5	40.05 60.05	605	200	3 4	500	603	\$ 60	\$02	<0.5
Dacthal	Non	<0.5	\$0.5	\$65	89	200	26.	96	600	000	¢0.5
Dalapon	1/0//	<0.5	<0.5	<0.5	<0.5	605	505	200	500	600	000
Dicamba	W	<0.5	<0.5	<0.5	\$65	605	505	ę	200	600	600
Dichloroprop (2-(2,4-Dichlorophenoxy)proponic acid)	701	<0.5	<0.5	40.5	605	6 05	202	500		COV	e e
Dinoseb	VOJ	<0.5	<0.5	<0.5	<0.5	40.5	905	505	200	600	000
WCPA	No.	¢10	<10	<10	<10	ما 0	¢10	65	900	210	
WCPP Designations	Vo.	ę	¢2	٠ 1 0	¢10	<10	<10	oto	eş.	01.0	910
Pentachiorophenoi	100	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	\$0.5	202
	1001	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	6 05	\$65

I * value is between the Method Detection Limit and the Practical Quantitation Level " = diuted sample, - = not analyzed

Colored Colo			05.15	1.50	- ct.8c	26.50	6.50	5.7		2	2700	
Marie Mari			09/27/98	09/27/98	09/27/98	10/11/98	10/11/98	10/11/98	<u> </u>	3 60	10/10/98	10/10/98
Participa Part		Chrits										
1974 4.05	PAH (#310)											
March Marc	1											
Marie Mari	- Methylnaphthalene	7/0/	Ç0.9	000	\$0.5	\$05	¢0.5	Ç02	\$0.5	\$0.5	\$00	¢0.6
Mar. 405	 Methylnaphthalene 	MO/L	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.6	۷0°6
Mark	cenaphthene	MOL	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	9.0>	9'0>
Marie Mari	cenaphithylene	NOV	-Q-	40.1	601	-0>	-0.1	<0.1	<0.1	40.1	ê.	60
March Marc	nthracene	701	40.5	<0.5	<0.5	\$0.5	<0.5	<0.5	<0.5	60.5	95.0>	¢0.6
Martin M	enzo(a)anthracene	70%	800	8	86	\$0.00	9000	\$0.0¢	90.09	0.07	90.09	9000
Marie Mari	enzo(a)ovrene	Vov	60.05 60.05	900	8,68	\$0.05 50.05	\$0.0°	\$0.0 \$	900	80.0	9000	5
Mathematical Control of the contro	on 200 hill increasing an	Į,	ę	Ę	ę	ç	Ę	5	Ę	Ę		5
Part Color	occeptate incompany											2 9
Marcella	enzo(g,n,gperyrene	700	9	8	9	S	\$	Ŝ	9	Ç0.1	G.1	<0.12
Marchen Myl 40 to 40 t	enzo(k)fluoranthene	707	900	90.0	<0.05	\$0.88	90.00	¢0.09	6 8	\$ 8	<0.06	<0.06
Marchelle May 401	hrysene	J/0/J	60.05	<0.05	40.05	<0.05	<0.05	<0.05	<0.05	90.0	<0.06	90.0>
Surrophies 1971 4011 4	ibenzo(a,h)anthracene	NO/	<0.1	<0.1	<0.1	<0.1	<0.1	1.0>	40.1	60.1	40.11	<0.12
Sumplified Applied A	uoranthene	764	6	601	60.1	6	\$0.1	¢0.1	601	0.12	11.0>	<0.12
Machine Mol. close clo	LOCATA	Von	601	0.0	60.1	9	9	401	é	0.131	100	20.12
Marchelea	denot 1 2 andhowene	Į,	Ş	چ	٤	5005	5	200	300	500	800	900
Mathematical Part Color	artifications	100	205	902	20.5	505	202	20.5	20.5	200	800	400
Marcopules	acastheses		200	20.0%	300	200	2000	300	200		2	2
Surrogates			3 5	3 5	3 4	3 5	3 5	3 5	3 5	2 5	8 8	8 8
Marriage Surrogates St. 100 126 120 120 120 110 120 110 120 110 120 110 120 110 120 110 120 110 120 110 120 110 120 110 120 110 120			80.00	20.00	30.00	30.00	97.00	\$0.05	30.05	6.131	80.05	8
Matchine % 86												
No. 100 100	+DCAA	*	છ	8	126	86	72	110	82	99	70	72
Name Name 68 82 85 85 86 86 105 107 Name Machin Name 107 75 60	4,5,6-TCMX	¥	æ	75	8	120	76	Ş	72	72	88	73
Name Marches Name 107 75 67 103 62 97 82 67 103 73 Name Marches mg/L -0.05 </td <td>90</td> <td>æ</td> <td>88</td> <td>83</td> <td>ಜ</td> <td>8</td> <td>8</td> <td>8</td> <td>88</td> <td>5</td> <td>101</td> <td>73</td>	90	æ	88	83	ಜ	8	8	8	88	5	101	73
Meeting Mineting Meeting a coope coope	terphenyl	%	107	75	67	103	82	87	82	99	74	78
March Marc	Metals											
Part Part	uminum	mg/L	¢0.08	0.058	¢0.02	0.090	60.0 5	0.096	0.061	<0.05	<0.05	0.32
mark 4.001	ubmony	mg/L	<0.005	<0.005	0.011	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005
Marchelle May Color Co	Beri c	mg/L	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	10.0>
Jumin (Asia) Asian (Asia) 4001<	White was a second seco	m0/L	<0.1	<0.1	-0.1	<0.1	c0.1	1.0>	<0.1	40.1	40.1	60.1
Wight 4001 40001	erythium.	₩/J	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	00.00
mg/L 72 89 580 96 120 64 49 49 120 nim (total) mg/L <001	admium	MQ/L	¢0:001	40.001	<0.001	<0.001	<0.001	<0.001	¢0.001	60.00	\$0.00°	¢0.001
Nimit (1008) Might 4,010 4,011 4,012	mode.	-NOW	22	2	260	8	120	23	3	3	120	1.4
1	hromium (total)	mg/L	¢0.01	40.01	¢0.01	40.0 ¹	<0.01	<0.01	¢0.01	\$0.0\$	<0.01	¢0.01
H Mg/L cd 06 cd 07 cd 06 cd 07 cd 0	tedo	AQF.	\$0.0°	\$0.05	40.05	<0.05	<0.05	\$0.0°	\$0.0¢	\$0.05 80.05	80.05 80.05	\$ 80.0%
mg/L 0.16 1.9 2.9 0.062 0.7 0.055 0.12 0.11 1.7 mg/L 0.005 0.005 0.005 0.005 0.005 0.005 0.005 0.005 0.005 mg/L 0.002 0.002 0.002 0.002 0.002 0.002 0.002 0.002 mg/L 0.002 0.002 0.002 0.002 0.002 0.002 0.002 0.002 mg/L 0.001 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.002 0.002 0.002 0.002 0.002 0.002 0.002 0.002 0.002 mg/L 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.002 0.002 0.002 0.002 0.002 0.002 0.002 0.002 0.002 mg/L 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 0.01 mg/L 0.01	0000	Z	\$0.08	\$0.0¢	\$0.0¢	90.00	\$0.00	\$0.0°	\$0.00	800	800	800
Maje Maje	00	AQE.	9.70	6.5	2.9	0.092	0.7	6 8	0.12	0.11	1,7	-
mg/L 3.1 2.7 730 2.5 6.2 2.1 <0.5 <0.5 <0.77 nmg/L <0.01 <0.01 0.054 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01	pee	A _O m	<0.005	\$00.0	¢0.005	<0.005	<0.005	<0.005	\$000	¢0005	<0.005	\$00.00 \$
Process Age 4001 4004 4001 4004 4001 <	agnesium	mo/L	3.4	2.7	730	2.5	6.2	2.1	40.5	40.5	7.7	0.0
ry colored col	anganese	Age.	\$0.0	\$0.0	0.15	¢0.01	280	\$0.0¥	\$0.0	600	6.03	¢0.01
sign c001 <th< td=""><td>ercury</td><td>ZoE</td><td><0.0002</td><td><0.0002</td><td><0.0002</td><td><0.0002</td><td><0.0002</td><td><0.0002</td><td><0.0002</td><td><0.0002</td><td>2000°</td><td>¢0 0002</td></th<>	ercury	ZoE	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	2000°	¢0 0002
win mg/L 1.8 <0.5 6.0 6.0 <0.5 1.3 1.4 wm mg/L 0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01	Ckel	Æ	¢0.01	<0.01	40.01	\$0.0	\$0.0	\$0.0	60.0	40.05	600	¢0.01
um mg/L 6.01 I <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01	otassium	Jour I	97	40.5	z	40.5	0.7	<0.5	6.1	1.3	7	<0.5
mg/L <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01	elenium	Age.	0.01	600	40.01	\$ 0.00	40.0	¢0.01	¢001	<0.01	\$0.00	40 O
m rg/L 12 22 8600 8.2 44 10 4.1 4.1 87 m	iver	70	40.0v	60.0	<0.01	40 O	×0.01	<0.01	40.0	<0.0	900	1000
n mg/L <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0.002 <0		٤	2	8	5600	E.2	7	10	-	7	4	8.2
Killum mg/L <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01	halism	Ē	20000>	2000>	2000	200.00	2000	0.0021	2000>	2000	2000	2000
mod 201 201 201 201 201 201 201 201 201 201		2	100	100	200	100	Ę	100	5	500	100	
		2	6	Ę	5	5	100	Ę	ę	6	۶	

		08-40	05.50	300							
		10/10/98	10/17/98	10/17/98	10/1/09	DUNE-1S	DUNE-11	DUNE-1D	DUNE-2S	DUNE:2	00 1816
Organochlorine Pesticides (8061)						08.00.80	09/18/98	09/15/96	09/21/98	09/21/98	11/27/04
4,4' - DOD											
4,4 · DOE	Š .	98	\$0.0g	<0.05	\$0.0×	200					
4.4' - DOT		ê 8	-0.05 -0.05	80.6	8	3 2	8	89	<0.05	90°0≥	Š
Aldrin	Į.	9	808	6.05	80.05	3 8	8 8	\$0.0g	\$0.0 8	88	3 6
Apha - BHC	1	8 8	8	<0.05	\$0.05	88	8 6	88	×0.08	\$0.0×	90.0
Beta - 84C	١	3 6	\$0.00	6 8	<0.05	88	38	9 5	60.08	\$0.0¢	60°8
Chlordane (alpha)		8 8	88	& 8	8 8	88	3 8	9	\$0.88	\$0.0°	8
Chlordane (gamma)		9 8	6 8	\$0.0¢	88	5	8 8	98	60.05	6.08	900
Chlordane (total)	To S	800	90.0v	<0.05	800	3 6	8 8	998	<0.05	6.08 80.08	8 6
Delta - BHC		⊽ .	v	دا	v	1	9,	900	<0.06	6. 86.	8
Dielchin	100	8	\$ 8	\$0.0°	800	8	7	v	₹	⊽	7
Endosultan I	100	8 8	8	÷0.05	\$0.0	8	8 8	8	\$0.05	\$0.0×	800
Endoeultan II (beta)	1	8 8	×0.05	<0.05	900	900	3 5	88	\$0.05	<0.05	8
Endosultan Sultate	1	3 8	900	<0.05	6 8	39.65	3 8	8	<0.05	<0.05	8
Endrin		3 2	9000	¢0.05	80.0×	88	3 8	89.69	¢0.08	<0.05	88
Endrin Aldehyde		9 8	8	<0.06	\$0.0°	9000	3 5	8 8	0000	<0.09	60.09
Endrin Ketone		8 6	6 8	6.08	-0.05	88	3 2	8	88	\$0.05	88
Gamma - BHC (Lindane)		8 6	\$ 8	<0.05	\$0.00	8	9 8	8	¢0.05	<0.05	999
Heptachlor	1	900	\$0.0g	<0.05	800	8	8 8	8	89.09	<0.05	800
Heptachlor Epoxide(a)	3	\$0.00	8	<0.05	800	8 8	9 5	6.8	<0.05	\$0.0g	5
Heptachior Epoxide(b)	2	89	<0.05	<0.05	89	3 8	9 2	8	\$0.05	\$0.0×	5
laodrin	1	80.0	88	<0.05	88	88	9 9	8	<0.06	\$0.0°	9000
Methoxychior		800	\$9. 98.	<0.05	89.65	Š	8 8	8	¢0.85	9.08	8000
Wirex		8	88	\$0.08	8	5	8 8	900	60 80	\$0.0°	8
Toxaphene		90.00	900	6.05 50.05	90.09	800	8 8	88	\$0.0g	\$0.05 80.05	868
Areclors (8062)		y	2	42	8	2	300	88	999	\$0.0°	800
1242	1						1	3	2	7	5
PC8-1221		•	ī	<1	₽	Ī	ļ				
PCB-1232		Į,	V	Þ	V	-			Þ	Þ	⊽
PCB-1246	1	•	·	۲	ī	Ī	1	Ţ	⊽	ī	⊽
PCB-1254			7	v	v	-	†	V	₹	۱۷	⊽
			V	٦	ī	ī			V	Þ	⊽
Chlorinated Herbicides (8151)		7	v	5	V	V	+	- -	⊽	⊽	⊽
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Savex)								1	ī	V	₽
2.4.5-Trichtorophenoxy acetic acid (2,4,5 - T)		99	\$0.5	425	<0.5	90.5	į				
2,4-Dichlorophenoxy acetic acid (2,4 - D)	ğ	200	50.0	425	<0.5	<0.5	40.5	6 6	992	<0.5	ē
J.P. C. S. C. C. C. C. C. C. C. C. C. C. C. C. C.	ž	992		969	\$0.5	<0.5	<0.5	205	600	800	⊽
14.1-Undicophenoxy)bulyinc acid (2,4 - DB)	Ž	<0.5	200	5	999	<0.5	9.5	\$65	2 4	9	V
Adherien	MA	<0.5	\$0.5		900	\$0.5	<0.5	\$0.5	\$65	8	1
Bentazon	ž	<0.5	<0.5	255	200	5	\$0.5	<0.5	<0.5	202	1
Chloramben	Z,	<0.5	<0.5	425			\$0.5	<0.5	\$05	505	
Dacha	ğ	60.5	<0.5	42.5	1	9	\$0.5	<0.5	\$0.5	505	
Datapon	ZQ.	<0.5	<0.5	42.5	100	9	902	40.5	<0.5	505	
Dicamba	Z.	60.5	<0.5	42.5	\$0.5	9 4	\$ 65	\$ 65	<0.5	<0.5	
Dichloroprop (2-(2,4-Dichlorophenoxylomomic acid)	Š	\$0.5	<0.5	<2.5	\$0.5	2 4	9	393	<0.5	\$00	T
Oinoseb		9.5	<0.5	<2.5	8	1	9	905	<0.5	<0.5	V
WCPA	į į	£ 5	\$0.5	425	<0.5	98	999	500	205	<0.5	v
WCPP	, co			ŝ	<10	95	95	000	605	<0.5	V
Pentachlorophenol	709	2	010	8	<10	410	95		2 9	ş	SS SS
COORT	701	505	0,6	623	<0.5	<0.5	\$05	20.5	200	9	\$
			,	6.3	<0.5	<0.5	<0.5	¢0.5		\$0.5	7

Limitodia properties Attition		_	4,00	70.50	D.C. 51	59.50	DUNE-1S	DONE:1	DUNE-10	DONE-2S	DUNE-21	DUNE-20
Marie 1948		<u> </u>	10/10/98	10/17/98	10/17/98	10/11/98	09/18/98	09/18/98	09/15/98	09/21/98	09/21/86	11/27/98
Partition Part		SP										
Marine M	(\$100) THE											
Mathematical Color	ı		ļ	3 0	49	,0 £	30%	505	<0.5	<0.5	<0.5	<0.5
Marie April Apri	1 - Methylnaphthalene	Š.	900	500	96	200	202	505	60.5	605	<0.5	<0.5
March Marc	2 - Methylnaphthalene	Į,	90.0	500	36	500	-0.5	6 05	<0.5	¢05	¢0.5	<0.5
Marie Mari	Acenaphthene	Š.	9	3	5	5	-0-1	9	40.1	601	50.	٧.
Marie Mari	Acenachthylene	TAGE!	99	Š	500	505	<0.5	\$ 65	40.5	<0.5	<0.5	<0.05
Marie April Continue Cont	Anthracene	1	8 8	200	300	900	40.05 40.05	900	<0.05	\$0.05	<0.05	<0.05
Marie April Apri	Benzo(a)anthracene		8 8	3 5	900	\$0.05	900	\$0.05	\$0.0¢	<0.05	\$0.0¢	<0.05
Marie Mari	Benzo(a)pyrene		8 4	3	5	ę	-0	<0.1	-01	-ç0 -	¢0.1	-0
March Marc	Benzo(b)fluoranthene	100	20.16	- 0	2	Ę	ç	50.1	40.1	40.1	6.	60.1
Marie Mari	Benzo(g,h,i)perylene	100	<0.12	100	20.1	200	2	50.0	90.00	1 30 0	3000	50.05
Marchine 1971 4006 400	Benzo(k)fluoranthene	ZQI	٠0.06	8	8	80.0	300	3 4	3 6	200	300	500
Marie Mari	Chrysene	101	¢0.08	\$ 8	60.0 5	80.00	800	90.00	20.00	300	3,5	3 5
Marie Marie April Apri	Diberzo(a hanthracene	7/0//	<0.12	<0.1	61	60	¢0.1	203	Ş	100	0	
Marchelle	Firecoothene	Z	<0.12	<0.1	\$0.1	601	-0 -	ç0.	6	60.7	20,	0
1971 1972 1975	7. Charles	Ę	<0.12	\$	40.1	0	6	60.1	¢0.1	¢0.1	40.1	Ç0)
Marche March Marche March	ruorene	۶	8000	\$0.05	\$00°	\$0.05	<0.05	<0.05	<0.05	<0.05	6.85	<0.05
Surrogues 400 -0.05 <	indend 1,2,3-culpyrene		800	50,	505	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Marie Mari	Naphthalene	1	200	500	6	\$0.0×	\$0.0°	60.08	\$0.05	90.0>	\$0.0×	<0.05
Surrogies X SE 67 94 82 109 109 109 109 109 109 109 109 109 109 109 109 109 109 109 109 109 109 100 109 100 109 100 109 100 <th>Phenanthrene</th> <th>1</th> <th>3 6</th> <th>800</th> <th>6</th> <th>9000</th> <th>\$0.0°</th> <th>900</th> <th>\$0.0\$</th> <th>40.05</th> <th>90:0></th> <th><0.05</th>	Phenanthrene	1	3 6	800	6	9000	\$0.0°	900	\$0.0\$	40.05	90:0>	<0.05
Municipalities % 92 96 62 94 62 100 40 108 Municipalities % 78 67 96 77 76 67 58 75 100 40 108 Municipality % 78 66 107 66 66 77 100 66 77 100 Municipality mg/L -0.05 0.65 1.3 0.082 0.07 6.07 6.07 6.05			97.00	333								
No. 100 No.	Surrogetee								,	90,	Ş	87
No. 10, 10, 10, 10, 10, 10, 10, 10, 10, 10,		*	28	88	2	a	28	3	2	3 8	3	3 6
Wetlie W	2.4.5.6.TCMX	*	78	67	86	22	9/	5	8 6	26	8 8	2
Part Malaja Na	280	×	8	42	8	107	28	3 8	*	3 8	2 8	413
Months	o-tercheny	*	76	8	82	2	2	8	ő	3		2
March May Colors Color												
March Marc		Į.	900	0.55	0.065	1.3	0.062	0.072	0.1	0.28	9.1	\$0.02 0.02
Marie Mari	Aummun		9000	9000	40 00s	¢0.005	\$0.00	0.011	<0.005	<0.005	<0.01	<0.005
mark mark	Arthmony		3	100	1000	1000	000	\$0.0	40.05	0.01	0.0241	0.0281
Table Tabl	Arsenic		Ę	ę	Ę	9	6	ê	1.05	<0.1	<0.1	40.1
Marie Mari	Barium		1000	i i	, we	10005	000	40.00	00.00	<0.001	<0.001	<0.001
Mariotan May 120	Beryflium	a l	300	3 6	100	1000	10002	\$0.00	1000	\$0.00	<0.001	<0.001
mg/L c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c001 c002 c005	Cadmium	mor.	1000	3	180	20.5	170	280	350	52-	530	530
mg/L c_006	Calcium	TO.	200	300	500	1000	1000	1000	600	¢0.04	<0.01	<0.01
mg/L c0.05	Chromium (total)	TO THE	5 9	200	500	6	900	800	90.00	\$0.0 \$	<0.05	<0.05
May I. 6,100 6,100 6,000 <t< th=""><th>Cobat</th><th>HQL.</th><th>800</th><th>3 6</th><th>3 6</th><th>5</th><th>90.09</th><th>80.00</th><th>800</th><th>\$0.05</th><th>0.1</th><th>90 O></th></t<>	Cobat	HQL.	800	3 6	3 6	5	90.09	80.00	800	\$0.05	0.1	90 O>
mg/L 4,005	Copper		36	350	-	0.2	900	\$ 90.00	0.069	0.22	×0.06	5.8
mg/L 2.7 0.04 50 1 120 660 1000 420 420 1000 420	lron	TANK T	300	1000	9000	40 002 40 005	<0.005	\$0.00°	<0.005	<0.005	0.007 1	0.006 t
mg/L dits	Lead	-	200	3	ş	-	120	9	1000	420	1600	1900
mg/L c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.0002 c_0.001 c_	Magnesium		100	ç	0 0.065	000	40.0	¢0.01	40.01	<0.01	0.056	0.3
mg/L c(0)1	Manganese		2000	20000	×0.0002	<0.0005 40.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002
1,2,1 1,5,1 1,1 34 200 290 180 1	Mentury		100	500	1000	600	600	6.0	<0.01	<0.01	<0.01	800
10	Nickel		4	46	3.7	=	3	200	280	150	200	540
mg/L c001 c002	Potassium			Ę	1000	0.0111	60.0	\$0.0¥	40.01	<0.01	0.01	\$0.1
mg/L viv. <th< th=""><th>Selentum</th><th>1</th><th>5 5</th><th>Ş</th><th>1002</th><th>40.01</th><th>0.0</th><th>\$0.0</th><th>40.01</th><th><0.01</th><th><0.01</th><th>0.01</th></th<>	Selentum	1	5 5	Ş	1002	40.01	0.0	\$0.0	40.01	<0.01	<0.01	0.01
1	Silver				95	2	98	975	7500	3300	12000	15000
um mg/L control contro	Sodium	TANK T	200	cwo,	20000	×0.002	\$0.00°	2000°	40.00v	<0.002	<0.002	<0.002
00m	Thairtin		3 5	5	1000	9.0	000	40.05	¢0.01	<0.01	<0.01	¢0.01
	Vanadium			5	6	é	6	6	40.1	<0.1	40.1	ç0.
	Zinc		200									

| = value is between the Method Detection Limit and the Practical Quantitation Level * * diluted sample, - = not analyzed

		000 700 000	9000001	00,00,00			2016	Sole	CONF	MARSH-1S
		26/17/60	26.00	10/09/36	96/60/01	10/06/98	10/09/08	10,000	- AMAMA	
	Chits					of springs	2000	888	10/06/98	09/19/98
Organochlorine Pesticides (8081)										
4.4 500	ŀ	į								
44.006		8	8	6. 8	60 80	-0.05	\$0.0×	898	, O.S.	Ş
100	ğ	<0.09 ~0.08	8.8	89.9	<0.05	40 GS	Š	2	3 3	3
100 - 11	M	8	90.00	6 8	50.05	Ş	2	3	80.0	900
MOTH.	ž	900	800	Š	Š	3	3	8	8	\$ 8
Apria - BHC	1/0//	98	200	2	3 8	8.5	8	\$0.00 80.00	\$ 8	& 8
Beta - BHC	1	3 2	3 3	900	99	<0.05	98	899	\$0.0¢	8
Chloritana (alriha)		3	8	90.00	900	& &	800	\$0.05	8	Ş
	ğ	\$0 \$0 \$0 \$0 \$0 \$0 \$0 \$0 \$0 \$0 \$0 \$0 \$0 \$	¢0.08	8 8	\$0.05	900	5	200	3	3
Circums (germa)	100/L	60.05	90.09	900	50.00	90.00	3	333	9	9
Chlordane (total)	764	V	7	,	,	3	37.00	9.6	69.88	9
Deta - BHC	Ş	20.00	,		۶	\$	V	⊽	V	⊽
Chicken		8.69	8	60.05	\$ 8	8	\$0.0×	8	500	2
CHENCHIL	1/0/	8	8	\$0.0×	Ş	200	2		3	80.00
Endoguran	Z,	<0.05	Ş	8	200	3	3	\$	\$	99.08
Endoeullan II (beta)	Ž	200		3	800	S	9	8 8	8	900
Enchounties Suitate		3	3	800	×0.05	8 8	800	\$0.05	20,00	Š
Endis.	Į.	60.09	600°	80 Q	60.09	00°0	Ş	8	8	3
CIONNI	1001	8.8	800	900>	ې	5		Bio	3	8 0.0
Endrin Aldehyde	1/0/	9000	ž	200	3 3	3	80.00	8.5	¢0.08	8. 8
Endrin Ketone	Ş	20.00	2		33	8.5	90.00	\$0.00	6. 86.	8
Gamma - BHC (Lindane)		333	20.00	80.00	80.08	\$0.0g	8	88	800	5
Hantschire		8	88	¢0.08	8	\$0.0°	800	30,00	5	3
	Į,	60.08	8	& 8	900>	900	200	200	3 2	3
representation oppositely	707	8	\$0.0°	900>	866	200	2		3	9
Heptachlor Epoxide(b)	7/0/	800	9	Š	3 3	3	8.5	88	\$0.0 \$	40.0 2
sodrin	/w/	Š	90.00	3	30.00	\$	8	\$0.08	8	8
Methowchlor		2	3 3	3	900	8	8	8.8	\$0.0¢	90 09
Mirex		3	8	800	\$9.	8.8	8.6	90.00	50.00	8
Toverhere		3	\$ 8	898	\$ 8	8.9	\$0.00 80.00	89.65	5	2
	Į.	4	4	42	3	9	0	S	,	3
Arociors (6082)										y
747 TO-00-1	101	~	₽	12	Ī	Ţ	ļ			
TV0-1621	NO.	⊽	٥	ī	Į	,	1		v	V
-CB-1222	W	V	⊽	V	\ \ \		7	•	ē	v
PCB-1246	707	ī	,			1	۱	ī	۲,	⊽
PCB-1254	Ş	-	,		-	ē	v	V	•	⊽
PCB-1260	3			7	۲	v	۲,	V	⊽	Į
Chlodostad Marchaelas (8454)			Ÿ	⊽	V	v	V	ī	⊽	
Commence (als)										
244,4,5-1 nonconferioxy)proprinc acid (2,4,5 - TP) (Silvex)	1/0/	<0.5	<0.5	<0.5	20°5	36	į	į		
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	707	<0.5	\$65	29	3			8	\$0.5	<0.5
2,4-Dichlorophenoxy acetic acid (2,4 - D)	ğ	205	36	2		9	60.0	93	<0.5	<0.5
3,50CBA	Ę	ť			6	90	40.5	9.5	<0.5	\$0.5
H2.4-Dichlorophenoxy)butvric acid (2.4., DR)			60.00	9	93	<0.5	6 5	<0.5	\$05	60.5
1 - Nitrochanol	\$ 1	603	65	695	<0.5	<0.5	40.5	805	805	Ę
Acidi poden		9	6.5	<0.5	<0.5	<0.5	<0.5	\$65	é	96
Section 1	Į.	6 05	c0.5	<0.5	<0.5	\$6.5	905	ş		
CONTROL	W	<0.5	<0.5	\$0.5	59.5	Ě			2	900
Chloramben	YOY	<0.5	\$65	Ę	36			9	\$	9.5
Dacthal	70/	505	92	200		200	co.	<0.5	<0.5	<0.5
Delapon	101	\$05	200		000	ê	605	885	<0.5	<0.5
Dicamba	Ş	50.5	300		600	83	92	80.5	<0.5	40.5
Dictrioroprop (2-(2,4-Dictriorophenoxy)oroponic acid	١	2 4		600	ę,	ê S	60.5	40.5	<0.5	\$05
Dinoseb			5	8	9	<0.5	<0.5	<0.5	\$65	905
MCPA			3	\$02	\$65	<0.5	<0.5	<0.5	\$65	605
MCPP				2	ę	<10	<10	01>	910	950
Pentachkoromenok	100	2	01,0	ę	¢10	<10	ot>	ę	9	950
Dichem	100	60.5	92	<0.5	<0.5	<0.5	\$0.5	865	30,	
	3	600	-	- C	344	200				,

I = value is between the Method Detection Limit and the Practical Quantitation Level * = diuled sample, · = not analyzed

	_	OLINE-3S	OUNE-4S	20.32		5	CONCASS	1	3.3	
	<u>.</u>	09/21/98	10/09/98	10/09/98	10/09/98	10/09/98	10/06/98	10/08/98	10/08/98	09/19/98
	Pinis					dulpicate				
	e di la									
PAH (6310)				3,0	Ş	300	20,	205	<0.5	<0.5
1 - Methylnaphthalene	NA.	60	603	600		200	Š	50.5	\$65	605
2 - Methylnaphthalene	W	9.5	600	\$0.5		200	3 4	200	9	<0.5
Acenachthene	76	<0.5	<0.5	69	40.5	000			2	Ę
Acenachtiviene	7/0//	<0.1	¢0.1	8	-0.1	00	9	200	2	į
Anthonone	707	<0.5	<0.5	40.5	<0.5	¢0.5	30.5	800	20.0	200
Dentalelanthracehe	র	0.13	90:0>	<0.05	¢0.65	60.05 0.05	\$0.00 \$0.00	8	8 6	3 8
Detail (a) and expend	Į (g	1900	\$0.0	\$0.0×	<0.05	6 8	¢0.05	900	8	8
Denzo(alpyrene	200	Ę	601	000	40.1	60.1	40.1	60.1	\$0.1	¢0.1
Benzo(b)ituoranthene		Ę	Ę	9	-0.1	0.0	60.1	<0.1	\$0.1	6 -
Benzo(g,h,i)perylene			5	۶	c0 05	¢0.05	\$0.05	<0.05	\$0.0×	<0.05
Benzo(k)fluoranthene	TANK!		3 9	5	50.05	\$0.0¢	\$9.69	<0.05	<0.05	90.09
Chrysene	io	2 0	3		Ę	é	6	60	¢0.1	40.t
Dibenzo(a,h)anthracene	100	5			,	ć	Ę	ê	0,0	60.1
Fluoranthene	No.	6	5			ę	Ę	é	601	6
Fluorene	Z,	9	ē		200	2	200	800	90.05	800
indeno(1,2,3-cd)pyrene	NOT NOT	5-0	3.5	900	300	3	36	50.5	20.5	60 5
Nachthalene	1/0//	c0.5	¢0.5	605	9	200	200	200	8	50.65
Phenanthrene	1/0//	6. 8	ê 8	900	8.6	3 5	3 5	3 6	8 6	50.05
Pyrene	M	40.06	60.0 8	¢9.09	80.89 V	30.00	20.00	3		
Surrocates										
5.4 PCAA	ļ	8	9/	8	130	118	8	146	202	8 3
V-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1	×	8	90	83	85	88	S	8	3	2
Z,4,0,0-10mX	*	28	ğ	88	86	112	88	Z	9	128
	,	8	88	86	70	8	91	ğ	8	
C-terpinenty										
	901	9	98	\$0.05 \$0.05	9.14	0.11	0.061	0.082	8	0.063
Auminum			S. C.	90002	0.012	\$0.0	<0.005	¢0.002	<0.005	\$0.00
Antimony	T T	1000	500	1000	0.0121	\$0.0	0.065	960.0	0.029 1	\$ 0.01
Arsenic		100	ę	-0	6	6.0	40.1	c0.1	¢0.1	Ç0.1
Barken		1800	i (ii)	100.05	1000	40.00	<0.001	0.0031	<0.001	6 8
Beryffum		200	1800	10000	40.001	¢0.00	0.002	0.00	0.001	00°0
Cedmium	TA COL		3	8	3	330	7.5	æ	130	610
Calcium	1,011	300	500	100	10.0>	\$0.05	\$0.0	0.01	<0.01	¢0.01
Chromium (total)	TANK T	5 6	5	98	900	800	\$0.0°	<0.05	<0.05	\$0.09 0.09
Cobett	TOWN TO SERVICE STATE OF THE PARTY OF THE PA	3 8	5	900	86	900	\$0.0°	<0.05	\$0.05	8 8
Copper		3 5	8	8	•	9.	\$0.0°	0.11	0.3	9
Lion		38	3000	9000	9000	- 900'0	<0.005	0.027	<0.005	<0.005
Lead	TOTAL STATE	3000		=	1100	2	6.2	3	210	160
Magnesium	100	3	0	100	0.15	0.15	0.042	0.011	0.037	0.13
Manganese	TO.	1000		2000	20000	40 0002	<0.0002	-0000S	<0.0002	<0.0002
Mercury	TO.	40.00k	1000	1000	1000	90.0	900	0.012	¢0.01	<0.01
Nickel		99.01	200		9	\$10	<0.5	21	120	9.6
Potassium	J.Ow.		200		200	200	ě	0.027	\$0.01	-0°0
Selenium	T/Ou	500	000		200	500	\$0.04	000	000	\$0.0×
Silver	mo/L	40.01	50.00	20.0		9000	25	200	2000	1780
Sodium	Agr.	8/4	2	200		2000	au o'	2000	\$0.00°	0.003
Thallium	AQ.	<0.002	S C	\$0.00¢	2000	3 6	500	0.013	40.01	900
Vanadium	J/Dir	\$0.0	000	S C	2		Ę	ę	6	6
	700			-			?			

I = value is between the Method Detection Limit and the Practical Quantitation Level * = diluted sample, : = not analyzed

		99413/98	09/13/98	9000	CZ-UCUVIII	00.1100	MARSH-3	MARSH-4S	H-4S	MARSHA	MADOL ES
	Units		duolicate	D6.81	08/27/88	09/24/96	09/24/98	08/2	09/22/98	09/22/98	10/06/08
Organochlorine Pesticides (8061)									duplicate		2
4,4: 000	1/0/	800	20.00								
44. DOE	Ş	3 6	8 8	800	\$0.09 80.09	<0.05	<0.05	\$0.0\$	300	Ş	
4,4°-00T	100	3 6	3 5	\$0.00	8 0 8	<0.06	\$0.0×	6 8	5	3 8	900
Adm	Į.	3 6	8 6	800	8	<0.05	\$0.0¢	\$0.05 80.05	50.05	3 2	8 8
Apha - BHC	100	3 8	9 8	8	ç 8	<0.05	-0.08 -0.08	60.05 0.05	50.00	3 2	8
Beta - 69+C		300	9	6 8	90:0°	<0.05	88	9009	36	88	\$0.08
Chlordane (sipha)		90.00	8	6 8	90°0	& 8	98	200	3 8	8 5	<0.08
Chlordane (gamma)		8 8	8	6 8	90.0°	800	90.00	500	3 5	8 8	40.08 80.08
Chlordane (total)		90.5	\$ 8	<0.05	90.0 >	800	5	3 6	8 3	88	ç0.0 8
Deta - Par	S.	⊽	₽	۶	<12	7	3	80.00	¢0.08	6.08	¢0.05
District	100	¢0.08	89.0	\$0.0°	800	2	7	ē	<1		⊽
Fados Man	W	<0.05	800	8000	800	3 8	39	6.08	<0.05	80.6	8000
Tri North and Tri	W)L	\$0.0\$	\$0.0¢	8	3 2	9.5	88	د0.05 د0.05	<0.05	800	50,00
crookinan II (beta)	704	\$0.00 80.00	5	200	8 8	8.0	998	<0.05	8	60.05	300
Endosuman Sumate	ğ	60 0°	800	3 5	9 3	×0.08	8	<0.05	8	800	3 6
Endrin	109	5005	3 6	800	(0)	80°	¢0.09	90°0	90.00	8	38
Endrin Aldehyde	Ş	3 6	3 5	90.00	8	<0.05	<0.05	88	20.00	3 6	8 8
Endrin Kelone	100	3 5	9 5	80.00	90 Q	<0.05	89.05	888	8	3 5	300
Gamma - BHC (Lindane)		9 2	90.00	\$0 \$0	×0.06	×0.06	8	900	3 8	3	\$0.00 0.000
Heptachior	2	9	\$0.0×	<0.05	90:0×	800	5	3 6	9 5	800	608 8
Heptachtor Enoxide(a)		99	\$0.0g	<0.05	90.0°	800	5	3 5	9 2	\$0.8	60.05 50.05
Heutschlor Enouglech)	100	\$	\$0.0¢	<0.05	90.00	50	2 6	3 8	8	6 8	<0.05
Bodrin	Š	8	\$0.0¢	<0.05	\$0.0°	99	3 8	8 8	99	\$9. 89.	<0.05
Methorophor	100	\$ 8	<0.05	-0.05	800	50.0	300	8 8	88	\$9.09 80.09	<0.05
Mark	YOU	8.	<0.05	\$0.05 80.05	9000	200	3 5	80.00	88	×0.08	\$0.0°
Toxachana	100	60.86	<0.05	\$0.0¢	\$0.0°	8	3 5	9 8	\$0.05	<0.05	<0.05
	Į,	4	4	8	<2.5		3	8,	90.00	8	<0.05
Arodors (BUEZ)								7	62	<2	42
PCB-10101242	7/07	<1	⊽	ī	:;						
1221-001	767	Þ	⊽	,			Ī	v	₹	₽	ī
PUB-1232	7/0//	Ţ	v	7		Ş .	V	v	دا	ī	-
FC8-1248	701	⊽	-			ē	⊽	5	ī	⊽	
PCB-1254	ş	ī			7	v	v	₽	ē	V	
PCB-1280	ž	v	\		Į,	Į	دا	₽	ī	V	1
Chlorinated Herbicides (8151)	ľ				7	Ţ	×1	V	⊽	v	
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	J/O/	205	į	ļ							
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	1/0/	Š		9	0	ç	<0.5	<0.5	\$0°5	ž	93 0
2,4-Dichlorophenoxy acetic acid (2,4 - D)	101	505		99	9	Ş	<0.5	<0.5	\$05	5	8.00
3,5-DCBA	705	505	26	9 5	22	2	<0.5	<0.5	\$05	500	8 8
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	704	202	3 6	9	8	çç	<0.5	<0.5	\$65	5	3 4
4 - Nitrophenol	ZQQ	60.5	200		Ş.	ę	<0.5	<0.5	999	205	35
Acituorien	ğ	\$05	200		\$	Ş	<0.5	<0.5	<0.5	\$05	3
Bentazon	100	50.5	3	9	0	\$	<0.5	<0.5	<0.5	202	250
Chloramben	18	200		9	0	\$	<0.5	<0.5	40.5	500	8 8
Dactha	VO.	é		9	8	ŝ	<0.5	60.5	605	36,	8 9
Oatapon	Ę	36	9	9	Ş	45	<0.5	\$05	205		8 8
Dicamba	٤			8	ç	\$	<0.5	¢0.5	Ş	200	90.00
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	Ş		000	69	\$	<5	<0.5	60.5	5	2	90.00
Dinoseb				\$0.5	Ş	ç	<0.5	\$05	46		8
MCPA	100		600	92	6	<5	<0.5	902	205	600	800
MCPP	Ş			8	8	×100	د10	95	9		80.38
Pentachlorophenol	100	150	010	9	× 18	×100	<10	9	ę	25	5
Pictoram	Vog	1			ę,	ŝ	<0.5	40.5	6.5	3,6	93 0
			I	Q.5	9	¢\$	<0.5	<0.5	\$05		8 9

! = value is between the Method Detection Limit and the Practical Quantitation Level * = diluted sample, - = not analyzed

1 - Methyknaphthalene		09/13/96 04/13/99 04/13/99 04/15/19/99 05 05 05 05 05	08/19/98	09/27/98	09/24/98	09/24/98	09/22/98	2/98 duplicate	09/22/96	10/06/98
PAH (\$310) 6 6 6 6 6 6 7 7 8 8 9 8 9 9 9 9 9 9 9 9 9 9 9 9 9 9				i c				duplicate		
PAH (6310) 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6			<0.5	30						
e e en e en e en e en e en e en e en e		H	<0.5	300						
ene ene Surrogates		\prod	,	COD	<0.5	<0.5	<0.5	<0.5	405	\$0.5
ene ene Surrogates			600	<0.5	<0.5	<0.5	\$00	<0.5	<0.5	<0.5
ene ene yne Surrogstes			605	40.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
e ene		-	-0	0	601	-01	<0.1	<0.1	<0.1	0.0
ene ene frie Surrogates		-	<0.5	<0.5	40.5	ć0.5	<0.5	<0.5	<0.5	<0.5
ene Ante	$\frac{1}{1}$		90.00	\$00°	\$0.05	90.0>	<0.05	<0.05	&0.85	<0.05
Surrogates	$\mathbb{H}\mathbb{H}$	-	800	900	9000	40.05 40.05	<0.05	<0.05	<0.05	\$0.00
Surrogates	+H	+	,	ę	9	00	-60.1	60.1	<0.1	<0.1
Surrogates	$oldsymbol{H}$	+	-	Ę	Ę	ę	601	6.0	<u>6</u>	40.1
Surrogates	+	1.00	2	200	8	Š	898	\$0.09 50.09	×0.05	\$0.05
Surrogetes		+	8 8	300	3 5	36	8	Ş	89	9000
Surrogites	-	+	8.5	8	30.00	3,5	3	1	400	100
Surrogates	-		(0)	5	9				Ç	5
d)pyrene Surrogalas		<0.1	9	ê	Ş	Ş	- CO.	2	9	9
Ojpyrene Surrogates	_		\$0.1	8	9	90.1	\$0.1	2	000	100
Surrogates	L	<0.05	<0.05	<0.05	\$0.0 \$	<0.0 6	¢0.05	\$0.0g	\$0.00	80.05
Surrogales	┞	L	<0.5	<0.5	<0.5	<0.5	<0.5	\$9.5	\$0.5	902
Surrogates	-		800	\$0.0°	<0.05	<0.05	<0.05	¢0.05	\$0 8	\$0.88
Surrogates	╀	90.00	\$0.05	\$0.05	<0.05	<0.05	¢0.09	<0.05	<0.05	<0.05
Surrogates	ŀ	l								
	$\frac{1}{1}$	-	30.	١		Ę	2	140	95	130
	+	1	3 3	٥	8	8	98	8	78	16
2,4,5,6-TCMX	+	98	8 8	3 8	38	8	87	1	88	92
	+	26	3 4	3	22	8	88	23	88	115
p-terphenyl	$\frac{1}{1}$	71/		2						
Metals	$\frac{1}{1}$	$\frac{1}{1}$						-	0.20	0.42
Atuminum M9/L	-	4	$\frac{1}{1}$	800	21.5	100	300	300 0	000	AUC C
Antimony	Н	-	¢0.09	0.000	c00.0>	500	40.00	COMOS	200	
Anamic mg/L			¢0.05	¢0.0	49.01	(0.0)	G.000	200.5	0.00	
Reckin mg/L	_			ê	0.41	100	9		- 00	100
£	\vdash	<0.001 <0.001		00 00 100 00	¢0.001	,0 9	40.001	(0.0)	\$ 0.00	989
	H			¢0.001	89	¢0.001	40.001	(0,00)	(C)	3
	Н	580 580	670	260	240	260	9.5	3.0	200	2 6
m (total)		Ц	-	\$0.00	40.01	40.01	9.03	0.018	200	9
	H	<0.05 <0.05	-	\$0.0°	80	90.09	8.6	35	8 2	3 5
			-	\$0.09	0000	0.038	800	9	8	9.00
	H		-	\$ 8	\$0.0 ₂	6.3	980	G.U/S		9000
		Ц		¢0.005	9000	0000	60.05	90.00	0.03	CONTO V
Erita	L			25	450	28	2	2	36	2
	H	H	L	0.015	1700	0.46	\$0.01	¢0.01	3,0	0.023
	H	<0.0002	\vdash	<0.0002	<0.0002	<0.0002	¢0.0002	<0.0002	<0.0002	<0.0002
	H	\vdash	H	<0.01	<0.01	<0.01	40 <u>.</u> 01	6 .01	40.01	40.01
	H	ŀ	L	170	160	120	28	27	8	\$
	ŀ		-	¢0.01	<0.01	<0.01	<0.01	<0.010	<0.010	\$
5	+			1000	900	<0.01	<0.01	<0.01	<0.01	¢0.01
	+	-	1300	999	3900	7100	95	510	14000	1400
	╀	0.00211 0.003	L	\$0.00°	<0.002	<0.002	<0.002	×0.002	<0.002	¢0.005
	+		¢0.01	¢0.01	¢0.01	<0.01	0.012	0.011	<0.01	<0.01
Own	+			\$0.5	60	¢0.1	c0.1	<0.1	<01	<0.1
Zwc										

I = value is between the Method Detection Limit and the Practical Quantitation Level * = diluted sample, - = not analyzed

		MARSH-51	MARSH-5D	MARSHAG	WIE CT 10						
	e e	10/06/96	10/06/98	09/20/98	10/07/98	10/07/9A	WEST-10	WES	WEST-2S	WEST-3S	WEST.3
Organochlorine Pesticides (8081)							200	2	10/21/98	10/16/98	10/16/98
	ļ								duplicate		
4.4' · DOE	Z.	ê 8	<0.05	\$0.0×	2000						
4.4. DOI	70	60 8	\$0.08	9000	200	9	6 8	\$0.05	800	\$ 6	
Aldrin	100	\$ 9	<0.05	\$0.0°	38	800	6 8	<0.05	9000	3 8	8
Alpha - BHC	100	6 8	900	800	3 5	9	6 8	89.8	500	3 2	8
Beta - BHC	100	\$0.0g	\$0.00 80.00	Š	8 6	8	608 8	6 8	6	8 6	8
Chlordane (alpha)	To To	¢0.09	6 86	60	3 5	8	<0.05	6.85	8	3 5	98
Chlordene (cemme)	Z,	9.09	898	200	3	\$0.09 80	¢0.05	\$0.05	ę	313	6 8
Chloriana (total)	101	6 8	800	3 2	8.5	\$ 8	900	6 8	3 2	8	6.88
Della Dia	Z	V		8	\$0.08	¢0.05	\$0.0°	200	8 8	999	÷0.05
	Ŕ	5	200	ē	Ţ	٧	₹	3	9	6 8	\$0.0°
	1/07	900	S S	88	\$0.00 80.00	800	5		Ÿ	₽	⊽
Choosinan	Ş	300	99	\$0.00 80	80.8	800	2	3	900	8.8	80.05
Endosultan II (beta)		8	¢0.08	\$0.0 5	800	2	3 3	60.08	<0.05	88	5
Endosultan Sulfate		80.00	¢0.05	\$ 80.05	8	3 2	9	6. 8	\$0.00	998	3 2
Endin	100	900	60 O>	8000	8 0	38	8	¢0.05	\$0.09 80.09	8	3 8
Endrin Adehyde	ZZ	& 8	88	89.65	8 6	SO.OS	89	60.0°	8	3 8	8
Endrin Kelone	707	<0.05	86	5	3 3	808	<0.05	80.0	ę	8 8	00.09
Gamma - B+C (Lindana)	100	40.06	800	200	3 3	8	99	90.00	5	3	8
Hectachio	700	\$0.0°	9000	3 6	8	\$0.00	<0.05	900	3 2	8	¢0.09
Manuachios Essentialist	NOV	8	200	3 5	98	<0.05	800	8	3 3	8	\$0.00 \$0.00
Harrist Etonogia)	ğ	8	3 5	8	\$ 8	8,0	60.00	3 8	9.5	8	60.05 50.05
specimo sportego)	ğ	89	3 8	8	899	9.00	998	3 8	3	98	\$0.0¢
Mothers	Ž	2000	3 8	8	898	\$0.05	\$0 GS	3 2	8	\$0.00	90.0V
A Manual Andrews	Ϋ́ S	8	3 5	8	90.00	\$0.00	8000	3 2	89	<0.09	<0.05
TO MAKE	Ş	200	8 8	88	<0.09	800	5	9	\$0.09 80.09	\$ 8	\$0.05 80.05
Loxaphane	707		9,0	89	\$0.05	88	É	9 5	\$0.00	\$0.0g	\$0.0¢
Araclars (8082)			8	7	42	3		Section	\$0.0	<0.05	\$0.0\$
PCB-1016/1242	Į.	ļ							8	ø	7
F08-1221	Ę			V	1	Ī	ļ				
PCB-1232		1	7	F	₽			٦	⊽	دا	ī
PCB-1248		1	V	V	ī		-	₹	V	⊽	
PCB-1254		۱	ē	۷.	ī	+	1	V	₽	⊽	
PCB-1280		Ĭ	ē	⊽	V	1	7	v	V	⊽	
Chlorinated Harbickies (8181)		⊽	•	V	·		V	₹	v	Ī	
2-(2,4,5-Trichlorophenoxyloropionic acid /2.4,5-Trichlorophenoxyloropionic					†	1	Į,	⊽	Ü	⊽	1
2,4,5-Trichlorophenoxy acytic acid (2,4,5-11) (Silvex)	Z Z	<0.5	<0.5	905	1	1					
2.4-Dichlorophenoxy spekt acid (2.4. D)	Z,	<0.5	\$05	902	000	92	<1.0	89.62	800	1	
3,5-DCBA	No.	<0.5	¢0.5	\$05	600	\$65	<1.0	\$ \$	29.65	6	60.5
4-(2,4-Dichlorophenoxy)butyric acid (2.4 - na)	100	<0.5	<0.5	<0.5	1	999	<1.0	29.0V	8	36	900
4 - Nitrophenol	100	60.5	<0.5	\$0.5	200	9	0.5	20.0°	\$ \$	505	9
Acifluorlen	2	\$00	<0.5	\$05	505	2 4		800	28.00	805	200
Bentazon		\$ 9	<0.5	<0.5	20.5	300		8	29.00	\$05	200
Chloramben		9	<0.5	<0.5	605	1	2	8 8	29.00	8	5
Dacthel		\$02	<0.5	<0.5	\$0.5		2	8	<0.62	\$0.5	200
Dalapon	į,	\$0.5	<0.5	<0.5	300		0.15 	30.62	8	20.0	
Dicamba	S.	<0.5	<0.5	\$0.5	500	999	0.2	8	& 8	\$05	000
Dichloroprop [2-(2,4-Dichlorophenoxyphynomic soid)	No.	c05	<0.5	\$0.5	200		410	29.0 2	29.05	292	
Dinoseb	SQ.	\$0.5	<0.5	605	500	699	010	900	\$ \$	\$65	600
WCPA		\$65	<0.5	<0.5	20.5	000	0.5	20.6 2	\$0.68	205	
MCPP		010	¢†0	¢10	900	200		89.09	<0.62	\$05	200
Pentachlorophenol		000	¢10	<10	55		8 5	<12	<12	95	200
Poloram		600	¢0.5	<0.5	800	1	8	<12	<12	ê	9
		<0.5	<0.5	<0.5	\$0.5	36	010	29.00	29 0>	\$65	200
						2.02	41.0	8	800	3,4	

		MARSH-51	MARSH-5D	MAHSH-00	WE31-13	WEST	000170770	10/21/86		10/16/1998	10/16/1998
		10/06/1998	10/06/1998	09/20/1998	10/07/1998	10/07/1998	01/12/1999	301	dendinate		
	Chrits								Outricals		
DAM (19910)									3 4	40.5	70.5
ı	VQ,	<0.5	<0.5	<0.5	<0.5	<0.5	¢0.5	\$0.5	200	250	50.5
- Wethynaphmalene	٤	40.5	<0.5	<0.5	<0.5	<0.5	\$0.5	600	600	2	300
- Methylnaphthalene	1	20,	505	<0.5	<0.5	<0.5	<0.5	\$0.5	600	20.3	
Acenaphthene	1		Ę	ç	<0.1	40.1	<1.0	\$0.1	\$0.1	40.1	9
Acenaphthylene	100		2	300	70.5	<0.5	¢0.05	<0.5	<0.5	<0.5	¢0.5
Anthracene	104	000	6	500	<0.05	¢0.05	<0.05	<0.05	0.06	0.16	<0.05
Benzo(a)anthracene	VOV	9	900	3 5	300	40.05	<0.05	<0.0>	0.07 1	0.14	<0.05
Benzo(a)ovrene	W	60.65	900	333	3	10,	691	¢0.1	1.0>	0.171	<0.1
Penzothilluoranthene	M	ç0.1	601	9	20.1		5	100	60.1	¢0.1	<0.1
Describe Described	VOV	¢0.1	40.1	\$0.5	100	- 60	200	70.05	50.05	0.11	\$0.05
Selection from	γoγ	<0.05	<0.05	<0.05	908	60.05	20.05	300	300	0.17	\$0.05
DOUT OF INTERIOR	7/0/7	\$0.05	<0.05	<0.05	<0.05	\$0.0°	SO O	3000		Ç	10
Chrysene	3	100	40.1	40.1	<0.1	¢0.1	ê	9	\$		
Dibenzo(a,h)anthracene	1 5	5	100	601	-0.1	40.1	¢0.1	Ç	Ç0.1	8 3	0
Fluoranthene	1		Ę	6	00	60.1	¢0.1	0,	6	100	9
Fluorene	100	200	200	<0.05	<0.05	<0.05	<0.05	c0.05	40.05	60.0	60.00
Indeno(1,2,3-cd)pyrene	100	300	26	<0.5	40.5	<0.5	<0.5	<0.5	60.5	\$02	600
Naphthalene	ě.	200	200	50.05	\$0.02 0.05	40.05	<0.05	<0.05	c0.05	¢0.05	500
Phenanthrene	Ž,	20.00	200	5005	\$0.05	¢0.05	<0.05	<0.05	<0.05	<0.05	CD 02
Pyrene	Į,	20.02	20.02								
Surrogates					905	٤	ř	122	09	114	8
DA-DCAA	3º	85	8	3	97	3	٤	25	122	120	111
DA F. R. TCLAY	%	114	87	20	36	2	3	3	9	100	8
	×	136	8	\$	£,		3 3	šä	22	85	88
o territorial	%	110	71	91	è		5				
Metals	L						į	1	2000	,005 5005	0.15
	TO/L	0.072	90.0	0.17	0.25	6.08	9	2000	300	2000	500
Aummum	ě	\$000°	\$00.0×	<0.005	<0.005	\$0.005	\$000	SO OS	2000	200	11000
Antimony	2	100	000	0.021	<0.01	٥٥٥٠	40.01	\$0.01	20.01	100	
Arsenic		0.43	ę	0.13	60.1	٥٠ م	۰01	\$0.1	9	20.1	100
Banum	100	1000	1000	0000	<0.001	<0.001	<0.001	\$0.00	00:00	1000	900
Beryflum	1	8 6	000	40.00	<0.001	<0.001	¢0.001	900	¢0.001	1000	00.00
Cadmium	TOE .	3	670	250	8	140	× 50	30	130	140	270
Calcium	TANK I	3	200	1000	1000	<0.01	<0.01	×0.01	400v	¢0.03	000
Chromium (total)	T ST	200	9000	\$0.05	¢0.05	<0.05	<0.05	<0.05	\$0.05	\$0.05	9 8
Cobalt	1	3 5	8	40.05 40.05	<0.05	<0.05	\$0.05	\$0.05	\$0.05	50.05	40.00
Copper		300	5005	9	1,4	1.4	<0.05	0.062	0.067	60.05	1.0
lron		3000	3000	\$0000	\$000	<0.005	<0.005	¢0.005	<0.005	<00.00 0.00	9000
Lead	T T	3	1500	965	**	3	<0.50	25	8	9.	3
Magnesium	E .	900	0.062	021	0.017	0.023	<0.010	0.021	0.021	0.01	0.041
Manganese	E C	0.078		20000	20000	<0.000×	<0.0002	<0.0002	<0.0002	×0 0005	¢0.0002
Mercury	Ē	SO.WK		800	000	000	\$0.0	<0.01	<0.0	<0.01	¢001
Nicke	Ę,	000	(O.O.)	3 2	9	**	60.50	12	12	1.3	9
Potassium	Z E	8	200	3	100	100	40.01	40.0	<0.01	<0.01	<0.01
Selenium	Σ	1000	000	200	2	100	000	\$0.0	<0.01	<0.01	40.01
Niver.	Z Z	1000	10.00	200	5	199	35	230	220	52	1200
Sodium	Age.	1980	DO S	300	2000	20002	<0.002	¢0.002	<0.002	<0.002	<0.002
Thalium	Age.	2000°	\$ C	20.00	4	100	5003	¢0.01	<0.01	0.01	<0.01
Vanadium	mo/L	\$	\$0.0°	1000	1000		5	6	6	00	60.1
						-	?				

| = value is between the Method Detection Limit and the Practical Quantitation Level * = dituled sample, · = not analyzed

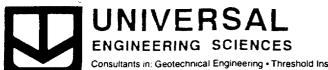
		WEST-30	WEST-4S	WEST-4	WEST-40	WEST-5S	WES	1.51	WEST-65
		10/16/1998	10/14/1998	10/14/1998	10/14/1998	09/29/1998	09/29/96	786	09/27/1998
	Units							duplicate	
Organochlorine Pesticides (8081)									
4.4 - 000	M	د0.05	<0.05	90·0>	<0.05	<0.05	<0.0×	\$0.05	50.05
4.4 - DOE	ZZ	د0 05 8	<0.05	<0.05	<0.05	<0.05	40.05	\$0.0¢	\$0.05
4.4 · UDI	707	8	\$0.05	<0.05	<0.05	<0.05	\$0.0¢	60.05 80.05	¢0.05
Adm	ž	¢0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	60 80	\$0.0°
April - Brita	VQV	\$0.05	<0.05	<0.05	\$0.0 5	<0.05	<0.05	\$0.05	\$0.0×
Chloridate (sloke)	707	\$0.00	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Chicago agus	707	¢0.05	<0.05	<0.05	د0.05 د0.05	<0.05	<0.05	<0.05	<0.05
Chicago (alt.)	70	800	¢0.05	¢0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Chicagonal (County)	70/	ē	⊽	v	7	<1	۲>	⊽	1>
	1/0/	¢0.05	<0.05	<0.05	\$0.05	<0.05	<0.05	\$0.05 50.05	<0.05
	707	\$0.0 ₅	د0.05	0.06 د0.06	<0.05	<0.05	<0.05	¢0.05	40.05 50.05
	1/0/	¢0.09	د0 وج	د0.05	<0.05	<0.05	<0.05	¢0.05	900
Endosulian II (beta)	Z,	\$0.05	<0.05	<0.05	<0.05	<0.05	\$0.0\$	<0.05	<0.05
Errodeuner Surene	100	800	¢0.09	<0.09	\$0.0°	<0.09	60.0°	60 Q2	<0.09
Engran	1/0//	60.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0°5	<0.05
Endrin Adenyoe	VOV	<0.05	<0.05	<0.05	<0.05	<0.05	c0.05	80%	<0.05
CITOTRI NEIGHB	100	<0.05	×0.05	<0.05	<0.05	<0.05	\$0.05	¢0.05	40.05 50.05
Carrena · Dr.C (Lindane)	NA NA	40.05	د0.05 د0.05	<0.05	<0.05	×0.05	<0.05	<0.05	<0.05
Ingulation	101	<0.05	<0.05	\$0.08	<0.05	<0.05	<0.05	<0.05	\$0.05
repartment Epotocog(a)	YO,	\$0.0 \$	<0.05	<0.05	<0.05	<0.05	<0.05	¢0.05	\$0.05 50.05
reparament Epotoce(U)	100	40.08	\$0.05	<0.05	<0.05	<0.05	<0.05	\$0.0°	¢0.05
Modern	Ŋ,	40.05 0.05	¢0.0\$	<0.05	<0.05	<0.05	<0.05	¢0.05	<0.05
Mice	No.	800	<0.05	<0.05	<0.05	<0.05	c0.05	<0.05	<0.05
Totachene	\$ 1	900	\$0.05	\$0.05	\$000	88	<0.05	<0.05	<0.05
Araclass (MM2)		,	*	2	3	\$	ζ,	<2	<2
PCB-1016/1942	ļ								
PCB.1221		•	- -	V	v	Ÿ	⊽	<1	7
DCB 1979	100	₹ ·	V	V	v	ē	F	.	دا
970 000	Š	5	5	v	₹	5	<1	Þ	⊽
DOD 1964	70	Ş.	٧	V	⊽	V	۲,	4>	دا
100 Look	S.	Į,	v	v	V	-	دا	1 2	٠,
Chlodinated Harbiddes (8151)	100	v	v	v	ī	ī	~	د1	<1
	ļ	3 4	,						
л _		000	\$0.5 5	000	\$0.5	¢0.5	<0.5	<0.5	<0.5
2.4-Dichlorophenoxy acetic acid (2.4 - D)	Ş	500	000	000	\$0.5	\$0.5	<0.5	<0.5	<0.5
3.5-DCBA	٤	9 6	60.0	000	000	\$0.5	<0.5	\$0.5	<0.5
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	76%	0.5	505	200	9 4	000	000	40.5	<0.5
4 - Nitrophenol	ğ	<0.5	605	902	202	3 6	600	603	000
Acifluorien	w	<0.5	<0.5	<0.5	40.5	205	502	200	600
Bentazon	VOV	<0.5	<0.5	<0.5	<0.5	605	605	505	200
Chloramben	707	<0.5	<0.5	<0.5	26.5	<0.5	6.65	200	200
Dacthal	1/0//	<0.5	<0.5	<0.5	<0.5	c0.5	505	300	
Dalapon	ZQ.	<0.5	<0.5	<0.5	<0.5	60.5	40.5	\$05	505
Dicamos	10	<0.5	<0.5	<0.5	<0.5	=	40.5	<0.5	505
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic acid]	ZQ.	<0.5	<0.5	<0.5	<0.5	\$0.5	<0.5	\$0.5	505
Diriosed	K	40.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
MCA	V.	ę	95	<10	د10	<10	<10	¢10	\$6
Designation	707	40	¢10	¢10	¢10	<10	<10	×10	¢10
Perhabition	700	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	605
rcocam	1/0/	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5
				,	20.00	10.00	2		0.00

Wilstand Politicist Wilstand Politicist				4						֡
Junis			WEST-30	10/14/1998	10/14/1998	10/14/1998	09/29/1998	62/60	- 1	09/27/1998
PAH14310 Cont.		Ligite	00000	200						
PANTONIO 1997. 405 405 405 405 405 405										
Marchite	PAH (8310)				į	30,	307	505	<0.5	<0.5
Marche March Color Col	- Methylnaphthalene	YOU	¢02	605	000	200	200	505	\$0.5	<0.5
Marie Mari	- Methylnaphthalene	You	¢05	600	603		300	200	50.5	405
Martin	senachthene	T/O/	<0.5	<0.5	\$0°	000	000		5	Ę
Marie Mari	cenachthylene	MON.	ç0.1	¢0.1	40.1	0)	0,0	100	100	200
Part Color	othracene	JON.	<0.5	<0.5	<0.5	¢0.5	cO5	50.0	200	
Marie Mari	a postporana	Von	\$0.0¢	<0.05	<0.05	<0.05	<0.05	\$0.02 \$0.02	\$0.05	80.00
Particle 601 60	enzu(a)ammerene	Von	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	\$0.05 CO 05
Material	enzo(a)pyrene	100	-	0.0	100	¢0.1	40.1	<0.1	¢0.1	¢0.1
Marie Mari	enzo(b)fluoranthene	NA.			Ę	6	6	60.1	40.1	<0.1
Marie Mari	enzo(g,h,i)perylene	Nov	Ç0.1	- 1	200	2 2	30 0	300	50.05	<0.05
Mature	enzo(k)fluoranthene	NOV	¢002	\$0.02	20.00	2000	300	900	200	200
Distriction	hosene	MO/L	<0.05	<0.05	c005	¢0.05	co os	80.00	3	300
National	ibonacia hisothracena	NON	<0.1	<0.1	¢0.1	¢01	601	9	9	0
National Part Colo Col	indirection in account	Von	601	¢0.1	<0.1	د0.1	601	¢01	0	Ç.
	UCITATION	Von	60	¢0.1	¢0.1	<0.1	<0.1	¢0.1	\$0.1	601
Surrogales	Norene	VO.	200	<0.05	\$0.05	\$0.0×	<0.05	<0.05	<0.05	<0.05
National Part Color Colo	deno(1,2,3-cd)pyrene	100	300	20%	\$05	<0.5	<0.5	<0.5	<0.5	<0.5
CALAX Surrogates	aphthalene	W		200	2002	\$0.05 50.05	<0.05	<0.05	<0.05	<0.05
Characteristic Paper COC	henanthrene	TANK!	36	200	50.05	<0.05	<0.05	<0.05	<0.05	<0.05
CAA Surrogates % 126 112 110 74 62 94 E-TCAAX % 96 115 106 94 66 131 E-TCAAX % 96 115 116 96 84 66 131 P-TCAAX % 96 115 116 97 150 151 P-TCAAX % 80 98 115 79 151 150 P-TCAAX % 80 98 97 102 71 115 Archard 400 98 97 102 77 115 115 Archard 400 98 97 400 4005	yrene	W	300	200						
COA % 126 112 116 44 66 73 G-TOMX % 96 115 166 94 66 115 Perory Metalis % 96 115 96 87 79 155 Prenry Metalis % 90 98 175 71 110 Annah Metalis mg/L c-0.05 0.074 c-0.05	Surrogates						į	į	g	æ
FETOMX % 9B 115 10B 94 00 15 10B 94 15 10B 94 155 10B 155 10B 115 10B 115 10B 115 116	4-DCAA	%	126	112	110	4/8	2 8	5	2	8
Price (A) Métalis % 81 119 96 87 71 150 Price (A) Métalis % 80 98 97 102 71 110 Price Right 4005 6005 6005 4005	4.5.6-TCMX	%	88	115	98	\$	8 8	335	3 6	2
Price IVA Mobilish % 80 98 97 102 7 102 7 102 7 102 7 102 7 102 7 102 7 102 7 102 7 102 <t< td=""><td>ည</td><td>%</td><td>ě</td><td>119</td><td>s</td><td>à</td><td></td><td>3</td><td>2</td><td>2</td></t<>	ည	%	ě	119	s	à		3	2	2
Machine Mach	-terphenyl	%	8	8	À	701				
May							300	30.0	30.0	٤
Majora Majora Control Contro	huminum	mg/L	¢0.09	0.074	\$0.05	2002	00.00	3000	3000	4000
May Color	oppour	mg/L	<0.005	<0.005	<0.005	\$0.005	C00.00	50000	2000	300
mg/L c001	Capic	mg/L	c0.01	¢0:01	¢0.01	40.01	\$0.01	1000	1000	
May	The last	Mg∕L	<0.1	¢0.1	601	0,	\$0.1	- G	200	2
mg/L c0 001 c0	military.	mg/L	<0.001	¢0.001	¢0.001	¢0.001	100.00	1000	00.00	38
May	adminus	mg/L	<0.001	c0.001	\$0.001	6 8	\$0.00¥	100.00	2000	0000
mg/L c001	Engle	-1/8m	330	100	<u>\$</u>	26	160	180	200	nez (
mg/L 40.05 40.05 40.05 40.05 40.05 40.05 40.05 mg/L 2.4 6.24 1.6 6.05 40.05 40.05 mg/L 2.4 6.24 1.6 6.05 40.05 40.05 mg/L 6.026 40.05 40.05 40.05 40.05 mg/L 6.032 40.002 40.002 40.002 40.002 mg/L 6.031 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 mg/L 40.01 40.01 40.01 40.01 40.01 40.01 40.02 40.00 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 40.01 4	Droming (lotal)	Λgm	<0.01	<0.01	٥.0٥	40.01	¢0.01	20.03	0.00	5 6
mg/L 0.28 <0.05 <0.05 <0.05 <0.05 <0.05 mg/L 0.028 <0.005 <0.005 <0.005 mg/L 1.00 13 6.2 6.6 5.3 6.7 mg/L 0.028 <0.0002 <0.0002 <0.0002 mg/L 0.039 <0.0002 <0.0002 <0.0002 mg/L 0.031 <0.011 <0.011 <0.011 <0.011 mg/L 0.031 <0.011 <0.011 <0.011 <0.011 mg/L 0.031 <0.011 <0.011 <0.011 <0.011 <0.011 mg/L 0.031 <0.011 <0.011 <0.011 <0.011 <0.011 mg/L 0.031 <0.011 <0.011 <0.011 <0.011 <0.011 mg/L 0.031 <0.011 <0.011 <0.011 <0.011 <0.011 mg/L 0.031 <0.011 <0.011 <0.011 <0.011 <0.011 mg/L 0.001 <0.011 <0.011 <0.011 <0.011 <0.011 mg/L 0.001 <0.001 <0.011 <0.011 <0.011 <0.011 mg/L 0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 mg/L 0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011 <0.011	obalt	mg/L	<0.05	<0.05	<0.05	c0 05	800	2000	90.00	3 6
May	Connec	mg/L	0.28	<0.05	co 05	\$0.05	000	60.	20.02	
May	uou	mg/L	2.4	0.24	9.	88.0	9.00	900	9000	3000
May	peo	₩.	0.028	<0.005	\$000°	60.00	cmny	40.000	2	8
May 0,063 0,005 0,007 0,000 0,007 0,000	Aaonesium	mg/L	110	13	62	3	200	6	370	4600
mg/L c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 0002 c0 001 c0 01	Mannanese	mg/L	0.063	0.035	0.076	0.071	800	0.000	2000	300
mg/L 0,031 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001 <001	April	mo/L	<0.0002	<0.0002	¢0.0002	2000°	<0.0002	20000x	SU UNUK	3000
mg/L 13 5.7 11 13 1.7 14 14 15 17 14 15 17 14 15 17 14 15 17 14 15 17 14 15 17 14 15 17 15 17 15 15 15 15	Walcu y	Aom	0.031	<0.01	<0.01	¢0.01	\$0.01	000	40.01	0.00
Mg/L	W. M. C. C. C. C. C. C. C. C. C. C. C. C. C.	mo/L	13	5.7	=	13	1.7	*	23	23
mg/L < 0.01 < 0.01 < 0.01 < 0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <	Coassium	mod.	¢0.01	¢001	<0.01	<0.01	<0.01	0°0ء	\$0.0	40.01
mg/L 1100 150 660 720 25 400 100 100 660 720 25 400 100	Coremon	Jour Tow	<0.01	¢0.01	<0.01	<0.01	<0.01	¢0.01	\$0.01	400
mg/L <0002 <0002 <0002 <0002 0001 0001	Civer	Joe T	100	8	099	720	25	400	380	3
1000 (000 1000 1000 1000 1000 1000 1000	Soorum	Vom	×0 002	40 00S	<0.002	<0.002	<0.002	0.0021	0.002	<0.002
	naikun.	1/00	1000	400	005	<0.0	<0.01	<0.01	¢0.01	\$ 0.05
, co 1 co 1 co 1 co 1 co 1	Vanadium	700	000	0,0	<0.1	60.	<0.1	<0.1	<0.1	<0.1

i = value is between the Method Detection Limit and the Practical Quantitation Level * « diluted sample, - = not analyzed

APPENDIX D

Universal Engineering Sciences Report for Kennedy Space Center Background Groundwater Well installation (Does not include IDW tracking logs from the report Universal submitted to NASA)



Consultants in: Geotechnical Engineering • Threshold Inspection Environmental Sciences • Construction Materials Testing

November 11, 1998

Offices in

 0.3530 Gainesville

Fort Myers

Rockledge

Tampa

Debary

 St. Augustine Daytona Beach West Palm Beach Jacksonville

National Aeronautics and Space Administration NASA Mail Code JJ-D John F. Kennedy Space Center Kennedy Space Center, Florida 32899

Attention:

Ms. Rosaly Santos-Ebaugh, Remediation Project Manager

Reference:

Summary of Field Work Activities

Basic Ordering Agreement NAS10-12206(B)/DO-5 Installation of 52 Permanent Monitoring Wells

Kennedy Space Center, Florida

Report No.: 98-1124 (35789)

Dear Ms. Santos-Ebaugh:

Universal Engineering Sciences, Inc. (UES) has completed the installation of monitor wells and lithological auger borings in accordance with BOA NAS10-12206(B)/DO-5. The following data is a report of UES's field work activities as completed during August and September 1998.

PROJECT OVERVIEW

The scope of work required the installation of 51 permanent investigation monitor wells and 24 standard penetration test (SPT) borings at 24 individual sites. Based upon NASA requirements for each individual site, UES installed and developed shallow, intermediate, and/or deep monitor wells for supplementary groundwater sampling by NASA and associated agencies. Please recall that the deep well at the T-10K site was canceled, thus a total of 51 wells as opposed to the initial plan of 52 wells. During the development tasks, UES obtained turbidity data from discrete groundwater samples collected from each well at variable time intervals. UES completed the standard penetration test borings to provide a lithological profile of the subsurface conditions at each project site. Table 1 presents a summary of each project site in terms of well depths, boring depths, and final well completion.

SUMMARY DOCUMENT CONTENTS

For each individual site, UES has prepared the following representative documents:

- A Boring Log depicting the subsurface lithology at each site. 1)
- A Well Completion Log depicting the construction details of each monitor well. 2)
- A Standard Well Development Form which outlines well development data such as 3) turbidity, groundwater appearance, and estimated gallons removed for each monitor well.
- A Drum Inventory Log which documents information regarding the number, type, contents, 4) and storage location of drums containing IDW derived at each project location.

Ms. Rosaly Santos-Ebaugh, Remediation Project Manager Summary of Field Work Activities Installation of 52 Permanent Monitoring Wells Kennedy Space Center, Florida Report No.: 98-1124 (35789)

Page 4

The aforementioned documents have been separated by site location and arranged as an Exhibit for preferential access. The Exhibit contents are as follows:

EXHIBIT	GRID	STUDY LOCATION
Exhibit I:	A-1	North of Canaveral National Seashore, Parking Area 4-5
Exhibit II:	A-2	Playalinda Beach: Parking Lots 1,2,5
Exhibit III:	A-3	Playalinda Beach: Parking Lots 1,2,5
Exhibit IV:	A-4	Dunes East of LC-41
Exhibit V:	A-5	North of Corrosion Test Site (Solar Panel)
Exhibit VI:	A-6	North of Pump Station 7 Road, west of Cape Road
Exhibit VII:	B-1	North of Haulover/Shiloh, Shiloh Scrub Restoration Site
Exhibit VIII:	B-2	Happy Creek: north or south of Camera Pad
Exhibit IX:	B-3	Happy Creek: north or south of Camera Pad
Exhibit X:	B-4	Jerome Road - East of SR3, Apiary Site #67
Exhibit XI:	B-5	Ranson Road, Apiary Site #10
Exhibit XII:	B-6	Schwartz Road north along firebreak
Exhibit XIII:	C-1	PAPI Lights, Apiary Site #47
Exhibit XIV:	C-2	Schwartz Road, West of SR3
Exhibit XV:	C-3	Roberts Road, West of SR3
Exhibit XVI:	C-4	Jerome Road, West of SR3
Exhibit XVII:	C-5	Shuttle Landing Facility
Exhibit XVIII:	C-6	Dummit tower
Exhibit XIX:	D-1	Shiloh Turnbull Creek Drainage
Exhibit XX:	D-2	Shiloh 1 Impoundment
Exhibit XXI:	D-3	T-10K area near tower
Exhibit XXII:	D-4	Playalinda - Mosquito Lagoon by 1st Camera Pad
Exhibit XXIII:	D-5	Banana Creek West of SR3
Exhibit XXIV:	D-6	T-24/T-16 dike road

PERMANENT INVESTIGATION MONITOR WELLS

MONITOR WELL INSTALLATION

UES installed the monitor wells using truck-mounted drill rigs and hollow-stem auger (6.25-inch I.D.) assemblies. UES steam-cleaned and decontaminated the individual auger assemblies and associated tools prior to their use at each monitor well location. UES completed the cleaning and decontamination tasks in accordance with NASA's Program Wide Sampling and Analysis Plan (PWSAP) and UES's Florida Department of Environmental Protection (FDEP) approved Comprehensive Quality Assurance Plan (CQAP) No. 860101G. Investigation Derived Wastes (IDW) generated during UES's fieldwork were collected in steel drums: fluids generated during the decontamination and/or development process were stored in closed-head 55-gallon drums; and, soils generated during drilling were stored in open-head 55-gallon drums. The drums used for this project were selected to be in compliance with Performance Oriented Packaging Standards (POP) and 49 CFR 173.24a, 178.504, 178.509, and 178.522 regulations. Upon completion of work at each project site, all drums were appropriately labeled, positioned on spill pallets, covered, and strapped for security. As a final step, UES labeled each spill pallet.



Ms. Rosaly Santos-Ebaugh, Remediation Project Manager Summary of Field Work Activities Installation of 52 Permanent Monitoring Wells Kennedy Space Center, Florida Report No.: 98-1124 (35789) Page 5

MONITOR WELL SPECIFICATIONS AND CONSTRUCTION

The intent of the monitor well installation tasks and construction specifications was to provide low turbidity well water for sampling organic compounds and inorganic metals. To address this requirement, UES constructed the monitor wells implementing 30/65 fine sand as a filter pack medium and utilized No. 6 (0.006-inch) slotted well screen to accommodate the 30/65 fine sand. Each monitor well is constructed with standard two-inch diameter Triloc PVC casing, screen, and well points. The well screen is 10 feet in length for water table investigation wells (shallow monitor wells) and intermediate monitor wells, and the well screen is five feet in length for the deep monitor wells. The final monitor well depths for the shallow monitor wells were based on observed water table depths at the time of well installation. An approximate two-feet section of slotted screen remains above the groundwater table (where feasible) to allow for seasonal fluctuations in the water table.

UES installed a 30/65 grade silica sand filter pack in the annular space between the well screen and the borehole walls, and extended the filter pack to height approximately two feet above the top of the well screen. An additional one-half to one foot of bentonite was added to serve as an effective seal above the well screen where applicable. The remaining annular space was grouted with neat Portland Cement. Each well was protected with a flush-mounted liquid-tight protective manhole cover or a steel above grade protective sleeve.

STANDARD PENETRATION TEST BORING INSTALLATION

To assess the subsurface lithology at each project location, UES installed one standard penetration test boring (SPT) utilizing a split spoon sampler to the depth of the deepest well slated for that corresponding site; therefore, one SPT boring is representative of all wells installed at a given site. The SPT boring designation directly corresponds to an individual project site.

WELL DEVELOPMENT AND TURBIDITY ASSESSMENT

WELL DEVELOPMENT

UES's primary objective when developing a monitor well is to provide groundwater which is suitable for sample collection and laboratory analysis. Well development creates a stable, effective filter zone in the annular space of the well boring, removes particulate matter from the filter media, increases the hydraulic conductivity of the aquifer surrounding the well filter pack, and reduces the turbidity of groundwater infiltrating the well screen.

UES developed each monitor well installed for DO-5 using a hydraulic powered rotary pump and 1.25-inch tremie pipe equipped with a foot valve. The monitor wells were developed prior to final completion of the upper well sections (cement grout, concrete pad, protective measures).

TURBIDITY ASSESSMENT

To assess the turbidity of the groundwater being removed from each monitor well, UES utilized a HF Scientific, Incorporated, DRT-15CE Turbidimeter. The DRT-15CE operates through three

Ms. Rosaly Santos-Ebaugh, Remediation Project Manager Summary of Field Work Activities Installation of 52 Permanent Monitoring Wells Kennedy Space Center, Florida Report No.: 98-1124 (35789) Page 6

sensitivity ranges: 0 - 10 nephelometric turbidity units (NTUs), 0 - 100 NTUs, and 0 - 1,000 NTUs. The referenced turbidimeter functions via the Nephelometric method where the intensity of light scattered by the groundwater sample is compared to the intensity of light scattered by a calibrated reference sample. UES utilized a 0.02 NTU reference sample to calibrate the DRT-15CE device at each monitor well location.

UES developed each monitor well until a DRT-15CE response of approximately 20 NTUs or less was achieved. When groundwater samples are to be analyzed for chlorinated solvent compounds and inorganic metals, the FDEP Waste Cleanup Section suggests that groundwater wells be developed until a turbidity level of 20 NTUs or less is obtained.

TURBIDITY ASSESSMENT - COMMENTARY

Six monitor wells, DuneSwale-IW-4S (B-4), DuneSwale-IW-5S (B-5), DuneSwale-IW-6S (B-6), Western-IW-1S and 1I (C-1), and Western-IW-6S (C-6), presented development complications based on the lithological formations present in the screened interval (see Boring Logs). The fine sand and silty conditions surrounding the well screens resulted in turbid effluent groundwater during the initial development period.

CONCLUSIONS

UES recommends that all investigation monitor wells associated with this report be purged prior to sampling with a peristaltic pump; this technique will limit the disturbance within the well screen and may assist in limiting turbidity conditions sometimes initiated by purging. Groundwater samples slated to be analyzed for inorganic metals should be collected using a peristaltic pump, as well.

UES appreciates this opportunity to offer environmental consulting services to the National Aeronautics and Space Administration, and we are looking forward to our continued association. Please contact either of the undersigned should you have questions.

Respectfully submitted,

Universal Engineering Sciences, Inc.

Melissa A. Albright Project Geologist

Ronald V. Sanzi

NASA Contract Manager

MAA/RVS:tlk

Attachments: Exhibit I through Exhibit XXIV



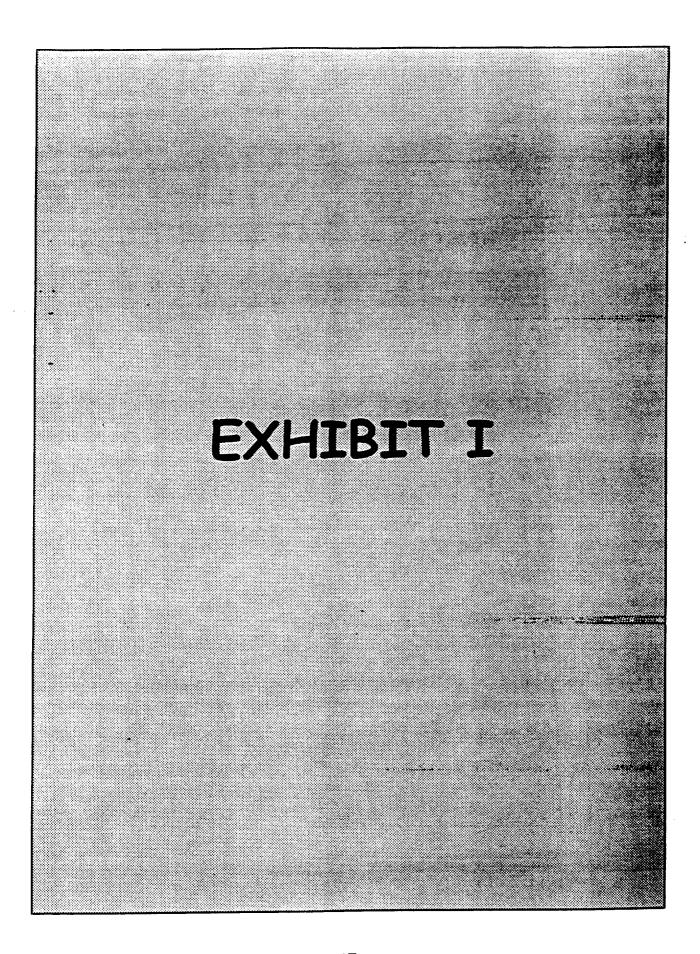
TABLE 1 Summary of Site Work and Estimated Units NASA DO-5

_		6	5	4	ω	2			თ	5	4	ω	2				8
West of SR3, road to PAPI Lights, Apiary Site #47.		Schwartz Road north along firebreak.	Ranson Road: E of SR3, Apiary Site #10.	Southern Ridge Jerome Road - E of SR3, Apiary Site #67	Happy Creek: north or south of Camera Pad.	Happy Creek: north or south of Camera Pad.	North of Haulover/Shiloh, Shiloh Scrub Restoration Site.		North of Pump Station 7 Road, west of Cape Road.	North of Corrosion Test (Solar Panel) site East of Road along dune ridge.	Dunes East of LC41.	North of KSC Security Zone/Playalinda Beach: shallow - several choices in disturbed areas near Parking Lots 1,2,5.	North of KSC Security Zone/Playalinda Beach: all three - several choices in disturbed areas near Parking Lots 1,2,5.	North end of Canaveral National Seashore, Parking Area 4-5.		•	BMVN BIS T MBLSASBIR
																	GRE
										•						BC45	RIG RIG
15		15	15	15	15	15	15		15	15	15	15	15	15		SHAL	
35	C. Wester	0	35	35	0	35	35	B. Dune-	35	35	0	0	35	35	A. Dui	WT	WELL
50	C. Western Plain Subsystem	0	0	50	0	50	50	B. Dune-Swale Subsystem	50	0	0	0	50	50	A. Dune Subsystem	DEEP	
100	system	15	50	100	15	100	100	ystem	100	50	15	15	100	100	ä		TOTAL FOOT.
50		15	35	50	15	50	50		50	35	15	15	50	50		mo>⊣c	07 \$
0		1	2	0	1	ω	3		ω	2		-	ω	3		2>5	WELL PROTECTION
ω		0	0	ω	0	0	0		0	0	0	0	0	0		Å G	CTION
9		3	9	9	3	9	9		ဖ	თ	ω	ω	9	6		402>	rr0#
7		2	4	7	2	7	7		7	4	2	2	7	7		L HQUID#I	IDW DRUMS
10		2	5	10	2	10	10		10	υ'n	2	2	10	10		SOLID	UMS
5		-1	2	5	_	5	U1		5	2	-		თ	СI		sp	SPILL PALLETS
5		_	2	5	_1	5	5		5	2	_	_	5	Ŋ		ŞTR	APS APS
								<u> </u>	28	5							

285

TABLE 1 Summary of Site Work and Estimated Units NASA DO-5

					1	-		ł									
	79	79	156	118	- 5	23 15 9	29		 	1,470	550	560	360			DRILLING UNIT REQUIREMENTS	
										Units	Tabulation Of Units	Tabula					
	_	-	2	2	3	1	0	5	15	15	c	-	ā				
!	5	5	10	-	- "		\	+			,		<u>ה</u>			T-24/T-16 dike road	6
!				,	<u>'</u>	•	ω		0 50	100	50	35	15			S GAG.	
	5	5	5	4	<u>ი</u>	2	0	35		50	0	35	15			Banana Creek West of SD3	5
	Ch	ۍ.	10	7	9	ω	0	1 8	_	-	1					Playalinda - Mosquito Lagoon by 1st Camera	4
		_	,		+	\bot	,	1	\downarrow	3		ж	15	<u> </u>		1-10K area near tower: all three	10
			ر د	2	<u>ω</u>	_	0	15		15	0	0	15				ა
	5	۲h	0			,		_	_	+	1	T				Shiloh 1 Impoundment	2
286			,	7	٥	ω	<u> </u>	50	100		50	35	15			or usa, wouth of US	
								-		ysiem	or mai an aubsystem			1		Shiloh Turnbull Creek Drainage, Road West	
·			2	^	<u> </u>	Ľ		L		- 1	+		-	-			1
		T,		,	<u> </u>		<u>.</u>	5	ゔ 		0	0	15		· · · · · · · · · · · · · · · · · · ·	tower.	
	J ,	3	5	4	6	0	2	35	50	(F)	0	35	ชี	+		West of SR3. North of SR406	6
	.n	5	10	7	9	ယ	0	50	8	+	g	ű	1 2	-	1	West side of SLF near tower.	55
	U	υ ₁	10	,	9	,	'		\bot	+		#	1,	+		Jerome Road, West of SR3, South side	4
	,				<u> </u>	s	>	50	00	_	50	35	15			Roberts Road, West of SR3, South side.	0
	_	-	2	2	ω		0	15	15			0	15			road along ditch, some cutting required.	ا د
								m		-						Schwartz Road, West of SR3, South side	2
~	STR	ş	anos	Sainer	.02	- 00>	z>z	ด≻⊣		4	DEEP	ž	R SHAL	BCA		-	
	& STRAPS	\$ \$1				Š		00.		1	1	1		7 7 8 8 8			
	PALLETS	SPILL	IDW DRUMS	d MGI	0.5		WELL ,	" <u> </u>	FOOTAL TOTAL	 של	OM T	FOOTAGE	1 7		GRID	& SITE NAME	
																	š





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

REPORT NO PAGE B-2 1

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION:

SECTION: TOWNSHIP:

A-1

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/21/98

WATER TABLE (ft): 0.0

DATE FINISHED:

8/21/98

DATE OF READING: 08/21/98

DRILLED BY:

UES - ORLANDO

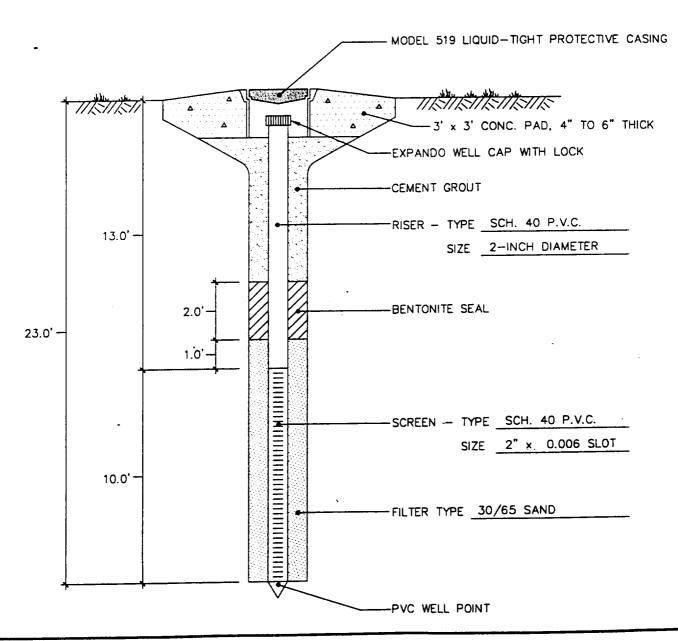
EST. W.S.W.T. (ft):

	S		1	_	S	EST. W.S.W.	Г. (ft):		YPE OF	SAMPLI	NG: ASTN	D-1586
(FT.)	A M P L	BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	w.t.	S Y M B	DESCRIPTION	-200 (%)	MC (%)		RBERG MITS	K (FT./	ORG CONT
	E				L		(,,,,	\ \~'	LL	PI	DAY)	(%)
0 -	+					Tan SAND; with shell [SP]					<u> </u>	
]					\cdots	Tan Onto, With Shell [SP]	1					
4					Δ Δ	Loose, SHELL; with trace of fine sand						İ
5 —	X 1.	2-3-4	7		ΔΔ	Loose, SHELL; with trace of fine sand						
	₹	4-4-4	8		Δ Δ Δ Δ							
→	{	4-4-6	10		Δ 4						į	
5	₹	4-5-7	12	- 1	Δ Δ Δ Δ							'
10	Ī		·····		ΔΔ	- medium dense						
4					4							
1	1	_		ľ			1				ĺ	
15-	44	8-11-13	24		ام ۵							
		1			Δ Δ Δ Δ					· · · · · · · · · · · · · · · · · · ·	••••••	
4_			1		4							
20	1	8-8-12	20									
					Δ Δ Δ							
		1			4					1	1	
- \	,	-11-17	20			•						
25		-1.1.1.7	28		7 7							
					4							
					Δ Δ							
30 — 📉	25	5-28-33	61	مإ	الم						1	
<u> </u>					$\therefore $	Dense, tan fine SAND; with shell [SP]			1			
				<u> </u>								
35	14	-21-18	39									
, TI												
_					$ \cdot $							
- ₩	16	16.00										
10 +	10	-16-22	38		· <u>·</u>							
7 /					$\cdot \cdot $							
+		1.							1		1	
15	20-	25-27	52		<u>:]</u>							
			- 1	[:·:	$\cdot \cdot $	`						
11			ĺ	1:	$: \mid$		-					į
X	5	-4-7 1	1	· . · Δ		CUEL					1	
0	 .			┈┝╼		oose, tan SHELL; with sand ORING TERMINATED AT 50 FEET	·····					
4						A SOLE						
1												
5 —											-	



PROJECT NO.	18467-005-01
REPORT NO	
PAGE:	

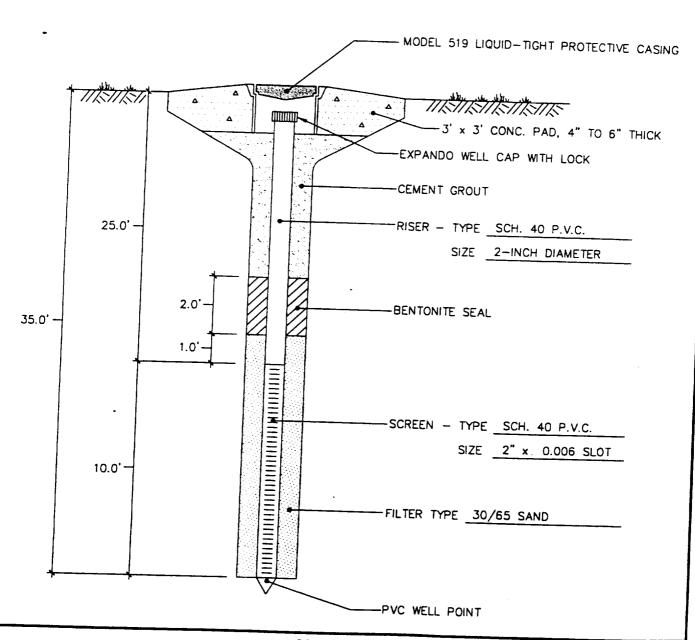
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/26/98
	BER: DUNE-IW-1S LOCATION: A-1	
	DY. UES - DAYTONA BEACH	





2	ROUEDT NO.	18467-005-01
R:	EPORT NO.:	
P	AGE:	

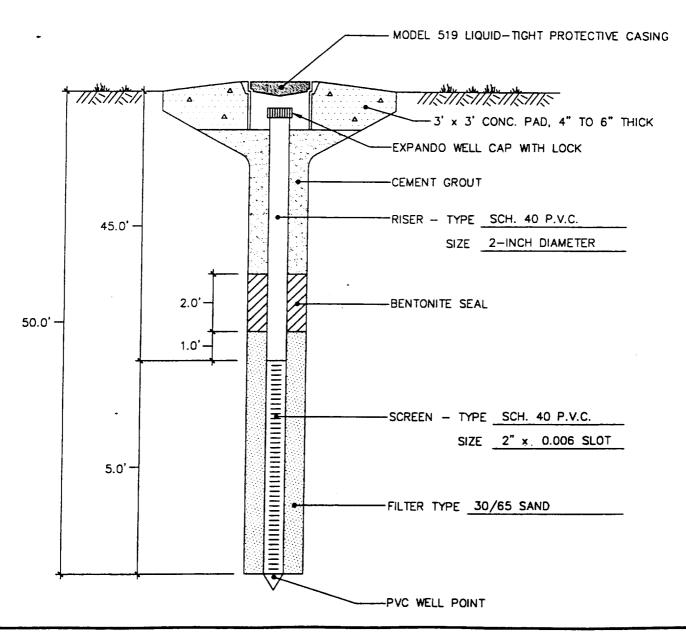
PROJECT: INSTALLATION 52 MONITOR WELLS	
KENNEDY SPACE CENTER, FLORIDA	
CLIENT: NASA JJ-D	DATE: <u>08/26/98</u>
WELL NUMBER: DUNE-IW-11 LOCATION: A-1	
INSTALLED BY: UES - DAYTONA BEACH	
	





PROJECT NO.: 18467-005-01
REPORT NO.:
PAGE:

PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
CLIENT: _	NASA JJ-D	DATE: 08/27/98	
WELL NUME	BER: DUNE-IW-1D LOCATION: A-1		





STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

	, KOJEC	INFORMATION	
Contract:	NAS10-12206(B)/DO-5/TD- 002	Client Name:	NASA
Reference:	Monitor Well Installation	Department:	JJ-D
Project No.:	18467-005-01	Client Contact:	Panely Court - m
Well Location:	A-1	Client Phone:	Rosaly Santos-Ebaugh, RPM
Date:	8-26-98		(407) 867-4121
		Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-1S	Total Depth:	23.0'
Screen Interval:	13.0 to 23.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad	Development Method:	Rotary Pump, Tremie Pipe

WELL DEVELOPMENT DATA			
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE	
1:40 p.m.	200+	Murky	
	87.3	Murky	
	51.2	Cloudy	
	37.5	Clearer	
	15.53	Clear	
1:55 p.m.	11.15	Clear	
Total Development Time:	15 minutes		
Final Turbidity Reading:	11.15 NTUs		
Estimated Volume Removed:	55 gallons		
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):	
		OVA-PID):	

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	A-1	Client Phone:	(407) 867-4121
Date:	8-26-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-1I	Total Depth:	35.0'
Screen Interval:	25.0 to 35.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
2:45 p.m.	200+	Murky
	185.3	Murky
	76.9	Cloudy
	37.2	Cloudy
	29.9	Clearer
	21.3	Clear
3:10 p.m.	8.9	Clear
Total Development Time:	25 minutes	
Final Turbidity Reading:	8.9 NTUs	
Estimated Volume Removed	: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
	-	

STANDARD WELL DEVELOPMENT LOG

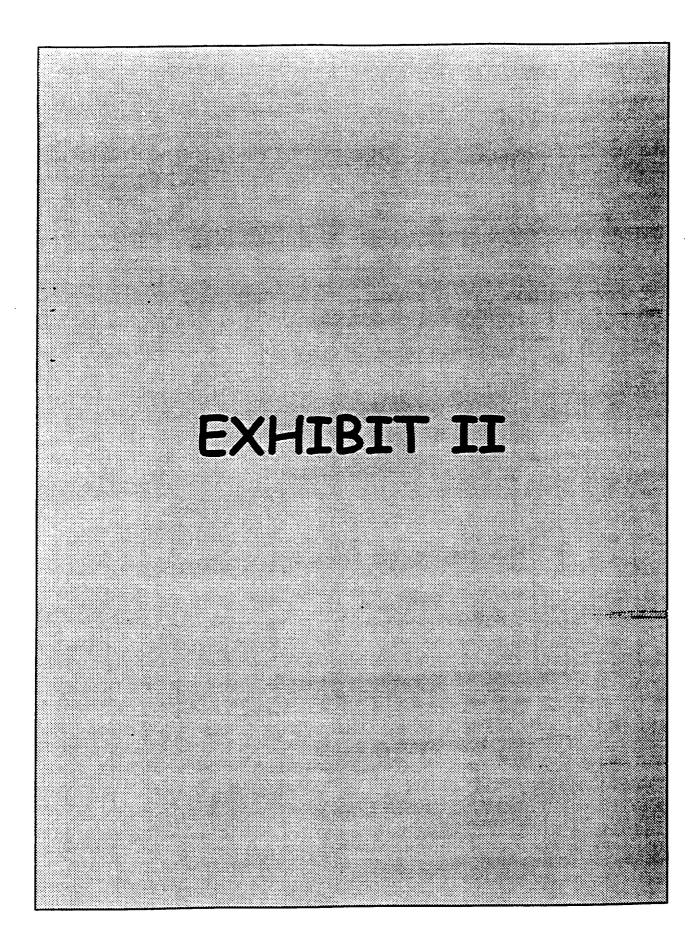
PROJECT INFORMATION

	PROJEC	TINFORMATION	
Contract:	NAS10-12206(B)/DO-5/TD- 002	Client Name:	NASA
Reference:	Monitor Well Installation	Department:	JJ-D
Project No.:	18467-005-01	Client Contact:	Pagetty 9
Well Location:	A-1	Client Phone:	Rosaly Santos-Ebaugh, RPM
Date:	8-27-98	Client Fax:	(407) 867-4121
		Cheffit Pax:	(407) 867-8040

MONITOR WELL DATA

Well ID:		WELL DATA	
-	Dune-IW-1D	Total Depth:	50.0' bis
Screen Interval:	45.0 to 50.0'	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad	Development Method:	Rotary Pump, Tremie Pipe

	WELL DE	VELOPMENT DATA
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
1:20 p.m.	200+	THE APPEARANCE
	137.3	
	82.4	
	54.6	
1:35 p.m.	14.23	
Total Development Time:	15 minutes	
Final Turbidity Reading:	14.23 NTUs	
Estimated Volume Removed:	55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
		COVA-PID):
	2	94





UNIVERSAL ENGINEERING SCIENCES BORING LOG

PROJECT NO 184-100501 REPORT NO PAGE 8.2.2

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION:

AS SPECIFIED BY CLIENT

REMARKS:

HAND AUGER FROM 0 FEET TO 4 FEET

BORING DESIGNATION:

SECTION:

A-2

TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/7/98

WATER TABLE (ft): 5.6

DATE FINISHED:

8/7/98

DATE OF READING: 08/07/98

DRILLED BY:

UES - D.B.

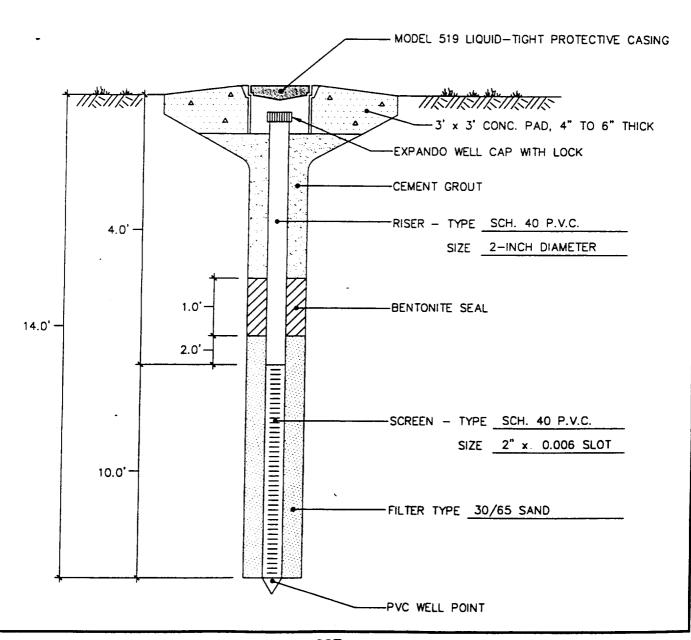
	S A BLOW	/s	AI.		S	EST. W.S.W.T	(ft):		T	YPE OF	SAMPLIN	G: ASTN	· D.B. • D-158
DEPTH (FT.)	P PER (L INCREM E	B' (BL	N OWS/ T.)	W.T.		DESCRIPTION		200 %)	MC (%)		RBERG MITS	K (FT./	OR COI
0										LL	PI	DAY)	19
						Gray-brown fine SAND [SP]		-+		 			
4			ĺ		· · .	;}		-					
5 —	4-5-7	1.	_		• • • •	gray dark gray							
·	5-6-6			_		gray							
太	4-8-9		- 1]:		Medium dense, light brown fine SAND; with shell [SP]	1				1		
- 1	3-4-3	- 1		-	•	loose						1	
10	3-4-3		······.			Dark gray fine SAND; with shell [SP-SM]	1				Ī	1	
7	1			ŀ			1						
1	1			ļ.		dark gray sand; with silt				ĺ		ı	
15 ———	2-3-4	7				dark gray sand; with silt							
_	1				$::\cdot $		 						
20-1	9-10-15	25	1	- :		Medium dense, gray SHELL; and sand trace silt	1						
												-	

\	12-21-23			· :	$\cdot \cdot \cdot \mid$				1			1	
5	12-21-23	44		····/		Dense dark con 6. Carre							
4		1				Dense, dark gray fine SAND; shell; trace silt [SP]							•••••
1										1			
) - 	14-17-17	34											
11						dark gray sand; trace shell	• • • • • • • • • • • • • • • • • • • •						
-												- 1	
14	11-14-15	29		·									
4						medium dense, with trace of shell and silt							
11					s	Medium dense, dark gray fine SAND; with							
- X	7-8-8	16		· : · :	1								
-		10	ļ	- 4	2 L	oose, dark gray SHELL; with some sand					1		
4				4		nd silt		1					•••••
\mathbb{H}									•				
-11	3-4-4	8		Δ.	4				-				
11				: : :	De	ense, dark gray fine SAND; trace silt and sell fragments [SP]	**********	ļ				<u> </u>	
-				·			į						
<u> 1</u>	2-15-17	32			1						1		
 					BC	PRING TERMINATED AT 50 FEET							
1													
11	- 1								1				
					1		ı		ı	1	1	1	- 1



PROJECT NO.: 1	8467-005-0
REPORT NO.:	
PAGE:	

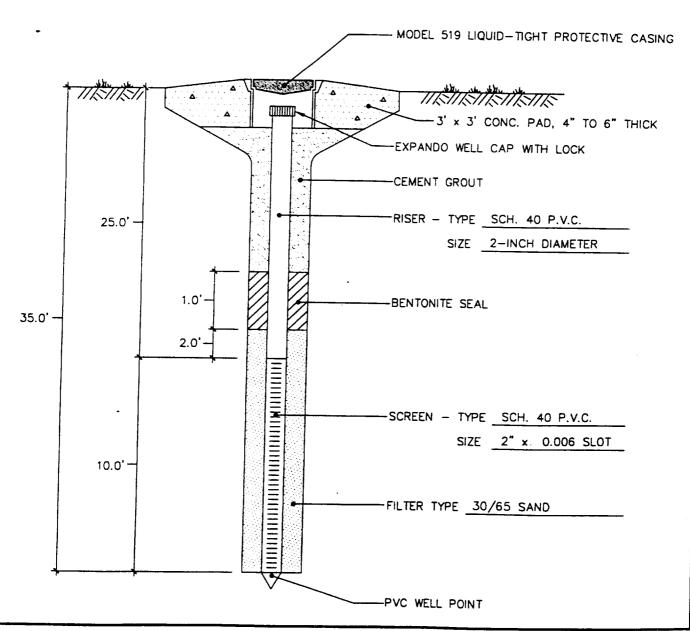
PROJECT:	INSTALLATION 52 MONITOR WELLS						
	KENNEDY SPACE CENTER, FLORIDA						
CLIENT: _	NASA JJ-D	DATE: 08/13/98					
WELL NUME	ER: DUNE-IW-2S LOCATION: A-2						
INSTALLED	BY: UES - DAYTONA BEACH						





PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

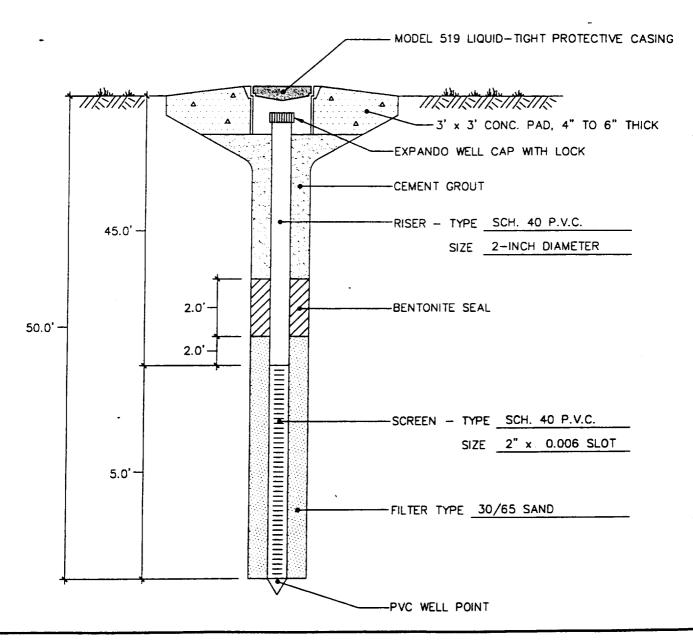
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/13/98
WELL NUMB	ER: DUNE-IW-2I LOCATION: A-2	
INSTALLED	BY: UES - DAYTONA BEACH	





PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS						
	KENNEDY SPACE CENTER, FLORIDA						
CLIENT: _	NASA JJ-D	DATE: 08/10/98					
WELL NUME	BER: DUNE-IW-2D LOCATION: A-2						
INSTALLED	BY: UES - ORLANDO						



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

	rkojet	INFORMATION	
Contract:	NAS10-12206(B)/DO-5/TD- 002	Client Name:	NASA
Reference:	Monitor Well Installation	Department:	JJ-D
Project No.:	18467-005-01	Client Contact:	Pagely Court
Well Location:	A-2	Client Phone:	Rosaly Santos-Ebaugh, RPM
Date:	9.44.00	Gildin Filolie.	(407) 867-4121
	8-14-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:		WELL DATA	
Well ID.	Dune-IW-2S	Total Depth:	14.0' bis
Screen Interval: -	4.0 to 14.0'	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad	Development Method:	Rotary Pump, Tremie Pipe

	WELL DEVE	ELOPMENT DATA
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
7:35 a.m.	200+	Murky/Cloudy
	137.6	Cloudy
	83.4	Clearer
	39.7	Clear
7:55 a.m.	10.1	Very clear
Total Development Time:	20 minutes	
Final Turbidity Reading:	10.1 NTUs	
Estimated Volume Removed:	50 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
		COVA-PID):
	_	

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	A-2	Client Phone:	(407) 867-4121
Date:	8-13-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-2I	Total Depth:	35.0'
Screen Interval:	25.0 to 35.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	LOFMENT DATA
111112	TORBIDITY (NTU)	GROUNDWATER APPEARANCE
2:20 p.m.	200+	Murky/Cloudy
	135.7	
	85.6	
	42.3	Clearer
2:35 p.m.	18.0	Clear
Total Development Time:	15 minutes	
Final Turbidity Reading:	18.0 NTUs	
Estimated Volume Removed	d: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

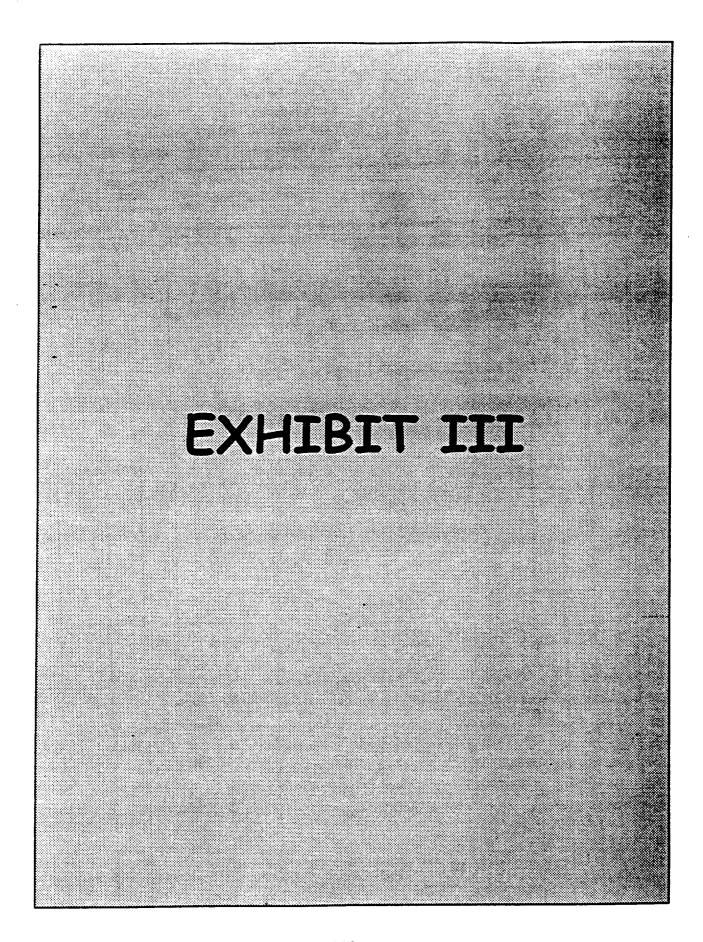
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	A-2	Client Phone:	(407) 867-4121
Date:	8-10-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-2D	Total Depth:	50.0'
Screen Interval <u>:</u>	45.0 to 50.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad	Development Method:	Rotary Pump, Tremie Pipe

TURBIDITY (NTU)	GROUNDWATER APPEARANCE
1,000+	
418	
61	
10.6	
19 minutes	
10.6 NTUs	
55 gallons	
	SOIL SCREENING RESULTS (OVA-PID):
	1,000+ 418 61 10.6 19 minutes 10.6 NTUs





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 13467-005-01 REPORT NO. PAGE B-2.3

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION:

AS SPECIFIED BY CLIENT

REMARKS:

HAND AUGER FROM 0 FEET TO 4 FEET

BORING DESIGNATION:

A-3 TOWNSHIP: SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

SECTION:

DATE STARTED:

8/7/98

WATER TABLE (ft): 5.8

DATE FINISHED:

8/7/98

DATE OF READING: 08/07/98

DRILLED BY:

UES - D.B.

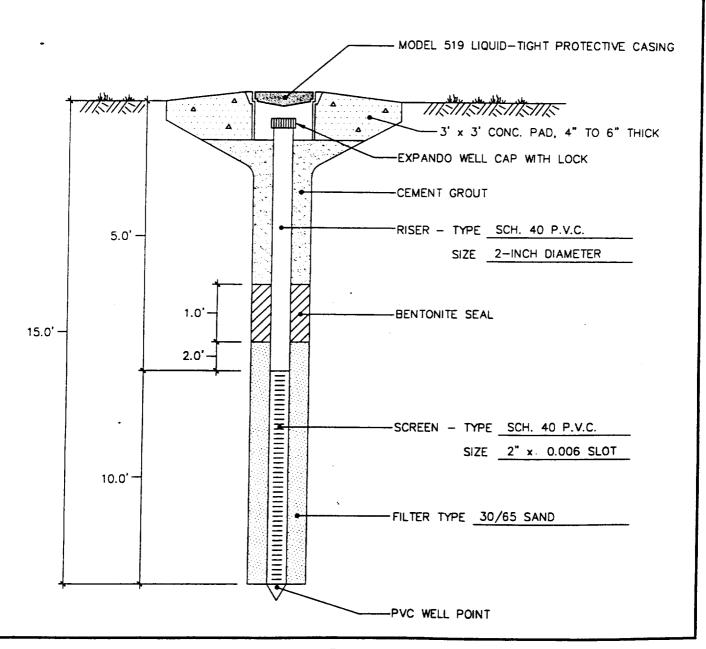
FST WSWT #1

	ſs	T	T		S	EST. W.S.W.T.	(ft):	TYPE	OF SA	MPLIN	IG: ASTI	M D-1586
DEPTI (FT.)	H MPLE	BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	W.T.	SYMBOL	DESCRIPTION	-20 (%		LIMI	BERG I'S	K (FT./ DAY)	ORG. CONT. (%)
0	+					Light brown fine SAND [SP]						\pm
	4				• • • •							
5 -	X	4-5-7	1.2			brown Medium dense, light brown-brown fine	4					
	$\frac{1}{4}$	4-4-6	10			SAND; trace shell Loose, brown fine SAND; with shell [SP]	7					
10 -	\Re	3-4-2 - 2-2-5	6 7	ļ		tan						
10-	\prod					Medium dense, dark gray fine SAND; with silt [SP-SM]	-	 			••••••	
	\forall			ŀ								
15 -	H	3-5-8	13			BORING TERMINATED AT 15 FEET		 				•••••
20 —												
-												•••••••••••
25 												
-								 				
-		1										
30						-		 				
=												
35							••••••					
]												
40			-									
1							•••••••	 ļ	ļ			
=												ĺ
45 —							••••••	 				
=												
50								İ				
†									• • • • • • • • • • • • • • • • • • • •	1		
55 —					<u> </u>			 	·····			



PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/10/98
WELL NUME	BER: DUNE-IW-3S LOCATION: A-3	
INSTALLED	BY: UES - DAYTONA BEACH	



STANDARD WELL DEVELOPMENT LOG

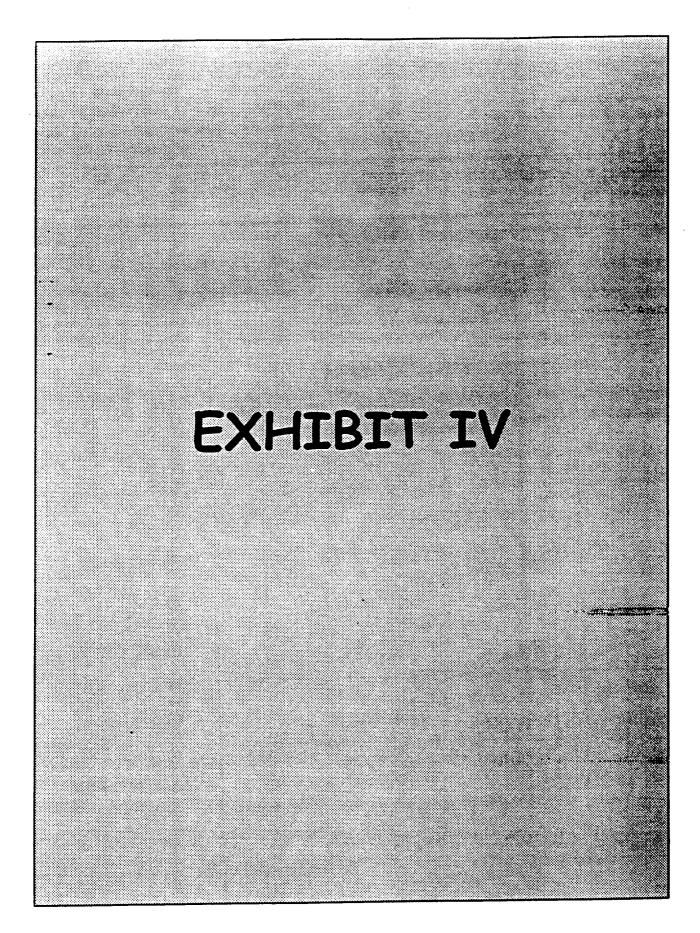
PROJECT INFORMATION

Contract:	NAS10-12206(B)/DO-5/TD-	Client Name:	NASA
Reference:	002 Monitor Well Installation	Department:	JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	A-3	Client Phone:	(407) 867-4121
Date:	8-10-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID;	Dune-IW-3S		
	Dulle-144-22	Total Depth:	15.0'
Screen Interval:	5.0 to 15.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad	Development Method:	Rotary Pump, Tremie Pipe

	WELL DEVELOPMENT DATA							
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE						
9:00 a.m.	200+	Murky/Cloudy						
9:04 a.m.	86.5	Cloudy						
9:09 a.m.	43.1	Cloudy/Milky						
9:12 a.m.	28.7	Clearer						
9:15 a.m.	11.3	Clear						
Total Development Time:	15 minutes							
Final Turbidity Reading:	11.3 NTUs							
Estimated Volume Removed:	55 gallons							
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):						
		OVA-PID):						





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 18467-005-01 REPORT NO PAGE. B-2.4

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION:

SECTION:

A-4 TOWNSHIP: SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/24/98

WATER TABLE (ft): 9.0

DATE FINISHED:

8/24/98

DATE OF READING: 08/24/98

DRILLED BY:

UES - ROCKLEDGE

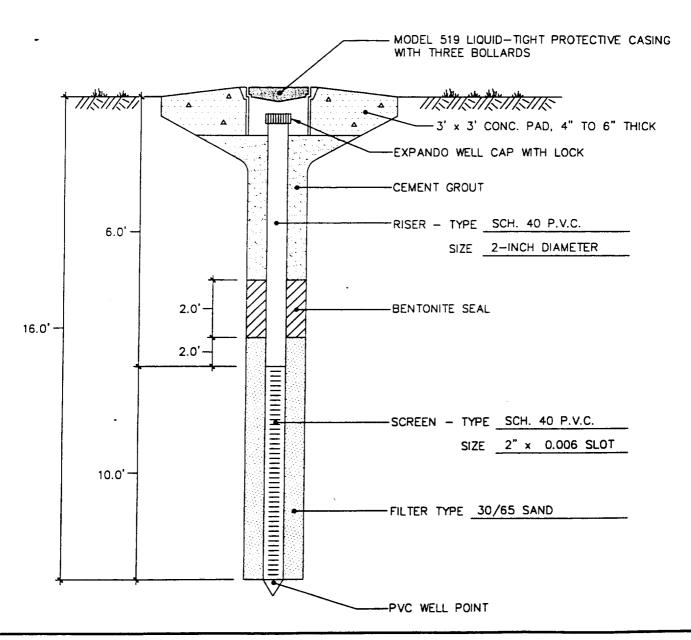
EST. W.S.W.T. (ft):

	S		T	т-	S	EST. W.S.W	/.T. (ft):		TYPE	OF SA	MPLIN	G: ASTA	1 D-1586
DEPTH (FT.)	SAMPLE	BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	W.T.	0	DESCRIPTION	-20 (%			TTER8 LIMIT	ERG 'S	K (FT./	ORG
0 —	Ī		 	 	+-				L	L	PI	DAY)	(%)
-	П					Light brown fine SAND; with shell [SP]			_	\perp			
7	1				 				İ	- 1	1		
7		٠	·					l	l				
5	X	2-2-2	4			very loose				1			
7	Ä	4-6-7	13			medium dense					1	**************	•••••
	Ä	7-7-7	14	_									
10	쟉	6-7-10	17	<u> </u>				İ					
-			1			Dense, light brown fine SAND [SP]				····-			
4	1			ŀ	\cdots	•							
15	X	12-18-21	39	-				ľ				ĺ	
13 7				·····		BORING TERMINATED AT 15 FEET							
					1								
7			1										
20 —	+											ļ	
7													•••••
-			- 1				1		1				
25													
1			1		- 1								
7			- 1										
٠,		1	- 1			-						1	
7	Ī		1			·							
_									i			ĺ	
4				-									
5	ļ												
71													••••••
+ 1							1					- 1	
,]			-							İ			
41			- 1	-				†					
71												1	
. + 1			ı		1							-	
7							ļ	ļ		•	ļ		
4					1								
1											1	1	
													ľ
1										••••••	1		
4													ı
11						j			-				
											ļ		



PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	-

PROJECT:	INSTALLATION 52 MONITOR WELLS							
	KENNEDY SPACE CENTER, FLORIDA							
CLIENT: _	NASA JJ-D	DATE: <u>09/01/98</u>						
WELL NUME	BER: DUNE-IW-4S LOCATION: A-4							
INSTALLED	BY: UES - ROCKLEDGE							



STANDARD WELL DEVELOPMENT LOG

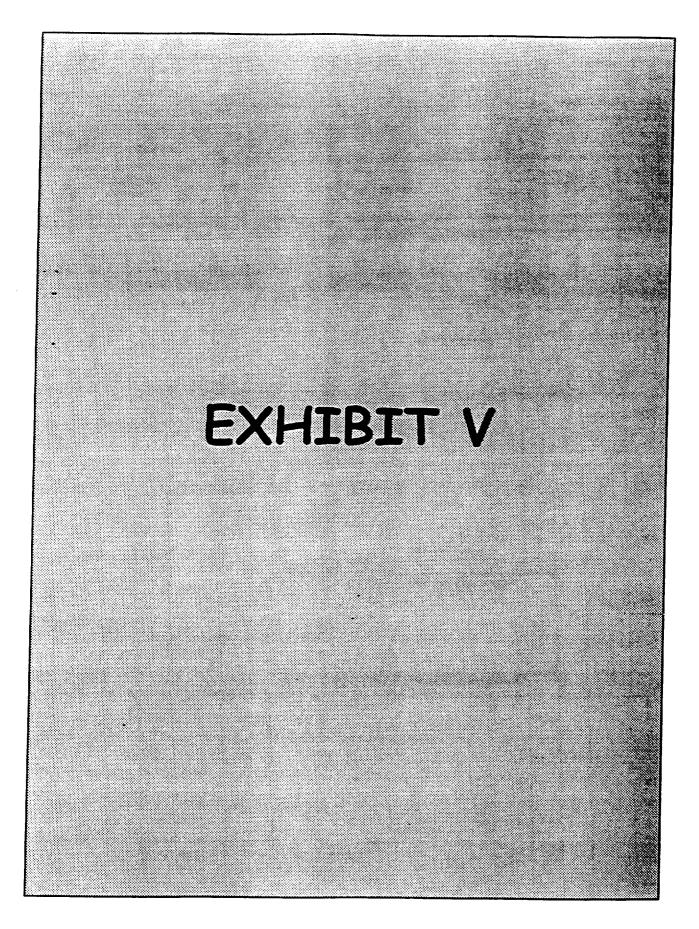
PROJECT INFORMATION

Contract:	NAS10-12206(B)/DO-5/TD-	Client Name:	NASA
Reference:	002 Monitor Well Installation	Department:	JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	A-4	Client Phone:	(407) 867-4121
Date:	9-1-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-4S	Total Depth:	16.0'
Screen Interval:	6.0 to 16.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

	TILL DEVE	LOPMENT DATA
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
12:40 p.m.	2.3	Clear
1:10 p.m.	2.4	
Total Development Time:	30 minutes	
Final Turbidity Reading:	2.4 NTUs	
Estimated Volume Removed:	55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
		COLE CONCENTION RESULTS (OVA-PID):
		10
	٠ 3	10





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 18467-005-01 REPORT NO PAGE B-2.5

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION:

A-5 TOWNSHIP: SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

SECTION:

DATE STARTED:

8/24/98

WATER TABLE (ft): 8.0

DATE FINISHED:

8/24/98

DATE OF READING: 08/24/98

DRILLED BY:

UES - ROCKLEDGE

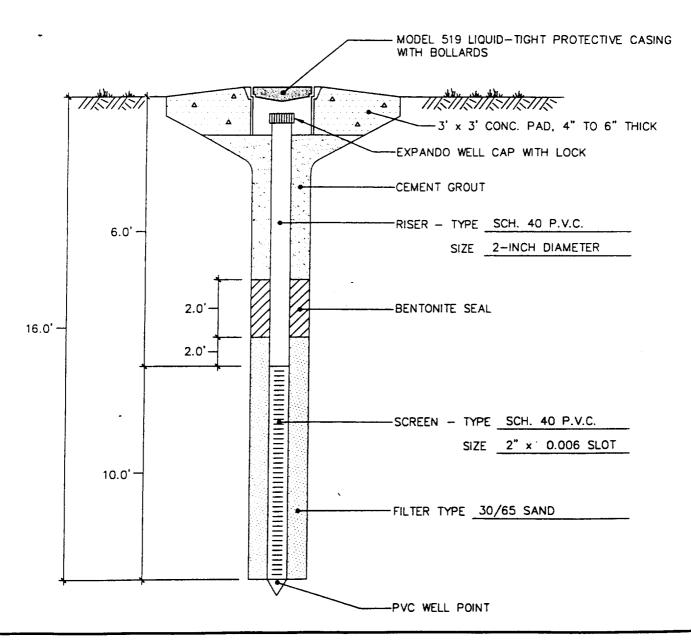
EST. W.S.W.T. (ft):

	s		Γ	T -	1 5	EST. W.S.W	.T. (ft):	1	TYPE OF	SAMPL	ING: ASTI	M D-158
DEPTH	SAMP	BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	w.T	S Y M B O	DESCRIPTION	-200 (%)		ATT	ERBERG MITS	K (FT./	OR
					Ľ		(70)	(%)	LL	PI	DAY)	(%
0十	Ť			-	 	Light brown fine SAND; with shell [SP]						
4					ļ:::							
1					ļ ::-			-				
5 —	} -	2-4-5	9 <u>.</u>			- loose						
+		4-5-5	10	_								
**	1	5-6-7	13	_		medium dense						
10	١	6-10-11	21									
7			1							1		
→	1										l	
15	٠	16-20-34	54									
4						very dense						
- -	١,	14 20 26										
20 —	^י	14-29-36	65									· • • • • • • • • • • • • • • • • • • •
-				-	· ·							
- 1 √	١,	5-19-27	46									
25		5-19-27	46			dense						
-				ľ		Dense, gray fine SAND [SP]	_				ĺ	
- - - - - - - - - -	۶	3-15-20	35	ŀ		·						
30			33					ļ				
[ŀ						ĺ		
				ŀ	:::							
15						BORING TERMINATED AT 35 FEET	-					
-										İ		
										1		
0 -	· · · · •											•••••
-								4.4				
_												
5	• • • •					,				·········.		
_												
]								ł	1			
7							 					
4										- 1		
7												
-	·····						ļļ.					



PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 09/02/98
WELL NUME	BER: DUNE-IW-5S LOCATION: A-5	
INSTALLED	BY: UES - ROCKLEDGE	





PROJECT NO.	18467-005-01
REPORT NO.:	

Р	Δ	G	F

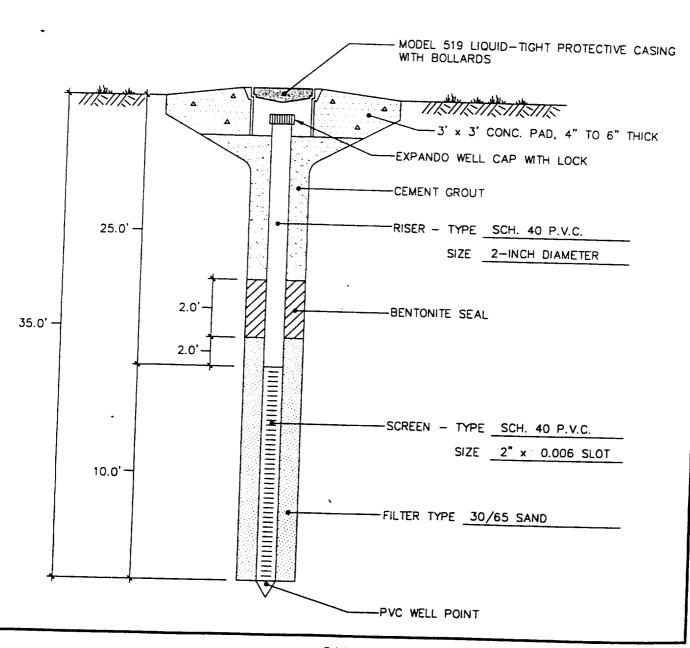
PROJECT: INSTALLATION 52 MONITOR WELLS

KENNEDY SPACE CENTER, FLORIDA

CLIENT: NASA JJ-D DATE: 09/02/98

WELL NUMBER: DUNE-IW-51 LOCATION: A-5

INSTALLED BY: UES - ROCKLEDGE



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	A-5	Client Phone:	(407) 867-4121
Date:	9-2-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID;	Dune-IW-5S	Total Depth:	16.0'
Screen Interval:	6.0 to 16.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
8:50 a.m.	60	
9:20 a.m.	22	
Total Development Time:	30 minutes	
Final Turbidity Reading:	22 NTUs	
Estimated Volume Removed	: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

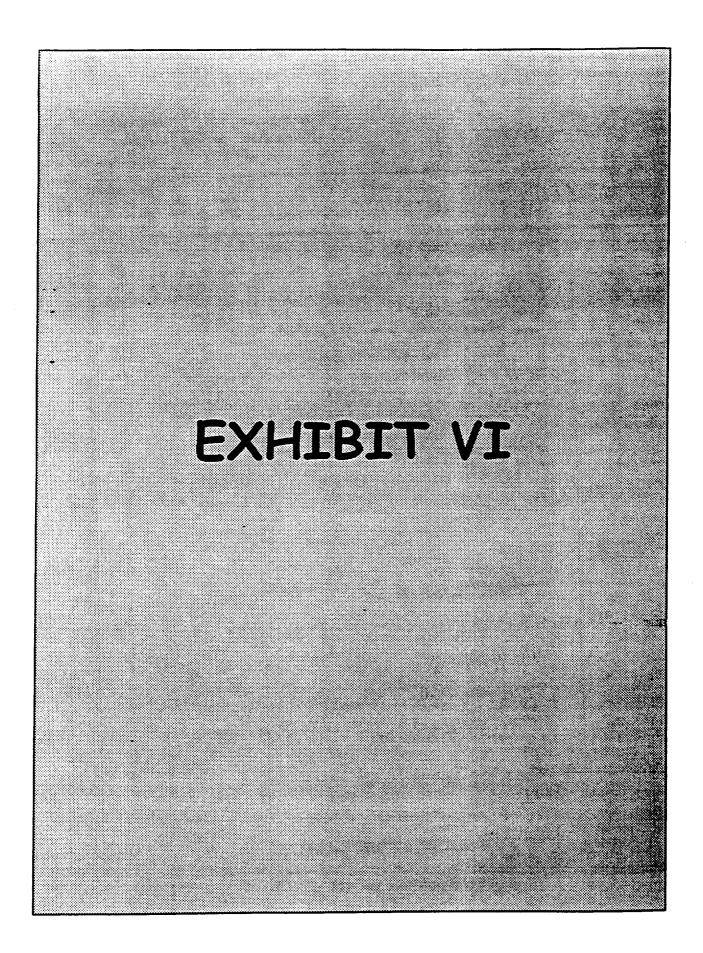
PROJECT	INFORMATION
---------	-------------

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	A-5	Client Phone:	(407) 867-4121
Date:	9-2-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-5I	Total Depth:	35.0'
Screen Interval:	25.0 to 35.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
12:15 p.m.	128	
12:25 p.m.	110	
12:30 p.m.	83	
12:40 p.m.	48	
12:45 p.m.	18	
·		
Total Development Time:	30 minutes	
Final Turbidity Reading:	18 NTUs	
Estimated Volume Removed:	55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 1,3457-005-01 REPORT NO. PAGE: B-2.6

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION:

A-6 TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

SECTION:

DATE STARTED:

8/24/98

WATER TABLE (ft): 8.0

DATE FINISHED:

8/24/98

DATE OF READING: 08/24/98

DRILLED BY:

UES - ROCKLEDGE

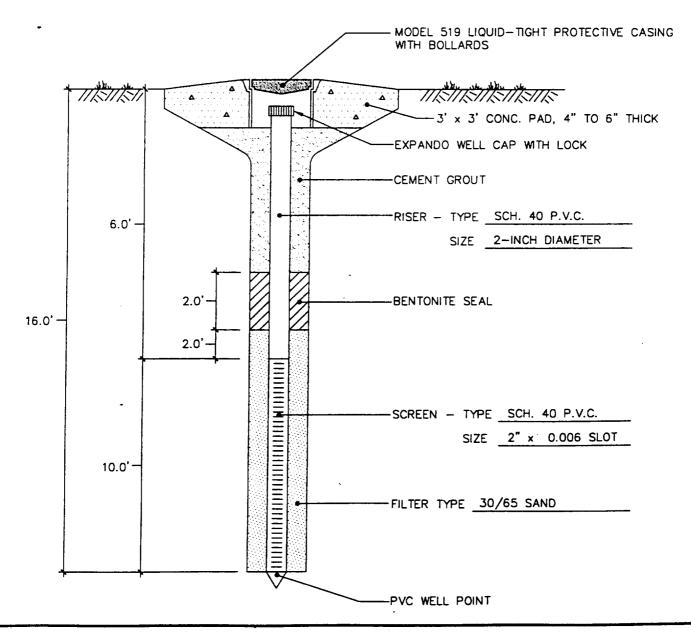
EST. W.S.W.T. (61)

Ts	: I		,	T e	EST. W.S.W.T.	(ft):	1	YPE OF	SAMPLIN	IG: ASTN	l D-1586
DEPTH M (FT.) P	INCREMENT	N (BLOWS/ FT.)	w.t.	S M B O	DESCRIPTION	-200 (%)	MC (%)		RBERG MITS	K (FT./	ORG.
E		<u> </u>	 	<u> </u>				LL	Pi	DAY)	(%)
0 1					Light brown fine SAND; with shell [SP]	-	 	-			
-											
7	-						-				
5 —	7-8-11				— medium dense	<u></u>					
 	14-15-12	27	_		Medium dense, dark brown fine SAND; with silt [SP-SM]]					!
-	8-11-13	24	▼		Medium dense, light brown fine SAND [SP]	1					
10	7 9-12-15	27			with shell						
7 1			ŀ				1		Í		***************************************
14			ŀ								
15	10-13-37	50	;								
7			•		Dense, gray fine SAND [SP]	1					
14			ŀ						1		
20	15-22-33	55			very dense						
1	[ŀ		voly delise						
\downarrow			ļ.								
25 —	15-17-23	40									
1					dense				1		
+	1		· .	\cdots	Dense, gray fine SAND [SP]						ĺ
30	18-24-25	49			-				l		
1			ļ.·								•••••
+			ļ:				1		j		
35	11-9-11	20					1				
] [medium dense			····			
			.1	 	Very loose, gray silt fine SAND; with clay						
40	1-1-1	2			and trace of shell [SM-SC]						j
				1.						•••••	
-[]	1]	oose, gray silt fine SAND; with shell [SM]		* *				
45	2-3-4	7	: -	:				ľ			
					`	·····	···				
]			-	· N	Medium dense, gray fine SAND; with shell						
50	6-6-5	11	:-	: [5	SPI						
30 1				В	ORING TERMINATED AT 50 FEET						
]											
											İ
55				<u></u>							
					210				1		



PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

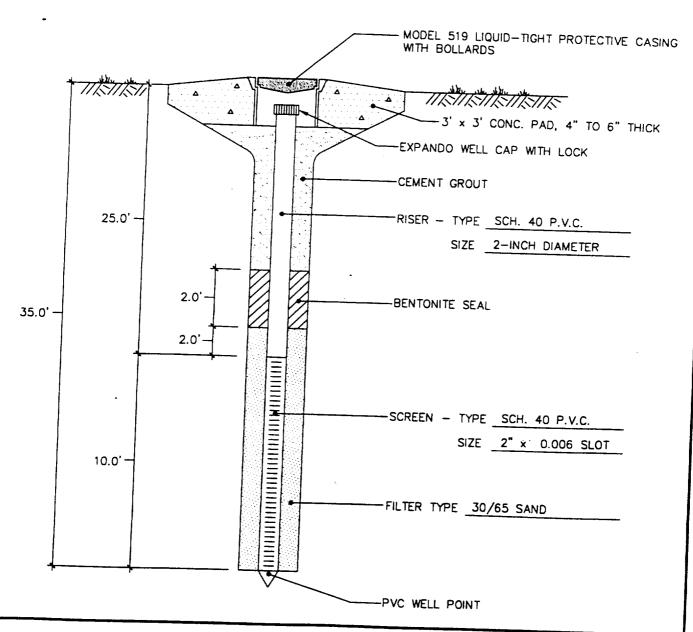
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/28/98
	BER: DUNE-IW-6S LOCATION: A-6	
	BY: UES - ROCKLEDGE	





PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT: INSTALLATION 52 MONITOR WELLS	
KENNEDY SPACE CENTER, FLORIDA	
CLIENT: NASA JJ-D	00 (00 (00
WELL NUMBER: DUNE-IW-61 LOCATION: A-6	DATE: 08/28/98
INSTALLED BY: UES - ROCKLEDGE	

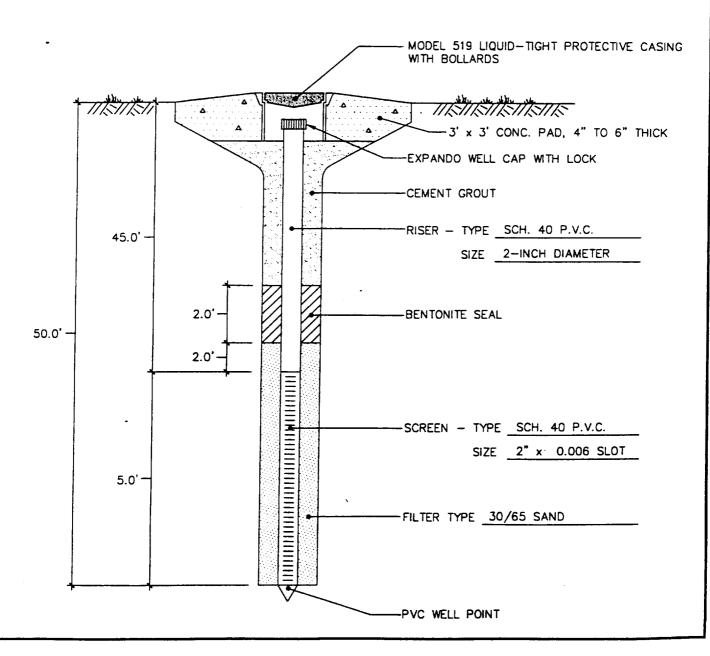




PROJECT NO	18467-305-01
REPORT NO.:	

PAGE:

PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
CLIENT: _	NASA JJ-D	DATE: 08/28/98	
WELL NUME	BER: DUNE-IW-6D LOCATION: A-6		
	BY: UES - ROCKLEDGE		



UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract:	NAS10-12206(B)/DO-5/TD-	Client Name:	NASA
Reference:	002 Monitor Well Installation	Department:	JJ-D
Project No.:	18467-005-01	Client Contact:	Posaly Santas The Land
Well Location:	A-6	Client Phone:	Rosaly Santos-Ebaugh, RPM (407) 867-4121
Date:	9-1-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-6S	Tetal Band	
Screen Interval:	6.0 to 16.0' bis	Total Depth: Well Type: Construction:	Permanent 2" PVC Riser
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	2" PVC Screen Rotary Pump, Tremie Pipe

	WELL DEVE	LOPMENT DATA
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
8:45 a.m.	42.7	Clear
9:15 a.m.	2.7	
· ·		
Total Development Time:	30 minutes	
Final Turbidity Reading:	2.7 NTUs	
Estimated Volume Removed:	55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
		COVAPID):
		<u> </u>

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	A-6	Client Phone:	(407) 867-4121
Date:	9-1-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-6I	Total Depth:	35.0'
Screen Interval:	25.0 to 35.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection: Flush mount with concrete pad and bollards		Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
8:30 a.m.	380	Clear
8:50 a.m.	75	
9:15 a.m.	19	
·		
	· · · · · · · · · · · · · · · · · · ·	
Total Development Time:	45 minutes	
Final Turbidity Reading:	19 NTUs	
Estimated Volume Removed	d: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

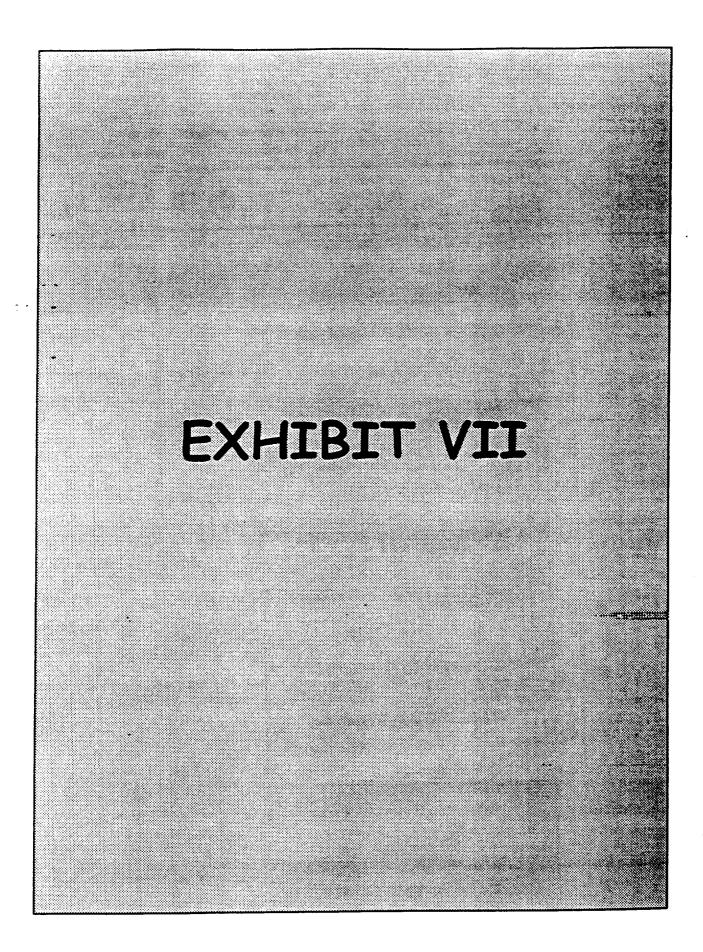
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Posaly Santos Ebourb David
Well Location:	A-6	Client Phone:	Rosaly Santos-Ebaugh, RPM (407) 867-4121
Date:	8-31-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Dune-IW-6D	Total Depth:	50.0'
Screen Interval:	45.0 to 50.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

WEEL DEVELOPMENT DATA			
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE	
9:00 a.m.	180		
9:05 a.m.	60		
9:30 a.m.	11		
Total Development Time:	30 minutes		
Final Turbidity Reading:	11 NTUs		
Estimated Volume Removed	: 55 gallons		
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):	
		==: NEOSE 10 (OVA-1D):	





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 19467-005-01 REPORT NO PAGE: B-2.7

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION:

AS SPECIFIED BY CLIENT

REMARKS:

BORING DESIGNATION:

B-1 TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

SECTION:

DATE STARTED:

8/3/98

WATER TABLE (ft): 7.3

DATE FINISHED:

8/3/98

DATE OF READING: 08/03/98 DRILLED BY:

EST. W.S.W.T. (ft):

TYPE OF SAMPLING:

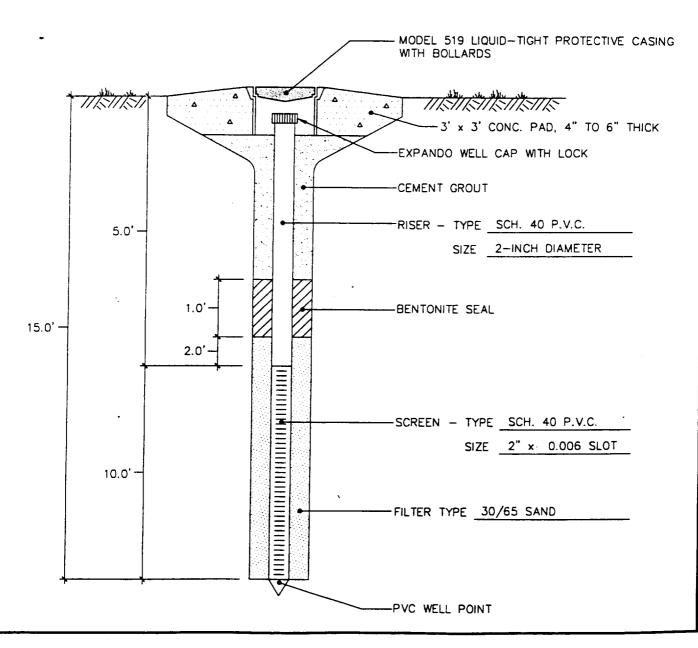
UES - ORLANDO

S				S	EST. W.S.W	.1. (rt):	T	YPE OF :	SAMPLIN	IG: ASTI	M D-1586
DEPTH M	PER 6"	N (BLOWS/ FT.)	w.t.	S Y M B	DESCRIPTION	-200 (%)	MC (%)		RBERG AITS	K (FT./	ORG.
E				L				LL	PI	DAY)	(%)
1 0 +					Very loose, orange fine SAND [SP]						
	2-2-2	4		• : • : •							
	- 3-2-3	5			loose						
5 - 1	5-2-3	5									
l 	5-4-4	8	_		Loose light and beautiful for						
1	4-4-4	8	[.	•:•:	Loose, light gray brown fine SAND [SP]						
10	4-2-10	12									
1 - 1			ŀ			-			1		
	0 * 1 * 1			11	Medium dense, light gray green silty shelly	_					
15	9-11-11	22			fine SAND [SM]						
			:								
X	9-9-9							1			
20	9-9-9	18					,				
4		ĺ									
74					Dense, light gray to white silty cemented	-				1	
25	18-24-16	40	[:]	-	SAND; with shell [SM]						
41									1		
₩	7.07				Medium dense, light gray green silty shelly fine SAND [SM]	7 1					
30	7-8-7	15			o Arto (ow)						
41				11	•						
₩.				11	Dense, light gray fine SAND; with shell [SP]	1	ļ				į
35	14-24-16	40			Solido, light gray line SAND; with shell (SP)						
4						1 1					
74				11	/ery loose, green silty fine SAND [SM]	1			ł		
40	0-1-1	2		[:]		ļ			-		
7											
74				s	oft, dark green gray CLAY [CH]	4 1					
45	1-2-1	3		//							
7							}				
#											- 1
50	3-3-3	6		Lo	pose, gray fine SAND; with shells and dark						
11				B	ay silt [SP] DRING TERMINATED AT 50 FEET				Ī		
] [. ==.		1				
55											
			<u> </u>	<u> </u>							



PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

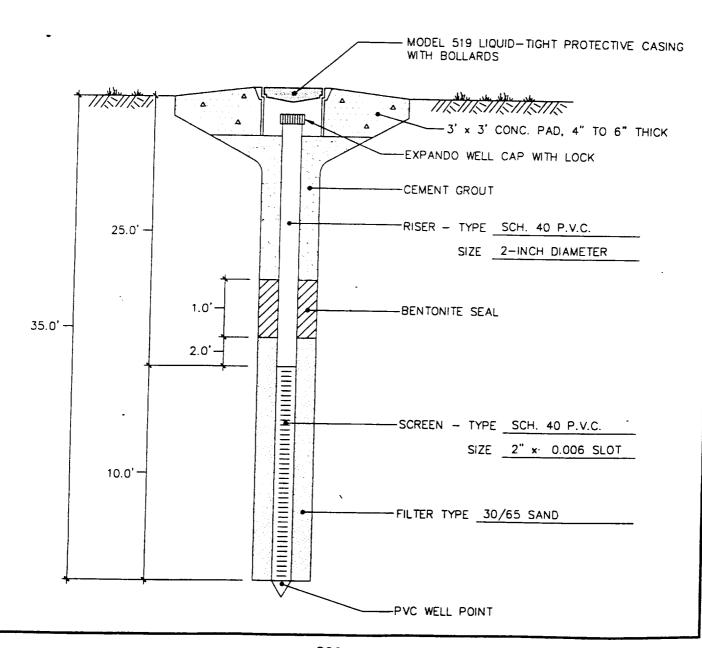
PROJECT:	INSTALLATION 52 MONITOR WELLS				
	KENNEDY SPACE CENTER, FLORIDA				
CLIENT: _	NASA JJ-D	DATE: 08/05/98			
WELL NUME	BER: DUNESWALE-IW-1S LOCATION: B-1				
INSTALLED	BY: UES - DAYTONA BEACH				





PROJECT NO: 18467-005-0:
REPORT NO
PAGE:

PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
CLIENT:	NASA JJ-D	DATE:	08/05/98
WELL NUMB	ER: DUNESWALE-IW-11 LOCATION: B-1		
INSTALLED	BY: UES - DAYTONA BEACH		



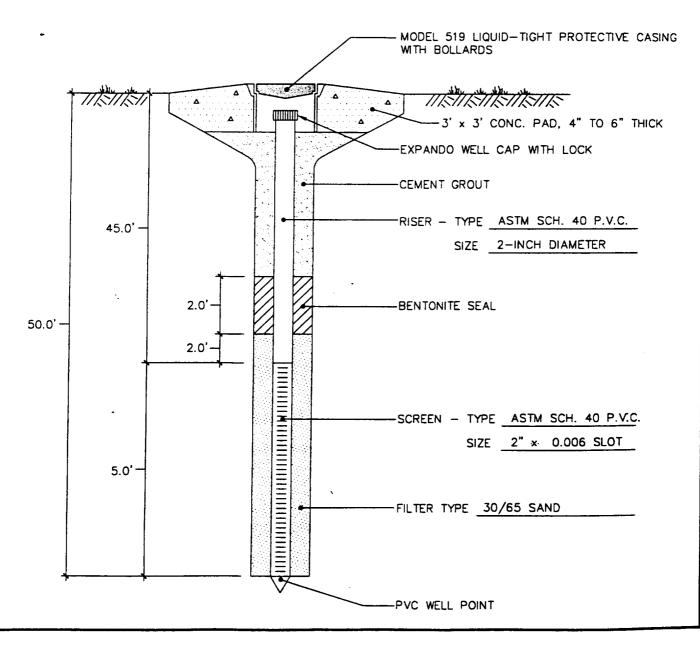


PROJECT NO.:	18467-005-01
REPORT NO.:	

PAGE:

PROJECT:	INSTALLATION 52 MONITOR WELLS					
	KENNEDY SPACE CENTER, FLORIDA					
CLIENT: _	NASA JJ-D	DATE: 08/03/98				
WELL NUME	BER: DUNESWALE-IW-1D LOCATION: B-1					
INSTALLED	BY: UES - ORLANDO					

WELL DIAGRAM - NOT TO SCALE



9

Š

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-1	Client Phone:	(407) 867-4121
Date:	8-5-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-1S	Total Depth:	15.0'
Screen Interval:	5.0 to 15.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
1:15 p.m.	200+	Cloudy/murky
1:18 p.m.	163.4	
1:21 p.m.	80.7	
1:23 p.m.	40.3	Clearer
1:25 p.m.	18.54	Clear
Total Development Time:	10 minutes	
Final Turbidity Reading:	18.54 NTUs	
Estimated Volume Removed	55 gallons	`
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
	4	

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-1	Client Phone:	(407) 867-4121
Date:	8-5-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-1I	Total Depth:	35.0'
Screen Interval:	25.0 to 35.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
11:40 a.m.	200+	Cloudy/Murky
11:43 a.m.	200+	
11:47 a.m.	200+	
11:50 a.m.	100.3	
11:53 a.m.	49.4	Clearer
11:57 a.m.	24.3	Clear
12:00 p.m.	18.1	Clear
Total Development Time:	20 minutes	
Final Turbidity Reading:	18.1 NTUs	
Estimated Volume Remove	d: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

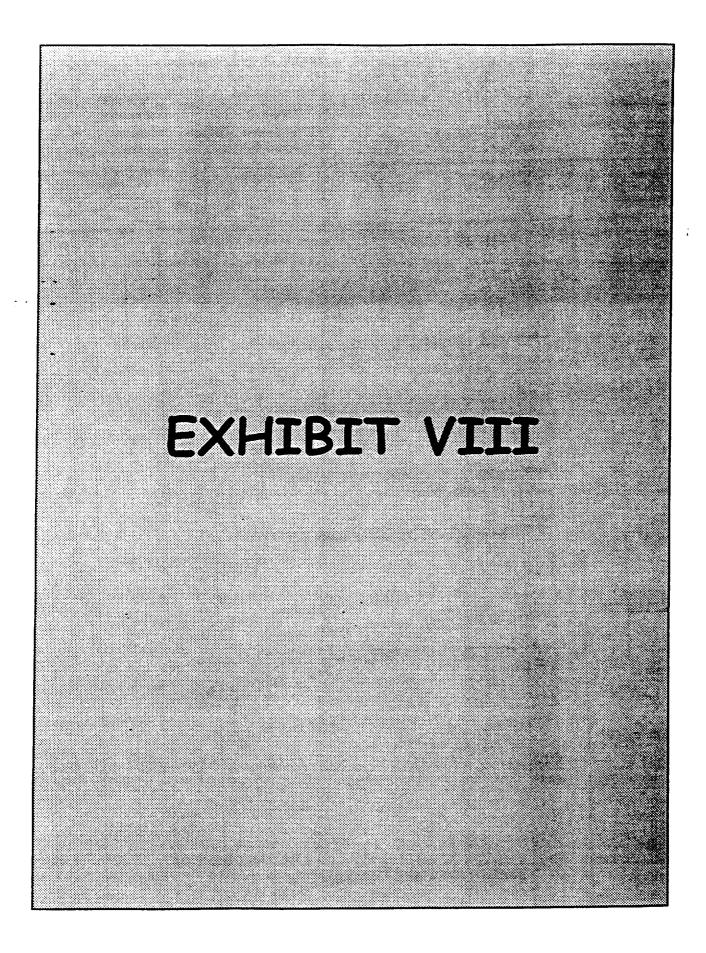
PROJECT INFORMATION

Contract:	NAS10-12206(B)/DO-5/TD-	Client Name:	NASA	
Reference:	Monitor Well Installation	Department:	JJ-D	
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM	
Well Location:	B-1	Client Phone:	(407) 867-4121	
Date:	8-3-98	Client Fax:		
			(407) 867-8040	

MONITOR WELL DATA

Well ID:	DuneSwale-IW-1D	Total Depth:	50.0'
Screen Interval:	45.0 to 50.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
3:05 p.m.	1,000+	SHOWATER AFFEARANCE
3:30 p.m.	101	Murky
3:45 p.m.	107	Cloudy
4:00 p.m.	56	Cloudy
4:05 p.m.	31.7	Clear
4:12 p.m.	27.4	Clear
4:18 p.m.	17.7	Clear
Total Development Time:	73 minutes	
Final Turbidity Reading:	17.7 NTUs	
Estimated Volume Removed:	100 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
		TESTER (OVA-PID):





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 12467-005-01 REPORT NO. PAGE: B-2.8

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER BREVARD, FLORIDA

BORING DESIGNATION: SECTION:

B-2 TOWNSHIP:

SHEET: 1 of 1

RANGE:

CLIENT:

NASA JJ-D

LOCATION:

AS SPECIFIED BY CLIENT

REMARKS:

HAND AUGER FROM 0 FEET TO 4 FEET

G.S. ELEVATION (ft):

DATE STARTED:

8/7/98

WATER TABLE (ft): 4.3

8/7/98

DATE OF READING: 08/03/98

DATE FINISHED: DRILLED BY:

UES - ORLANDO

EST. W.S.W.T. (ft):

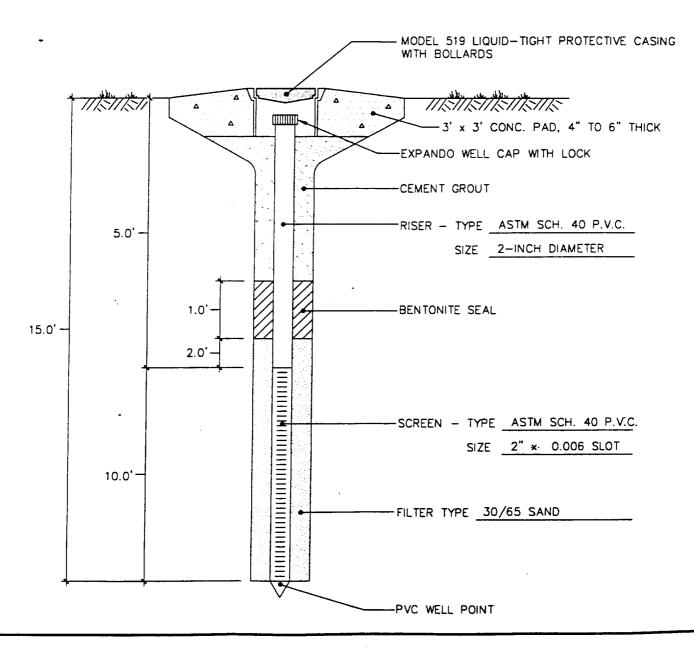
TYPE OF SAMPLING: ASTM DUESE

	S	1		T	S Y			4 D-1586				
DEPTH (FT.)	P		N (BLOWS/ FT.)	w.T.		DESCRIPTION	DESCRIPTION -200 MC LIMITS		K (FT./	ORG.		
<u> </u>	E		 	-					LL	PI	DAY	(%)
0-	1					Light gray fine SAND [SP]						
	-										•	
	1-	-	1 1	≖	• • •		1					
5 —	X	2-3-4	7			Very light gray to white fine SAND [SP]	_					
-	$\langle \rangle$	5-4-4	8	ŀ								
] [X	3-3-3	6	ŀ		Loose, very dark red brown fine SAND; with	_	İ	İ			
10 —	M	4-5-9	14			silt [SP-SM] — medium dense						
] [1				•:•:	medium dense						
-				Ė	<u> </u>	Loose very light and have C. Calle in	_	1		1		
15 —	Д	1-3-2	5			Loose, very light gray brown fine SAND [SP]		1				
			-	ŀ			1		1	·····-		••••••
_				Ŀ	· · ·					1		
20	X.	2-7-12	19			Medium dense, dark green silty fine SAND; with shells {SM}	1			ł		
-		[With shore town		•				
1		1	ł									
25 —	X	6-6-5	11		$ \cdot $	Medium dense, light green fine SAND; with shells [SP]	1					
23 -					• : •		·····		ļ	······································		
1			1		$ \cdot $							
	$\sqrt{}$	8-12-14	26	[:		•				1		j
30 -				· ·		dense						
											ı	1
	7	3-1-2	3			Soft, green CLAY; with shells [CH]			1			
35 -	Ť											
+								ļ				- 1
- 1 √	7	0-0-0	WÖR									
40-	لا	U-U-U	WOR			dark green				<u> </u>		
7						-						
*	1						j					
45	لإ	0-1-2	3									
7		-					}			Ī		
4]					
50 - 💢	.	1-0-1	1		4							
		1			- '	BORING TERMINATED AT 50 FEET			1	····		
						1		f				
55	ļ											
					1							



PROJECT NO:	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/10/98
WELL NUME	BER: DUNESWALE-IW-2S LOCATION: B-2	
INSTALL'ED	BY: UES - ROCKLEDGE	

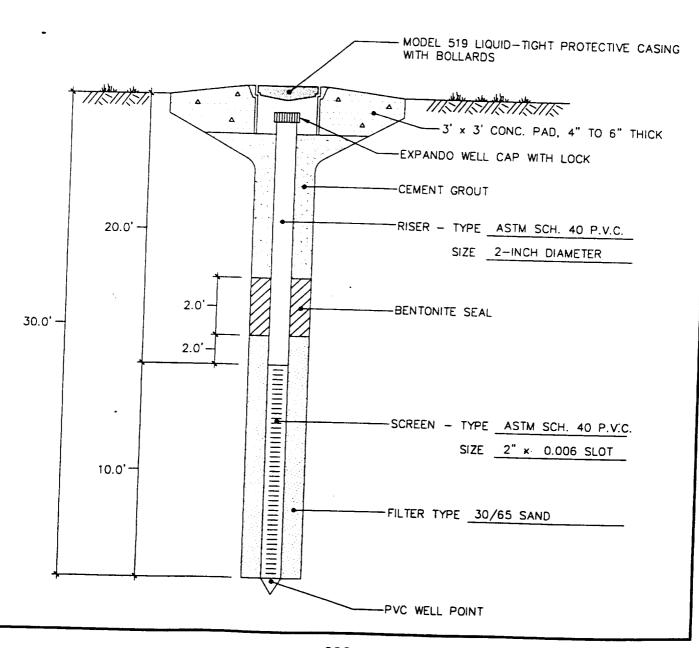




PROJECT NO.	15467-005-0:
REPORT NO.:	

PAGE:

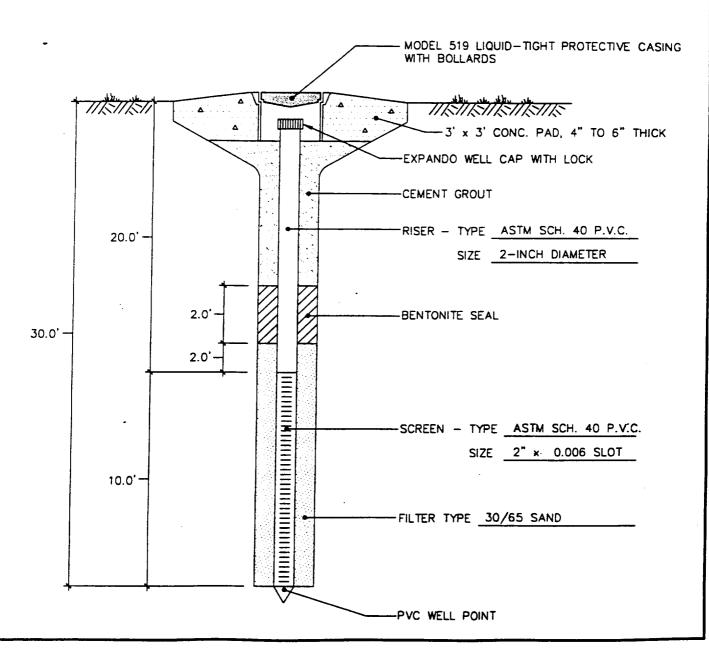
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT:	NASA JJ-D	DATE: 08/11/98
WELL NUMBI	ER: DUNESWALE-IW-21 LOCATION: B-2	50 T. (1971)
	BY: UES - ROCKLEDGE	





PROJECT NO.	15467-005-01
REPORT NO.:	
PAGE:	

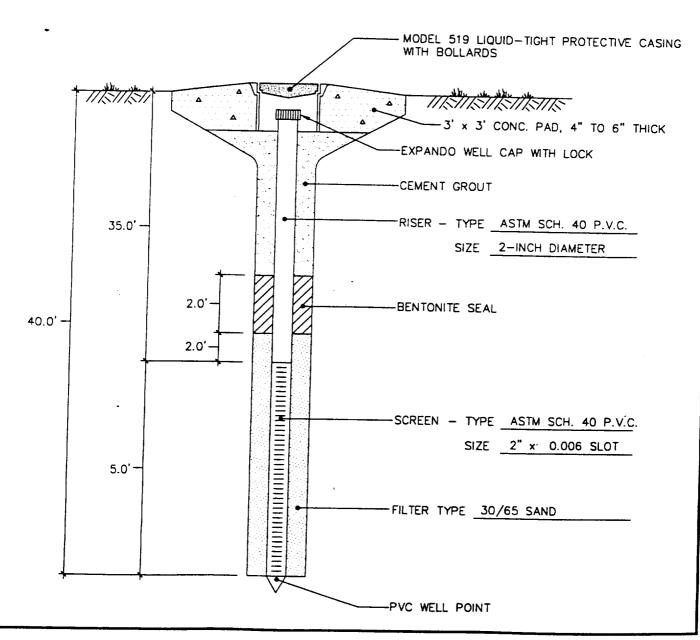
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/11/98
WELL NUME	BER: DUNESWALE-IW-21 LOCATION: B-2	





	PROJECT NO.: 18467-005-01
	REPORT NO.:
ı	PAGE:

PROJECT:	INSTALLATION 52 MONITOR WELLS							
	KENNEDY SPACE CENTER, FLORIDA							
CLIENT: _	NASA JJ-D	DATE:	08/07/98					
WELL NUME	ER: DUNE-SWALE-1W-2D LOCATION: B-2							
INSTALLED	BY: UES - ROCKLEDGE							



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-2	Client Phone:	(407) 867-4121
Date:	8-10-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-2S	Total Depth:	15.0'
Screen Interval:	5.0 to 15.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
1:55 p.m.	1,000+	Brown
2:00 p.m.	140	Light brown
2:02 p.m.	12.7	Clear
	A	
.		
Total Development Time:	7 minutes	
Final Turbidity Reading:	12.7 NTUs	
Estimated Volume Remove	d: gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract:	NAS10-12206(B)/DO-5/TD-	Client Name:	NACA
Reference:	002 Monitor Well Installation	Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Pacely Court
Well Location:	B-2	Client Phone:	Rosaly Santos-Ebaugh, RPM
Date:	8-11-98		(407) 867-4121
		Client Fax:	(407) 867-8040

MONITOR WELL DATA

Weil ID:	DuneSwale-IW-2I	Total Depth:	30.0'		
Screen Interval:	20.0 to 30.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen		
Above Grade Protection:	Flush mount with concrete pad with bollards	Development Method:	Rotary Pump, Tremie Pipe		

	WELL DEVE	ELOPMENT DATA
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
9:58 a.m.	1,000+	Grey
10:00 a.m.	447	Light grey
10:02 a.m.	217	Grey
10:04 a.m.	95.9	Grey
10:09 a.m.	73.4	Grey
10:12 a.m.	45.0	Grey
10:14 a.m.	40.4	Grey
10:16 a.m.	31.3	Grey
10:18 a.m.	21.1	Grey
Total Development Time:	20 minutes	
Final Turbidity Reading:	21.1 NTUs	
Estimated Volume Removed:	110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

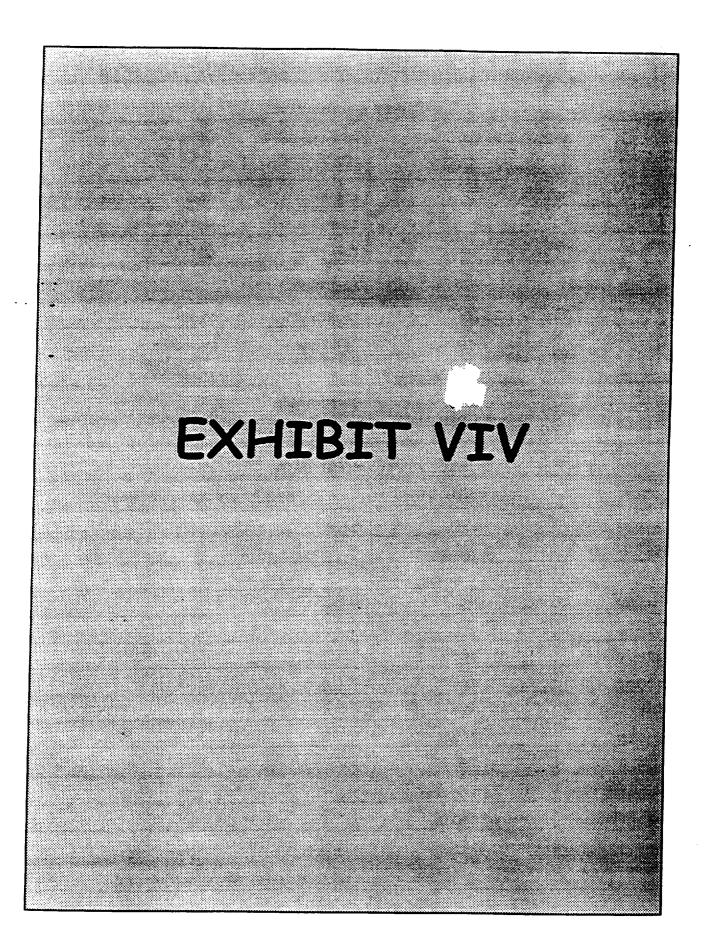
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-2	Client Phone:	(407) 867-4121
Date:	8-7-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-2D	Total Depth:	40.0'
Screen Interval:	35.0 to 40.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
3:48 p.m.	1,000+	Cloudy
3:52 p.m.	No reading	
Stop		
4:09 p.m.	166.4	Cloudy
4:13 p.m.	77.2	
4:15 p.m.	55.7	
4:17 p.m.	48.5	
4:19 p.m.	35.9	
4:20 p.m. ·	34.9	
4:22 p.m.	19	
Total Development Time:	17 minutes	
Final Turbidity Reading:	19 NTUs	
Estimated Volume Remove	d: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 19467-005-01 REPORT NO .: PAGE: B-2.9

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS:

HAND AUGER FROM 0 FEET TO 4 FEET

BORING DESIGNATION:

SECTION:

B-3

SHEET: 1 of 1

TOWNSHIP: RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/10/98

WATER TABLE (ft): 5.0 DATE OF READING: 08/10/98

DATE FINISHED:

8/10/98

DRILLED BY:

UES - ROCKLEDGE

EST. W.S.W.T. (ft):

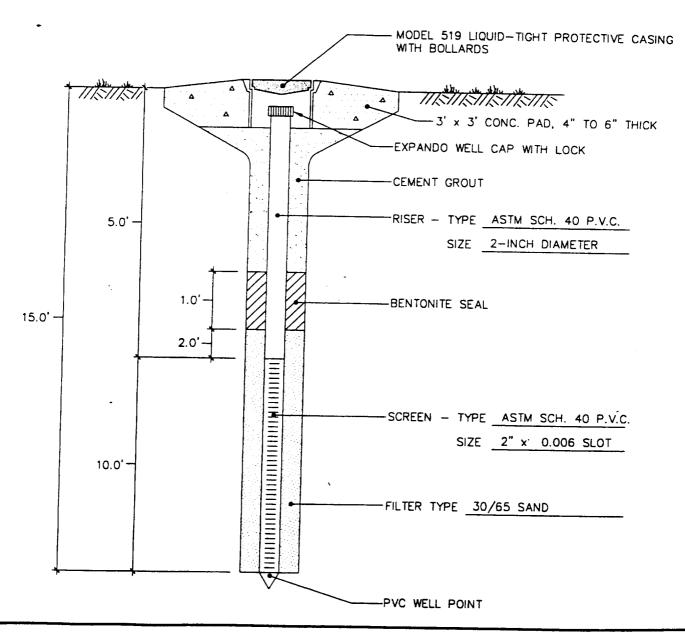
TYPE OF SAMPLING: ASTM D-1586

DEPTH	S A M P		N (BLOWS/	w.t.		DESCRIPTION	-200 (%)	MC (%)		RBERG MITS	K (FT./	ORG.
(F 1.)	E	INCREMENT	FT.)		Į.		(70)	(90)	LL PI		DAY)	(%)
0 –						Brown fine SAND; with roots (SP)						
-] - M		8			gray sand Loose, brown fine SAND; with silt [SP-SM] light brown	-					
5 - -		6-5-5	10			Loose, light gray brown fine SAND [SP]						
10 —		3-4-4 5-6-6	8 12									
- - -												
15 —						Light brown fine SAND (SP) BORING TERMINATED AT 15 FEET			<u> </u>		• • • • • • • • • • • • • • • • • • • •	
-												İ
20	}										•••••	
25 —												
25		·										
30 —						- 						
-				!								
35												•••••
<u> </u>												
40												•••••
15										Ì		
45 —												
50 —												
1												
55 —												



PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
CLIENT: _	NASA JJ-D	DATE:	08/10/98
WELL NUMB	ER: DUNESWALE-IW-3S LOCATION: B-3		
INSTALLED	BY: UES - ROCKLEDGE		



STANDARD WELL DEVELOPMENT LOG

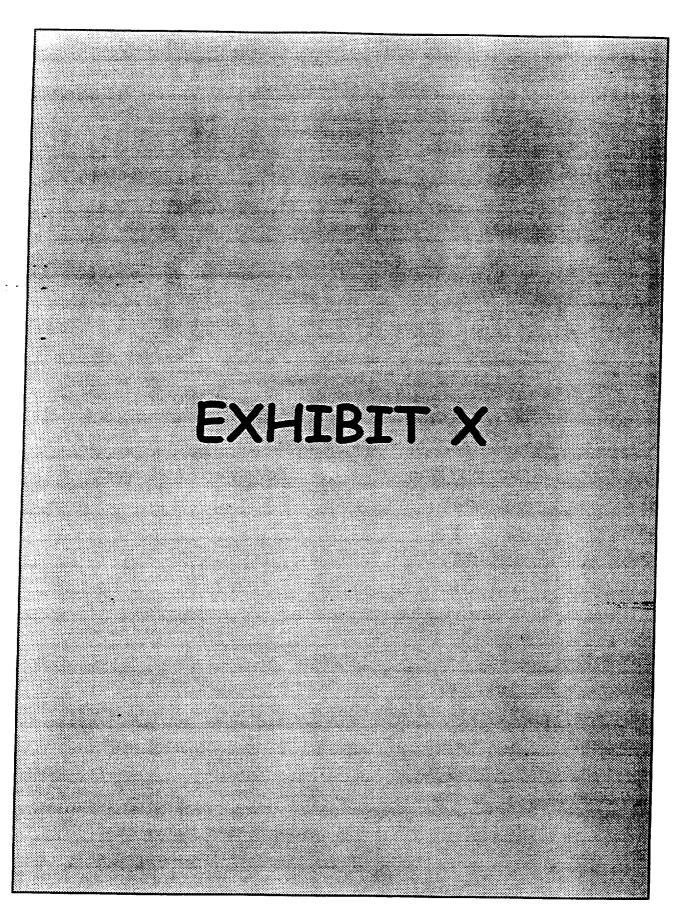
PROJECT INFORMATION

Contract:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-3	Client Phone:	(407) 867-4121
Date:	8-10-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-3S	Total Depth:	15.0'
Screen Interval:	5.0 to 15.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad with bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
9:09 a.m.	1,000+	
9:13 a.m.	22.5	
9:14 a.m.	10.9	
		<u> </u>
Total Development Time:	5 minutes	· · · · · · · · · · · · · · · · · · ·
Final Turbidity Reading:	10.9 NTUs	,
Estimated Volume Remove	d: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO .. 18467-005-01

REPORT NO .:

PAGE:

B-2.10

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION:

SECTION:

B-4 TOWNSHIP: SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/11/98

WATER TABLE (ft): 6.11

DATE FINISHED:

8/11/98

DATE OF READING: 08/10/98

DRILLED BY:

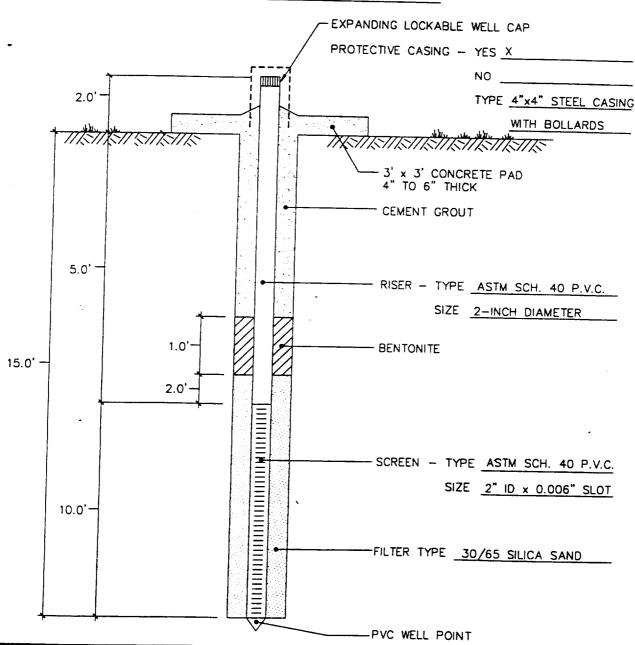
UES - ORLANDO

	BLOWS PER 6"	N (BLOWS/	w.t.		DESCRIPTION	-200 (%)	MC (%)		RBERG MITS	K (FT./	ORG CON1
[FT.)		ů				u	Pi	DAY)	(%)
0+	-				Mixed gray brown fine SAND [SP]						
1	.],				Light gray fine SAND [SP]						
5 —	3-3-4	7			loose, with roots	ļ	ļ				
\rightarrow	5-5-5	10	T		Loose, very light brown fine SAND [SP]	1					
7	7-8-9	17			Medium dense, dark brown fine SAND; with silt [SP-SM]	1	i				
10 7	9-10-15	25			Medium dense, dark gray brown fine SAND						
				· · · · · · · · · · · · · · · · · · ·	[SP]						
15	3-3-4	7			Loose, dark reddish brown fine SAND [SP]						
1											
<u></u>	2-2-2	4	į	• • • •	Very loose, gray fine SAND [SP]						
20 -	1			-							•••••
1											
25	10-13-23	36	[
.5					dense, with shells						••••••
I X	7-9-17	26			·····						
~ 					medium dense, with fine sand				Ī		
			ŀ								
5 📉	9-10-9	19		:::	not as much shell						
					not as miden stan						
		.	Ŀ	. · · ·							
<u>، ۲</u> ۲	2-3-5			 	Loose, greenish gray SILT; with few shells {ML}						
+			Ц	Щ							
5 🐴	6-8-4	12			Medium dense, very light gray fine SAND; with few shells {SP}						
1			[:						ŀ		
+			<u>.</u>						j		
<u> </u>	15-21-24	45	<u>'</u>	<u></u>	BORING TERMINATED AT 50 FEET						
					DOTTING TERMINATED AT 50 FEET						
4							1				
5 🚽 📗	1			- 1					1	1	



	PROJECT NO.:	18467-005-01
	REPORT NO.:	
1	PACE:	

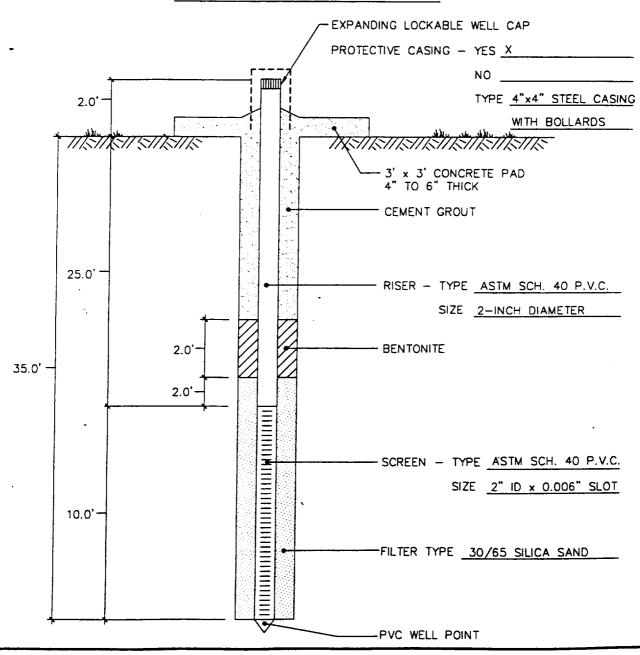
PROJECT: _	INSTALLATION 52 MONITO	R WELLS				
	KENNEDY SPACE CENTER,	FLORIDA				
CLIENT:!	NASA JJ-D			DATE:	08/14/98	
WELL NUMBER	R: DUNESWALE-IW-45	LOCATION:	8-4	DATE:	30/14/30	
INSTALLED BY	UES - ROCKLEDGE					





PROJECT NO.:	18467-005-01
REPORT NO.:	
PACE:	

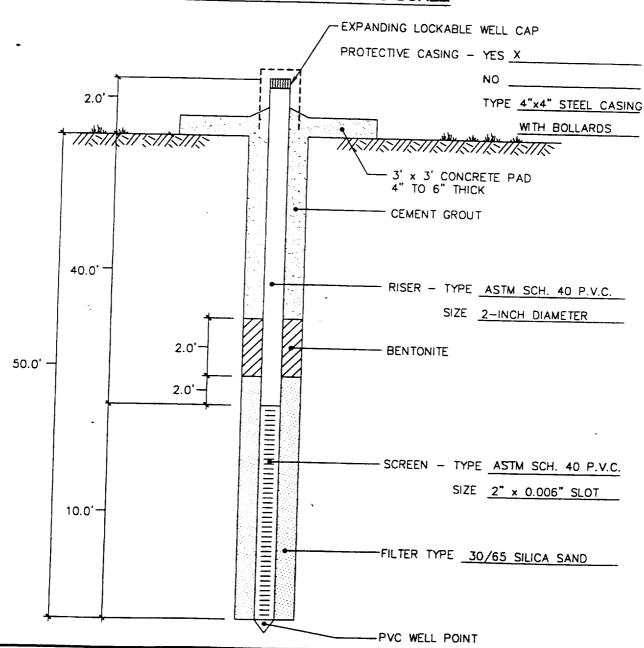
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/14/98
WELL NUME	BER: DUNESWALE-IW-4I LOCATION: B-4	
INSTALLED	BY: UES - ROCKLEDGE	





	PROJECT NO. 18467-005-01
ĺ	REPORT NO.:
	PAGE:

PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
CLIENT: _	NASA JJ-D	DATE:	08/11/98
WELL NUMB	ER: DUNE SWALE-1W-4D LOCATION: B-4		
INSTALLED I	BY. UES - ORLANDO		



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-4	Client Phone:	(407) 867-4121
Date:	8-14-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-4S	Total Depth:	15.0'
Screen Interval <u>:</u>	5.0 to 15.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4"x4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
10:20 a.m.	1,000+	Dark brown
10:30 a.m.	1,000+	
10:45 a.m.	550	Brown
10:55 a.m.	300	Light brown
11:00 a.m.	220	
11:05 a.m.	190	
11:10 a.m.	190	
Total Development Time:	50 minutes	
Final Turbidity Reading:	190 NTUs	
Estimated Volume Removed	i: 165 gallons	\
COMMENTS: Well screen	installed in silty sand and	SOIL SCREENING RESULTS (OVA-PID):
fine sand		

UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Posaly Santon Ebourb Santon
Well Location:	B-4	Client Phone:	Rosaly Santos-Ebaugh, RPM (407) 867-4121
Date:	8-14-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-4I	Total Depth:	35.0'
Screen Interval <u>:</u>	25.0 to 35.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

WELL DEVELOPMENT DATA						
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE				
12:55 p.m.	1,000+	Dark grey				
1:02 p.m.	60	Clear				
1:25 p.m.	33	Clear				
Total Development Time:	30 minutes					
Final Turbidity Reading:	33 NTUs					
Estimated Volume Removed	1: 55 gallons					
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):				

UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

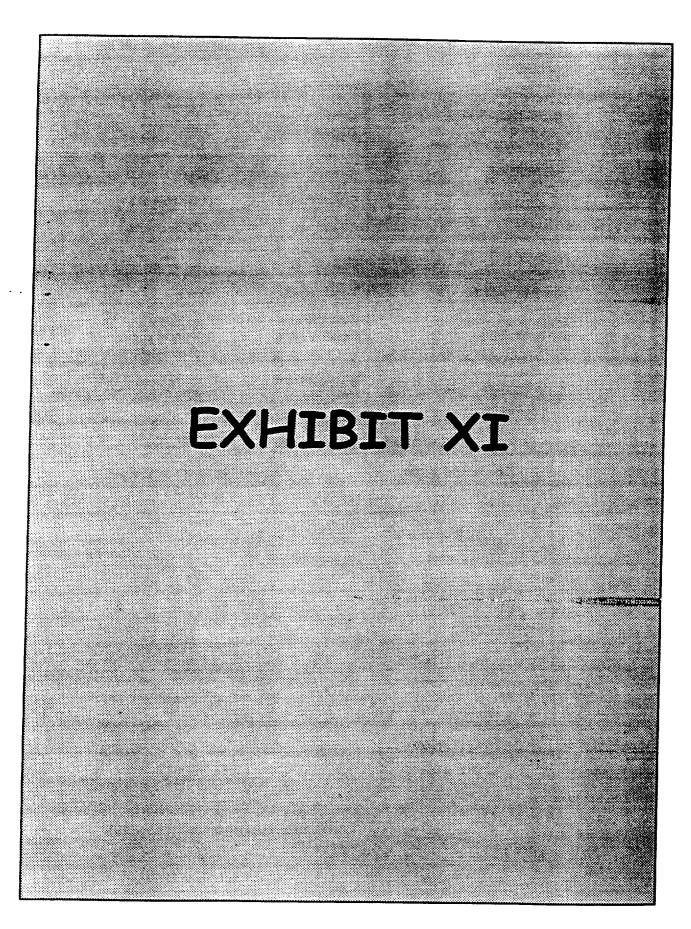
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-4	Client Phone:	(407) 867-4121
Date:	8-11-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-4D	Total Depth:	50.0'
Screen Interval;	45.0 to 50.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
1:02 p.m.	No reading	
1:04 p.m.	310	Cloudy grey
1:08 p.m.	59.6	
1:10 p.m.	72.1	
1:12 p.m.	26.8	
1:14 p.m.	15.6	
		,
Total Development Time:	12 minutes	
Final Turbidity Reading:	15.6 NTUs	
Estimated Volume Remove	d: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





UNIVERSAL ENGINEERING SCIENCES BORING LOG

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET

BORING DESIGNATION:

SECTION: TOWNSHIP:

B-5

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/13/98

WATER TABLE (ft):

DATE FINISHED:

8/13/98

DATE OF READING: 08/ /98

DRILLED BY:

UES - ROCKLEDGE

EST. W.S.W.T. (ft):

TYPE OF SAMPLING: ASTM D-1586

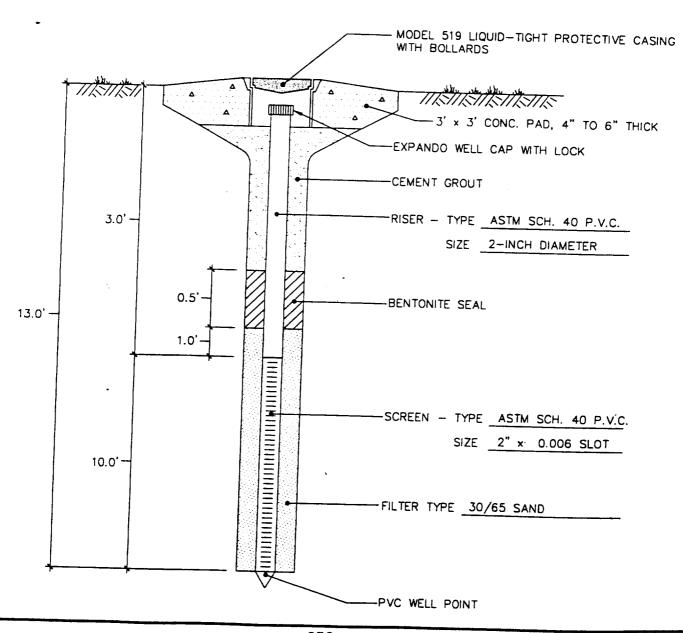
DEPTH M (FT.)	BLOWS PER 6"	N (BLOWS/	w.t.	, ,	DESCRIPTION -200 (%)		MC (%)	ATTERBERG LIMITS		K (FT./	ORG.
E	INCREMENT	FT.)		L L		(~ /	1701	LL	PI	DAY)	(%)
0-								 			
¥ -					Brown fine SAND; with roots [SP]						
-						ļ.	1				
1-	•				light gray						
5 —X	2-2-4	6		,	Loose, light brown fine SAND; with roots						
±X	4-4-6	10			Loose, gray brown fine SAND; with silt	7					İ
$-\mathbf{X}$	6-6-10	16			∖[SP-SM]	\mathcal{A}					
	6-7-8	15		[.].	Medium dense, dark brown silty fine SAND;						
10					with roots (SM)					••••••••••	
4 1			ľ	111							
]	ŀ	- [- [-]							
15 TX	2-7-14	21			Medium dense, gray fine SAND; with shells		j				
			- 1		(SP)						
							İ				
₩	40.40.45		.	:::							
20-	10-18-15	33		-:-:	dense		ļ				
		1	ŀ		35.133	1					
]	ļ		l.	$\cdot \cdot \cdot \cdot $							
😾	14-10-14	24	-		Medium dense, gray fine SAND; with shells	4					
25	45.195.14	24			[SP]	·					••••••••
41		Ì	ŀ							1	
-		ĺ].								
30 - 📉	10-11-10	21	<u> </u> :	:: <u>:</u> :							
4			į.	$\cdot \cdot \cdot $	gray						
			ŀ	$ \cdot $						I	
- ₩	242								1		
35	3-1-2	3	·····		Very loose, gray silty fine SAND (SM) BORING TERMINATED AT 35 FEET	-	·····	ļ			••••
] [1				BORING TERMINATED AT 35 FEET						
41		.								-	
. + 1		ļ	-	-					ŀ	- 1	
40									·····		
4									-		
] []			[
45					\						
									1	-	
]						[1	
11											
50											
]											
11]	
		1									
55	1							1		1	



PROJECT NO.	18467-005-01
REPORT NO .:	

PAGE:

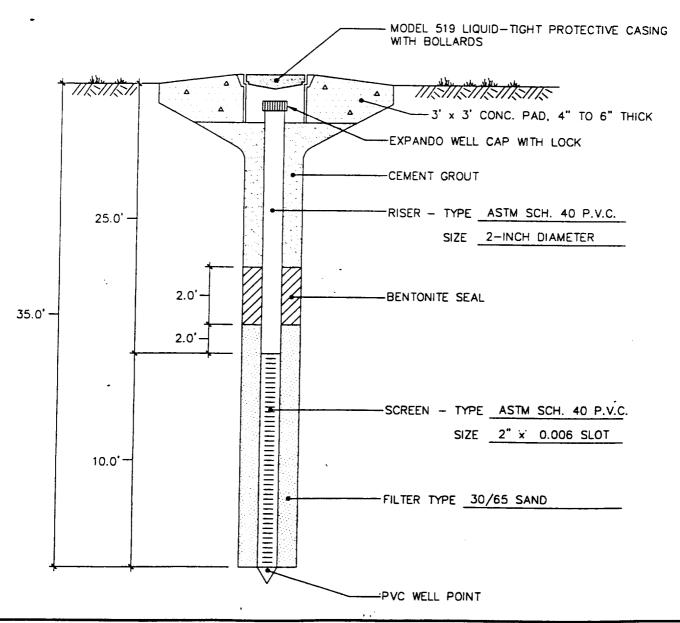
PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
CLIENT: _	NASA JJ-D	DATE:	08/13/98
WELL NUME	BER: DUNESWALE-IW-5S LOCATION: B-5	DATE:	307.0730
	BY: UES - ROCKLEDGE		





PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS				
	KENNEDY SPACE CENTER, FLORIDA				
CLIENT:	NASA JJ-D	DATE: <u>08/13/98</u>			
WELL NUMBE	R: DUNESWALE-IW-5I LOCATION: B-5				
INSTALLED F	gy. UES - ROCKLEDGE				



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-5	Client Phone:	(407) 867-4121
Date:	8-13-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-5S	Total Depth:	13.0'
Screen Interval:	3.0 to 13.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
6:34 p.m.	1,100	Brown
7:03 p.m.	800	Brown
7:15 p.m.	330	Brown
7:30 p.m.	150	
7:37 p.m.	180	Well surge
8:00 p.m.	75	
Total Development Time:	86 minutes	
Final Turbidity Reading: 75 NTUs		
Estimated Volume Removed: 165 gallons		
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
Well screen installed in silty sand.		

UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

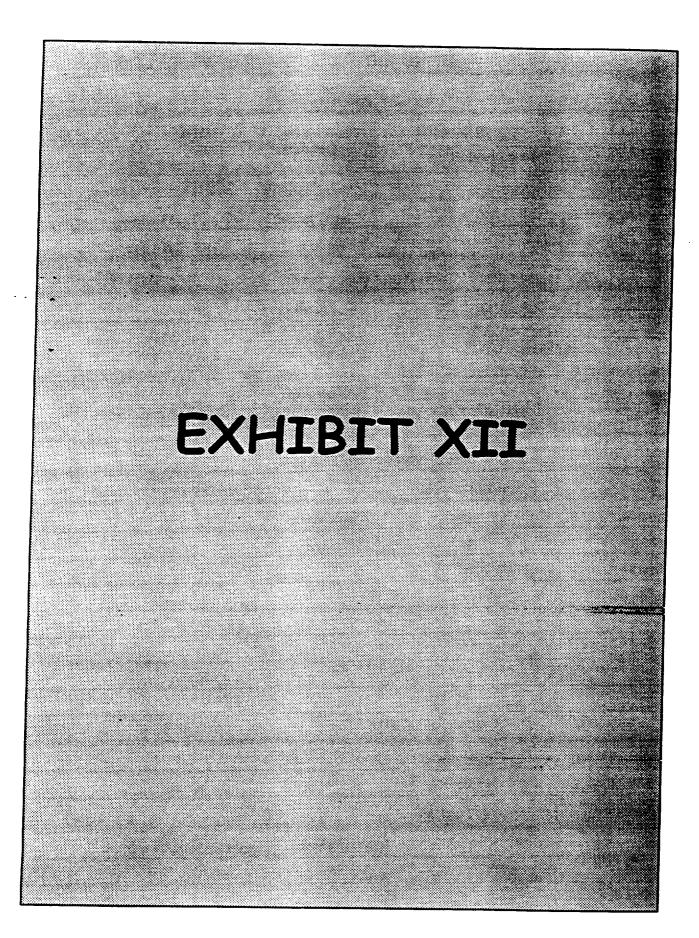
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-5	Client Phone:	(407) 867-4121
Date:	8-13-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-5I	Total Depth:	35.0'
Screen Interval:	25.0 to 35.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
3:00 p.m.	1,000+	
3:17 p.m.	450	Grey silty
3:20 p.m.	65	Light grey
4:02 p.m.	17	
7.02 p.m.	17	
·		
	· · · · · · · · · · · · · · · · · · ·	
·		
Total Development Time:	62 minutes	
Final Turbidity Reading:	17 NTUs	
Estimated Volume Removed	d: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
-		





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 18467-005-01 REPORT NO. PAGE: B-2.12

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION:

AS SPECIFIED BY CLIENT

REMARKS:

HAND AUGER FROM 0 FEET TO 4 FEET

BORING DESIGNATION:

SECTION:

B-6

SHEET: 1 of 1

TOWNSHIP:

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/12/98

WATER TABLE (ft): 4.5

DATE FINISHED:

8/12/98

DATE OF READING: 08/12/98

DRILLED BY:

UES - ORLANDO

EST. W.S.W.T. (ft):

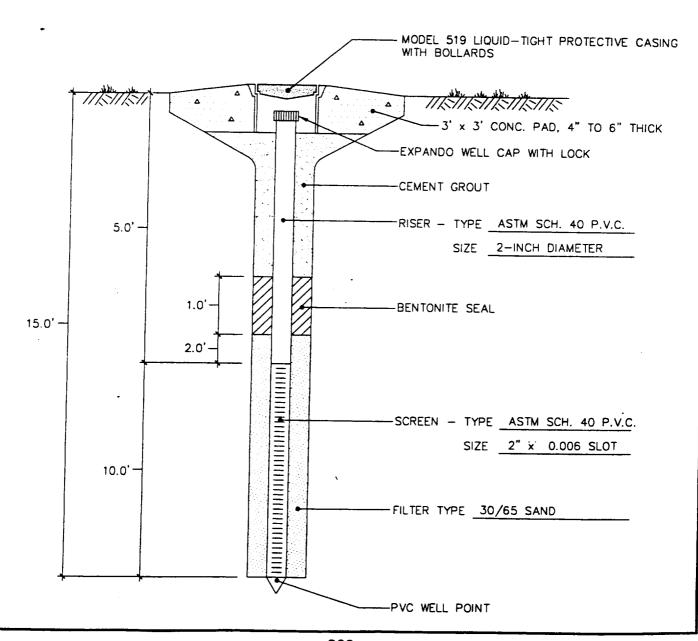
TYPE OF SAMPLING: ASTM D-1586

<u> </u>	S			ī	S	ES1. W.S.W.1	1		T -		NG: ASTI	M D-1586
DEPTI (FT.)	H AMPLE	BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	w.t.		DESCRIPTION	-200 (%)	MC (%)	ATT	ERBERG IMITS	(FT./	ORG.
	E				Ĺ				LL	Pi	DAY)	(%)
0	+			-		Light gray fine SAND; with roots [SP]						
1	4											
	1-	•			\cdots	brown						
5	$-\!$	8-16-20	36			dark brown						
1	$\frac{1}{4}$	26-28-29	57			dense				1		
	K.	10-11-29	40	İ		Very dense, dark gray fine SAND; with silt [SP-SM]						
10-	1	26-28-32	60			very dense						
	11			ŀ		401 y 401130						
Ì	从											
15 -	X	10-12-12	24			medium-dense						
	1					BORING TERMINATED AT 15 FEET						
	+								<u> </u>			
20 -] .							1.				
	1			İ	1							***************************************
,	$ \cdot $	1		l								
25 –												
	$\left\{ \ \right\}$				ĺ							
-]			- 1	l						İ	
30	1					· ·]			
-	1							1		·····		
-	11		İ		ł			İ			l	1
35 					- 1							ı
-								†	ļ <u>-</u>			
-			_									
40							İ					
40 —							````		 			
												1
						,						Ì
45 —	1								ļļ.			
4												1
7												
50												
7												
1								į				
55 —												
		·				···						



PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT: _	NSTALLATION 52 MONITOR WELLS	
_ +	KENNEDY SPACE CENTER, FLORIDA	
_		
CLIENT: N	NASA JJ-D	DATE: <u>08/12/98</u>
WELL NUMBER	DUNESWALE-IW-6S LOCATION: B-6	
INSTALLED BY	UES - ROCKLEDGE	



STANDARD WELL DEVELOPMENT LOG

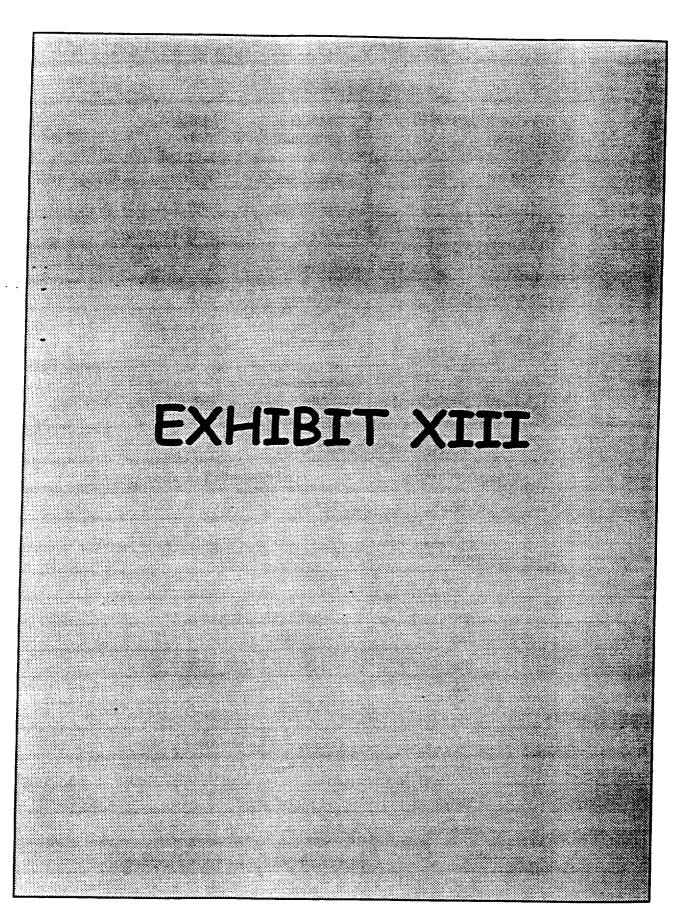
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	B-6	Client Phone:	(407) 867-4121
Date:	8-12-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	DuneSwale-IW-6S	Total Depth:	15.0'
Screen Interval:	5.0 to 15.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad with bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
11:55 a.m.	1,000+	
12:00 p.m.	1,000+	
12:06 p.m.	650	
12:09 p.m.	380	
12:15 p.m.	290	
12:20 p.m.	240	
1:25 p.m.	300	·
1:30 p.m.	110	
1:39 p.m.	55	<u> </u>
Total Development Time:	104 minutes	
Final Turbidity Reading:	55 NTUs	`
Estimated Volume Removed	l: 165 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





UNIVERSAL ENGINEERING SCIENCES BORING LOG

PROJECT NO.: 18457-005-01

REPORT NO.

PAGE: B-2.13

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS:

BORING DESIGNATION:

SECTION:

C-1

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/12/98

WATER TABLE (ft): 5.3

D.

DATE FINISHED: 8/12/98

DATE OF READING: 08/12/98

DRILLED BY:

UES - D.B.

EST. W.S.W.T. (ft):

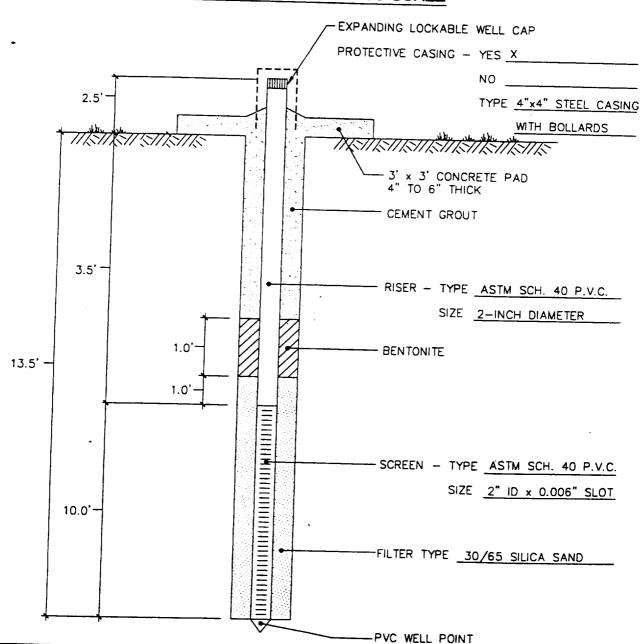
TYPE OF SAMPLING: ASTM D-1586

DEPTH M (FT.) P	PER 6"	N (BLOWS/	W.T.	, ,	DESCRIPTION	-200 (%)	MC (%)		RBERG MITS	Κ (FT./	ORG.
E	INCREMENT	FT.)		Ç		(%)	(70)	LL	PI	DAY)	(%)
0				 	Loose, gray fine SAND; trace roots [SP]						
1	2-3-3	6		· · · ·		1					
1	3-4-7	11		· ·	Medium dense, dark brown fine SAND [SP]; hardpan						
5 — 🔀	3-3-4	7	<u>.</u>	,,,	Loose, dark brown fine SAND; trace silt [SP]	ļ	ļ	ļ			
1	4-5-3	8		• • • •	dedictions of						
1	3-2-2	4			dark brown						
10	2-1-2	3		,	very loose, brown		ļ				
- \[\]	1167			Δ Δ Δ Δ	Medium dense, light gray cemented SHELL; with sand						
15	11-6-7	13		ΔΖ	Medium dense, gray cemented SHELL; with						
				Δ Δ Δ Δ	sand						
→	0.10.10			ΔΔ							
20	9-10-12	22		$\Delta \Delta \Delta$							
7			-	4							
₩	44 40 45				İ						
25	11-12-15	27		^ ^	Medium dense, gray fine SAND; with shell		• • • • • • • • • • • • • • • • • • • •				
7 1			F		[SP]	į			İ	l	
74	8-11-12	23	ļ		. •						
30	0-11-12	23	······	• •	Medium dense, blueish dark gray fine SAND;						
			F		with shell [SP]						
74	10-8-9		ŀ								
35	10-8-9	17	······	• • •	Medium dense, blueish dark gray fine SAND;						
4	1		ŀ		shell, and trace silt [SP]					1	•
₩	7-8-8	16	ŗ	::].		
40	/-0-8	10	······	• • • •	Medium dense, gray fine SAND; trace shell,						••••
4 }					clay		• •				•
74	6-5-5	10		::							
45	0-0-0	10	Ż		Loose, blue gray CLAY; some shell [CH]						
7											
\forall	4-4-5										
50	4-4-3	9			BORING TERMINATED 50 FEET						·•············
	-										
					1						
55 —											



PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

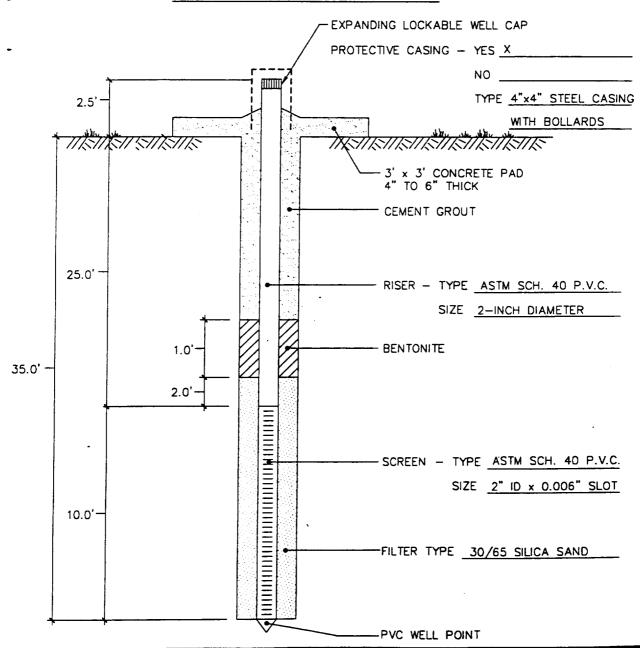
PROJECT:	INSTALLATION 52 MONITOR WELLS						
	KENNEDY SPACE CENTER, FLORIDA						
CLIENT: _	NASA JJ-D DATE: _08/13/98						
WELL NUMB	ER: WESTERN-IW-1S LOCATION: C-1						
INSTALLED	BY: UES - DAYTONA BEACH						





PROJECT NO. 18467-005-01	
REPORT NO.:	
PACE:	

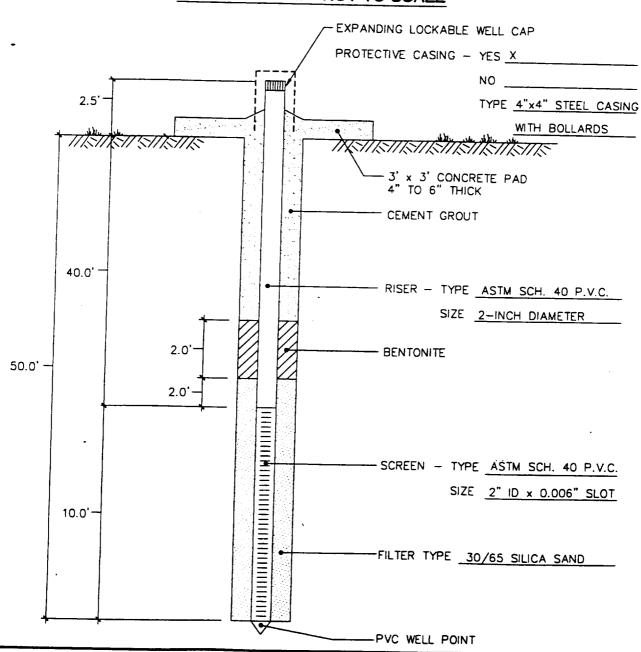
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/12/98
WELL NUME	BER: WESTERN-IW-11 LOCATION: C-1	
INSTALLED	BY: UES - DAYTONA BEACH	





PROJECT NO	18467-005-01
REPORT NO.:	
PAGE:	

INSTALLATION 52 MONITOR WELLS		
KENNEDY SPACE CENTER, FLORIDA		
NASA JJ-D	DATE: 08/	19/98
ER: WESTERN-IW-1D LOCATION: C-1	0/11L,	
ay. UES - ORLANDO		
	KENNEDY SPACE CENTER, FLORIDA NASA JJ-D ER: WESTERN-IW-1D LOCATION: C-1	NASA JJ-D DATE: 08/



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-1	Client Phone:	(407) 867-4121
Date:	8-13-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Western-IW-1S	Total Depth:	15.0'
Screen Interval:	3.5 to 13.5' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
8:50 a.m.	200+	Murky/Cloudy
	200+	
	158.9	
	100.5	Cloudy
	83.6	
9:50 a.m.	58.9	
Total Development Time:	60 minutes	• • • • • • • • • • • • • • • • • • • •
Final Turbidity Reading:	58.9 NTUs	
Estimated Volume Remove	ed: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
Well screen installed in sand, silty sand, and sand		
with shell.		

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

		THE ORIGINATION	
Contract:	NAS10-12206(B)/DO-5/TD- 002	Client Name:	NASA
Reference:	Monitor Well Installation	Department:	JJ-D
Project No.:	18467-005-01	Client Contact:	Possily Sentes The Land
Well Location:	C-1	Client Phone:	Rosaly Santos-Ebaugh, RPM
Date:	0.40.00	Guerra none.	(407) 867-4121
Date.	8-12-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Western-IW-1I	Total Depth:	25.01
Screen Interval;	25.0 to 35.0' bls	Well Type: Construction:	2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

	TO THE TAX PROPERTY OF THE PRO
200+	GROUNDWATER APPEARANCE Murky/Cloudy
200+	
200+	
200+	
200+	
200+	
200+	Murky/Cloudy
30 minutes	
200+ NTUs	
165 gallons	
	SOIL SCREENING RESULTS (OVA-PID):
n sheli.	NEGOCIO (OVATID).
	200+ 200+ 200+ 200+ 200+ 200+ 30 minutes 200+ NTUs

STANDARD WELL DEVELOPMENT LOG

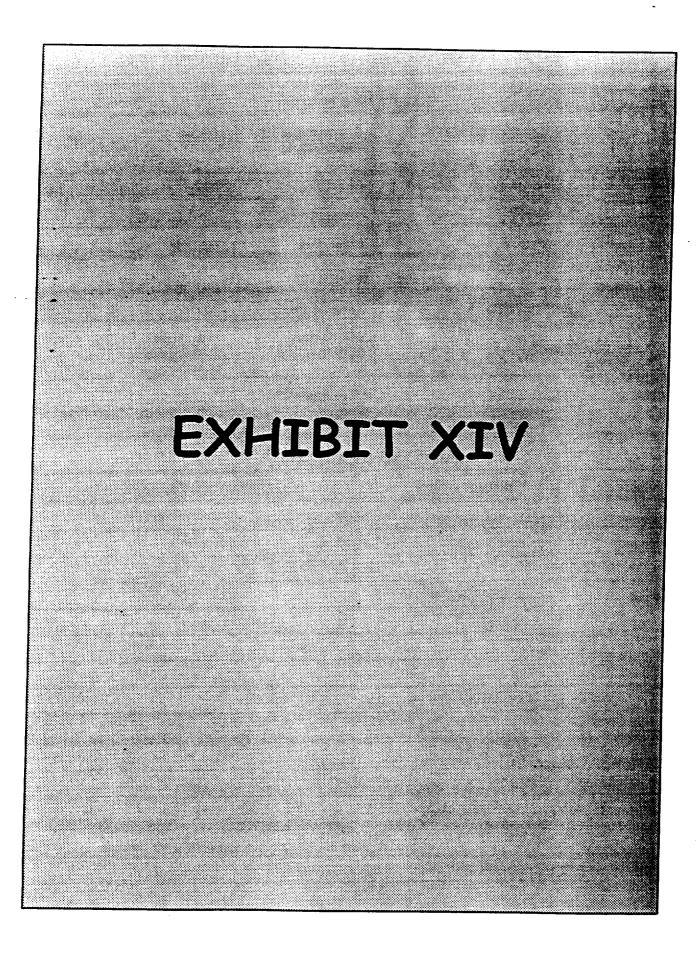
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-1	Client Phone:	(407) 867-4121
Date:	8-19-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Western-IW-1D	Total Depth:	50.0'
Screen Interval:	45.0 to 50.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
1:17 p.m.	No reading	Milky grey - opaque
1:20 p.m.	473	Cloudy grey
1:28 p.m.	106	
1:38 p.m.	25.6	
	<u>,</u>	
Total Development Time:	21 minutes	
Final Turbidity Reading:	25.6 NTUs	
Estimated Volume Removed	d: 50 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

REPORT NO. PAGE. 8-2.14

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS:

HAND AUGER FROM 0 FEET TO 4 FEET

BORING DESIGNATION:

G.S. ELEVATION (ft):

C-2 SECTION: TOWNSHIP: SHEET: 1 of 1

RANGE:

DATE STARTED:

8/5/98

WATER TABLE (ft): 0.0

DATE FINISHED:

8/5/98

DATE OF READING: 08/05/98

DRILLED BY:

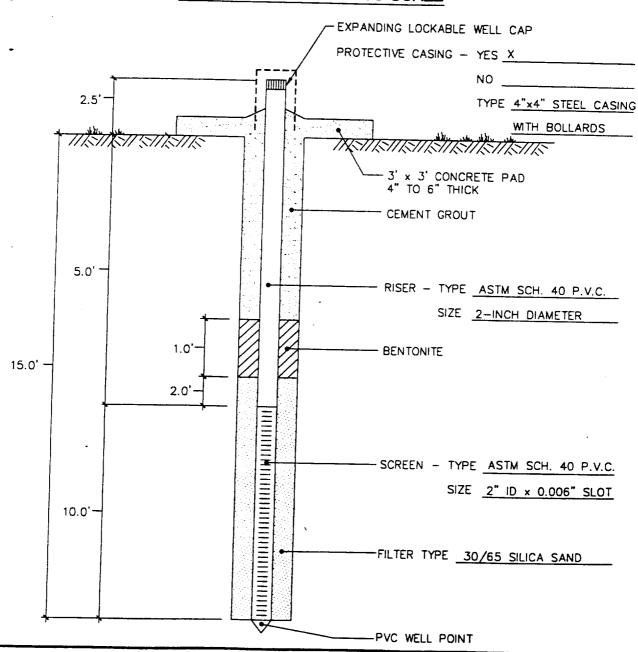
UES - ROCKLEDGE

	7.			,		EST. W.S.W.T.	(ft):	Т	YPE OF	SAMPLI	NG: ASTM	1 D-1586
DEPTH (FT.)	S A M P	BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	w.t.		DESCRIPTION	-200 (%)				(FT./	ORG.
	E	INCUENCIA!	F1.,		ι C				LL	PI	DAY)	(%)
0 -	+			Y	 	Gray fine SAND; with roots [SP]		-	-	ļ		-
	∄				111	Dark gray brown silty fine SAND [SM]	-					
	╧	•				dark brown						
5 -	M	3-4-5	9			gray						ļ
	X	4-4-4	8		· · · ·	Loose, light gray fine SAND; with silt	۱			1		
- 10	X	4-4-5 4-5-4	9 9			Loose, gray brown fine SAND; with shell and silt [SP]						
- 10	\prod											
-	1				• : • : •							
45	V	7-10-12	22			non-fine and distance de	-					
15 —						BORING TERMINATED AT 15 FEET			1			
-]			İ								
-	1								Ì			
20 —	1 1							†				
-				1	Ī							
_												
25											••••	••••••
-												
-		1				•						
30						· · · · · · · · · · · · · · · · · · ·		ļ	ļ			
=					-			Ę				
				ŀ								
35 —												
1											ļ	
+												
40									ļ	ļ		
j				ļ								
-}		1				}						
45												
-						·				- 1	j	
7												
50										}		
~7								***************************************				
55												



PROJECT NO 18467-205-01
REPORT NO:
PAGE:

PROJECT: INSTALL	ATION 52 MONITOR WELLS		
KENNED	Y SPACE CENTER, FLORIDA		
CLIENT: NASA J	J-D	DATF.	08/05/98
WELL NUMBER: WES	STERN-IW-2S LOCATION: C-2		
INSTALLED BY: UES	- ROCKLEDGE		



STANDARD WELL DEVELOPMENT LOG

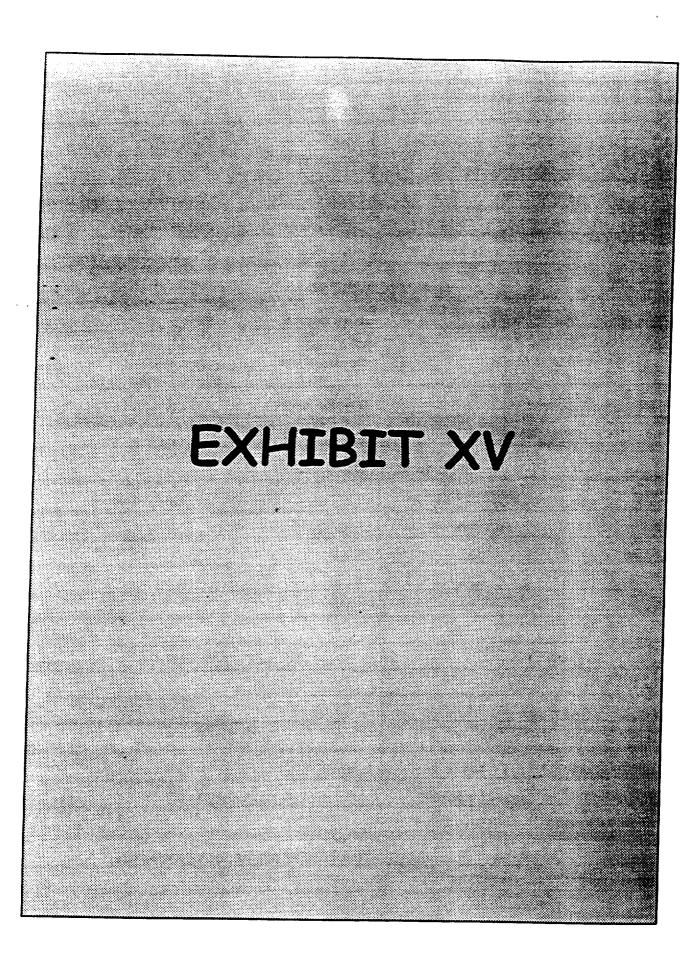
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-2	Client Phone:	(407) 867-4121
Date:	8-5-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Western-IW-2S	Total Depth:	15.0'
Screen Interval:	5.0 to 15.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
8:35 a.m.	1,000+	Dark grey
8:41 a.m.	137	Cloudy
8:47 a.m.	62.6	Cloudy
8:50 a.m.	15.8	.Clear
		·
Total Development Time:	15 minutes	
Final Turbidity Reading:	15.8 NTUs	
Estimated Volume Remove	d: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





THALLING OCIENCES **BORING LOG**

REPORT NO. PAGE. B-2.15

PROJECT

25

30

35

40

45

50

55

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS:

BORING DESIGNATION:

G.S. ELEVATION (ft):

SECTION:

C-3 TOWNSHIP: SHEET. 1 of 1

RANGE:

DATE STARTED:

8/12/98

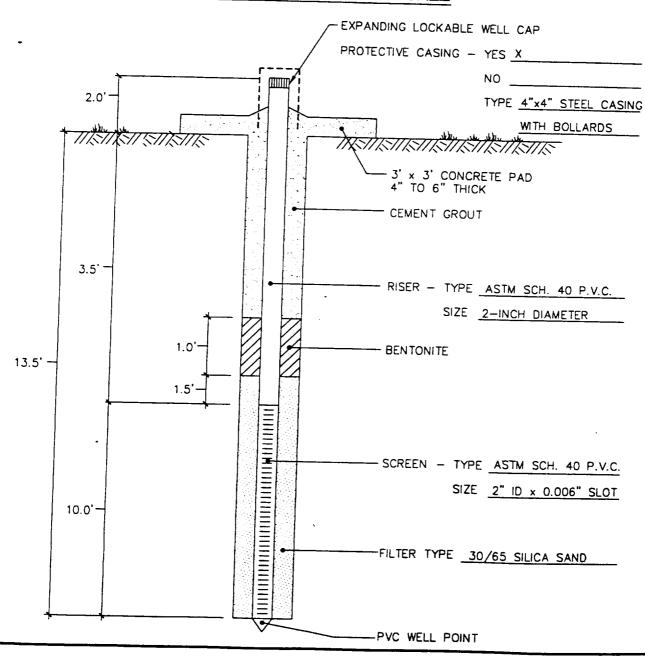
DATE FINISHED: 8/12/98

WATER TABLE (ft): 5.5 HAND AUGER FROM 0 FEET TO 4 FEET DATE OF READING: 08/12/98 DRILLED BY: **UES - ROCKLEDGE** EST. W.S.W.T. (ft): TYPE OF SAMPLING: ASTM D-1586 Ă **BLOWS ATTERBERG** DEPTH ORG. -200 PER 6" (BLOWS/ W.T. MC LIMITS DESCRIPTION (FT.) В (FT./ CONT. (%) INCREMENT (%) FT.) DAY) (%) LL 0 Gray fine SAND; with roots [SP] Gray brown clayey fine SAND [SC] Light brown cemented fine SAND [SP] Medium dense, light brown fine SAND; with 6-6-11 ..:1.7... gravel [SP] 10-12-15 27 - light gray brown 9-10-13 26 --- light gray 7-8-10 18 - trace shell 10 5-5-5 10 Loose, gray fine SAND; with silt [SP-SM] 15 Medium dense, gray fine SAND; with gravel 3-4-8 12 20 13-20-20 40 Dense, light gray fine SAND [SP] 13-30-24 54 Very dense, gray fine SAND; with shell [SP] 7-6-5 Medium dense, grayish fine SAND; with shell 11 ISPI 3-3-3 6 Loose, gray silty fine SAND; with shell [SM] 7-6-4 10 Loose, gray silty fine SAND; with shell [SM] 5-4-3 7 BORING TERMINATED AT 50 FEET



PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT: INSTALLATION 52 MONITOR WELLS	T: INSTALLATION 52 MONITOR WELLS		
KENNEDY SPACE CENTER, FLORIDA			
CLIENT: NASA JJ-D	DATE: 08/19/98		
WELL NUMBER: WESTERN-IW-3S LOCATION: C-3	DAIL. <u>-997.10788</u>		
INSTALLED BY: UES - ROCKLEDGE			

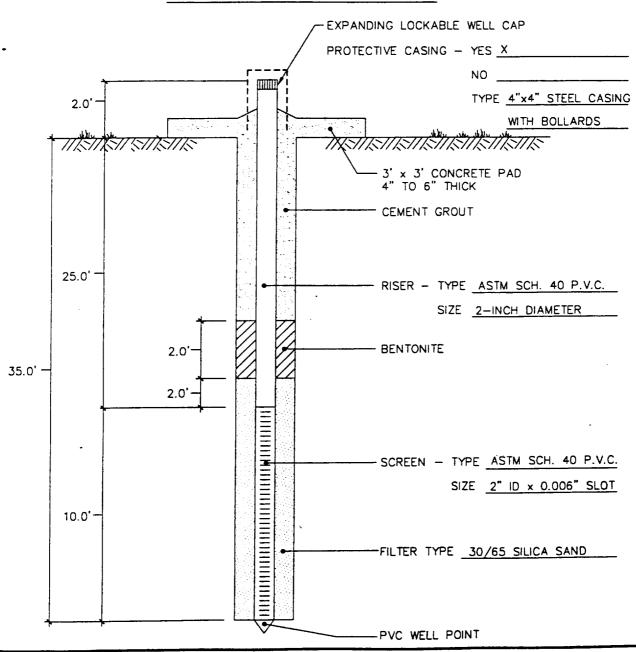




PROJECT NO.:	18467-005 - 01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: .	NASA JJ-D	DATE: 08/19/98
OLILITY	BER: WESTERN-IW-3I LOCATION: C-3	
INSTALLED	BY: UES - ROCKLEDGE	

WELL DIAGRAM - NOT TO SCALE

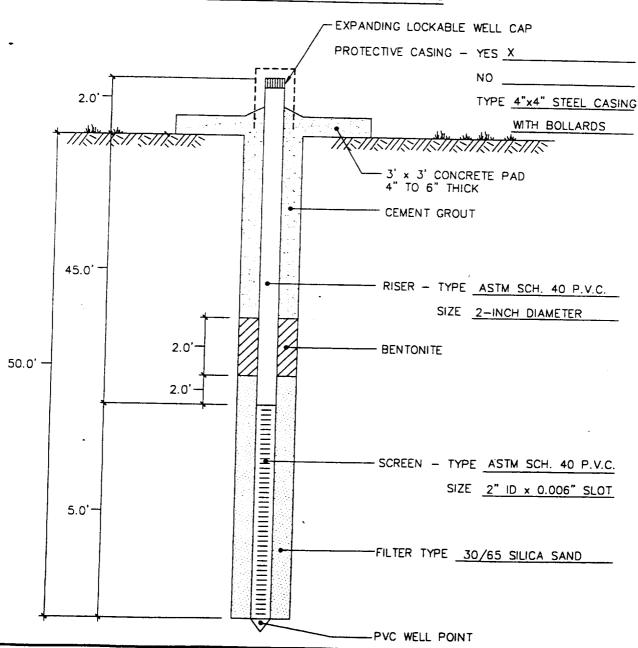


, , , , , , ,



PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS				
	KENNEDY SPACE CENTER, FLORIDA				
	NACA III D				
CLIENT: _	NASA JJ-D	•		DATE:	08/14/98
WELL NUMB	ER: WESTERN-IW-3D	LOCATION:	C-3		
INSTALLED	BY: UES - ORLANDO				



STANDARD WELL DEVELOPMENT LOG

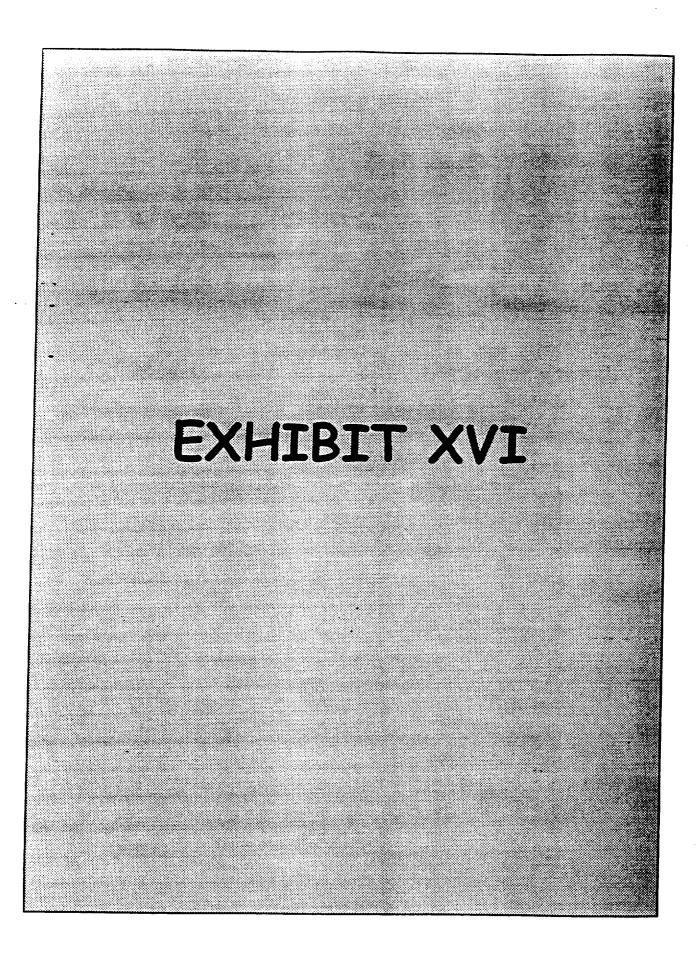
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-3	Client Phone:	(407) 867-4121
Date:	8-19-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Western-IW-3S	Total Depth:	13.5'
Screen Interval:	3.5 to 13.5' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
9:20 a.m.	638	
9:25 a.m.	312	
9:33 a.m.	114	
9:35 a.m.	83	
9:40 a.m.	44	
9:45 a.m.	18	
Total Development Time:	25 minutes	
Final Turbidity Reading:	19 NTUs	
Estimated Volume Removed	f: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





BORING LOG

REPORTING

PAGE: B-2.16

PROJECT

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET

BORING DESIGNATION:

SECTION:

C-4
TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/12/98

WATER TABLE (ft):

t): 4.6

DATE FINISHED:

8/12/98

DATE OF READING: 08/12/98

DRILLED BY:

UES - ORLANDO

EST. W.S.W.T. (ft):

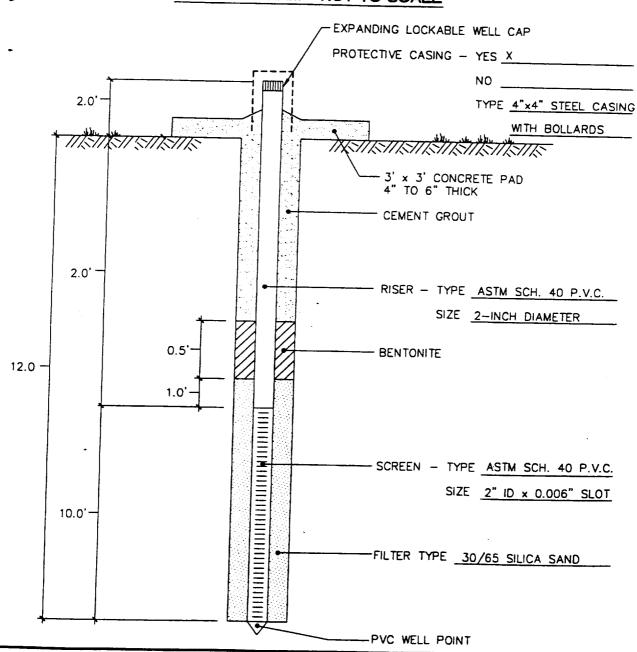
TYPE OF SAMPLING: ASTM D-1586

DEPTH A	BLOWS			S Y M B	DESCRIPTION	-200	мс		RBERG AITS	K (FT./	ORG.
(FT.) P	INCREMENT	FT.)		ļ		(%)	(%)	LL	PI	DAY) (%)	
0	5-6-6	12			Medium dense, light gray to white cemented SILT [ML]						
- 1	5-6-5 7-5-7	11	≖		with limestone						
3 T	3-4-3	7			Medium dense, light yellow silty fine SAND [SM]					••••••••••	
1	1-1-4	5			Loose, light green gray silty fine SAND; with broken shell [SM]						
10	2-1-1	2			Very loose, gray fine SAND; with silt						
15	2-3-3	6			Loose, gray fine SAND; with shells [SP]						
	7										
20	12-13-15	28			medium dense						
25	6-6-8	14		Δ Δ Δ Δ	Medium dense, SHELLS; with fine sand						
				Δ Δ Δ Δ							
30	2-2-2	4			Very loose, green fine SAND; with shells						
35	5-7-7	14			Medium dense, gray green silty fine SAND; with shell (SM)						
- - -									<u>.</u>		
40	4-3-3	6			loose, with seams of clay						
. 🔻	2-3-5	8									
45											
50	6-4-6	10			BORING TERMINATED AT 50 FEET						
-											
55											



PROJECT NO	:8487-005-01
REPORT NO.:	
DACE.	

PROJECT: INSTALLATION 52 MONITOR WELLS	
KENNEDY SPACE CENTER, FLORIDA	
CLIENT: NASA JJ-D	DATE: 08/19/98
WELL NUMBER: WESTERN-IW-4S LOCATION: C-4	
INSTALLED BY: UES - ROCKLEDGE	

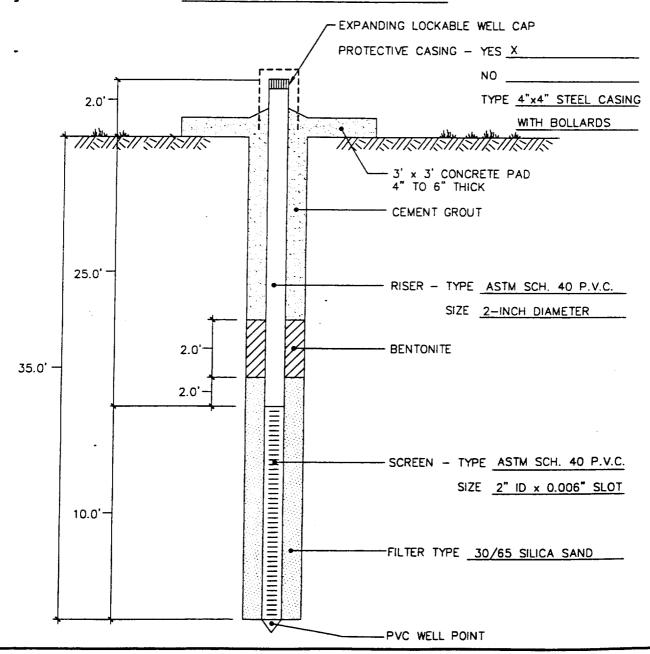




PROJECT NO	18467-005-01
REPORT NO.:	

PAGE:

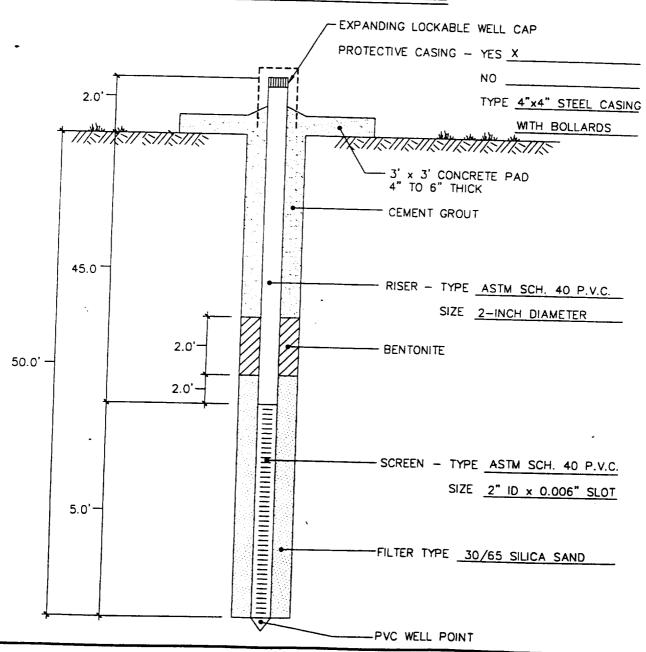
PROJECT:	INSTALLATION 52 MONITOR WELLS					
	KENNEDY SPACE CENTER, FLORIDA					
CLIENT: _	NASA JJ-D	DATE: <u>08/19/98</u>				
WELL NUME	BER: WESTERN-IW-4I LOCATION: C-4					
INSTALLED	BY: UES - ROCKLEDGE					





PROJECT NO	18467-008-01
REPORT NO.:	
PACE:	

DJECT: INSTALLATION 52 MONITOR WELLS					
KENNEDY SPACE CENTER, FLORIDA					
NACA III B					
CLIENT: NASA JJ-D	DATE: 08/13/98				
WELL NUMBER: WESTERN-IW-4D LOCATION: C-4					
INSTALLED BY: UES - ORLANDO	·				



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-4	Client Phone:	(407) 867-4121
Date:	8-19-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Westem-IW-4S	Total Depth:	12.0'
Screen Interva <u>l</u> :	2.0 to 12.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
8:15 a.m	1,000+	Grey, cloudy
8:19 a.m.	110	
8:30 a.m.	120	
8:35 a.m.	45	
8:45 a.m.	12	
	·	
Total Development Time:	30 minutes	
Final Turbidity Reading:	12 NTUs	
Estimated Volume Remove	d: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-4	Client Phone:	(407) 867-4121
Date:	8-19-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Western-IW-4I	Total Depth:	35.0'
Screen Interval:	25.0 to 35.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
10:50 a.m.	1,000+	Grey
10:55 a.m.	140	J. Grey
11:05 a.m.	55	
11:15 a.m.	19	Clear
·		
·		
Total Development Time:	25 minutes	
Final Turbidity Reading:	19 NTUs	
Estimated Volume Removed	: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
		1000000

STANDARD WELL DEVELOPMENT LOG

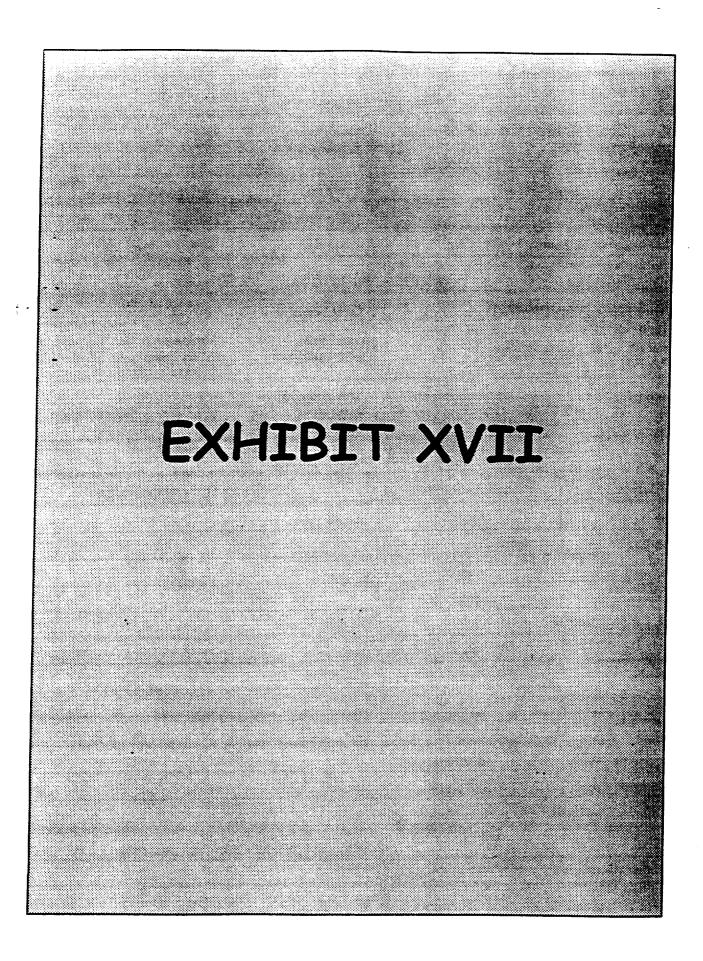
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-4	Client Phone:	(407) 867-4121
Date:	8-13-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Western-IW-4D	Total Depth:	50.0'
Screen Interval <u>:</u>	45.0 to 50.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

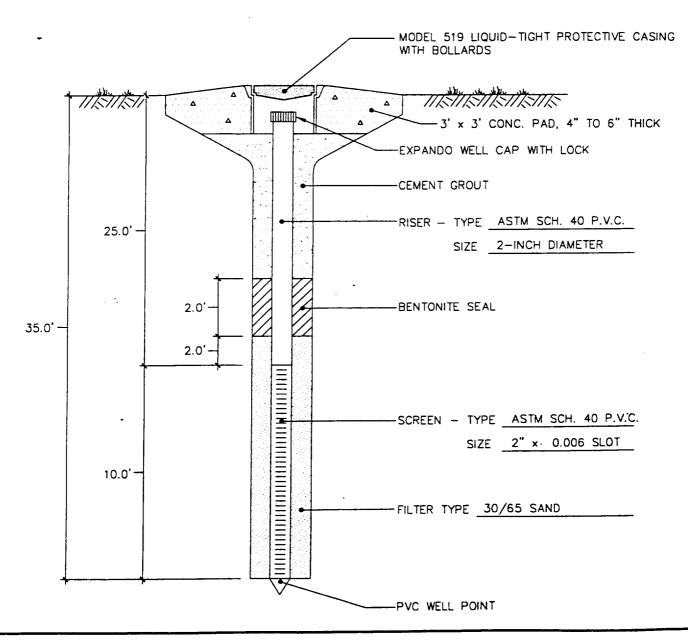
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
9:00 a.m.	783	Cloudy, grey
9:10 a.m.	312	
9:20 a.m.	148	
9:25 a.m.	132	
9:35 a.m.	87	. Murky
9:41 a.m.	63	Clear
9:45 a.m.	20.7	Clear
Total Development Time:	45 minutes	
Final Turbidity Reading:	20.7 NTUs	
Estimated Volume Remove	d: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





PROJECT NO	15467-005 - 01
REPORT NO	
PAGE:	

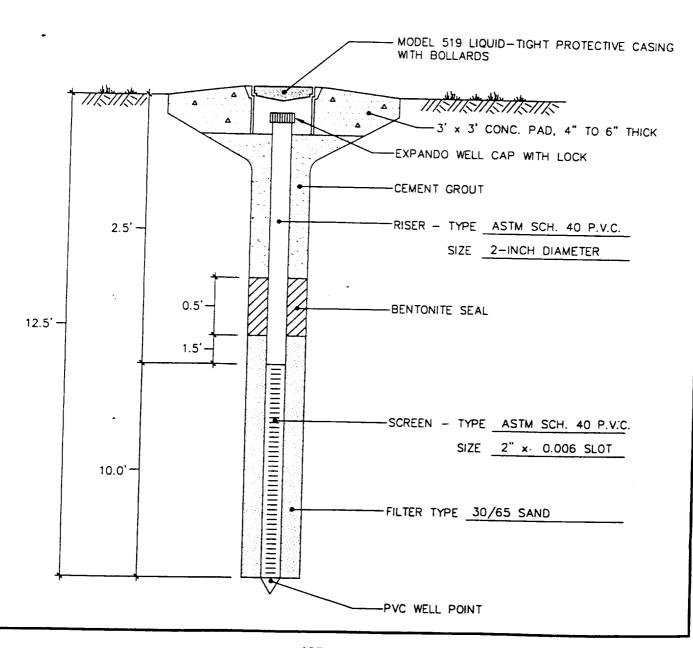
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
	NASA JJ-D	DATE: 08/03/98
CLIENT: _		DATE:
WELL NUMI	BER: MARSH-IW-51 LOCATION: D-5	
INSTALLED	BY: UES - ROCKLEDGE	





PROJECT NO.:	18467-005-01
REPORT NO.:	
PACE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/03/98
WELL NUMB	ER: MARSH-IW-5S LOCATION: D-5	
INSTALLED I	BY: UES - ROCKLEDGE	





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 18467-005-01 REPORT NO. B-2.22 PAGE:

PROJECT:

CLIENT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION: SECTION:

D-5 TOWNSHIP: SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/3/98

WATER TABLE (ft): 4.5

DATE OF READING: 08/03/98

DATE FINISHED:

8/3/98

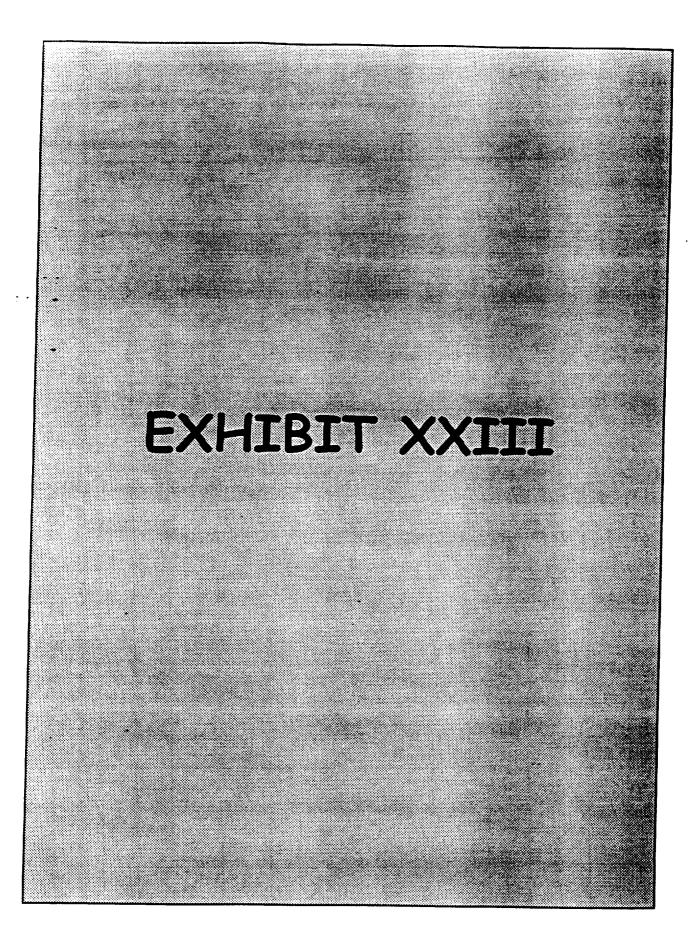
FST W S W T (#1)

DRILLED BY:

UES - ROCKLEDGE

TYPE OF SAMPLING: ASTM D-1596

		,	, <u></u>	,		EST. W.S.W.T. ((ft):	T	YPE OF S	AMPLIN	G: ASTM	D-1586
DEPTH (FT.)	P PER 6" (BLOWS/ W.T. B		DESCRIPTION	-200	MC	ATTERBERG LIMITS		K (FT./	ORG.			
	E	INCREMENT	FT.)		l L		(%)	(%)	LL	PI	DAY)	(%)
0 -	+					Grass and brown SAND						
	1			1	· : :	Light gray fine SAND [SP]	1					
	4					<u> </u>]					
5	X	2-1-2	3	¥		Very loose, dark brown silty fine SAND; with roots [SM]						
¥ -	-12	3-3-4	7			Loose, gray brown fine SAND; with silt and		***************************************			••••••	•••••
,	X	3-3-3	6			roots [SP-SM]						
	-	- 4-5-6	11			gray brown Medium dense, light gray fine SAND [SP]						
10 -					,	Medicin dense, light gray fine SAND [SP]				······		•••••
	-	ı										
•	H				: : : :							
15 -	H	8-4-5	9			Loose, gray fine SAND; and shell [SP]						
] [ļ	· · · ·							
-	┧			ŀ	• : : :							
20 —	M	5-2-5	7		ΔΖ	Loose, gray SHELL; with sand				1		
	-			ļ	ΔΔ					Ī		*****************
-	1		- 1		ΔΔ		İ				Ī	
-	X	10-12-13	25	-	Δ.	Medium dense, gray fine SAND; and shells	ĺ		l		j	
25 —	H	10-12-10	- 25			(SP)			-			
-	1		1	ŀ	<i>:::</i> ·							
_	Ы		-	ļ.	•							
30 —	Α.	15-14-23	37			-						
_	1			ŀ	<i>::::</i>	•				İ		
-		ļ	İ	ļ.			1	İ			1	
35 —	M.	7-10-17	27	ŀ							.]	
-			1	ŀ	:::							
			ŀ		$\cdot \cdot \cdot \cdot$							
4	\forall	11-10-12	22	Ė		Medium dense, gray fine SAND ISPI	1]				
40	4	11-10-12	- 22			Medidin dense, gray line SAND (SF)				······································		
4			İ	- 1	:-:·					ļ		
	\forall				\cdots							
45 -	Д	5-6-5	11			Madissa da se se cabillo de dia						
				<u> </u>		Medium dense, gray fine SAND; with silt [SP-SM]			ļ			
4				F:	$ \cdot $		-				- 1	
50	XI	5-5-5	10	<u>-</u>		Loose, gray fine SAND; with shells [SP]	-					
-	7					BORING TERMINATED AT 50 FEET	1			-		
7												
55 —			·····				······································					



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-4	Client Phone:	(407) 867-4121
Date:	8-17-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Marsh-IW-41	Total Depth:	33.0'
Screen Interval:	23.0 to 33.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
12:05 p.m.	No reading	
12:08 p.m.	39.6	
12:10 p.m.	52.1	
12:20 p.m.	253	
12:24 p.m.	67.3	
12:30 p.m.	37.5	
12:35 p.m.	32	·
12:40 p.m.	23.2	
Total Development Time:	35 minutes	
Final Turbidity Reading:	23.2 NTUs	
Estimated Volume Removed		
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-4	Client Phone:	(407) 867-4121
Date:	8-17-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

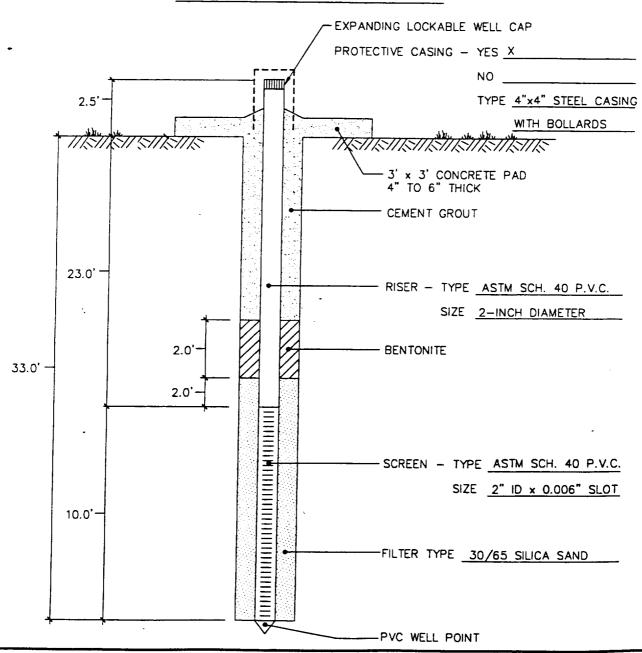
Well ID:	Marsh-IW-4S	Total Depth:	12.5'
Screen Interval:	2.5 to 12.5' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" × 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
7:30 a.m.	No reading	
7:42 a.m.	72.9	
8:04 a.m.	194	
8:09 a.m.	142	
Stopped - rain		
11:10 a.m.	No reading	
11:18 a.m.	123	
11:31 a.m.	83.8	
11:45 a.m.	49.3	
12:05 p.m.	22.3	
Total Development Time:	94 minutes	
Final Turbidity Reading:	22.3 NTUs	
Estimated Volume Removed:	70 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):



PROJECT NO.:	18467-005-0
REPORT NO.:	
PAGE:	

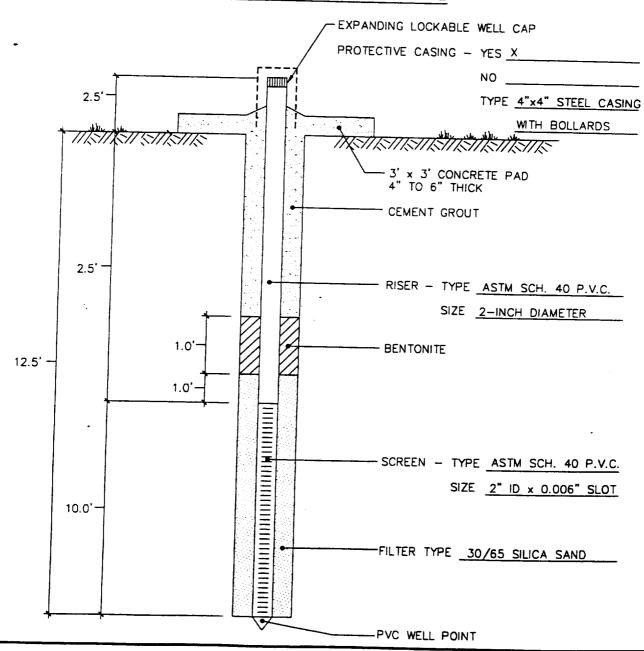
PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
CLIENT:	NASA JJ-D	DATE:	08/17/98
WELL NUME	BER: MARSH-IW-41 LOCATION: D-4		
INSTALLED	BY: UES - ORLANDO		





PROJECT NO : 18467-005-01
REPORT NO.:
PAGE:

PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
0.15	NASA LI D		
CLIENT: _	NASA JJ-D	DATE:	08/17/98
WELL NUMB	ER: MARSH-IW-4S LOCATION: D-4		
INSTALLED	BY: UES - ORLANDO		





UNIVERSAL ENGINEERING SCIENCES BORING LOG

REPORT NO.:
PAGE. B-2.22

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET. MARSH

BORING DESIGNATION: D-4

SECTION: TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/17/98

WATER TABLE (ft): 3.0

DATE FINISHED:

8/17/98

DATE OF READING: 08/17/98

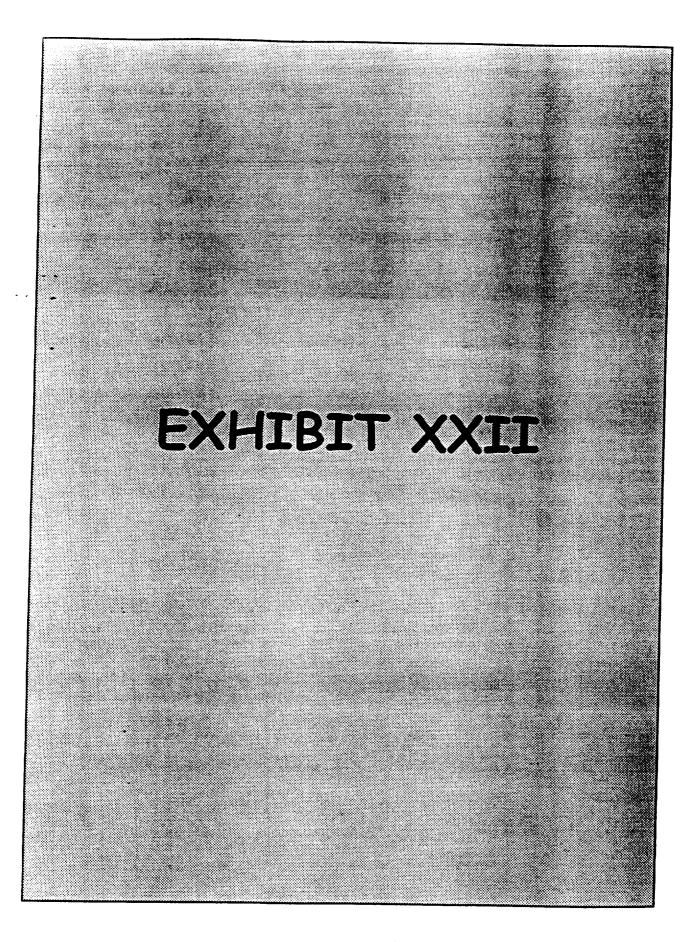
DRILLED BY:

UES - D.B.

EST. W.S.W.T. (ft):

TYPE OF SAMPLING: ASTM D-1586

SA	TH M BLOWS N Y M ATTERBERG					-					
DEPTH M (FT.) P L	o reno j	" (BLOWS/ \	LOWS/ W.T.	ŏ	DESCRIPTION	-200 (%)	MC (%)	LIMITS		K (FT./ DAY)	ORG. CONT (%)
			-				 	LL	PI		1,70,
0	· ·· · · · · · · · · · · · · · · · · ·		<u> </u>	· · ·	Greenish gray fine SAND [SP]		 	+	ļ	<u> </u>	
4											
1_	•	_	_								
5 - 🛚	1-1-2	3			with shells						
	2-2-2	4				j					
$\overline{\mathbb{X}}$	3-5-5	10									
10	- 3-1-1	2		• : • : •	loose	-	1				
10 77					very loose					•••••	
-		ļ									
H		_]									1
15	4-4-4	8	[•••••	
]			ŀ	:::	loose, brown						
-			-						ı		
20	5-6-6	12									İ
			ŀ	:::	medium dense, little lighter						
] [ļ.	$ \cdot \cdot $							
\	10-21-22	43	Ŀ							İ	
25				::::	dense, greenish gray with shells		ļ	ļ	·····		
41		1		$\cdot : \cdot :$					ľ		
			ŀ							į	
30	8-19-21	40			•		ļ				
					more shells						
			ŀ					}			
35	10-15-12	27		• • • •							
"					BORING TERMINATED AT 35 FEET				···········		
				- 1					į		
]		j		- 1						Ī	
40											
] [-				1 1		·		-	
4								ļ		İ	
45											
4				İ	١		-		1		
11		1									
_] [-		
50						 					
41								-			
]							ľ			-	
55 —							<u>]</u>				



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-3	Client Phone:	(407) 867-4121
Date:	8-11-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Marsh-IW-3I	Total Depth:	34.0'
Screen Interval <u>:</u>	24.0 to 34.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
		GROUNDWATER AFFEARANCE
9;05 a.m.	200+	
	85.7	
	62.4	
	32.6	
9:20 a.m.	14.5	
Total Development Time:	15 minutes	
Final Turbidity Reading:	14.5 NTUs	
Estimated Volume Remove	d: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-3	Client Phone:	(407) 867-4121
Date:	8-10-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

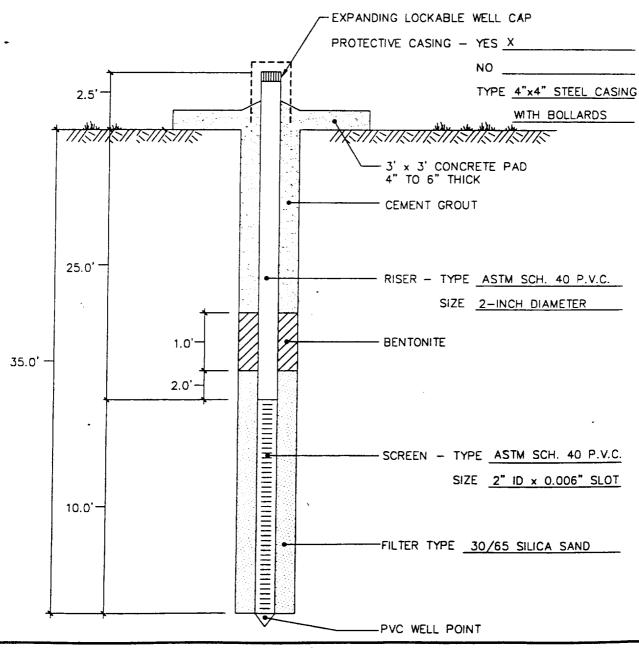
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe
Screen Interval;	2.5 to 12.5' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Well ID:	Marsh-IW-3S	Total Depth:	12.5'

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
11:10 a.m.	200+	
	135.7	
	73.4	
	34.6	
11:25 a.m.	15.3	
		
Total Development Time:	15 minutes	
Final Turbidity Reading:	15.34 NTUs	
Estimated Volume Removed:	110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
-1-2		THE REPORT OF THE PART OF THE



PROJECT NO.	8487-	225	- ĵ.	
REPORT NO.:				
PAGE:				

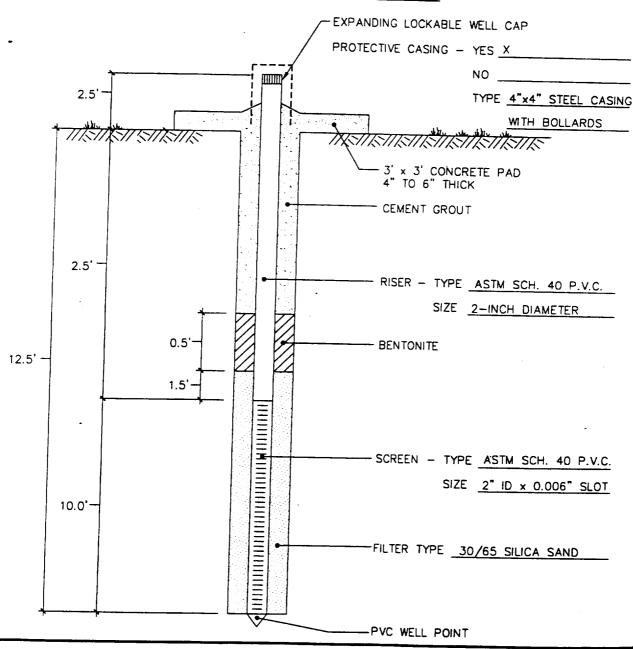
PROJECT:	INSTALLATION 52 MONITOR WELLS											
	KENNEDY SPACE CENTER, FLORIDA											
CLIENT: _	NASA JJ-D	DATE: _08/11/98										
WELL NUME	BER: MARSH-IW-3I LOCATION: D-3											
INSTALLED	BY: UES - DAYTONA BEACH											





PROJECT NO. :8467-005-01
REPORT NO.:
PAGE:

PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
	NACA III D		
CLIENT:	NASA JJ-D	DATE:	08/10/98
WELL NUMB	ER: MARSH-IW-3S LOCATION: D-3		
INSTALLED E	BY: UES - DAYTONA BEACH	 	





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

13407-005-01 REPORT NO PAGE: B-2.21

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS:

BORING DESIGNATION: TOWNSHIP:

SECTION:

D-3

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/10/98 8/10/98

WATER TABLE (ft):

DATE OF READING: 08/10/98

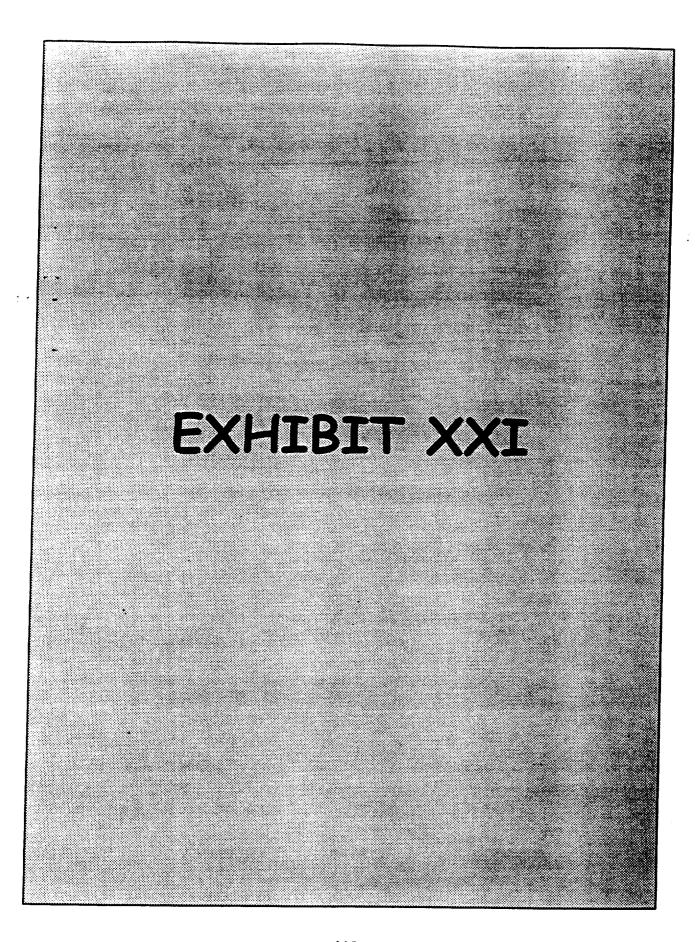
DATE FINISHED: DRILLED BY:

UES - D.B.

EST. W.S.W.T. (ft):

TYPE OF SAMPLING: ASTM D-1586

DEPTH M	BLOWS PER 6"	N (BLOWS/	w.t.	S Y M B	DESCRIPTION	-200	мс		rberg IITs	K (FT./	ORG.
(F1.) E	INCREMENT			ρ		(%)	(%)	LL	PI	DAY)	(%)
0-	ļ			ļ.,,,,							
-[2	4-5-3	8			Loose, gray and light gray fine SAND; with clay [SP-SC]		!				
. 🏵	2-1-4	5		• •	Loose, dark brown muck trace fine SAND; \and organics (SP)	7					
	3-2-3 3-4-4	5 8			Loose, gray-brown fine SAND; with roots	/	•••••••••••••••••••••••••••••••••••••••			••••••	
-[X	3-4-5	9		• • • •	Loose, gray-brown fine SAND; trace silt						
10	- 3-2-3	5			Loose, gray fine SAND; trace silt [SP-SM]	1					
					Very loose, dark gray fine SAND; trace silt [SP-SM]		••••••••••••				
15	2-1-2	3		· · ·	Medium dense, dark gray fine SAND; with	4		ļ			· · · · • · · · · · · · · · · · · · · ·
1	770	15			shell [SP]						
20	7-7-8	15	······ė	1111	Very soft, dark gray CLAY; with shell	-					
25	1-0-1	1									
30	0-1-1	2			Very soft, dark gray CLAY; trace shell [CL]						
	2.2.4				Firm, dark gray CLAY; with sand and shell [CH]						
35	3-2-4	6									
40	2-1-2	3.			soft						
	3-4-2	6									
15					firm						
50 7	2-2-2				soft BORING TERMINATED AT 50 FEET					4	•••••
55								1			



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-2	Client Phone:	(407) 867-4121
Date:	8-6-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

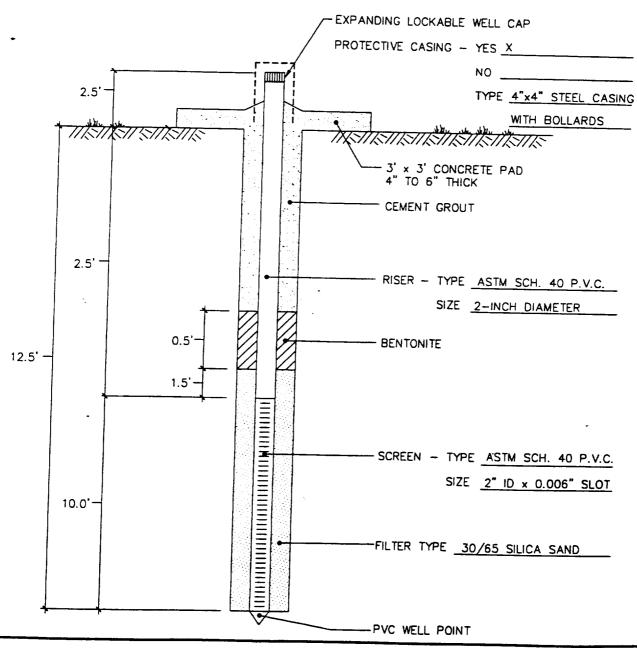
Well ID:	Marsh-IW-2S	Total Depth:	12.5'
Screen Interval;	2.5 to 12.5' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
10:57 a.m.	200+	Silty/Cloudy
11:06 a.m.	200+	
11:15 a.m.	200+	
11:20 a.m.	136.7	Murky
11:38 a.m.	65.4	Clearer/Cloudy
11:47 a.m.	33.8	
11:55 a.m.	21.9	Clear
12:00 p.m.	17.9	
Total Development Time:	63 minutes	
Final Turbidity Reading:	17.9 NTUs	
Estimated Volume Remove	d: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):



PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
	NACA III D	
<u></u>	NASA JJ-D	DATE: 08/06/98
	ER: MARSH-IW-2S LOCATION: D-2	
INSTALLED I	BY: UES - DAYTONA BEACH	





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 18467-005-01 REPORT NO PAGE. B-2.20

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION:

AS SPECIFIED BY CLIENT

REMARKS:

BORING DESIGNATION:

SECTION:

D-2 TOWNSHIP: SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/6/98 8/6/98

WATER TABLE (ft): 0.0

DATE FINISHED:

DATE OF READING: 08/06/98

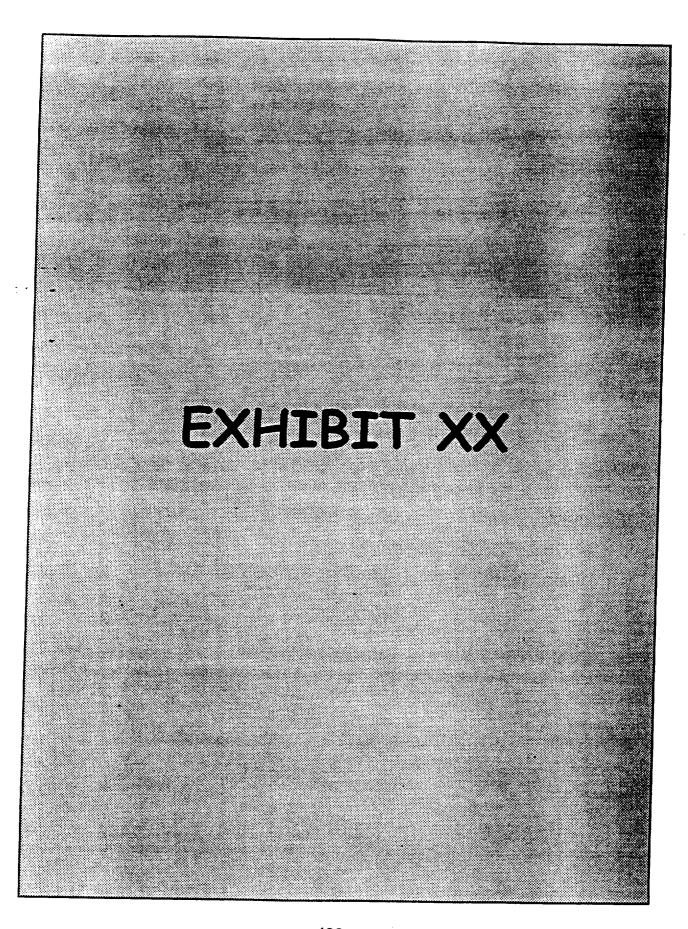
DRILLED BY:

UES - D.B.

EST. W.S.W.T. (ft):

TYPE OF SAMPLING: ASTM D-1586

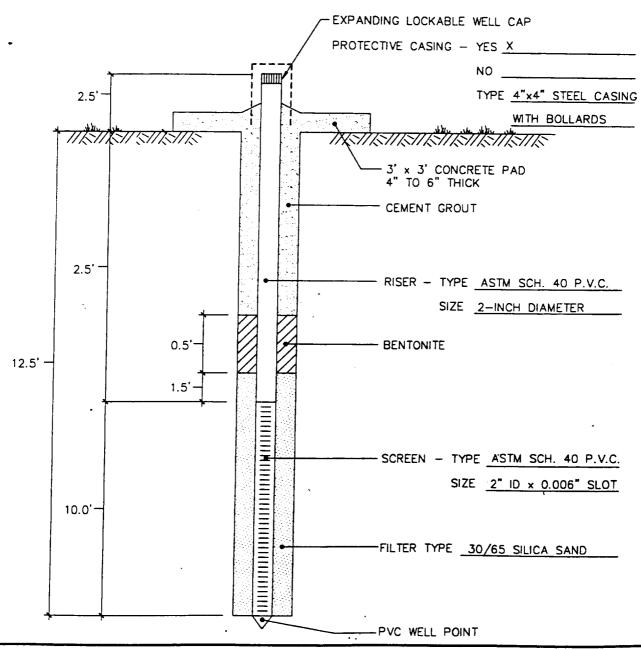
DEPTH (FT.)		BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	w.t.	, ,	M B DESCRIPTION -200 MC LIMITS						
	È	STONEMEN!	F1.)		° C		(70)	(00)	LL	PI	DAY)	(%)
0.	+i				 	Loose, gray-light gray fine SAND; some shell						
	$\overline{\mathbb{A}}$	2-3-4	7		· · · ·	[57]			ĺ	1 .		
	\square	3-3-5	8 .			Firm, dark gray CLAY [CH]	1	İ				
5 -	-M	1-1-2	3			Very loose, dark gray SILT; trace sand [ML]	-					
:	M	1-1-1	2			dark brown-dark gray						
	-	1-1-1	2		• . • . •	dark gray, trace shell						
10-	14	1-1-2	3			Very loose, dark gray fine SAND; with silt [SP-SM]						
	1				Δ Δ						1	
	-					Medium dense, light gray SHELL; trace sand						[
15 -	1	5-8-11	19		Δ <u>4</u>		1					
	$\dashv \mid$					BORING TERMINATED AT 15 FEET	-					
•	7				İ							
20 <i>~</i>	1											
								1				****************
-	1						l					
25 -	1	1										
25 -	7						••••••	· 				
-	1				ĺ			1				
	1					_					1	
30 —	1 1		: -						·			
-				İ		·						
					l							
35	-							ļ				
-				-	ļ			į				
4				ł				!			[
40			•							<u> </u>		
7											1	
								!		-	1	
45												
1											- 1	
-						1						
50							<u> </u>					
4										1		
7												
55 —	1								Ī		}	
							••••••		····	······································		





PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/06/98
WELL NUMB	ER: MARSH-IW-2S LOCATION: D-2	
INSTALLED	BY: UES - DAYTONA BEACH	



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-1	Client Phone:	(407) 867-4121
Date:	8-4-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Marsh-IW-1D	Total Depth:	50.0'
Screen Interval;	45.0 to 50.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
3:00 p.m.	234	Milky grey
3:15 p.m.	198	Milky grey
Stopped		Pumped well dry
3:40 p.m.	No reading	
3:45 p.m.	41	
Stopped		Pumped well dry
4:02 p.m.	No reading	
4:10 p.m.	47	Murky grey
4:25 p.m.	52	
4:36 p.m.	56	
Total Development Time:	54 minutes	
Final Turbidity Reading:	56 NTUs	
Estimated Volume Removed	l: 35 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-1	Client Phone:	(407) 867-4121
Date:	8-4-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Marsh-IW-1I	Total Depth:	35.0'
Screen Interval:	25.0 to 35.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
10:45 a.m.	200+	
10:50 a.m.	200+	
10:54 a.m.	132.5	
10:58 a.m.	67.5	
11:02 a.m.	27.3	
11:05 a.m.	14.7	
Total Development Time:	20 minutes	
Final Turbidity Reading:	14.7 NTUs	
Estimated Volume Removed	d: 80 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-1	Client Phone:	(407) 867-4121
Date:	8-3-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

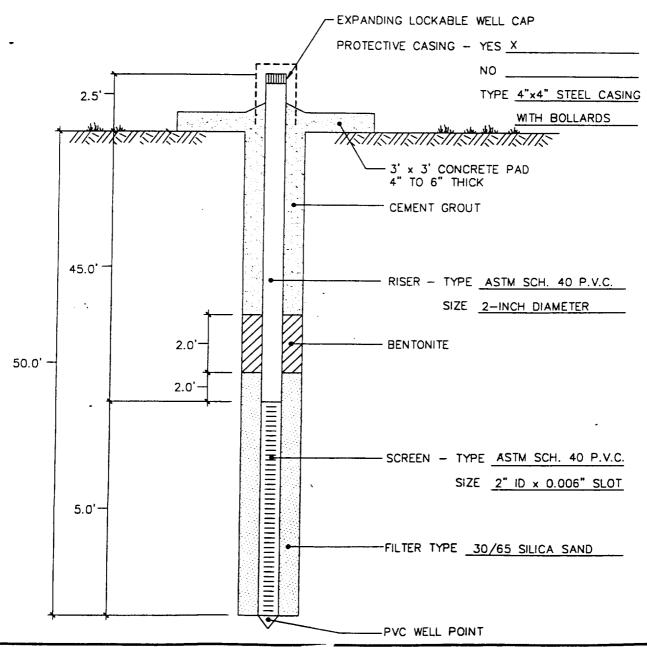
Well ID:	Marsh-IW-1S	Total Depth:	13.5'
Screen Interval:	3.5 to 13.5' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
4:45 p.m.	200+	
4:48 p.m.	57.2	
4:52 p.m.	31.9	
5:00 p.m.	15.2	
· .		
·		
Total Development Time:	15 minutes	
Final Turbidity Reading:	15.2 NTUs	
Estimated Volume Removed	: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):



PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

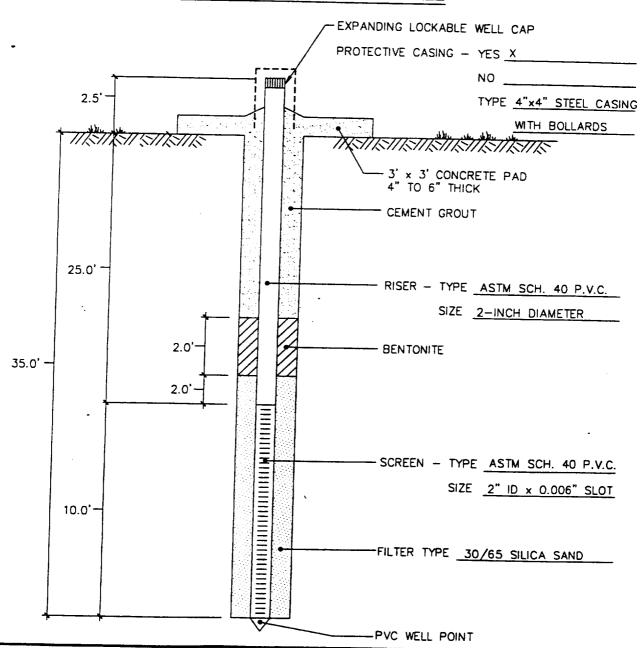
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: _08/07/98
WELL NUME	BER: MARSH-IW-1D LOCATION: D-1	
INSTALLED	BY: UES - ORLANDO	





PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

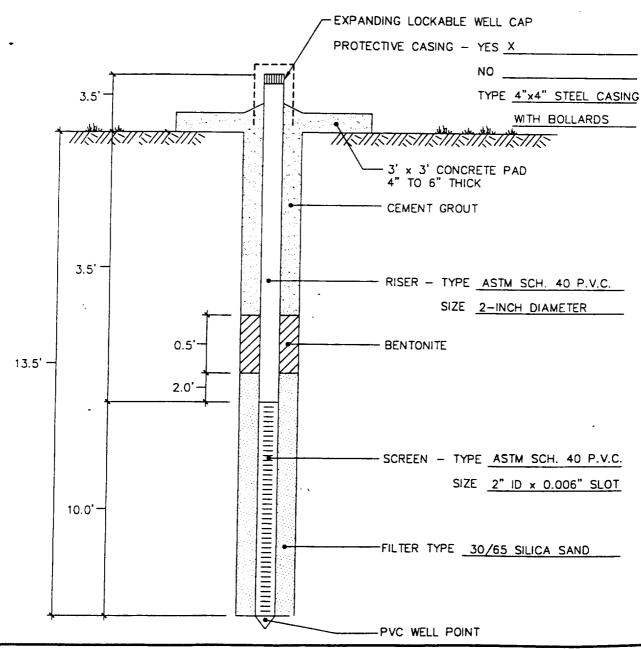
PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/04/98
WELL NUMB	ER: MARSH-IW-11 LOCATION: D-1	





PROJECT NO.:	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS		
	KENNEDY SPACE CENTER, FLORIDA		
CLIENT: _	NASA JJ-D	DATE: <u>C</u>	08/03/98
WELL NUMB	ER: MARSH-IW-1S LOCATION: D-1		
INSTALLED	BY: UES - DAYTONA BEACH		





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

PROJECT NO 18467-005-01 REPORT NO PAGE: B-2.19

PROJECT:

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION:

AS SPECIFIED BY CLIENT

REMARKS:

BORING DESIGNATION:

SECTION:

D-1 TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/3/98

WATER TABLE (ft): 5.3

DATE FINISHED:

8/3/98

DATE OF READING: 08/03/98

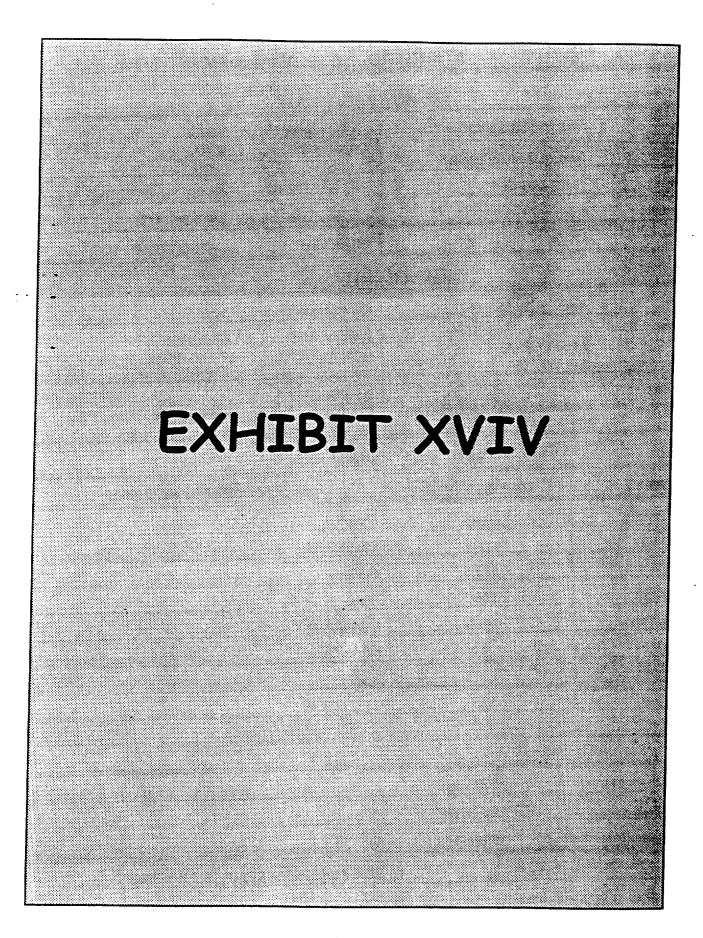
DRILLED BY:

UES - D.B

EST. W.S.W.T. (ft):

TYPE OF SAMPLING: ASTM D-1506

	51	T	T	S	EST. W.S.W.1	. (rt):	T	YPE OF	SAMPLIN	IG: ASTI	M D-1586
DEPTH M	PER 6"	N (BLOWS/ FT.)	W.T.	S Y M B O	DESCRIPTION	-200 (%)	MC (%)		RBERG MITS	K (FT./	ORG.
	<u> </u>			<u> </u>				LL	PI	DAY)	(%)
0				· · ·	Loose, dark gray fine SAND; with silt						
1 -2	2-3-2	5		• • •	[SP-SM]		}	1		-	
	2-3-3	6			·]						1
5 - X	3-3-4	7		·	light gray		1	1			
│	2-2-3	5		• :- :	blue gray			•••••••••			
	2-1-3	4		• •	dark gray with some silt	\mathcal{A}					
	3-4-5	9	-		Very loose, dark gray fine SAND; with silt	4					
10			·······[:		Loose, dark gray fine SAND; with silt trace						
-					Loose, gray SHELL	_					
		1			Very loose, blue dark gray CLAY: with shell	1			1		
15	2-1-1		<u>/</u>		(CH)						}
-			[·]		Loose, dark gray fine SAND; trace silt and shell [SP-SM]	1	1	······	·····	2	
]			[3.1011 (37-310)		1 1				
\	2-2-5			• :						ł	
20 -					Loose, dark gray fine SAND; some shell [SP]	-				7	
			[:	$\cdot : \cdot \mid$	(C,)						
4				: -:						ł	
25 —	9-11-15		<u>. </u>							26	
1				$\cdot \mid \cdot \mid$	Medium dense, dark gray fine SAND; some silt [SP]				·····		
-[_]										-	ı
30	11-9-15		[::	$ \cdot $					l		1
			. · .		Very loose, dark gray fine SAND; with silt	·····				24	
					trace shell [SP-SM]					İ	1
-	221						1				
35	2-2-1		777		Soft, dark gray CLAY; with shell [CH]					3	
]					, gray CDAT, WILL SHAIL [CM]						
-						-	1				- 1
40	2-2-2						1				1
4	•	T								4	
] [1										- 1
-	1-2-2	1						}			- 1
45	1-2-2			/						4	
7							1				
-											
50	2-2-3						1			5	l
- 41				77	BORING TERMINATED AT 50 FEET					<u></u>	
1					1						j
4											- 1
55 —											
					401						



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION	OJECT IN	FORMATIO	N
---------------------	----------	----------	---

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-6	Client Phone:	(407) 867-4121
Date:	8-3-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

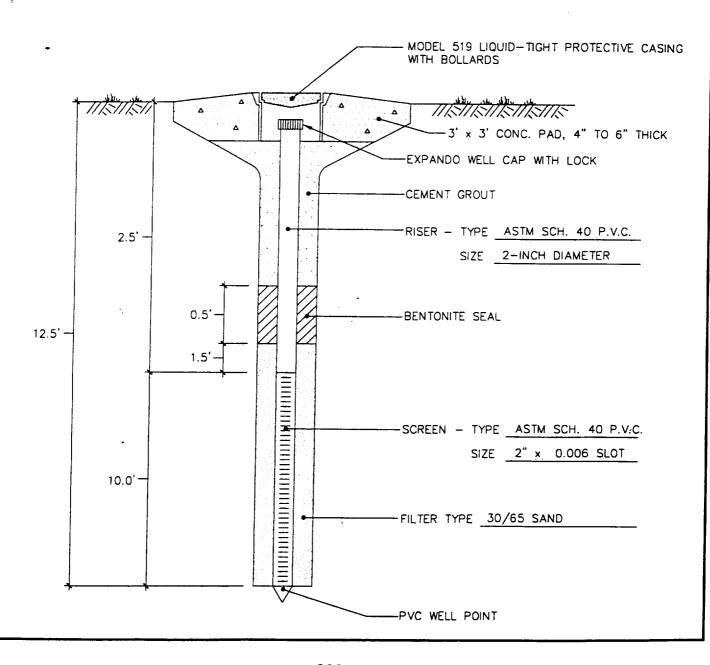
Well ID:	Western-IW-6S	Total Depth:	12.5'
Screen Interval:	2.5 to 12.5' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
11:55 a.m.	200+	Milky
12:05 p.m.	200+	Milky/Cloudy
12:12 p.m.	170.5	Cloudy
12:20 p.m.	200+	Milky
12:25 p.m.	101.3	. Murky/Clear
12:28 p.m.	79.2	Murky/Clear
·		
Total Development Time:	33 minutes	
Final Turbidity Reading:	79.2 NTUs	
Estimated Volume Removed	: 165 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
Well screen installed in sand w	rith shell.	, , , , , , , , , , , , , , , , , , , ,



PROJECT NO	18467005-01
REPORT NO.	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT:	NASA JJ-D	DATE: 08/03/98
WELL NUME	BER: WESTERN-IW-6S LOCATION: C-6	
INSTALLED	BY: UES - DAYTONA BEACH	





OTT A FLICOUS BINDHAFFUNA DOLENCED

BORING LOG

REPORT NO. PAGE. B-2.18

PROJECT.

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION:

SECTION:

C-6 TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

8/3/98

WATER TABLE (ft): 3.8

DATE FINISHED:

8/3/98

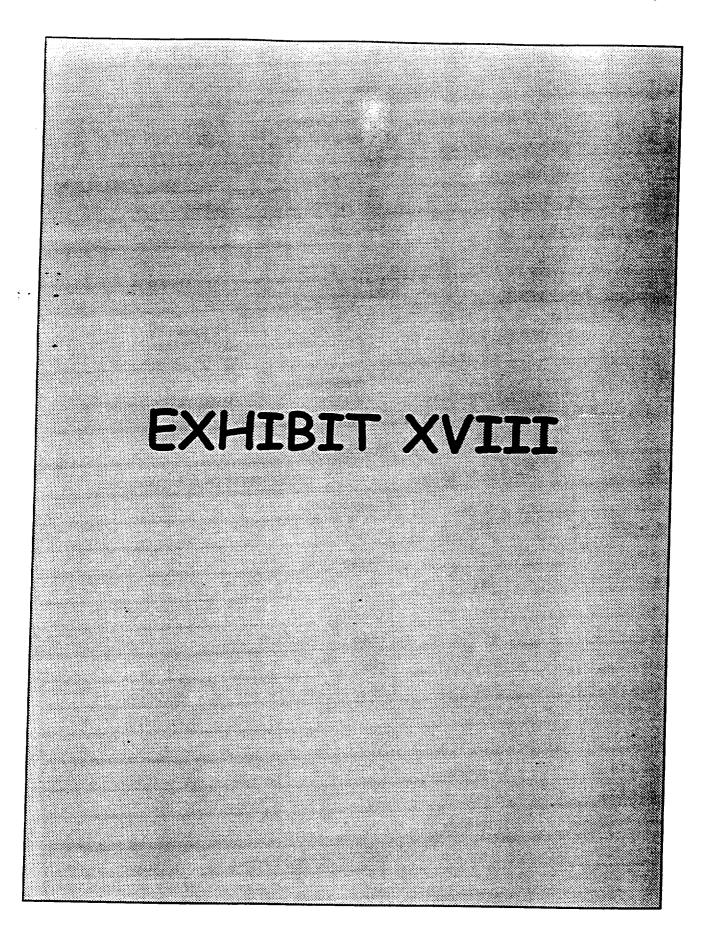
DATE OF READING: 08/03/98

UES - D.B.

EST. W.S.W.T. (fr)

DRILLED BY:

	Is	<u> </u>	η	ı	1 e	EST. W.S.W.T	. (ft):		YPE OF	SAMPLI	NG: ASTN	A D-1586
DEPTH (FT.)	AM PL	BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	w.t.	S Y M B O	DESCRIPTION	-200 (%)			ERBERG MITS	K (FT./	ORG
	E				Ľ				LL	Pi	DAY)	(%)
0 –	-			-	···	Dark gray fine SAND [SP]						
-	- - - -	•	-	_		gray dark brown						
5 	M	4-4-8	1.2			brown						
· -	Ø	7-9-11	20		ΔΖ	gray-brown Medium dense, gray SHELL; with sand	7			†		ļ
_	M	10-12-12	24			Medium dense, gray fine SAND with shall	4					
-	Ø	13-11-14	25			[SP]	1		1			
10 —						Loose, light gray fine SAND; with shell [SP]						
_				}		o o , a same, menonen for f						
j	H				: : : i							
15	Д.	2-2-3	5				1					
_					1	BORING TERMINATED AT 15 FEET		1	1			
7		ŀ	ŀ			•						
					- 1							
20	···· † ··											
4			1									
Ė			ĺ									
25 —										1		
+								1				
7												
{										- 1		
30					······							
7										ı		
1			- 1								1	
35 —									1	- 1		
-			- 1									
			İ							- 1		
4					- 1							
.0 —												••••••
7			-					<i>.</i> .			1	
\dashv			ŀ					- 1	ı		ŀ	
5	1											
ٽ ا "					<u> </u>				·····			• • • • • • • • • • • • • • • • • • • •
\dashv						`		ĺ			-	
1												
0-	ļ											
7			1			1		ļ				
							- 1					
5 —												·····



STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-5	Client Phone:	(407) 867-4121
Date:	8-6-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Western-IW-5I	Total Depth:	35.0'
Screen Interval <u>:</u>	25.0 to 35.0' bls	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
9:51 a.m.	1,000+	Milky
9:56 a.m.	728	Cloudy
10:01 a.m.	368	Cloudy
10:06 a.m.	142	Clear
10:12 a.m.	83.1	Clear
10:18 a.m.	48.5	Clear
10:20 a.m.	30.1	Clear
10:30 a.m.	20.1	Clear
Total Development Time:	39 minutes	
Final Turbidity Reading:	20.1 NTUs	
Estimated Volume Removed	: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
		Estimo Resolts (OVAPID).

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	C-5	Client Phone:	(407) 867-4121
Date:	8-6-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

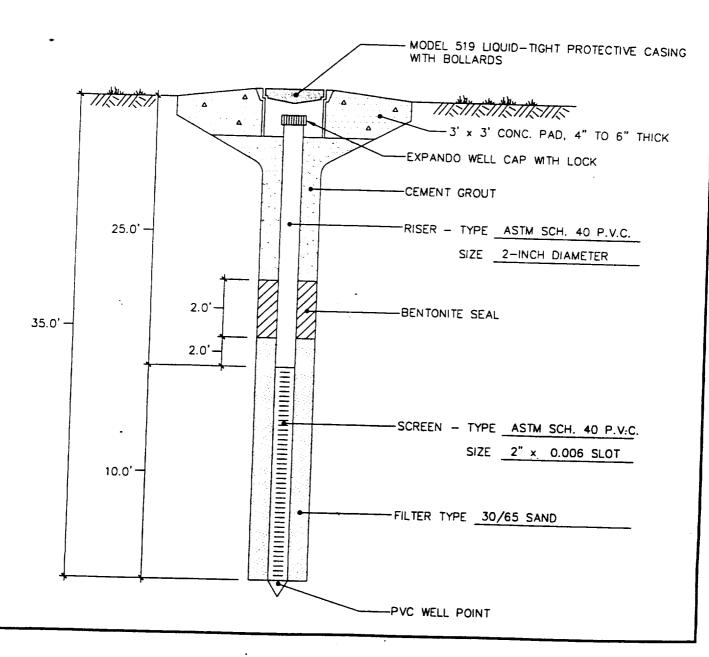
Well ID:	Western-IW-5S	Total Depth:	13.5'
Screen Interval:	3.5 to 13.5' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
1:53 p.m.	1,060	Cloudy
1:57 p.m.	40.6	
1:59 p.m.	18.7	
Total Development Time:	6 minutes	
Final Turbidity Reading:	18.7 NTUs	
Estimated Volume Remove	d: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):
	<u></u>	



PROLECT NO	18467-005-01
REPORT NO.	
PAGE:	

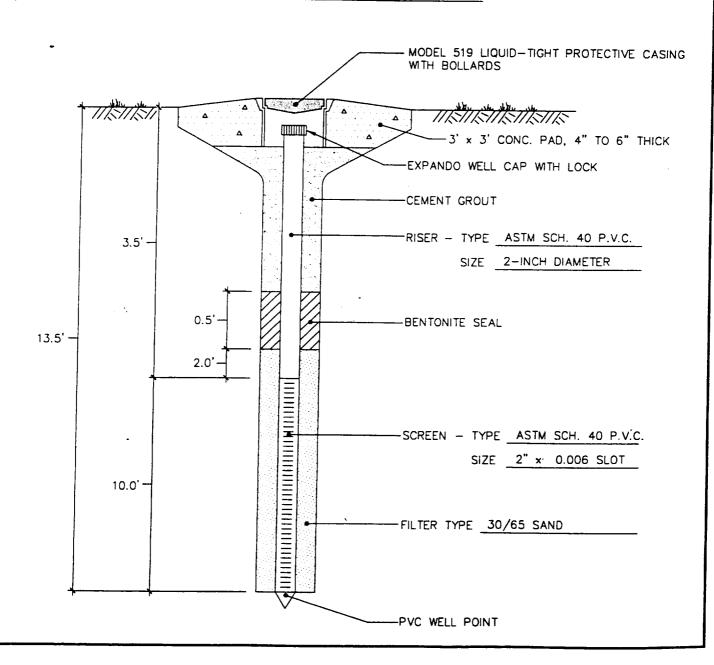
PROJECT:	ROJECT: INSTALLATION 52 MONITOR WELLS						
	KENNEDY SPACE CENTER, FLORIDA						
CLIENT: _	NASA JJ-D	DATE:	08/06/98				
WELL NUME	ER: WESTERN-IW-5I LOCATION: C-5	DATE.					
	BY: UES - ROCKLEDGE						





PROJECT NO.:	18467-005-01
REPORT NO.:	
PACE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: 08/06/98
WELL NUMB	ER: WESTERN-IW-5S LOCATION: C-5	
INSTALLED	BY: UES - ROCKLEDGE	





BORING LOG

REPORT NO PAGE. B-2.17

PROJECT

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT:

NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION:

C-5 TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft): WATER TABLE (ft):

SECTION:

DATE STARTED:

8/5/98

8/5/98

5.5 DATE OF READING: 08/05/98 DATE FINISHED:

DRILLED BY:

UES - ROCKLEDGE

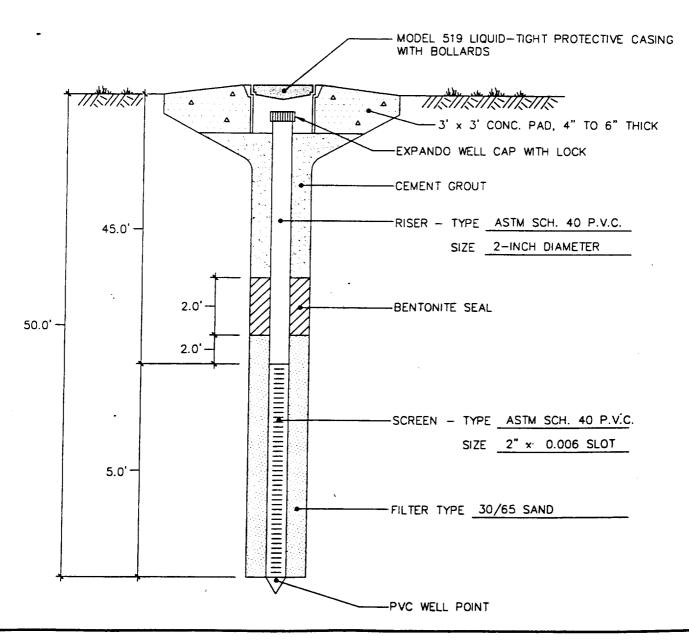
FST WSWT (4)

	Ts	T	-		1 A	EST. W.S.W.T	`. (ft):		TYPE OF	SAMPLIN	IG: ASTN	1 D-1586
DEPTH (FT.)		BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	W.T.	Ö	DESCRIPTION	-200 (%)	MC (%)	l u	ERBERG MITS	K (FT./	ORG.
_	- -			-	L		<u> </u>		LL.	PI	DAY)	(%)
0-	7					Gray fine SAND; with roots [SP]				 		
	1				· ·	Light brown fine SAND; with roots [SP]	_					
_	\downarrow					Brown clayey fine SAND; with roots [SC]	1					
5 -	\Rightarrow	4-5-10 10-12-15	15 27			Medium dense, brown fine SAND; with shell	┧			-	***************************************	
	从	7-8-9	17		• • •	Medium dense, gray fine SAND; with shell	//					
	 	-10-14-12	26	ŀ		į visti	/					
10-	1					Medium.dense.gray.brown.fine.SAND; with				ļ		•
	1			:								
45	N	7-10-13	23	į.								
15 –										ļ		
-	1			ŀ								
-	\square	18-21-34	55	ŀ								
20 —	\prod					very dense	-	ļ				
-	1			ŀ								
25 —	X	14-19-23	42									
25						dense		ļ				
-												İ
30	X	5-10-13	23	<u> </u>								ł
30 —						medium dense			· 			
	Í			F:								
35	XI.	7-9-12	21	÷	: 	Dense, gray fine SAND; with silt [SP-SM]]		İ	- 1
337	7					BORING TERMINATED AT 35 FEET		•••••••••	ļļ.			
1												- 1
40 +												
40 -			•					••••••		<u> </u>		
1												
4												ļ
45 —							·····					
4						`	1					
_ +												
50	1			-								
1												
						İ						j
55 —	1											



PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS							
	KENNEDY SPACE CENTER, FLORIDA							
CLIENT: _	NASA JJ-D	DATE: 08/06/98						
	BER: MARSH-IW-5D LOCATION: D-5							
INSTALLED	BY: UES - ORLANDO							



UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-5	Client Phone:	(407) 867-4121
Date:	8-3-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Marsh-IW-5S	Total Depth:	12.5'
Screen Interval:	2.5 to 12.5' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
2:15 p.m.	1,000+	Dark grey
2:20 p.m.	409	Grey
2:27 p.m.	130	Light grey
2:32 p.m.	87.3	Light grey
2:42 p.m.	59.3	Light grey
2:47 p.m.	41.9	Clear
2:51 p.m.	31.7	Clear
Total Development Time:	36 minutes	
Final Turbidity Reading:	31.7 NTUs	
Estimated Volume Removed	: 110 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-5	Client Phone:	(407) 867-4121
Date:	8-4-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Marsh-IW-5I	Total Depth:	35.0'
Screen Interval <u>:</u>	25.0 to 35.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
9:50 a.m.	1,000+	Dark grey
9:59 a.m.	490	Light brown
10:09 a.m.	12.9	Clear
÷		
Total Development Time:	19 minutes	
Final Turbidity Reading:	12.9 NTUs	
Estimated Volume Removed	i: 165 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):

STANDARD WELL DEVELOPMENT LOG

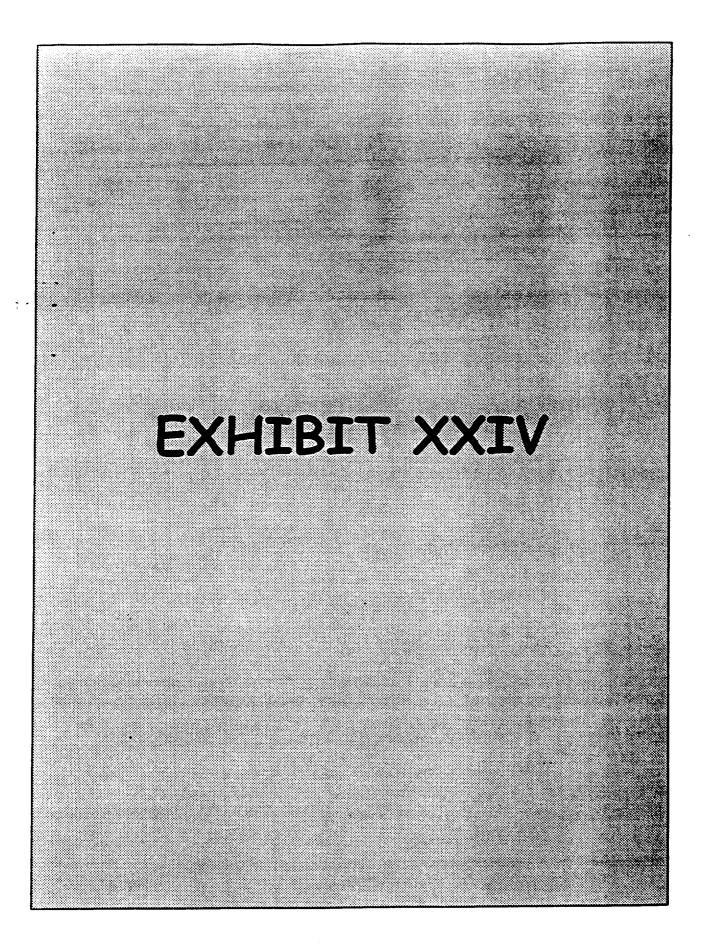
PROJECT INFORMATION

Contract: Reference:	NAS10-12206(B)/DO-5/TD- 002 Monitor Well Installation	Client Name: Department:	NASA JJ-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-5	Client Phone:	(407) 867-4121
Date:	8-6-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Marsh-IW-5D	Total Depth:	50.0'
Screen Interval <u>:</u>	45.0 to 50.0' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	Flush mount with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE
9:15 a.m.	No reading	CHOCKDIVATER AFFEARANCE
9:20 a.m.	67.5	Clear
9:25 a.m.	11.7	
·		
Total Development Time:	10 minutes	
Final Turbidity Reading:	11.7 NTUs	
Estimated Volume Removed	i: 55 gallons	
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):





UNIVERSAL ENGINEERING SCIENCES **BORING LOG**

REPORT NO	
PAGE.	B-2.22

INSTALLATION OF 52 MONITOR WELLS

KENNEDY SPACE CENTER

BREVARD, FLORIDA

CLIENT: NASA JJ-D

LOCATION: AS SPECIFIED BY CLIENT

REMARKS: HAND AUGER FROM 0 FEET TO 4 FEET BORING DESIGNATION: SECTION:

D-6 TOWNSHIP:

SHEET: 1 of 1

RANGE:

G.S. ELEVATION (ft):

DATE STARTED:

9/2/98 9/2/98

WATER TABLE (ft): 2.8

DATE OF READING: 09/02/98

DATE FINISHED: DRILLED BY:

UES - ROCKLEDGE

	S			EST. W.S.W.T					TYPE (OF SAMP	LING: ASTI	M D-1586
DEPTH (FT.)	AMP LE	BLOWS PER 6" INCREMENT	N (BLOWS/ FT.)	w.T.		DESCRIPTION	-200 (%)		<u> </u>	TERBER LIMITS	(FT./	ORG.
0-	-							_		L PI	-	(%)
	11					Dark gray fine SAND [SP]						
				▼.	· : :	Dark gray fine SAND; with roots (muck)			-			
5 -	-X	4-1-1	2			very loose						
,	X	0-0-0	0			Gray clayey fine SAND; with trace sand	_					
10-	X	1-3-8 7-7-6	11 13			medium dense, shell						
- -				ļ								••••••
15 	\boxtimes	6-7-7	14	Ŀ								
-	1		Ĭ			BORING TERMINATED AT 15 FEET						
_]		İ									
20 —												
-												***************************************
25 —												
1												
30 🚽			···									
7		İ				·						
]												
35 —									ļ	ļ		
10			.									Ī
7									ļ	ļ		
7												
15	ļ											
1						¥						
_								į				
io —	ļ											
7										1	*****	
‡												
5 —												

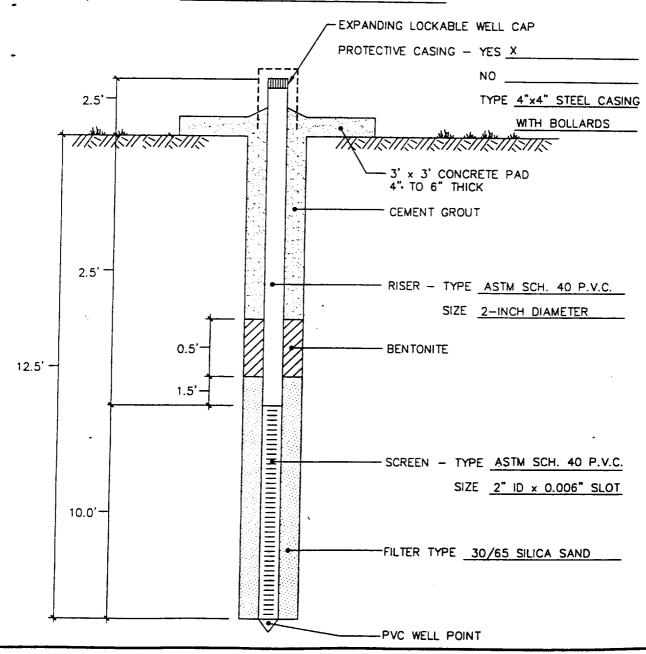


UNIVERSAL ENGINEERING SCIENCES WELL COMPLETION LOG

PROJECT NO.	18467-005-01
REPORT NO.:	
PAGE:	

PROJECT:	INSTALLATION 52 MONITOR WELLS	
	KENNEDY SPACE CENTER, FLORIDA	
CLIENT: _	NASA JJ-D	DATE: <u>09/02/98</u>
WELL NUMB	ER: MARSH-IW-6S LOCATION: D-6	
INSTALLED	BY: UES - ROCKLEDGE	

WELL DIAGRAM - NOT TO SCALE



UNIVERSAL ENGINEERING SCIENCES STANDARD WELL DEVELOPMENT LOG

PROJECT INFORMATION

Contract:	NAS10-12206(B)/DO-5/TD-	Client Name:	NASA
Reference:	002 Monitor Well Installation	Department:	71-D
Project No.:	18467-005-01	Client Contact:	Rosaly Santos-Ebaugh, RPM
Well Location:	D-6	Client Phone:	(407) 867-4121
Date:	9-2-98	Client Fax:	(407) 867-8040

MONITOR WELL DATA

Well ID:	Marsh-IW-6S	Total Depth:	12.5'
Screen Interva <u>l</u> :	2.5' to 12.5' bis	Well Type: Construction:	Permanent 2" PVC Riser 2" PVC Screen
Above Grade Protection:	4" x 4" Steel riser with concrete pad and bollards	Development Method:	Rotary Pump, Tremie Pipe

WELL DEVELOPMENT DATA

WELL DEVELOPMENT DATA						
TIME	TURBIDITY (NTU)	GROUNDWATER APPEARANCE				
9:00 a.m.	1,000+	Milky				
9:15 a.m.	250	Cloudy				
9:40 a.m.	90	Milky, Purged well dry				
10:10 a.m.	12					
:						
Total Development Time:	60 minutes					
Final Turbidity Reading:	12 NTUs					
Estimated Volume Removed:	40 gallons					
COMMENTS:		SOIL SCREENING RESULTS (OVA-PID):				
	43	36				

APPENDIX E

Surface Water Data for Kennedy Space Center

Table E-1. List of sample site descriptions for KSC Background Study surface water sampling locations.

Sampling Station	Classification	Description of site
AVE001	Ditch	End of Avenue E (or Radio Tower Road)
BCE004	Banana Creek	East Banana Creek
BCE012	Banana Creek	East side of Route 3 culvert
BCW014	Banana Creek	West of Route 3 culvert
BCW15	Banana Creek/Moore	North end where Moore Creek enters Banana Creek.
	Creek	The state of the s
BDP001	Black Point Dr.	Just before viewing site #3 near cedar tree
BPD002	Black Point Dr.	Before viewing tower near culvert
BRC09	Banana River	Pepper Flats channel seagrass transect
BRN18	Banana River	KSC Long Term Water station
BRN020	Banana River	SE corner of the manantee enclosures
BRS017	Banana River	Banana River East of radio tower
BRS019	Banana River	Banana River south of Hangar AF islands
HPC001	Happy Creek	Boat ramp on side road
IRL001	Indian River Lagoon	Indian River Lagoon
IRL002	Indian River Lagoon	Indian River Lagoon
IRL003	Indian River Lagoon	Indian River Lagoon south of Haulover near SE
		corner
IRL004	Indian River Lagoon	North side of Indian River by Titusville
IRL005	Indian River Lagoon	South side of Indian River by bridge
JER001	Ditch	Jerome Road
KPI001	Borrow Pit	Kars Park II
MAX01		KSC Long Term Water station
MLN001	Mosquito Lagoon	60 degrees east of channel marker 21
	Mosquito Lagoon	South side of channel marker 34
	Mosquito Lagoon	KSC Long Term Water station
	Moore Creek	Moore Creek in C15-E
	Ditch	North side of NASA causeway
	Borrow Pit	Avenue 10 south of O&C building
	Banana River	Guard guard boat dock
	Banana River	Pepper Flats, south of PEF10 seagrass transect
		Inside the refuge by Indian River
		Robert's Road
		St. John's Water Management District station
SJML02		St. John's Water Management District station
		South side of the tow road before SLF
		Ditch on the side road by the Lightning lab
		Ditch on the SE corner of the SR 3 security station
		Corner of Schwartz Road east and Static Test Road
SRW001		Schwartz Road west of Route 3 about 1 mile E of the
CDWOOD		left curve
		Borrow Pit area towards Robert's Road
TEL004	Ditch	East side of road near cattle farm area

Table E-2. List of latitude, longitude, and Florida State Plane NAD27 coordinates (X and Y) for surface water sampling locations. Coordinates were collected with a Global Positioning System (GPS).

STATION	Latitude	Longitude	X Coordinate	Y Coordinate
AVE001	80.69322849	28.52090342	598514.128	1522152.539
BCE004	80.64351194	28.60581045	614387.966	1553067.047
BCE012	80.65900196	28.5989994	609424.635	1550576.197
BCW014	80.66519987	28.58589169	607449.071	1545804.885
BCW15	80.70782029	28.58526396	593771.175	1545540.843
BDP001	80.76606	28.65621191	575029.403	1571295.007
BPD002	80.77242619	28.67575843	572974.088	1578397.828
BRC09	80.60487976	28.57026397	626826.801	1540181.955
BRN18	80.60961147	28.52954257	625356.205	1525371.543
BRN020	80.59903326	28.50545843	628782.222	1516626.302
BRS017	80.64335991	28.46280848	614591.345	1501074.951
BRS019	80.60319886	28.4716421	627484.919	1504327.068
HPC001	80.65780787	28.61459092	609791.604	1556246.064
IRL001	80.83889609	28.75297154	551621.808	1606436.57
IRL002	80.79451931	28.75797648	565838.152	1608278.182
IRL003	80.74988309	28.71763042	580170.897	1593636.268
IRL004	80.78751796	28.62900032	568164.959	1561388.551
IRL005	80.7873466	28.62737116	568220.988	1560796.318
JER001	80.67294948	28.49271873	605054.318	1511922.49
KP1001	80.66836547	28.52125525	606498.129	1522301.693
MAX01	80.64482416	28.65136981	613917.706	1569630.363
MLN001	80.78613466	28.8272189	568479.428	1633458.667
MLN002	80.77062248	28.77136094	573485.593	1613158.581
MLS03	80.67589247	28.6873698	603917.396	
MOC021	80.71047434	28.56152436	592940.249	1536907.581
NSC001	80.70887827	28.52626672	593483.727	1524089.979
OCA010	80.64316548	28.50784925	614605.174	1517450.808
PEF20	80.62070578	28.59374787	621719.838	
PEF024	80.60367124	28.59328304	627187.023	
REF001	80.77177926	28.63270232	573211.434	
ROB001	80.6902763	28.54077212	599443.522	1529378.755
SJB02	80.63383414	28.43975227	617677.594	1492701.541
SJML02	80.71741362	28.7317818		
SLF001	80.66544776	28.59314977	607362.141	1548443.55
SR3-D1	80.72085799	28.70525556		
SR3-D2	80.74468642	28.72509064	581830.801	1596352.224
SRE001	80.61020229	28.56248671	625127.547	
SRW001	80.69254774	28.56282572	598693.66	
SRW002	80.66817001	28.55321287	606528.749	
TEL004	80.67544479	28.47626649	604268.937	1505938.711

Table E-3. Parameters that were determined by the NASA/KSC Remediation Program team to be treated as below detection limits (non-detect) for all surface water locations based on low frequency of detection.

PAHs (8310)
Fluorene
Napthalene
Metals
Copper

Table E-4. Minimum, maximum, mean, and standard deviation of all surface water sampling stations with salinity 0-5 ppt. Certain values have combined means based on frequency of detection which was *<6, and **<12.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Dev.
Organochlorine Pesticides (8081)						
4,4'-DDD	μg/L	13	ND	ND	ND	ND
4,4'-DDE	μg/L	13	ND	ND	ND	ND
4,4'-DDT	μg/L	13	ND	ND	ND	ND
Aldrin	μg/L	13	ND	ND	ND	ND
Alpha - BHC	μg/L	13	ND	ND	ND	ND
Beta - BHC	μg/L	13	ND	ND	ND	ND
Chlordane (alpha)	μg/L	13	ND	ND	ND	ND
Chlordane (gamma)	μg/L	13	ND	ND	ND	ND
Chlordane (total)	μg/L	13	ND	ND	ND	ND
Delta - BHC	μg/L	13	ND	ND	ND	ND
Dieldrin	μg/L	13	ND	ND	ND	ND
Endosulfan I	μg/L	13	ND	ND	ND	ND
Endosulfan II (beta)	μg/L	13	ND	ND	ND	ND
Endosulfan Sulfate	μg/L	13	ND	ND	ND	ND
Endrin	μg/L	13	ND	ND	ND	ND
Endrin Aldehyde	μg/L	13	ND	ND	ND	ND
Endrin Ketone	μg/L	13	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/L	13	ND	ND	ND	ND
Heptachlor	μg/L	13	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/L	13	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/L	13	ND	ND	ND	ND
Isodrin	μg/L	13	ND	ND	ND	ND
Methoxychlor	μg/L	13	ND	ND	ND	ND
Mirex	μg/L	13	ND	ND	ND	ND
Toxaphene	µg/L	13	ND	ND	ND	ND
Aroclors (8082)						
PCB - 1016/1242	μg/L	13	ND	ND	ND	ND
PCB - 1221	μg/L	13	ND	ND	ND	ND
PCB - 1232	μg/L	13	ND	ND	ND	ND
PCB - 1248	μg/L	13	ND	ND	ND	ND
PCB - 1254	μg/L	13	ND	ND	ND	ND
PCB - 1260	μg/L	13	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
Silvex (2,4,5 - TP) (2-(2,4,5 -	μg/L	13	ND	ND	ND	ND
Trichlorophenoxy)) propionic acid		<u> </u>			ļ <u>.</u>	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic	μg/L	13	ND	ND	ND	ND
acid)		ļ		<u> </u>	1	115
2,4 - D (2,4 - Dichlorophenoxy acetic	μg/L	13	ND	ND	ND	ND
acid)		1-20	1.5	10	AID.	ND
3,5 - DCBA	μg/L	13	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid)	μg/L	13	ND	ND	ND	
MCPP	μg/L	13	ND	ND	ND	ND

Table E-4. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Dev.
Chlorinated Herbicides (8151) (cont.)						
4 - Nitrophenol	μg/L	13	ND	ND	ND	ND
Acifluorfen	μg/L	13	ND	ND	ND	ND
Bentazon	μg/L	13	ND	ND	ND	ND
Chloramben	μg/L	13	ND	ND	ND	ND
Dacthal	μg/L	13	ND	ND	ND	ND
Dalapon	μg/L	13	ND	ND	ND	ND
Dicambia	μg/L	13	ND	ND	ND	ND
Dichlorprop [2-(2,4-Dichlorophenoxy)	μg/L	13	ND	ND	ND	ND
proponic acid]	~3-		,,,,		.,,,	
Dinoseb	μg/L	13	ND	ND	ND	ND
MCPA	μg/L	13	ND	ND	ND	ND
Pentachlorophenol	μg/L	13	ND	ND	ND	ND
Picloram	μg/L	13	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						
1-Methylnaphthalene	μg/L	13	ND	ND	ND	ND
2-Methylnaphthalene	μg/L	13	ND	ND	ND	ND
Acenaphthylene	μg/L	13	ND	ND	ND	ND
Anthracene	μg/L	13	ND	ND	ND	ND
Benzo(a)anthracene	μg/L	13	ND	ND	ND	ND
Benzo(a)pyrene	μg/L	13	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/L	13	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/L	13	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/L	13	ND	ND	ND	ND
Chrysene	μg/L	13	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/L	13	ND	ND	ND	ND
Fluoranthene	μg/L	13	ND	ND	ND	ND
Fluorene	μg/L	13	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/L	13	ND	ND	ND	ND
Naphthalene	μg/L	13	ND	ND	ND	ND
Phenanthrene	μg/L	13	ND	ND	ND	ND
Pyrene	μg/L	13	ND	ND	ND	ND
Metals	<u> </u>					
Aluminum	mg/L	13	0.025	1.4	0.42	0.51
Antimony *	mg/L	13	0.003	0.026	0.008	0.008
Arsenic	mg/L	13	ND	ND	ND	ND
Barium	mg/L	13	ND	ND	ND	ND
Beryllium	mg/L	13	ND	ND	ND	ND
Cadmium	mg/L	13	ND	ND	ND	ND
Calcium	mg/L	13	22	340	98	85
Chloride	mg/L	13	20	14000	1527	3789
Chromium	mg/L	13	ND ND	ND	ND	ND
Cobalt	mg/L	13	ND	ND	ND	ND
Copper	mg/L	13	ND ND	ND	ND	ND
Iron	mg/L	13	0.05	2.3	0.49	0.61
	9/∟	10	0.00	2.0	U. 7 3	U.U I

Table E-4. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Dev.
Metals (cont.)						
Lead *	mg/L	13	0.003	0.013	0.004	0.003
Magnesium	mg/L	13	2.3	1400	139	380
Manganese **	mg/L	13	0.005	0.27	0.04	0.07
Mercury	mg/L	13	ND	ND	ND	ND
Nickel	mg/L	13	ND	ND	ND	ND
Potassium	mg/L	13	2.3	290	33	78
Selenium	mg/L	13	ND	ND	ND	ND
Silver	mg/L	13	ND	ND	ND	ND
Sodium	mg/L	13	9.8	7700	895	2064
Thallium	mg/L	13	ND	ND	ND	ND
Vanadium	mg/L	13	ND	ND	ND	ND
Zinc	ma/L	13	ND	ND	ND	ND

Table E-5. Minimum, maximum, mean, and standard deviation of all surface water sampling stations with salinity \geq 6 ppt. Certain values have combined means based on frequency of detection which was *<16, **<10, and ***<5.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Dev.
Organochlorine Pesticides (8081)		 		Maximum	IVIGALL	Sid. Dev.
4,4'-DDD	μg/L	33	ND	ND	ND	ND
4,4'-DDE	μg/L	33	ND	ND	ND	ND
4,4'-DDT	μg/L	33	ND	ND	ND	ND
Aldrin	μg/L	33	ND	ND	ND	ND
Alpha - BHC	μg/L	33	ND	ND	ND	ND
Beta - BHC	μg/L	33	ND	ND	ND	ND
Chlordane (alpha)	μg/L	33	ND	ND	ND	ND
Chlordane (gamma)	μg/L	33	ND	ND	ND	ND
Chlordane (total)	μg/L	33	ND	ND	ND	ND
Delta - BHC	μg/L	33	ND	ND	ND	ND
Dieldrin ***	μg/L	33	0.025	0.39	0.037	0.064
Endosulfan I	μg/L	33	ND	ND	ND	ND
Endosulfan II (beta)	μg/L	33	ND	ND	ND	ND
Endosulfan Sulfate	µg/L	33	ND	ND	ND	
Endrin	μg/L	33	ND	ND	ND	ND
Endrin Aldehyde	μg/L	33	ND	ND	ND	ND ND
Endrin Ketone	µg/L	33	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/L	33	ND	ND	ND	ND
Heptachlor	μg/L	33	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/L	33	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/L	33	ND	ND	ND	ND
Isodrin	μg/L	33	ND	ND	ND	ND
Methoxychlor	μg/L	33	ND	ND	ND	ND
Mirex	μg/L	33	ND	ND	ND	ND
Toxaphene	μg/L	33	ND	ND	ND	ND
Aroclors (8082)	1 - 3 -			- 110	-10	IND
PCB - 1016/1242	μg/L	33	ND	ND	ND	ND
PCB - 1221	μg/L	33	ND	ND	ND	ND
PCB - 1232	μg/L	33	ND	ND	ND	ND ND
PCB - 1248	μg/L	33	ND	ND	ND	ND
PCB - 1254	μg/L	33	ND	ND	ND	ND
PCB - 1260	μg/L	33	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
Silvex (2,4,5 - TP) (2-(2,4,5 - richlorophenoxy)) propionic acid	μg/L	33	ND	ND	ND	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	μg/L	33	ND	ND	ND	ND
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/L	33	ND	ND	ND	ND
5,5 - DCBA	μg/L	33	ND	ND	ND	ND
,4 - DB (4 - (2,4 - Dichlorophenoxy) outyric acid)	μg/L	33	ND	ND	ND	ND
1CPP	μg/L	33	ND	ND	ND	ND

Table E-5. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Dev.
Chlorinated Herbicides (8151) (cont.)						
4 - Nitrophenol	μg/L	33	ND	ND	ND	ND
Acifluorfen	μg/L	33	ND	ND	ND	ND
Bentazon	μg/L	33	ND	ND	ND	ND
Chloramben	μg/L	33	ND	ND	ND	ND
Dacthal	μg/L	33	ND	ND	ND	ND
Dalapon	μg/L	33	ND	ND	ND	ND
Dicambia	μg/L	33	ND	ND	ND	ND
Dichlorprop [2-(2,4-Dichlorophenoxy)	μg/L	33	ND	ND	ND	ND
proponic acid]	7.3-					
Dinoseb	μg/L	33	ND	ND	ND	ND
MCPA	μg/L	33	ND	ND	ND	ND
Pentachlorophenol	μg/L	33	ND	ND	ND	ND
Picloram	μg/L	33	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)	<i>P</i> 9 -	- 00				
1-Methylnaphthalene	μg/L	33	ND	ND	ND	ND
2-Methylnaphthalene	μg/L	33	ND	ND	ND	ND
Acenaphthylene	μg/L	33	ND	ND	ND	ND
Anthracene	μg/L	33	ND	ND	ND	ND
Benzo(a)anthracene ***	μg/L	33	0.025	0.13	0.05	0.02
Benzo(a)pyrene ***	μg/L	33	0.025	0.22	0.06	0.04
Benzo(b)fluoranthene	μg/L	33	ND	ND	ND	ND
		33	ND	ND	ND	ND
Benzo(g,h,i)perylene Benzo(k)fluoranthene ***	μg/L	33	0.025	0.21	0.05	0.03
Chrysene ***	μg/L	33	0.025	0.21	0.05	0.03
	μg/L	33	0.023 ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/L	33	ND	ND	ND	ND
Fluoranthene	μg/L	33	ND	ND	ND	ND
Fluorene	μg/L	33	0.025	0.13	0.05	0.02
Indeno(1,2,3-cd)pyrene ***	μg/L	33	0.025 ND	ND	ND	ND
Naphthalene	μg/L		ND ND	ND	ND	ND
Phenanthrene	μg/L	33	ND	ND	ND	ND
Pyrene	μg/L	33	ND	IND	IND	ing.
Metals		22	0.025	1.9	0.448	0.503
Aluminum	mg/L	33	0.025	0.033	0.448	0.009
Antimony *	mg/L	33		0.033	0.012	0.006
Arsenic **	mg/L	33	0.005	ND	ND	ND
Barium	mg/L	33	ND 0.0005		0.0005	0.0004
Beryllium **	mg/L	33	0.0005	0.002 ND	ND	ND
Cadmium	mg/L	33	ND 140	<u> </u>	304	89
Calcium	mg/L	33	140	550	12097	4803
Chloride	mg/L	33	2500	25000		4803 ND
Chromium	mg/L	33	ND	ND	ND	ND
Cobalt	mg/L	33	ND	ND	ND	<u> </u>
Copper	mg/L	33	ND	ND 1.05	ND	ND 0.35
Iron	mg/L	33	0.025	1.25	0.26	0.35

Table E-5. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Dev.
Metals (cont.)						
Lead *	mg/L	33	0.003	0.014	0.003	0.002
Magnesium	mg/L	33	0.005	0.065	0.035	0.014
Manganese ***	mg/L	33	0.005	0.065	0.012	0.014
Mercury	mg/L	33	ND	ND	ND	ND
Nickel	mg/L	33	ND	ND	ND	ND
Potassium	mg/L	33	55	600	311	142
Selenium ***	mg/L	33	0.005	0.025	0.017	0.006
Silver ***	mg/L	33	0.0002	0.005	0.001	0.0009
Sodium	mg/L	33	1700	12000	7006	2576
Thallium ***	mg/L	33	0.001	0.005	0.002	0.0009
Vanadium	mg/L	33	ND	ND	ND	ND
Zinc	mg/L	33	ND	ND	ND	ND

Table E-6. Minimum, maximum, mean, and standard deviation of each parameter analyzed in surface water samples classified as Banana Creek.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)		-				
4,4'-DDD	μg/L	7	ND	ND	ND	ND
4,4'-DDE	μg/L	7	ND	ND	ND	ND
4.4'-DDT	μg/L	7	ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/L	7	ND	ND	ND	ND
Aldrin	μg/L	7	ND	ND	ND	ND
Alpha – BHC	μg/L	7	ND	ND	ND	ND
Beta – BHC	μg/L	7	ND	ND	ND	ND
Chlordane (alpha)	μg/L	7	ND	ND	ND	ND
Chlordane (gamma)	μg/L	7	ND	ND	ND	ND
Chlordane (total)	μg/L	7	ND	ND	ND	ND
Delta – BHC	μg/L	7	ND	ND	ND	ND
Dieldrin	μg/L	7	ND	ND	ND	ND
Endosulfan I	μg/L	7	ND	ND	ND	ND
Endosulfan II (beta)	μg/L	7	ND	ND	ND	ND
Endosulfan Sulfate	μg/L	7	ND	ND	ND	ND
Endrin	μg/L	7	ND	ND	ND	ND
Endrin Aldehyde	μg/L	7	ND	ND	ND	ND
Endrin Ketone	μg/L	7	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/L	7	ND	ND	ND	ND
Heptachlor	μg/L	7	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/L	7	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/L	7	ND	ND	ND	ND
Isodrin	μg/L	7	ND	ND	ND	ND
Methoxychlor	μg/L	7	ND	ND	ND	ND
	μg/L	7	ND	ND	ND	ND
Mirex		7	ND	ND	ND	ND
Toxaphene	μg/L	<u> </u>	145	 		
Aroclors (8082)	110/1	7	ND	ND	ND	ND
PCB - 1016/1242 PCB - 1221	μg/L μg/L	7	ND	ND	ND	ND
PCB - 1232	μg/L	7	ND	ND	ND	ND
PCB - 1232	μg/L	7	ND	ND	ND	ND
PCB - 1254	μg/L	7	ND	ND	ND	ND
PCB - 1260	μg/L	7	ND	ND	ND	ND
Chlorinated Herbicides (8151)	pyr	 	1.0			
Silvex (2,4,5 - TP) (2-(2,4,5 -	μg/L	7	ND	ND	ND	ND
Trichlorophenoxy)) propionic acid						
2,4,5 - T (2,4,5 - Triclorophenoxy	μg/L	7	ND	ND	ND	ND
acetic acid) 2,4 - D (2,4 - Dichlorophenoxy acetic	μg/L	7	ND	ND	ND	ND
	μg/L	'				
acid) 3,5 - DCBA	μg/L	7	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy)	μg/L	7	ND	ND	ND	ND
butyric acid)					<u> </u>	<u> </u>

Table E-6. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)						
4 - Nitrophenol	μg/L	7	ND	ND	ND	ND
Acifluorfen	μg/L	7	ND	ND	ND	ND
Bentazon	μg/L	7	ND	ND	ND	ND
Chloramben	μg/L	7	ND	ND	ND	ND
Dacthal	μg/L	7	ND	ND	ND	ND
Dalapon	μg/L	7	ND	ND	ND	ND
Dicambia	μg/L	7	ND	ND	ND	ND
Dichloroprop [2-(2,4-	μg/L	7	ND	ND	ND	ND
Dichlorophenoxy) proponic acid]						
Dinoseb	μg/L	7	ND	ND	ND	ND
MCPA	μg/L	7	ND	ND	ND	ND
MCPP	μg/L	7	ND	ND	ND	ND
Pentachlorophenol	μg/L	7	ND	ND	ND	ND
Picloram	μg/L	7	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/L	7	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						
1-Methylnaphthalene	μg/L	7	ND	ND	ND	ND
2-Methylnaphthalene	μg/L	7	ND	ND	ND	ND
Acenaphthylene	μg/L	7	ND	ND	ND	ND
Anthracene	μg/L	7	ND	ND	ND	ND
Benzo(a)anthracene	μg/L	7	ND	ND	ND	ND
Benzo(a)pyrene	μg/L	7	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/L	7	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/L	7	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/L	7	ND	ND	ND	ND
Chrysene	μg/L	7	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/L	7	ND	ND	ND	ND
Fluoranthene	μg/L	7	ND	ND	ND	ND
Fluorene	μg/L	7	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/L	7	ND	ND	ND	ND
Naphthalene	μg/L	7	ND	ND	ND	ND
Phenanthrene	μg/L	7	ND	ND	ND	ND
Pyrene	μg/L	7	ND	ND	ND	ND
Sum of Polyaromatic Hydrocarbons	μg/L	7	ND	ND	ND	ND
Metals						
Aluminum	mg/L	7	0.24	1.9	1.1	0.61
Antimony	mg/L	7	0.003	0.015	0.006	0.004
Arsenic	mg/L	7	0.005	0.016	0.008	0.004
Barium	mg/L	7	ND	ND	ND	ND
Beryllium	mg/L	7	ND	ND	ND	ND
Cadmium	mg/L	7	ND	ND	ND	ND
Calcium	mg/L	7	220	380	311.43	76.69
Chloride	mg/L	7	8000	18000	12414.29	4046.57
Chromium	mg/L	7	ND	ND	ND	ND

Table E-6. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Cobalt	mg/L	7	ND	ND	ND	ND
Copper	mg/L	7	ND	ND	ND	ND
Iron	mg/L	7	0.07	1.2	0.64	0.42
Lead	mg/L	7	ND	ND	ND	ND
Magnesium	mg/L	7	580	1100	857.14	244.45
Manganese	mg/L	7	0.005	0.042	0.013	0.013
Mercury	mg/L	7	ND	ND	ND	ND
Nickel	mg/L	7	ND	ND	ND	ND
Potassium	mg/L	7	150	340	258.57	92.45
Selenium	mg/L	7	ND	ND	ND	ND
Silver	mg/L	7	ND	ND	ND	ND
Sodium	mg/L	7	4400	8900	6714.29	2110.63
Thallium	mg/L	7	ND	ND	ND	ND
Vanadium	mg/L	7	ND	ND	ND	ND
Zinc	mg/L	7	ND	ND	ND	ND
Other Parameters						
Dissolved Oxygen	mg/L	7	3.72	10.25	6.88	2.70
Total Dissolved Solids	mg/L	7	14000	28000	21000	6531.97
Total Organic Carbon	mg/L	7	22	48	33.57	10.56

Table E-7. Minimum, maximum, mean, and standard deviation of each parameter analyzed in surface water samples classified as Banana River.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)	1	<u> </u>			<u> </u>	
4,4'-DDD	μg/L	8	ND	ND	ND	ND
4,4'-DDE	μg/L	8	ND	ND	ND	ND
4,4'-DDT	μg/L	8	ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/L	8	ND	ND	ND	ND .
Aldrin	μg/L	8	ND	ND	ND	ND
Alpha - BHC	μg/L	8	ND	ND	ND	ND
Beta - BHC	µg/L	8	ND	ND	ND	ND
Chlordane (alpha)	μg/L	8	ND	ND	ND	ND
Chlordane (gamma)	μg/L	8	ND	ND	ND	ND
Chlordane (total)	μg/L	8	ND	ND	ND	ND
Delta - BHC	μg/L	8	ND	ND	ND	ND
Dieldrin	μg/L	8	ND	ND	ND	ND
Endosulfan i	μg/L	8	ND	ND	ND	ND
Endosulfan II (beta)	μg/L	8	ND	ND	ND	ND
Endosulfan Sulfate	μg/L	8	ND	ND	ND	ND
Endrin	μg/L	8	ND	ND	ND	ND
Endrin Aldehyde	μg/L	8	ND	ND	ND	ND
Endrin Ketone	μg/L	8	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/L	8	ND	ND	ND	ND
Heptachlor	μg/L	8	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/L	8	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/L	8	ND	ND	ND	ND
Isodrin	μg/L	8	ND	ND	ND	ND
Methoxychlor	μg/L	8	ND	ND	ND	ND
Mirex	μg/L	8	ND	ND	ND	ND
Toxaphene	μg/L	8	ND	ND	ND	ND
Aroclors (8082)	-					
PCB - 1016/1242	μg/L	8	ND	ND	ND	ND
PCB - 1221	μg/L	8	ND	ND	ND	ND
PCB - 1232	μg/L	8	ND	ND	ND	ND
PCB - 1248	μg/L	8	ND	ND	ND	ND
PCB - 1254	μg/L	8	ND	ND	ND	ND
PCB - 1260	μg/L	8	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
Silvex (2,4,5 - TP) (2-(2,4,5 - Trichlorophenoxy)) propionic acid	μg/L	8	ND	ND	ND	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	μg/L	8	ND	ND	ND	ND
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/L	8	ND	ND	ND	ND
3,5 - DCBA	μg/L	8	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid)	μg/L	8	ND	ND	ND	ND

Table E-7. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)						
4 - Nitrophenol	μg/L	8	ND	ND	ND	ND
Acifluorfen	μg/L	8	ND	ND	ND	ND
Bentazon	μg/L	8	ND	ND	ND	ND
Chloramben	μg/L	8	ND	ND	ND	ND
Dacthal	μg/L	8	ND	ND	ND	ND
Dalapon	μg/L	8	ND	ND	ND	ND
Dicambia	μg/L	8	ND	ND	ND	ND
Dichloroprop [2-(2,4-	μg/L	8	ND	ND	ND	ND
Dichlorophenoxy) proponic acid]						
Dinoseb	μg/L	8	ND	ND	ND	ND
MCPA	μg/L	8	ND	ND	ND	ND
MCPP	μg/L	8	ND	ND	ND	ND
Pentachlorophenol	μg/L	8	ND	ND	ND	ND
Picloram	μg/L	8	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/L	8	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						
1-Methylnaphthalene	μg/L	8	ND	ND	ND	ND
2-Methylnaphthalene	μg/L	8	ND	ND	ND	ND
Acenaphthylene	μg/L	8	ND	ND	ND	ND
Anthracene	μg/L	8	ND	ND	ND	ND
Benzo(a)anthracene	μg/L	8	ND	ND	ND	ND
Benzo(a)pyrene	μg/L	8	0.05	0.18	0.07	0.05
Benzo(b)fluoranthene	μg/L	8	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/L	8	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/L	8	0.05	0.21	0.07	0.06
Chrysene	μg/L	8	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/L	8	ND	ND	ND	ND
Fluoranthene	μg/L	8	ND	ND	ND	ND
Fluorene	μg/L	8	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/L	8	ND	ND	ND	ND
Naphthalene	μg/L	8	ND	ND	ND	ND
Phenanthrene	μg/L	8	ND	ND	ND	ND
Pyrene	μg/L	8	ND	ND	ND	ND
Sum of Polyaromatic Hydrocarbons	μg/L	8	ND	ND	ND	ND
Metals						
Aluminum	mg/L	8	0.025	0.72	0.207	0.255
Antimony	mg/L	8	0.003	0.021	0.008	0.007
Arsenic	mg/L	8	0.005	0.023	0.015	0.007
Barium	mg/L	8	ND	ND	ND	ND
Beryllium	mg/L	8	0.001	0.002	0.001	0.001
Cadmium	mg/L	8	ND	ND	ND	ND
Calcium	mg/L	8	200	250	226.25	15.98
Chloride	mg/L	8	6200	11000	8212.5	1956.26
Chromium	mg/L	8	ND	ND	ND	ND

Table E-7. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Cobalt	mg/L	8	ND	ND	ND	ND
Copper	mg/L	8	ND	ND	ND	ND
iron	mg/L	8	0.025	0.41	0.12	0.14
Lead	mg/L	8	0.003	0.008	0.004	0.002
Magnesium	mg/L	8	520	650	606.25	46.58
Manganese	mg/L	8	0.005	0.042	ND	0.013
Mercury	mg/L	8	ND	ND	ND	ND
Nickel	mg/L	8	ND	ND	ND	ND
Potassium	mg/L	8	170	320	238.75	50.55
Selenium	mg/L	8	ND	ND	ND	ND
Silver	mg/L	8	ND	ND	ND	ND
Sodium	mg/L	8	4300	5400	4925	452.77
Thallium	mg/L	8	ND	ND	ND	ND
Vanadium	mg/L	8	ND	ND	ND	ND
Zinc	mg/L	8	ND	ND	ND	ND
Other Parameters						
Dissolved Oxygen	mg/L	8	5.7	8.74	7.25	1.06
Total Dissolved Solids	mg/L	8	14000	18000	16125	1552.65
Total Organic Carbon	mg/L	8	15	24	18.5	3.12

Table E-8. Minimum, maximum, mean, and standard deviation of each parameter analyzed in ditches with salinity values of 0-5 ppt. for surface water sampling locations.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)						
4,4'-DDD	μg/L	7	ND	ND	ND	ND
4,4'-DDE	μg/L	7	ND	ND	ND	ND
4,4'-DDT	μg/L	7	ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/L	7	ND	ND	ND	ND
Aldrin	μg/L	7	ND	ND	ND	ND
Alpha - BHC	μg/L	7	ND	ND	ND	ND
Beta - BHC	μg/L	7	ND	ND	ND	ND
Chlordane (alpha)	μg/L	7	ND	ND	ND	ND
Chlordane (gamma)	μg/L	7	ND	ND	ND	ND
Chlordane (total)	μg/L	7	ND	ND	ND	ND
Delta - BHC	μg/L	7	ND	ND	ND	ND
Dieldrin	μg/L	7	ND	ND	ND	ND
Endosulfan I	μg/L	7	ND	ND	ND	ND
Endosulfan II (beta)	μg/L	7	ND	ND	ND	ND
Endosulfan Sulfate	μg/L	7	ND	ND	ND	ND
Endrin	μg/L	7	ND	ND	ND	ND
Endrin Aldehyde	μg/L	7	ND	ND	ND	ND
Endrin Ketone	μg/L	7	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/L	7	ND	ND	ND	ND
Heptachlor	μg/L	7	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/L	7	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/L	7	ND	ND	ND	ND
Isodrin	μg/L	7	ND	ND	ND	ND
Methoxychlor	μg/L	7	ND	ND	ND	ND
Mirex	μg/L	7	ND	ND	ND	ND
Toxaphene	μg/L	7	ND	ND	ND	ND
Aroclors (8082)						
PCB - 1016/1242	μg/L	7_	ND	ND	ND	ND
PCB - 1221	μg/L	7	ND	ND	ND	ND
PCB - 1232	μg/L	7	ND	ND	ND	ND
PCB - 1248	μg/L	7	ND	ND	ND	ND
PCB - 1254	μg/L	7	ND	ND	ND	ND
PCB - 1260	μg/L	7	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
Silvex (2,4,5 - TP) (2-(2,4,5 - Trichlorophenoxy)) propionic acid	μg/L	7	ND	ND	ND	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	μg/L	7	ND	ND	ND	ND
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/L	7	ND	ND	ND	ND
3,5 - DCBA	μg/L	7	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid)	μg/L	7	ND	ND	ND	ND

Table E-8. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)						
4 - Nitrophenol	μg/L	7	ND	ND	ND	ND
Acifluorfen	μg/L	7	ND	ND	ND	ND
Bentazon	µg/L	7	ND	ND	ND	ND
Chloramben	μg/L	7	ND	ND	ND	ND
Dacthal	μg/L	7	ND	ND	ND	ND
Dalapon	μg/L	7	ND	ND	ND	ND
Dicambia	μg/L	7	ND	ND	ND	ND
Dichloroprop [2-(2,4- Dichlorophenoxy) proponic acid]	μg/L	7	ND	ND	ND	ND
Dinoseb	μg/L	7	ND	ND	ND	ND
MCPA	μg/L	7	ND	ND	ND	ND
MCPP	μg/L	7	ND	ND	ND	ND
Pentachlorophenol	μg/L	7	ND	ND	ND	ND
Picloram	μg/L	7	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/L	7	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						
1-Methylnaphthalene	μg/L	7	ND	ND	ND	ND
2-Methylnaphthalene	μg/L	7	ND	ND	ND	ND
Acenaphthylene	μg/L	7	ND	ND	ND	ND
Anthracene	μg/L	7	ND	ND	ND	ND
Benzo(a)anthracene	μg/L	7	ND	ND	ND	ND
Benzo(a)pyrene	μg/L	7	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/L	7	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/L	7	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/L	7	ND	ND	ND	ND
Chrysene	μg/L	7	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/L	7	ND	ND	ND	ND
Fluoranthene	μg/L	7	ND	ND	ND	ND
Fluorene	μg/L	7	ND	ND	ND	ND
ndeno(1,2,3-cd)pyrene	μg/L	7	ND	ND	ND	ND
Naphthalene	μg/L	7	ND	ND	ND	ND
Phenanthrene	μg/L	7	ND	ND	ND	ND
Pyrene	μg/L	7	ND	ND	ND	ND
Sum of Polyaromatic Hydrocarbons	μg/L	7	ND	ND	ND	ND
Metals						
Aluminum	mg/L	7	0.03	1.4	0.42	0.51
Antimony	mg/L	7	0.003	0.026	0.008	0.009
Arsenic	mg/L	7	ND	ND	ND	ND
Barium	mg/L	7	ND	ND	ND	ND
Beryllium	mg/L	7	ND	ND	ND	ND
Cadmium	mg/L	7	ND	ND	ND	ND
Calcium	mg/L	7	22	340	98.38	85.19
Chloride	mg/L	7	20	14000	1527.15	3789.13
Chromium	mg/L	7	ND	ND	ND	ND

Table E-8. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Cobalt	mg/L	7	ND	ND	ND	ND
Copper	mg/L	7	ND	ND	ND ND	ND
Iron	mg/L	7	0.05	2.3	0.49	0.61
Lead	mg/L	7	0.003	0.013	0.004	0.003
Magnesium	mg/L	7	2.3	1400	139.2	380.3
Manganese	mg/L	7	0.005	0.270	0.039	0.077
Mercury	mg/L	7	ND	ND	D	ND
Nickel	mg/L	7	ND	ND	ND	ND
Potassium	mg/L	7	2.3	290	32.6	77.8
Selenium	mg/L	7	ND	ND	ND	ND
Silver	mg/L	7	ND	ND	ND	ND
Sodium	mg/L	7	9.8	7700	894.7	2064.1
Thallium	mg/L	7	ND	ND	ND	ND
Vanadium	mg/L	7	ND	ND	DN	ND
Zinc	mg/L	7	ND	ND	ND	ND
Other Parameters						
Dissolved Oxygen	mg/L	13	3.4	106.9	22.3	36.6
Total Dissolved Solids	mg/L	13	155	27000	3138.9	7253.45
Total Organic Carbon	mg/L	13	9	68	32.7	16.7

Table E-9. Minimum, maximum, mean, and standard deviation of each parameter analyzed in surface water samples classified as ditches with salinity values \geq 6 ppt.

Parameter	Units	N	Minimum	Maximum	Mean	Std Davieties
Organochlorine Pesticides (8081)		 ''	TVIII III I GIST	IVIAXIITUITI	Mean	Std. Deviation
4,4'-DDD	μg/L	5	ND	ND	ND	ND.
4,4'-DDE	µg/L	5	ND	ND	ND	ND ND
4,4'-DDT	µg/L	5	ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/L	5	ND	ND	ND	ND ND
Aldrin	μg/L	5	ND	ND	ND	ND
Alpha - BHC	μg/L	5	ND	ND	ND	ND
Beta - BHC	μg/L	5	ND	ND	ND	ND
Chlordane (aipha)	μg/L	5	ND	ND	ND	ND ND
Chlordane (gamma)	μg/L	5	ND	ND	ND	ND
Chlordane (total)	μg/L	5	ND	ND	ND	ND
Delta - BHC	μg/L	5	ND	ND	ND	ND ND
Dieldrin	μg/L	5	0.025	0.07	0.034	0.02
Endosulfan I	μg/L	5	ND	ND	ND	ND
Endosulfan II (beta)	μg/L	5	ND	ND	ND	ND ND
Endosulfan Sulfate	μg/L	5	ND	ND	ND	ND
Endrin	μg/L	5	ND	ND	ND	ND
Endrin Aldehyde	μg/L	5	ND	ND	ND	ND
Endrin Ketone	μg/L	5	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/L	5	ND	ND	ND	ND
Heptachlor	μg/L	5	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/L	5	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/L	5	ND	ND	ND	ND
Isodrin	μg/L	5	ND	ND	ND	ND
Methoxychlor	μg/L	5	ND	ND	ND	ND
Mirex	μg/L	5	ND	ND	ND	ND
Toxaphene	μg/L	5	ND	ND	ND	ND
Aroclors (8082)						
PCB - 1016/1242	μg/L	5	ND	ND	ND	ND
PCB - 1221	μg/L	5	ND	ND	ND	ND
PCB - 1232	μg/L	5	ND	ND	ND	ND
PCB - 1248	μg/L	5	ND	ND	ND	ND
PCB - 1254	μg/L	5	ND	ND	ND	ND
PCB - 1260	μg/L	5	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
Silvex (2,4,5 - TP) (2-(2,4,5 - Trichlorophenoxy)) propionic acid	µg/L	5	ND	ND	ND	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	μg/L	5	ND	ND	ND	ND
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/L	5	ND	ND	ND	ND
3,5 - DCBA	μg/L	5	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy) outyric acid)	μg/L	5	ND	ND	ND	ND

Table E-9. (cont.)

Chlorinated Herbicides (8151) (cont.) 4 - Nitrophenol Acifluorfen Bentazon Chloramben Dacthal	μg/L μg/L μg/L μg/L μg/L	5 5 5	ND ND	ND	ND	
4 - Nitrophenol Acifluorfen Bentazon Chloramben	μg/L μg/L μg/L	5			ND	<u></u>
Acifluorfen Bentazon Chloramben	μg/L μg/L μg/L	5			ND	
Bentazon Chloramben	μg/L μg/L		ND			ND
Chloramben	μg/L	5		ND	ND	ND
			ND	ND	ND	ND
Dacthal	<i>μ</i> η/Ι	5	ND	ND	ND	ND
		5	ND	ND	ND	ND
Dalapon	μg/L	5	ND	ND	ND	ND
Dicambia	μg/L	5	ND	ND	ND	ND
Dichloroprop [2-(2,4-	μg/L	5	ND	ND	ND	ND
Dichlorophenoxy) proponic acid]						<u> </u>
Dinoseb	μg/L	5	ND	ND	ND	ND
MCPA	μg/L	5	ND	ND	ND	ND
MCPP	μg/L	5	ND	ND	ND	ND
Pentachlorophenol	μg/L	5	ND	ND	ND	ND
Picloram	μg/L	5	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/L	5	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						<u> </u>
1-Methylnaphthalene	μg/L	5	ND	ND	ND	ND
2-Methylnaphthalene	μg/L	5	ND	ND	ND	ND
Acenaphthylene	μg/L	5	ND	ND	ND	ND
Anthracene	μg/L	5	ND	ND	ND	ND
Benzo(a)anthracene	μg/L	5	ND	ND	ND	ND
Benzo(a)pyrene	μg/L	5	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/L	5	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/L	5	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/L	5	ND	ND	ND	ND
Chrysene	μg/L	5	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/L	5	ND	ND	ND	ND
Fluoranthene	μg/L	5	ND	ND	ND	ND
Fluorene	μg/L	5	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/L	5	ND	ND	ND	ND
Naphthalene	μg/L	5	ND	ND	ND	ND
Phenanthrene	μg/L	5	ND	ND	ND	ND
Pyrene	μg/L	5	ND	ND	ND	ND
Sum of Polyaromatic Hydrocarbons	μg/L	5	ND	ND	ND	ND
Metals	F3-					
Aluminum	mg/L	5	0.03	1.25	0.43	0.48
Antimony	mg/L	5	0.003	0.033	0.017	0.012
Arsenic	mg/L	5	0.005	0.017	0.007	0.005
Barium	mg/L	5	ND	ND	ND	ND
Beryllium	mg/L	5	ND	ND	ND	ND
Cadmium	mg/L	5	ND	ND	ND	ND
Calcium	mg/L	5	140	400	266	120.7
Chloride	mg/L	5	2500	17000	9520	6961.1
Chromium	mg/L	5	ND	ND	ND	ND

Table E-9. (cont.)

Parameter	Units	N	Minimum	Maximum	Mean	Ctd Davietie
Metals (cont.)				Maximon	Weall	Std. Deviation
Cobalt	mg/L	5	ND	ND	ND	
Copper	mg/L	5	ND	ND	ND	ND
iron	mg/L	5	0.06		. ND	ND
Lead	mg/L	5	ND	1.25	0.32	0.52
Magnesium	mg/L	5	200	ND 5000	ND ND	ND
Manganese	mg/L	5		5200	1628	2050.8
Mercury			ND	ND	ND	ND
Nickel	mg/L	5	ND	ND	ND	ND
	mg/L	5	ND	ND	ND	ND
Potassium	mg/L	5	55	560	257.2	217.1
Selenium	mg/L	5	ND	ND	ND	ND
Silver	mg/L	5	ND	ND	ND	ND
Sodium	mg/L	5	1700	9800	5480	
Thallium	mg/L	5	ND	ND	ND	3945.5
Vanadium	mg/L	5	ND	ND	ND	ND
Zinc	mg/L	5	ND	ND		ND.
Other Parameters	- g	-		IND	ND	ND
Dissolved Oxygen	mg/L	5	4.09	69.4	20.4	
Total Dissolved Solids	mg/L	5	2750		22.4	31.5
Total Organic Carbon		5		33000	17390	14081.39
- 3	mg/L	<u> </u>	11	33	23.8	8.0

Table E-10. Minimum, maximum, mean, and standard deviation of each parameter analyzed in surface water samples classified as Indian River Lagoon.

Organochlorine Pesticides (8081)	Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
4,4*-DDD μg/L 7 ND	Organochlorine Pesticides (8081)						
4,4*-DDE μg/L 7 ND ND ND ND 4,4*-DDT μg/L 7 ND		μα/L	7	ND	ND	ND	ND
4,4*DDT μg/L 7 ND ND ND ND Sum 4,4*DDD + 4,4*DDE + 4,4*DDT μg/L 7 ND ND ND ND Aldrin μg/L 7 ND ND ND ND Aldrin - Aldrin μg/L 7 ND ND ND ND Aldrin - Aldrin μg/L 7 ND ND ND ND Aldrin - Aldrin μg/L 7 ND ND ND ND Aldrin - Aldrin μg/L 7 ND ND ND ND Aldrin - Aldrin μg/L 7 ND ND ND ND Chlordane (alpha) μg/L 7 ND ND ND ND Chlordane (gamma) μg/L 7 ND ND ND ND ND Delta - BHC μg/L 7 ND ND ND ND ND ND ND ND ND ND <td< td=""><td></td><td></td><td></td><td>ND</td><td>ND</td><td>ND</td><td>ND</td></td<>				ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT					ND	ND	ND
Aldrin μg/L 7 ND ND ND ND Alpha - BHC μg/L 7 ND ND ND ND ND Chlordane (alpha) μg/L 7 ND ND ND ND Chlordane (alpha) μg/L 7 ND ND ND ND Chlordane (alpha) μg/L 7 ND ND ND ND Chlordane (alpha) μg/L 7 ND ND ND ND Chlordane (alpha) μg/L 7 ND ND ND ND Chlordane (alpha) μg/L 7 ND ND ND ND Chlordane (alpha) μg/L 7 ND ND ND ND Deleta BHC μg/L 7 ND ND ND ND Deleta BHC μg/L 7 ND ND ND ND Endosultan L μg/L 7 ND ND <						ND	ND
Alpha - BHC						ND	ND
Beta - BHC							
Chlordane (alpha)		*					<u> </u>
Chiordane (gamma)							<u> </u>
Chiordane (total)							
Delta - BHC							
Dieldrin							
Endosulfan I μg/L 7 ND ND ND ND Endosulfan II (beta) μg/L 7 ND ND ND ND Endosulfan Sulfate μg/L 7 ND ND ND ND Endrin Sulfate μg/L 7 ND ND ND ND Endrin Aldehyde μg/L 7 ND ND ND ND Endrin Ketone μg/L 7 ND ND ND ND Endrin Ketone μg/L 7 ND ND ND ND Endrin Ketone μg/L 7 ND ND ND ND Endrin Ketone μg/L 7 ND ND ND ND Heptachlor μg/L 7 ND ND ND ND Heptachlor μg/L 7 ND ND ND ND Heptachlor Epoxide (a) μg/L 7 ND ND ND							
Endosulfan II (beta) μg/L 7 ND ND ND ND Endosulfan Sulfate μg/L 7 ND ND ND ND Endrin μg/L 7 ND ND ND ND Endrin Aldehyde μg/L 7 ND ND ND ND Endrin Aldehyde μg/L 7 ND ND ND ND Endrin Aldehyde μg/L 7 ND ND ND ND Endrin Aldehyde μg/L 7 ND ND ND ND Endrin Aldehyde μg/L 7 ND ND ND ND Endrin Aldehyde μg/L 7 ND ND ND ND Endrin Aldehyde μg/L 7 ND ND ND ND Heptachlor μg/L 7 ND ND ND ND Heptachlor μg/L 7 ND ND ND							
Endosultan Suifate		•					
Endrin							
Endrin Aldehyde							
Find Find							
Gamma -BHC (Lindane) μg/L 7 ND ND ND ND Heptachlor μg/L 7 ND ND ND ND Heptachlor Epoxide (a) μg/L 7 ND ND ND ND Heptachlor Epoxide (b) μg/L 7 ND ND ND ND Heptachlor Epoxide (b) μg/L 7 ND ND ND ND Isodrin μg/L 7 ND ND ND ND ND Methoxychlor μg/L 7 ND ND ND ND ND Mirex μg/L 7 ND ND ND ND ND Toxaphene μg/L 7 ND ND ND ND ND PCB - 1016/1242 μg/L 7 ND ND ND ND ND PCB - 1221 μg/L 7 ND ND ND ND ND ND ND							
Heptachlor							
Heptachlor Epoxide (a)							
Heptachlor Epoxide (b)							
No							
Methoxychlor							
Mirex μg/L 7 ND ND ND ND ND ND ND							
Toxaphene μg/L 7 ND ND ND ND ND Aroclors (8082) PCB - 1016/1242 μg/L 7 ND ND ND ND ND ND ND ND ND ND ND ND ND	The state of the s						<u> </u>
Aroclors (8082) PCB - 1016/1242 μg/L 7 ND ND ND ND PCB - 1221 μg/L 7 ND ND ND ND PCB - 1232 μg/L 7 ND ND ND ND PCB - 1248 μg/L 7 ND ND ND ND PCB - 1254 μg/L 7 ND ND ND ND PCB - 1260 μg/L 7 ND ND ND ND Silvex (2,4,5 - TP) (2-(2,4,5 - Triclorophenoxy)) propionic acid μg/L 7 ND ND ND ND 2,4 - D (2,4 - Dichlorophenoxy acetic acid) μg/L 7 ND ND ND ND 3,5 - DCBA μg/L 7 ND ND ND ND 2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid) μg/L 7 ND ND ND ND					<u> </u>		
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		μg/L		ND	ND	140	110
PCB - 1221 μg/L 7 ND ND ND ND PCB - 1232 μg/L 7 ND ND ND ND PCB - 1248 μg/L 7 ND ND ND ND PCB - 1254 μg/L 7 ND ND ND ND PCB - 1260 μg/L 7 ND ND ND ND Silvex (2,4,5 - TP) (2-(2,4,5 - Triclorophenoxy)) propionic acid μg/L 7 ND ND ND ND Silvex (2,4,5 - Triclorophenoxy) propionic acid μg/L 7 ND ND ND ND 2,4,5 - T (2,4,5 - Triclorophenoxy) acetic acid μg/L 7 ND ND ND ND 2,4 - D (2,4 - Dichlorophenoxy acetic acid) μg/L 7 ND ND ND ND 3,5 - DCBA μg/L 7 ND ND ND ND 2,4 - DB (4 - (2,4 - Dichlorophenoxy) μg/L 7 ND ND ND ND <t< td=""><td></td><td></td><td>7</td><td>ND</td><td>ND</td><td>NID</td><td>ND</td></t<>			7	ND	ND	NID	ND
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$							
PCB - 1248 $\mu g/L 7 ND ND ND ND$ PCB - 1254 $\mu g/L 7 ND ND ND ND$ PCB - 1260 $\mu g/L 7 ND ND ND ND$ PCB - 1260 $\mu g/L 7 ND ND ND ND$ Chlorinated Herbicides (8151) $Silvex (2,4,5-TP) (2-(2,4,5-\mu g/L) 7 ND ND ND ND$ Silvex (2,4,5 - TP) (2-(2,4,5 - \mu g/L) 7 ND ND ND ND Trichlorophenoxy)) propionic acid $2,4,5-T (2,4,5-Triclorophenoxy \mu g/L 7 ND ND ND ND ND$ acetic acid) $2,4-D (2,4-Dichlorophenoxy acetic \mu g/L 7 ND ND ND ND ND$ $3,5-DCBA \mu g/L 7 ND ND ND ND ND$ butyric acid) $ND ND ND ND ND ND ND ND $							
PCB - 1254 $\mu g/L$ 7 ND ND ND ND ND ND ND ND ND ND ND ND ND							
PCB - 1260 $\mu g/L$ 7 ND ND ND ND ND ND ND Chlorinated Herbicides (8151) $\mu g/L$ 7 ND ND ND ND ND ND ND ND Trichlorophenoxy) propionic acid $\mu g/L$ 7 ND ND ND ND ND ND ND ND ND ND ND ND ND					<u> </u>		
Chlorinated Herbicides (8151) Silvex (2,4,5 - TP) (2-(2,4,5 - μg/L 7 ND ND ND ND ND Trichlorophenoxy)) propionic acid μg/L 7 ND ND ND ND ND ND ND ND ND ND ND ND ND	The second secon				1		
Silvex (2,4,5 - TP) (2-(2,4,5 - $\mu g/L$ 7 ND ND ND ND ND Trichlorophenoxy)) propionic acid 2,4,5 - T (2,4,5 - Triclorophenoxy $\mu g/L$ 7 ND ND ND ND ND ND acetic acid) 2,4 - D (2,4 - Dichlorophenoxy acetic acid) $\mu g/L$ 7 ND ND ND ND ND ND acid) 3,5 - DCBA $\mu g/L$ 7 ND ND ND ND ND ND ND ND Dtyric acid)		μg/L	'	NU	ND	ND	NO
Trichlorophenoxy)) propionic acid μg/L 7 ND ND ND ND 2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid) μg/L 7 ND ND ND ND 2,4 - D (2,4 - Dichlorophenoxy acetic acid) μg/L 7 ND ND ND ND 3,5 - DCBA μg/L 7 ND ND ND ND 2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid) μg/L 7 ND ND ND ND				ND	ND	ND	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid) μg/L 7 ND ND ND ND 2,4 - D (2,4 - Dichlorophenoxy acetic acid) μg/L 7 ND ND ND ND 3,5 - DCBA μg/L 7 ND ND ND ND 2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid) μg/L 7 ND ND ND ND		μg/L	′	ND	IND	ND	
2,4 - D (2,4 - Dichlorophenoxy acetic acid) μg/L 7 ND ND ND ND ND 3,5 - DCBA μg/L 7 ND ND ND ND ND 2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid) μg/L 7 ND ND ND ND	2,4,5 - T (2,4,5 - Triclorophenoxy	μg/L	7	ND	ND	ND	ND
3,5 - DCBA μg/L 7 ND ND ND ND 2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid) μg/L 7 ND ND ND ND	2,4 - D (2,4 - Dichlorophenoxy acetic	μg/L	7	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy) µg/L 7 ND ND ND ND butyric acid)		μα/l	7	ND	ND	ND	ND
	2,4 - DB (4 - (2,4 - Dichlorophenoxy)						
	4 - Nitrophenol	μg/L	7	ND	ND	ND	ND

Table E-10. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)						
Acifluorfen	μg/L	7	ND	ND	ND	AID
Bentazon	μg/L	7	ND	ND	ND	ND
Chloramben	µg/L	7	ND	ND	ND	ND
Dacthal	µg/L	7	ND	ND	+	ND
Dalapon	μg/L	7	ND	ND	ND	ND
Dicambia	μg/L	7	ND	ND	ND ND	ND
Dichloroprop [2-(2,4-Dichlorophenoxy) proponic acid]	µg/L	7	ND	ND	ND	ND ND
Dinoseb	μg/L	7	ND	ND	ND	AID.
MCPA	μg/L	7	ND	ND	ND	ND
MCPP	μg/L	7	ND	ND	ND	ND
Pentachiorophenol	μg/L	7	ND	ND	ND	ND
Picloram	μg/L	7	ND	ND	ND	ND ND
Sum of Chlorinated Herbicides	μg/L	7	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)	F-5-	<u> </u>	 	IND	ND	ND
1-Methylnaphthalene	μg/L	7	ND	ND	ND	ND
2-Methylnaphthalene	μg/L	7	ND	ND	ND	ND
Acenaphthylene	μg/L		ND	ND	ND	ND
Anthracene	μg/L	7	ND	ND	ND	ND
Benzo(a)anthracene	μg/L	7	ND	ND	ND	ND
Benzo(a)pyrene	μg/L	7	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/L	7	ND	ND	ND	ND ND
Benzo(g,h,i)perylene	μg/L	7	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/L	7	ND	ND	ND	ND ND
Chrysene	μg/L	7	ND	ND	ND	ND ND
Dibenzo(a,h)anthracene	μg/L	7	ND	ND	ND	
luoranthene	μg/L	7	ND	ND	ND	ND ND
luorene	μg/L	7	ND	ND ND	ND	ND ND
ndeno(1,2,3-cd)pyrene	μg/L	7	ND	ND	ND	ND ND
laphthalene	μg/L	7	ND	ND	ND	ND ND
Phenanthrene	μg/L	7	ND	ND	ND	ND
Pyrene	μg/L	7	ND	ND	ND	ND
um of Polyaromatic Hydrocarbons	μg/L	7	ND	ND	ND	ND
Metals	79-			110	145	IND
luminum	mg/L	7	0.03	0.74	0.21	0.25
ntimony	mg/L	7	0.003	0.028	0.012	0.25
rsenic	mg/L	7	0.005	0.016	0.007	0.004
arium	mg/L	7	ND ND	ND	ND	0.004 ND
eryllium	mg/L	7	ND	ND	ND	ND ND
admium	mg/L	7	ND	ND	ND	ND ND
alcium	mg/L	7	280	420	341	49
hloride	mg/L	7	13000	17000	14857	1464
hromium	mg/L	7	ND	ND	ND	ND
obalt	mg/L	7	ND	ND	ND	ND ND

Table E-10. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Copper	mg/L	7	ND	ND	ND	ND
Iron	mg/L	7	0.03	0.42	0.11	0.15
Lead	mg/L	7	ND	ND	ND	ND
Magnesium	mg/L	7	800	1300	990	172
Manganese	mg/L	7	0.005	0.065	0.024	0.022
Mercury	mg/L	7	ND	ND	ND	ND
Nickel	mg/L	7	ND	ND	ND	ND
Potassium	mg/L	7	230	390	303	58
Selenium	mg/L	7	0.005	0.024	0.013	0.007
Silver	mg/L	7	0.000	0.002	0.001	0.001
Sodium	mg/L	7	7100	9300	8300	879
Thallium	mg/L	7	0.001	0.003	0.002	0.001
Vanadium	mg/L	7	ND	ND	ND	ND
Zinc	mg/L	7	ND	ND	ND	ND
Other Parameters						
Dissolved Oxygen	mg/L	7	5.29	12.28	8.18	2.18
Total Dissolved Solids	mg/L	7	20000	30000	25285.7	3860.7
Total Organic Carbon	mg/L	7	11	31	21.6	6.2

Table E-11. Minimum, maximum, mean, and standard deviation of each parameter analyzed in surface water samples classified as Mosquito Lagoon.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)						Old. Dovidion
4,4'-DDD	μg/L	6	ND	ND	ND	ND
4,4'-DDE	μg/L	6	ND	ND	ND	ND
4,4'-DDT	μg/L	6	ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/L	6	ND	ND	ND	ND
Aldrin	μg/L	6	ND	ND	ND	ND
Alpha - BHC	μg/L	6	ND	ND	ND	ND
Beta - BHC	μg/L	6	ND	ND	ND	ND
Chlordane (alpha)	μg/L.	6	ND	ND	ND	ND
Chlordane (gamma)	μg/L	6	ND	ND	ND	ND
Chlordane (total)	μg/L	6	ND	ND	ND	ND
Delta - BHC	μg/L	6	ND	ND	ND	ND
Dieldrin	μg/L	6	ND	ND	ND	ND
Endosulfan I	μg/L	6	ND	ND	ND	ND
Endosulfan II (beta)	μg/L	6	ND	ND	ND	ND
Endosulfan Sulfate	μg/L	6	ND	ND	ND	ND
Endrin	μg/L	6	ND	ND	ND	ND
Endrin Aldehyde	μg/L	6	ND	ND	ND	ND
Endrin Ketone	μg/L	6	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/L	6	ND	ND	ND	ND
Heptachlor	μg/L	6	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/L	6	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/L	6	ND	ND	ND	ND
Isodrin	μg/L	6	ND	ND	ND	ND
Methoxychlor	μg/L	6	ND	ND	ND	ND
Mirex	μg/L	6	ND	ND	ND	ND
Toxaphene	μg/L	6	ND	ND	ND	ND
Aroclors (8082)	<u> </u>				110	ND
PCB - 1016/1242	μg/L	6	ND	ND	ND	ND
PCB - 1221	μg/L	6	ND	ND	ND	ND ND
PCB - 1232	μg/L	6	ND	ND	ND	ND
PCB - 1248	μg/L	6	ND	ND	ND	ND
PCB - 1254	μg/L	6	ND	ND	ND	ND
PCB - 1260	μg/L	6	ND	ND	ND	ND
Chlorinated Herbicides (8151)	79-				-140	NO
Silvex (2,4,5 - TP) (2-(2,4,5 -	μg/L	6	ND	ND	ND	ND
Trichlorophenoxy)) propionic acid						
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	μg/L	6	ND	ND	ND	ND
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/L	6	ND	ND	ND	ND
3,5 - DCBA	μg/L	6	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid)	μg/L	6	ND	ND	ND	ND
4 - Nitrophenol	μg/L	6	ND	ND	ND	ND

Table E-11. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151)						
(cont.)						
Acifluorfen	μg/L	6	ND	ND	ND	ND
Bentazon	μg/L	6	ND	ND	ND	ND
Chloramben	μg/L	6	ND	ND	ND	ND
Dacthal	μg/L	6	ND	ND	ND	ND
Dalapon	μg/L	6	ND	ND	ND	ND
Dicambia	μg/L	6	ND	ND	ND	ND
Dichloroprop [2-(2,4-	μg/L	6	ND	ND	ND	ND
Dichlorophenoxy) proponic acid]						
Dinoseb	μg/L	6	ND	ND	ND	ND
MCPA	μg/L	6	ND	ND	ND	ND
MCPP	μg/L	6	ND	ND	ND	ND
Pentachiorophenol	μg/L	6	ND	ND	ND	ND
Picloram	μg/L	6	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/L	6	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						
1-Methylnaphthalene	μg/L	6	ND	ND	ND	ND
2-Methylnaphthalene	μg/L	6	ND	ND	ND	ND
Acenaphthylene	μg/L	6	ND	ND	ND	ND
Anthracene	μg/L	6	ND	ND	ND	ND
Benzo(a)anthracene	μg/L	6	0.025	0.13	0.07	0.04
Benzo(a)pyrene	μg/L	6	0.025	0.22	0.09	0.07
Benzo(b)fluoranthene	μg/L	6	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/L	6	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/L	6	0.05	0.13	0.06	0.03
Chrysene	μg/L	6	0.05	0.21	0.08	0.06
Dibenzo(a,h)anthracene	μg/L	6	ND	ND	ND	ND
Fluoranthene	μg/L	6	ND	ND	ND	ND
Fluorene	μg/L	6	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/L	6	0.025	0.13	0.07	0.04
Naphthalene	μg/L	6	ND	ND	ND	ND
Phenanthrene	μg/L	6	ND	ND	ND	ND
Pyrene	μg/L	6	ND	ND	ND	ND
Sum of Polyaromatic Hydrocarbons	μg/L	6	ND	ND	ND	ND
Metals	~ 3					
Aluminum	mg/L	6	0.18	0.6	0.31	0.15
Antimony	mg/L	6	0.003	0.030	0.008	0.011
Arsenic	mg/L	6	ND	ND	ND	ND
Barium	mg/L	6	ND	ND	ND	ND
Beryllium	mg/L	6	ND	ND	ND	ND
Cadmium	mg/L	6	ND	ND	ND	ND
	mg/L	6	340	550	388	81
Calcium	mg/L	6	12000	25000	15833	4792
Chloride		6	ND	ND	ND	ND ND
Chromium	mg/L		ND	ND	ND	ND
Cobalt	mg/L	6	טאו ן	שאו	ַ עווו	1 110

Table E-11. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						Juli Doviduon
Copper	mg/L	6	ND	ND	ND	ND
Iron	mg/L	6	0.06	0.3	0.14	0.09
Lead	mg/L	6	0.005	0.014	0.008	0.003
Magnesium	mg/L	6	1000	2300	1267	513
Manganese	mg/L	6	ND	ND	ND	ND
Mercury	mg/L	6	ND	ND	ND	ND
Nickel	mg/L	6	ND	ND	ND	ND
Potassium	mg/L	6	450	600	523	57
Selenium	mg/L	6	ND	ND	ND	ND
Silver	mg/L	6	ND	ND	ND	ND
Sodium	mg/L	6	8700	12000	9883	1347
Thallium	mg/L	6	0.001	0.002	0.001	0.001
Vanadium	mg/L	6	ND	ND	ND	ND
Zinc	mg/L	6	ND	ND	ND	ND
Other Parameters					··	
Dissolved Oxygen	mg/L	6	5.6	10.9	7.2	2.1
Total Dissolved Solids	mg/L	6	27000	47000	33667	7581
Total Organic Carbon	mg/L	6				

Table E-10. KSC Background raw data for surface water locations.

		AVE001	BPD001	BPD002	JER001	MAX01	ROB001	100	IRL004	IRL005	OCA010	10
		07/22/1998	07/17/1998	07/17/1998	8	07/17/1998	7/22/98	/98	09/03/1998	09/03/1998 09/03/1998	96/20/60	98
								dulpicate				duplicate
Organochlorine pesticides (8081)												
4,4' - DDD	M	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
4,4' - DDE	/W/	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
4,4' - DDT	Mg/L	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Aldrin	VOV	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Alpha - BHC	ηδη.	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Beta - BHC	MI	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Chlordane (alpha)	Mg/L	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
าล)	MOL	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Total)	Mg/L	~	√ 1	V	د ا	<1	<1	<1	<۱	<1	۲>	~
Delta - BHC	MOL	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Dieldrin	MOL	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endosultan I	Mg/L	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	40.05
Endosulfan II (beta)	√6⁄/	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endosultan Sulfate	7/6/	<0.09	60:0>	60:0>	<0.09	<0.09	<0.09	<0.09	<0.09	60·0>	<0.09	¢0.09
Endrin	MON	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endrin Aldehyde	<i>μ</i> g/L	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
	VB//	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Gamma - BHC (Lindane)	Mg/L	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Heptachlor	M	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Heptachlor Epoxide (a)	Mg/L	<.050	<.050	<.050	<.050	<.050	<.050	<.050	<.050	<.050	<.050	<.050
Heptachlor Epoxide (b)	M	<.050	<.050	<.050	<.050	<.050	<.050	<.050	050'>	<.050	<.050	<.050
Isodrin	MOL	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Methoxychior	M	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Mirex	γØγ	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Toxaphene	MOL	₹	~	<2	~ 5	<2 2	7	ζ>	7	<2	42	\$
Aroclors (8082)												
PCB-1016/1242	MOL	-1	1 >	<1	1>	<1	<1	<1	1>	<1	₽	٧
PCB-1221	MOL	₹	7	<1		<1	~	\	<۱		<1	-
PCB-1232	76 <i>i</i>	₹	₩.	7	7	√1	~	<1	~	<1	<1	۲,
PCB-1248	767	₹	₹	7	₹	₹	√	\	⊽	۲۰	<1	<1
PCB-1254	Wg/L	۲۰	۲	<1	۲ ۰	<1	~ 1	<1	<1	<1	۲>	⊽
PCB-1260	JQJ	₹	7		-	.	1 >	<1	-1	<1	۲	~
Chlorinated Herbicides (8151)												
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	VOI	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5
2,4,5-Trichlorophenoxy acetic acid (2,4,5 -	Ş	505	20.5	<0.5	20.5	<0.5	20.5	, O.5	3.05	3.5	4	4
											2	2

t= value is between the Method Detection Limit and the Practical Quanitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background

screening values.

Table E-10. KSC Background raw data for surface water locations.

		AVE001	BPD001			MAX01	ROB001	100	IRL004	IRL 005	OCA010	910
	_	01122/1990		9661/1/10	0//22/1998	07/17/1998	7/22/98	98	09/03/1998	09/03/1998 09/03/1998	96/20/60	96
Chlorinated Herbicides (8151) (cont.)								duplicate				duplicate
2,4-Dichlorophenoxy acetic acid (2.4 -												
	MON	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	20.5	ć	ç	ç
3,5-DCBA	W	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	40°5	<0.5	20.5	200	6
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - IDB)	V0 /	505	3.5	3 07	3	1					200	6.5
4 - Nitrophenol	1/0/	\$ 65	505	200	500	000	5 5	000	<0.5	<0.5	<0.5	\$0.5
Acifluorfen	10,	200	3 4	500	50.5	000	60.5	8	<0.5	<0.5	<0.5	<0.5
Bentazon	1/0/	5 6	60.5	50.5	00.0	40.5	\$0.5	9.5	<0.5	<0.5	<0.5	<0.5
Chloramban		20.5	0.0	Q.5	69.5	<0.5	<0.5	85	<0.5	<0.5	<0.5	<0.5
Darthal	7 (5)	Q.5	0.00	\$0.5	<0.5	<0.5	9.5	<0.5	<0.5	<0.5	<0.5	40.5
Dalaren	₹ •	\$ 5.5	9.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	<0.5
District	\$	Q.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	<0.5
	8	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	505
Dichlorophop [2-(2,4- Dichlorophenoxy)proponic acid]	Ą	<0.5	<0.5	<0.5	20.5	70 2	, ,	ç	2		;	
Dinoseb	Ϋ́	<0.5	<0.5	<0.5	505	200	2 4	3 4	50.5	500	Q.5	50.5
MCPA	784	×10	<10	×10	9	707		3 5	20.3	0.00	5	\$0.5
MCPP	Ź	×10	950	95	010	2 5	=======================================	2 5	0 0		2	010
Pentachlorophenol	γg	<0.5	40.5	00.5	5.5	2 4	2 4	2 4	200	Oly Second	₽	×10
Picloram	Ŋ	40.5	905	205	25.6	500	2 4	200	50.5	9.5	60.5	\$0.5
PAHs (8310)					?	2007	50.5	8.5	€S	<0.0>	Q.5	<0.5
1 - Methylnarchthalana	3	3.5										
2 - Mathidaechthologo	7	000	500	Q.5	40.5	40.5	\$0.5	6 0.5	<0.5	<0.5	40.5	<0.5
A CONTRACTOR RESIDENCE	100/	6.5	9.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	8.5	<0.5
Acertapriumene	100	9.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	80.5	<0.5
Acettapriuryiene	1/0/r	9.1	9.4	Ç.	\$0.1	\$0.1	40.1	<0.1	<0.1	\$0.1	8	\$
Possolo	18	40.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5
Bonno (a) minor	100	\$ 50.05 50.05	\$0.05	<0.05	<0.05	<0.05	40.05	<0.05	<0.05	<0.05	<0.05	<0.05
Bonzo(h) ingraphene	3	39.03	S.03	<0.05	<0.05	<0.05	\$0.05	40.05	<0.05	<0.05	<0.05	<0.05
Benzola h Bendenn	100	9.5	9	50,	9.1	\$0.1	<u>6</u>	6	<0.1	<0.1	6.1	&
Ponto(b)#inconthing		- 18	5	-G-1	9.1	60.1	<u>6</u>	& ±.	<0.1	<0.1	8	8.1
Der LO (K)III OF AN III OF	100	\$0.05	<0.05	<0.05	<0.05	0.13	<0.05	<0.05	<0.05	<0.05	\$0.05	& 605
	ड्ड	<0.05	<0.05	<0.05	<0.05	0.07	<0.05	<0.05	<0.05	<0.05	\$0.05	<0.05
Inmracene	र्य	8	6.1	69.1	8	<0.1	<0.1	<0.1	69.1	6. 1.	8	9
Mene	Š	6.	\$0.1	49.1	-0.1	<0.1	-0.1	6.1	\$0.1	6.1	é	8
	78	\$ 1 .	c 0.1	0.23	<0.1	<0.1	6 1.0	8	\$ 0.1	69.	6	ę
3-cd)pyrene	Ş,	40.05	9.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	900	<0.05
Napuralene	ड्डो	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	9.5
Frienantriene	100	40.05	40.05	40.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	20.05	2002
Lyrene	100	40.05	\$0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	0.0 5

l= value is between the Method Detection Limit and the Practical Quanitation Limit.
 * = value is an outlier and not used in the analysis to develop the KSC Background screening values.

466

Table E-10. KSC Background raw data for surface water locations.

		AVE001	BPD001	BPD002	JER001	MAX01	ROB001	100	IRL004	IRL005	OCA010	10
		07/22/1998	07/17/1998	07/22/1998 07/17/1998 07/17/1998	07/22/1998 07/17/1998	07/17/1998	7/22/98	.88	09/03/1998	09/03/1998 09/03/1998	86/20/60	98
								dulpicate				duplicate
Metais												
Aluminum	mg/L	<0.05	<0.05	0.18	0.083	0.22	1.4	1.3	0.14	0.74	0.77	0.95
Antimony	mg/L	900'0>	1 800.0	0.016 1	>0.006	0.03	<0.006	<0.006	0.009	<0.006	<0.006	<0.006
Arsenic	mg/L	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Barium	mg/L	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1
Beryllium	mg/L	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001
Cadmium	mg/L	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001
Calcium	mg/L	48	180	340	9	550	120	120	310	280	6/	73
Chloride (total)	mg/L	9/	2000	14000	42	25000	870	860	14000	14000	730	099
Chromium (total)	mg/L	-0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Cobalt	mg/L		<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Copper	mg/L	_	<0.004	<0.004	<0.004	<0.004	0.007	<0.004	<0.008	<0.008	<0.008	<0.008
Iron	mg/L	0.084	0.24	<0.1	0.85	0.16	0.58	0.53	0.11	0.42	0.54	0.72
Lead	mg/L		0.005	0.013	0.007	0.014	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005
Magnesium	mg/L	8.4	110	1400	4	2300	53	54	906	830	8	67
Manganese	mg/L		0.019	0.016 1	0.02	<0.01	0.034	0.034	0.065	0.037	0.011	0.012
Mercury	mg/L		<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002
Nickel	mg/L		<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	c0.01	<0.01
Potassium	mg/L		28 I	290	7.7	260	16	16	290	260	21	77
Selenium	mg/L	_	<0.02	<0.02	<0.01	<0.02	<0.01	<0.01	0.024	0.012	\$0.01	<0.01
Silver	mg/L	_	<0.002	<0.002	<0.002	<0.002	<0.002	<0.002	0.0014	0.0011	<0.0005	<0.0005
Sodium	mg/L		1 006	7700 I	75	12000 1	460	490	8000	7100	260	550
Thallium	mg/L	•	<0.004	<0.004	<0.004	<0.004	<0.004 4	<0.004	<0.006	-0.006	×0.006	×0.006
Vanadium	mg/L	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	-0.01	<0.01	A0.01	<0.01	<0.01
Zinc	mg/L		<0.1	<0.1	<0.1	4 0.1	60.1	6 0.1	6 0.1	60.1	-60.1	<0.1
Other Parameters												
Dissolved Oxygen	mg/L	7.	11.05	8.59	4.25	5.6	6.78	NA	8.19	8.56	4.21	4.34
Total Dissolved Solids	mg/L	430	3800	27000	410	47000	2000	2000	22000	22000	1900	1900
Total Organic Carbon	mg/L	19	47	30	55	48	42	38	25	19	27	27

l= value is between the Method Detection Limit and the Practical Quanitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

	, ,										
	MEF001	1 EL004		KP1001	NSC001	BRN020 KPI001 NSC001 PEF20 SLF001	SLF001	SRE001		_	BRC09
		200		000011330	0661/0000	0001/1990	00/02/1888	8861/10/90	06/01/1998	06/01/1998	6/10/98
Organochlorine paeticidae (8004)											
000 - +'+	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
4,4 - DDE	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	20.05	300
4,4' - DDT	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	20.07	80.0
Aldrin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	40.05 0.05	2002	90.02	300	50.03
Alpha - BHC	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	50.6	800	20.03	\$0.03	\$0.05
Beta - BHC	<0.05	<0.05	<0.05	9005	<0.05	20.05	300	20.05	0.00	\$0.05	\$0.05 10.05
Chlordane (alpha)	<0.05	40 05 0 05	2005	70 OF	20.05	300	3 5	20.03	\$0.05	<0.05	<0.05
Chlordane (gemma)	200	20.0	30.0	50.00	20.05	20.05	\$0.02 0.03	<0.05	<0.05	<0.05	<0.05
Chlordane (Total)	8.5	50.03	20.00	ςς. (Ω)	\$0.02	40.05	<0.05	<0.05	<0.05	<0.05	<0.05
One of the purchase	7	v	· ·	⊽		⊽	v	~1	₩	₽	⊽
	40.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Diekarin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	000
Endosuran	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endosultan II (beta)	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endosultan Sultate	69.0 0	<0.09	<0.05	<0.05	<0.05	<0.05	<0.05	40.09	60.0>	90.02	200
Endrin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	0.05	20.05	2000
Endrin Aldehyde	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	40 05	200	20.00
Endrin Ketone	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	40.05	\$0.05	20.05	20.00	300
Gamma - BHC (Lindane)	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	200	20.0	3 5
Heptachlor	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	00.05	30.00	3 6
Heptachlor Epoxide (a)	<.050	<.050	<.050	<.050	<.050	<.050	<.050	<0.05	2005	20.00	800
Heptachlor Epoxide (b)	<.050	<.050	<.050	<.050	<.050	<.050	<.050	<0.05	<0.05	50.05	36,6
Isodrin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	00.05	2005	20.00
Methoxychlor	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	50.05	8 8
Mirex	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	50.05	20.00
Toxaphene	<2	\$	<2	\$	<2	\$	Ŋ	2	8		200
Aroclors (8082)											
PCB-1016/1242	ī	۲	7	~	<1	د1	⊽	۲×	⊽	 ->	-
PCB-1221	⊽	⊽	⊽	₽	<1	د ا	⊽	⊽	⊽	⊽	V
PCB-1232	⊽	⊽	7	~ 1	-1	۲	⊽	⊽	⊽	V	7
PCB-1248	⊽		۲	<1	<1	V	⊽	V	V	V	1
PCB-1254	₹	-1	-	-	۲	⊽	⊽	V	V	V	7
PCB-1260	-1	۲	-	1 >	1 >	⊽	⊽	V	7	7	7
Chlorinated Herbicides (8151)											7
2-(2,4,5-Trichlorophenoxy)propionic acid											
(2,4,5 - TP) (Silvex)	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	~	7	7	7
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	<0.5	505	20.5	4	4	4		,	,	,	
				200	200	200	50.3	Ī	⊽	- -	~

L= value is between the Method Detection Limit and the Practical Quanitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

	2001	TE: 004	0001400	10000	100001	00110	1001	10000	70074100	000000	00000
	09/03/1998	09/03/1998	0	06/05/1998	06/05/1998	06/01/1998	SEF001 06/05/1998	SHEUU1 06/01/1998		SHW002 06/01/1998	6/10/98
Chlorinated Herbicides (8151) (cont.)											
2,4-Dichlorophenoxy acetic acid (2,4 -											
D)	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	۲>	<1	<1
3,5-DCBA	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	<1	<1	<1
4-(2,4-Dichlorophenoxy)butyric acid (2,4 -											
DB)	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	₹	⊽	₹	7
4 - Nitrophenol	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	<1	<1	<1
Acifluorfen	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	<1	^	Ţ
Bentazon	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	1>	₹	v
Chloramben	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	1>	<1	~ 1
Dacthal	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	1 >	<1	۲>
Dalapon	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	<1	۲۷	۲۷
Dicamba	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	5 .0>	1>	L>	.	Þ
Dichloroprop [2-(2,4-Dichlorophapacy)	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<١	₽	! >	₽
Dinoseb	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	V	V	V	7
MCPA	<10	<10	÷	\$	Ş	\$	<5	×50	×50	<50	<10
MCPP	×10	×10	\$	\$	\$	Ş	<5	<50	<50	<50	<10
Pentachlorophenol	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	۲>	۲۷	<u>د</u>	۲×
Pictoram	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<1	!>	<1	د ا
PAHs (8310)											
1 - Methylnaphthalene	<0.5	<0.5	-<1	~ 1	<1	<1	<1	<0.5	<0.5	<0.5	~1
2 - Methylnaphthalene	<0.5	<0.5	<1	~	<1	<1	د ا	<0.5	<0.5	<0.5	^
Acenaphthene	<0.5	<0.5	۲	⊽	<u>م</u>		۲۰	<0.5	<0.5	<0.5	7
Acenaphthylene	<0.1	<0.1	<0.2	<0.2	<0.2	<0.2	<0.2	1>	<1	I>	<0.2
Anthracene	<0.5	<0.5	۲۷	<1	<1	<1	<1	<0.05	<0.05	<0.05	~ 1
Benzo(a)anthracene	<0.05	<0.05	<0.1	~ 0.1	c 0.1	۵.1 د	8.1	<0.05	<0.05	<0.05	6 0.1
Benzo(a)pyrene	<0.05	<0.05	40.1	-0.1	<0.1	0.18	<0.1	<0.05	<0.05	<0.05	40.1
Benzo(b)fluoranthene	40.1	& 1.	<0.2	<0.2	<0.2	<0.2	<0.2	\$0.1	8.1	40.1	<0.2
Benzo(g,h,i)perylene	<0.1	<0.1	<0.2	<0.2	<0.2	<0.2	<0.2	6 .1	-0.1	۵.1	<0.2
Benzo(k)fluoranthene	<0.05	<0.05	<0.1	<0.1	<0.1	0.21	-0.1	<0.05	<0.05	<0.05	<0.1
Chrysene	<0.05	<0.05	<0.1	<0.1	<0.1	-0.1	-0.1	<0.05	<0.05	<0.05	<0.1
Dibenzo(a,h)anthracene	40.1	<0.1	<0.2	<0.2	<0.2	<0.2	<0.2	<0.1	<0.1	<0.1	<0.2
Fluoranthene	<0.1	<0.1	<0.2	<0.2	<0.2	<0.2	<0.2	<0.1	<0.1	<0.1	<0.2
Fluorene	<0.1	<0.1	<0.2	<0.2	<0.2	<0.2	<0.2	-0.1	<0.1	<0.1	<0.2
Indeno(1,2,3-cd)pyrene	<0.05	<0.05	<0.1	<0.1	<0.1	<0.1	<0.1	<0.05	<0.05	<0.05	<0.1
Napthalene	<0.5	<0.5	~	۲	7	⊽	7	-0.5	<0.5	<0.5	^
Phenanthrene	<0.05	<0.05	₽	<0.1	69.1	60.1	-0.1	<0.05	<0.05	<0.05	40.1
Pyrene	<0.05	<0.05		<0.1	<0.1	40.1	<0.1	<0.05	<0.05	<0.05	<0.1

l= value is between the Method Detection Limit and the Practical Quanitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

	פהביסס	TCION	CONTACO	,00,00							
	09/03/1998 09	09/03/1998		06/01/1998 06/05/1998 06/05/1998 06/01/1998 06/05/1998	06/05/1998	PEF20 06/01/1998	SLF001 06/05/1998	SRE001 06/01/1998	SRW001	SRW002	BRC09
									066171000	0001/1000	98/01/0
Metals											
Aluminum	0.25	0.22	0.066	0.097	0.23	0.72	0.32	3 6	1000	300	
Antimony	0.01	40.00e	0.021	0.003	0.033	950	3000	0.30	70.00	\$0.05	0.48
Arsenic	<0.01	<0.01	0.01	1000	200	20.0	0.020	20.00	<0.005	<0.005	0.006
Barium	60	\$	5	5	5		20.01) IO:O	<0.03	<0.01	0.022
Beryllium	5	500	- 6		- 120	-	Q.1	\$0.1	<0.1	40.1	٠ <u>0</u> .1
Cadmium	9	300	00.00	100.00	20.00	40.001	<0.001	<0.001	<0.001	<0.001	0.001
Calcium	300	30.5	20.00	20.00	£0.001	40.001	<0.001	<0.001	<0.001	<0.001	<0.001
Chloride (total)	010	7 2	730	29	250	210	22	140	110	40	220
Chromium (lotal)	AUS.	2	6700	\$20	9300	6200	150	2500	320	- 20 10 10 10 10 10 10 10 10 10 10 10 10 10	6700
Cobalt	20.01	5.0	\$0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Copper	8.63	\$0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
iados.	\$0.00g	40.008	<0.002	<0.00 4	<0.004	<0.002	<0.004	<0.05	<0.05	<0.05	40 00 A
51	0.15	2.3*	<0.05	0.051	0.084	0.41	0.18	<25	0 14	0 13	200
Lead	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	0005	9000		0.63
Magnesium	800	2.3	620	မ	5200*	560	5.2	200	23	20.00	0.003
Manganese	0.034	0.27	<0.01	<0.01	<0.01	<0.01	20.03	0013	0.022	- 6	266
Mercury	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.000	20000		2000	0000	10.05
Nickel	<0.01	40.01	<0.055	40.01	<0.01	<0.055 0.055	2000	20.000	40.00Z	<0.0002 0.0002	<0.0002
Potassium	260	4.2	180	2.7	ş	170	96	200	40.01	50.01	40.01
Selenium	0.018	\$0.0	90	500	200	2 6	200	S	4.7	2.3	240
Silver	<0.0005	<0.0005	<0.0015	2000	2000	1000	20.00	50.0	\$0.01	<0.01	<0.03
Sodium	7200	80	4700	30	340	2000	20.00	Q.01	۲0.0×	40.01	<0.0006
Thallium	<0.006	9000°	7000	300	300	355	0/6	30/1	160	54	4900
Vanadium	5003	500	1000	700	70.00	40.00	\$0.005	<0.002	<0.002	<0.002	<0.002
Zinc	Ç			5,00	10.00	Q.01	<0.01	<0.01	<0.01	40.04	40.0
	-90-	- O	<0.1	<0.1	<0.1	<0.1	40.1	<0.1	60.1	6 -	\$0.1
Other Parameters											
Dissolved Oxygen	5.29	3.4	86	93.6	69.4	89.4	1060	11 93	99 9	, 0,	77.0
Total Dissolved Solids	20000	220	16000	290	10000	14000	280	75500	800	10.	47000
Total Organic Carbon	24	89	45	16	36	3	3 8	335	200	SIU	000/-
			2	2	83	77	22	33	5	<u>-</u>	- 28

l= value is between the Method Detection Limit and the Practical Quanitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

L	BRN18	BRS017 6/10/08	BRS019	IRL003	MLN001	MLN002	MLS03	03	PEF024 6/10/98	SJB02 6/10/98	SJML02 06/11/1998	IRL001
		8	+-	2001	22		5	duplicate		_		
Organochiorine pesticides (8081)												
4.4' - DDD	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
4.4' - DDE	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
4,4' - DDT	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Aldrin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Alpha - BHC	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Beta - BHC	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Chlordane (alpha)	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Chlordane (gamma)	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Chlordane (Total)	۲>	<1	<1	<1	<1	<1	۲	<1	~	<	1>	<1
Delta - BHC	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Dieldrin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	0.39
Endosulfan I	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endosultan II (beta)	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endosulfan Sulfate	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	50.0>	<0.05
Endrin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endrin Aldehyde	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endrin Ketone	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Gamma - BHC (Lindane)	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Heptachlor	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Heptachlor Epoxide (a)	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Heptachlor Epoxide (b)	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Isodrin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Methoxychior	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Mirex	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	~0.05	<0.05	<0.05	<0.05	<0.05
Toxaphene	7	8	٧	<2	<2	<2	7	2	7	7	<2	<2
Aroclors (8082)												
PCB-1016/1242	-1	√	V	7	₹	7	⊽	⊽	⊽	⊽	⊽	⊽
PCB-1221	~	v	V	۲	5	⊽	v	V	⊽	V	⊽	⊽
PCB-1232	7	~1	V	۲>	⊽	⊽	⊽	⊽	V	⊽	⊽	⊽
PCB-1248	⊽	⊽	~1	۲>	₹	⊽	⊽	⊽	Ÿ	V	⊽	⊽
PCB-1254	<1	<1	<1	<1	∵	₩	⊽	V	V	v	~	۲
PCB-1260	₽	۲>	۲۷	۲>		⊽	⊽	v	-	~	۲	7
Chlorinated Herbicides (8151)												
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	⊽	⊽	⊽	7	⊽	⊽	•	1>	1>	۲	₹	<0.5
2,4,5-Trichlorophenoxy acetic acid (2,4,5 -	⊽	⊽	⊽	⊽	⊽	⊽	⊽	⊽	₹	⊽	⊽	<0.5

I= value is between the Method Detection Limit and the Practical Quanitation Limit.
• = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

	BRN18	BRS017	BRS019	IRL003	MLN001	MLN002	MLS03	503	PFF024	SIRO	S IMI 00	100
	6/10/98	6/10/98		œ	06/11/1998	0	06/11/98	98	6/10/98	6/10/98	06/11/1998	06/17/1998
								duplicate				
Chlorinated Herbicides (8151) (cont.)												
2,4-Dichlorophenoxy acetic acid (2,4 -												
(D)	~	⊽	⊽	⊽	<1	<u>د</u> 1	٧	⊽	⊽	⊽	7	<0.5
3,5-DCBA	₹	⊽	V	₽	<1	1>	V	⊽	⊽	⊽	V	<0.5
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	7	7	*	•	•	,	•	,				
4 Nijembono	,		,	7	V	v	ī	~	v	V	<1	<0.5
A : Idition and	V	V	V	⊽		V	₹		~	<1		<0.5
Acimonten	V	V	v	⊽	٠ <u>-</u>	~	~	<1	7	ŗ	⊽	<0.5
bentazon	⊽	7	V	₹	<u>۲</u>	۲	<1	⊽	! >	₹	⊽	<0.5
Chloramben	⊽	V	v		~1	1>	⊽	⊽	v	⊽	V	<0.5
Dacthal	⊽	v	⊽	⊽	<1	<1	V	⊽	⊽	⊽	V	<0.5
Dalapon	v	<u>^</u>	⊽	<1	د ا	1>	⊽	₹	⊽	⊽	⊽	20.5
Dicamba	V	v	⊽	-	<1	<1	<1	⊽	⊽	⊽	₽	<0.5
Dichlorophop (2-(2,4-	⊽	⊽	⊽	⊽	⊽	₹	۲	⊽	⊽	⊽	⊽	
Direcah	Ţ	•	,	ļ								<0.5
DI CONTROL	v	Į.	⊽ :	V	·	V	⊽	V	⊽	⊽	~	<0.5
MCFA	012	ę	ę	ę ₽	<10	<10	<10	~ 10	<10	<10	<10	\$
MCFF	×10	ş	e V	<10	<10	<10	<10	<10	<10	5	<10	\$
Pentachlorophenol	⊽	V	⊽	٧	۸	<1	-	⊽	⊽	⊽	₽	<0.5
Pictoram	ţ	V	~	۲ <u>۰</u>	<1	⊽	~	<1	<1	7	⊽	<0.5
PAHs (8310)												
1 - Methylnaphthalene	ا	₽	-	ţ>	<1	۲۰	⊽	⊽	⊽	v	⊽	-V
2 - Methylnaphthalene	⊽	V	V	7	<1	۲۰	1 >	⊽	⊽	⊽	⊽	V
Acenaphthene	V	V	V	⊽	₹	-	~		۷1	7	⊽	⊽
Acenaphmyiene	<0.2	Q.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2
Anuracene	V	V	V	V	·	⊽	⊽	V	v	·	<1 <	<u>۲</u>
Denizo(a)anunacene	Ş.	Ş.		60.1	40.1	\$	8	0.12	<u>۾</u>	60.1	0.13 1	-0.1
Desizo(k)	-S-	÷	- - - - -	40.1	40.1	9.1	8	0.13	8	60.1	0.22	-0.1
Desizo(a h Beerdes)	20.2	9.5	8.2	\$0.2	20.5	<0.2	40.2	825	40.2	<0.2	<0.2	<0.2
Derizo(g,ii,ij)eryierie	20.2	\$0.5	8	39.5	<0.5	<0.2	9.5	49.5	40.2	<0.2	<0.2	<0.2
Denzo (k) illorarimene	Q-1	Ş.	8	40.1	9.1	& 4	8	6	8	٥°.1	<0.1	-0.1
Cilysere	\$-	÷	5	6.1	8.1	\$9.1	8	6.	\$0.1	6.1	0.21	<0.1
Ulbenzo(a,n)anmracene	<0.2	<0.2	8	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2
riuoranmene	<0.2	40.2	8	40.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	40.2	<0.2
Fluorene	40.2	\$0.2	8	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	29.5	<0.2
Indeno(1,2,3-cd)pyrene	40.1	\$	<u>6</u>	\$0.1	40.1	\$0.1	ئ 1.0	0.12	c0.1	40.1	0.13	6.1
Napthalene	· ·	v	V	⊽	₹	⊽	⊽	ţ	1	-	7	⊽
Phenanmrene	9.1	& 	8	\$ 0.1	\$0.1 1	~0.1	40.1	40.1	40.1	<0.1	1.0	6.1
Pyrene	\$0.1	9.	& 1-1	69.1	<0.1	<0.1	<0.1	<0.1	<0.1	60.1	60.1	40.1

l= value is between the Method Detection Limit and the Practical Quantitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

	BRN18	BRS017	BRS019	IRL003	MLN001	MLN002	MLS03	89	PEF024	SJB02	SJML02	IRL001
	6/10/98	6/10/98	6/10/98	06/11/1998	06/11/1998 06/11/1998 06/11/1998	06/11/1998	06/11/98	88/	6/10/98	6/10/98	6/10/98 06/11/1998 06/17/1998	06/17/1998
								duplicate				
Metals												
Aluminum	0.095	<0.05	0.068	<0.05	9.0	0.26	0.25	0.32	0.18	<0.05	0.18	0.23
Antimony	<0.006	900'0>	<0.006	900'0>	900'0>	>0.006	0.007	<0.006	0.008	<0.006	<0.006	0.028
Arsenic	0.021	0.014 1	0.019	0.016	<0.01	<0.01	<0.01	<0.01	<0.01	0.023 1	<0.01	<0.01
Barium	\$0.1	40.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1	<0.1
Beryllium	0.002	<0.001	0.002	<0.001	<0.001	<0.001	<0.001	<0.001	0.001	0.002	<0.001	<0.001
Cadmium	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001
Calcium	230	230	250	320	390	360	340	350	200	240	340	330
Chloride (total)	10000	8800	10000	14000	17000	14000	13000	14000	6300	11000	12000	16000
Chromium (total)	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Cobalt	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Copper	<0.004	<0.004	<0.004	<0.004	<0.004	<0.004	<0.004	<0.004	<0.004	<0.004	<0.004	<0.004
Iron	<0.05	<0.05	<0.05	<0.05	0.3	0.097	0.081	0.14	0.16	<0.05	0.059	<0.05
Lead	<0.005	<0.005	0.008	<0.005	0.006	0.007	0.006	0.007 1	0.006	0.005 1	0.005 1	<0.006
Magnesium	620	640	650	1000	1200	1100	1000	1000	520	650	1000	1000
Manganese	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.02
Mercury	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002
Nickel	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Potassium	250	260	320	<460	009	550	470	450	210	280	510	340
Selenium	<0.03	<0.05	<0.025	<0.03	<0.02	<0.03	<0.034	<0.02	<0.03	<0.03	<0.02	<0.01
Silver	>0.0006	<0.0006	<0.0006	<0.004	<0.004	<0.004	<0.004	<0.004	<0.0006	<0.0006	<0.004	<0.004
Sodium	5200	5400 1	5400	8800 1	11000	1 0086	8700	8900	4300 1	5200	8900 I	8700 1
Thallium	<0.002	<0.002	<0.002	0.005	<0.002	<0.002	<0.002	<0.002	<0.002	<0.002	0.002	<0.002
Vanadium	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	40.01
Zinc	<0.1	<0.1	<0.1	<0.1	c 0.1	c 0.1	80.1	۵ 1.	٥ 1.	40.1	-0.1	6 0.1
Other Parameters												
Dissolved Oxygen	7.31	6.82	5.7	12.28	9.9	6.18	10.88	ΑN	8.05	6.88	6.7	7.44
Total Dissolved Solids	16000	18000	18000	27000	34000	29000	28000	37000	14000	16000	27000	28000
Total Organic Carbon	15	16	16	21	12	12	14	16	19	24	12	20

|= value is between the Method Detection Limit and the Practical Quanitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

	IRL002 06/17/1009	SR3-D1	<u>.</u>	SR3-D2		BCE004	BCW15	115	BCW014	HPC001	100	MOC021
	001//1880	88//1/90	98//	06/17/1998	06/24/1998	06/24/1998	6/25/98	86/	06/25/1998	06/24/98	86/1	06/25/199R
Organich lering and state of the state of th			duplicate					duplicate			duplicate	200
A 1 DDD												
000 - 4,4	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	/0 O.F	90.0	100
4,4 · DUE	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	905	50.05 P.O.F.	3 5	50.05	0.05
4,4' - DDT	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	000	3 5	50.50	\$ 6	<0.05
Aldrin	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	005	200	50.00	3 4	00.00	\$0.05
Alpha - BHC	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	90.05	3 5	\$0.03	00.00	0.05 5	40.05
Beta - BHC	<0.05	<0.05	<0.05	<0.05	<0.05	20 05	30.6	3 4	50.03	00.00	\$0.05	<0.05
Chlordane (alpha)	<0.05	<0.05	<0.05	<0.05	<0.05	200	3 5	00.00	\$0.00	<0.05	\$0.05	<0.05
Chlordane (gamma)	<0.05	<0.05	<0.05	2005	30.0	50.0	3.5	S :	<0.05	<0.05	<0.05	<0.05
Chlordane (Total)	1>	V	3 7	200	60.05	\$0.05	Ş.	90.02	<0.05	<0.05	\$0.05	<0.05
Delta - BHC	<0.05	7005	700	7 9	200	V	⊽	⊽	₹	~	۷	٧
Dieldrin	20.05	30.02	30.00	\$0.00	\$ 50.05	<0.05	40.05	9.05	<0.05	<0.05	<0.05	<0.05
Endosulian	20.00	30.00	\$0.00	0.0	<0.05	<0.05	6 0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Endosulfan II (beta)	30.00	20.03	20.00	\$3.05	<0.05	<0.05	40.05	<0.05	<0.05	<0.05	A.05	<0.05
Endosulfan Sulfate	3 5	50.03	\$ 50.05	\$ 50.02	<0.05	<0.05	40.05	<0.05	<0.05	40.05	\$0.05	<0.05
Endrin	20.03	8.63	S-50	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	40.05	\$0.05	<0.05
Frdrin Aldehvde	0.00	\$ 50.00	\$0.02	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	<0.05
Endrin Katona	0.03	30.03	60.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	40.0 5	40.05	60.05
Gamma - BHC (Lindone)	0.03	30.00	50.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	40.05	80.05	0.05
Hartechlor	20.03	9.63	30.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	6 0 0 0 0 0 0	00.05
Hertachlor Ecovido (a)	20.05	89.9	89.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05 0.05	20.05
Hontachlor Epovide (h)	20.03	\$0.05	\$0.02 1	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	\$0.05	20 05
teodin	\$ 50.05	\$.65 5.65	9.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	40.05	\$0.05	2005
Methoxychlor	\$0.05	5 5	\$0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	80.05	\$0.05	40.05
Miray	\$0.05	30.05	9.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	\$0.05	\$0.05 \$0.05
Toxanhaoa	\$0.00 \$	S.03	€ 6.6	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	\$0.05	\$0.05	<0.05
Aradore (8003)	, Y	7	7	22	7	7	7	42	8	3	2	3
PCB-1016/1949												
PCR-1921	v *	₹	√,	V	v	⊽	⊽	 	۷.	₽	⊽	V
PCB-1939	1	Į.	v ,	⊽	V	⊽	⊽	₽	V	⊽	⊽	⊽
PCB-1248	7	Į,	V	·	⊽	⊽	⊽	⊽	⊽	⊽	⊽	V
100 1054 000 1054	V	⊽	<u>-</u>	₹		~	₽	⊽	⊽	⊽	V	7
#C71-001	V	⊽	⊽	▽	7	<1	⊽	⊽	⊽	V	1	1
rcb-1280	<1	⊽	⊽	۲>	⊽	₹	⊽	V	V	7		/
Chlorinated Herbicides (8151)											\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	-
2-(2,4,5-Trichlorophenoxy)propionic acid												
(c,+,3 - 1r) (Silvex)	40.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	6 0.5	25	, r	4
2,4,5-1 richiorophenoxy acetic acid (2,4,5-17)	0.5	<0.5	Z.	4	4	ų		;		2:0		0.00
			2.2	2.2	200	6.9	6.5	Q.5	60.5	<0.5	<0.5	<0.5

l= value is between the Method Detection Limit and the Practical Quanitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

	IRL002	SR3-D1	5	SR3-D2	BCE012	BCE004	BCW15	15	BCW014	HPC001	100	MOC021
	06/17/1998	06/17/98	/98	06/17/1998	06/24/1998	06/24/1998	6/22/98	98	06/25/1998	06/24/98	86/	06/25/1998
			duplicate					duplicate			duplicate	
Chlorinated Herbicides (8151) (cont.)												
2,4-Dichlorophenoxy acetic acid (2,4 -	4	ų	30	3 4	30	,						
3.5-DCBA	0.5	<0.5 0.5	0.5	0.5	0.5	0 0	0.05	000	0.0	6.5	\$ 6 5 4	\$0.5 40.5
4-(2,4-Dichlorophenoxy)butyric acid (2,4 -								200	2:27	22	200	6.0
08)	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
4 - Nitrophenol	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Acifluorfen	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Bentazon	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Chloramben	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Dacthal	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	<0.5
Dalapon	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Dicamba	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Dichloroprop [2-(2,4- Dichlorophenoxy)proponic acid]	<0.5	<0.5	6 .5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Dinoseb	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	40.5	<0.5
MCPA	\$	\$	<5	<5	<5	\$	\$	\$	\$	\$	\$	\$
MCPP	<5	\$>	Ŝ	^ 5	\$	\$	ŝ	Ş	\$	\$	\$	49
Pentachlorophenol	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5
Pictoram	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	<0.5	40.5	<0.5
PAHs (8310)												
1 - Methytnaphthalene	1 >	⊽	V	7	<1	<1	<1	-1	<1	<1	⊽	×1.1
2 - Methylnaphthalene	~	V	Ţ	٧	7	₹		⊽	<1	<1	<1	<1.1
Acenaphthene	٧	⊽	v	₹	<u>۲</u>	₽	V	~	<1	-		<1.1
Acenaphthylene	<0.2	40.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2	<0.2
Anthracene	⊽	⊽	⊽	۲۷	<1	₽	⊽	v	د ا	7	~	<1.1
Benzo(a)anthracene	\$0.1	ξ. 1.	& 	<0.1	<0.1	6 0.1	\$0.1	\$0.1	40.1	8	<0.1	<0.11
Benzo(a)pyrene	40.1	8	8	8.1	<0.1	<0.1	40.1	\$0.1	49.1	60.1	-0.1	<0.11
Benzo(b)fluoranthene	<0.2	<0.2	97	<0.2	<0.2	<0.21	<0.2	<0.2	<0.2	<0.2	<0.21	<0.22
Benzo(g,h,i)perylene	<0.2	Q.2	40.2	<0.2	<0.2	<0.21	<0.2	40.2	<0.2	<0.2	<0.21	<0.22
Benzo(k)fluoranthene	8	& 1.0	& 	6 0.1	<0.1	Å.1	6.1	8.1	-0.1	40.1	40.1	<0.11
Chrysene	8.1	& 1.1	م 1.0	<0.1	<0.1	~0.1	40.1	<0.1	<0.1	<0.1	<0.1	<0.11
Dibenzo(a,h)anthracene	<0.2	<0.2	<0.2	<0.2	<0.2	<0.21	<0.2	<0.2	<0.2	<0.2	<0.21	<0.22
Fluoranthene	<0.2	<0.2	<0.2	<0.2	<0.2	<0.21	<0.2	<0.2	<0.2	<0.2	<0.21	<0.22
Fluorene	<0.2	<0.2	\$0.5	<0.2	<0.2	<0.21	<0.2	<0.2	<0.2	<0.2	<0.21	<0.22
Indeno(1,2,3-cd)pyrene	40.1	6 0.1	& 1.1	40.1	-0 .1	40.1	<0.1	40.1	<0.1	<0.1	40.1	<0.11
Napthalene	٧	⊽	₹	1.6	۸1		₹	۲	<1	<1	<1	<1.1
Phenanthrene	40.1	80.1	ج 1.	<0.1	-0.1	6 0.1	8.1	<0.1	<0.1	<0.1	<0.1	<0.11
Pyrene	40.1	\$0.1	40.1	<0.1	<0.1	<0.1	ح0.1	<0.1	<0.1	<0.1	<0.1	<0.11

l= value is between the Method Detection Limit and the Practical Quantitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

Table E-10. KSC Background raw data for surface water locations.

	E COO	SB2.D1	خ	CD3.D3	00000	DOE OUT	4.00	27	1,00,000			
	1000	5 5	2	20-010			CI MOD	2	BCW014		5	MOC021
	06/17/1998	06/17/98	7/98	06/17/1998	06/24/1998 06/24/1998	06/24/1998	6/25/98	98	06/25/1998	06/24/98		06/25/1998
			duplicate					duplicate			duplicate	
Metals												
Aluminum	<0.05	0.32	0.3	<0.05	1.5	1.9	0.71	0.65	0.24	-	1.7	100
Antimony	0.028	0.021	0.01	0.016	<0.012	<0.012	÷0.00e	<0.012	<0.006	\$0.00°	0.015 1	<0.012
Arsenic	40.01	<0.01	<0.01	<0.01	<0.01	0.016	40.01	<0.01	<0.02	40.05	\$ 000 000	40.016
Barium	49.1	<0.1	<0.1	<0.1	<0.1	49.1	¢0.1	6.1	\$0.1	6	6	£
Beryllium	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	c0.001	40.001	<0.001	\$0.00	00.00	0000
Cadmium	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	<0.001	40.00	<0.001	40.00	900	0000
Calcium	420	400	380	160	370	240	380	360	380	230	220	390
Chloride (total)	16000	17000	17000	4800	14000	8800	15000	15000	18000	8100	800	17000
Chromium (total)	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.02	40.01	\$0.0	\$0.01
Cobalt	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05	<0.05
Copper	₹0.00	<0.004	<0.004	<0.004	<0.004	<0.004	<0.002	<0.002	<0.002	60.05 40.00	40.00	<0.002
Iron	<0.05	0.11	0.083	0.058	0.73	1.1	0.38	97.0	0.074	0.72	12	c 0.05
Lead	×0.006	<0.006	<0.005	<0.006	<0.005	<0.005	<0.005	<0.005	<0.005	<0.005	40.005	<0.005
Magnesium	1300	120	1200	340	1000	630	1100	1000	1100	290	280	1100
Manganese	<0.02	<0.02	<0.05	<0.01	<0.02	0.042	<0.02	<0.02	<0.02	\$0.01	40.01	<0.02
Mercury	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002	<0.0002
Nickel	<0.01	60.0	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	40.01	\$0.01	<0.01
Potassium	380	8	260	61	330	170	340	330	330	8	150	350
Selenium	<0.01	c0.01	<0.05	<0.01	40.01	<0.01	<0.01	<0.01	<0.01	c0.01	<0.02	<0.02
Säver	<0.004	<0.004	<0.004	<0.004	<0.002	<0.002	<0.001	<0.001	<0.001	<0.002	<0.002	<0.001
Sodium	9300	- 0086	9700	2800	7600	4700	8900	9600	8400	4400	4400	0006
Thallium	<0.002	<0.002	<0.002	<0.002	<0.002	<0.002	<0.002	<0.002	0.005	<0.002	40.002	<0.002
Vanadium	\$0.01	60.0	\$0.04	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	¢0.01	<0.01	<0.01
Zinc	<0.1	<0.1	<0.1	40.1	<0.1	<0.1	& 1.0	8.	<0.2	<u>6</u>	8	\$0.1
Other Parameters												
Dissolved Oxygen	6.67	4.78	ΑĀ	4.09	6.41	8.96	3.72	¥	10.25	5.05	ΑN	8.86
Total Dissolved Solids	30000	32000	33000	9200	22000	15000	27000	28000	27000	4000	14000	28000
Total Organic Carbon	=	24	25	11	29	37	22	22	31	46	84	31

l= value is between the Method Detection Limit and the Practical Quanitation Limit.
 = value is an outlier and not used in the analysis to develop the KSC Background screening values.

APPENDIX F

Sediment Data for Kennedy Space Center

Table F-1. Minimum, maximum, mean, and standard deviation for each parameter analyzed in all sediment samples.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)	-		T. C. C. C. C. C. C. C. C. C. C. C. C. C.	maximum	IVIGALI	Std. Deviation
4,4'-DDD	μg/kg	46	ND	ND	ND	ND
4,4'-DDE	µg/kg		ND	ND	ND	ND ND
4,4'-DDT	μg/kg		ND	ND	ND	ND ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/kg		ND	ND	ND	ND
Aldrin	μg/kg		ND	ND	ND	
Alpha - BHC	μg/kg		ND	ND	ND	ND
Beta - BHC	μg/kg		ND	ND	ND	ND
Chlordane (alpha)	μg/kg		ND	ND	ND	ND
Chlordane (gamma)	μg/kg		ND	ND	ND	ND
Chlordane (total)	μg/kg		ND	ND	ND	ND
Delta - BHC	μg/kg		ND	ND		ND
Dieldrin	μg/kg		ND	ND	ND	ND
Endosulfan I	μg/kg		ND	ND	ND ND	ND
Endosulfan II (beta)	μg/kg		ND	ND	ND	ND.
Endosulfan Sulfate	μg/kg		ND		ND	ND
Endrin	μg/kg		ND	ND	ND	ND
Endrin Aldehyde	μg/kg		ND	ND	ND	ND
Endrin Ketone		_		ND	ND ND	ND
Gamma -BHC (Lindane)	μg/kg		ND	ND	ND	ND
Heptachlor	μg/kg	46	ND ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	46	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	46	ND	ND	ND	ND
Isodrin	μg/kg	46	ND	ND	ND	ND
Methoxychlor	μg/kg	46	ND	ND	ND	ND
Mirex	μg/kg	46	ND	ND	ND	ND
	μg/kg	46	ND	ND	ND	ND
Toxaphene Arcologo (2022)	μg/kg	46	ND	ND	ND	ND
Aroclors (8082) PCB - 1016/1242	<u> </u>					
	μg/kg	46	ND	ND	ND	ND
PCB - 1221	µg/kg	46	ND	ND	ND	ND
PCB - 1232	μg/kg	46	ND	ND	ND	ND
PCB - 1248	μg/kg	46	ND	ND	ND	ND
PCB - 1254	μg/kg	46	ND	ND	ND	ND
PCB - 1260	μg/kg	46	ND	ND	ND	ND
Chlorinated Herbicides (8151)						
Silvex (2,4,5 - TP) (2-(2,4,5 -	μg/kg	46	ND	ND	ND	ND
Trichlorophenoxy) propionic acid)						
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	µg/kg	46	3	27	8.45	4.32
2,4 - D (2,4 - Dichlorophenoxy acetic acid)		46				
3,5 - DCBA	μg/kg	46	3	27	8.38	3.91
2,4 - DB (4 - (2,4 - Dichlorophenoxy)	μg/kg	46	ND	ND	ND	ND
butyric acid)	µg/kg	46	ND	ND	ND	ND
4 - Nitrophenol	μg/kg	46	ND	ND	- ND	NID
Acifluorfen		46	ND		ND	ND
	μg/kg	40	ואט	ND	ND	ND

Table F-1. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (cont.)						
Bentazon	μg/kg	46	ND	ND	ND	ND
Chloramben	μg/kg	46	ND	ND	ND	ND
Dacthal	μg/kg	46	ND	ND	ND	ND
Dalapon	μg/kg	46	ND	ND	ND	ND
Dicambia	μg/kg	46	ND	ND	ND	ND
Dichlorprop [2-(2,4-Dichlorophenoxy)	μg/kg	46	ND	ND	ND	ND
proponic acid]						
Dinoseb	μg/kg	46	ND	ND	ND	ND
MCPA	μg/kg	46	ND	ND	ND	ND
MCPP	μg/kg	46	ND	ND	ND	ND
Pentachlorophenol	μg/kg	46	ND	ND	ND	ND
Picloram	μg/kg	46	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/kg	46	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						
1-Methylnaphthalene	μg/kg	46	ND	ND	ND	ND
2-Methylnaphthalene	μg/kg	46	ND	ND	ND	ND
Acenaphthene	μg/kg	46	ND	ND	ND	ND
Acenaphthylene	μg/kg	46	ND	ND	ND	ND
Anthracene	μg/kg	46	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	46	0.95	12	1.64	1.79
Benzo(a)pyrene	μg/kg	46	1	11	1.80	1.55
Benzo(b)fluoranthene	μg/kg	46	1.5	14	2.99	2.75
Benzo(g,h,i)perylene	μg/kg	46	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/kg	46	1	10	1.86	1.70
Chrysene	μg/kg	46	0.95	42	2.71	6.26
Dibenzo(a,h)anthracene	μg/kg	46	ND	ND	ND	ND
Fluoranthene	μg/kg	46	1.95	67	4.14	9.55
Fluorene	μg/kg	46	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	46	ND	ND	ND	ND
Naphthalene	μg/kg	46	9.5	67	15.37	16.33
Phenanthrene	μg/kg	46	0.95	10	1.68	2.92
Pyrene	μg/kg	46	0.95	34	2.23	4.59
Metals						
Aluminum	mg/kg	46	210	2800	977	632
Antimony	mg/kg	46	ND	ND	ND	ND
Arsenic	mg/kg	46		3.3	0.82	0.71
Barium	mg/kg	46		ND	ND	ND
Beryllium	mg/kg			ND	ND	ND
Cadmium	mg/kg	46		ND	ND	ND
Calcium	mg/kg	46		220000	52208	63423
Chromium	mg/kg	46		6.4	2.47	1.59
Cobalt	mg/kg	46	ND	ND	ND	ND
Copper	mg/kg	46		7	3.25	1.15
Iron	mg/kg	46	150	3200	967	756

Table F-1. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)		1			1010411	Ota. Deviation
Lead	mg/kg	46	0.5	6	2.35	1.36
Magnesium	mg/kg	46	42	3800	974	882
Manganese	mg/kg	46	0.5	110	11.31	17.85
Mercury	mg/kg	46	0.005	0.019	0.008	0.004
Nickel	mg/kg	46	ND	ND	ND	ND
Potassium	mg/kg	46	16	1200	354	299
Selenium		$\overline{}$	ND	ND	ND	ND ND
Silver	mg/kg	46	1	3.4	1.61	0.57
Sodium	mg/kg		19	10000	2865	2149
Thallium		-	0.5	5.4	0.82	0.82
Vanadium	mg/kg	_	0.5	7.7	2.58	1.86
Zinc			3	27	7.26	6.12

Table F-2. Minimum, maximum, mean, and standard deviation for each parameter analyzed in Banana Creek sediment samples.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)						•
4,4'-DDD	μg/kg	8	ND	ND	ND	ND
4,4'-DDE	μg/kg	8	ND	ND	ND	ND
4,4'-DDT	μg/kg	8	ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/kg	8	ND	ND	ND	ND
Aldrin	μg/kg	8	ND	ND	ND	ND
Alpha - BHC	μg/kg	8	ND	ND	ND	ND
Beta - BHC	μg/kg	8	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	8	ND	ND	ND	ND
Chlordane (gamma)	μg/kg	8	ND	ND	ND	ND
Chiordane (total)	μg/kg	8	ND	ND	ND	ND
Delta - BHC	μg/kg	8	ND	ND	ND	ND
Dieldrin	μg/kg	8	ND	ND	ND	ND
Endosulfan I	μg/kg	8	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	8	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	8	ND	ND	ND	ND
	μg/kg	8	ND	ND	ND	ND ND
Endrin	μg/kg	8	ND	ND	ND	ND ND
Endrin Aldehyde	μg/kg	8	ND	ND	ND	ND ND
Endrin Ketone		8	ND	ND	ND	ND ND
Gamma -BHC (Lindane)	μg/kg	8	ND	ND	ND	ND ND
Heptachlor	μg/kg	-	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	8	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	8		ND	ND	ND
Isodrin	μg/kg	8	ND		ND	ND
Methoxychlor	μg/kg	8	ND	ND ND		ND
Mirex	μg/kg	8	ND	ND	ND	
Toxaphene	µg/kg	8	ND	ND	ND_	ND
Aroclors (8082)	ļ <u>.</u>	Ļ		110	110	ND
PCB - 1016/1242	μg/kg	8	ND	ND	ND	ND ND
PCB - 1221	μg/kg	8	ND	ND	ND	ND ND
PCB - 1232	μg/kg	8	ND	ND	ND	ND
PCB - 1248	μg/kg	8	ND	ND	ND	ND
PCB - 1254	µg/kg	8	ND	ND	ND	ND ND
PCB - 1260	µg/kg	8	ND	ND	ND	ND
Chlorinated Herbicides (8151)	ļ	_				<u> </u>
Silvex (2,4,5 - TP) (2-(2,4,5 -	μg/kg	8	ND	ND	ND	ND
Trichlorophenoxy) propionic acid)	- ()-	ا ہ	ND -	ND	ND	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	μg/kg	8	ND	ND	ND	ND
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/kg	8	ND	ND	ND	ND
3,5 - DCBA	μg/kg	8	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy)	μg/kg	8	ND	ND	ND	ND
butyric acid)		\perp		ļ		
4 - Nitrophenol	μg/kg	8	ND	ND	ND	ND
Acifluorfen	μg/kg	8	ND	ND	ND	ND

Table F-2. (cont.)

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (cont.)		T				
Bentazon	μg/kg	8	ND	ND	ND	ND
Chloramben	μg/kg	8	ND	ND	ND	ND
Dacthal	μg/kg	8	ND	ND	ND	ND
Dalapon	μg/kg	8	ND	ND	ND	ND
Dicambia	μg/kg	8	ND	ND	ND	ND
Dichlorprop [2-(2,4-Dichlorophenoxy)	μg/kg	8	ND	ND	ND	ND
proponic acid]				<u> </u>		
Dinoseb	µg/kg	8	ND	ND	ND	ND
MCPA	µg/kg	8	ND	ND	ND	ND
MCPP	µg/kg	8	ND	ND	ND	ND
Pentachlorophenol	μg/kg	8	ND	ND	ND	ND
Picloram	μg/kg	8	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/kg	8	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						
1-Methylnaphthalene	μg/kg	8	ND	ND	ND	ND
2-Methylnaphthalene	μg/kg	8	ND	ND	ND	ND
Acenaphthene	μg/kg	8	ND	ND	ND	ND
Acenaphthylene	μg/kg	8	ND	ND	ND	ND
Anthracene	µg/kg	8	ND	ND	ND	ND
Benzo(a)anthracene	µg/kg	8	ND	ND	ND	ND
Benzo(a)pyrene	µg/kg	8	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/kg	8	0.002	0.004	0.002	0.001
Benzo(g,h,i)perylene	μg/kg	8	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/kg	8	ND	ND	ND	ND
Chrysene	μg/kg	8	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/kg	8	ND	ND	ND	ND
Fluoranthene	μg/kg	8	0.002	0.009	0.003	0.003
Fluorene	μg/kg	8	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	8	ND	ND	ND	ND
Naphthalene	μg/kg	8	ND	ND	ND	ND
Phenanthrene	μg/kg	8	ND	ND	ND	ND
Pyrene	μg/kg	8	ND	ND	ND	ND
Metals						
Aluminum	mg/kg	8	620	1900	1119	453
Antimony	mg/kg	8	ND	ND	ND	ND
Arsenic	mg/kg	_	0.35	2.3	1.36	0.87
Barium	mg/kg	8	ND	ND	ND	ND
Beryllium	mg/kg		ND	ND	ND	ND
Cadmium	mg/kg		ND	ND	ND	ND
Calcium		8	3700	160000	68100	69314
Chromium	mg/kg	8	1.7	4.4	2.56	0.94
Cobalt	mg/kg	_	ND	ND	ND	ND
Copper	mg/kg		0.5	3.4	1.85	1.16
ron	mg/kg	_	440	1700	1021	447
ead	mg/kg	_	0.5	3	2.05	1.01

Table F-2. (cont.)

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Magnesium	mg/kg	8	110	1900	969	566
Manganese	mg/kg	8	4.2	10	6.6	2.3
Mercury	mg/kg	8	ND	ND	ND	ND
Nickel	mg/kg		ND	ND	ND	ND
Potassium	mg/kg	8	100	680	471	183
Selenium	mg/kg	8	ND	ND	ND	ND
Silver	mg/kg	8	1	3	1.96	0.80
Sodium	mg/kg	8	130	3600	2791	1115
Thallium	mg/kg		0.5	1.8	0.8	0.5
Vanadium	mg/kg		1.8	4.1	2.6	0.9
Zinc	mg/kg		3	19	5.9	5.6

Table F-3. Minimum, maximum, mean, and standard deviation for each parameter analyzed in Banana River sediment samples.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)	-	+	Total III	WICKITICHT	IVIGALI	Sid. Deviation
4,4'-DDD	μg/kg	13	ND	ND	ND	ND
4,4'-DDE	μg/kg	_	ND	ND	ND	ND
4,4'-DDT	μg/kg		ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/kg	_	ND	ND	ND	ND
Aldrin	μg/kg		ND	ND	ND	ND
Alpha - BHC	μg/kg		ND	ND	ND	ND
Beta - BHC	μg/kg		ND	ND	ND	ND
Chlordane (alpha)	μg/kg		ND	ND	ND	ND
Chlordane (gamma)	μg/kg		ND	ND	ND	ND
Chlordane (total)	μg/kg		ND	ND	ND	ND
Delta - BHC	μg/kg		ND	ND	ND	ND ND
Dieldrin	μg/kg	13	ND	ND	ND	ND
Endosulfan I	μg/kg	13	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	13	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	13	ND	ND	ND	ND
Endrin	μg/kg	13	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	13	ND	ND	ND	ND
Endrin Ketone	μg/kg	13	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/kg	13	ND	ND	ND	ND
Heptachlor	μg/kg	13	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	13	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	13	ND	ND	ND	ND
Isodrin	μg/kg	13	ND	ND	ND	ND ND
Methoxychlor	μg/kg	13	ND	ND	ND	ND
Mirex	μg/kg	13	ND	ND	ND	ND ND
Toxaphene	μg/kg	13	ND	ND	ND	ND
Aroclors (8082)	1	1			140	IND
PCB - 1016/1242	μg/kg	13	ND	ND	ND	ND
PCB - 1221	μg/kg	13	ND	ND	ND	ND
PCB - 1232	μg/kg	13	ND	ND	ND	ND
PCB - 1248	μg/kg	13	ND	ND	ND	ND
PCB - 1254	μg/kg	13	ND	ND	ND	ND ND
PCB - 1260	μg/kg	13	ND	ND	ND	ND
Chlorinated Herbicides (8151)	1	1			140	NO
Silvex (2,4,5 - TP) (2-(2,4,5 - Trichlorophenoxy) propionic acid)	μg/kg	13	ND	ND	ND	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	µg/kg	13	0.005	0.027	0.010	0.006
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/kg	13	0.005	0.001	0.010	0.005
B,5 - DCBA	μg/kg μg/kg	13	0.005 ND	0.021	0.010	0.005
2,4 - DB (4 - (2,4 - Dichlorophenoxy)	μg/kg μg/kg	13	ND ND	ND	ND	ND
outyric acid)	#9/kg	13	טאו	ND	ND	ND
- Nitrophenol	μg/kg	13	ND	ND	ND	ND
Acifluorfen	μg/kg	13	ND	ND	ND	ND
	1 - 99			170	טוו	110

Table F-3. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (cont.)						
Bentazon	μg/kg	13	ND	ND	ND	ND
Chloramben	μg/kg	13	ND	ND	ND	ND
Dacthal	μg/kg	13	ND	ND	ND	ND
Dalapon	μg/kg	13	ND	ND	ND	ND
Dicambia	μg/kg	13	ND	ND	ND	ND
Dichlorprop [2-(2,4-Dichlorophenoxy)	μg/kg	13	ND	ND	ND	ND
proponic acid]						
Dinoseb	µg/kg	13	ND	ND	ND	ND
MCPA	μg/kg	13	ND	ND	ND	ND
MCPP	μg/kg	13	ND	ND	ND	ND
Pentachlorophenol	µg/kg	13	ND	ND	ND	ND
Picloram	µg/kg	13	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/kg	13	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)	1					
1-Methylnaphthalene	μg/kg	13	ND	ND	ND	ND
2-Methylnaphthalene	μg/kg	13	ND	ND	ND	ND
Acenaphthene	μg/kg	13	ND	ND	ND	ND
Acenaphthylene	μg/kg	13	ND	ND	ND	ND
Anthracene	μg/kg	13	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	13	0.001	0.006	0.002	0.002
Benzo(a)pyrene	μg/kg	13	0.001	0.003	0.002	0.001
Benzo(b)fluoranthene	μg/kg	13	0.002	0.014	0.003	0.003
Benzo(g,h,i)perylene	μg/kg	13	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/kg	13	0.001	0.009	0.002	0.002
Chrysene	μg/kg	13	0.001	0.013	0.003	0.003
Dibenzo(a,h)anthracene	μg/kg	13	ND	ND	ND	ND
Fluoranthene	μg/kg	13	0.002	0.067	0.007	0.018
Fluorene	μg/kg	13	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	13	ND	ND	ND	ND
Naphthalene	μg/kg	13	ND	ND	ND	ND
Phenanthrene	μg/kg	13	0.001	0.007	0.002	0.002
Pyrene	μg/kg	13	0.001	0.034	0.004	0.009
Metals						
Aluminum	mg/kg	13	210	2200	915	580
Antimony	mg/kg	13	ND	ND	ND	ND
Arsenic	mg/kg	13	0.35	2	0.86	0.49
Barium	mg/kg	13		ND	ND	ND
Beryllium	mg/kg	13		ND	ND	ND
Cadmium	mg/kg	13	ND	ND	ND	ND
Calcium	mg/kg			220000	50540	60407
Chromium	mg/kg			5.2	2.9	1.0
Cobalt	mg/kg	_		ND	ND	ND
Copper	mg/kg			ND	ND	ND
Iron	mg/kg	13	150	3200	1166	930

Table F-3. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Lead	mg/kg	13	2	6	3.4	1.4
Magnesium	mg/kg	13	42	2100	892	633
Manganese	mg/kg	13	1	27	10.6	8.2
Mercury	mg/kg	_	0.005	0.01	0.007	0.003
Nickel	mg/kg	13	ND	ND	ND	ND
Potassium	mg/kg		19	790	286	241
Selenium	mg/kg	_	ND	ND	ND	ND
Silver	mg/kg	_	1	2.4	1.5	0.4
Sodium		_	19	4800	1940	1498
Thallium	mg/kg	13	0.5	5.4	0.99	1.34
Vanadium	mg/kg	13	0.5	7.7	3.0	2.1
Zinc			3	27	10.8	6.2

Table F-4. Minimum, maximum, mean, and standard deviation for each parameter analyzed in Indian River sediment samples.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochiorine Pesticides (8081)						
4,4'-DDD	μg/kg	16	ND	ND	ND	ND
4,4'-DDE	μg/kg	16	ND	ND	ND	ND
4,4'-DDT	μg/kg	16	ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/kg	16	ND	ND	ND	ND
Aldrin	μg/kg	16	ND	ND	ND	ND
Alpha - BHC	μg/kg	16	ND	ND	ND	ND
Beta - BHC	μg/kg	16	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	16	ND	ND	ND	ND
	μg/kg	16	ND	ND	ND	ND
Chlordane (gamma)	μg/kg	16	ND	ND	ND	ND
Chlordane (total)	μg/kg	16	ND	ND	ND	ND
Delta - BHC		16	ND	ND	ND	ND
Dieldrin	μg/kg	16	ND	ND	ND	ND
Endosulfan I	μg/kg			ND	ND	ND ND
Endosulfan II (beta)	μg/kg	16	ND		ND	ND
Endosulfan Sulfate	μg/kg	16	ND	ND		ND ND
Endrin	μg/kg	16	ND	ND	ND	ND ND
Endrin Aldehyde	μg/kg	16	ND	ND	ND	
Endrin Ketone	μg/kg	16	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/kg	16	ND	ND	ND	ND
Heptachlor	μg/kg	16	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	16	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	16	ND	ND	ND	ND
Isodrin	μg/kg	16	ND	ND	ND	ND
Methoxychlor	μg/kg	16	ND	ND	ND	ND
Mirex	µg/kg	16	ND	ND	ND	ND
Toxaphene	μg/kg	16	ND	ND	ND	ND
Arociors (8082)						
PCB - 1016/1242	μg/kg	16	ND	ND	ND	ND
PCB - 1221	μg/kg	16	ND	ND	ND	ND
PCB - 1232	μg/kg	16	ND	ND	ND	ND
PCB - 1248	μg/kg	16	ND	ND	ND	ND
PCB - 1254	μg/kg	16	ND	ND	ND	ND
PCB - 1260	μg/kg	16	ND	ND	ND	ND
Chlorinated Herbicides (8151)				-	_	
Silvex (2,4,5 - TP) (2-(2,4,5 -	μg/kg	16	ND	ND	ND	ND
Trichlorophenoxy) propionic acid)						
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	μg/kg	16	ND	ND	ND	ND
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/kg	16	ND	ND	ND	ND
3,5 - DCBA	μg/kg	16	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy) butyric acid)	μg/kg	16	ND	ND	ND	ND
4 - Nitrophenol	μg/kg	16	ND	ND	ND	ND
Acifluorfen	μg/kg	16	ND	ND	ND	ND

Table F-4. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (8151) (cont.)		1		1		- Cita Bornanon
Bentazon	μg/kg	16	ND	ND	ND	ND
Chloramben	µg/kg	16	ND	ND	ND	ND
Dacthal	μg/kg	16	ND	ND	ND	ND
Dalapon	μg/kg	16	ND	ND	ND	ND
Dicambia	μg/kg	16	ND	ND	ND	ND
Dichlorprop [2-(2,4-Dichlorophenoxy) proponic acid]	μg/kg	16	ND	ND	ND	ND
Dinoseb	μg/kg	16	ND	ND	ND	ND
MCPA	μg/kg	16	ND	ND	ND	ND
MCPP	μg/kg	16	ND	ND	ND	ND
Pentachlorophenol	μg/kg	16	ND	ND	ND	ND
Picloram	μg/kg	16	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/kg	16	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)	1					1
1-Methylnaphthalene	μg/kg	16	ND	ND	ND	ND
2-Methylnaphthalene	μg/kg	16	ND	ND	ND	ND
Acenaphthene	μg/kg	16	ND	ND	ND	ND
Acenaphthylene	μg/kg	16	ND	ND	ND	ND
Anthracene	μg/kg	16	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	16	0.001	0.012	0.002	0.003
Benzo(a)pyrene	µg/kg	16	0.001	0.011	0.002	0.003
Benzo(b)fluoranthene	μg/kg	16	0.002	0.013	0.003	0.004
Benzo(g,h,i)perylene	μg/kg	16	ND	ND	ND	ND
Benzo(k)fluoranthene	μg/kg	16	0.001	0.01	0.002	0.002
Chrysene	μg/kg	16	0.001	0.042	0.004	0.010
Dibenzo(a,h)anthracene	μg/kg	16	ND	ND	ND	ND
Fluoranthene	μg/kg	16	0.002	0.005	0.003	0.001
Fluorene	μg/kg	16	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	16	ND	ND	ND	ND
Naphthalene	μg/kg	16	0.011	0.067	0.016	0.014
Phenanthrene	μg/kg	16	ND	ND	ND	ND
Pyrene	μg/kg	16	ND	ND	ND	ND
Metals						
Aluminum	mg/kg	16	230	2800	837	739
Antimony		16	ND	ND	ND	ND
Arsenic	mg/kg	16	0.3	3.3	0.7	0.8
Barium		16	ND	ND	ND	ND
Beryllium	mg/kg		ND	ND	ND	ND
Cadmium	mg/kg		ND	ND	ND	ND
Calcium		16	450	220000	50145	68188
Chromium	mg/kg	16	0.5	6.4	1.7	1.7
Cobalt	mg/kg	$\overline{}$	ND	ND	ND	ND
Copper	mg/kg		0.5	7	3.4	1.3

Table F-4. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)						
Iron	mg/kg	16	150	1700	555	431
Lead	mg/kg	16	0.5	4.7	2.2	1.3
Magnesium	mg/kg	16	63	1800	562	498
Manganese	mg/kg	16	0.5	110	10.9	26.6
Mercury	mg/kg	16	0.005	0.017	0.008	0.005
Nickel	mg/kg	16	ND	ND	ND	ND
Potassium	mg/kg	16	16	820	211	212
Selenium	mg/kg	16	ND	ND	ND	ND
Silver	mg/kg	16	ND	ND	ND	ND
Sodium	mg/kg	16	26	6700	2424	2048
Thallium	mg/kg	16	0.5	2.9	0.7	0.6
Vanadium	mg/kg	16	0.5	5.5	1.8	1.7
Zinc	mg/kg	16	3	25	6.4	6.8

Table F-5. Minimum, maximum, mean, and standard deviation for each parameter analyzed in Mosquito Lagoon sediment samples.

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Organochlorine Pesticides (8081)		+-			, , , , ,	Old. Deviation
4,4'-DDD	μg/kg	9	ND	ND	ND	ND
4,4'-DDE	μg/kg		ND	ND	ND	ND
4,4'-DDT	μg/kg		ND	ND	ND	ND
Sum 4,4'-DDD + 4,4'-DDE + 4,4'-DDT	μg/kg		ND	ND	ND	ND
Aldrin	μg/kg	9	ND	ND	ND	ND
Alpha - BHC	μg/kg	9	ND	ND	ND	ND
Beta - BHC	μg/kg	9	ND	ND	ND	ND
Chlordane (alpha)	μg/kg	9	ND	ND	ND	ND
Chlordane (gamma)	μg/kg	9	ND	ND	ND	ND
Chlordane (total)	μg/kg	9	ND	ND	ND	ND
Delta - BHC	μg/kg	9	ND	ND	ND	ND
Dieldrin	μg/kg	9	ND	ND	ND	ND
Endosulfan i	μg/kg	9	ND	ND	ND	ND
Endosulfan II (beta)	μg/kg	9	ND	ND	ND	ND
Endosulfan Sulfate	μg/kg	9	ND	ND	ND	ND
Endrin	μg/kg	9	ND	ND	ND	ND
Endrin Aldehyde	μg/kg	9	ND	ND	ND	ND
Endrin Ketone	μg/kg	9	ND	ND	ND	ND
Gamma -BHC (Lindane)	μg/kg	9	ND	ND	ND	ND
Heptachlor	μg/kg	9	ND	ND	ND	ND
Heptachlor Epoxide (a)	μg/kg	9	ND	ND	ND	ND
Heptachlor Epoxide (b)	μg/kg	9	ND	ND	ND	ND
Isodrin	μg/kg	9	ND	ND	ND	ND
Methoxychlor	μg/kg	9	ND	ND	ND	ND
Mirex	μg/kg	9	ND	ND	ND	ND
Toxaphene	μg/kg	9	ND	ND	ND	ND
Aroclors (8082)	1					110
PCB - 1016/1242	μg/kg	9	ND	ND	ND	ND
PCB - 1221	μg/kg	9	ND	ND	ND	ND
PCB - 1232	μg/kg	9	ND	ND	ND	ND
PCB - 1248	μg/kg	9	ND	ND	ND	ND
PCB - 1254	μg/kg	9	ND	ND	ND	ND
PCB - 1260	μg/kg	9	ND	ND	ND	ND
Chlorinated Herbicides (8151)		Ħ				
Silvex (2,4,5 - TP) (2-(2,4,5 - Trichlorophenoxy) propionic acid)	μg/kg	9	ND	ND	ND	ND
2,4,5 - T (2,4,5 - Triclorophenoxy acetic acid)	μg/kg	9	ND	ND	ND	ND
2,4 - D (2,4 - Dichlorophenoxy acetic acid)	μg/kg	9	ND	ND	ND	ND
3,5 - DCBA	μg/kg	9	ND	ND	ND	ND
2,4 - DB (4 - (2,4 - Dichlorophenoxy) outyric acid)	μg/kg	9	ND	ND	ND	ND
1 - Nitrophenol	μg/kg	9	ND	ND	ND	ND
Acifluorfen	μg/kg	9	ND	ND	ND	ND

Table F-5. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Chlorinated Herbicides (cont.)	1	<u> </u>				
Bentazon	μg/kg	9	ND	ND	ND	ND
Chloramben	μg/kg	9	ND	ND	ND	ND
Dacthal	μg/kg	9	ND	ND	ND	ND
Dalapon	μg/kg	9	ND	ND	ND	ND
Dicambia	μg/kg	9	ND	ND	ND	ND
Dichlorprop [2-(2,4-Dichlorophenoxy) proponic	μg/kg	9	ND	ND	ND	ND
acid]						
Dinoseb	μg/kg	9	ND	ND	ND	ND
MCPA	µg/kg	9	D	ND	ND	ND
MCPP	μg/kg	9	ND	ND	ND	ND
Pentachlorophenol	μg/kg	9	ND	ND	ND	ND
Picloram	μg/kg	9	ND	ND	ND	ND
Sum of Chlorinated Herbicides	μg/kg	9	ND	ND	ND	ND
Polyaromatic Hydrocarbons (8310)						
1-Methylnaphthalene	μg/kg	9	ND	ND	ND	ND
2-Methylnaphthalene	µg/kg	9	ND	ND	ND	ND
Acenaphthene	µg/kg	9	ND	ND	ND	ND
Acenaphthylene	µg/kg	9	ND	ND	ND	ND
Anthracene	μg/kg	9	ND	ND	ND	ND
Benzo(a)anthracene	μg/kg	9	ND	ND	ND	ND
Benzo(a)pyrene	μg/kg	9	ND	ND	ND	ND
Benzo(b)fluoranthene	μg/kg	9	ND	ND	ND	ND
Benzo(g,h,i)perylene	μg/kg	9	ND	ND	ND	ND
Benzo(k)fluoranthene	µg/kg	9	ND	ND	ND	ND
Chrysene	μg/kg	9	ND	ND	ND	ND
Dibenzo(a,h)anthracene	μg/kg	9	ND	ND	ND	ND
Fluoranthene	μg/kg	9	ND	ND	ND	ND
Fluorene	μg/kg	9	ND	ND	ND	ND
Indeno(1,2,3-cd)pyrene	μg/kg	9	ND	ND	ND	ND
Naphthalene	μg/kg	9	0.012	0.054	0.023	0.016
Phenanthrene	μg/kg	9	0.001	0.01	0.002	0.003
Pyrene	μg/kg	9	0.001	0.015	0.003	0.005
Metals						
Aluminum	mg/kg	9	500	2300	1196	671
Antimony	mg/kg	9	ND	ND	ND	ND
Arsenic	mg/kg	9	0.35	0.9	0.54	0.25
Barium	mg/kg	9	ND	ND	ND	ND
Beryllium	mg/kg	9	ND	ND	ND	ND
Cadmium	mg/kg	9	ND	ND	ND	ND
Calcium	mg/kg	9	19	180000	44158	62393
Chromium	mg/kg	9	0.5	6	3.2	2.2
Cobalt	mg/kg	9	ND	ND	ND	ND
Copper	mg/kg	9	ND	ND	ND	ND

Table F-5. (cont.).

Parameter	Units	N	Minimum	Maximum	Mean	Std. Deviation
Metals (cont.)				····coxiii ···	Widan	Old. Deviation
Iron	mg/kg	9	500	2800	1363	915
Lead	mg/kg	9	0.5	2.3	1.4	0.7
Magnesium	mg/kg	9	280	3800	1938	1376
Manganese	mg/kg	9	3	55	17	17
Mercury	mg/kg	9	0.005	0.019	0.009	0.005
Nickel	mg/kg	9	ND	ND	ND	ND ND
Potassium	mg/kg	9	140	1200	635	413
Selenium	mg/kg	9	ND	ND	ND	ND
Silver	mg/kg	9	1	3.4	1.7	0.7
Sodium	mg/kg	9	2200	10000	5325	2449
Thallium	mg/kg	9	ND	ND	ND	ND
Vanadium	mg/kg	9	0.5	6	3.2	2.2
Zinc	mg/kg	9	3	13	5	3.1

Table F-6. KSC Background raw data for sediment locations.

		AVE001	BPD001	BPD002	JER001	MAXO1	ROB001	10	IRL004	IRL005
		07/22/1998	07/17/1998	07/17/1998	07/22/1998	07/17/1998	07/22/98	98	09/03/1998	09/03/1998
								duplicate		
Organochlorine pesticides (8081)										
4,4' - DDD	1/g/kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
4,4' - DDE	10/kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
4,4' - DDT	νδγkg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Aldrin	<i>μ</i> φ/kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Alpha - BHC	₩6/	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Beta - BHC	<i>µ</i> g∕kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Chlordane, alpha	µg∕kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Chlordane, gamma	µg∕kg	-6.5	<4.6	<4.3	<4.8	6.4>	<4.3	<4.3	<4.5	<4.3
Chlordane (Total)	<i>μ</i> g/kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Delta - BHC	µg∕kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Diełdrin	µg∕kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Endosulfan I	<i>μ</i> g/kg	<6.5	6.45	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Endosulfan II (beta)	₽9/kg	46.5	64.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Endosulfan Sulfate	09∕kg	-6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Endrin	µg∕kg	9:9>	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Endrin Aldehyde	Ø⁄kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Endrin Ketone	₩0/kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Gamma - BHC (Lindane)	₽9/kg	<6.5	64.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Heptachlor	µg∕kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Heptachlor Epoxide(a)	<i>μ</i> g/kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Heptachlor Epoxide(b)	μg/kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Isodrin	ng∕kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Methoxychlor	υ g/k g	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Mirex	ng∕kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Toxaphene	<i>μ</i> g/kg	<130	<92	<86	96>	<98	<87	-88 -88	<90	<86
Aroclors (8082)										
PCB-1016/1242	1/0/kg	<65	<46	<43	<48	<49	<43	<43	<45	<43
PCB-1221	₩ô/kg	<65	<46	<43	<48	<49	<43	<43	<45	<43
PCB-1232	μg/kg	<65	<46	<43	<48	<49	<43	<43	<45	<43
PCB-1248	µg∕kg	<65	<46	<43	<48	<49	<43	<43	<45	<43
PCB-1254	μg/kg	<65	<46	<43	<48	<49	643	<43	<45	<43
PCB-1260	₽\$/6⁄7	<65	<46	<43	<48	<49	<43	<43	<45	<43

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

		AVE001	BPD001	BPD002	JER001	MAXO1	ROBOOT	-	IBI OOA	300 101
		07/22/1998	07/17/1998	07/17/1998	07/22/1998	07/17/1998	07/22/98	88	09/03/1998	09/03/1998
								duplicate		
Chlorinated Herbicides (8151)										
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	w/ka	07.7	717	67,	,					
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	A D		-	2	v	615	ę	ę	<14	<13
	<i>µ</i> g∕kg	<10	<14	<13	1 >	<15	φ	ę	×14	57
2,4-Dichlorophenoxy acetic acid (2,4 - D)	<i>μ</i> g/kg	<10	<14	<13	12	<15 <15	9	4	144	2 5
3,5-DCBA	6¥⁄6⁄⁄	<10	<14	<13	12	<15	ę	9 9	212	2 5
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	µg∕kg	<10	<14	<13	<7	<15	ę	9	4	÷
4 - Nitrophenol	Mg/kg	×10	<14	<13	<7	<15	ę	9>	^14 414	×13
Actiluorien	MO/Kg	<10	<14	<13	- 2	<15	9>	9>	×14	<13
Bertazon	Mg/kg	<10	<14	<13	<7	<15	9	9	×14	×13
Chloramben	ης/γ γ	<10	<14	<13	<i>L></i>	<15	9	9>	41×	×13
Dachal	₩ _O W	~10	<14	<13	- 2	<15	9	ş	414	×13
Dalapon	₩0	~ 10	<14	<13	<i>L</i> >	<15	ę	9	4.	×13
Dicamba	MVK9	<10	<14	<13	<i>L</i> >	<15	9	9	×14	773
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic										2
acidj	₩ø/	<10	<14	<13	<7	<15	\$	ę	41^	×13
Dinoseb	₩Q/	<10	<14	<13	-22	<15	ę	9	41^	×13
MCPA	Mg/Kg	~ 5 00	<140	<130	<140	<150	230 230	<130	<270	<260
MCPP	<i>µ</i> g∕kg	<200	<140	<130	<140	<150	<130	×130	<270	2,560
Pentachlorophenol	Mg/kg	<10	<14	<13	-<7	<15	ę	ę	<14	×13
ridoram	₩g/kg	<10	<14	<13	<i>-</i> 22	<15	æ	9	<14	<13
Polyaromatic Hydrocarbons (8310)										
1 - Methylnaphthalene	µg∕kg	<33	<23	<22	424	<25 -25	22	222	562	200
2 - Methylnaphthalene	µg∕kg	<33	<23	<22	424	425	422	8	8	60
Acenaphthene	µg∕kg	<33	<23	<22	<24	<25	25	422	423	223
Acenaphthylene	₩ _Q /kg	<6.5	<4.6	<4.3	<4.8	<4.9	<4.3	<4.3	<4.5	<4.3
Anthracene	Wg/kg	<33	<24	<22	<25	<25	25	25	\$3	22
Benzo(a)anthracene	₩0/kg	<3.3	<2.3	<2.2	<2.4	<2.5	2.2	<2.2	12	222
Benzo(a)pyrene	<i>µ</i> g∕kg	4>	\$3	\$	<3	ဗ	8	2	1	2
Benzo(b)fluoranthene	Mg/kg	Ŷ	4	<4	<4	4	4	4	12	13

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.
• = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

		AVE001	BPD001	BPD002	JER001	MAX01	HOB001	21	IRL004	IRL005
		07/22/1998	07/17/1998	07/17/1998	07/22/1998	07/17/1998	07/22/98	98	09/03/1998	09/03/1998
								duplicate		
Polyaromatic Hydrocarbons (8310) (cont.)										
Benzo(g,h,i)perylene	µg∕kg	<6.5	<4.6	<4.3	<4.8	\$	<4.4	<4.3	=	<4.3
Benzo(k)fluoranthene	ng∕kg	^ 4	<3	<2	<3	دع	స	<2 2	10	2 1
Chrysene	µg∕kg	<3.3	<2.3	<2.2	<2.4	<2.5	<2.2	<2.2	42	6.9
Dibenzo(a,h)anthracene	μg/kg	<6.5	<4.6	<4.3	<4.8	<5	<4.4	<4.3	<4.6	<4.3
Fluoranthene	µg∕kg	<6.5	<4.6	<4.3	<4.8	\$ >	<4.4	<4.3	<4.6	<4.3
Fluorene	µg∕kg	<6.5	<4.6	<4.3	<4.8	\$ >	<4.4	<4.3	<4.6	<4.3
Indeno(1,2,3-cd)pyrene	µ9∕kg	<3.3	<2.3	<2.2	<2.4	<2.5	<2.2	<2.2	6	<2.2
Naphthalene	µg∕kg	<33	<23	<22	<24	<25	<22	<22	<23	<22
Phenanthrene	µg/kg	<3.3	<2.3	<2.2	<2.4	<2.5	<2.2	<2.2	<2.3	<2.2
Pyrene	μg/kg	<3.3	<2.3	<2.2	<2.4	<2.5	<2.2	<2.2	<2.3	<2.2
Surrogates										
2,4,5,6-TCMX	%	114	62	85	86	99	123	126	81	65
DBC	%	128	78	102	66	62	125	139	100	103
2,4-DCAA	%	98	100	58	108	104	106	82	88	124
p-terphenyi	%	89	84	93	84	104	06	49	78	73
Metals										
Aluminum	mg/Kg	1600	440	350	2000	610	999	740	260	230
Antimony	mg/Kg	<4	<3	<2	જી	<3	<3	~5	<3	42
Arsenic	mg/Kg	<1	<0.7	6.0	<0.7	2.0>	<0.6	9.0>	<0.7	<0.6
Barlum	mg/Kg	66>	<28	<26	<29	0E>	<26	<26	<27	<26
Beryllium	mg/Kg	<2	<1	-	~1	*	~	.	^1	-
Cadmium	mg/Kg	<2	<1	-1	<1	<1	<1	1 >		~1
Calcium	mg/Kg	32000	22000	14000	62000	7000	2100	1400	220000	120000
Chromium	mg/Kg	2.9	<1	1 >	3.6	-	1.4	1.6	~	~
Cobalt	mg/Kg	<10	<7	9>	<7	<i>L</i> >	9>	9>	/ >	9>
Copper	mg/Kg	<10	2 >	9>	<7	/ 2	Ŷ	9>	/>	380*
Iron	mg/Kg	1200	150	220	1700	520	340	400	250	260
Lead	mg/Kg	5.1	1.8	1.7	4.6	1	1.6	1.7	3.4	3.2
Magnesium	mg/Kg	510	620	430	380	096	110	130	550	440
Manganese	mg/Kg	27	2.4	2.1	14	7.6	1.7	2.1	4.9	5
Mercury	mg/Kg	<0.02	0.017	0.014	<0.01	0.019	<0.01	<0.02	<0.01	<0.01

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.

* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

		AVEOUT	BPD001	BPD002	JER001	MAXO1	ROBOS	2	, 90	
		07/22/1998	07/17/1008	07/47/4000	1			5	IHLU04	H_005
			_	0////1990	9881/22//0	0//1//1998	07/22/98	86	09/03/1998	09/03/1000
										000
Metale (Accet)								duplicate		
		-								
Nicke	Well Co	Ç								
Dotocolium	2	212	>	ę	' >	12	ý	97		
	E XoE	82	140	446	8		?	eş V	\	8
Selenium	Wow.	,		2	S	210	<33	35	140	9
Cilvor	2	*	ડ	ℽ	8	ç	63	Ş	,	
Cilver	E X	×4	87	۲	,		?	7	Ş	₩
Sodium			?	,	ંડ	Ş	Ÿ	8	8	9
	I MONG	320	3300	2800	86	7000	\$			y
- namum	mo/Ko	2	,			3	3	310	2200	3000
Vanadium		,	7	v		⊽	⊽	V	7	
7.	mg/Kg	3.5	⊽	⊽	5.5	•	-			Ī
ZinC	mo/Kg	13	22	2	Ş	-	V	4.	·	⊽
Other Parameters					2	/>	9>	9>	22	9
Persont Colida										
- dicalit collus	%	5.	7.2	1.	18					
Total organic carbon	1/10	20200	3,	;	ŝ	29	9/	77	73	77
	BUOM	>20/00	23000	7400	22000	16600	2200	2400	8	1000
								34.5	300	9500

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.

• = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	OCA010	H	REF001	TEL004	BRN020	KP1001	NSC001	PEF20	SLF001	SRE001	SRW001
	09/03/98	0/60	<u></u>	09/03/1998	06/01/1998	06/05/1998	06/05/1998	06/01/1998	06/05/1998	06/01/1998	06/01/1998
	duplicate	cate									
Organochlorine pesticides (8081)											
4,4' - DDD	64.4 <4	<4.5	<4.8	\$	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
4,4' - DDE	<4.4 <4	<4.5	<4.8	\$	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
4,4' - DDT	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Aldrin	× 4.4 <4	<4.5	<4.8	\$	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Alpha - BHC	<4.4 <4	<4.5	<4.8	Ş	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Beta - BHC	<4.4 <<	_	<4.8	\$	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Chlordane, alpha	<4.4	<4.5	<4.8	\$	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Chlordane, gamma	<4.4 <	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Chlordane (Total)	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Delta - BHC	<4.4	<4.5	<4.8	\$	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Dieldrin	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Endosulfan I	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Endosulfan II (beta)	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Endosulfan Sulfate	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Endrin	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Endrin Aldehyde	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Endrin Ketone	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Gamma - BHC (Lindane)	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Heptachlor	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Heptachlor Epoxide(a)	<4.4	<4.5	<4.8	5 >	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	4.4
Heptachlor Epoxide(b)	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Isodrin	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	4.4
Methoxychior	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Mirex	<4.4	<4.5	<4.8	<5	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Toxaphene	\$> 68>	06>	96>	<100	<3.8	<4.1	<4.1	<4.3	<4.3	<4.2	<4.4
Aroclors (8082)											
PCB-1016/1242	> 44	<45	<48	<50	<38	<41	<41	<43	<43	<42	<44
PCB-1221	<44	<45	<48	<50	<38	<41	<41	<43	<43	<42	44
PCB-1232	<44 ×	<45	<48	<50	<38	<41	<41	<43	<43	<42	<44
PCB-1248	<44	<45	<48	<50	<38	<41	<41	<43	<43	<42	<44
PCB-1254	> 747	<45	<48	<50	<38	<41	<41	<43	<43	<42	^44
PCB-1260	×44	<45	<48	<50	<38	<41	<41	<43	<43	<42	<44

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.

* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	OCA010	010	REF001	TEL004	BRN020	KPI001	NSCOO	DEESO	1003		
	96/20/60	86/1	09/03/1998	09/03/1998	06/01/1998	06/05/1998	06/05/1998	06/01/1998	06/05/1998	SHE001	SRW001
		duplicate							200	0001/1000	0001/1880
Chlorinated Herbicides (8151)											
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 -											
IP) (Silvex)	×14	×14	×14	<15	612	107	ç	ç	(,	
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)						5	3	77.5	222	<21	<22
	×14	<14	<14	<15	8	8	2	60	Ę	8	
2,4-Uchlorophenoxy acetic acid (2,4 - D)	<14	<14	<14	<15	83	100		118	3 8	3 8	53
3,5-DCBA	<14	<14	<14	<15	620	251	100	2 50	3 6	3 8	423
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	<14	<14	<14	<15	<19	2	5 5	200	235		\$23
4 - Mirophenol	<14	<14	<14	<15	0%	200	5	1 8	3 5	7	73
Acifluorien	<14	<14	×14	<15	<19		7	3 8	225	225	<23
Bentazon	<14	<14 <14	<14	1.45	95,		9 2	775	77.7	<21	<22
Chloramben	×14	<14	414	145	6	7 2	7	<22	<22	<21	<22
Dacthal	412	77.		2 2	SI V	5	\$	<22	<22	<21	<22
Dalapon	1		* ;	S S	<19	421	<21	<22	<22	<21	<22
Dicamba		*	414 1	<15	0Z>	421	<21	<22	<22	<22	<23
Dioplomatical to 10 4 District	4 4	<14	<14	<15	<20	<21	<21	422	222	507	200
acid	•	,									3
Dinosah	\$.	\$1\$	<14 	<15	83	-21	<21	~ 55	<22	<22	<23
MCPA	4 4 015	<14 0.25	<14	<15	<19	<21	<21	<22	<25	421	255
MCPP	0/2/0	<2/0	×290	900 *300	<190	<210	<210	<220	<220	<210	220
Pentachlorooberol	0/25	6270 ,;	<290	<300	×190	<210	<210	<220	<220	<210	220
Pictoram	410	\$15	×14	<15	82	<21	<21	<22>	422	422	233
Polysomethy Livings (0210)	4 l 4	<14 <14	<14	<15	<20	<21	<21	<22	225	422	233
Manual Information (6310)											
i meuryarapriirarene	<23	<23	<24	<25	<19	21	-23	800	733		6
z - Metnylnaphthalene	<23	<23	<24	<25	<19	12/2	100	187	3 8	y c	3 8
Acenaphthene	<23	<23	<24	<25	<19	100	160	١٤	3) 6	¥ 3	3
Acenaphthylene	4.4>	<4.5	<4.8	3	43.8	1	+	3);	3	3	422
Anthracene	<23	500	36/	30,	3			<4.3	<4.3	<4.2	<4.4
Benzo(a)anthracene	203	203		9 6	3	(2)	12	\$	<22	<22	<23
Benzo(a)pvrene	3			6,3	6.15	42.1	42.1	4.7	<2.2	5.9	<2.2
Renzo(h)filozanthene	3	3	3	3	7	<2	~	3	23	\$	Ş
	*	4 -	45	4	63	4	<4	4	4	14	4
										- T	

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.

* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	OCA010	٥	REF001	TEL004	BRN020	KP1001	NSC001	PEF20	SLF001	SRE001	SRW001
	09/03/98	8	09/03/1998	09/03/1998	06/01/1998	06/05/1998	06/05/1998	06/01/1998	06/05/1998	06/01/1998	06/01/1998
		duplicate									
Polyaromatic Hydrocarbons (8310) (cont.)											
Benzo(g,h,i)perylene	<4.5	<4.6	<4.8	<5	<3.9	<4.2	<4.2	<4.3	<4.3	<4.2	<4.4
Benzo(k)fluoranthene	¢3	3.1	<3	جع جع	2>	?>	<2	<2	7	o	63
Chrysene	5.3	3.6	<2.4	<2.5	<1.9	<2.1	<2.1	4.3	<2.2	13	<2.2
Dibenzo(a,h)anthracene	<4.5	<4.6	<4.8	<5	<3.9	<4.2	<4.2	<4.3	<4.3	<4.2	<4.4
Fluoranthene	<4.5	<4.6	<4.8	5>	6.6>	<4.2	<4.2	<4.3	9.6	29	4.51
Fluorene	<4.5	<4.6	<4.8	\$ >	<3.9	<4.2	<4.2	<4.3	<4.3	<4.2	<4.4
Indeno(1,2,3-cd)pyrene	<2.3	<2.3	<2.4	<2.5	<1.9	<2.1	<2.1	<2.2>	<2.2>	<2.1	<2.2
Naphthalene	<23	<23	<24	<25	<19	<21	<21	<22	~2 2	<21	67
Phenanthrene	<2.3	<2.3	<2.4	<2.5	<1.9	<2.1	<2.1	<2.2	<2.2	7.5	<2.2
Pyrene	<2.3	<2.3	<2.4	<2.5	<1.9	<2.1	<2.1	<2.2	<2.2	34	<2.2
Surrogates											
2,4,5,6-TCMX	69	74	89	99	100	111	94	81	127	92	93
DBC	102	105	86	104	109	101	26	69	137	109	104
2,4-DCAA	124	132	108	128	89	08	120	114	136	84	96
p-terphenyl	88	92	85	84	87	80	85	100	06	95	113
Metals											
Aluminum	1400	2200	480	210	300	2800	069	440	620	650	290
Antimony	છ	S.	63	\$3	7>	~ 5	~	?>	?>	<2	63
Arsenic	<0.7	<0.7	<0.7	5.4*	2	<0.6	21	0.91	1.4.1	11	<0.7
Barium	<27	<27	<29	<30	<23	<25	<25	<26	<26	<25	<27
Beryllium	<1	₽	~ 1	<2	-1	~1	-1	-1	۲>	<1	7
Cadmium	<1	<1	1 >	\$	~1	-	~1	7	· ·	<1	7
Calcium	2300	2900	40000	820	220000	450	120000	26000	11000	44000	1300
Chromium	3.6	5.2	۲۰	3	2	6.4	2.2	1.8	1.8	1.8	1
Cobait	<i>L</i> >	' 2	<i>L</i> >	8>	9>	9>	9>	9>	9	9>	- 2
Copper	<i>L</i> >	<i>L</i> >	<i>L</i> >	18.	9>	3	9>	9>	9>	9>	<i>L</i> >
Iron	3000	3200	550	150	540	096	920	490	440	260	410
Lead	3.2	2.9	2	5	2.9	2	3.1	2.1	۲۰	2	~
Magnesium	300	260	520	42	1500	78	260	570	110	400	63
Manganese	1.8	1.8	2.8	\$	15	~	7.1	4.7	4.2	7.2	1.6
Mercury	<0.01	0.010	<0.01	<0.02	<0.01	0.0151	<0.01	<0.01	<0.01	<0.01	<0.01

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.
• = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	OCA010	10	RFF001	TEION	PDAINOR	VDIOO	7000074				
	-	9			DINNEN		300	PEF20	S.F831	SREDO1	S COWER
	OBVOSVBB	98	09/03/1998	09/03/1998	06/01/1998	06/05/1998	06/05/1998	06/01/1008	08/08/4000		
								2001/1000		00/01/1888	06/01/1998
		duplicate									
Metals (cont.)											
NICKE	434	76>	~14	av	ą	ļ	ļ				
Potassium				9	Ş	ę	ê	9	9	9	27
	2	25	170	- 38 438	560	40	380	250	5	45	
Selenium	×14	<14	9	5	,	1		3	3	₹	ς <u>ξ</u>
Silver	,	ļ		,	y	7	<2	7	7	8	ç
	Ş	3	Ç	చ	2.4	0	?	ç	,	ľ	
Sodium	USC.	400	2000				,	ץ	۲	3	♡
T	3	B	3800	\$ \$	3000	- - - - - - -	2200	1600	130	1300	F
I Damor	⊽	⊽	⊽	5.4*	7	,	3			33	;
Vanadium	7.7	•	ļ			7	£.3	·	1.8	⊽	⊽
7:50		:	ų.	7	 5.	<u>-</u>	~	4.4	1.8	,	,
Z.II.C	<7	€0	1 >	27	ş	ų	4	ç	•	•	7
Other Parameters							2	71	0>	ę	<7
Percent Solids	74	73	99	ge	30	8					
Total organic carbon	00000	00100		3	8	8	80	11	- 12	62	75
	><0/W	>20/00	386	>56700	5	55	8	S	55	8	
							•	3			

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	SRW002	BRC09	BRN18	BRS017	BRS019	IRL003	MLN001	MLN002	MLSO3	83	PEF024
	8	06/10/1998	06/10/1998	06/10/1998	06/10/1998	06/11/1998	06/11/1998	06/11/1998	6/11/98	38	06/10/1998
										duplicate	
Organochlorine pesticides (8081)											
4,4' - DDD	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	<5	<4.7
4,4' - DDE	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	5>	<4.7
4,4' - DDT	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	<5	<4.7
Aldrin	<9.2	<5.2	<4.7	64.6	<5.2	<4.4	-6	<4.6	<5.3	<5	<4.7
Alpha - BHC	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	9.4>	<5.3	<5	<4.7
Beta - BHC	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	9.4>	<5.3	\$ >	<4.7
Chlordane, alpha	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	-<6	6.45	<5.3	\$ >	<4.7
Chlordane, gamma	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	9'4>	<5.3	<5>	<4.7
Chlordane (Total)	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	<6	<4.6	<5.3	<5	<4.7
Delta - BHC	<9.2	<5.2	<4.7	<4.6	<5.2	4.4	9>	<4.6	<5.3	\$>	<4.7
Dieldrin	<9.2	<5.2	<4.7	7.9	<5.2	<4.4	<و	<4.6	<5.3	<و	<4.7
Endosulfan I	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	ç>	<4.7
Endosulfan II (beta)	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	<5	<4.7
Endosulfan Sulfate	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	<5>	<4.7
Endrin	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	<5	<4.7
9 Endrin Aldehyde	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	<6	<4.6	<5.3	<5	<4.7
U Endrin Ketone	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	>4.6	<5.3	<5	<4.7
Gamma - BHC (Lindane)	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	- e6	<4.6	<5.3	<5	<4.7
Heptachlor	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	<5	<4.7
Heptachlor Epoxide(a)	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	<5.3	<5	<4.7
Heptachlor Epoxide(b)	<9.2	<5.2	<4.7	<4.6	<5.2	4.4>	9>	<4.6	<5.3	~	<4.7
Isodrin	<9.2	<5.2	<4.7	<4.6	<5.2	4.4>	9>	<4.6	<5.3	\$	<4.7
Methoxychlor	<9.2	<5.2	<4.7	<4.6	<5.2	4.4	9>	<4.6	<5.3	<5	<4.7
Mirex	<9.2	<5.2	<4.7	<4.6	<5.2	4.4	9>	<4.6	<5.3	\$	<4.7
Toxaphene	<9.2	<5.2	<4.7	<4.6	<5.2	4.4>	9>	<4.6	<5.3	<5	<4.7
Aroclors (8082)											
PCB-1016/1242	<92	<52	<47	<46	<52	<44	09>	<46	<53	<50	<47
PCB-1221	<92	<52	<47	<46	<52	<44	09>	<46	<53	<50	<47
PCB-1232	<92	<52	<47	<46	<52	<44	09>	<46	<53	² 50	<47
PCB-1248	<92	<52	<47	<46	<52	<44	09>	<46	<53	<50	<47
PCB-1254	<92	<52	<47	<46	<52	<44	09>	<46	<53	< 5 0	<47
PCB-1260	Z6>	<52	<47	<46	<52	<44	09>	<46	<53	<50	<47

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.

• = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	SRW002	BRC09	BRN18	BRS017	BRS019	IBL003	MI NO01	MI NOO2	MISOS	5	700100
	06/01/1998	06/10/1998	06/10/1998	06/10/1998	06/10/1998	06/11/1998	06/11/1998	06/11/1998	6/11/98	2 8 E	06/10/1998
						•				duplicate	
Chlorinated Herbicides (8151)											
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 -											
TP) (Silvex)	<46	<16	×14	414	×16	×13	44	45	4	Ť	,
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)							2	!	21/	CI V	414
	<47	<16	<14	41^	17	<13	×18	<14		7	22
2,4-Dichlorophenoxy acetic acid (2,4 - D)	<47	<16	<14	41.	19	<13	×18	×14	219	1,15	16
3,5-DCBA	<47	<16	<14	×14	<16	<13	×18	×14	746	5 7	177
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	<46	<16	<14	×14	<16	<13	<18	×14	95	215	
4 - Nitrophenol	<47	<16	<14	<14	×16	<13	√ 418	×14	416	7.5	1
Acifluorien	<46	<16	<14	<14	<16	<13	<18	41×	416	<15	7
Bentazon	<46	<16	<14	<14	<16	<13	<18	<u> </u>	912	715	7
Chloramben	<46	<16	<14	<14	<16	<13	¢18	41×	×16	7.15	1
Dacthal	<46	<16	<14	<14	<16	<13	×18	×14	2 4	34	
Dalapon	<47	<16	41.	<14 <14	<16	×13	V 18	<14	3 4	2 4	***
Dicamba	<47	<16	41×	41×	<16	<13	×18	114	31,		
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic								::		2	4+4
acid]	<47	<16	41>	×14	<16	<13	×18	412	4	7	7
Dinoseb	<46	<16	<14	<14 <14	<16	<13	×18	×14	94	34,	
MCPA	<460	<160	<140	<140	×168	<130	×180	×140	89.	150	140
МСРР	<460	<160	<140	<140	^ 160	<130	×180	×140	89	<150	140
Pentachlorophenol	<47	<16	<14	<14	<16	<13	×18	×14	912	\$ 25	×14
Pidoram	<47	<16	<14	<14	<16	<13	<18	414	<16	<15 <15	<14
Polyaromatic Hydrocarbons (8310)											
1 - Methylnaphthalene	<46	<26	<24	<24	<26	<22	<30	-24	707	36,	10.
2 - Methylnaphthalene	<46	<26	<24	<24	<26	422	630	424	123	3 8	3
Acenaphthene	~46	<26	<24	<24	<26	<22	<30	<24	727	255	204
Acenaphthylene	<9.2	<5.2	<4.7	<4.6	<5.2	<4.4	9>	<4.6	653	3	147
Anthracene	<47	427	<24	<24	72 >	<23	431	424	727	2,56	707
Benzo(a)anthracene	<4.6	<2.6	<2.4	<2.4	<2.6	<2.2	65	42.4	42.7	255	20.4
Benzo(a)pyrene	ş	ç	3.1	લ્ડ	<3	3.1	4>	63	8	8	S
Benzo(b)fluoranthene	8	<5	^	4	<5	4	\$	4>	25	1	, ,
									,		7

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.

* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

Polyagomenic hydrocentoms (8519) (Cont.) Polyagomenic hydrocento		SRW002	BRC09	BRN18	BRS017	BRS019	IRI 003	MI NO01	MI NOO2	MI SO3	83	PFF024
Polyanomatic hydrocarbone (8310) (cont.) 4.3 4.4 4.5 4.4 4.6 4.6 4.7 4.5 4.6 4.7 4.5 4.6 4.7 4.5 4.6 4.7 4.7 4.7 4.7 4.7 4.7 4.7 4.7 4.7		06/01/1998	06/10/1998		06/10/1998	06/10/1998	06/11/1998	06/11/1998	06/11/1998	6/11/	98	06/10/1998
Polygroundite Pydicearthorse (3310) [cont.) Act. Act											duplicate	
Bercolg Disponsivene 642 643 643 644 66 647 654 65 Chrystein Strong Disponsivene 643 643 643 643 644 65 645 645 645 645 645 645 645 645 645 645 645 645 645 647 645 647 645 647 645 647 645 647 645 647 645 647 645 647 645 647 648 647 648 647 668 647 646 647 648 647 653 644 66 647 654 65 647 656 647 656 647 656 647 656 647 656 647 656 647 656 647 656 657 648 647 656 647 656 647 657 648 657 648 657 648 657 648 657	Polyaromatic Hydrocarbons (8310) (cont.)											
Descrict(Introntinene	Benzo(g,h,i)perylene	<9.2	<5.3	<4.8	<4.7	<5.3	<4.4	99	<4.7	<5.4	\$	<4.8
Objective languisties of a color	Benzo(k)fluoranthene	9>	<3	6>	<3	<3	<3	<4	<3	೪	દુ	<3
Difference (a) Difference (b) Difference (c) Diff	Chrysene	<4.6	<2.6	<2.4	<2.4	<2.6	<2.2	<3	<2.4	<2.7	<2.5	<2.4
Fluctuation	Dibenzo(a,h)anthracene	<9.2	<5.3	<4.8	<4.7	<5.3	<4.4	9>	<4.7	<5.4	<5	<4.8
Fluctene C-0.2 C	Fluoranthene	<9.2	<5.3	<4.8	<4.7	<5.3	<4.4	<6	<4.7	<5.4	\$>	<4.8
Indenci(1.23-cd)pyrene <46 <2.56 <2.4 <2.4 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.6 <2.	Fluorene	<9.2	<5.3	<4.8	<4.7	<5.3	<4.4	9>	<4.7	<5.4	\$	<4.8
Najorthaliene <46	Indeno(1,2,3-cd)pyrene	<4.6	<2.6	<2.4	<2.4	<2.6	<2.2	<3	<2.4	<2.7	<2.5	<2.4
Phenanthrene 4-6 2-2 6 7.3 1 2-4 2-2 6 2-2 7 2-2 7 2-2 5 2-2 6 2-2 6 2-2 6 2-2 6 2-2 6 2-2 6 2-2 7	Naphthalene	<46	<26	<24	<24	<26	<22	<30	<24	<27	<25	<24
Surrogates 4.6 (a.) (a.) (a.) (a.) (a.) (a.) (a.) (a.)	Phenanthrene	<4.6	<2.6	7.3 1	<2.4	<2.6	<2.2	<3	<2.4	<2.7	<2.5	42.4
Surrogates Surrogates 6 6 6 6 6 6 6 7 7 24.56-TCMX 100 84 67 67 68 66 69 78 55 71 24.56-TCMX 110 182 67 68 126 69 170 178 170	Pyrene	<4.6	<2.6	<2.4	<2.4	<2.6	<2.2	<3	<2.4	<2.7	<2.5	<2.4
2.4.5.6-TCMX 100 84 67 67 68 66 69 78 55 71 DBC DBC 117 122 88 126 99 103 121 128 96 107 D-BC 2.4-DCAA 90 118 40 96 136 121 128 96 107 D-Ierptieny Metals 83 82 78 82 82 82 88 96 107 D-Ierptieny Metals 83 82 78 78 87 87 88 89 88 88	Surrogates											
DBC 117 122 88 126 103 121 128 96 103 121 128 96 103 121 128 96 107 24-DCAA 90 118 40 96 136 40 82 72 118 98 Petrphenyl Matals 83 82 78 82 87 118 98 100 Aluminum Matals 83 82 78 82 87 89 88 88 88 88 88 89 108 89 118 96 118 96 118 96 118 80 88 88 88 88 88 88 88 83	2,4,5,6-TCMX	100	84	- 67	- 67	89	99	69	78	55	7.1	69
2.4-DCAA 90 118 40 96 136 40 62 72 118 98 P-lerphenyl Metals 83 82 78 82 82 87 80 85 180 85 82 87 80 85 88 82 88 82 82 82 87 80 85 82 87 80 85 82 87 80 85 82 87 80 85 82 8	DBC	117	122	88	126	66	103	121	128	96	107	96
Prietryhenyly Metals 85 82 78 82 87 89 89 88 Aluminum Metals 300 450 1000 830 560 1200 280 1600 4500* 1400 1800 Aluminum 630 1000 830 560 1200 280 1600 4500* 1400 1800 Aluminum 66 63 63 62 63 63 63 63 63 63 Arsenic 66 63 63 62 63 62 63 62 63	_	06	118	40	96	136	40	82	72	118	86	96
Metals Signor 1000 830 560 1200 1600 4500* 1400 1800 Antimony <6 <3 <3 <3 <3 <4 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3 <3		95	83	82	78	82	82	87	80	85	88	91
m 630° 430 560 1200 280 1600 450° 1400 1800 y -6 -3 -4 -3 -3 -3 -3 -3 -3 -4 -3 -4 -3 -3 -3 -3 -3 -3 -3 -3 -3 -3 -3 -3 -3 -4 -2 -4 -2 -4 -2 -4 -2 -4 -2 -4												
yy c6 c3 c4 c3 c4 c2 c4<	Aluminum	5300*	1000	830	260	1200	280	1600	4500*	1400	1800	410
13.31 1.11 0.71 0.71 1.11 0.71 1.11 0.71 1.11 0.71 1.11 0.71 1.11 0.72 0.29 0.27 0.29 0.27 0.29 0.27 0.27 0.29 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.29 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 0.27 0.29 <t< th=""><th>Antimony</th><th>9></th><th><3</th><th>6></th><th><3</th><th><3</th><th><3</th><th><4</th><th><3</th><th><3</th><th>دع</th><th>63</th></t<>	Antimony	9>	<3	6>	<3	<3	<3	<4	<3	<3	دع	63
n <56	Arsenic	3.31	1.1 1	0.7 1	0.7 1	1.1 i	<0.7	<0.9	4.0>	<0.8	1 8.0	<0.7
III <2	Barium	>56	<32	<28	<28	<32	<27	<36	<28	<32	<30	<28
m <3	Beryllium	<3	₹	<1	√1	<2	٧٠	<2	1>	<2	~	<1
Imm 160000 43000 12000 58000 100000 3100 6400 120000 21000 23000 23000 Imm 13° 3 3.4 1.8 3.8 <1 4.7 11° 4.5 5.4 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 6 7 6 7 6 7 6 7 6 7 6 7 6 7 6	Cadmium	<3	<2	-	<1	<2	~ 1	<2	<۱	\$	<i>7</i> >	٧
Imm 13* 3.4 1.8 3.8 <1	Calcium	160000	43000	12000	58000	100000	3100	6400	120000	21000	23000	11000
414 68 77 68 77 68 77 68 78 68 78 68 78 68 78 68 78 68 78 68 78 68 78 68 78 68 78 68 78 68 78 68 78 68 78 68 78	Chromium	13*	E	3.4	1.8	3.8	<1	4.7	-11	4.5	5.4	1.8
414 68 7 68 7 69 7 68 68 67 69 7 68 68 68 68 7 68 7 68 7 68 7 68 7 68 7 68 7 68 7 1800 2100 </th <th>Cobalt</th> <th><14</th> <th>8></th> <th><i>L</i>></th> <th><7</th> <th><8</th> <th><i>L</i>></th> <th>6></th> <th><i>L</i>></th> <th>8</th> <th>8></th> <th>1></th>	Cobalt	<14	8>	<i>L</i> >	<7	<8	<i>L</i> >	6>	<i>L</i> >	8	8>	1 >
660° 1000 1100 630 1200 270 2000 540° 1800 2100 2100 ium 4.7 3.5 2 2.5 6 <1 <2 2.2 <2 1.8 iese 1600 1500 1200 930 2100 690 3100 6500 2600 3200 iese 110 10 12 12 19 2.5 16 55 16 20 iese <0.03 <0.03 <0.01 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02	Copper	<14	8>	<i>L</i> >	<7	<8	<i>L</i> >	6>	<i>L</i> >	<8	8>	<7
4.7 3.5 2 2.5 6 <1	Iron	2600*	1000	1100	630	1200	270	2000	2400	1800	2100	490
1600 1600 1200 930 2100 690 3100 6500 2600 3200 3200 110 10 12 12 19 2.5 16 55 16 20 <0.03 <0.03 <0.01 <0.01 <0.01 <0.01 <0.01 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.03 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.03 <0.02 <0.03 <0.03 <0.02 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0	Lead	4.7	3.5	2	2.5	9	~ 1	<2	2.2	<2	1.8	2.1
110 10 12 19 2.5 16 55 16 20 <0.03 <0.02 <0.01 <0.02 <0.01 <0.02 <0.01 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.02 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03 <0.03	Magnesium	1600	1600	1200	930	2100	069	3100	0059	2600	3200	069
<0.03	Manganese	110	10	12	12	19	2.5	16	22	16	20	4.7
	Mercury	<0.03	<0.02	<0.01	<0.01	<0.02	<0.01	<0.02	<0.01	<0.02	<0.02	<0.01

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.

• = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	SRW002	BRC09	BRN18	BRS017	BRS019	IRLOO3	MI NOO1	MI NOO2	MI COS	5	7001
	06/01/1998	06/01/1998 06/10/1998	06/10/1000	06/40/4000						?	FEF-024
	300	0001/01/00	0661/01/00	00/10/1998	00/10/1998	06/11/1998		06/11/1998 06/11/1998	6/11/98	æ	06/10/199R
Marie A										duplicate	
Metals (cont.)											
Nickel	714	٩	1,	,							
Dotessium		9	V	v	8	<7	6 8	'	\$	89	<7
- Crassini	360	570	310	370	790	270	1200	2100	220	010	240
Selenium	9	65	స	63	63	٢	17	Ç		2 6	240
Silver	4	,	ľ	,		?	,	3	Ş	\$	<u>چ</u>
	2	Ş	Ş	Ş	છ	Q	4	જ	\$3	8	5
Sodium	470	3300	2300	2800	4800	3300	12000	200	2000	1	
Thallium	5	۶	-		,		3	3	3300	2300	2000
//ene-di.m-		y	7	V	\$	<1	7	⊽	7	7	V
Validation	11"	2.8	5.6	1.7	3.5	1>	4.4	12.	4.4	2 2	\\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\
Zinc	<14	<25	623	<25 25 25 25 25 25 25 25 25 25 25 25 25 2	225	1	٩	: :		3,	v
Other Parameters						,	ŝ	2	9>	ς,	423
Percent Solids	36	63	02	71	63	7.5	55	ļ			
Total organic carbon	170	1000	2002	0000		2	3	-	3	8	2
	2	386	M/c	30000	34000	4000	14000	38000	0006	16000	8000
											2

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.

• = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	SIBOS	COMMIS	100	200 101	0000						
	06/10/1998	06/11/1998	06/17/1908	06/17/1000	543-D1	_ 9	SH3-D2	BCE012	BCE004	BCW15	5
	000		000 1/1 1000	9861//100	06/1/98	20	06/17/1998	06/24/1998	06/24/1998	6/25/98	88
						duplicate					duplicate
Organochlorine pesticides (8081)											
4,4' - DDD	<4.9	<5.1	<4.5	<5.5	<4.4	<4.4	<4.8	<4.8	75.2	14.3	14.0
4,4' - DDE	<4.9	<5.1	<4.5	<5.5	<4.4	<4.4	<4.8	×4.8	5.5	2 4	24.5
4,4' - DDT	<4.9	<5.1	<4.5	<5.5	<4.4	<4.4	<4.8	×4.8	25.2 25.2	2 6	5.4.5
Aldrin	<4.9	<5.1	<4.5	<5.5	4.4	<4.4	<4.8	44.8	1,50	2	2
Alpha - BHC	<4.9	<5.1	<4.5	<5.5	<4.4	<4.4	<4.8	64.8	4,50	2 5	4.3
Beta - BHC	<4.9	<5.1	<4.5	<5.5	<4.4	×4.4	×4.8	<4.8	3,5,0	2 5	5.45
- Chlordane, alpha	<4.9	<5.1	<4.5	<5.5	<4.4	<4.4	<4.8	<4.8	1,50		44.5
Chlordane, gamma	<4.9	<5.1	<4.5	<5.5	×4.4	<4.4	×4.8	<4.8	55.2	2 5	54.5
Chlordane (Total)	<4.9	<5.1	<4.5	<5.5	4.4	<4.4	<4.8	<4.8	<5.2 25.2	643	2 7
Defta - BHC	<4.9	<5.1	<4.5	<5.5	<4.4	4.1	<4.8	<4.8	1,4,0	5	7.40
Diekdrin	<4.9	<5.1	<4.5	<5.5	<4.4	<4.1	<4.8	8 4 9	25.44	2 6	54.5
Endosulfan I	<4.9	<5.1	<4.5	<5.5	<4.4	<4.1	<4.8	64 B	1,50	2 5	2.5
Endosulfan II (beta)	<4.9	<5.1	<4.5	<5.5	<4.4	<4.1	×4.8	8 4 9	25.5	2 5	64.0
Endosulfan Sulfate	<4.9	<5.1	<4.5	<5.5	<4.4	4.1	×4.8	648	35.5		44.3
Endrin	<4.9	<5.1	<4.5	<5.5	<4.4	4.1	<4.8	<4.8	1,5,0	2 5	24.5
Endrin Aldehyde	<4.9	<5.1	<4.5	<5.5	<4.4	4.	<4.8	×4.8	55.2	2 5	24.5
Endrin Ketone	<4.9	<5.1	<4.5	<5.5	<4.4	<4.1	<4.8	<4.8	<52	243	2 7
Gamma - BHC (Lindane)	<4.9	<5.1	<4.5	<5.5	<4.4	4.1	<4.8	<4.8	<5.2	<4.3	×4.3
Heptachlor	<4.9	<5.1	<4.5	<5.5	<4.4	¢4.1	<4.8	<4.8	<5.2	<4.3	<4.3
Heptachior Epoxide(a)	<4.9	<5.1	<4.5	<5.5	<4.4	<4.1	<4.8	<4.8	<5.2	<4.3	< 43
Heptachlor Epoxide(b)	<4.9	<5.1	<4.5	<5.5	<4.4	<4.1	<4.8	<4.8	<5.2	<4.3	¢4.3
ISOOFIN	<4.9	5 .1	<4.5	<5.5	<4.4	<4.1	<4.8	<4.8	<5.2	<4.3	<4.3
Meuroxycrio	<4.9	5 .1	<4.5	<5.5	<4.4	<4.1	<4.8	<4.8	<5.2	<4.3	<4.3
MIFEX	<4.9	5 .1	<4.5	<5.5	<4.4	<4.1	<4.8	<4.8	<5.2	<4.3	<4.3
i oxaprierie	<4.9	<5.1	<4.5	<5.5	<4.4	<4.1	<4.8	<4.8	<5.2	<4.3	<4.3
Aroclors (8082)					_						
PCB-1016/1242	<49	<51	<45	<55	^44	145	<48	<48	c52	43	243
PCB-1221	<49	<51	<45	<55	×44	14	<48	×48	265	2 E P	243
PCB-1232	<49	<51	<45	<55	×44	<41	<48	<48	1 54	3	
PCB-1248	<49	-55	<45	<55	<44	<41	847	48	3 5	3	2
PCB-1254	<49	-55	<45	<55	4	<44 44	×48	48	25,65	\$ \{	\$ 6
PCB-1260	¢49	<51	<45	655	744	1	94.		3,5	\$45	<43
			,	7	F	•	0#5	×48	<52	<43	<43

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

^{* =} value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	3000	SUM LOS	R 601	- TE 000	SB2.D1	7	500	0.0100			
	06/10/1998	06/11/1998	06/17/1998	06/17/1998	06/17/98	98	06/17/1998	06/24/1998	BCE004 06/24/1998	BCW15 6/25/98	യയ
						duplicate					
Chlorinated Herbicides (8151)											duplicate
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	<15	715	417	7.							
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)					414	Z12	~15	<	×16	<13	<13
	<15	<15	<14	<17	<14 41	<12	<15	7,	4	Ç	ų,
2,4-Dichlorophenoxy acetic acid (2,4 - D)	<15	<15	<14	<17	<14	<14	<15	<15	2 4	2 5	\$ \$ \$
JUBA	<15	<15	<14	<17	<14	<14	<15	<15	2 4	2 5	5 5
4-(2,4-Uraniorophenoxy)butyric acid (2,4 - DB)	<15	<15	<14	<17	<14	<14	<15	<15 <15	×16	3 5	2 5
A - Annoparenta	<15	<15	<14	<17	<14	<14	<15	<15	416	2 5	5 5
Podraco	<15	<15	<14	<17	<14	<14	<15	<15 <15	216	2 5	2 5
azul	<15	<15	<14	<17	<14	<14	<15	×15	218	3	515
Cindamoeri	<15	<15	<14	<17	<14	<14	<15	<15	216	2 5	2 5
Dacuel	<15	<15	<14	<17	×14	<12	<15	<15	146	218	5 5
Uatapon	<15	<15	<14	<17	<14	<12	<15 15	15	2 4	2 8	2 5
Dicamba	<15	<15	<14	<17	14	414	14			2	× 13
Dichloroprop (2-(2,4-Dichlorophenoxy)proponic							2	CI >	9	<13	<13
	<15	<15	41>	<17	^ 4!^	<12	7.	<u>u</u>	4	Ş	,
Dinoseb	<15	<15	×14	<17	×14	252	2 45	2 4	0 0		<u>\$</u>
MCPA	<150	<150	<140	<170	<140	22	5.5	150	200	2 5	\$13
МСРР	<150	<150	<140	<170	<140	7140	3 5	3 5	3 5	35	<130
Pentachlorophenoi	<15	<15	<14	<17	<14	54,	3 4	200	2015	2138 238 238 238 238 238 238 238 238 238 2	<u>ب</u>
Picloram	<15	<15	<14	<17	-14	100		CI V	۹۱۶	<13	<13
Polyaromatic Hydrocarbons (8310)						71.	612	<15	<16	< 1 3	<13
- Methylnaphthalene	<25	- Se	523	867	55	1					
2 - Methyinaphthalene	<25	~26	233	200	3 6	ÿ	9	S	929	<22	<22
Acenaphthene	<25	426	623	867	3 8	y 8	3	c;	\$25	<22	<22
Acenaphthylene	×4.9	-52	145			3	Ş	623	<26	<22	<22
Anthracene	26/	ac	2 5	3 8	7.00	<4.4	<4.8	<4.8	<5.2	<4.3	<4.3
Benzo(alanthracene	3,5	96	3 3	87	\$23	423	<25	<25	<27	<22	<22
Benzo(a)ovrene	6.3	Q.Y	623	<2.8	<2.3	<2.3	<2.5	<2.5	<2.6	42.2	42.2
Benzothiffioranthooa	3	3	3	ę	ς,	<3	<3	63	ప	2	83
ole/moralisment	4	Ŷ	*	4	7	•	,				

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

^{* =} value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	SJB02	SJML02	IRL001	IRL002	SR3-D1	2	SR3-D2	BCE012	BCE004	BCW15	5
	06/10/1998	06/11/1998	06/17/1998	06/17/1998	06/17/98	88	06/17/1998	06/24/1998	06/24/1998	6/25/98	8
						duplicate					duplicate
Polyaromatic Hydrocarbons (8310) (cont.)											
Benzo(g,h,i)perylene	\$	<5.1	<4.6	<5.6	<4.5	<4.5	<4.9	<4.9	<5.3	<4.3	<4.3
Benzo(k)fluoranthene	<3	જ	જ	65	ę	బ	53	63	\$3	3	8
Chrysene	<2.5	<2.6	<2.3	<2.8	<2.3	<2.3	<2.5	<2.5	<2.6	<2.2	<2.2
Dibenzo(a,h)anthracene	<5	<5.1	<4.6	>5.6	<4.5	<4.5	<4.9	<4.9	<5.3	<4.3	<4.3
Fluoranthene	<5	<5.1	<4.6	9:5>	<4.5	<4.2	<4.9	<4.9	<5.3	<4.3	<4.3
Fluorene	<5	<5.1	<4.6	9:5>	<4.5	<4.2	<4.9	<4.9	<5.3	<4.3	<4.3
Indeno(1,2,3-cd)pyrene	<2.5	<2.6	<2.3	<2.8	<2.3	<2.1	<2.5	<2.5	<2.6	<2.2	<2.2
Naphthalene	<25	<26	<23	<28	54	24 !	47	<25	<26	422	<22
Phenanthrene	<2.5	10	<2.3	<2.8	<2.3	<2.1	<2.5	<2.5	<2.6	<2.2	<2.2
Pyrene	<2.5	15	<2.3	<2.8	<2.3	<2.1	<2.5	<2.5	<2.6	<2.2	<2.2
Surrogates											
2,4,5,6-TCMX	69	89	71	58	72	62	75	55	65	75	99
DBC	105	107	130	119	130	119	150	107	118	110	120
2,4-DCAA	112	96	72	88	80	8	80	120	96	04	104
p-terphenyl	82	83	89	68	92	82	98	72	101	82	76
Metais											
Aluminum	1100	2300	1300	1400	640	720	200	1300	1900	750	700
Antimony	<3	<ع	<3	<3	<3	છ	<3	<3	63	3	ę
Arsenic	1.3 1	1 6:0	<0.7	8.0>	<0.7	<0.7	0.0	<0.7	<0.8	2.3	1.8
Barlum	<30	<31	<27	<33	<27	<27	<29	<29	<32	~ 26	426
Beryflium	7	<2	<1	~	~ 1	₽	<1	<1	₹	۲-	₽
Cadmium	٨	8	₹	2	<u>.</u>	-	<1	1>	~	Þ	۲
Calcium	96000	<38	970	32000	16000	24000	180000	3700	4100	140000	130000
Chromium	3.7	9	2.6	2.7	1.9	2	<1	8	4.4	1.8	1.7
Cobalt	<7	8	<7	8 >	<7	2 2	<7	L >	8>	9>	9>
Copper	<7	8>	1 >	8>	<i>L</i> >	2 >	<7	1>	<2	8.	1.6
Iron	1400	2800	480	1100	280	290	200	1200	1700	069	640
Lead	4.8	2.3	<1	1.8	<1	۲	1.9	2.4	1.9	2.7	2.5
Magnesium	1500	3800	810	1800	880	089	280	1400	1900	540	570
Manganese	21	32	5.1	60	9	3	3.5	7	10	4.9	4.6
Mercury	40.01	<0.02	<0.01	<0.02	<0.01	<0.01	<0.01	<0.01	<0.02	<0.01	<0.01
											ł

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	SJB02	SJML02	1BL001	IRL002	SR3-D1	7	SR3-D2	BCE012	BCE004	BCW15	3
	06/10/1998	06/10/1998 06/11/1998	06/17/1998	06/17/1998	06/17/98	86	06/17/1998	0	06/24/1998	6/25/98	· &
						duplicate					diminate
Metals (cont.)											Commence
Nickel	-77	٩	7.	ļ		,					
	,	?	7	8	</td <td>Ŷ</td> <td>/></td> <td>/></td> <td>æ</td> <td>9</td> <td>φ</td>	Ŷ	/>	/ >	æ	9	φ
Potassium	900	1100	450	820	400	350	140	200	570	440	ARO
Selenium	6	\$	છ	ట	63	<2 2	63	8	5	ç	3
Silver	53	8	5	5	,	,				,	3
			?	3	3	7	5.4	\$	3	2.7	က
Sodium	3400	5700	3300	6700	4300	2800	2200	3100	2700	3500	3600
Thallium	7	8	⊽	3	1 ∨	٧	V	7	12	7	3
Vanadium	3.3	9	3	3.5	1.9	1.9	V	e e	. 7		7
Zinc	<24	89	<i>L</i> >	80	7	99	1/2	2 0	q	£3	£.1
Other Parameters									9	8	e V
Percent Solids	29	65	73	09	7.4	æ	89	89	63	7.7	1
Total organic carbon	30000	18000	11000	22000	22000	28700	26700	3	3)
				2003	2000	30.55	20/02	2400	36.7	2010	>26700

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

^{* =} value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	BCW014	HPC001	201	MOC021
	06/25/1998	06/24/98	86/	06/25/1998
			duplicate	
Organochlorine pesticides (8081)				
4,4' - DDD	<4.7	<5.1	<4.8	<4.7
4,4' - DDE	<4.7	<5.1	<4.8	<4.7
4,4' - DDT	<4.7	<5.1	<4.8	<4.7
Aldrin	<4.7	<5.1	<4.8	<4.7
Alpha - BHC	<4.7	<5.1	<4.8	<4.7
Beta - BHC	<4.7	<5.1	<4.8	<4.7
Chlordane, alpha	<4.7	<5.1	<4.8	<4.7
Chlordane, gamma	<4.7	<5.1	<4.8	<4.7
Chlordane (Total)	<4.7	<5.1	<4.8	<4.7
Delta - BHC	<4.7	5 .1	<4.8	<4.7
Dieldrin	<4.7	<5.1	<4.8	<4.7
Endosulfan I	<4.7	<5.1	<4.8	<4.7
Endosulfan II (beta)	<4.7	<5.1	<4.8	<4.7
Endosulfan Sulfate	<4.7	<5.1	<4.8	<4.7
Endrin	<4.7	<5.1	<4.8	<4.7
Endrin Aldehyde	<4.7	<5.1	<4.8	<4.7
Endrin Ketone	<4.7	<5.1	<4.8	<4.7
Gamma - BHC (Lindane)	<4.7	<5.1	<4.8	<4.7
Heptachlor	<4.7	<5.1	<4.8	<4.7
Heptachlor Epoxide(a)	<4.7	<5.1	<4.8	<4.7
Heptachlor Epoxide(b)	<4.7	<5.1	<4.8	<4.7
Isodrin	<4.7	<5.1	<4.8	<4.7
Methoxychlor	<4.7	<5.1	<4.8	<4.7
Mirex	<4.7	<5.1	<4.8	<4.7
Toxaphene	<4.7	<5.1	<4.8	<4.7
Aroclors (8082)				
PCB-1016/1242	<47	<51	<48	<47
PCB-1221	<47	<51	<48	<47
PCB-1232	<47	<51	<48	<47
PCB-1248	<47	<51	<48	<47
PCB-1254	<47	<51	<48	<47
PCB-1260	<47	<51	<48	<47

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

^{* =} value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	BCW014	HPC001	5	MOC021
	06/25/1998	06/24/98	/98	06/25/1998
			duplicate	
Chlorinated Herbicides (8151)				
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 -				
TP) (Silvex)	×14	~15	~14	×14
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)				
	<14	<15	41×	×14
2,4-Dichlorophenoxy acetic acid (2,4 - D)	<14	<15	<14	<14 414
3,5-DCBA	<14	<15	<14	×14
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	<14	<15	<14	×14
4 - Nitrophenol	<14	<15	<14	×14
Acifluorfen	<14	<15	<14	<14
Bentazon	×14	<15	<14	<14
Chloramben	<14	<15	<14	<14
Dacthal	<14	<15	<14	<14
Dalapon	<14	<15	<14	<14
Dicamba	<14	<15	<14	<14
Dichloroprop [2-(2,4-Dichlorophenoxy)proponic				
acid]	<14	<15	<14	<14
Dinoseb	<14	<15	<14	<14
MCPA	<140	<150	<140	<140
MCPP	<140	<150	<140	<140
Pentachlorophenol	<14	<15	<14	<14
Picloram	<14	<15	<14	<14
Polyaromatic Hydrocarbons (8310)				
1 - Methylnaphthalene	<24	<26	424	<24
2 - Methylnaphthalene	<24	426	<24	<24
Acenaphthene	<24	₹	<24	<24
Acenaphthylene	<4.7	<5.1	<4.8	<4.7
Anthracene	<24	<26	<25	<24
Benzo(a)anthracene	<2.4	<2.6	<2.4	<2.4
Benzo(a)pyrene	<3	<3	ج3	\$
Benzo(b)fluoranthene	42	<5	<4	77

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	BCW014	HPC001	5	MOC021
	06/25/1998	06/24/98	98	06/25/1998
			duplicate	
Polyaromatic Hydrocarbons (8310) (cont.)				
Benzo(g,h,i)perylene	<4.8	<5.1	<4.8	<4.8
Benzo(k)fluoranthene	63	ŝ	છ	63
Chrysene	<2.4	<2.6	<2.4	<2.4
Dibenzo(a,h)anthracene	<4.8	<5.1	<4.8	<4.8
Fluoranthene	<4.8	<5.1	<4.8	<4.8
Fluorene	<4.8	5.1	<4.8	<4.8
Indeno(1,2,3-cd)pyrene	<2.4	<2.6	<2.4	<2.4
Naphthalene	<24	<26	<24	<24
Phenanthrene	<2.4	<2.6	<2.4	<2.4
Pyrene	<2.4	<2.6	<2.4	<2.4
Surrogates				
2,4,5,6-TCMX	96	62	99	09
DBC	601	132	110	110
2,4-DCAA	80	72	35	72
p-terphenyl	67	82	88	79
Metals				
Aluminum	880	1500	1300	340
Antimony	<3	<3	છ	63
Arsenic	<0.7	2	2.3	<0.7
Barium	<28	<31	~ 29	<28
Beryllium	1 >	4	۲>	<1
Cadmium	1>	<2	<1	۲
Calcium	4000	92000	160000	3000
Chromium	2	3.2	2.6	⊽
Cobalt	<i>t></i>	<8	<i>L</i> >	/>
Copper	<1	3.4	3	<1
Iron	800	1500	1200	310
Lead	<1	2.9	3	2.1
Magnesium	930	1300	1000	510
Manganese	4.8	6	8.1	4.7
Mercury	<0.01	<0.02	<0.01	<0.01

I = value is between the Method Detection Limit and the Practical

Quantitation Limit.
* = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Table F-6. KSC Background raw data for sediment locations.

	BCW014	HPC001	5	MOC021
	06/25/1998	06/24/98	96/	06/25/1998
			duplicate	
Metals (cont.)				
Nickel	<i>L</i> >	<8	<i>L></i>	<i>L</i> >
Potassium	370	989	650	230
Selenium	\$	63	દર	63
Silver	çş	গু	9	ęş
Sodium	3100	2900	3300	3100
Thallium	^	\$	۲	⊽
Vanadium	1.8	2.8	2.3	1.6
Zinc	<7	8.3	L >	<i>L</i> >
Other Parameters				
Percent Solids	70	65	69	02
Total organic carbon	11900	20000	>26700	>26700

I = value is between the Method Detection Limit and the Practical Quantitation Limit.

Quantitation Limit.

• = value is an outlier and not used in the analysis to deveop the KSC Background screening values.

Appendix G

John F. Kennedy Space Center Baseline Chemical Characterization Database Evaluation and Validation

Prepared by Dynamac Corporation

Prepared for the NASA KSC Remediation Program Office

TABLE OF CONTENTS

SECTION	Page
Table of Contents	514
Abstract	515
List of Tables List of Figures	516 517
1.0 Introduction 1.1 Data Quality Objectives 1.2 Laboratory Qualifications	519 519 520
2.0 Measurement System 2.1 Field Sampling 2.1.1 Sample Site Selection 2.1.2 Sample Sizes 2.1.3 Sample Collection Efficiency 2.1.4 Equipment Blanks	520 520 520 520 521 522
2.2 Laboratory Performance 2.2.1 Holding Times 2.2.2 Method Blanks 2.2.3 Surrogate Recovery 2.2.4 Matrix Spikes (duplicate precision) 2.2.5 Spike Recoveries	522 522 522 523 524 524
2.3 Reporting 2.3.1 Chain of Custody 2.3.2 Narratives 2.3.3 Electronic Data transfer	525 525 525 526
3.0 Summary	526
4.0 References	529

Abstract

This appendix provides a summary of the KSC Baseline Chemical Characterization Database. Emphasis is placed on sampling and analytical procedures, data collection efficiency, chain of custody record keeping, and overall data quality and validity. The project specific database includes 42073 data points out of an expected 42076. This represents a 99.99% data collection efficiency. All field methods utilized during sample collection followed procedures described in the contractor laboratory Florida DEP approved Quality Assurance Plan. In addition, groundwater sampling was conducted in compliance with NASA/KSC Sampling and Analysis Plan for RCRA Facility Investigations. All samples were analyzed utilizing appropriate EPA approved methods and all sample holding times were met. Chain of custody records were maintained throughout the project. Date entry errors were found on four Chain of Custody sheets, but no unusual data were reported for the sampling events in question. Signed laboratory reports were provided for each analytical event. Laboratory narratives were provided with laboratory reports. Results of equipment blank analyses indicate no significant procedural contamination of field samples. One equipment blank was reported to have organic contamination however all organic results for the field samples in question were reported as below detection. Results for method blank analyses indicated possible metal contamination in five sample runs. Low level sodium and calcium contamination did not meaningfully impact estimates of mean values. Aluminum contamination in the laboratory process may have produced a positive 4.25% bias in the mean concentration estimate for the coastal soil class.

All laboratory blanks for the EPA 8081, 8082, 8151, and 8310 organic analyses were reported as below detection indicating no laboratory process contamination of organic samples. Assessment of surrogate and spike recovery data show the Laboratory had good quality control. Metal analyses were high quality with few problems noted. Matrix interferences were an issue in several coastal soil samples, which are known to be high in sodium, magnesium, calcium, chloride, and aluminum. Organic analyses displayed a greater degree of variability than metals, as expected with the more complex procedures and methodologies. Organic analyses results are in general of good quality. Sample collection and processing induced errors have little influence on estimates of mean concentrations in the different media or the calculations of screening levels.

List of Tables

		Page
Table G-1.	Summary of surface water and groundwater parameter collection efficiency for the KSC Baseline Chemical Characterization. The expected numbers include duplicate samples where appropriate.	529
Table G-2.	Summary of soil and sediment parameter collection efficiency for the KSC Baseline Chemical Characterization. The expected numbers include duplicate samples where appropriate.	532
Table G-3.	Results of equipment blank analysis. Values presented are for all parameters reported above detection limits. All other parameters analyses results reported as below detection.	535
Table G-4.	Summary of laboratory test data for blanks, surrogate recoveries, and relative percent differences of matrix spike duplicates.	536
Table G-5.	Summary laboratory spike recovery data expressed as a percent. Recoveries ranked as Excellent = all values within control limits, Good > 90% within control limits, Fair > 80% within control limits and Poor < 80% within control limits.	547
Table G-6.	Summary of chain of custody records and sample holding time information.	549

List of Figures

		Page
Figure G.1.	Spike recovery frequency histogram for 2,4,-DB.	551
Figure G.2.	Spike recovery frequency histogram for 2,4,-DCAA	551
Figure G.3.	Spike recovery frequency histogram for 2,4,5,6-TCMX.	552
Figure G.4.	Spike recovery frequency histogram for 2,4-D.	552
Figure G.5.	Spike recovery frequency histogram for 4,4'-DDT.	553
Figure G.6.	Spike recovery frequency histogram for 2,4,5-TP (Silvex).	553
Figure G.7.	Spike recovery frequency histogram for Acenaphthene.	554
Figure G.8.	Spike recovery frequency histogram for Aldrin.	554
Figure G.9.	Spike recovery frequency histogram for Aluminum.	555
Figure G.10.	Spike recovery frequency histogram for Antimony.	555
Figure G.11.	Spike recovery frequency histogram for Arsenic.	556
Figure G.12.	Spike recovery frequency histogram for Barium.	556
Figure G.13.	Spike recovery frequency histogram for Benzo(g,h,i)perylen	e. 557
Figure G.14.	Spike recovery frequency histogram for Benzo(a)pyrene.	557
Figure G.15.	Spike recovery frequency histogram for Berylium.	558
Figure G.16.	Spike recovery frequency histogram for Cadmium	558
Figure G.17.	Spike recovery frequency histogram for Calcium.	559
Figure G.18.	Spike recovery frequency histogram for Chromium.	559
Figure G.19.	Spike recovery frequency histogram for Cobalt.	560
Figure G.20.	Spike recovery frequency histogram for Copper.	560
Figure G.21.	Spike recovery frequency histogram for Dalapon.	561
Figure G.22.	Spike recovery frequency histogram for DBC.	561

		Page
Figure G.23.	Spike recovery frequency histogram for Dicmba.	562
Figure G.24.	Spike recovery frequency histogram for Endrin.	562
Figure G.25.	Spike recovery frequency histogram for gamma-BHC.	563
Figure G.26.	Spike recovery frequency histogram for Heptachlor.	563
Figure G.27.	Spike recovery frequency histogram for Iron.	564
Figure G.28.	Spike recovery frequency histogram for 2,4,5,6-TCMX.	564
Figure G.29.	Spike recovery frequency histogram for Magnesium.	565
Figure G.30.	Spike recovery frequency histogram for Manganese.	565
Figure G.31.	Spike recovery frequency histogram for Mercury.	566
Figure G.32.	Spike recovery frequency histogram for Napthalene.	566
Figure G.33.	Spike recovery frequency histogram for Nickel.	567
Figure G.34.	Spike recovery frequency histogram for PCB-1221.	567
Figure G.35.	Spike recovery frequency histogram for Potassium.	568
Figure G.36.	Spike recovery frequency histogram for p-Terphenyl.	568
Figure G.37.	Spike recovery frequency histogram for Selenium.	569
Figure G.38.	Spike recovery frequency histogram for Silver.	569
Figure G.39.	Spike recovery frequency histogram for Sodium.	570
Figure G.40.	Spike recovery frequency histogram for Thallium.	570
Figure G.41.	Spike recovery frequency histogram for Total Organic Carbon	571
Figure G.42.	Spike recovery frequency histogram for Vanadium.	571
Figure G.43.	Spike recovery frequency histogram for Zinc.	572
Figure G.44.	Spike recovery frequency histogram for Metals.	572

1.0 Introduction

The Kennedy Space Center Environmental Program Office (EPO) conducted a Baseline Chemical Characterization to define background concentrations of selected chemicals in the environment at the space center. The objective of the project was to create a data set describing ambient concentrations of metals, pesticides, herbicides and polycyclic aromatic hydrocarbons (PAH) in soils, sediments, surface waters and ground waters. These chemical data can be used in the decision-making and project development process for remediation activities, contamination assessments, ecological risk determinations, ecological research, and other regulatory and NEPA activities. This document provides a review and assessment of field and laboratory data generation processes. The assessment evaluates the extent to which the data set satisfies the overall project objective of defining baseline conditions in the four media.

1.1 Data Quality Objectives

Definition of data quality objectives is recognized as the single most important aspect of any data collection activity (Keith, 1991). Without clear data quality objectives and good planning, data may be unusable no matter how good the field crews or laboratories perform. The goal of this project was to create a statistically valid data set, describing as realistically as possible within cost considerations, the baseline or ambient chemical concentrations of metals, pesticides, herbicides and PAH in soil, sediment, surface water and groundwater of KSC. Data quality objectives established prior to project initiation included:

- Environmental samples must be collected in locations free from past KSC operational activities in a fashion (stratified) that optimizes data utility and enhances knowledge about KSC conditions.
- Data must be collected that is statistically robust and scientifically meaningful so a valid estimate of the mean chemical concentration in background media can be developed.
- Field and laboratory procedures and processes must be conducted under a Florida approved Comprehensive Quality Assurance Program and documented according to approved and accredited methods.
- Chain of custody records must be maintained, field sampling equipment and sample containers must be free from contamination, equipment to perform analyses must be rigorously calibrated and readily available.
- Personnel must show competence to perform the tests,
- A computerized data management system must be in place to allow for the electronic transfer of data, eliminating potential data entry errors.

1.2 Laboratory Qualifications

The contractor laboratory selected to conduct field sampling and laboratory analyses for this project was Environmental Conservation Laboratories, Inc. (ENCO) of Orlando Florida. ENCO is a full service laboratory that operates under approved Florida Department of Environmental Protection Standard Operating Procedures for Laboratory Operations and Sample Collection Activities. Their Comprehensive QA Plan is # 960038. In addition to the Florida State certification, ENCO participates in the USEPA Water Pollution (WP) proficiency program twice a year and the American Association for Laboratory Accreditation for non-water matrices. A detailed description of the ENCO Quality Control and Quality Assurance procedures can be found in the Environmental Conservation Laboratories, Inc. Quality Assurance Program Plan (Inman, 1997 with annual updates).

2.0 Overall Measurement Systems

2.1 Field Sampling

2.1.1 Sample Site Selection

Sample site selection was defined based on statistical sampling theory that included a stratified design by media type (Cochran, 1953; Sokal and Rohlf,1969). Sampling was to provide data of sufficient quality and quantity to satisfy possible future statistical test requirements. Media (soil, groundwater, surface water, and sediments) were stratified according to unique features of each media type based on existing knowledge of the KSC area. All sample sites were located away from industrial activities with a minimal history of disturbance by man. The history of activity was determined by examining historical aerial photography (1943-1996) for KSC in combination with current land use maps and GIS layers. GIS data included information on soils, vegetation types, and groundwater systems. Once sample sites were selected and locations were identified in the field, coordinates were obtained using a Trimble Pathfinder Professional Global Positioning System using real-time differential—correction procedures (Trimble Navigation Limited,1994). This method has been shown to be accurate in the X,Y plane to less than one meter for the KSC area.

2.1.2 Sample Sizes

Determination of sample size in each strata was based on the law of diminishing returns and cost (Levin and Rubin, 1980). The best sample number is defined as the largest that can be achieved economically given that the precision of the estimate of the mean increases with sample size. In addition, the standard error of the mean decreases in proportion to the square root of the sample number. For example, an increase in sample size from 10 to 20 reduces the standard

error (and the width of the confidence limits) by approximately one third (Levin and Ruben, 1980). To achieve another one third reduction in the width of the confidence limits would require a sample size of about 45, more than doubling the cost of laboratory analysis and significantly increasing field sampling requirements.

Based on existing knowledge, media types were stratified as follows, soils (10), surface water (2), sediment class (2) and groundwater (4). For soils, surface waters, and sediments, the sample size per strata was set at 20. This sample size provided statistically robust results and a quality estimate of mean values within cost limitations of the project. Because of cost associated with groundwater well installation, the sample number was limited to 10 per strata. For groundwater sampling, a vertical stratification was also requested and an additional 11 wells were installed to accommodate sampling from different depths. This produced 51 groundwater samples.

2.1.3 Sample Collection Efficiency

All field sampling activities followed protocols described in the ENCO Florida State approved Quality Assurance Plan. LSSC staff determined all sample site selections and accompanied ENCO staff during all sampling events for soils, surface waters, and sediments, verifying use of proper procedures. In addition, a meeting was held with a KSC Remediation Site contractor to ensure groundwater well sampling was conducted in the same fashion as ongoing and planned remediation sampling. These procedures are defined in the NASA/KSC Sampling and Analysis Plan for RCRA Facility Investigations at Kennedy Space Center, Florida and Cape Canaveral Air Station, Florida, Volume 4 (NASA, 1996)

Results of sampling activities are summarized in Tables F.1 and F.2, which describe expected and actual number of samples by parameter for each media type. Data collection success for this survey was very high, at greater than 99.99%. For soils, 200 samples and 20 duplicates were collected producing a total of 220 data points per parameter. One bulk density and one pH sample were lost during sample processing. This resulted in a total of 23098 measurements out of an expected 23100 (99.99%). A total of 40 sediment samples sites were sampled with the inclusion of 6 duplicates. All 4738 (100%) sediment parameters were collected. Surface water sampling was conducted at 40 stations. There were 6 duplicates and one equipment blank for a total of 47 possible samples. Dissolved oxygen, pH and specific conductivity were collected at the 40 surface water stations with no field duplication. Surface water sampling produced 4531 measurements. Groundwater sampling was conducted at 51 well locations. Field crews collected six duplicate samples and 24 equipment blanks for a total of 81 samples. The turbidity value for one well was not recorded, producing 7706 possible measurements out of an expected 7707 (99.99%). Overall sampling success was very high.

2.1.4 Equipment Blanks

During the sampling program 25 equipment blanks were processed to test the possibility of procedure induced contamination of samples. Results for all parameters observed above detection limits in equipment blanks are presented in Table F.3. For sample OR3879, several organic compounds were reported in concentrations above the detection limit in the equipment blank, however the same compounds were reported as below detection in the associated groundwater samples. Therefore, possible equipment contamination had no meaningful influence on the sample data.

Detectable values of chlorides were observed in equipment blanks for OR3706, OR3836, and OR4456. Equipment Blank values were below 95 mg/l. Sample results for these three stations ranged between 3300 and 7500 mg/l limiting possible contamination effects to less than 1.5%.

Three metals, copper, lead and zinc were reported as present in the equipment blank for sample OR3890. Reported sample results for these metals were all below detection. Based on the Equipment Blank analysis, significant sample contamination did not occur during field operations.

2.2 Laboratory Performance

2.2.1 Holding Times

Review of the database information on time elapsed between sample collection, processing, and analysis indicates that all samples were processed within required method specific holding times. There are no issues associated with laboratory sample processing rates for the Baseline Chemical Characterization survey.

2.2.2 Laboratory Blanks

Results of method blank analyses are presented in Table 4. Laboratory blank detects were observed in five sample runs for metals. No lab blank detections were reported for the EPA 8081, 8082, 8310 or 8151 organic analyses methods.

In soil sample report OR2540 sodium was detected at 28 mg/kg. Soil sample values for this report ranged between 9,300 and 56,000 mg/kg. The blank contamination level was below the level of significant digits recorded for the analysis and therefore had no meaningful impact on results. Similarly, calcium was observed in the soil sample lab blank for report OR2613 at 47 mg/kg.

Sample results ranged between 5100 and 6700 mg/kg and were not meaningfully influenced by the possible low level contamination.

In report OR2613 aluminum was report in the lab blank as 69 mg/kg while the three coastal soil samples from the report were between 180 and 320 mg/kg. This possible error in 3 of the 23 total samples used to calculate the mean (282.6 mg/kg) in the coastal soil class could produce a positive bias of up to 4.25 %. This possible positive bias in the mean value estimate is minor considering the large estimate of the standard deviation (262.7 mg/kg) for aluminum. Calcium was also reported present in the lab blank.

Cadmium was reported in the laboratory blank for surface water laboratory reports OR2525 and OR3023. Cadmium concentrations in associated field samples were reported as below detection. Similarly, lead and zinc were reported in the laboratory blank for surface water report OR4986. The field sample results data for these stations were reported as below detection.

2.2.3 Surrogate Recovery

Surrogate recovery results, summarized by lab number and method, are presented in Table F.4. For the 268 surrogate analysis, 19 or 7% were found to be out side expected control limits. Eight of the 67 runs of the EPA 8081 procedure for organochlorine pesticides were beyond control limits. One control limit violation was low and 7 were high suggesting a possible slight positive bias in the estimated mean values of organochlorine pesticides. The impact of this bias on statistical estimates of baseline values is minimal because nearly all parameters were reported as below detection in field samples.

Two surrogate control limit violations were observed in the 67 lab runs using the EPA 8082 method for aroclors. Both violations were below the lower control limit for the method suggesting actual values could potentially be slightly higher than reported values. Impacts of this possible bias on this group of parameters is not significant because all field samples were reported as below detection.

Nine of the surrogate control limits violations were associated with the 67 runs of the EPA 8151 analysis procedure for chlorinated herbicides. All nine of the out of limit recoveries were biased high suggesting estimated mean values for these parameters could be somewhat higher than true baseline concentrations. The importance of this to the Baseline Database is minimal since the majority of sample values for herbicides were reported as below detection.

There were no surrogate recovery control limit violations for the EPA 8310 method for polycyclic aromatic hydrocarbons.

2.2.4 Matrix Spikes (duplicate precision)

Matrix spike duplicate analysis results are presented in Table F.4. Fifty-five of the 268 (20.5%) analyses had one or more parameters reported above the control limit. In most cases, the reported sample result values for these analyses were below detection and the poor precision displayed by the laboratory in analyzing duplicate matrix spikes had no influence on database quality. For the EPA Method 8081, 4,4-DDT, Heptachlor, Lindane, and Aldrin analysis were noted on one or more occasions as being outside desired control limits. In each case the associated field sample results were reported as below detection.

All matrix spike duplicate analyses for the EPA Method 8082 were within the control limit, and there were no issues with precision for this procedure.

Matrix spike duplicate data for the EPA Method 8151 indicated poor precision for 2,4-D on soils laboratory report OR3429. Field sample results for all other parameters where precision was out of control limits were reported as below detection. There were no data quality impacts from poor laboratory precision on this EPA Method. Matrix spike duplicate data for EPA Method 8310 were outside control limits for Naphthalene in soils laboratory reports OR2665 and OR3082.

Interpretation of results for matrix spike duplicate analyses for metals procedures is confounded by the problem of matrix interferences with spike recoveries. The coastal environment of Florida with the associated high levels of sodium, chloride, magnesium. aluminum, and silica often result in uncontrolled interference with laboratory procedures. This problem introduces error in analyses and directly influences quantification of low concentration level recoveries. This in turn influences precision estimates. For this database, the poor matrix spike recoveries for sodium, calcium, magnesium and aluminum are considered inconsequential. Matrix spike RPD values beyond control limits for arsenic, barium, cadmium, chromium, iron, nickel and zinc are not significant because field sample values are reported as below detection on the laboratory reports.

2.2.5 Spike Recoveries

Spike recovery is one of the best indicators of laboratory procedure, equipment, and operator performance accuracy. Spike recovery analyses data are summarized in Figures F.1. through F.41. Figures are presented in alphabetical order. For this assessment, we considered spike recoveries to be acceptable if they were between 55 and 145% for organics and 65 and 135% for metals. If all recoveries fell within the range and the sampling distribution appeared normal (little skewness or kurtosis) the procedure was ranked as excellent. If a plot had less that 10% of values outside the control range and the sampling distribution appeared normal, it was ranked as good. A plot with between 10 and 20% of

values outside the control range was ranked fair. A rank of poor was assigned to any plot that had more than 20% of results outside the control limits.

A summary of reported results is presented in Table F.5. Spike recovery for the surrogate 2,4-DCAA is presented in Figure F.2. Spike recovery was fair with 25 of 266 samples being below the 55% lower limit and five values being above the 145% upper limit. The sampling distribution appears slightly biased low by approximately 10%. These surrogate results are considered representative of the performance of sampling instruments and procedures for many organic analysis. Issues associated with the fair performance include matrix interferences and poor spike extraction due to tight bonding with materials in the sample. This may be especially problematic in soil and sediment spike recoveries where the introduced material is mixed with the complex soil matrix prior to extraction.

Metals spike recovery rates were generally very good with fifteen being ranked as excellent (Table F.5) In contrast, only two of the organic recoveries ranked excellent.

2.3 Reporting

2.3.1 Chain of Custody Records

Chain of custody record evaluation is presented in Table F.5. All records were maintained by ENCO staff during all phases of the baseline program. Examination of chain of custody data sheets revealed that signatures were present for sample bottle pick-up, sample check-in, and sample processing.

Four data sheets were found to have data entry errors. These included mistakes in the date fields (wrong month), for the date of sampling and in the sample bottle check out signature field. In the case of the date issue the field personnel wrote that samples were collected in the eighth month instead of the ninth month. All other dates on the sheet were correct.

Sample bottles and equipment were prepared and provided by the laboratory on an as needed basis. All sample bottle storage, when needed, was conducted at the Cape Dispensary building 49635, inside the security of CCAFS. These human errors in data entry were not associated with any unusual or inconsistent chemical data values.

2.3.2 Narratives

Laboratory narratives describing issues associated with sample collection and processing were provided to explain unusual results and procedures. The text

provided was often brief, lacking detail necessary to assess data issues. This resulted in the need to make many calls to the laboratory for additional information and clarification.

2.3.3 Electronic Data Transfer

This area represents the single weakest aspect of the overall project. Data were provided by ENCO in electronic format as requested. However, loss of the LSSC Database Administrators early in the project prior to completion of database development and loading produced many delays. In addition, discrepancies were found between the electronic data and the several hardcopy laboratory reports. This required extensive database review, data re-entry and verification to correct all problems.

3.0 Summary

Overall this chemical characterization provides a statistically rigorous data set describing mean background chemical concentration values for metals, herbicides, pesticides, and PAH in soils, sediments, surface water, and groundwater of KSC. A large number of parameters were found to be below detection limits of the EPA required methods. The spatially stratified approach to sampling with moderate to large sample sizes (n= 10-20) allows for definition of expected ranges of parameters within strata and provides a data set that can be used to characterize concentration difference between strata and potential contamination sites. Estimates of mean values, minimum values, maximum values, and standard deviations are representative of baseline conditions for these parameters in background areas of KSC at the time of sampling. The lack of temporal sampling within a media type precludes definition of seasonal variability that may be associated with changing environmental conditions such as rainfall patterns, severe storm events, wildfires, biological activity, and temperature.

Field and laboratory performance on this project are considered good with the exception of the electronic data transfer procedure and onsite database development activities. Field and laboratory procedures introduced little error in the data as indicated by equipment and laboratory blanks and matrix spike recoveries. Chain-of-custody records were maintained with few human errors, Sample holding times were met and appropriate EPA methods were utilized in all analyses. Laboratory and field procedures followed Florida DEP and NASA approved procedures. Metals analysis were found to be of high quality with few exceptions. Organic analysis were more problematic due to procedural complexities; however, the majority of analyses were reported as below detection as would be expected in background areas of KSC and the Merritt Island

National Wildlife Refuge. One set of cation exchange capacity measurements for coastal soils were found to be an order of magnitude higher than previously measured values or values reported in the scientific literature for coastal soils. The laboratory was not able to identify a procedural cause for the error so data were excluded from analysis as not representative of background conditions.

4.0 References

Cochran, W. G., 1963, Sampling Techniques. John Wiley & Sons Inc., New York, NY., 413 pp.

Keith, L. H., 1991, Environmental Sampling and Analysis, A Practical Guide. Lewis Publishers, Chelsea, MI., 143 pp.

Levin, R. I. and D. S. Rubin, 1980, Applied Elementry Statistics. Prentice-Hall Inc. Englewood Cliffs, NJ. 578 pp.

Sokal, R.R., and F. J. Rohlf, 1969, Biometry. W. H. Freeman Co., San Francisco, CA., 776 pp.

Table G-1. Summary of surface water and groundwater parameter collection efficiency for the KSC Baseline Chemical Characterization. The expected numbers include duplicate samples where appropriate.

Parameter Name	Su	ırface Water		Groundwater			
	Expected	Collected	Percent	Expected	Collected	Percent	
1 - Methylnaphthalene	47	47	100.00	81	81	100.00	
2-(2,4,5-Trichlorophenoxy)propionic acid	47	47	100.00	81	81	100.00	
(2,4,5 - TP) (Silvex) 2 - Methylnaphthalene	47	47	100.00	81	81	100.00	
2 - Methymaphthalene 2,4,5,6-TCMX	47	47	100.00	81	81	100.00	
	47	47 47	100.00	81	81	100.00	
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)		47 47	100.00	81	81	100.00	
2,4-DCAA	47			81	81	100.00	
2,4-Dichlorophenoxy acetic acid (2,4 - D)	47	47	100.00				
3,5-DCBA	47	47	100.00	81	81	100.00	
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	47	47	100.00	81	81	100.00	
4-(2-Methyl-4-chlorophenoxy)butyric acid (MCPP)	47	47	100.00	81	81	100.00	
4 - Nitrophenol	47	47	100.00	81	81	100.00	
4,4' - DDD	47	47	100.00	81	81	100.00	
4,4' - DDE	47	47	100.00	81	81	100.00	
4,4' - DDT	47	47	100.00	81	81	100.00	
Acenaphthene	47	47	100.00	81	81	100.00	
Acenaphthylene	47	47	100.00	81	81	100.00	
Acifluorfen	47	47	100.00	81	81	100.00	
Aldrin	47	47	100.00	81	81	100.00	
alpha - BHC	47	47	100.00	81	81	100.00	
Aluminum	47	47	100.00	81	81	100.00	
Anthracene	47	47	100.00	81	81	100.00	
Antimony	47	47	100.00	81	81	100.00	
Aroclor-1254 (noncarcinogenic)	47	47	100.00	81	81	100.00	
Arsenic (as carcinogen)	47	47	100.00	81	81	100.00	
Barium	47	47	100.00	81	81	100.00	
Bentazon	47	47	100.00	81	81	100.00	
Benzo(a)anthracene	47	47	100.00	81	81	100.00	
Benzo(a)pyrene	47	47	100.00	81	81	100.00	
Benzo(b)fluoranthene	47	47	100.00	81	81	100.00	
Benzo(g,h,i)perylene	47	47	100.00	81	81	100.00	
Benzo(k)fluoranthene	47	47	100.00	81	81	100.00	
Beryllium	47	47	100.00	81	81	100.00	
beta - BHC	47	47	100.00	81	81	100.00	
Bulk Density	NA	NA	NA	NA	NA	NA	
Cadmium	47	47	100.00	81	81	100.00	
Calcium	47	47	100.00	81	81	100.00	
CEC (as Na)	NA	NA	NA	NA	NA	NA	
Chloramben	47	47	100.00	81	81	100.00	
Chlordane (Total)	47	47	100.00	81	81	100.00	
Chlordane, alpha or gamma	47	47	100.00	81	81	100.00	
Chloride, Total	47	47	100.00	81	81	100.00	
Chromium (total)	47	47	100.00	81	81	100.00	
Chrysene	47	47	100.00	81	81	100.00	
Cobalt	47	47	100.00	81	81	100.00	

Table G-1. (cont.).

Parameter Name	Sı	urface Water	· · · · · · · · · · · · · · · · · · ·	Groundwater			
	Expected	Collected Percent		Expected	Collected		
Copper	47	47	100.00	81	81	100.00	
Dacthal	. 47	47	100.00	81	81	100.00	
Dalapon	47	47	100.00	81	81	100.00	
DBC	47	7	100.00	81	81	100.00	
delta - BHC	47	47	100.00	81	81	100.00	
Dibenz(a,h)anthracene	47	47	100.00	81	81	100.00	
Dicamba	47	47	100.00	81	81	100.00	
Dichloroprop [2-(2,4- Dichlorophenoxy)proponic acid]	47	47	100.00	81	81	100.00	
Dieldrin	47	47	100.00	81	81	100.00	
Dinoseb	47	47	100.00	81	81	100.00	
Dissolved Oxygen	40	40	100.00	51	51	100.00	
Endosulfan I	47	47	100.00	81	81	100.00	
Endosulfan II (beta)	47	47	100.00	81	81	100.00	
Endosulfan sulfate	47	47	100.00	81	81	100.00	
Endrin	47	47	100.00	81	81	100.00	
Endrin aldehyde	47	47	100.00	81	81	100.00	
Endrin Ketone	47	47	100.00	81	81	100.00	
Fluoranthene	47	47	100.00	81	81	100.00	
Fluorene	47	47	100.00	81	81	100.00	
gamma - BHC (Lindane)	47	47	100.00	81	81	100.00	
Heptachlor	47	47	100.00	81	81	100.00	
Heptachlor Epoxide(a)	47	47	100.00	81	81	100.00	
Heptachlor Epoxide(b)	47	47	100.00	81	81	100.00	
ndeno(1,2,3-cd)pyrene	47	47	100.00	81	81	100.00	
ron	47	47	100.00	81	81	100.00	
sodrin	47	47	100.00	81	81	100.00	
.ead	47	47	100.00	81	81	100.00	
Magnesium	47	47	100.00	81	81	100.00	
Manganese	47	47	100.00	81	81	100.00	
<i>I</i> CPA	47	47	100.00	81	81	100.00	
Mercury (inorganic)	47	47	100.00	81	81	100.00	
Methoxychior	47	47	100.00	81	81	100.00	
f irex	47	47	100.00	81	81	100.00	
laphthalene	47	47	100.00	81	81	100.00	
lickel	47	47	100.00	81	81	100.00	
-terphenyl	47	47	100.00	81	81	100.00	
CB-1016/1242	47	47	100.00	81	81	100.00	
CB-1221	47	47	100.00	81	81	100.00	
CB-1232	47	47	100.00	81	81	100.00	
CB-1248	47	47	100.00	81	81	100.00	
CB-1260	47	47	100.00	81	81		
entachlorophenol	47	47	100.00	81	81	100.00 100.00	
ercent Solids	NA	NA	NA	NA	NA		
н	40	40	100.00	51	51	NA 100.00	
henanthrene	47	47	100.00	81		100.00	
icloram	47	47	100.00	81	81 81	100.00	
otassium	47	47	100.00	81	81 81	100.00 100.00	

Table G-1. (cont.).

Parameter Name	Sı	urface Water		Groundwater		
	Expected	Collected	Percent	Expected	Collected	Percent
Pyrene	47	47	100.00	81	81	100.00
Resistivity	NA	NA	NA	NA	NA	NA
Selenium	47	47	100.00	81	81	100.00
Silver	47	47	100.00	81	81	100.00
Sodium	47	47	100.00	81	81	100.00
Spec Cond-Field	40	40	100.00	51	51	100.00
Temp-field	40	40	100.00	51	51	100.00
Texture (No.4)	NA	NA	NA	NA	NA	NA
Texture (No. 10)	NA	NA	NA	NA	NA	NA
Texture (No. 40)	NA	NA	NA	NA	NA	NA
Texture (No. 60)	NA	NA	NA	NA	NA	NA
Texture (No. 100)	NA	NA	NA	NA	NA	NA
Texture (No. 200)	NA	NA	NA	NA	NA	NA
Thallium	47	47	100.00	81	81	100.00
Total Dissolved Solids, Filterable residue	47	47	100.00	81	81	100.00
Total organic carbon	47	47	100.00	81	81	100.00
Toxaphene	47	47	100.00	81	81	100.00
Turbidity	47	47	100.00	51	50	98.04
Vanadium	47	47	100.00	81	81	100.00
Zinc	47	47	100.00	81	81	100.00
Total Number	4531	4531	100.00	77 07	7706	99.99

NA = Not Analyzed

Table G-2. Summary of soil and sediment parameter collection efficiency for the KSC Baseline Chemical Characterization. The expected numbers include duplicate samples where appropriate.

Parameter Name	The expected numbers include duplicate samples where appropriate Soil Sediment							
	Expected	Collected	Percent	Expected	Collected	Percent		
1 - Methylnaphthalene	220	220	100.00	46	46	100.00		
2-(2,4,5-Trichlorophenoxy)propionic acid (2,4,5 - TP) (Silvex)	220	220	100.00	46	46	100.00		
2 - Methylnaphthalene	220	220	100.00	46	46	100.00		
2,4,5,6-TCMX	220	220	100.00	46	46	100.00		
2,4,5-Trichlorophenoxy acetic acid (2,4,5 - T)	220	220	100.00	46	46	100.00		
2,4-DCAA	220	220	100.00	46	46	100.00		
2,4-Dichlorophenoxy acetic acid (2,4 - D)	220	220	100.00	46	46	100.00		
3,5-DCBA	220	220	100.00	46	46	100.00		
4-(2,4-Dichlorophenoxy)butyric acid (2,4 - DB)	220	220	100.00	46	46	100.00		
4-(2-Methyl-4-chlorophenoxy)butyric acid (MCPP)	220	220	100.00	46	46	100.00		
4 - Nitrophenol	220	220	100.00	46	46	100.00		
4,4' - DDD	220	220	100.00	46	46	100.00		
4,4' - DDE	220	220	100.00	46	46	100.00		
4,4' - DDT	220	220	100.00	46	46	100.00		
Acenaphthene	220	220	100.00	46	46	100.00		
Acenaphthylene	220	220	100.00	46	46	100.00		
Acifluorfen	220	220	100.00	46	46	100.00		
Aldrin	220	220	100.00	46	46	100.00		
alpha - BHC	220	220	100.00	46	46	100.00		
Aluminum	220	220	100.00	46	46	100.00		
Anthracene	220	220	100.00	46	46	100.00		
Antimony	220	220	100.00	46	46	100.00		
Aroclor-1254 (noncarcinogenic)	220	220	100.00	46	46	100.00		
Arsenic (as carcinogen)	220	220	100.00	46	46	100.00		
Barium	220	220	100.00	46	46	100.00		
Bentazon	220	220	100.00	46	46	100.00		
Benzo(a)anthracene	220	220	100.00	46	46	100.00		
Benzo(a)pyrene	220	220	100.00	46	46	100.00		
Benzo(b)fluoranthene	220	220	100.00	46	46	100.00		
Benzo(g,h,i)perylene	220	220	100.00	46	46	100.00		
Benzo(k)fluoranthene	220	220	100.00	46	46	100.00		
Beryllium	220	220	100.00	46	46	100.00		
peta - BHC	220	220	100.00	46	46	100.00		
Bulk Density	220	219	99.55	NA	NA	NA		
Cadmium	220	220	100.00	46	46	100.00		
Calcium	220	220	100.00	46	46	100.00		
CEC (as Na)	220	220	100.00	NA	NA	NA		
Chloramben	220	220	100.00	46	46	100.00		
chlordane (Total)	220	220	100.00	46	46	100.00		
chlordane, alpha or gamma	220	220	100.00	46	46	100.00		
chloride, Total	NA	NA	NA	NA	NA	NA		
chromium (total)	220	220	100.00	46	46	100.00		
Chrysene	220	220	100.00	46	46	100.00		
Cobalt	220	220	100.00	46	46	100.00		
Copper Copper	220	220	100.00	46	46	100.00		

Table G-2. (cont.).

Parameter Name		Soil		Sediment			
	Expected	Collected	Percent	Expected	Collected	Percent	
Dacthal	220	220	100.00	46	46	100.00	
Daiapon	220	220	100.00	46	46	100.00	
DBC	220	220	100.00	46	46	100.00	
delta - BHC	220	220	100.00	46	46	100.00	
Dibenz(a,h)anthracene	220	220	100.00	46	46	100.00	
Dicamba	220	220	100.00	46	46	100.00	
Dichloroprop [2-(2,4- Dichlorophenoxy)proponic acid]	220	220	100.00	46	46	100.00	
Dieldrin	220	220	100.00	46	46	100.00	
Dinoseb	220	220	100.00	46	46	100.00	
Dissolved Oxygen	NA	NA	NA	NA	NA	NA	
Endosulfan I	220	220	100.00	46	46	100.00	
Endosulfan II (beta)	220	220	100.00	46	46	100.00	
Endosulfan sulfate	220	220	100.00	46	46	100.00	
Endrin	220	220	100.00	46	46	100.00	
Endrin aldehyde	220	220	100.00	46	46	100.00	
Endrin Ketone	220	220	100.00	46	46	100.00	
Fluoranthene	220	220	100.00	46	46	100.00	
Fluorene	220	220	100.00	46	46	100.00	
gamma - BHC (Lindane)	220	220	100.00	46	46	100.00	
Heptachlor	220	220	100.00	46	46	100.00	
Heptachlor Epoxide(a)	220	220	100.00	46	46	100.00	
Heptachlor Epoxide(b)	220	220	100.00	46	46	100.00	
Indeno(1,2,3-cd)pyrene	220	220	100.00	46	46	100.00	
Iron	220	220	100.00	46	46	100.00	
Isodrin	220	220	100.00	46	46	100.00	
Lead	220	220	100.00	46	46	100.00	
Magnesium	220	220	100.00	46	46	100.00	
Manganese	220	220	100.00	46	46	100.00	
MCPA	220	220	100.00	46	46	100.00	
Mercury (inorganic)	220	220	100.00	46	46	100.00	
Methoxychlor	220	220	100.00	46	46	100.00	
Mirex	220	220	100.00	46	46	100.00	
Naphthalene	220	220	100.00	46	46	100.00	
Nickel	220	220	100.00	46	46	100.00	
p-terphenyi	220	220	100.00	46	46	100.00	
PCB-1016/1242	220	220	100.00	46	46	100.00	
PCB-1221	220	220	100.00	46	46	100.00	
PCB-1232	220	220	100.00	46	46	100.00	
PCB-1248	220	220	100.00	46	46	100.00	
PCB-1260	220	220	100.00	46	46	100.00	
Pentachlorophenol	220	220	100.00	46	46	100.00	
Percent Solids	220	220	100.00	46	46	100.00	
рН	220	219	99.55	46	46	100.00	
Phenanthrene	220	220	100.00	46	46	100.00	
Picloram	220	220	100.00	46	46	100.00	
Potassium	220	220	100.00	46	46	100.00	
Pyrene	220	220	100.00	46	46	100.00	

Table G-2. (cont.).

Parameter Name		Soil			Sediment	
	Expected	Collected	Percent	Expected	Collected	Percent
Resistivity	220	220	100.00	46	46	100.00
Selenium	220	220	100.00	46	46	100.00
Silver	220	220	100.00	46	46	100.00
Sodium	220	220	100.00	46	46	100.00
Spec Cond-Field	NA	NA	NA	NA	NA	NA.
Temp-field	NA	NA	NA	NA	NA	NA.
Texture (No.4)	220	220	100.00	46	46	100.00
Texture (No. 10)	220	220	100.00	46	46	100.00
Texture (No. 40)	220	220	100.00	46	46	100.00
Texture (No. 60)	220	220	100.00	46	46	100.00
Texture (No. 100)	220	220	100.00	46	46	100.00
Texture (No. 200)	220	220	100.00	46	46	100.00
Fhallium .	220	220	100.00	46	46	100.00
Total Dissolved Solids, Filterable residue	NA	NA	NA	NA	NA	NA
Total organic carbon	220	220	100.00	46	46	100.00
Toxaphene	220	220	100.00	46	46	100.00
Furbidity	NA	NA	NA	NA	NA	NA
/anadium	220	220	100.00	46	46	100.00
Zinc Zinc	220	220	100.00	46	46	100.00
otal Number	23100	23098	99.99	4738	4738	100.00

NA = Not Analyzed

Table G-3. Results of equipment blank analysis. Values presented are for all parameters reported above detection limits. All other parameters analyses results reported as below detection.

Sample Number	Chemical Name	Concentration	Units
OR3706	Chloride, Total	4	mg/L
OR3836	Chloride, Total	92	mg/L
OR3836	Total Dissolved Solids, Filterable residue	40	mg/L
OR3879	Benzo(a)anthracene	0.09	ug/L
OR3879	Benzo(a)pyrene	0.14	ug/L
OR3879	Chrysene	0.16	ug/L
OR3879	Indeno(1,2,3-cd)pyrene	0.11	ug/L
OR3890	Copper	0.44	mg/L
OR3890	Lead	0.28	mg/L
OR3890	Zinc	1.9	mg/L
OR3955	Total organic carbon	8	mg/L
OR4456	Chloride, Total	6	mg/L

Table G-4. Summary of laboratory test data for blanks, surrogate recoveries, and relative percent differences of matrix spike duplicates.

Lab Report	Media	Analysis	I oh Blasti	10	
Number	IVIOLIA	Method	Lab Blank Results	Surrogate Recovery	
OR2472	SW	8081	ND	Hecovery	Difference
OR2472		8082	ND	 -	
OR2472		8151	ND	HIGH	2,4-D, 2,4-DB RPD high. Sample
				1	values reported below detection.
OR2472		8310	ND		The second secon
OR2472		METALS	ND		Calcium, Magnesium, Potassium, Sodium with matrix interference. Copper and iron RPD high.
000470	CED	0004	+		
OR2473	SED	8081	ND		4,4-DDT RDP high. Sample reported below detection.
OR2473		8082	ND		•
OR2473		8151	ND		2,4 D RDP high. Sample reported below detection.
OR2473		8310	ND		
OR2473		METALS	ND		
					
OR2524	CED	0004			
	SED	8081	ND	HIGH	4,4-DDT RDP high. Sample reported below detection.
OR2524		8082	ND		
OR2524		8151	ND		2,4-D RDP high. Sample reported below detection.
OR2524		8310	ND		
OR2524		METALS	ND		
OR2525	sw	8081	ND		
OR2525		8082	ND		
OR2525		8151	ND	HIGH	2,4-D, 2,4-DB RPD high. Samples reported below detection.
OR2525		8310	ND		Acenaphthene RPD high. Samples
ODOSOS		1457416			reported below detection.
OR2525		METALS	cadmium		
OR2540	so	8081	ND ND	HICH	
OR2540	-55	8082	ND ND	HIGH	
OR2540		8151	ND		24D BBD High Completed to
					2,4 D RPD High. Sample value reported below detection.
OR2540		8310	ND		
OR2540		METALS	sodium		
OR2559	so	8081	ND		
OR2559		8082	ND		
OR2559		8151	ND		Dicamba RPD high. Sample value reported below detection.
OR2559		8310	ND		reported below detection.

Table G-4. (cont.).

Lab Report	Media	Analysis	Lab Blank	Surrogate	Matrix Spike Relative Percent
Number	Media	Method	Results	Recovery	Difference
OR2559		METALS	ND		Aluminum, Iron, and Sodium matrix
		_			interference. Iron not detected in
					samples.
OR2575	SW	8081	ND		
OR2575		8082	ND		
OR2575		8151	ND	HIGH	2,4-D, and 2,4-DB RPD high. Samples
					reported below detection.
OR2575		8310	ND		
OR2575		METALS	ND		
OR2576	SED	8081	ND		Heptachlor RPD high. Samples reported below detection.
OR2576		8082	ND		
OR2576		8151	ND	HIGH	Dicamba RPD high. Samples reported below detection.
OR2576		8310	ND		Benzo(a)pyrene RPD high. Samples
000570		METALO	ND		reported below detection.
OR2576		METALS	ND		
OBSESS	SW	8081	ND		
OR2592 OR2592	344	8082	ND		
OR2592		8151	ND	[
OR2592		8310	ND ND		
OR2592		METALS	ND ND		
0112332		WILTALO	110		
OR2593	SED	8081	ND		
OR2593		8082	ND		
OR2593		8151	ND		Dicamba RPD high. Samples reported
011233					below detection.
OR2593		8310	ND		Benzo(a)pyrene RPD high. Samples
					reported below detection.
OR2593		METALS	ND		
			,		
OR2613	SO	8081	ND		Heptachlor RPD high. Sample values reported below detection.
OR2613		8082	ND		
OR2613		8151	ND		
OR2613		8310	ND		Benzo(a)pyrene RPD high. Sample
					values reported below detection.
OR2613		METALS	aluminum,		Magnesium RPD high. Matrix
			calcium	<u> </u>	interference.
000000		0004	ND		Hentopher PPD high Sample values
OR2633	so	8081	ND		Heptachlor RPD high. Sample values reported below detection.
OBJECT		8082	ND		reported below detection.
OR2633 OR2633		8151	ND		
OR2633		8310	ND ND		
UNZOSS	L	1 0010	140	<u> </u>	<u></u>

Table G-4. (cont.).

Lab Report	Media	Analysis	Lab Blank	Surrogate	Matrix Spike Relative Percent
Number	Micaia	Method	Results	Recovery	Difference
OR2633		METALS	ND	riccovery	Directore
0112000			.,,,,		
OR2642	SW	8081	ND		
OR2642		8082	ND		
OR2642		8151	ND		
OR2642		8310	ND		
OR2642		METALS	ND		Calcium, Magnesium, Manganese, Potassium, and Sodium RPD high. Matrix interference.
OR2644	SED	8081	ND		
OR2644		8082	ND	·	
OR2644		8151	ND		
	-				
OR2644	-	8310	ND		
OR2644		METALS	ND		
OR2645	so	8081	ND		
OR2645		8082	ND		
OR2645		8151	ND		Dalpon RPD high. Sample values reported below detection.
OR2645		8310	ND		
OR2645		METALS	ND		
OR2665	SO	8081	ND		
OR2665		8082	ND		
OR2665		8151	ND		Dalapon RPD high. Samples reported below detection.
OR2665		8310	ND		RPD high. Naphthalene poor precision, Acenaphthene sample values reported below detection.
OR2665		METALS	ND		Sodium, Aluminum, Calcium and Iron matrix interference. Iron not detected in sample.
OR2693	so	8081	ND		
OR2693	30	8082	ND ND		
OR2693		8151	ND ND		Dalpon RPD high. Sample values
0,12000		0.51	110		reported below detection.
OR2693		8310	ND		Naphthalene, Acenaphthene RPD high. Sample values reported below detection.
OR2693		METALS	ND		
OR2725	SW	8081	ND		
OR2725		8082	ND		
OR2725		8151	ND		

Table G-4. (cont.).

Number OR2725				Surrogate	Matrix Spike Relative Percent
		Method	Results	Recovery	Difference
000		8310	ND		
OR2725		METALS	ND		Iron, Nickel RPD high. Samples
					reported below detection.
			· <u> </u>		
OR2726	SED	8081	ND		
OR2726		8082	ND		
OR2726		8151	ND		
OR2726		8310	ND		
OR2726		METALS	ND		
OR2742	SW	8081	ND		
OR2742		8082	ND		
OR2742		8151	ND		
OR2742		8310	ND		
OR2742		METALS	ND		Iron, Nickel RPD high. Samples
					reported below detection.
				···	
OR2743	SED	8081	ND		
OR2743		8082	ND		
OR2743		8151	ND		
OR2743		8310	ND		
OR2743		METALS	ND		
OR2763	so	8081	ND		
OR2763		8082	ND		
OR2763		8151	ND		
OR2763		8310	ND		
OR2763		METALS	ND ND		Copper RPD high. Poor precision.
OR2779	so	8081	ND	·	
OR2779		8082	ND		
OR2779		8151	ND ND		December 1
OR2779		8310	ND		Benzo(a)pyrene RPD high.
OR2779		METALS	ND		Calcium RPD high. Poor precision.
OB0704		9004			
OR2784	SO	8081	ND ND		
OR2784		8082	ND ND	LIOL	Disamba DDD high Complexions
OR2784	Ţ	8151	ND	HIGH	Dicamba RPD high. Sample values reported below detection.
OR2784		8310	ND		raported balow detection.
		METALS	ND ND		Calaium BBD high Boor procision
OR2784		IVIETALS	טא		Calcium RPD high. Poor precision.
OBSSS	so	8081	ND		
OR2836 OR2836	-30	8082	ND		
OR2836		8151	ND ND		Dicamba RPD high. Sample values
UN2030		0131	טאו		reported below detection.
OR2836		8310	ND		reported below detection.

Table G-4. (cont.).

Lab Report	Media	Anglimia	Joh Dissi	16	la di la di
Number	IVIOUIA	Analysis Method	Lab Blank Results	Surrogate	Matrix Spike Relative Percent
OR2836		METALS	ND	Recovery	Difference
0112000		MILIALS	IND	 	
OR2843	so	8081	ND		
OR2843	- 00	8082	ND	ļ	
OR2843		8151	ND	 	D:
0112040		0.51	IND		Dicamba RPD high. Sample values reported below detection.
OR2843		8310	ND		reported below detection.
OR2843		METALS	ND	 	Calcium and Iron matrix interference.
					Calcium and non matrix interreference.
OR2859	SO	8081	ND		
OR2859		8082	ND		
OR2859		8151	ND		
OR2859		8310	ND		
OR2859	*	METALS	ND		Calcium RPD high. Matrix interference.
					Calolain The Bright. Matrix litter erice.
OR2923	SO	8081	ND		
OR2923		8082	ND		
OR2923		8151	ND		
OR2923		8310	ND		
OR2923		METALS	ND		
OR2927	SO	8081	ND		
OR2927		8082	ND		
OR2927		8151	ND		
OR2927		8310	ND		
OR2927		METALS	ND		Calcium RPD high. Matrix interference.
OR2985	SW	8081	ND		
OR2985		8082	ND		
OR2985		8151	ND		
OR2985		8310	ND		
OR2985		METALS	ND		Arsenic, Barium, Cadmiun, Chromium,
	ĺ			i	Nickel RPD high. Samples reported
					below detection.
OBOOC	- CEC	9004			
OR2986 OR2986	SED	8081	ND		
OR2986		8082	ND		
OR2986		8151	ND ND		
OR2986		8310 METALS	ND		
UN2300		IVIE I ALS	ND		
OR3023	SW	8081	ND		
OR3023	344	8082	ND ND		
OR3023		8151	ND		
OR3023		8310	ND ND		
OR3023		METALS	cadmium		Potossium PPD high Dana andicia
00020		.n.L I ALO	Jaganian		Potassium RPD high. Poor precision.
OR3024	SED	8081	ND	HIGH	
J. 1004-7	<u> </u>	0001		пап	

Table G-4. (cont.).

Number	Lab Report	ort Media	Analysis	Lab Blank	Surrogate	Matrix Spike Relative Percent
OR3024 8082 ND OR3024 8151 ND OR3024 8310 ND OR3024 METALS ND OR3069 SO 8081 ND OR3069 8082 ND OR3069 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample RPD high. Poor precision. Acenaphthene RPD high. Poor precision. Acenaphthene RPD high. Poor precision. Acenaphthene RPD high. Poor precision. Acenaphthene RPD high. Poor precision. Acenaphthene not detected in sample RPD high. Poor precision. Acenaphthene not detected in sample RPD high. Poor precision. Acenaphthene ND OR3082 8310 ND ND OR3082 METALS ND ND OR3082 METALS ND ND OR3082 METALS ND ND OR3251 8081 ND HIGH OR3251 8082 ND ND OR3251 8310 ND ND OR3285 8081 ND ND OR3285 8082 ND ND OR3285		-				
OR3024 8151 ND OR3024 8310 ND OR3024 METALS ND OR3069 SO 8081 ND OR3069 8082 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample OR3069 B310 ND Acenaphthene not detected in sample OR3069 METALS ND 2,4-D, Naphthalene, and Acenaphthene RPD high. Poor precision. Acenaphthene RPD high. Poor precision. Acenaphthene RPD high. Poor precision. Acenaphthene RPD high. Poor precision. Acenaphthene not detected in sample OR3082 8310 ND ND OR3082 METALS ND ND OR3082 METALS ND ND OR3251 8082 ND ND OR3251 8082 ND ND OR3251 8310 ND HIGH OR3251 8310 ND OB3251 OR3285 8082 ND OB3285 OR3285 8082 ND OB3285 OR3285 8081					1.000.0.7	
OR3024 8310 ND OR3024 METALS ND OR3069 SO 8081 ND OR3069 8082 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample OR3069 8310 ND Acenaphthene not detected in sample OR3069 METALS ND Acenaphthene not detected in sample OR3082 SO 8081 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample OR3082 8310 ND ND ND OR3082 METALS ND ND ND OR3082 METALS ND ND ND OR3082 METALS ND HIGH ND OR3251 8081 ND HIGH ND OR3251 8151 ND HIGH ND OR3251 8310 ND ND ND OR3285 8082 ND ND ND OR3285 8082						
OR3024 METALS ND OR3069 SO 8081 ND OR3069 8082 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample ND OR3069 8310 ND Acenaphthene not detected in sample ND OR3069 METALS ND Acenaphthene not detected in sample ND OR3082 SO 8081 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene ND OR3082 8151 ND ND ND OR3082 METALS ND ND OR3082 METALS ND ND OR3082 METALS ND ND OR3251 SO 8081 ND HIGH OR3251 8151 ND HIGH OR3251 METALS ND OR3285 8082 OR3285 8081 ND OR3285 8151 OR3285 8310 ND OR3285 8310 OR3285 8310 ND OR328						
OR3069 SO 8081 ND OR3069 8082 ND OR3069 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample OR3069 8310 ND ND OR3082 SO 8081 ND OR3082 8082 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene RPD high. Poor precision. Acenaphthe						
OR3069 8082 ND OR3069 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample of the samp	<u> </u>		METALS.			
OR3069 8082 ND OR3069 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample of the samp	OR3069	9 SO	8081	ND		
OR3069 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample of the sampl				ND		
RPD high. Poor precision. Acenaphthene not detected in sample						2.4-D. Naphthalene, and Acenaphthene
OR3069 8310 ND OR3069 METALS ND OR3069 METALS ND OR3082 SO 8081 ND OR3082 8082 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample RPD high	011000					
OR3069 METALS ND OR3082 SO 8081 ND OR3082 8082 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample ND OR3082 8310 ND ND OR3082 METALS ND ND OR3082 METALS ND HIGH OR3082 SO 8081 ND HIGH OR3251 SO 8081 ND HIGH OR3251 8151 ND HIGH OR3251 8310 ND OR3251 OR3251 METALS ND OR3251 OR3285 SO 8081 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 SO 8081 ND						Acenaphthene not detected in samples.
OR3082 SO 8081 ND OR3082 8082 ND OR3082 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample ND OR3082 8310 ND ND OR3082 METALS ND ND OR3251 SO 8081 ND HIGH OR3251 8151 ND HIGH OR3251 8310 ND ND OR3251 METALS ND ND OR3251 METALS ND ND OR3285 SO 8081 ND OR3285 8151 ND HIGH OR3285 8151 ND HIGH OR3285 8310 ND OR3285 OR3285 8310 ND OR3285 OR3362 SO 8081 ND OR3362 8082 ND	OR3069	j9	8310	ND		
OR3082 SO 8081 ND OR3082 8082 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample ND OR3082 8310 ND ND OR3082 METALS ND ND OR3082 METALS ND HIGH OR3251 SO 8081 ND HIGH OR3251 8151 ND HIGH OR3251 8310 ND OR3251 OR3251 METALS ND OR3251 METALS ND OR3252 SO 8081 ND OR3285 SO 8081 ND OR3285 8310 ND OR3285 OR3285 8310 ND OR3285 METALS ND OR3285 8310 ND OR3362 SO 8081 ND OR3362 SO 8081 ND			METALS	ND		
OR3082 8082 ND OR3082 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample ND OR3082 8310 ND ND OR3082 METALS ND ND OR3082 METALS ND ND OR3251 SO 8081 ND HIGH OR3251 8151 ND HIGH OR3251 8310 ND ND OR3251 METALS ND ND OR3285 SO 8081 ND OR3285 8082 ND ND OR3285 8310 ND ND OR3285 METALS ND ND OR3285 METALS ND ND OR3362 SO 8081 ND OR3362 SO 8081 ND						
OR3082 8082 ND OR3082 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample ND OR3082 8310 ND ND OR3082 METALS ND ND OR3251 SO 8081 ND HIGH OR3251 8082 ND OR3251 RIST OR3251 8310 ND ND OR3251 METALS ND OR3251 OR3251 METALS ND OR3285 OR3285 8082 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 SO 8081 ND	OR3082	2 SO	8081	ND		
OR3082 8151 ND 2,4-D, Naphthalene, and Acenaphthe RPD high. Poor precision. Acenaphthene not detected in sample ND OR3082 8310 ND ND OR3082 METALS ND ND OR3251 SO 8081 ND HIGH OR3251 8082 ND ND OR3251 RIST ND OR3251 8310 ND ND OR3251 METALS ND OR3251 METALS ND ND OR3285 SO 8081 ND OR3285 8082 ND ND HIGH OR3285 METALS ND OR3285 8310 ND ND OR3285 METALS ND OR3285 METALS ND OR3285 METALS ND OR3362 SO 8081 ND ND OR3362 SO 8081 ND				ND		
RPD high. Poor precision. Acenaphthene not detected in sample ND ND ND						2,4-D, Naphthalene, and Acenaphthene
OR3082 8310 ND ND OR3082 METALS ND ND OR3082 METALS ND HIGH OR3251 8082 ND ND OR3251 8151 ND HIGH OR3251 METALS ND ND OR3285 SO 8081 ND OR3285 8082 ND ND OR3285 8310 ND HIGH OR3285 8310 ND OR3285 OR3285 METALS ND OR3362 SO 8081 ND OR3362 SO 8081 ND	•	_				
OR3082 METALS ND OR3251 SO 8081 ND HIGH OR3251 8082 ND ND HIGH OR3251 8310 ND ND OR3251 METALS ND OR3251 METALS ND ND OR3285 SO 8081 ND OR3285 8082 ND ND HIGH OR3285 8310 ND HIGH OR3285 8310 ND OR3285 OR3285 METALS ND OR3285 OR3362 SO 8081 ND OR3362 SO 8081 ND						Acenaphthene not detected in samples.
OR3082 METALS ND OR3251 SO 8081 ND HIGH OR3251 8082 ND HIGH OR3251 8151 ND HIGH OR3251 8310 ND OR3251 METALS ND OR3285 SO 8081 ND OR3285 8082 ND OR3285 8310 ND OR3285 8310 ND OR3285 METALS ND OR3285 METALS ND	OR3082	32	8310	ND		ND
OR3251 SO 8081 ND HIGH OR3251 8082 ND ND OR3251 8151 ND HIGH OR3251 8310 ND ND OR3251 METALS ND ND OR3285 SO 8081 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 SO 8081 ND				ND		
OR3251 8082 ND OR3251 8151 ND HIGH OR3251 8310 ND OR3251 METALS ND OR3285 SO 8081 ND ND OR3285 8151 ND HIGH OR3285 8310 ND ND OR3285 METALS ND OR3285 METALS ND ND OR3362 SO 8081 ND OR3362 SO 8081 ND ND OR3362 ND						
OR3251 8082 ND OR3251 8151 ND HIGH OR3251 8310 ND OR3251 METALS ND OR3285 SO 8081 ND ND OR3285 8151 ND HIGH OR3285 8310 ND ND OR3285 METALS ND OR3285 METALS ND ND OR3362 SO 8081 ND OR3362 SO 8081 ND ND OR3362 ND	OR3251	i1 SO	8081	ND	HIGH	
OR3251 8151 ND HIGH OR3251 8310 ND OR3251 METALS ND OR3285 SO 8081 ND OR3285 8082 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 SO 8082 ND				ND		
OR3251 8310 ND OR3251 METALS ND OR3285 SO 8081 ND OR3285 8082 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 8082 ND				ND	HIGH	
OR3251 METALS ND OR3285 SO 8081 ND OR3285 8082 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 8082 ND				ND		
OR3285 SO 8081 ND OR3285 8082 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 8082 ND				ND		
OR3285 8082 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 8082 ND						
OR3285 8082 ND OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 8082 ND	OR3285	5 SO	8081	ND		
OR3285 8151 ND HIGH OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 8082 ND				ND		
OR3285 8310 ND OR3285 METALS ND OR3362 SO 8081 ND OR3362 8082 ND				ND	HIGH	
OR3285 METALS ND OR3362 SO 8081 ND OR3362 8082 ND						
OR3362 SO 8081 ND OR3362 8082 ND						
OR3362 8082 ND					1	
OR3362 8082 ND	OR3362	2 SO	8081	ND		
OR3362 8151 ND 2,4-D RPD high. Sample value report	OR3362		8151	ND		2,4-D RPD high. Sample value reported
below detection.		[
OR3362 8310 ND	OR3362	32	8310	ND		
OR3362 METALS ND				ND		
OR3363 SO 8081 ND	OR3363	3 SO	8081	ND	1	
OR3363 8082 ND				ND	1	
OR3363 8151 ND 2,4-D RPD high. Sample values				·		
reported below detection.					<u></u>	
OR3363 8310 ND	OR3363	33	8310	ND		
OR3363 METALS ND				ND		
OR3429 SO 8081 ND	OR3429	29 SO	8081	ND		

Table G-4. (cont.).

Lab Report	Media	Analysis	Lab Blank	Curronata	Marin Orito Batain B
Number	modia	Method	Results	Recovery	Matrix Spike Relative Percent Difference
OR3429		8082	ND	riccovery	Difference
OR3429		8151	ND		2,4-D RPD high. Poor precision.
OR3429	_	8310	ND	 	2,4-b RFb High. Foor precision.
OR3429		METALS	ND		Calaium PRD high Matrix interference
		1,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	110		Calcium RPD high. Matrix interference.
OR3491	SW	8081	ND		
OR3491		8082	ND		
OR3491		8151	ND		2,4,5-TP silvex RPD high. Sample
OR3491		8310	ND		values reported below detection. Naphthalene and Acenaphthene RPD high. Sample values reported below detection.
OR3491		METALS	ND		detection.
OR3492	SED	8081	ND		
OR3492		8082	ND		
OR3492		8151	ND		
OR3492		8310	ND		
OR3492		METALS	ND		
OR3564	GW	8081	ND		
OR3564		8082	ND		
OR3564		8151	ND		2,4,5-TP silvex RPD high. Samples reported below detection.
OR3564		8310	ND		Naphthalene, Acaphthene RPD high. Samples reported below detection.
OR3564		METALS	ND		
OR3584	GW	8081	ND		
OR3584		8082	ND		
OR3584		8151	ND		2,4-DB RPD high. Samples reported below detection.
OR3584		8310	ND		
OR3584		METALS	ND		
OR3629	GW	8180	ND		
OR3629		8082	ND		
OR3629		8151	ND		2,4-DB RPD high. Samples reported below detection.
OR3629		8310	ND		DOION GOLGOLION.
OR3629		METALS	ND		
OR3634	GW	8081	ND		
OR3634		8082	ND		
OR3634		8151	ND		
OR3634		8310	ND		
OR3634		METALS	ND		

Table G-4. (cont.).

Lab Report	Media	Analysis	Lab Blank	Surrogate	Matrix Spike Relative Percent
Number		Method	Results	Recovery	Difference
OR3635	GW	8081	ND		
OR3635		8082	ND		
OR3635		8151	ND		
OR3635		8310	ND		
OR3635		METALS	ND		
OR3660	GW	8081	ND		
OR3660		8082	ND		
OR3660		8151	ND		
OR3660		8310	ND		
OR3660		METALS	ND		Sodium RPD high. Matrix interference.
OR3681		8081	ND	LOW	
OR3681		8082	ND	LOW	
OR3681		8151	ND		
OR3681		8310	ND		
OR3681		METALS	ND		
				<u> </u>	
OR3706	GW	8081	ND	LOW	
OR3706		8082	ND	LOW	
OR3706		8151	ND		
OR3706		8310	ND		Naphthalene RPD high. Samples reported below detection.
OR3706		METALS	ND		
0110700					
OR3717	GW	8081	ND		
OR3717		8082	ND		
OR3717		8151	ND		
OR3717		8310	ND		Naphthalene RPD high. Samples
					reported below detection.
OR3717		METALS	ND		
			· <u>-</u>		
OR3718	GW	8081	ND		
OR3718		8082	ND		
OR3718		8151	ND		
OR3718		8310	ND		Naphthalene RPD high. Samples
					reported below detection.
OR3718		METALS	ND		
OR3737	GW	8081	ND	 	4,4-DDT, Gamma BHC Lindane RPD
0113737	avv -	5501	.,,,,		high. Samples reported below detection.
OR3737		8082	ND		
OR3737		8151	ND		
OR3737		8310	ND		Naphthalene RPD high. Samples
					reported below detection.
OR3737		METALS	ND		

Table G-4. (cont.).

Lab Report	Media	Analysis	I ob Diant	10	
Number	IVIBUIA	Analysis Method	Lab Blank Results	Surrogate Recovery	
OR3835	GW	8081	ND	necovery	Difference 4,4-DDT, Gamma BHC Lindane RPD
			''		high. Samples reported below detection.
OR3835		8082	ND	·	ringin. Camples reported below detection.
OR3835		8151	ND		
OR3835		8310	ND		
OR3835		METALS	ND		
OR3836	GW	8081	ND		4,4-DDT, Gamma BHC Lindane RPD
OR3836		8082	ND		high. Samples reported below detection.
OR3836		8151	ND		
OR3836		8310	ND ND		
OR3836		METALS	ND ND		
0.10000		WILTALS	ND	·	
OR3879	GW	8081	ND	·	Aldrin DDD high Complex
		0001			Aldrin RPD high. Samples reported below detection.
OR3879		8082	ND		Delow detection.
OR3879		8151	ND		Dalapon, Dicamba, 2,4-D,2,4,5-TP
					silvex, 2,4-DB RPD high. Samples
					reported below detection.
OR3879		8310	ND		Toponiou Dolow Goldonom.
OR3879		METALS	ND		
OR3883	GW	8081	ND		Aldrin RPD high. Samples reported below detection.
OR3883		8082	ND		
OR3883		8151	ND		
OR3883		8310	ND		
OR3883		METALS	ND		
OR3889		0004			
OR3889	GW	8081 8082	ND		
OR3889		8151	ND		
0113009	1	0131	ND		All parameters RPD high. Samples
OR3889		8310	ND		reported below detection.
		00.0			Acenaphthene RPd high. Samples reported below detection.
OR3889		METALS	ND		reported below detection.
OR3890	GW	8081	ND		
OR3890		8082	ND		
OR3890		8151	ND		All parameters RPD high. Samples reported below detection.
OR3890		8310	ND		Acenaphthene RPD high. Samples reported below detection.
OR3890		METALS	ND		The second desired in
OR3891	GW	8081	ND		
OR3891		8082	ND		

Table G-4. (cont.).

	NA - dia	Amaluaia	Lab Black	Currente	Matrix Spike Relative Percent
Lab Report	Media	Analysis	Lab Blank	Surrogate	Difference
Number		Method	Results	Recovery	Dalapon, Dicamba, 2,4-D, 2,4,5-TP
OR3891		8151	ND		silvex, 2,4-DB RPD high. Samples
					reported below detection.
00001		0010	ND		Acenaphthene RPD high. Samples
OR3891	į	8310	טא		reported below detection.
0.50001		METALO	ND		reported below detection.
OR3891		METALS	ND		
		0004	ND		
OR3916	GW	8081	ND		
OR3916		8082	ND ND		All agreements DDD high Comples
OR3916		8151	ND		All parameters RPD high. Samples reported below detection.
0.0010		0010	ND		reported below detection.
OR3916		8310	ND		
OR3916		METALS	ND_	ļ. <u> </u>	
OR3955	GW	8081	ND		
OR3955		8082	ND		Di di DDD bish Ossanles seconded
OR3955		8151	ND		Dieldrin RPD high. Samples reported below detection.
OR3955		8310	ND		
OR3955		METALS	ND		
OR3962	GW	8081	ND		
OR3962		8082	ND		
OR3962		8151	ND		Dalapon RPD high. Samples reported below detection.
OR3962		8310	ND		
OR3962		METALS	ND		Calcium RPD high. Samples reported
					below detection.
OR4013	GW	8081	ND	HIGH	
OR4013		8082	ND		
OR4013	T	8151	ND		
OR4013		8310	ND		
OR4013		METALS	ND	I	
OR4456	GW	8081	ND	HIGH	
OR4456	 ~~~	8082	ND		
OR4456	 	8151	ND		
OR4456	 	8310	ND		Naphthalene, Acenaphthene RPD high.
					Samples reported below detection.
OR4456	ļ	METALS	ND		
	 	9551			
OR4986	SW	8081	ND_		
OR4986	<u> </u>	8082	ND	-	
OR4986	ļ	8151	ND		
OR4986		8310	ND	<u> </u>	The state of the s
OR4986	1	METALS	lead, zinc	1	Lead, Sodium, and Zinc RPD high.
			<u> </u>		Sample value reported below detection.

Table G-4. (cont.).

Lab Report	Media	Analysis	Lab Blank	Surrogate	Matrix Spike Relative Percent
Number		Method	Results		
OR5062	GW	8081	ND		
OR5062		8082	ND		
OR5062		8151	ND		
OR5062		8310	ND		
OR5062		METALS	ND		

Table G-5. Summary laboratory spike recovery data expressed as a percent. Recoveries ranked as Excellent = all values within control limits, Good > 90% within control limits, Fair > 80% within control limits and Poor < 80% within control limits.

Parameter	Comment
2-(2,4,5-Trichlorophenoxy)propionic	Poor spike recovery. Six values below 55% and one value
acid (2,4,5 -TP) (Silvex)	above 145%
2,4,5,6-TCMX	Fair spike recovery. Fifteen values below 55% and two
	values above 145%
2,4-Dichlorophenoxy acetic acid (2,4 -	Fair spike recover. Nine values below 55% and one value
D)	above 145%
2,4-DCAA	Fair spike recovery. Twenty five values below 55% and five
	values above 145%. Method biased low approximately 10%
4-(2,4-Dichlorophenoxy)butyric acid	Poor spike recovery. Thirteen values below 55% and one
(2,4 - DB)	above 145%
4,4' - DDT	Good spike recovery. One value above 145%
Acenaphthene	Good spike recover but method biased low approximately
	25%
Aldrin	Excellent.
Aluminum	Poor spike recovery. Nine values above 135%. Matrix
	interferences cited.
Antimony	Good spike recovery but method biased high approximately
	10%
Arsenic (as carcinogen)	Excellent
Barium	Excellent
Benzo(a)pyrene	Excellent spike recovery but method biased low
	approximately 10%
Benzo(g,h,i)perylene	Excellent spike recovery but method biased low
	approximately 10%
Beryllium	Excellent. Method biased high approximately 10%
Cadmium	Excellent
Calcium	Fair spike recovery. One value below 65% and five values
	above 135%. Matrix interferences cited.
Cobalt	Excellent
Copper	Excellent
Dalapon	Fair spike recovery. Five values below 55% and four values
	above 145%
DBC	Fair spike recovery. Two values below 55% and thirteen
	values above 145%
Dicamba	Fair spike recovery. Four values below 55% and three values
	above 145%
Endrin	Good spike recovery
gamma - BHC (Lindane)	Good spike recovery. One value above 145%
Heptachlor	Good spike recovery. Two values below 55%
Iron	Fair spike recovery. Four values below 65% and four values
	above 145%. Matrix interference cited.
Lead	Excellent
Magnesium	Good spike recovery. One value below 65% and one value
	above 150 %. Matrix interference cited.
Manganese	Good spike recovery. One value below 55%.

Table G-5. (cont.).

Parameter	Comment
Mercury (inorganic)	Excellent.
Naphthalene	Good spike recovery. Four values below 55%. Method biased low approximately 25%
Nickel	Excellent
p-terphenyl	Excellent
PCB-1221	Good spike recovery. One value below 55% and one value above 145%.
Potassium	Fair spike recovery. Eight values above 145%. Matrix interference cited.
Selenium	Excellent
Silver	Excellent
Sodium	Good spike recovery. One value below 65%
Thallium	Excellent
Total organic carbon	Excellent
/anadium	Excellent
Zinc	Excellent

Table G-6. Summary of chain of custody records and sample holding time information.

Report Number OR2472	Sampled	Date Submitted	Date Reported	Holding	Signed	
OR2472		00000	neported	Time		Custody Intact
	06/01/98	06/02/98	08/04/98	OK	YES	YES
OR2473	06/01/98	06/02/98	06/26/98	OK	YES	YES
OR2524	06/05/98	06/05/98	06/26/98	OK	YES	YES
OR2525	06/05/98	06/05/98	08/04/98	OK	YES	YES
OR2540	06/08/98	06/09/98	07/14/98	OK	YES	YES
OR2559	06/09/98	06/10/98	07/20/98	OK	YES	YES
OR2575	06/10/98	06/11/98	07/06/98	ОК	YES	YES
OR2576	06/10/98	06/11/98	07/09/98	OK	YES	YES
OR2592	06/11/98	06/12/98	07/06/98	OK	YES	YES
OR2593	06/11/98	06/12/98	07/16/98	OK	YES	YES
OR2613	06/12/98	06/12/98	07/15/98	OK	YES	YES
OR2633	06/16/98	06/17/98	07/17/98	OK	YES	YES
OR2642	06/17/98	06/18/98	08/04/98	OK	YES	YES
OR2644	06/17/98	06/18/98	07/17/98	OK	YES	NO
OR2645	06/17/98	06/18/98	07/21/98	OK	YES	YES
OR2665	06/18/98	06/19/99	07/21/98	OK	YES	YES
OR2693	06/22/98	06/23/98	07/22/98	ОК	YES	YES
OR2725	06/24/98	06/25/98	07/29/98	ОК	YES	YES
OR2726	06/24/98	06/25/98	07/22/98	OK	YES	YES
OR2742	06/25/98	06/26/98	08/13/98	ОК	YES	YES
OR2743	06/25/98	06/26/98	07/23/98	ОК	YES	YES
OR2763	06/26/98	06/27/98	07/23/98	ОК	YES	YES
OR2779	06/29/98	06/30/98	07/23/98	OK	YES	YES
OR2784	06/30/98	07/01/98	07/30/98	ОК	YES	YES
OR2836	07/06/98	07/06/98	08/04/98	OK	YES	YES
OR2843	07/07/98	07/08/98	08/07/98	ОК	YES	YES
OR2859	07/08/98	07/09/98	08/31/98	ОК	YES	YES
OR2923	07/14/98	07/14/98	08/07/98	OK	YES	YES
OR2927	07/15/98	07/15/98	08/19/98	ОК	YES	YES
OR2985	07/17/98	07/18/98	08/10/98	OK	YES	YES
OR2986	07/17/98	07/18/98	08/19/98	OK	YES	YES
OR3023	07/22/98	07/22/98	08/11/98	ОК	YES	YES
OR3024	07/22/98	07/22/98	08/24/98	ОК	YES	YES
OR3069	07/27/98	07/27/98	08/28/98	OK	YES	YES
OR3082	07/28/98	07/28/98	08/28/98	ОК	YES	YES
OR3251	08/12/98	08/13/98	09/18/98	ОК	YES	NO
OR3285	08/13/98	08/17/98	09/18/98	OK	YES	YES
OR3362	08/20/98	08/21/98	09/18/98	OK	YES	YES
OR3363	08/21/98	08/21/98	09/22/98	OK	YES	YES
OR3429	08/28/98	08/28/98	09/21/98	OK	YES	NO
OR3491	08/03/98	09/03/98	09/25/98	ОК	YES	YES
OR3492	08/03/98	09/03/98	09/23/98	ОК	YES	YES
OR3564	09/13/98	09/14/98	09/28/98	OK	YES	YES

Table G-6. (cont.).

Ponet	Doto	1		T		
Report Number	Date	Date	Date	Holding	Signed	Chain of
	Sampled	Submitted		Time		Custody Intact
OR3584	09/15/98	09/16/98	09/28/98	OK	YES	YES
OR3629	09/18/98	09/19/98	09/30/98	OK	YES	YES
OR3634	09/19/98	09/21/98	09/30/98	OK	YES	YES
OR3635	09/20/98	09/21/98	09/30/98	OK	YES	YES
OR3660	09/21/98	09/23/98	10/01/98	ОК	YES	YES
OR3681	09/22/98	09/24/98	10/01/98	ОК	YES	YES
OR3706	missing	09/25/98	10/08/98	OK	YES	NO
OR3717	09/27/98	09/29/98	10/09/98	OK	YES	YES
OR3718	09/27/98	09/29/98	10/14/98	OK	YES	YES
OR3737	09/29/98	09/29/98	10/21/98	OK	YES	YES
OR3835	10/07/98	10/08/98	10/23/98	ОК	YES	YES
OR3836	10/06/98	10/08/98	10/23/98	ОК	YES	YES
OR3879	10/09/98	10/10/98	10/27/98	OK	YES	YES
OR3883	10/08/98	10/09/98	10/22/98	ОК	YES	YES
OR3889	10/11/98	10/12/98	11/02/98	ОК	YES	YES
OR3890	10/10/98	10/12/98	11/05/98	ОК	YES	YES
OR3891	10/11/98	10/12/98	11/03/98	ОК	YES	YES
OR3916	10/14/98	10/15/98	11/09/98	ОК	YES	YES
OR3955	10/16/98	10/17/98	11/03/98	ОК	YES	YES
OR3962	10/17/98	10/19/98	10/30/98	ОК	YES	YES
OR4013	10/21/98	10/22/98	11/13/98	ОК	YES	YES
OR4456	11/27/98	11/28/98	12/18/98	ОК	YES	YES
OR4986	01/06/99	01/07/99	01/22/99	ОК	YES	YES
OR5062	01/12/99	01/13/99	01/29/99	ОК	YES	YES

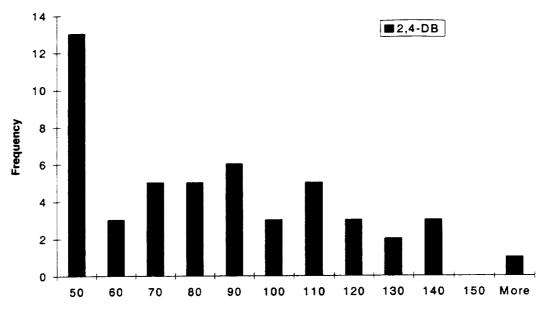


Figure G.1. Spike recovery frequency histogram for 2,4,-DB.

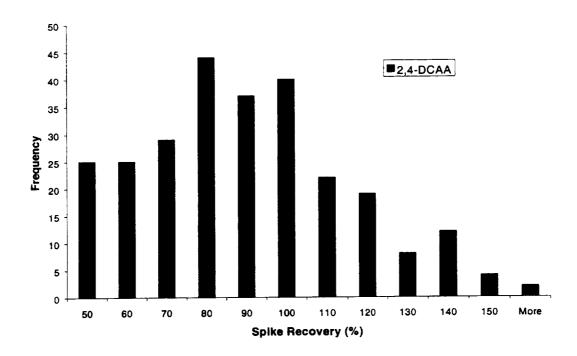


Figure G.2. Spike recovery frequency histogram for 2,4,-DCAA

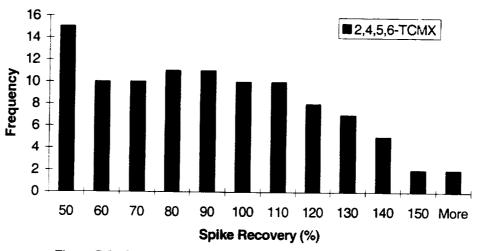


Figure G.3. Spike recovery frequency histogram for 2,4,5,6-TCMX..

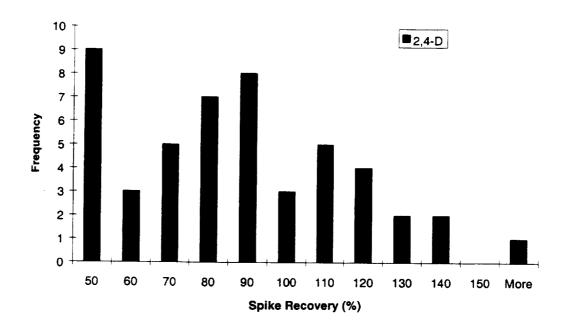


Figure G.4. Spike recovery frequency histogram for 2,4-D.

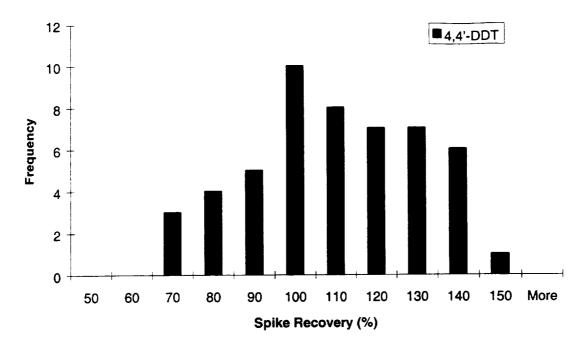


Figure G.5. Spike recovery frequency histogram for 4,4'-DDT.

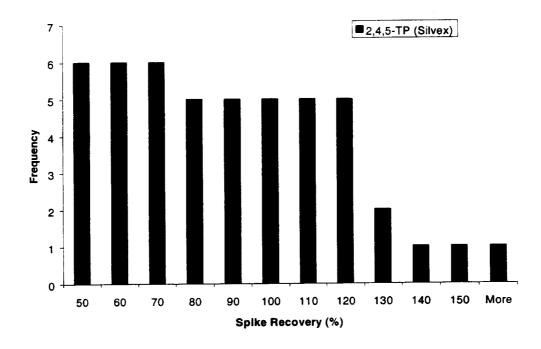


Figure G.6. Spike recovery frequency histogram for 2,4,5-TP (Silvex).

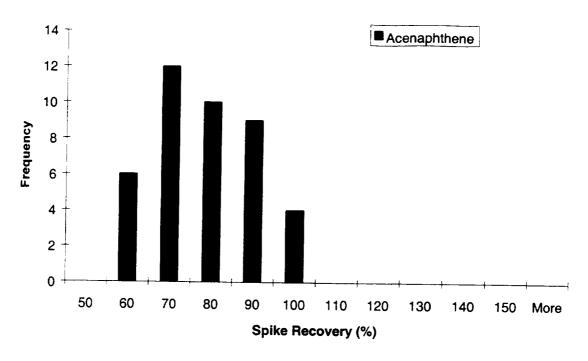


Figure G.7. Spike recovery frequency histogram for Acenaphthene.

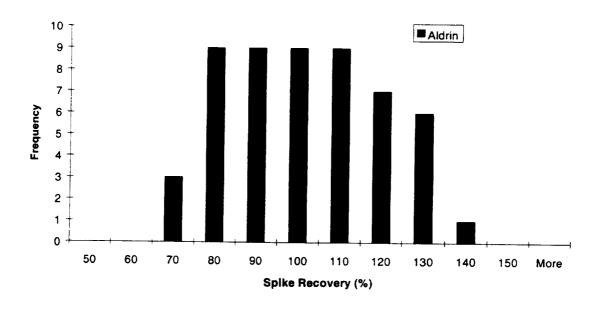


Figure G.8. Spike recovery frequency histogram for Aldrin.

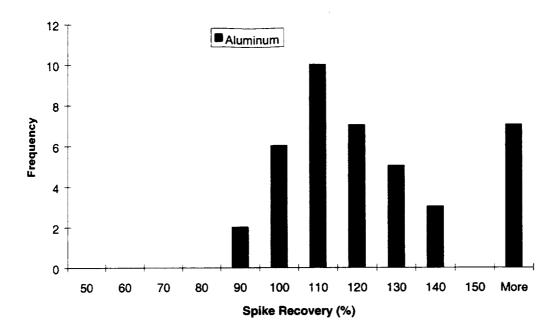


Figure G.9. Spike recovery frequency histogram for Aluminum.

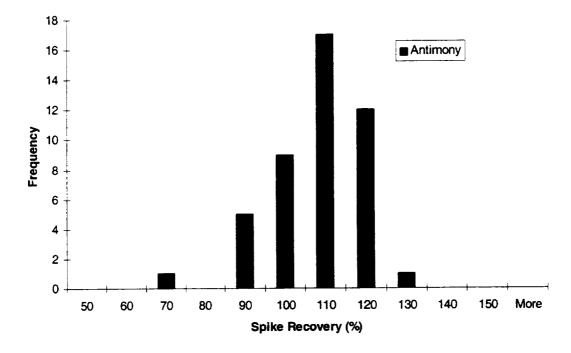


Figure G.10. Spike recovery frequency histogram for Antimony.

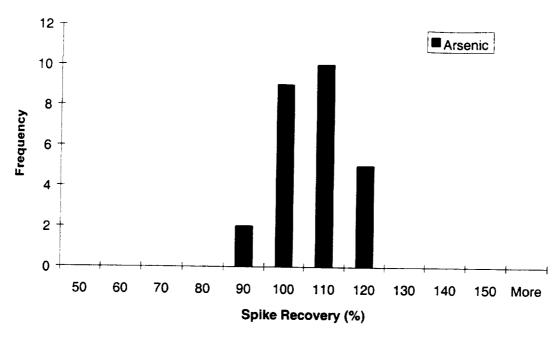


Figure G.11. Spike recovery frequency histogram for Arsenic.

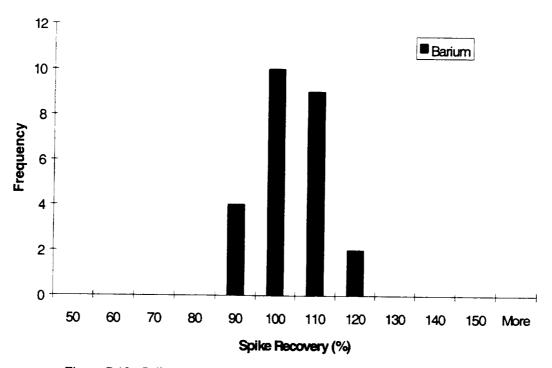


Figure G.12. Spike recovery frequency histogram for Barium.

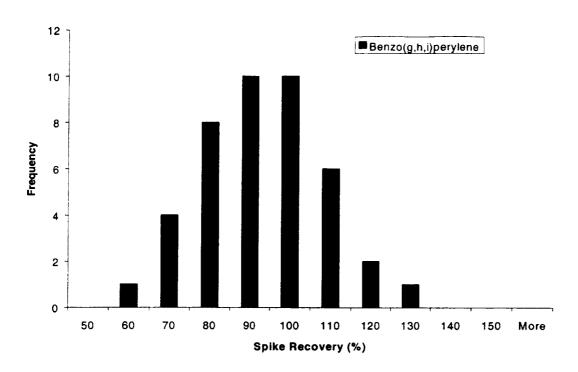


Figure G.13. Spike recovery frequency histogram for Benzo(g,h,i)perylene.

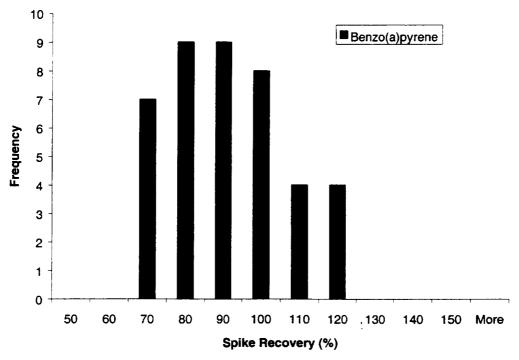


Figure G.14. Spike recovery frequency histogram for Benzo(a)pyrene.

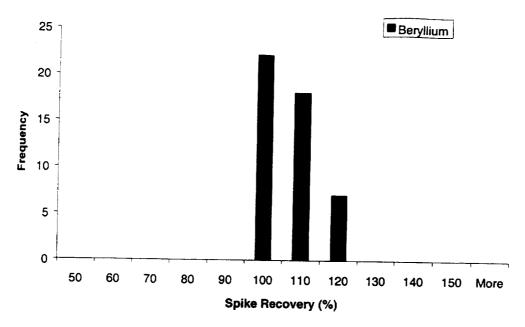


Figure G.15. Spike recovery frequency histogram for Berylium.

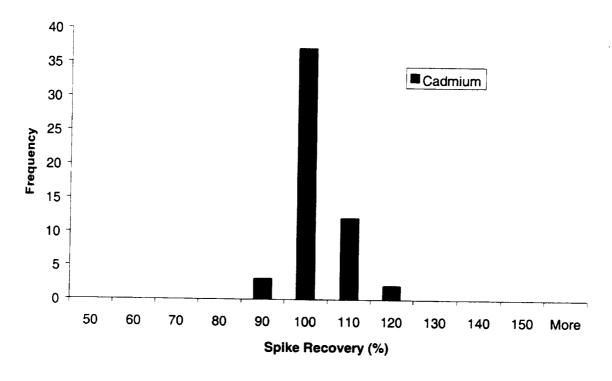


Figure G.16. Spike recovery frequency histogram for Cadmium.

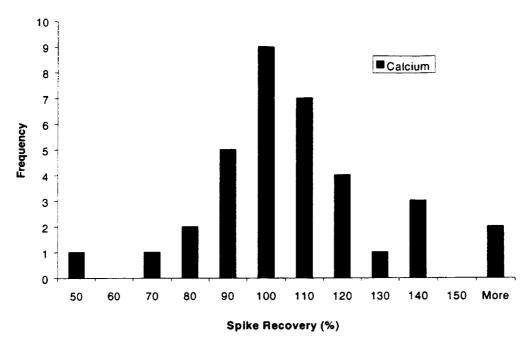


Figure G.17. Spike recovery frequency histogram for Calcium.

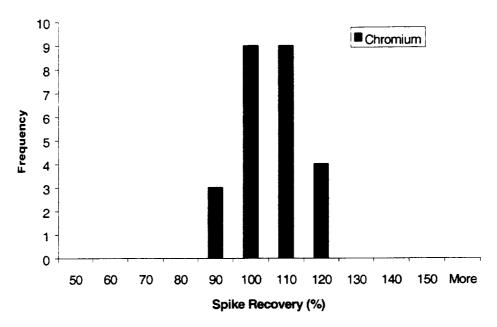


Figure G.18. Spike recovery frequency histogram for Chromium.

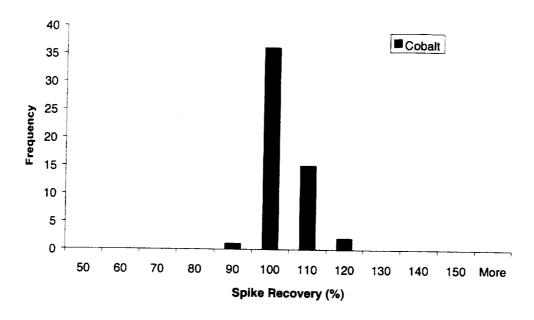


Figure G.19. Spike recovery frequency histogram for Cobalt.

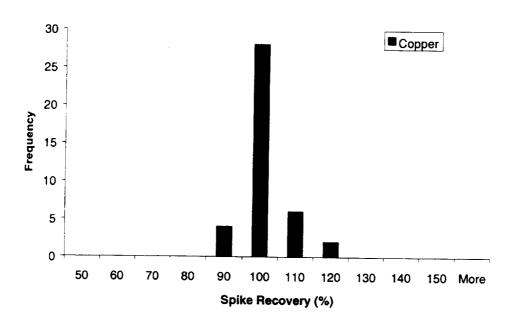


Figure G.20. Spike recovery frequency histogram for Copper.

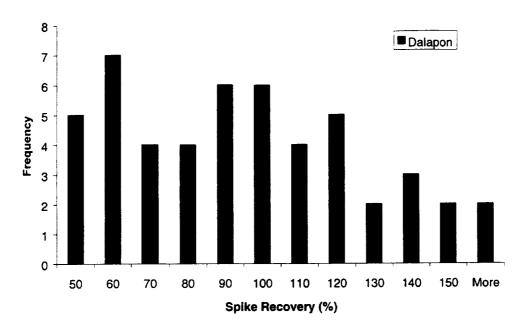


Figure G.21. Spike recovery frequency histogram for Dalapon.

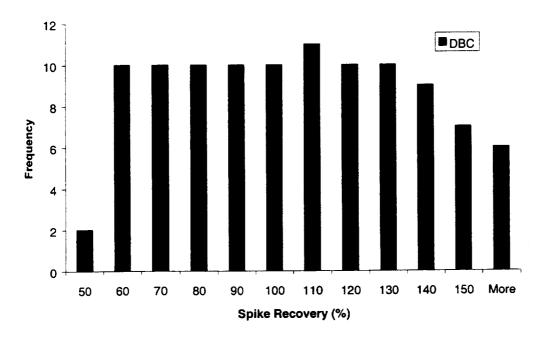


Figure G.22. Spike recovery frequency histogram for DBC.

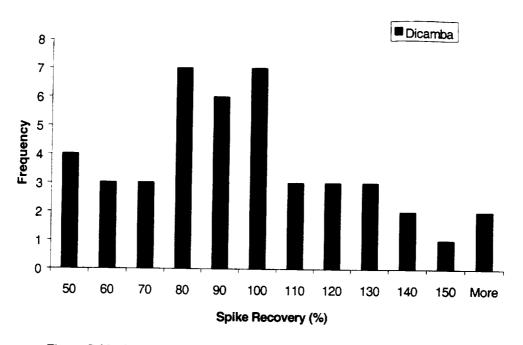


Figure G.23. Spike recovery frequency histogram for Dicamba.

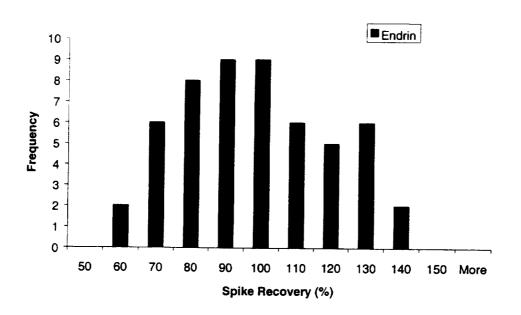


Figure G.24. Spike recovery frequency histogram for Endrin.

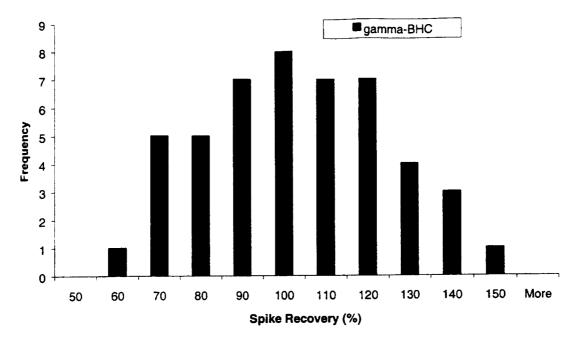


Figure G.25. Spike recovery frequency histogram for gamma-BHC.

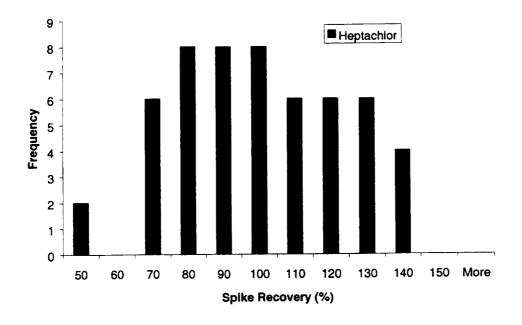


Figure G.26. Spike recovery frequency histogram for Heptachlor.

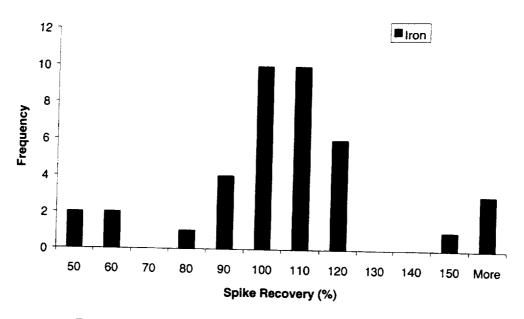


Figure G.27. Spike recovery frequency histogram for Iron.

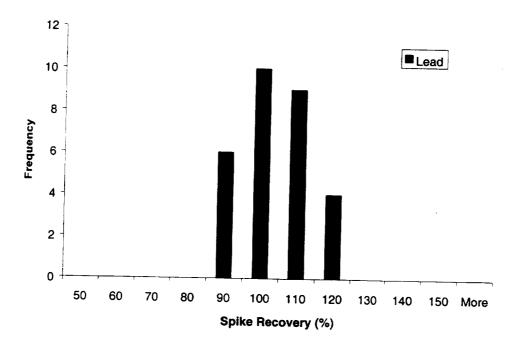


Figure G.28. Spike recovery frequency histogram for Lead.

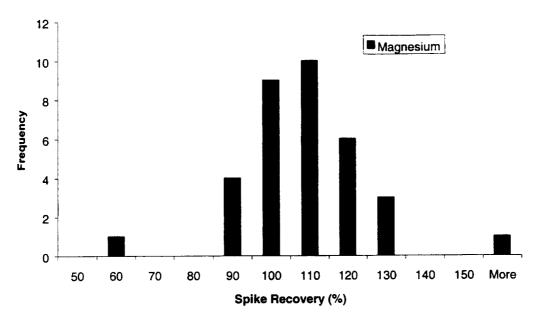


Figure G.29. Spike recovery frequency histogram for Magnesium.

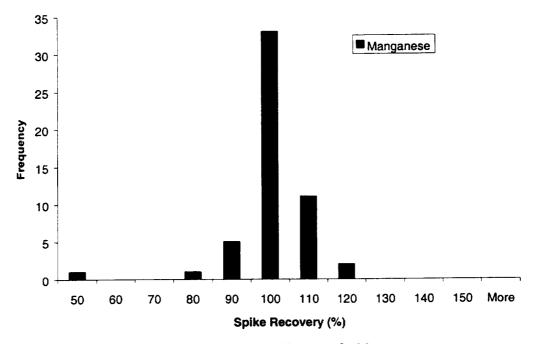


Figure G.30. Spike recovery frequency histogram for Manganese.

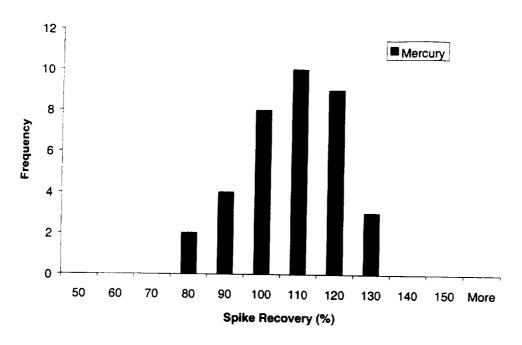


Figure G.31. Spike recovery frequency histogram for Mercury.

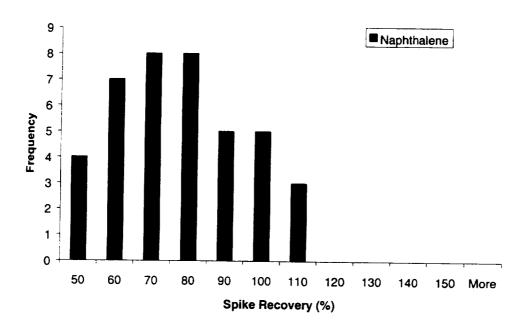


Figure G.32. Spike recovery frequency histogram for Napthalene.

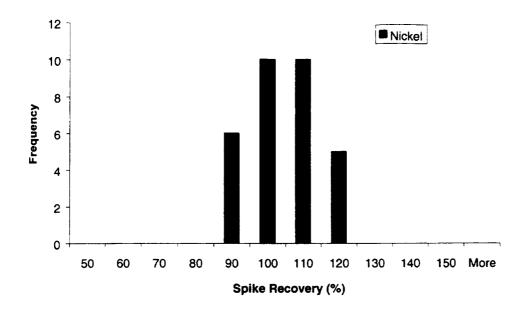


Figure G.33. Spike recovery frequency histogram for Nickel.

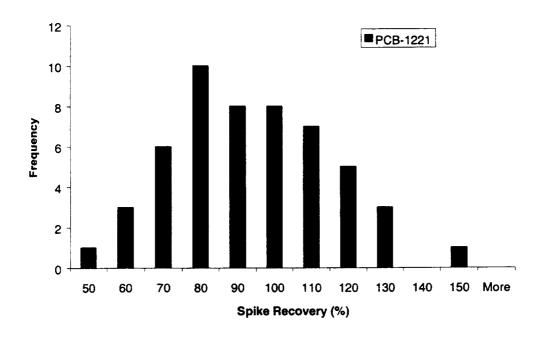


Figure G.34. Spike recovery frequency histogram for PCB-1221.

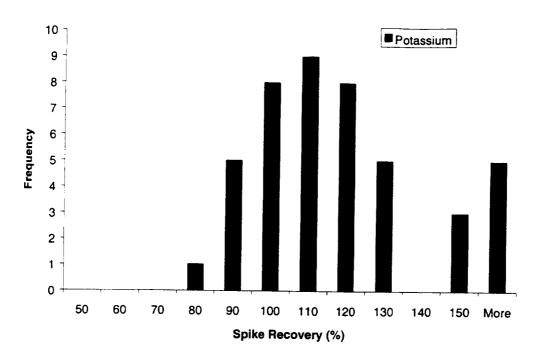


Figure G.35. Spike recovery frequency histogram for Potassium.

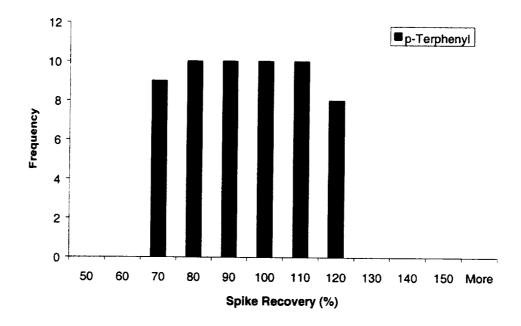


Figure G.36. Spike recovery frequency histogram for p-Terphenyl.

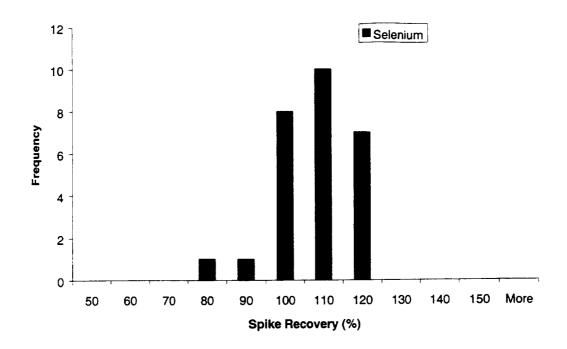


Figure G.37. Spike recovery frequency histogram for Selenium.

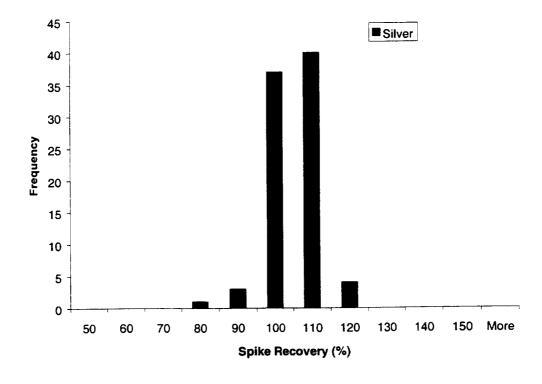


Figure G. 38. Spike recovery frequency histogram for Silver.

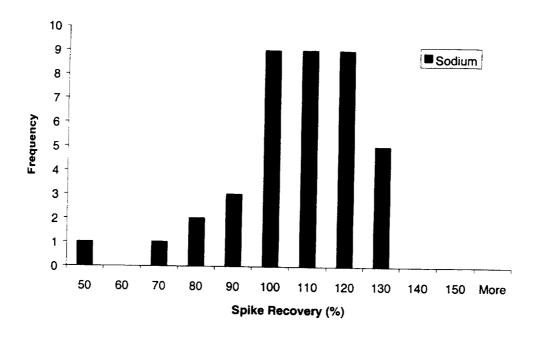


Figure G.39. Spike recovery frequency histogram for Sodium.

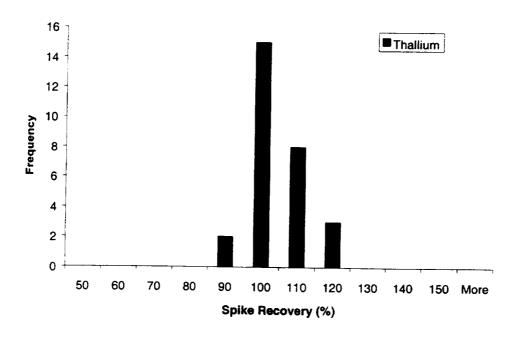


Figure G.40. Spike recovery frequency histogram for Thallium.

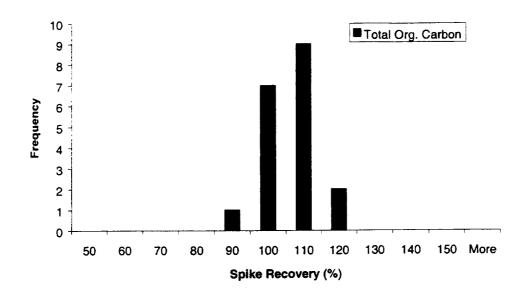


Figure G.41. Spike recovery frequency histogram for Total Organic Carbon.

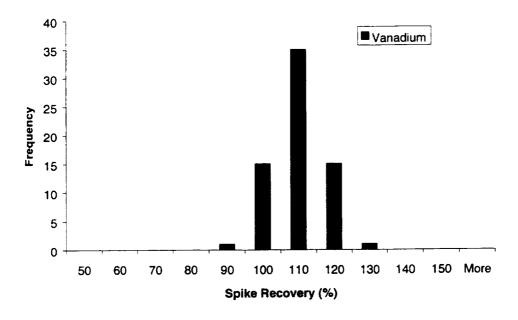


Figure G.42. Spike recovery frequency histogram for Vanadium.

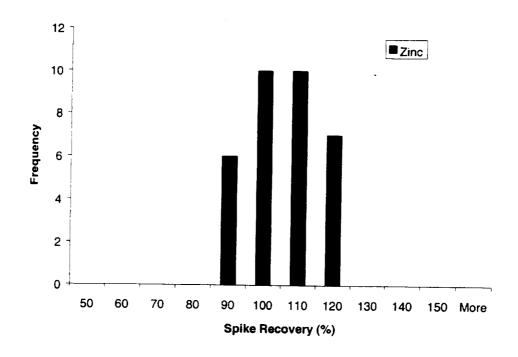


Figure G.43. Spike recovery frequency histogram for Zinc.

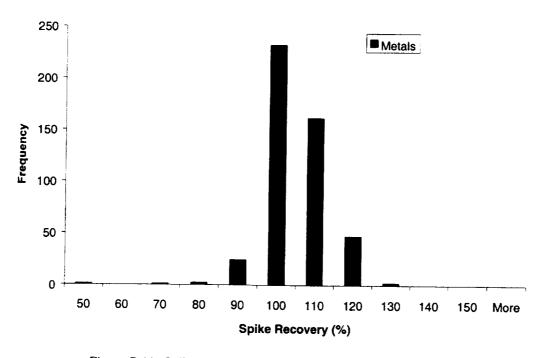


Figure G.44. Spike recovery frequency histogram for Metals.

REPORT DOCUMENTATION PAGE

Form Approved
OMB No. 0704-0188

Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden. to Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Artington, VA 22202-4302, and to the Office of Management and Budget, Paperwork Reduction Project (0704-0188), Washington, DC 20503.

1. AGENCY USE ONLY (Leave bla		3. REPORT TYPE AND DATE	COVERED
1. AGENCI UJE ONEI (LEGITE DIG	June 2000		
4. TITLE AND SUBTITLE	1 June 2000	5. FUN	DING NUMBERS
Kennedy Space Center	urface Water, and Sedi , Florida: Background	ments of Chemical and	
Physical Characteris 6. AUTHOR(S)	tics		
1	lissa A. Hensley, Mari Colleen A. Dunlevy	o Mota,	
7. PERFORMING ORGANIZATION	NAME(S) AND ADDRESS(ES)		FORMING ORGANIZATION
Dynamac Corporation		REP	ORT NUMBER
Mail Code DYN-2 Kennedy Space Center	, Florida 32899		
9. SPONSORING/MONITORING AC	ENCY NAME(S) AND ADDRESS(ES)		ONSORING / MONITORING
NASA		AG	ENCY REPORT NUMBER
Mail Code JJ Kennedy Space Center	r, Florida 32899	NA:	SA/TM
44 611001 5415117402 310756			
11. SUPPLEMENTARY NOTES			
Background, Chemical Soil, Surface Water	Characterization, Gro	oundwater, Metals, Org	ganics, Sediment,
12a. DISTRIBUTION / AVAILABILITY	STATEMENT	12b. D	ISTRIBUTION CODE
13. ABSTRACT (Maximum 200 wor	rds)		
This study documented	d background chemical o	composition of soils,	groundwater, surface
water, and sediments	of Kennedy Space Center of 10 soil classes. Fit	er. Two hundred soll	samples were
in A subsquifers of	the Surficial Aquifer a	and sampled: there we	re 24 shallow, 16
intermediate, and 11	deep wells. Forty sur	face water and sedim	ent samples were
collected in major wa	atershed basins. All s	samples were away fro	n sites of known
contamination. Samp	les were analyzed for o	organochlorine pestic	ides, aroclors,
chlorinated herbicide	es, polycyclic aromatic	hydrocarbons (PAH),	media. Some
other parameters. A	ll aroclors (6) were be cides were detected at	elow detection in all	in soil, sediment.
organochiorine pesti	Chlorinated herbicides	were detected at ver	v low frequencies in
soil and sediments.		requencies in soil, s	hallow groundwater,
surface water, and se	ediments. Concentration	ons of some metals di	ffered among soil
classes, with subagu	ifers and depths, and a	among watershed basin	s for surface water
	Most of the variation		ns was natural, but
	eased Cr, Cu, Mn, and	Zn.	15. NUMBER OF PAGES
14. SUBJECT TERMS			595
			16. PRICE CODE
AT COCHOTA CLASSICATION	10 CECUDITY CLASCICICATION	19. SECURITY CLASSIFICATION	20. LIMITATION OF ABSTRACT
17. SECURITY CLASSIFICATION OF REPORT	18. SECURITY CLASSIFICATION OF THIS PAGE	OF ABSTRACT	Lo. Chimira ilon or restrict
Unclassified	Unclassified	Unclassified	

GENERAL INSTRUCTIONS FOR COMPLETING SF 298

The Report Documentation Page (RDP) is used in announcing and cataloging reports. It is important that this information be consistent with the rest of the report, particularly the cover and title page. Instructions for filling in each block of the form follow. It is important to stay within the lines to meet optical scanning requirements.

- Block 1. Agency Use Only (Leave blank).
- Block 2. Report Date. Full publication date including day, month, and year, if available (e.g. 1 Jan 88). Must cite at least the year.
- Block 3. Type of Report and Dates Covered. State whether report is interim, final, etc. If applicable, enter inclusive report dates (e.g. 10 Jun 87 30 Jun 88).
- Block 4. <u>Title and Subtitle</u>. A title is taken from the part of the report that provides the most meaningful and complete information. When a report is prepared in more than one volume, repeat the primary title, add volume number, and include subtitle for the specific volume. On classified documents enter the title classification in parentheses.
- Block 5. Funding Numbers. To include contract and grant numbers; may include program element number(s), project number(s), task number(s), and work unit number(s). Use the following labels:

C - Contract PR - Project
G - Grant TA - Task
PE - Program WU - Work Unit
Element Accession No.

- Block 6. <u>Author(s)</u>. Name(s) of person(s) responsible for writing the report, performing the research, or credited with the content of the report. If editor or compiler, this should follow the name(s).
- Block 7. <u>Performing Organization Name(s) and Address(es)</u>. Self-explanatory.
- Block 8. <u>Performing Organization Report</u>
 <u>Number</u>. Enter the unique alphanumeric report
 number(s) assigned by the organization
 performing the report.
- Block 9. <u>Sponsoring/Monitoring Agency Name(s)</u> and <u>Address(es)</u>. Self-explanatory.
- Block 10. <u>Sponsoring/Monitoring Agency</u> Report Number. (If known)
- Block 11. Supplementary Notes. Enter information not included elsewhere such as: Prepared in cooperation with...; Trans. of...; To be published in.... When a report is revised, include a statement whether the new report supersedes or supplements the older report.

Block 12a. <u>Distribution/Availability Statement</u>. Denotes public availability or limitations. Cite any availability to the public. Enter additional limitations or special markings in all capitals (e.g. NOFORN, REL, ITAR).

DOD - See DoDD 5230.24, "Distribution Statements on Technical Documents."

DOE - See authorities.

NASA - See Handbook NHB 2200.2.

NTIS - Leave blank.

Block 12b. Distribution Code.

DOD - Leave blank.

DOE - Enter DOE distribution categories from the Standard Distribution for Unclassified Scientific and Technical Reports.

NASA - Leave blank. NTIS - Leave blank.

- Block 13. Abstract Include a brief (Maximum 200 words) factual summary of the most significant information contained in the report.
- Block 14. <u>Subject Terms</u>. Keywords or phrases identifying major subjects in the report.
- Block 15. <u>Number of Pages</u>. Enter the total number of pages.
- Block 16. <u>Price Code</u>. Enter appropriate price code (NTIS only).
- Blocks 17. 19. <u>Security Classifications</u>. Self-explanatory. Enter U.S. Security Classification in accordance with U.S. Security Regulations (i.e., UNCLASSIFIED). If form contains classified information, stamp classification on the top and bottom of the page.
- Block 20. <u>Limitation of Abstract</u>. This block must be completed to assign a limitation to the abstract. Enter either UL (unlimited) or SAR (same as report). An entry in this block is necessary if the abstract is to be limited. If blank, the abstract is assumed to be unlimited.

****** · *******

		· ···································
	i.	